Supporting Information

Binuclear Ni catalyzed ethylene copolymerization with short chain alkenol monomers

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1. Materials and Methods

Toluene, *n*-hexane, dichloromethane (DCM), and tetrahydrofuran (THF) were purified by MBraun SPS-800 system. Polymerization-grade ethylene was purified through an ethylene purification system (developed by the Dalian Institute of Chemical Physics, CAS). Allyl alcohol (A-ol), 3-buten-1-ol (HAA), 4-penten-1-ol (PA), 9-decen-1-ol (DA) were dried over anhydrous CaSO₄ and distilled. Allyl acetate (AAc) was dried over CaH₂ and distilled. Diethylaluminum chloride (DEAC) were purchased from Jiangsu Yongjian Chemical Co., Ltd as neat liquids and diluted with toluene to 1.0 M solutions before use. Poly(E-*co*-HAA), poly(E-*co*-PA) and poly(E-*co*-DA) were acetylated before GPC measurement. **1a** [1], **1b** [1] and **2a** [2] were known and synthesized according to literature. Compounds **3a** and **4a** were reported in our previous work [3].

All air and moisture-sensitive manipulations were carried out under high purity N₂ using standard Schlenk techniques or in a glovebox. ¹H NMR, ¹³C NMR, DEPT 135, COSY, HSQC and HMBC spectra were recorded on an Agilent Technologies spectrometer (400 MHz, 600 MHz) or a Varian spectrometer (400 MHz). Elemental analysis was performed by the Analytical Laboratory of the Shanghai Institute of Organic Chemistry (CAS). M_n , M_w , and dispersity index (Đ) were determined with Agilent Technologies PL-GPC 220 High temperature Gel Permeation Chromatography at 150 °C (polystyrene calibration, 1,2,4-trichlorobenzene as solvent at 150°C). X-Ray crystallographic data were collected using a Bruker AXSD8 X-ray diffractometer. Mass spectra were obtained using an HP5989A spectrometer.

2. Synthesis of compounds

(1) Synthesis of **3b** and **4b**



To a solution of compound **1b** (7.7 g, 25.0 mmol) in DCM (200 mL) was added a solution of mCPBA (10.2 g, 50.0 mmol) in DCM (200 mL). The reaction was monitored by TLC. Saturated NaHCO₃(aq) was added after complete conversion of the substrate (about 10 min). The organic phase was washed with saturated NaHCO₃(aq) two more times, dried over Na₂SO₄, filtered,

concentrated under reduced pressure, and purified by flash chromatography to afford crude **2b** as an orange oil (not stable), which was used in the next step without further purification.

To a solution of the above crude **2b** in 100 mL of anhydrous THF was added a MeMgBr solution (7.7 mL, 23.1 mmol, 3.0 M in Et₂O) at 0 °C. The reaction mixture was warmed to room temperature and stirred for 2 h being quenched with saturated NH₄Cl (aq). The organic phase was extracted with ethyl acetate three times. The combined organic layers were washed with brine, dried over Na₂SO₄, filtered, and concentrated under reduced pressure. The crude product was recrystallized from MeOH repeatedly until product **3b** was obtained as a white solid (1.4 g, 17%, two steps). ¹H NMR (400 MHz, CDCl₃) δ 7.05-7.03 (m, 4 H), 6.94-6.91 (m, 2 H), 6.16 (s, 1 H), 2.02 (s, 6 H), 1.99 (s, 6 H), 1.82 (s, 6 H), 1.80 (s, 3 H). ¹³C NMR (101 MHz, CDCl₃) δ 173.14, 147.39, 128.28, 128.24, 125.98, 125.66, 123.49, 80.27, 24.31, 18.21, 18.15, 15.74. HR-MS (ESI) [M+Na]⁺ Calcd for C₂₂H₂₈N₂ONa, 359.20938; Found, 359.20999.



Figure S1 ¹H NMR spectrum (400 MHz) of 3b in CDCl₃.



3b



(2) Synthesis of **3c** and **4c**



To a solution of compound **2a** (6.5 g, 15.1 mmol) in 75 mL of anhydrous THF was added an EtMgBr solution (25.0 mL, 75.5 mmol, 3.0 M in Et₂O) at 0 °C. The reaction mixture was warmed to room temperature and stirred overnight before being quenched with saturated NH₄Cl(aq). The organic phase was extracted with ethyl acetate three times. The combined organic layers were washed with brine, dried over Na₂SO₄, filtered, and concentrated under reduced pressure. The product and unreacted substrate were isolated by flash chromatography. The recovered substrate (4.0 g, 61%) was processed in the same way as above to obtain more products. The products of two batches were combined and recrystallized from MeOH to afford the product **3c** as a white solid (1.4 g, 20% overall two batches). ¹H NMR (400 MHz, CDCl₃) δ 7.16-7.14 (m, 4 H), 7.11-7.07 (m, 2H), 6.13 (s, 1 H), 2.69 (hep, *J* = 7.1 Hz, 4 H), 2.38 (q, *J* = 7.2 Hz, 2H), 1.84 (s, 6H), 1.17-1.12 (m, 24H), 1.05 (t, *J* = 7.2 Hz, 3H). ¹³C NMR (101 MHz, CDCl₃) δ 173.00, 145.13, 136.28, 136.09, 124.12, 123.20, 123.16, 83.50, 29.68, 28.19, 28.16, 23.64, 23.53, 23.20, 22.78, 16.16, 7.84. **HR-MS (ESI)** [M+H]⁺ Calcd for C₃₁H₄₇ON₂, 463.3683; Found, 463.3682.



230 220 210 200 190 180 170 160 150 140 130 120 110 100 90 80 70 60 50 40 30 20 10 0 -10 δppm

Figure S4¹³C NMR spectrum (400 MHz) of 3c in CDCl₃.



1. KH (2 equiv), THF, r.t., overnight 2. NiBr₂(DME) (2 equiv), THF, r.t.,12 h



3c

To a suspension of potassium hydride (KH) (0.08 g, 2.0 mmol, 2.0 equiv.) in 10 mL of dry THF was added a solution of **3c** (0.46 g, 1.0 mmol) in 10 mL of THF at room temperature. The resulting suspension was stirred overnight. It was filtered into a suspension of NiBr₂(DME) (0.62 g, 2.0 mmol, 2.0 equiv.) in THF at room temperature, and the suspension was stirred for 12 h. The solvent was removed under reduced pressure, and the residue was dissolved in dry DCM. After filtration, the filtrate was concentrated under reduced pressure, and the residue was dissolved in THF for recrystallization to afford the product **4c** (yellow solid, 0.63 g, 65%). Anal. Calcd. for $C_{35}H_{53}Br_3N_2Ni_2O_2$ (one THF coordination): C, 47.19; H, 6.00; N, 3.14; Found: C, 47.34; H, 6.18; N, 3.15.

(3) Synthesis of **3d** and **4d**



To a solution of compound **2b** (5.9 g, 18.3 mmol) in 100 mL of anhydrous THF was added an ^{*i*}PrMgBr solution (33.0 mL, 91.5 mmol, 2.8 M in 2-methyltetrahydrofuran) at 0 °C. The reaction mixture was warmed to room temperature and stirred overnight before being quenched with saturated NH₄Cl(aq). The organic phase was extracted with ethyl acetate three times. The combined organic layers were washed with brine, dried over Na₂SO₄, filtered, and concentrated under reduced pressure. The crude product was recrystallized from MeOH to afford the product **3b** as a white solid (1.9 g, 29%). ¹**H** NMR (400 MHz, CDCl₃) δ 7.05-7.02 (m, 4 H), 6.94-6.91 (m, 2 H), 6.01 (s, 1 H), 3.09 (hep, *J* = 6.6 Hz, 1 H), 2.04 (s, 6 H), 1.99 (s, 6 H), 1.88 (s, 6 H), 1.09 (d, *J* = 6.6 Hz, 6 H). ¹³C NMR (101 MHz, CDCl₃) δ 173.81, 147.64, 128.43, 128.37, 126.14, 125.82, 123.52, 85.58, 33.28, 19.00, 18.89, 17.73, 16.64. **HR-MS (ESI)** [M+Na]⁺ Calcd for C₂₄H₃₂N₂ONa, 387.24068; Found, 387.24036.



Figure S6¹³C NMR spectrum (400 MHz) of 3d in CDCl₃.



