

Supporting Information

Designing CoS₂-Mo₂C and CoS₂-W₂C hybrids for high-performance supercapacitors and hydrogen evolution reactions

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S1. Characterization

Field emission scanning electron microscopy (FESEM) images and energy-dispersive X-ray spectroscopy (EDX) mapping images were obtained using (HITACHI S-4700), 5 kV. The Raman spectroscopy measurements were accomplished using Renishaw Invia RE04, Ar laser - 512 nm at room temperature. The structural properties were characterized by Rigaku X-ray diffractometer with Cu-K α radiation (0.154 nm) at 40 kV and 40 mA in the scanning range of 10-80° (2 θ). X-ray powder diffraction (XRD) was conducted with Cu-K α radiation (0.154 nm) at 40 kV and 40 mA in the scanning range of 5-80° (2 θ). The XPS measurements were performed using an Ulvac PHI X-tool spectrometer with Al K α X-ray radiation (1486.6 eV). The atomic structures were characterized by a JEOL-2010F transmission electron microscopy with an accelerating voltage of 200 keV. 3Flex surface characterization analyzer for nitrogen adsorption and desorption measurement at 77 K (Micromeritics, USA).

The HER potential values were converted for reversible hydrogen electrode (RHE) by the given formula: $E(\text{RHE})_{\text{Ag/AgCl}} = E(\text{vs Ag/AgCl}) + E^0_{(\text{Ag/AgCl})} + 0.0592 \times \text{pH}$ for acidic medium and $E(\text{RHE})_{\text{HgO}} = E(\text{vs Hg/HgO}) + E^0_{(\text{Hg/HgO})} + 0.0592 \times \text{pH}$ for KOH medium.

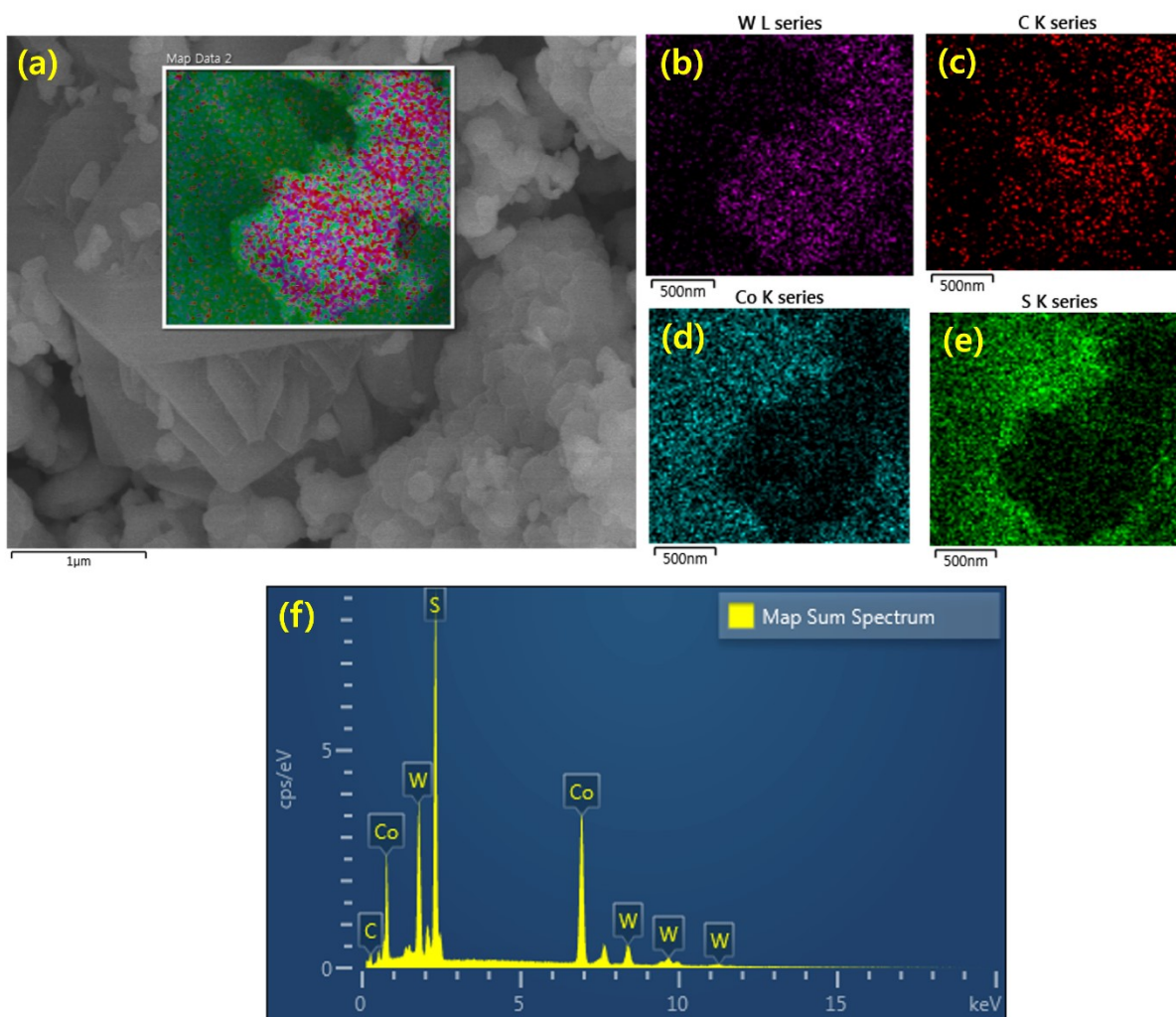


Figure S1. (a) FESEM elemental mapping and (b) W, (c) C, (d) Co and (e) S elements, (f) EDS profile for the $\text{CoS}_2\text{-W}_2\text{C}$ hybrids.

