Supporting Information

Pb_{3.5}GeS₄Br₃: The first phase-matching thiogermanate halide IR nonlinear optical material

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Table S1. Atomic coordinates, equivalent isotropic displacement parameters ($Å^2 \times 10^3$) of Pb₃GeS₄Br₂.

Atoms	х	У	Ζ	$U_{ m eq}$	Wyckoff positions
Pb1	8013.3(4)	6928.8(5)	6484.4(4)	38.56(14)	4 <i>e</i>
Pb2	8459.7(3)	2662.5(7)	4120.2(5)	45.47(15)	4 <i>e</i>
Pb3	5460.9(3)	-944.2(4)	3366.2(4)	27.99(12)	4 <i>e</i>
Gel	6254.4(7)	2948.2(10)	5160.6(8)	15.87(18)	4 <i>e</i>
S 1	6672.5(17)	4938(3)	4097(2)	21.1(4)	4 <i>e</i>
S2	7485.3(18)	915(3)	5463(2)	23.5(4)	4 <i>e</i>
S3	4500.7(16)	1914(2)	3982.2(19)	16.0(4)	4 <i>e</i>
S4	6511(2)	3843(3)	7074(2)	29.0(5)	4 <i>e</i>
Br1	8735.7(10)	6875.4(15)	9343.5(10)	38.8(3)	4 <i>e</i>
Br2	9674.3(8)	4356.4(13)	6993.3(10)	35.8(2)	4 <i>e</i>

 $\label{eq:solution} \mbox{Table S2. Atomic coordinates, equivalent isotropic displacement parameters (\mbox{\AA}^2 \times 10^3) of \mbox{Pb}_{3.5} \mbox{GeS}_4 \mbox{Br}_3.$

Atoms	Х	У	Z	$U_{ m eq}$	Wyckoff positions
Pb1	1453.8(3)	4040.6(3)	8784.7(6)	24.32(13)	6 <i>c</i>
Pb2	0	0	1517(2)	22.0(2)	2a
Gel	3333.33	6666.67	4759(2)	13.0(3)	2b
S 1	3333.33	6666.67	1140(5)	19.0(7)	2b
S2	1230.4(19)	6000(2)	6184(3)	15.4(4)	6 <i>c</i>
Br1	786.3(7)	2606.3(7)	4002.8(15)	23.33(19)	2b

Table S3. Selected bond lengths of Pb₃GeS₄Br₂.

Atom	Atom	Length/Å	Atom	Atom	Length/Å
Pb1	S1	2.980(2)	Pb3	S3 ¹	3.120(2)
Pb1	Br14	3.1346(11)	Pb3	$S1^2$	3.030(2)
Pb1	Br1	2.9899(11)	Pb3	S2	3.046(2)
Pb1	Br2	2.8432(10)	Pb3	S4 ³	2.939(2)
Pb2	S1	2.928(2)	Ge1	S1	2.218(2)
Pb2	S2	2.760(2)	Ge1	S2	2.194(2)
Pb2	S43	2.841(2)	Ge1	S3	2.216(2)
Pb3	S 3	2.8430(19)	Gel	S4	2.193(2)

¹1-X,-Y,1-Z; ²1-X,-1/2+Y,1/2-Z; ³+X,1/2-Y,-1/2+Z; ⁴+X,3/2-Y,-1/2+Z

Table S4. Selected bond lengths of Pb_{3.5}GeS₄Br₃.

Atom	Atom	Length/Å	Atom	Atom	Length/Å
Pb1	S1 ²	2.9490(16)	Pb2	$Br1^7$	2.9613(10)
Pb1	S2 ¹	2.8960(19)	Pb2	$Br1^8$	2.9613(10)
Pb1	S2	2.7797(19)	Pb2	$Br1^4$	2.9702(11)
Pb1	Br1	3.2024(10)	Gel	S1	2.191(3)
Pb2	Br1	2.9613(10)	Gel	S2	2.2243(19)
Pb2	Br1 ⁵	2.9702(11)	Gel	S2 ¹	2.2243(19)
Pb2	Br1 ⁶	2.9702(11)	Ge1	S2 ⁹	2.2243(19)

¹1-Y,1+X-Y,+Z; ²+X,+Y,1+Z; ³-X,-Y,1/2+Z; ⁴-X,-Y,-1/2+Z; ⁵+Y,-X+Y,-1/2+Z; ⁶-Y+X,+X,-1/2+Z; ⁷+Y-X,-X,+Z; ⁸-Y,+X-Y,+Z; ⁹+Y-X,1-X,+Z

Table S5. Selected angles (°) of Pb₃GeS₄Br₂.