1. KH (2 equiv), THF, r.t., overnight 2. NiBr₂(DME) (2 equiv), THF, r.t., 12 h



3d

To a suspension of potassium hydride (KH) (0.08 g, 2.0 mmol, 2.0 equiv.) in 10 mL of dry THF was added a solution of **3d** (0.36 g, 1.0 mmol) in 10 mL of THF at room temperature. The resulting suspension was stirred overnight. It was filtered into a suspension of NiBr₂(DME) (0.62 g, 2.0 mmol, 2.0 equiv.) in THF at room temperature, and the suspension was stirred for 12 h. The solvent was removed under reduced pressure, and the residue was dissolved in dry DCM. After filtration, the filtrate was concentrated under reduced pressure, and the residue was dissolved in THF for recrystallization to afford the product **4d** (red solid, 0.40 g, 46%). Anal. Calcd. for $C_{32}H_{47}Br_3N_2Ni_2O_3$ (two THF coordination): C, 44.44; H, 5.48; N, 3.24; Found: C, 44.43; H, 5.57; N, 3.38.

3. General Procedure for Olefin Copolymerizations with Polar Monomer

Under 1 atm of ethylene, a flame-dried two-necked flask equipped with a stir bar was charged with the desired amount of toluene and DEAC solution at desired temperature. After 10 min, the polar monomer was added dropwise. The mixture was stirred for 10 min, and the desired amount of Ni catalyst in toluene was added to initiate polymerization. The reaction was quenched by pouring it into 250 mL of acidified methanol (concentrated HCl/MeOH, 1/10, v/v) and stirred at room temperature overnight. The precipitated polymer was collected, washed with methanol, and dried under vacuum at 60 °C to a constant weight.

4. Acetylation Procedure for Poly(E-co-Alkenol)

To a round-bottom flask equipped with a stir bar was added the poly(E-*co*-alkenol), 20 mL of 1,2-dichlorobenzene, 5 mL of glacial acetic acid, 5 drops of concentrated H_2SO_4 sequentially. A reflux condenser was installed, and the mixture was heated to 120 °C for 6 h. The hot solution was poured into 100 mL of methanol, and an additional 10 mL of acidified methanol (concentrated HCl/MeOH, 1/10, v/v) was added to precipitate the copolymer. The precipitated copolymer was filtered, washed with methanol, dried under vacuum at 60 °C to a constant weight, and further characterized by ¹H NMR, confirming the complete conversion of hydroxyl to acetate groups.

5. Supplementary Notes

1) NMR Assignments of Poly(E-co-HAA) and Comonomer Content Calculation



Figure S7. ¹H NMR spectrum (600 MHz) of poly(E-*co*-HAA) (Entry 8, Table 2) in 1,1,2,2-tetrachloroethane- d_2 at 120 °C.

The comonomer content was calculated from quantitative ¹H NMR spectrum analysis . The mole ratios of HAA and ethylene in the copolymer are x and 1-x, respectively. The integration $I_{b+b'+c+c'+d+e+f}$ is set as n, and the integration $I_{a+a'}$ is set as 1.

$$\frac{2x}{5x+4(1-x)} = \frac{1}{n}$$
$$x = \frac{4}{2n-1} \times 100\%$$



Figure S8. ¹³C NMR spectrum (150 MHz) of poly(E-*co*-HAA) (Entry 8, Table 2) in 1,1,2,2-tetrachloroethane- d_2 at 120 °C.

The signals of ¹H and ¹³C NMR are assigned according to our previous work [3].

2) NMR Assignments of Poly(E-co-PA) and Comonomer Content Calculation



Figure S9. ¹H NMR spectrum (600 MHz) of poly(E-*co*-PA) (Entry 12, Table 2) in 1,1,2,2-tetrachloroethane- d_2 at 120 °C.

The comonomer content was calculated from quantitative ¹H NMR spectrum analysis. The mole ratios of PA and ethylene in the copolymer are x and 1-x, respectively. The integration $I_{b+b'+c+c'+d+e}$ is set as n, and the integration $I_{a+a'}$ is set as 1.

$$\frac{2x}{7x + 4(1 - x)} = \frac{1}{n}$$
$$x = \frac{4}{2n - 3} \times 100\%$$



Figure S10. ¹³C NMR spectrum (150 MHz) of poly(E-*co*-PA) (Entry 12, Table 2) in 1,1,2,2-tetrachloroethane- d_2 at 120 °C.

The signals of ¹H and ¹³C NMR are assigned according to our previous work [3].

3) NMR Assignments of Poly(E-co-DA) and Comonomer Content Calculation



Figure S11. ¹H NMR spectrum (600 MHz) of poly(E-*co*-DA) (Entry 16, Table 2) in 1,1,2,2-tetrachloroethane- d_2 at 120 °C.

The comonomer content was calculated from quantitative ¹H NMR spectrum analysis. The mole ratios of DA and ethylene in the copolymer are x and 1-x, respectively. The integration $I_{b+c+d+e}$ is set as n, and the integration I_a is set as 1.

$$\frac{2x}{17x + 4(1 - x)} = \frac{1}{n}$$
$$x = \frac{4}{2n - 13} \times 100\%$$



Figure S12. ¹³C NMR spectrum (150 MHz) of poly(E-*co*-DA) (Entry 16, Table 2) in 1,1,2,2-tetrachloroethane- d_2 at 120 °C.

The signals of ¹H and ¹³C NMR are assigned according to our previous work [3].

4) Microstructure Analysis of Poly(E-co-A-ol) and Comonomer Content Calculation

According to quantitative ¹H, ¹³C and 2D NMR (Figures S24-S29), as well as literature [3-6], the resulting poly(E-*co*-A-ol) exhibits microstructures as shown in the following scheme. Olefinic groups [5, 7, 8] and aromatic groups [9-12] were also detected in ¹H NMR, which were assigned according to literature.



Scheme S1. Microstructure of Poly(E-co-A-ol)



Figure S13. ¹H NMR spectrum (600 MHz) of poly(E-*co*-A-ol) (Entry 3, Table 2) in 1,1,2,2-tetrachloroethane- d_2 at 120 °C.

The comonomer content was calculated from quantitative ¹H NMR spectrum analysis. The mole ratios of A-ol and ethylene in the copolymer are x and 1-x, respectively. The integration $I_{b+b'+c+c'+d+e}$ is set as n, and the integration $I_{a+a'}$ is set as 1. Note that the olefinic and aromatic groups probably from β -OAl elimination were not taken into quantitative calculation.

$$\frac{2x}{3x+4(1-x)} = \frac{1}{n}$$
$$x = \frac{4}{2n+1} \times 100\%$$



Figure S14. ¹³C NMR spectrum (150 MHz) of poly(E-*co*-A-ol) (Entry 3, Table 2) in 1,1,2,2-tetrachloroethane- d_2 at 120 °C.



Figure S15. DEPT 135 spectrum of poly(E-*co*-A-ol) (Entry 3, Table 2) in 1,1,2,2-tetrachloroethane- d_2 at 120 °C.



Figure S16. COSY spectrum of poly(E-*co*-A-ol) (Entry 3, Table 2) in 1,1,2,2-tetrachloroethane-*d*₂ at 120 °C.



Figure S17. HSQC spectrum of poly(E-*co*-A-ol) (Entry 3, Table 2) in 1,1,2,2-tetrachloroethane-*d*₂ at 120 °C.



Figure S18. HMBC spectrum of poly(E-*co*-A-ol) (Entry 3, Table 2) in 1,1,2,2-tetrachloroethane d_2 at 120 °C.

5) NMR Assignments of Poly(E-co-AAc) and Double Bond Content Calculation



Figure S19. ¹H NMR spectrum (600 MHz) of poly(E-*co*-AAc) (Entry 2, Table 4) (little -OAc content but double bond detected) in 1,1,2,2-tetrachloroethane- d_2 at 120 °C.

Negligible peaks assigned to the -OAc group were observed in the ¹H NMR spectra of the poly(E-*co*-AAc) samples. The double bond content was calculated from quantitative ¹H NMR spectrum analysis. The mole ratios of double bond and ethylene in the copolymer are x and 1-x, respectively. The integration $I_{b+c+e+h}$ is set as n; the integration I_{d+f} is set as 1, and the integration I_f is set as m.

$$\frac{2x}{5x+4(1-x)} = \frac{1}{n+1+m/2}$$
$$x = \frac{4}{2n+m+1} \times 100\%$$

Note that the aromatic groups were also detected in ¹H NMR but were not taken into quantitative calculation. The double bond content calculated based on the equation above in Table 4 ranged from 0.2~0.5 mol%, which could be expressed as trace content, so their quantitative analysis was not pursued either.