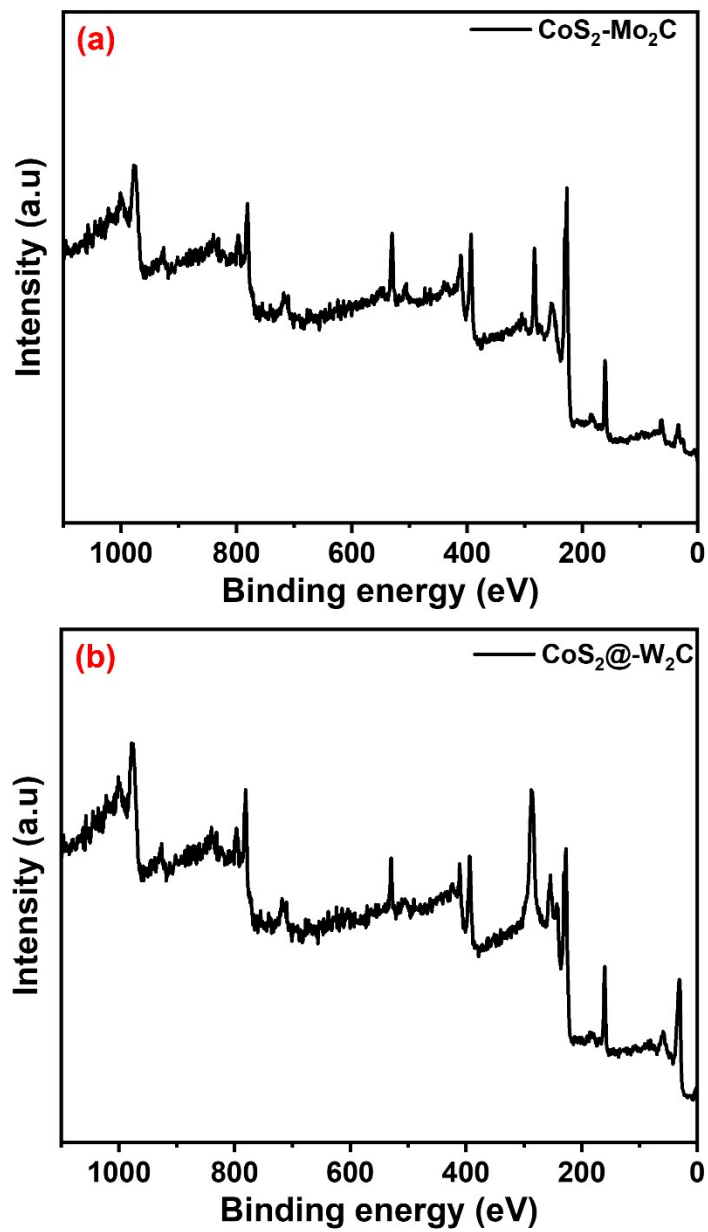


Figure S2. XPS survey spectra for (a) CoS₂-Mo₂C and (b) CoS₂-W₂C hybrids.