Atom	Atom	Atom	Angle/°	Atom	Atom	Atom	Angle/°
S1	Pb1	Br1	140.79(5)	S1 ²	Pb3	S31	125.98(5)
S1	Pb1	Br1 ⁴	72.30(5)	S1 ²	Pb3	S2	136.75(6)
Br2	Pb1	Br1	85.53(3)	S2	Pb3	S31	73.25(6)
Br2	Pb1	Br1 ⁴	86.00(3)	S4 ³	Pb3	S31	145.30(6)
Br2	Pb1	S1	82.80(5)	S4 ³	Pb3	S1 ²	79.75(6)
Br1	Pb1	Br1 ⁴	143.89(4)	S4 ³	Pb3	S2	72.16(6)
S2	Pb2	S1	77.13(6)	S3	Gel	S 1	111.29(8)
S2	Pb2	S4 ³	78.02(7)	S2	Gel	S3	106.52(8)
S4 ³	Pb2	S1	81.79(7)	S2	Gel	S 1	107.06(9)
S3	Pb3	S31	77.44(6)	S4	Ge1	S3	113.45(9)
S 3	Pb3	S1 ²	74.27(6)	S4	Gel	S 1	110.33(10)
S3	Pb3	S2	73.68(5)	S4	Gel	S2	107.86(9)
S 3	Pb3	S4 ³	90.29(7)				

 $\overline{{}^{1}1\text{-}X,-Y,1\text{-}Z;\,{}^{2}1\text{-}X,-1/2\text{+}Y,1/2\text{-}Z;\,{}^{3}\text{+}X,1/2\text{-}Y,-1/2\text{+}Z;\,{}^{4}\text{+}X,3/2\text{-}Y,-1/2\text{+}Z;\,{}^{5}\text{+}X,3/2\text{-}Y,1/2\text{+}Z;\,{}^{6}1\text{-}X,1/2\text{+}Y,1/2\text{-}Z;\,{}^{7}\text{+}X,1/2\text{-}Y,1/2\text{+}Z;\,{}^{6}\text{+}X,3/2\text{-}Y,1/2\text{+}Z;\,{}^{6}\text{+}X,3/2\text{-}Y,1/2\text{+}Z;\,{}^{6}1\text{-}X,1/2\text{+}Y,1/2\text{-}Z;\,{}^{7}\text{+}X,1/2\text{-}Y,1/2\text{+}Z;\,{}^{6}\text{+}X,3/2\text{-}Y,1/2\text{+}Z;\,{}^{6}\text{+}X,3/2\text{-}Y,1/2\text{+}Z;\,{}^{6}1\text{-}X,1/2\text{+}Y,1/2\text{-}Z;\,{}^{7}\text{+}X,1/2\text{-}Y,1/2\text{+}Z;\,{}^{6}\text{+}X,3/2\text{-}Y,1/2\text{+}Z;\,{}^{6}\text{+}X,3/2\text{-}Y,1/2\text{+}Z;\,{}^{6}1\text{-}X,1/2\text{+}Y,1/2\text{-}Z;\,{}^{7}\text{+}X,1/2\text{-}Y,1/2\text{+}Z;\,{}^{7}\text{+}X,1/2\text{-}Y,1/2\text{+}Z;\,{}^{6}\text{+}X,3/2\text{-}Y,1/2\text{+}Z;\,{}^{6}1\text{-}X,1/2\text{+}Y,1/2\text{-}Z;\,{}^{7}\text{+}X,1/2\text{-}Y,1/2\text{+}Z;\,{}^{7}\text{+}X,1/2\text{+}Y,1/2\text{-}X;\,{}^{7}\text{+}X,1/2\text{+}Y,1/2\text{-}X;\,{}^{7}\text{+}X,1/2\text{+}Y,1/2\text{+}X$

Table S6. Selected angles (°) of Pb_{3.5}GeS₄Br₃.

Atom	Atom	Atom	Angle/°	Atom	Atom	Atom	Angle/°
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S2	Pb1	Br1	77.49(4)	Br1	Pb2	Br1 ⁸	83.721(9)
S2 ¹	Pb1	Br1	70.33(4)	$Br1^7$	Pb2	$Br1^8$	179.71(6)
S2	Pb1	$S2^1$	77.43(7)	Br1 ³	Pb2	Br1 ⁵	96.09(4)
S2	Pb1	S1 ²	78.67(6)	Br1 ⁶	Pb2	Br1 ³	83.722(9)
S21	Pb1	S1 ²	76.87(6)	Br1 ⁵	Pb2	Br1 ⁸	96.09(4)
S1 ²	Pb1	Br1	142.78(6)	$Br1^7$	Pb2	Br1 ³	83.722(9)
Br1 ³	Pb2	Br1 ⁸	96.09(4)	Br1	Pb2	Br1 ⁵	83.721(9)
Br1	Pb2	Br1 ³	179.71(6)	S2	Ge1	S21	105.93(6)
Br1 ⁶	Pb2	Br1 ⁸	83.722(9)	S2 ⁹	Ge1	S21	105.93(6)
Br1 ⁶	Pb2	$Br1^7$	96.47(4)	S2	Ge1	S2 ⁹	105.93(6)
Br1 ⁶	Pb2	Br1 ⁵	179.71(6)	S1	Ge1	S21	112.82(6)
Br1 ⁷	Pb2	Br1 ⁵	83.722(9)	S1	Ge1	S29	112.82(6)
Br1 ⁶	Pb2	Br1	96.47(4)	S 1	Ge1	S2	112.82(6)
$Br1^7$	Ph2	Br1	96 47(4)				