6) Branching Analysis from ¹H NMR

The amount of methyl groups ($N_{Me groups/1000C}$) based on ¹H NMR spectra were analyzed according to literature [7, 8].

6. Polymer Characterizations



Figure S20. ¹H NMR spectrum (600 MHz) of poly(E-*co*-HAA) (Entry 2, Table 1) in 1,1,2,2-tetrachloroethane- d_2 at 120 °C.



Figure S21. ¹H NMR spectrum (600 MHz) of poly(E-*co*-HAA) (Entry 3, Table 1) in 1,1,2,2-tetrachloroethane- d_2 at 120 °C.



Figure S22. ¹H NMR spectrum (600 MHz) of poly(E-*co*-HAA) (Entry 4, Table 1 & Entry 6, Table 2) in 1,1,2,2-tetrachloroethane-*d*₂ at 120 °C.



Figure S23. ¹H NMR spectrum (600 MHz) of poly(E-*co*-HAA) (Entry 5, Table 1) in 1,1,2,2-tetrachloroethane- d_2 at 120 °C.



Figure S24. ¹H NMR spectrum (600 MHz) of poly(E-*co*-HAA) (Entry 6, Table 1) in 1,1,2,2-tetrachloroethane- d_2 at 120 °C.



Figure S25. ¹H NMR spectrum (600 MHz) of poly(E-*co*-HAA) (Entry 7, Table 1) in 1,1,2,2-tetrachloroethane- d_2 at 120 °C.



Figure S26. ¹H NMR spectrum (600 MHz) of poly(E-*co*-HAA) (Entry 8, Table 1) in 1,1,2,2-tetrachloroethane- d_2 at 120 °C.



Figure S27. ¹H NMR spectrum (600 MHz) of poly(E-*co*-HAA) (Entry 9, Table 1) in 1,1,2,2-tetrachloroethane- d_2 at 120 °C.



Figure S28. ¹H NMR spectrum (600 MHz) of poly(E-*co*-HAA) (Entry 10, Table 1) in 1,1,2,2-tetrachloroethane- d_2 at 120 °C.



Figure S29. ¹H NMR spectrum (600 MHz) of poly(E-*co*-HAA) (Entry 11, Table 1) in 1,1,2,2-tetrachloroethane- d_2 at 120 °C.



Figure S30. ¹H NMR spectrum (600 MHz) of poly(E-*co*-HAA) (Entry 12, Table 1) in 1,1,2,2-tetrachloroethane- d_2 at 120 °C.



Figure S31. ¹H NMR spectrum (600 MHz) of poly(E-*co*-HAA) (Entry 13, Table 1) in 1,1,2,2-tetrachloroethane- d_2 at 120 °C.



Figure S32. ¹H NMR spectrum (600 MHz) of poly(E-*co*-HAA) (Entry 14, Table 1) in 1,1,2,2-tetrachloroethane- d_2 at 120 °C.



Figure S33. ¹H NMR spectrum (600 MHz) of poly(E-*co*-HAA) (Entry 15, Table 1) in 1,1,2,2-tetrachloroethane- d_2 at 120 °C.



Figure S34. ¹H NMR spectrum (600 MHz) of poly(E-*co*-HAA) (Entry 16, Table 1) in 1,1,2,2-tetrachloroethane- d_2 at 120 °C.



Figure S35. ¹H NMR spectrum (600 MHz) of poly(E-*co*-HAA) (Entry 7, Table 2) in 1,1,2,2-tetrachloroethane- d_2 at 120 °C.



Figure S36. ¹H NMR spectrum (600 MHz) of poly(E-*co*-HAA) (Entry 9, Table 2) in 1,1,2,2-tetrachloroethane- d_2 at 120 °C.

2) NMR Spectra of Poly(E-co-PA)



Figure S37. ¹H NMR spectrum (600 MHz) of poly(E-*co*-PA) (Entry 10, Table 2) in 1,1,2,2-tetrachloroethane- d_2 at 120 °C.



Figure S38. ¹H NMR spectrum (600 MHz) of poly(E-*co*-PA) (Entry 11, Table 2) in 1,1,2,2-tetrachloroethane- d_2 at 120 °C.



Figure S39. ¹H NMR spectrum (600 MHz) of poly(E-*co*-PA) (Entry 13, Table 2) in 1,1,2,2-tetrachloroethane- d_2 at 120 °C.

3) NMR Spectra of Poly(E-co-DA)



Figure S40. ¹H NMR spectrum (600 MHz) of poly(E-*co*-DA) (Entry 14, Table 2) in 1,1,2,2-tetrachloroethane- d_2 at 120 °C.



Figure S41. ¹H NMR spectrum (600 MHz) of poly(E-*co*-DA) (Entry 15, Table 2) in 1,1,2,2-tetrachloroethane- d_2 at 120 °C.



Figure S42. ¹H NMR spectrum (600 MHz) of poly(E-*co*-DA) (Entry 17, Table 2) in 1,1,2,2-tetrachloroethane- d_2 at 120 °C.

4) NMR Spectra of Poly(E-co-A-ol)



Figure S43. ¹H NMR spectrum (600 MHz) of poly(E-*co*-A-ol) (Entry 1, Table 2) in 1,1,2,2-tetrachloroethane- d_2 at 120 °C.



Figure S44. ¹H NMR spectrum (600 MHz) of poly(E-*co*-A-ol) (Entry 2, Table 2) in 1,1,2,2-tetrachloroethane- d_2 at 120 °C.



Figure S45. ¹H NMR spectrum (600 MHz) of poly(E-*co*-A-ol) (Entry 4, Table 2) in 1,1,2,2-tetrachloroethane- d_2 at 120 °C.


Figure S46. ¹H NMR spectrum (600 MHz) of poly(E-*co*-A-ol) (Entry 5, Table 2) in 1,1,2,2-tetrachloroethane- d_2 at 120 °C.

5) NMR Spectra of Poly(E-co-AAc)



Figure S47. ¹H NMR spectrum (600 MHz) of poly(E-*co*-AAc) (Entry 1, Table 4) (little -OAc content but double bond detected) in 1,1,2,2-tetrachloroethane- d_2 at 120 °C.



Figure S48. ¹H NMR spectrum (600 MHz) of poly(E-*co*-AAc) (Entry 3, Table 4) (no -OAc incorporation but double bond detected) in 1,1,2,2-tetrachloroethane- d_2 at 120 °C.



Figure S49. ¹H NMR spectrum (600 MHz) of poly(E-*co*-AAc) (Entry 4, Table 4) (no -OAc incorporation but double bond detected) in 1,1,2,2-tetrachloroethane- d_2 at 120 °C.



Figure S50. ¹H NMR spectrum (600 MHz) of poly(E-*co*-AAc) (Entry 5, Table 4) (no -OAc incorporation but double bond detected) in 1,1,2,2-tetrachloroethane- d_2 at 120 °C.



Figure S51. ¹H NMR spectrum (600 MHz) of poly(E-*co*-AAc) (Entry 6, Table 4) (no -OAc incorporation but double bond detected) in 1,1,2,2-tetrachloroethane- d_2 at 120 °C.



Figure S52. ¹H NMR spectrum (600 MHz) of poly(E-*co*-AAc) (Entry 7, Table 4) (no -OAc incorporation but double bond detected) in 1,1,2,2-tetrachloroethane- d_2 at 120 °C.





Figure S53. ¹H NMR spectrum (600 MHz) of poly(E-*co*-HAA) (Entry 8, Table 2) after acetylation in 1,1,2,2-tetrachloroethane- d_2 at 120 °C.



Figure S54. ¹H NMR spectrum (600 MHz) of poly(E-*co*-PA) (Entry 12, Table 2) after acetylation



Figure S55. ¹H NMR spectrum (600 MHz) of poly(E-*co*-DA) (Entry 16, Table 2) after acetylation in 1,1,2,2-tetrachloroethane- d_2 at 120 °C.

7) GPC Traces of Polymer Samples



Figure S56. GPC trace of poly(E-co-HAA) (Entry 2, Table 1) after acetylation.