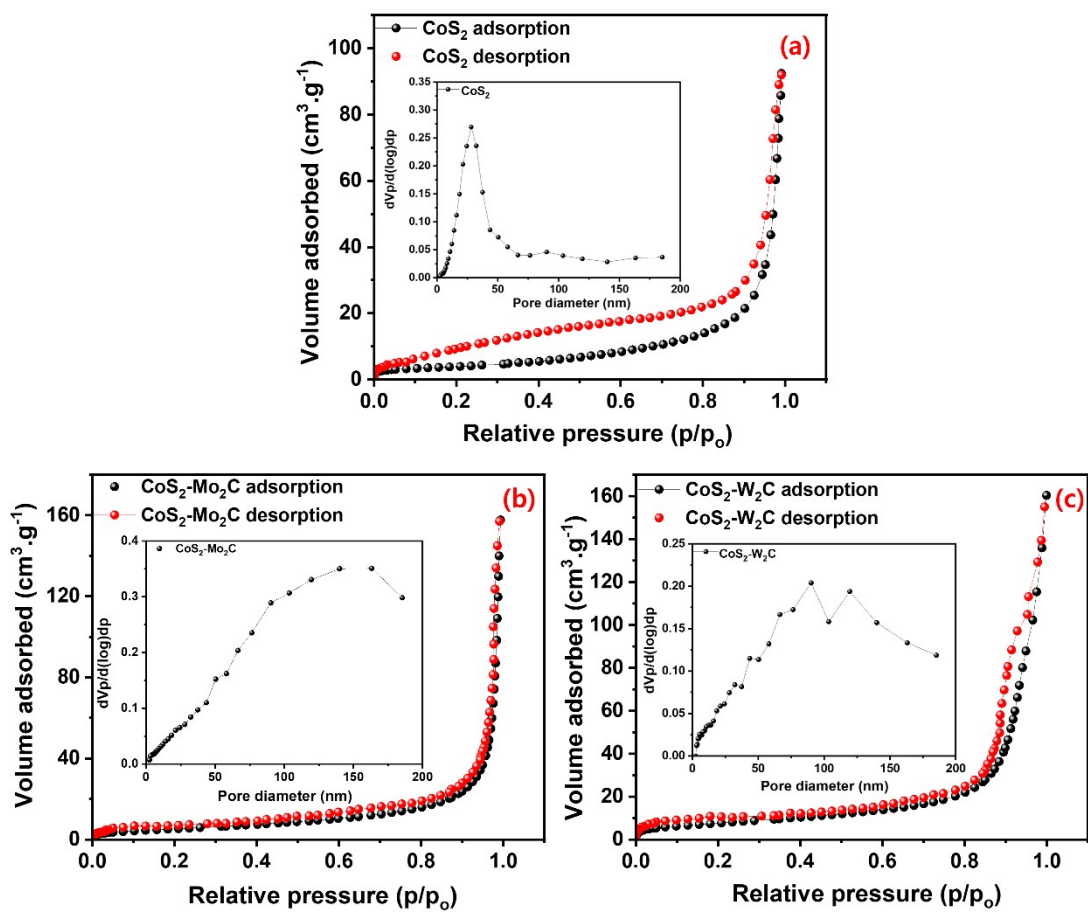


Figure S3. (a-c) BET isotherms curves for (c) CoS_2 , (d) $\text{CoS}_2\text{-Mo}_2\text{C}$ and (e) $\text{CoS}_2\text{-W}_2\text{C}$ hybrid nanocomposites (Inset: BJH profiles).

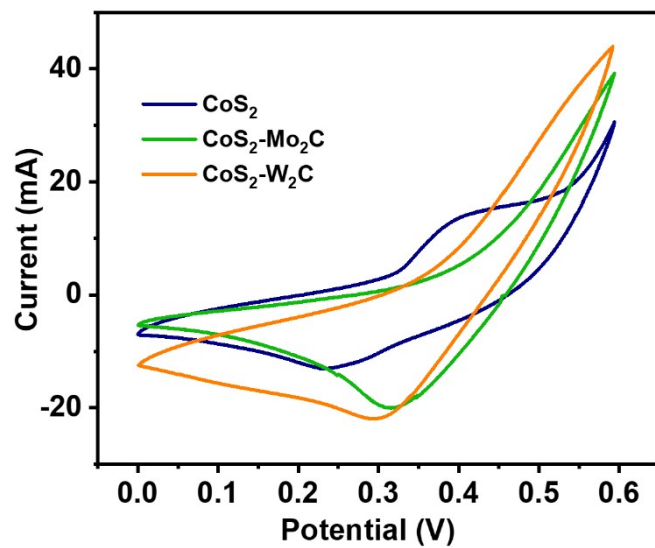


Figure S4. CVs at 100 mV/s for CoS₂, CoS₂-W₂C and CoS₂-Mo₂C electrodes

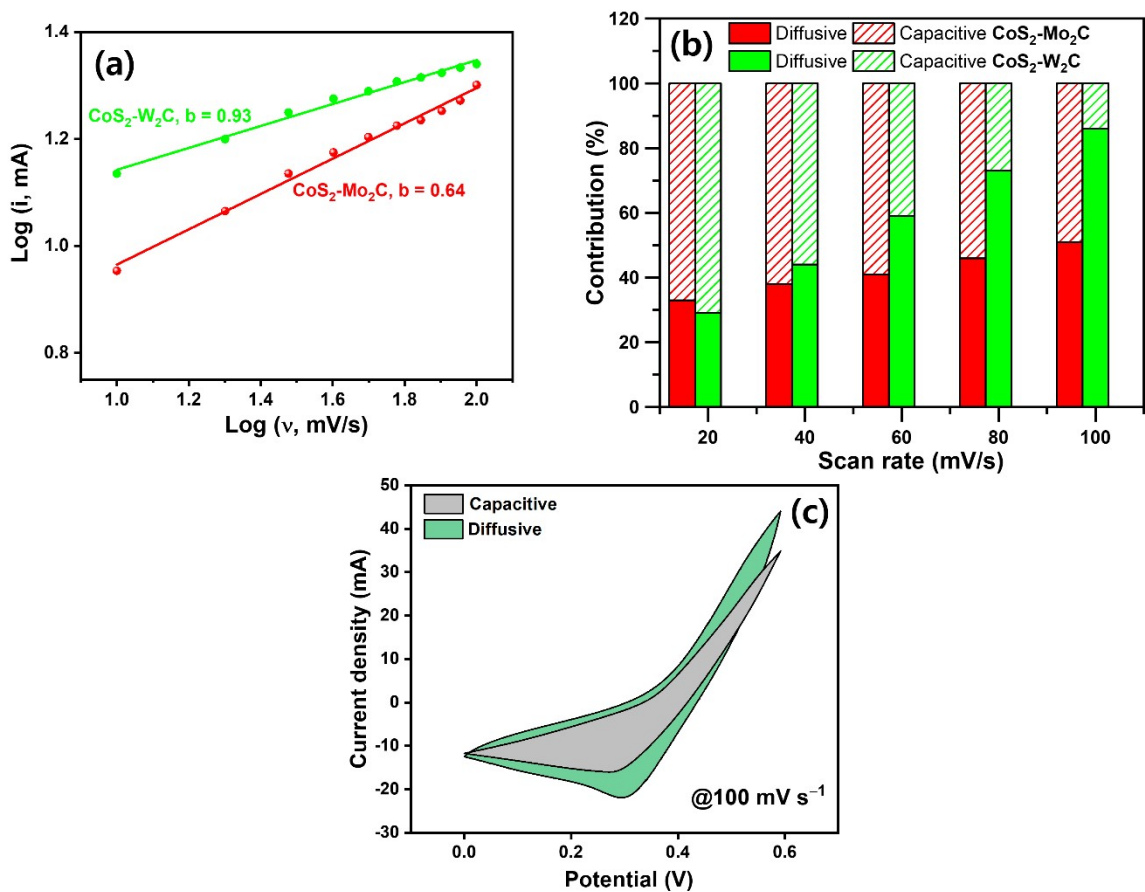


Figure S5. (a) Logarithmic relationship between peak current and scan rates for the hybrid electrodes; (b) Capacitive and diffusive process contribution percentage at different scan rates for hybrid electrodes; (c) Capacitive and diffusion contribution curves for $\text{CoS}_2\text{-W}_2\text{C}$ hybrid electrode at 100 mV/s .