¹1-Y,1+X-Y,+Z; ²+X,+Y,1+Z; ³-X,-Y,-1/2+Z; ⁴-X,-Y,1/2+Z; ⁵-Y+X,+X,-1/2+Z; ⁶+Y-X,-X,+Z; ⁷-Y,+X-Y,+Z; ⁸+Y,-X+Y,-1/2+Z; ⁹+Y-X,1-X,+Z; ¹⁰+Y-X,1-X,-1+Z; ¹¹+X,+Y,-1+Z; ¹²1-Y,1+X-Y,-1+Z

Table S7. Anisotropic displacement parameters ($Å^2 \times 10^3$) for Pb₃GeS₄Br₂. The anisotropic displacement factor exponent takes the form: - $2\pi^2[h^2a^{*2}U_{11}+2hka^*b^*U_{12}+...]$.

Atom	U ₁₁	U ₂₂	U ₃₃	U ₂₃	U ₁₃	U ₁₂
Pb1	38.5(2)	48.7(3)	25.6(2)	-1.34(16)	11.60(17)	18.96(17)
Pb2	25.7(2)	74.7(3)	40.7(3)	-11.1(2)	18.88(18)	-10.34(18)
Pb3	34.3(2)	26.06(19)	27.5(2)	2.63(13)	17.18(16)	2.76(13)
Ge1	18.0(4)	18.3(4)	11.6(4)	0.2(3)	6.8(3)	0.2(3)
S1	21.1(9)	21.9(9)	16.8(10)	3.5(7)	5.3(7)	-2.0(7)
S2	20.2(9)	20.8(9)	26.7(11)	2.2(8)	7.7(8)	5.2(7)
S3	16.0(8)	18.8(9)	13.6(9)	-0.3(7)	6.9(7)	-0.1(6)
S4	30.1(11)	41.1(13)	11.7(10)	-4.4(8)	5.5(8)	15.7(9)
Br1	41.3(5)	59.5(7)	23.3(5)	-6.7(4)	21.2(4)	-14.9(5)
Br2	26.1(4)	36.5(5)	31.0(5)	-9.8(4)	0.0(4)	11.4(4)

Table S8. Anisotropic displacement parameters (Å $^2 \times 10^3$) for Pb_{3.5}GeS_4Br_3. The anisotropic displacement factor exponent takes the form: - $2\pi^2[h^2a^{*2}U_{11}+2hka^{*}b^{*}U_{12}+...].$

Atom	U11	U22	U33	U23	U13	U12
Pb1	27.61(19)	21.53(18)	23.92(19)	4.86(17)	1.42(18)	12.36(13)
Pb2	17.6(3)	17.6(3)	30.9(6)	0	0	8.78(14)
Ge1	14.3(4)	14.3(4)	10.4(6)	0	0	7.1(2)
S1	23.3(11)	23.3(11)	10.5(17)	0	0	11.7(6)
S2	13.5(9)	16.6(10)	16.7(10)	-0.3(8)	0.4(7)	7.9(8)
Br1	22.3(4)	24.0(4)	21.5(4)	-2.1(4)	-4.3(4)	10.0(3)



Fig. S1 The optical image of Pb_{3.5}GeS₄Br₃ crystals.



Fig. S2 The EDS (a) and Raman spectra (b) of $Pb_3GeS_4Br_2$ and $Pb_{3.5}GeS_4Br_3.$



Fig. S3 The structures of $Pb_3GeS_4Br_2$ and $Pb_{3,5}GeS_4Br_3$. (a) The isolated [GeS₄] pseudo-layer structure alignment along the glide surface in $Pb_3GeS_4Br_2$; (b) The opposite-arranged [GeS₄] tetrahedra in $Pb_3GeS_4Br_2$; (c) The isolated [GeS₄] tetrahedral arranged around the [PbS₆] column configuration in $Pb_{3,5}GeS_4Br_3$; (d) The whole structure of $Pb_{3,5}GeS_4Br_3$; (e) The well-arranged [GeS₄] tetrahedra in $Pb_{3,5}GeS_4Br_3$.







Fig. S6 The RID measurement of $Pb_{3.5}GeS_4Br_3$. (a) The plane indices determined by single crystal XRD. (b) The RID was measured to 0.05@546 nm (c) The thickness of the single crystal is 20.495 μ m measured by an optical microscope.



Fig. S7 The band structure of $Pb_3GeS_4Br_2$.