Figure S57. GPC trace of poly(E-co-HAA) (Entry 3, Table 1) after acetylation.



Figure S58. GPC trace of poly(E-co-HAA) (Entry 4, Table 1; Entry 6, Table 2) after acetylation.



Figure S59. GPC trace of poly(E-co-HAA) (Entry 5, Table 1) after acetylation.



Figure S60. GPC trace of poly(E-co-HAA) (Entry 6, Table 1) after acetylation.



Figure S61. GPC trace of poly(E-co-HAA) (Entry 7, Table 1) after acetylation.



Figure S62. GPC trace of poly(E-co-HAA) (Entry 8, Table 1) after acetylation.



Figure S63. GPC trace of poly(E-co-HAA) (Entry 9, Table 1) after acetylation.



Figure S64. GPC trace of poly(E-co-HAA) (Entry 10, Table 1) after acetylation.



Figure S65. GPC trace of poly(E-co-HAA) (Entry 11, Table 1) after acetylation.



Figure S66. GPC trace of poly(E-co-HAA) (Entry 12, Table 1) after acetylation.



Figure S67. GPC trace of poly(E-co-HAA) (Entry 13, Table 1) after acetylation.



Figure S68. GPC trace of poly(E-co-HAA) (Entry 14, Table 1) after acetylation.



Figure S69. GPC trace of poly(E-co-HAA) (Entry 15, Table 1) after acetylation.



Figure S70. GPC trace of poly(E-co-HAA) (Entry 16, Table 1) after acetylation.



Figure S71. GPC trace of poly(E-co-A-ol) (Entry 1, Table 2).



Figure S72. GPC trace of poly(E-co-A-ol) (Entry 2, Table 2).



Figure S73. GPC trace of poly(E-co-A-ol) (Entry 3, Table 2).



Figure S74. GPC trace of poly(E-co-A-ol) (Entry 4, Table 2).



Figure S75. GPC trace of poly(E-co-A-ol) (Entry 5, Table 2).



Figure S76. GPC trace of poly(E-co-HAA) (Entry 7, Table 2) after acetylation.



Figure S77. GPC trace of poly(E-co-HAA) (Entry 8, Table 2) after acetylation.



Figure S78. GPC trace of poly(E-co-HAA) (Entry 9, Table 2) after acetylation.



Figure S79. GPC trace of poly(E-co-PA) (Entry 10, Table 2) after acetylation.



Figure S80(a). GPC trace of poly(E-co-PA) (Entry 11, Table 2) after acetylation.



Figure S80(b). DSC trace of poly(E-*co*-PA) (Entry 11, Table 2)



Figure S81. GPC trace of poly(E-co-PA) (Entry 12, Table 2) after acetylation.



Figure S82. GPC trace of poly(E-co-PA) (Entry 13, Table 2) after acetylation.



Figure S83. GPC trace of poly(E-co-DA) (Entry 14, Table 2) after acetylation.



Figure S84(a). GPC trace of poly(E-co-DA) (Entry 15, Table 2) after acetylation.



Figure S84(b). DSC trace of poly(E-co-DA) (Entry 15, Table 2)



Figure S85. GPC trace of poly(E-co-DA) (Entry 16, Table 2) after acetylation.



Figure S86. GPC trace of poly(E-co-DA) (Entry 17, Table 2) after acetylation.



Figure S87. GPC trace of poly(E-co-AAc) (Entry 1, Table 4).



Figure S88. GPC trace of poly(E-co-AAc) (Entry 2, Table 4).



Figure S89. GPC trace of poly(E-co-AAc) (Entry 3, Table 4).



Figure S90. GPC trace of poly(E-co-AAc) (Entry 4, Table 4).



Figure S91. GPC trace of poly(E-co-AAc) (Entry 5, Table 4).



Figure S92. GPC trace of poly(E-co-AAc) (Entry 6, Table 4).

Figure S93. GPC trace of poly(E-co-AAc) (Entry 7, Table 4).

7. X-ray Crystallography

Identification code	mo d8v18750 0m
Empirical formula	C39 H61 Br3 N2 Ni2 O3
Formula weight	963.04
Temperature	296(2) K
Wavelength	0.71073 Å
Crystal system	Monoclinic
Space group	P 21/c
Unit cell dimensions	a = 14.9658(9) Å
	b = 19.0661(9) Å
	c = 16.9642(9) Å
Volume	4786.8(4) Å ³
Ζ	4
Density (calculated)	1.336 Mg/m ³
Absorption coefficient	3.323 mm ⁻¹
F(000)	1976
Crystal size	0.200 x 0.170 x 0.130 mm ³
Theta range for data collection	2.541 to 26.000°.
Index ranges	-15<=h<=18, -23<=k<=23, -17<=l<=20
Reflections collected	23698
Independent reflections	9366 [R(int) = 0.0416]
Completeness to theta = 25.242°	99.6 %
Absorption correction	Semi-empirical from equivalents
Max. and min. transmission	0.7456 and 0.5430
Refinement method	Full-matrix least-squares on F ²
Data / restraints / parameters	9366 / 0 / 454
Goodness-of-fit on F ²	0.975
Final R indices [I>2sigma(I)]	R1 = 0.0453, WR2 = 0.1112
R indices (all data)	R1 = 0.0793, WR2 = 0.1222
Extinction coefficient	0.0026(3)
Largest diff. peak and hole	0.820 and -0.535 e.Å ⁻³

Table S1. Crystal data and structure refinement for $4 c \$

Table S2. Bond lengths [Å] for 4c

Ni(1)-O(1)	1.988(3)	С(17)-Н(17)	0.9800
Ni(1)-N(1)	2.031(3)	C(18)-H(18A)	0.9600
Ni(1)-O(2)	2.093(3)	C(18)-H(18B)	0.9600
Ni(1)-Br(1)	2.4350(6)	C(18)-H(18C)	0.9600
Ni(1)-Br(2)	2.5959(6)	C(19)-H(19A)	0.9600
Ni(2)-O(1)	1.979(2)	C(19)-H(19B)	0.9600
Ni(2)-N(2)	2.054(3)	C(19)-H(19C)	0.9600
Ni(2)-O(3)	2.106(3)	C(20)-C(25)	1.386(6)
Ni(2)-Br(3)	2.4058(7)	C(20)-C(21)	1.397(6)
Ni(2)-Br(2)	2.4609(7)	C(21)-C(22)	1.389(6)
N(1)-C(1)	1.286(5)	C(21)-C(29)	1.545(7)
N(1)-C(8)	1.456(5)	C(22)-C(23)	1.373(7)