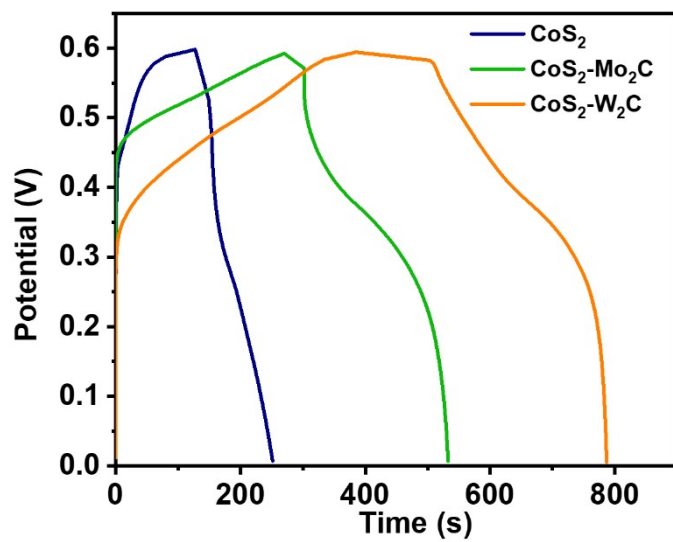


Figure S6. GCDs at 2 A/g for CoS_2 , $\text{CoS}_2\text{-W}_2\text{C}$ and $\text{CoS}_2\text{-Mo}_2\text{C}$ electrodes

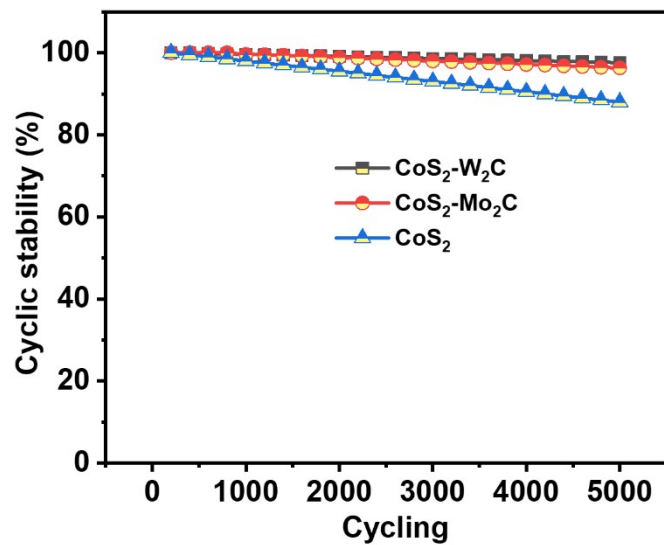


Figure S7. Cycling stability of CoS₂, CoS₂-W₂C and CoS₂-Mo₂C by half-cell

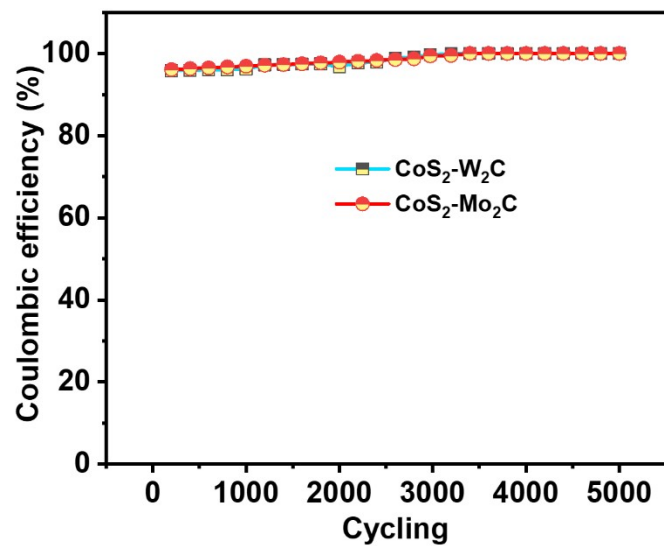


Figure S8. Coulombic efficiency of CoS₂-W₂C and CoS₂-Mo₂C hybrid

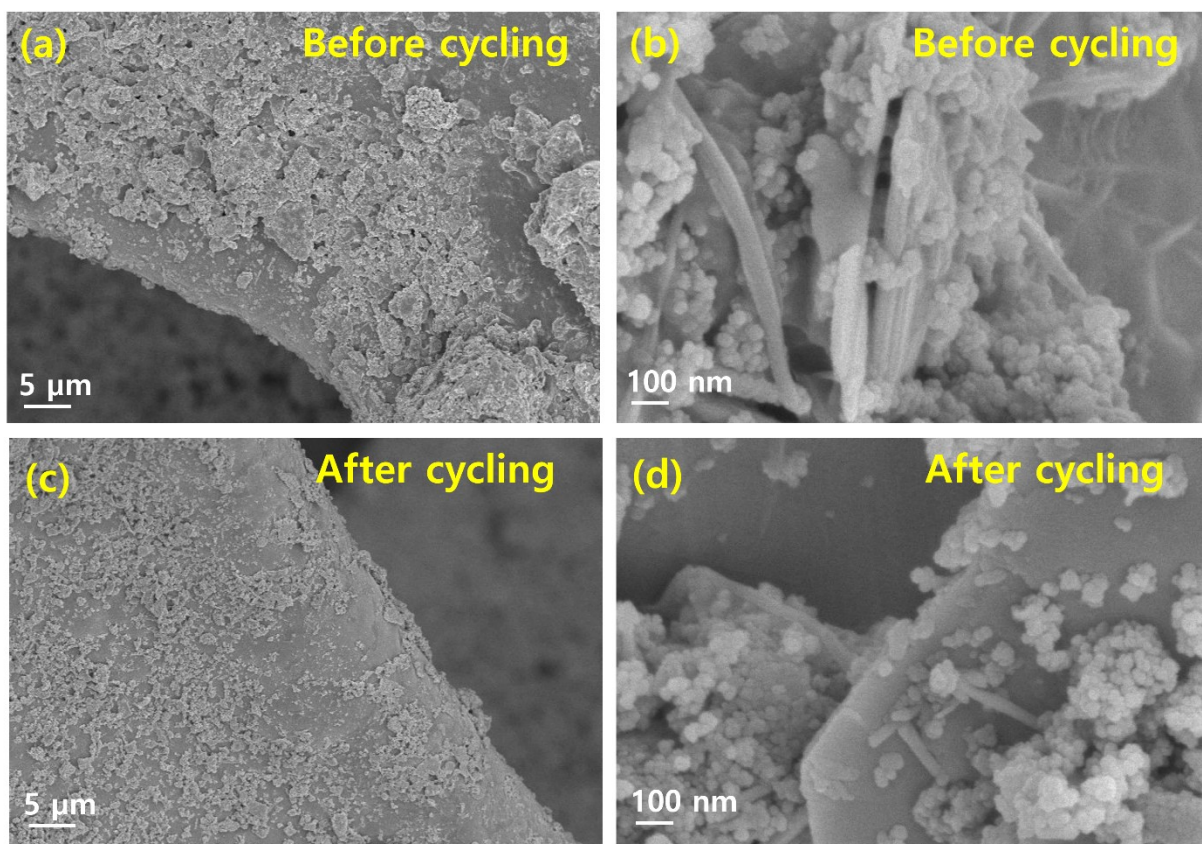


Figure S9. FESEM images (a-b) before and (c-d) after cycling of $\text{CoS}_2\text{-W}_2\text{C}$ coated NF supercapacitor electrode

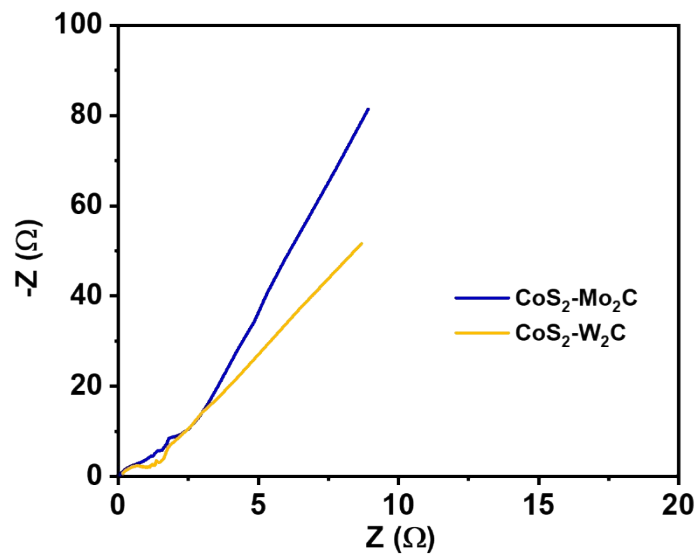


Figure S10. EIS profiles $\text{CoS}_2\text{-W}_2\text{C}$ and $\text{CoS}_2\text{-Mo}_2\text{C}$ ASC devices