N(2)-C(3)	1.283(5)	C(22)-H(22)	0.9300
N(2)-C(20)	1.445(5)	C(23)-C(24)	1.353(7)
O(1)-C(2)	1.396(4)	C(23)-H(23)	0.9300
O(2)-C(35)	1.433(6)	C(24)-C(25)	1.395(6)
O(2)-C(32)	1.452(6)	C(24)-H(24)	0.9300
O(3)-C(39)	1.440(5)	C(25)-C(26)	1.508(7)
O(3)-C(36)	1.446(5)	C(26)-C(27)	1.518(8)
C(1)-C(5)	1.514(6)	C(26)-C(28)	1.519(9)
C(1)-C(2)	1.537(6)	C(26)-H(26)	0.9800
C(2)-C(3)	1.518(6)	C(27)-H(27A)	0.9600
C(2)-C(6)	1.561(5)	C(27)-H(27B)	0.9600
C(3)-C(4)	1.509(6)	C(27)-H(27C)	0.9600
C(4)-H(4A)	0.9600	C(28)-H(28A)	0.9600
C(4)-H(4B)	0.9600	C(28)-H(28B)	0.9600
C(4)-H(4C)	0.9600	C(28)-H(28C)	0.9600
C(5)-H(5A)	0.9600	C(29)-C(31)	1.503(7)
C(5)-H(5B)	0.9600	C(29)-C(30)	1.521(7)
C(5)-H(5C)	0.9600	C(29)-H(29)	0.9800
C(6)-C(7)	1.500(6)	C(30)-H(30A)	0.9600
C(6)-H(6A)	0.9700	C(30)-H(30B)	0.9600
C(6)-H(6B)	0.9700	C(30)-H(30C)	0.9600
C(7)-H(7A)	0.9600	C(31)-H(31A)	0.9600
C(7)-H(7B)	0.9600	C(31)-H(31B)	0.9600
C(7)-H(7C)	0.9600	C(31)-H(31C)	0.9600
C(8)-C(13)	1.390(6)	C(32)-C(33)	1.507(8)
C(8)-C(9)	1.399(6)	C(32)-H(32A)	0.9700
C(9)-C(10)	1.389(6)	C(32)-H(32B)	0.9700
C(9)-C(17)	1.527(7)	C(33)-C(34)	1.428(9)
C(10)-C(11)	1.370(8)	C(33)-H(33A)	0.9700
C(10)-H(10)	0.9300	C(33)-H(33B)	0.9700
C(11)-C(12)	1.340(8)	C(34)-C(35)	1.477(8)
C(11)-H(11)	0.9300	C(34)-H(34A)	0.9700
C(12)-C(13)	1.403(7)	C(34)-H(34B)	0.9700
С(12)-Н(12)	0.9300	C(35)-H(35A)	0.9700
C(13)-C(14)	1.509(7)	C(35)-H(35B)	0.9700
C(14)-C(16)	1.507(7)	C(36)-C(37)	1.464(7)
C(14)-C(15)	1.546(8)	C(36)-H(36A)	0.9700
C(14)-H(14)	0.9800	C(36)-H(36B)	0.9700
C(15)-H(15A)	0.9600	C(37)-C(38)	1.482(8)
C(15)-H(15B)	0.9600	C(37)-H(37A)	0.9700
C(15)-H(15C)	0.9600	C(37)-H(37B)	0.9700
C(16)-H(16A)	0.9600	C(38)-C(39)	1.490(7)
C(16)-H(16B)	0.9600	C(38)-H(38A)	0.9700
С(16)-Н(16С)	0.9600	C(38)-H(38B)	0.9700
C(17)-C(19)	1.514(7)	C(39)-H(39A)	0.9700
C(17)-C(18)	1.542(7)	C(39)-H(39B)	0.9700

Table S3. Bond angles [°] for 4c

O(1)-Ni(1)-N(1)	81.74(12)	C(17)-C(18)-H(18A)	109.5
O(1)-Ni(1)-O(2)	170.52(11)	C(17)-C(18)-H(18B)	109.5
N(1)-Ni(1)-O(2)	100.29(13)	H(18A)-C(18)-H(18B)	109.5
O(1)-Ni(1)-Br(1)	96.05(7)	C(17)-C(18)-H(18C)	109.5
N(1)-Ni(1)-Br(1)	113.36(9)	H(18A)-C(18)-H(18C)	109.5
O(2)-Ni(1)-Br(1)	91.64(9)	H(18B)-C(18)-H(18C)	109.5
O(1)-Ni(1)-Br(2)	83.91(7)	C(17)-C(19)-H(19A)	109.5
N(1)-Ni(1)-Br(2)	97.27(9)	C(17)-C(19)-H(19B)	109.5
O(2)-Ni(1)-Br(2)	86.64(9)	H(19A)-C(19)-H(19B)	109.5
Br(1)-Ni(1)-Br(2)	149.10(3)	C(17)-C(19)-H(19C)	109.5
O(1)-Ni(2)-N(2)	79.45(11)	H(19A)-C(19)-H(19C)	109.5
O(1)-Ni(2)-O(3)	83.53(10)	H(19B)-C(19)-H(19C)	109.5
N(2)-Ni(2)-O(3)	154.10(12)	C(25)-C(20)-C(21)	122.6(4)
O(1)-Ni(2)-Br(3)	165.73(8)	C(25)-C(20)-N(2)	117.6(4)
N(2)-Ni(2)-Br(3)	99.32(9)	C(21)-C(20)-N(2)	119.8(4)
O(3)-Ni(2)-Br(3)	92.42(8)	C(22)-C(21)-C(20)	117.4(5)
O(1)-Ni(2)-Br(2)	87.79(8)	C(22)-C(21)-C(29)	119.2(4)
N(2)-Ni(2)-Br(2)	109.64(9)	C(20)-C(21)-C(29)	123.3(4)
O(3)-Ni(2)-Br(2)	88.91(9)	C(23)-C(22)-C(21)	121.1(5)
Br(3)-Ni(2)-Br(2)	105.85(3)	C(23)-C(22)-H(22)	119.5
Ni(2)-Br(2)-Ni(1)	75.60(2)	C(21)-C(22)-H(22)	119.5
C(1)-N(1)-C(8)	120.8(3)	C(24)-C(23)-C(22)	119.9(5)
C(1)-N(1)-Ni(1)	114.1(3)	C(24)-C(23)-H(23)	120.1
C(8)-N(1)-Ni(1)	125.1(3)	С(22)-С(23)-Н(23)	120.1
C(3)-N(2)-C(20)	119.3(3)	C(23)-C(24)-C(25)	122.4(5)
C(3)-N(2)-Ni(2)	115.4(3)	C(23)-C(24)-H(24)	118.8
C(20)-N(2)-Ni(2)	125.3(3)	C(25)-C(24)-H(24)	118.8
C(2)-O(1)-Ni(2)	117.4(2)	C(20)-C(25)-C(24)	116.5(5)
C(2)-O(1)-Ni(1)	115.3(2)	C(20)-C(25)-C(26)	122.9(4)
Ni(2)-O(1)-Ni(1)	102.88(11)	C(24)-C(25)-C(26)	120.6(4)
C(35)-O(2)-C(32)	108.9(4)	C(25)-C(26)-C(27)	112.1(5)
C(35)-O(2)-Ni(1)	117.9(3)	C(25)-C(26)-C(28)	110.2(5)
C(32)-O(2)-Ni(1)	122.0(3)	C(27)-C(26)-C(28)	112.9(5)
C(39)-O(3)-C(36)	109.0(3)	C(25)-C(26)-H(26)	107.1
C(39)-O(3)-Ni(2)	112.0(3)	C(27)-C(26)-H(26)	107.1
C(36)-O(3)-Ni(2)	124.5(3)	C(28)-C(26)-H(26)	107.1
N(1)-C(1)-C(5)	122.4(4)	C(26)-C(27)-H(27A)	109.5
N(1)-C(1)-C(2)	117.3(3)	C(26)-C(27)-H(27B)	109.5
C(5)-C(1)-C(2)	120.3(4)	H(27A)-C(27)-H(27B)	109.5
O(1)-C(2)-C(3)	109.3(3)	C(26)-C(27)-H(27C)	109.5
O(1)-C(2)-C(1)	110.2(3)	H(27A)-C(27)-H(27C)	109.5
C(3)-C(2)-C(1)	112.8(3)	H(27B)-C(27)-H(27C)	109.5
O(1)-C(2)-C(6)	107.8(3)	C(26)-C(28)-H(28A)	109.5
C(3)-C(2)-C(6)	111.7(3)	C(26)-C(28)-H(28B)	109.5
C(1)-C(2)-C(6)	105.0(3)	H(28A)-C(28)-H(28B)	109.5