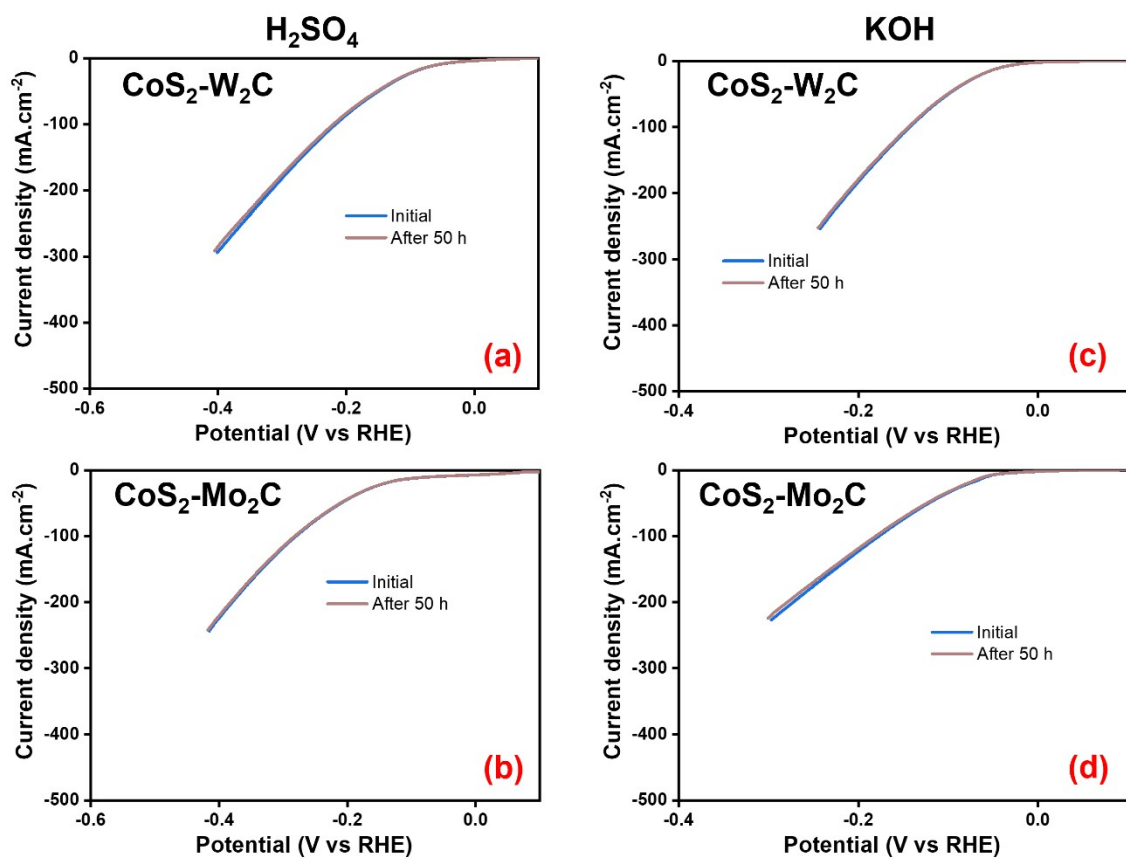


Figure S11. LSV profiles (a,c) CoS₂-W₂C and (b,d) CoS₂-Mo₂C under acid and KOH environment.

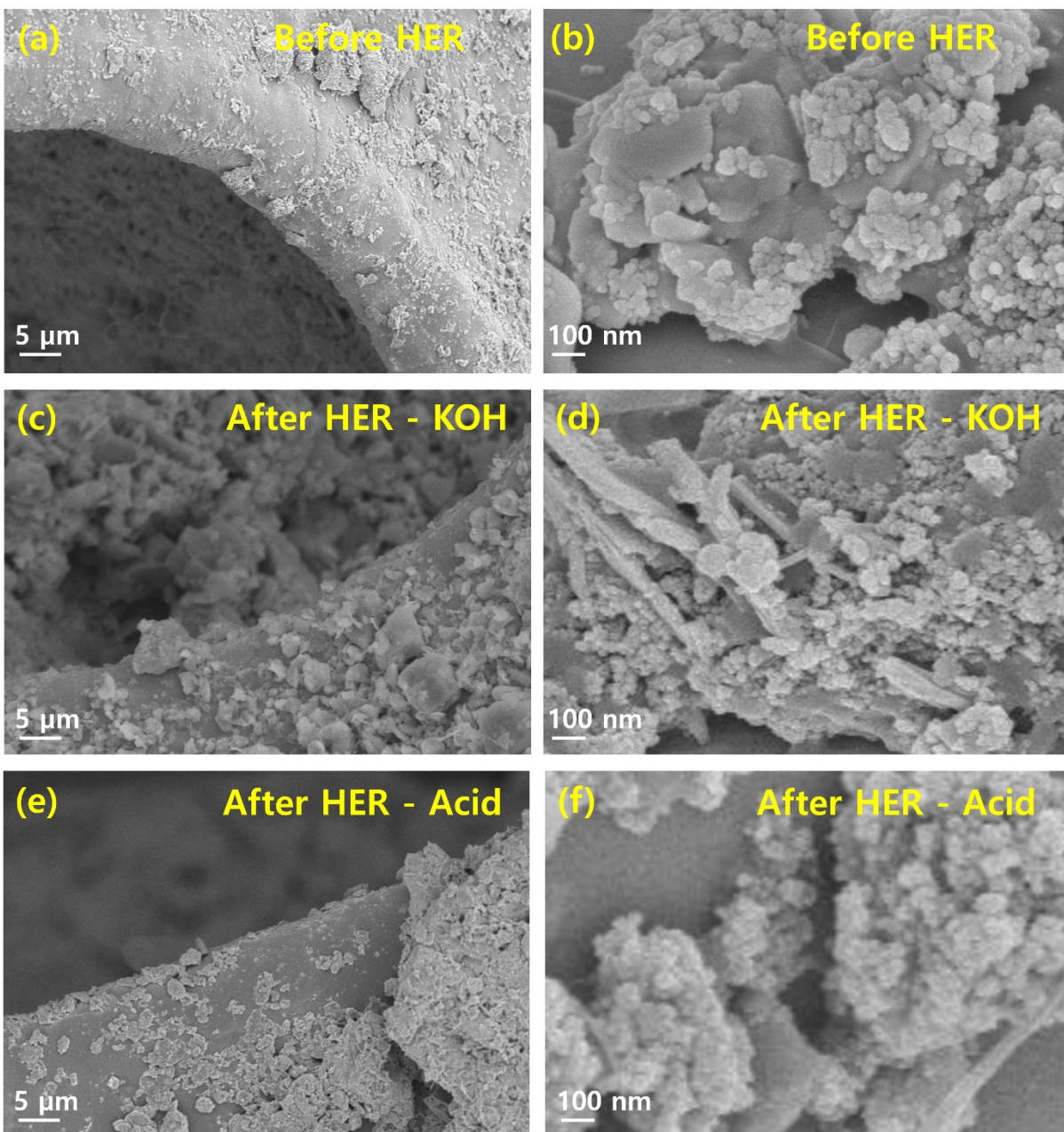


Figure S12. FESEM images (a-b) before HER and after HER in (c-d) KOH and (e-f) acid environment for $\text{CoS}_2\text{-W}_2\text{C}$ electrocatalyst coated NF.

Table S1. Half-cell supercapacitor performances of TMDs and TMCs based electrodes

Electrode materials	Electrolyte	Specific capacitance / capacity	Energy density	Power density	Capacitance retention (%) / cycles	Ref.
<i>CoS₂-W₂C</i>	<i>3 M KOH</i>	<i>720 C/g at 2 A/g</i>	-	-	<i>97/5000</i>	<i>This work</i>
3D graphene/ MoS ₂ composite	1.0 M Na ₂ SO ₄	410 F·g ⁻¹ @ 1 A·g ⁻¹	-	-	80.3/10000	1
Mo ₂ C/MoS ₂ hybrid	1 M KOH	1040 F·g ⁻¹ @ 0.5 A·g ⁻¹	-	-	94/5000	2
W ₂ C/MoS ₂ hybrid	1 M KOH	681 F·g ⁻¹ @ 0.5 A·g ⁻¹	-	-	90/5000	2
WS ₂ /RGO hybrids	1.0 M Na ₂ SO ₄	350 F/g@ 2mV/s	~49 Wh/kg	-	-	3
WS ₂ @MXene/GO	1 KOH	1111 F·g ⁻¹ @ 2 A·g ⁻¹	-	-	97.15/5000	4
3D graphene-MoS ₂ hybrid	1.0 M KOH	169.3 F/g	28.43 Wh/Kg	10.18 W/kg	-	5
MoSe ₂ -Mo ₂ C hybrid nanoarrays	1 M KOH	850 F·g ⁻¹ @ 2.5 A·g ⁻¹	-	-	98/10000	6
Mo ₂ C/NCF	6.0 M KOH	1250 F·g ⁻¹ @ 1 A·g ⁻¹	-	-	100/500	7
MXene-NiCo ₂ S ₄ @ NF	3 M KOH	1147.47 F·g ⁻¹ @ 1 A·g ⁻¹	-	-	80.4/3000	8
MoS ₂ /CNS	1 M Na ₂ SO ₄	108 F g ⁻¹ @ 1 A g ⁻¹	74 Wh/kg	3700 W/Kg	-	9
MoS ₂ /MXene nanohybrid	3 M KOH	583 F·g ⁻¹ @ 1 A·g ⁻¹	-	-	82.5/3000	10
MoS ₂ /MWCNT	1 M Na ₂ SO ₄	452.7 F g ⁻¹ @ 1 A g ⁻¹	-	-	95.8/ 1000	11
NiCo ₂ S ₄ -g-MoS ₂	1.0 M KOH	1270 F g ⁻¹ @1 A g ⁻¹	-	-	94.8/4000	12
MoS ₂ -graphene	1.0 M KOH	756 F·g ⁻¹ @ 0.5 A·g ⁻¹	6 Wh/Kg	125 W/kg	88/10000	13

Table S2. Asymmetric supercapacitors performances of TMDs and TMCs based electrodes

Electrode materials	Specific capacitance	Energy density	Power density	Capacitance retention (%) / cycles	Ref.
<i>CoS₂-W₂C</i>	<i>423 F/g at 2 A/g</i>	<i>150 Wh/Kg</i>	<i>4.5 kW/Kg</i>	<i>94.1/5000</i>	<i>This work</i>
MXene/MoSe ₂	350 F·g ⁻¹ @ 1 A·g ⁻¹	48 Wh/Kg	500 W/kg	93/5000	14
MXene-NiCo ₂ S ₄ @NF	-	27.24 Wh/Kg	0.48 kW/kg	-	8
MXene/CuS	49.3 F·g ⁻¹ @ 0.5 A·g ⁻¹	15.4 Wh/Kg	750.2 W/kg	82.4/5000	15
1T-MoS ₂ / MXene	386.7F·g ⁻¹ @ 1 A·g ⁻¹	-	-	91.1/20000	16
MXene/NiCo ₂ S ₄	621F·g ⁻¹ @ 1 A·g ⁻¹	72.82 Wh/Kg	0.635k W/Kg	90.88/20000	17
1T-VS ₂ /MXene Hybrid	115.7 F·g ⁻¹ @ 0.8 A·g ⁻¹	41.13 Wh/Kg	793.50 W/kg	85/5000	18
Cu _{0.5} Co _{0.5} Se ₂ //MXene	321 F·g ⁻¹ @ 1 A·g ⁻¹	84.19 Wh/Kg	715.12 W/kg	91.1/10000	19
MXene-MoO ₂	3 F cm ⁻³ @ 2 mV s ⁻¹	9.7 mW h cm ⁻³	0.198 W cm ⁻³	88/10000	20
NiMoO ₄ /Ti ₃ C ₂ T _x	137.3 F·g ⁻¹ @ 0.5 A·g ⁻¹	33.36 Wh/Kg	400.08 W/kg	72.6/10000	21
Ti ₃ C ₂ /Ni-Co-Al-LDH	128.89 F·g ⁻¹ @ 0.5 A·g ⁻¹	23.6 Wh/Kg	6.93 kW/kg	97.8/10000	22
NiCoS/d-Ti ₃ C ₂	95.2 F·g ⁻¹ @ 0.5 A·g ⁻¹	22.6 Wh/Kg	0.4 kW/kg	91.2/10000	23
graphene/MXene hydrogel	226.7 F·g ⁻¹ @ 1 A·g ⁻¹	9.3 Wh/Kg	500 W/kg	-	24
NiCo ₂ -LDHs@MXene/rGO	240 F·g ⁻¹ @ 0.5 A·g ⁻¹	65.3 Wh/Kg	700 W/kg	92.8/10000	25
MXene/MoSe ₂ /ASC	156.3 F·g ⁻¹ @ 0.5 A·g ⁻¹	55.6 Wh/Kg	800.3 W/kg	94.1/5000	26
V ₂ NT _x MXene	112.8 F·g ⁻¹ @ 1.85 mA/cm ²	15.66 Wh/Kg	3748.4 W/kg	96/10000	27
MXene@Ni-Mn LDH	56 F·g ⁻¹ @ 1 A·g ⁻¹	44.7 Wh/Kg	800 W/kg	90.3/5000	28
Ni-S/d-Ti ₃ C ₂ nanohybrid	69.4 c·g ⁻¹ @ 0.5 A·g ⁻¹	5.1 Wh/Kg	10k W/kg	71.4/10000	29