N(2)-C(3)-C(4)	121.2(4)	C(26)-C(28)-H(28C)	109.5
N(2)-C(3)-C(2)	117.0(3)	H(28A)-C(28)-H(28C)	109.5
C(4)-C(3)-C(2)	121.7(4)	H(28B)-C(28)-H(28C)	109.5
C(3)-C(4)-H(4A)	109.5	C(31)-C(29)-C(30)	110.6(5)
C(3)-C(4)-H(4B)	109.5	C(31)-C(29)-C(21)	112.9(4)
H(4A)-C(4)-H(4B)	109.5	C(30)-C(29)-C(21)	110.7(5)
C(3)-C(4)-H(4C)	109.5	C(31)-C(29)-H(29)	107.5
H(4A)-C(4)-H(4C)	109.5	C(30)-C(29)-H(29)	107.5
H(4B)-C(4)-H(4C)	109.5	C(21)-C(29)-H(29)	107.5
C(1)-C(5)-H(5A)	109.5	C(29)-C(30)-H(30A)	109.5
C(1)-C(5)-H(5B)	109.5	C(29)-C(30)-H(30B)	109.5
H(5A)-C(5)-H(5B)	109.5	H(30A)-C(30)-H(30B)	109.5
C(1)-C(5)-H(5C)	109.5	C(29)-C(30)-H(30C)	109.5
H(5A)-C(5)-H(5C)	109.5	H(30A)-C(30)-H(30C)	109.5
H(5B)-C(5)-H(5C)	109.5	H(30B)-C(30)-H(30C)	109.5
C(7)-C(6)-C(2)	112.1(4)	C(29)-C(31)-H(31A)	109.5
C(7)-C(6)-H(6A)	109.2	C(29)-C(31)-H(31B)	109.5
C(2)-C(6)-H(6A)	109.2	H(31A)-C(31)-H(31B)	109.5
C(7)-C(6)-H(6B)	109.2	C(29)-C(31)-H(31C)	109.5
C(2)-C(6)-H(6B)	109.2	H(31A)-C(31)-H(31C)	109.5
H(6A)-C(6)-H(6B)	107.9	H(31B)-C(31)-H(31C)	109.5
C(6)-C(7)-H(7A)	109.5	O(2)-C(32)-C(33)	104.9(5)
C(6)-C(7)-H(7B)	109.5	O(2)-C(32)-H(32A)	110.8
H(7A)-C(7)-H(7B)	109.5	C(33)-C(32)-H(32A)	110.8
C(6)-C(7)-H(7C)	109.5	O(2)-C(32)-H(32B)	110.8
H(7A)-C(7)-H(7C)	109.5	C(33)-C(32)-H(32B)	110.8
H(7B)-C(7)-H(7C)	109.5	H(32A)-C(32)-H(32B)	108.8
C(13)-C(8)-C(9)	121.6(4)	C(34)-C(33)-C(32)	106.0(5)
C(13)-C(8)-N(1)	117.5(4)	C(34)-C(33)-H(33A)	110.5
C(9)-C(8)-N(1)	120.8(4)	C(32)-C(33)-H(33A)	110.5
C(10)-C(9)-C(8)	117.7(5)	C(34)-C(33)-H(33B)	110.5
C(10)-C(9)-C(17)	118.7(4)	C(32)-C(33)-H(33B)	110.5
C(8)-C(9)-C(17)	123.6(4)	H(33A)-C(33)-H(33B)	108.7
C(11)-C(10)-C(9)	120.6(5)	C(33)-C(34)-C(35)	106.6(5)
С(11)-С(10)-Н(10)	119.7	C(33)-C(34)-H(34A)	110.4
C(9)-C(10)-H(10)	119.7	C(35)-C(34)-H(34A)	110.4
C(12)-C(11)-C(10)	121.6(5)	C(33)-C(34)-H(34B)	110.4
C(12)-C(11)-H(11)	119.2	C(35)-C(34)-H(34B)	110.4
C(10)-C(11)-H(11)	119.2	H(34A)-C(34)-H(34B)	108.6
C(11)-C(12)-C(13)	120.7(5)	O(2)-C(35)-C(34)	107.4(5)
С(11)-С(12)-Н(12)	119.6	O(2)-C(35)-H(35A)	110.2
С(13)-С(12)-Н(12)	119.6	C(34)-C(35)-H(35A)	110.2
C(8)-C(13)-C(12)	117.8(5)	O(2)-C(35)-H(35B)	110.2
C(8)-C(13)-C(14)	122.3(4)	C(34)-C(35)-H(35B)	110.2
C(12)-C(13)-C(14)	119.8(5)	H(35A)-C(35)-H(35B)	108.5
C(16)-C(14)-C(13)	115.2(5)	O(3)-C(36)-C(37)	106.1(5)
C(16)-C(14)-C(15)	108.7(5)	O(3)-C(36)-H(36A)	110.5

C(13)-C(14)-C(15)	110.8(5)	C(37)-C(36)-H(36A)	110.5
C(16)-C(14)-H(14)	107.2	O(3)-C(36)-H(36B)	110.5
C(13)-C(14)-H(14)	107.2	C(37)-C(36)-H(36B)	110.5
C(15)-C(14)-H(14)	107.2	H(36A)-C(36)-H(36B)	108.7
С(14)-С(15)-Н(15А)	109.5	C(36)-C(37)-C(38)	104.5(5)
C(14)-C(15)-H(15B)	109.5	C(36)-C(37)-H(37A)	110.9
H(15A)-C(15)-H(15B)	109.5	C(38)-C(37)-H(37A)	110.9
С(14)-С(15)-Н(15С)	109.5	C(36)-C(37)-H(37B)	110.9
H(15A)-C(15)-H(15C)	109.5	C(38)-C(37)-H(37B)	110.9
H(15B)-C(15)-H(15C)	109.5	H(37A)-C(37)-H(37B)	108.9
C(14)-C(16)-H(16A)	109.5	C(37)-C(38)-C(39)	103.5(5)
C(14)-C(16)-H(16B)	109.5	C(37)-C(38)-H(38A)	111.1
H(16A)-C(16)-H(16B)	109.5	C(39)-C(38)-H(38A)	111.1
C(14)-C(16)-H(16C)	109.5	C(37)-C(38)-H(38B)	111.1
H(16A)-C(16)-H(16C)	109.5	C(39)-C(38)-H(38B)	111.1
H(16B)-C(16)-H(16C)	109.5	H(38A)-C(38)-H(38B)	109.0
C(19)-C(17)-C(9)	110.7(5)	O(3)-C(39)-C(38)	105.7(4)
C(19)-C(17)-C(18)	110.1(5)	O(3)-C(39)-H(39A)	110.6
C(9)-C(17)-C(18)	110.9(4)	C(38)-C(39)-H(39A)	110.6
С(19)-С(17)-Н(17)	108.4	O(3)-C(39)-H(39B)	110.6
C(9)-C(17)-H(17)	108.4	C(38)-C(39)-H(39B)	110.6
C(18)-C(17)-H(17)	108.4	H(39A)-C(39)-H(39B)	108.7

Table S4. Crystal data and structure refinement for $\mathbf{4d}$

Identification code	mj22149 0m
Empirical formula	C32 H47 Br3 N2 Ni2 O3
Formula weight	864.86
Temperature	235.00 K
Wavelength	1.34139 Å
Crystal system	Triclinic
Space group	P-1
Unit cell dimensions	a = 9.1461(2) Å
	b = 9.5974(2) Å
	c = 20.7405(5) Å
Volume	1758.26(7) Å ³
Ζ	2
Density (calculated)	1.634 Mg/m ³
Absorption coefficient	8.674 mm ⁻¹
F(000)	876
Crystal size	0.07 x 0.07 x 0.05 mm ³
Theta range for data collection	4.048 to 55.017°.
Index ranges	-10<=h<=11, -11<=k<=11, -25<=l<=25
Reflections collected	19462
Independent reflections	6668 [R(int) = 0.0605]
Completeness to theta = 53.594°	99.2 %

Absorption correction	Semi-empirical from equivalents
Max. and min. transmission	0.7508 and 0.4483
Refinement method	Full-matrix least-squares on F ²
Data / restraints / parameters	6668 / 0 / 387
Goodness-of-fit on F ²	1.079
Final R indices [I>2sigma(I)]	R1 = 0.0463, WR2 = 0.1273
R indices (all data)	R1 = 0.0492, wR2 = 0.1302
Extinction coefficient	n/a
Largest diff. peak and hole	0.752 and -1.088 e.Å ⁻³