$\text{Ni}_{1.5}\text{Co}_{1.5}\text{S}_4$ 5@ Ti_3C_2	140 $\text{A}\cdot\text{g}^{-1}$ @ 1 $\text{A}\cdot\text{g}^{-1}$	49.8 Wh/Kg	800 W/kg	90/8000	30
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Table S3. HER catalytic performances TMDs and TMCs-based electrocatalysts

Electrocatalyst	Electrolyte	η (mV)	Tafel Slope (mV \cdot dec ⁻¹)	j_0 (mA \cdot cm ⁻²)	Ref
CoS₂-W₂C	1 M KOH	42	27	1.35	This work
CoS₂-W₂C	0.5 M H₂SO₄	50	42	1.2	
Reduced GO-Mo ₂ C composites	0.5 M H ₂ SO ₄	206 @ 10 mA/cm ²	52	-	31
MoSe ₂ /NiSe ₂ composite nanowires	0.5 M H ₂ SO ₄	249 @ 100 mA/cm ²	46.9	-	32
MoO ₂ / α -Mo ₂ C heterojunction	0.5 M H ₂ SO ₄ & 1 M KOH	152 & - 100 @ 10 mA/cm ²	65 & 50	4.42 \times 10 ⁻²	33
Mo ₂ C Nanoparticles	0.5 M H ₂ SO ₄ & 1 M KOH	180 & 210 @ 10 mA/cm ²	49 & 48	3 \times 10 ⁻³	34
WS ₂ /W ₂ C heterostructure	0.5 M H ₂ SO ₄	126 @ 10 mA/cm ²	68	0.501	35
MoSe ₂ @MoS ₂	0.5 M H ₂ SO ₄	161 @ 10 mA/cm ²	60	-	36
Mo ₂ C/mesoporous carbon	0.1 M KOH	165 @ 10 mA/cm ²	63.3	-	37
Mo ₂ C/ N doped carbon nanotubes	0.5 M H ₂ SO ₄	147 @ 10 mA/cm ²	71	72.7 @ 200 mV	38
MoC-Mo ₂ C Heteronanowires	0.5 M H ₂ SO ₄ & 1 M KOH	126 & 120 @ 10 mA/cm ²	43 & 42	1.1 \times 10 ⁻²	39
CoSe ₂ /MoSe ₂ heterostructures	1 M KOH	218 @ 10 mA/cm ²	76	-	40
Mo ₂ C/CNT	1.0 M HClO ₄	64 @ 1 mA/cm ²	52.2	1.4 \times 10 ⁻²	41
MoSe ₂ /Bi ₂ Se ₃ hybrids	0.5 M H ₂ SO ₄	300 mV @ 85 mA/cm ²	44	-	42
MoP/Mo ₂ C@C	0.5 M H ₂ SO ₄	89 @ 10 mA/cm ²	45	0.215	43
Mo ₂ C/MoS ₂	0.5 M H ₂ SO ₄ & 1 M KOH	93 & 98 @ 10 mA/cm ²	67 & 68	0.952 & 1.32	2
MoSSe Nanoflake	0.5 M H ₂ SO ₄	164 @ 10 mA/cm ²	48	-	44
MoSSe@rGO composite	0.5 M H ₂ SO ₄	135 @ 5 mA/cm ²	51	-	45
Mo ₂ C Nanoparticles/ N doped porous carbon nanofibers	0.5 M H ₂ SO ₄ & 1 M KOH	85 & 90 @ 1 mA/cm ²	68 & 60.2	0.178	46
Mo ₂ C/Graphene Nanoribbons	0.5 M H ₂ SO ₄ & 1 M KOH	167 & 217 @ 10 mA/cm ²	63 & 64	-	47

Mo ₂ C/CNT-graphene	0.5 M H ₂ SO ₄	130@ 10 mA/cm ²	58	6.20×10 ⁻²	48
MoC _x nano- octahedrons	0.5 M H ₂ SO ₄ and 1M KOH	142 & 151@ 10 mA/cm ²	53 & 59	0.023 & 0.029	49

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