Table S5. Bond lengths [Å] for 4d	
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Br(1)-Ni(1)	2.4215(5)	C(13)-C(14)	1.406(5)
Br(2)-Ni(1)	2.4574(5)	C(14)-C(15)	1.488(6)
Br(2)-Ni(2)	2.5337(5)	C(15)-H(15A)	0.9700
Br(3)-Ni(2)	2.4119(6)	C(15)-H(15B)	0.9700
Ni(1)-O(1)	1.979(2)	C(15)-H(15C)	0.9700
Ni(1)-O(2)	2.081(2)	C(16)-H(16A)	0.9700
Ni(1)-N(1)	2.040(3)	C(16)-H(16B)	0.9700
Ni(2)-O(1)	2.024(2)	C(16)-H(16C)	0.9700
Ni(2)-O(3)	2.137(2)	C(17)-C(18)	1.386(5)
Ni(2)-N(2)	2.007(3)	C(17)-C(22)	1.396(5)
O(1)-C(1)	1.388(3)	C(18)-C(19)	1.393(5)
O(2)-C(25)	1.430(6)	C(18)-C(24)	1.500(5)
O(2)-C(28)	1.441(4)	C(19)-H(19)	0.9400
O(3)-C(29)	1.433(4)	C(19)-C(20)	1.375(6)
O(3)-C(32)	1.453(5)	C(20)-H(20)	0.9400
N(1)-C(2)	1.286(4)	C(20)-C(21)	1.368(6)
N(1)-C(9)	1.446(4)	C(21)-H(21)	0.9400
N(2)-C(4)	1.290(4)	C(21)-C(22)	1.396(5)
N(2)-C(17)	1.458(4)	C(22)-C(23)	1.505(6)
C(1)-C(2)	1.545(4)	C(23)-H(23A)	0.9700
C(1)-C(4)	1.546(4)	C(23)-H(23B)	0.9700
C(1)-C(6)	1.573(4)	C(23)-H(23C)	0.9700
C(2)-C(3)	1.497(4)	C(24)-H(24A)	0.9700
C(3)-H(3A)	0.9700	C(24)-H(24B)	0.9700
C(3)-H(3B)	0.9700	C(24)-H(24C)	0.9700
C(3)-H(3C)	0.9700	C(25)-H(25A)	0.9800
C(4)-C(5)	1.493(5)	C(25)-H(25B)	0.9800
C(5)-H(5A)	0.9700	C(25)-C(26)	1.526(6)
C(5)-H(5B)	0.9700	C(26)-H(26A)	0.9800
C(5)-H(5C)	0.9700	C(26)-H(26B)	0.9800
C(6)-H(6)	0.9900	C(26)-C(27)	1.478(7)
C(6)-C(7)	1.506(5)	C(27)-H(27A)	0.9800
C(6)-C(8)	1.530(5)	C(27)-H(27B)	0.9800
C(7)-H(7A)	0.9700	C(27)-C(28)	1.519(7)

C(7)-H(7B)	0.9700	C(28)-H(28A)	0.9800
C(7)-H(7C)	0.9700	C(28)-H(28B)	0.9800
C(8)-H(8A)	0.9700	C(29)-H(29A)	0.9800
C(8)-H(8B)	0.9700	C(29)-H(29B)	0.9800
C(8)-H(8C)	0.9700	C(29)-C(30)	1.505(6)
C(9)-C(10)	1.396(5)	C(30)-H(30A)	0.9800
C(9)-C(14)	1.391(5)	C(30)-H(30B)	0.9800
C(10)-C(11)	1.395(6)	C(30)-C(31)	1.450(7)
C(10)-C(16)	1.493(6)	C(31)-H(31A)	0.9800
C(11)-H(11)	0.9400	C(31)-H(31B)	0.9800
C(11)-C(12)	1.383(8)	C(31)-C(32)	1.440(6)
C(12)-H(12)	0.9400	C(32)-H(32A)	0.9800
C(12)-C(13)	1.378(7)	C(32)-H(32B)	0.9800
C(13)-H(13)	0.9400		

Table S6. Bond angles [°] for 4d

Ni(1)-Br(2)-Ni(2)	77.619(17)	C(9)-C(14)-C(13)	116.8(4)
Br(1)-Ni(1)-Br(2)	100.924(19)	C(9)-C(14)-C(15)	122.9(3)
O(1)-Ni(1)-Br(1)	169.83(6)	C(13)-C(14)-C(15)	120.3(4)
O(1)-Ni(1)-Br(2)	88.79(6)	C(14)-C(15)-H(15A)	109.5
O(1)-Ni(1)-O(2)	85.46(9)	C(14)-C(15)-H(15B)	109.5
O(1)-Ni(1)-N(1)	80.10(9)	C(14)-C(15)-H(15C)	109.5
O(2)-Ni(1)-Br(1)	90.45(7)	H(15A)-C(15)-H(15B)	109.5
O(2)-Ni(1)-Br(2)	96.23(8)	H(15A)-C(15)-H(15C)	109.5
N(1)-Ni(1)-Br(1)	99.78(8)	H(15B)-C(15)-H(15C)	109.5
N(1)-Ni(1)-Br(2)	106.44(8)	C(10)-C(16)-H(16A)	109.5
N(1)-Ni(1)-O(2)	152.70(11)	C(10)-C(16)-H(16B)	109.5
Br(3)-Ni(2)-Br(2)	132.48(2)	C(10)-C(16)-H(16C)	109.5
O(1)-Ni(2)-Br(2)	85.70(6)	H(16A)-C(16)-H(16B)	109.5
O(1)-Ni(2)-Br(3)	97.69(6)	H(16A)-C(16)-H(16C)	109.5
O(1)-Ni(2)-O(3)	170.29(9)	H(16B)-C(16)-H(16C)	109.5
O(3)-Ni(2)-Br(2)	87.97(7)	C(18)-C(17)-N(2)	118.4(3)
O(3)-Ni(2)-Br(3)	92.02(7)	C(18)-C(17)-C(22)	122.2(3)
N(2)-Ni(2)-Br(2)	112.06(7)	C(22)-C(17)-N(2)	119.3(3)
N(2)-Ni(2)-Br(3)	115.21(7)	C(17)-C(18)-C(19)	118.0(3)
N(2)-Ni(2)-O(1)	79.97(9)	C(17)-C(18)-C(24)	121.3(3)
N(2)-Ni(2)-O(3)	95.64(10)	C(19)-C(18)-C(24)	120.7(3)
Ni(1)-O(1)-Ni(2)	102.79(9)	C(18)-C(19)-H(19)	119.6
C(1)-O(1)-Ni(1)	117.77(18)	C(20)-C(19)-C(18)	120.9(4)
C(1)-O(1)-Ni(2)	116.39(17)	C(20)-C(19)-H(19)	119.6
C(25)-O(2)-Ni(1)	116.8(2)	C(19)-C(20)-H(20)	119.9
C(25)-O(2)-C(28)	110.0(3)	C(21)-C(20)-C(19)	120.3(4)
C(28)-O(2)-Ni(1)	125.0(3)	C(21)-C(20)-H(20)	119.9
C(29)-O(3)-Ni(2)	130.0(2)	C(20)-C(21)-H(21)	119.4
C(29)-O(3)-C(32)	107.0(3)	C(20)-C(21)-C(22)	121.2(4)
C(32)-O(3)-Ni(2)	122.8(2)	C(22)-C(21)-H(21)	119.4

C(2)-N(1)-Ni(1)	116.1(2)	C(17)-C(22)-C(23)	122.0(3)
C(2)-N(1)-C(9)	119.9(3)	C(21)-C(22)-C(17)	117.5(4)
C(9)-N(1)-Ni(1)	123.8(2)	C(21)-C(22)-C(23)	120.5(3)
C(4)-N(2)-Ni(2)	117.6(2)	C(22)-C(23)-H(23A)	109.5
C(4)-N(2)-C(17)	120.2(3)	C(22)-C(23)-H(23B)	109.5
C(17)-N(2)-Ni(2)	122.2(2)	C(22)-C(23)-H(23C)	109.5
O(1)-C(1)-C(2)	109.8(2)	H(23A)-C(23)-H(23B)	109.5
O(1)-C(1)-C(4)	110.2(2)	H(23A)-C(23)-H(23C)	109.5
O(1)-C(1)-C(6)	109.8(2)	H(23B)-C(23)-H(23C)	109.5
C(2)-C(1)-C(4)	110.4(3)	C(18)-C(24)-H(24A)	109.5
C(2)-C(1)-C(6)	108.5(2)	C(18)-C(24)-H(24B)	109.5
C(4)-C(1)-C(6)	108.1(2)	C(18)-C(24)-H(24C)	109.5
N(1)-C(2)-C(1)	116.1(3)	H(24A)-C(24)-H(24B)	109.5
N(1)-C(2)-C(3)	122.2(3)	H(24A)-C(24)-H(24C)	109.5
C(3)-C(2)-C(1)	121.5(3)	H(24B)-C(24)-H(24C)	109.5
C(2)-C(3)-H(3A)	109.5	O(2)-C(25)-H(25A)	110.6
C(2)-C(3)-H(3B)	109.5	O(2)-C(25)-H(25B)	110.6
C(2)-C(3)-H(3C)	109.5	O(2)-C(25)-C(26)	105.6(4)
H(3A)-C(3)-H(3B)	109.5	H(25A)-C(25)-H(25B)	108.7
H(3A)-C(3)-H(3C)	109.5	C(26)-C(25)-H(25A)	110.6
H(3B)-C(3)-H(3C)	109.5	C(26)-C(25)-H(25B)	110.6
N(2)-C(4)-C(1)	115.9(3)	C(25)-C(26)-H(26A)	111.2
N(2)-C(4)-C(5)	122.3(3)	C(25)-C(26)-H(26B)	111.2
C(5)-C(4)-C(1)	121.8(3)	H(26A)-C(26)-H(26B)	109.1
C(4)-C(5)-H(5A)	109.5	C(27)-C(26)-C(25)	102.7(4)
C(4)-C(5)-H(5B)	109.5	C(27)-C(26)-H(26A)	111.2
C(4)-C(5)-H(5C)	109.5	C(27)-C(26)-H(26B)	111.2
H(5A)-C(5)-H(5B)	109.5	C(26)-C(27)-H(27A)	111.2
H(5A)-C(5)-H(5C)	109.5	C(26)-C(27)-H(27B)	111.2
H(5B)-C(5)-H(5C)	109.5	C(26)-C(27)-C(28)	102.7(4)
C(1)-C(6)-H(6)	108.5	H(27A)-C(27)-H(27B)	109.1
C(7)-C(6)-C(1)	111.1(3)	C(28)-C(27)-H(27A)	111.2
C(7)-C(6)-H(6)	108.5	C(28)-C(27)-H(27B)	111.2
C(7)-C(6)-C(8)	109.3(3)	O(2)-C(28)-C(27)	104.0(4)
C(8)-C(6)-C(1)	110.7(3)	O(2)-C(28)-H(28A)	111.0
C(8)-C(6)-H(6)	108.5	O(2)-C(28)-H(28B)	111.0
C(6)-C(7)-H(7A)	109.5	C(27)-C(28)-H(28A)	111.0
C(6)-C(7)-H(7B)	109.5	C(27)-C(28)-H(28B)	111.0
C(6)-C(7)-H(7C)	109.5	H(28A)-C(28)-H(28B)	109.0
H(7A)-C(7)-H(7B)	109.5	O(3)-C(29)-H(29A)	110.4
H(7A)-C(7)-H(7C)	109.5	O(3)-C(29)-H(29B)	110.4
H(7B)-C(7)-H(7C)	109.5	O(3)-C(29)-C(30)	106.6(4)
C(6)-C(8)-H(8A)	109.5	H(29A)-C(29)-H(29B)	108.6
C(6)-C(8)-H(8B)	109.5	C(30)-C(29)-H(29A)	110.4
C(6)-C(8)-H(8C)	109.5	C(30)-C(29)-H(29B)	110.4
H(8A)-C(8)-H(8B)	109.5	C(29)-C(30)-H(30A)	110.6
H(8A)-C(8)-H(8C)	109.5	C(29)-C(30)-H(30B)	110.6

H(8B)-C(8)-H(8C)	109.5	H(30A)-C(30)-H(30B)	108.8
C(10)-C(9)-N(1)	118.2(3)	C(31)-C(30)-C(29)	105.5(4)
C(14)-C(9)-N(1)	118.5(3)	C(31)-C(30)-H(30A)	110.6
C(14)-C(9)-C(10)	123.2(3)	C(31)-C(30)-H(30B)	110.6
C(9)-C(10)-C(16)	122.7(3)	C(30)-C(31)-H(31A)	109.9
C(11)-C(10)-C(9)	117.6(4)	C(30)-C(31)-H(31B)	109.9
C(11)-C(10)-C(16)	119.7(4)	H(31A)-C(31)-H(31B)	108.3
C(10)-C(11)-H(11)	119.6	C(32)-C(31)-C(30)	108.9(4)
C(12)-C(11)-C(10)	120.9(4)	C(32)-C(31)-H(31A)	109.9
C(12)-C(11)-H(11)	119.6	C(32)-C(31)-H(31B)	109.9
С(11)-С(12)-Н(12)	119.9	O(3)-C(32)-H(32A)	110.3
C(13)-C(12)-C(11)	120.1(4)	O(3)-C(32)-H(32B)	110.3
C(13)-C(12)-H(12)	119.9	C(31)-C(32)-O(3)	106.9(4)
C(12)-C(13)-H(13)	119.3	C(31)-C(32)-H(32A)	110.3
C(12)-C(13)-C(14)	121.4(5)	C(31)-C(32)-H(32B)	110.3
C(14)-C(13)-H(13)	119.3	H(32A)-C(32)-H(32B)	108.6

References

[1] J. Zhang, H. Gao, Z. Ke, F. Bao, F. Zhu, Q. Wu, Investigation of 1-hexene isomerization and oligomerization catalyzed with β -diketiminato Ni(II) bromide complexes/methylaluminoxane system, J. Mol. Catal. A: Chem., 231 (2005) 27-34.

[2] M. Tripathi, V. Regnier, Z. Ziani, M. Devillard, C. Philouze, D. Martin, Metal free oxidation of vinamidine derivatives: a simple synthesis of α -keto- β -diimine ligands, RSC Adv., 8 (2018) 38346-38350.

[3] G. Ji, Z. Chen, X.-Y. Wang, X.-S. Ning, C.-J. Xu, X.-M. Zhang, W.-J. Tao, J.-F. Li, Y. Gao, Q. Shen, X.-L. Sun, H.-Y. Wang, J.-B. Zhao, B. Zhang, Y.-L. Guo, Y. Zhao, J. Sun, Y. Luo, Y. Tang, Direct Copolymerization of Ethylene with Protic Comonomers Enabled by Multinuclear Ni Catalysts, Nat. Commun., 12 (2021) 6283.

[4] J.-i. Imuta, N. Kashiwa, Y. Toda, Catalytic Regioselective Introduction of Allyl Alcohol into the Nonpolar Polyolefins: Development of One-Pot Synthesis of Hydroxyl-Capped Polyolefins Mediated by a New Metallocene IF Catalyst, J. Am. Chem. Soc., 124 (2002) 1176-1177.

[5] S. Ito, M. Kanazawa, K. Munakata, J.-i. Kuroda, Y. Okumura, K. Nozaki, Coordination–Insertion Copolymerization of Allyl Monomers with Ethylene, J. Am. Chem. Soc., 133 (2011) 1232-1235.

[6] M. Li, X. Wang, Y. Luo, C. Chen, A Second-Coordination-Sphere Strategy to Modulate Nickel- and Palladium-Catalyzed Olefin Polymerization and Copolymerization, Angew. Chem. Int. Ed., 56 (2017) 11604-11609.

[7] T. Wiedemann, G. Voit, A. Tchernook, P. Roesle, I. Gottker-Schnetmann, S. Mecking, Monofunctional Hyperbranched Ethylene Oligomers, J. Am. Chem. Soc., 136 (2014) 2078-2085.
[8] Y. Gao, J. Chen, Y. Wang, D. Pickens, A. Motta, Q.J. Wang, Y.-W. Chung, T.L. Lohr, T.J. Marks, Highly Branched Polyethylene Oligomers via Single-Site Polymerization in Very Nonpolar Media, Nat. Catal., 2 (2019) 236-242.

[9] C. Obuah, B. Omondi, K. Nozaki, J. Darkwa, Solvent and Co-catalyst Dependent Pyrazolylpyridinamine and Pyrazolylpyrroleamine Nickel(II) Catalyzed Oligomerization and Polymerization of Ethylene, J. Mol. Catal. A: Chem., 382 (2014) 31-40.

[10] G. Si, Y. Na, C. Chen, Ethylene (co)Oligomerization by Phosphine-Pyridine Based Palladium and Nickel Catalysts, ChemCatChem, 10 (2018) 5135-5140.

[11] X. Ma, X. Hu, Y. Zhang, H. Mu, L. Cui, Z. Jian, Preparation and In Situ Chain-End-Functionalization of Branched Ethylene Oligomers by Monosubstituted α-Diimine Nickel Catalysts, Polym. Chem., 10 (2019) 2596-2607.

[12] S.V. Zubkevich, V.A. Tuskaev, S.C. Gagieva, A.A. Pavlov, V.N. Khrustalev, F. Wang, L. Pan, Y. Li, D. Saracheno, A.A. Vikhrov, D.N. Zarubin, B.M. Bulychev, Trapping the Short-Chain Odd Carbon Number Olefins Using Nickel(II)-Catalyzed Tandem Ethylene Oligomerization and Friedel-Crafts Alkylation of Toluene, Chin. J. Chem., 41 (2023) 2855-2865.