

*Supporting information*

**A Polar Two-Dimensional Lead-Free Hybrid Perovskite for Self-Powered Polarization-Sensitive Photodetection**

*Cheng-Dong Liu, Chang-Chun Fan, Bei-Dou Liang, and Wen Zhang\**

Jiangsu Key Laboratory for Science and Applications of Molecular Ferroelectrics and School of Chemistry and Chemical Engineering, Southeast University, Nanjing 211189, China.

E-mail: zhangwen@seu.edu.cn

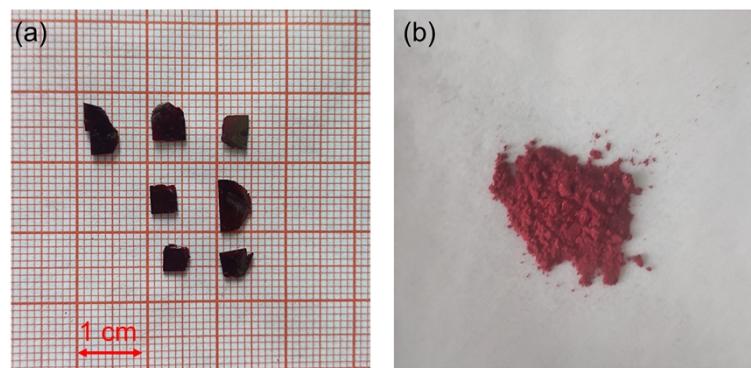
## **Experimental Section**

### **General characterizations**

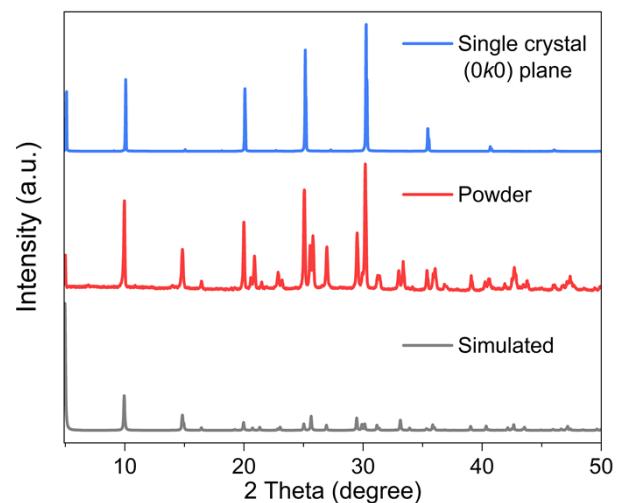
Powder X-ray diffraction (PXRD) patterns were obtained on a Rigaku SmartLab X-ray diffraction instrument. Ultraviolet-vis (UV-vis) absorption spectra were measured with Shimadzu UV-2600 equipped with ISR-2600Plus integrating sphere. The single crystal growth was carried out using a home-made hydrothermal growth system.

### **Single-crystal X-ray Diffraction.**

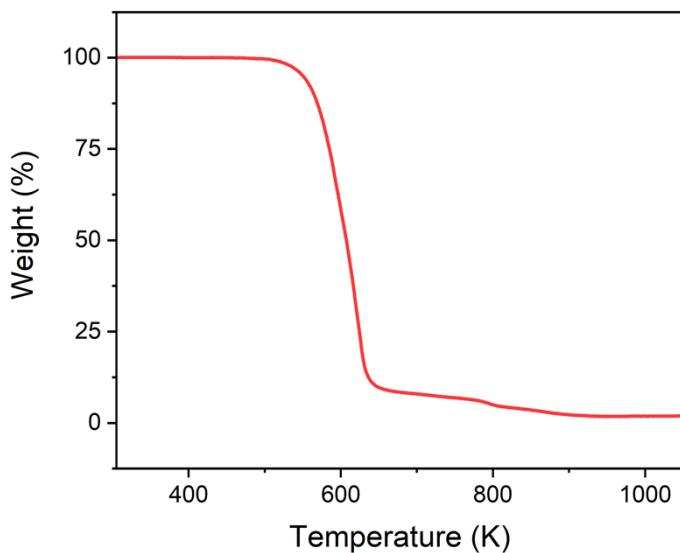
Crystallographic data of the compound were collected on a Rigaku Oxford Diffraction Supernova Dual Source, Cu at Zero equipped with an AtlasS2 CCD using Mo K $\alpha$  radiation and an XtaLAB Synergy R, DW system, HyPix diffractometer. Rigaku CrysAlisPro software was used to collect data, refine cell, and to reduce data. SHELXL-2018 with the OLEX2 interface was used to solve the structures by direct methods. All non-hydrogen atoms were refined anisotropically. The positions of hydrogen atoms were generated geometrically.



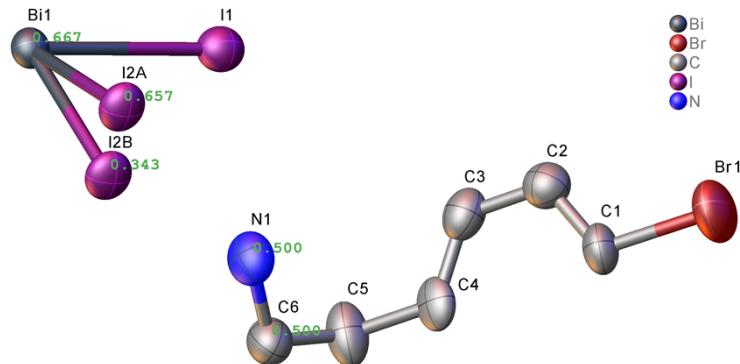
**Figure S1.** (a) Single crystal and (b) powder sample of  $(4\text{BrPEA})_3\text{BiI}_6$ .



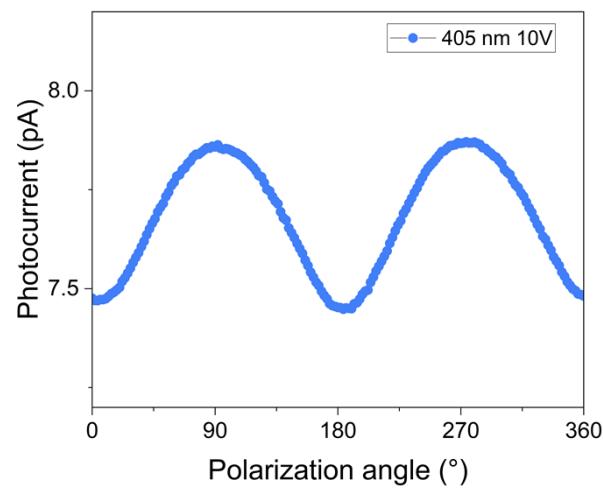
**Figure S2.** PXRD patterns of  $(4\text{BrPEA})_3\text{BiI}_6$  single crystal and powder sample.



**Figure S3.** Thermogravimetric analysis of  $(4\text{BrPEA})_3\text{BiI}_6$ .



**Figure S4.** The asymmetric unit of  $(4\text{BrPEA})_3\text{BiI}_6$ . The black label represents the atomic number while the green label represents the chemical occupancy.



**Figure S5.** Polarization-dependent photocurrent measured at 10 V bias.

**Table S1.** Crystal data and structure refinement details for  $(4\text{BrPEA})_3\text{BiI}_6$ .

Empirical formula	$\text{C}_8\text{H}_{11}\text{NBrBi}_{0.33}\text{I}_2$
Formula weight	524.63
Temperature / K	292(2)
Crystal system	Orthorhombic
Space group	$Fmm2$
a / Å	8.5646(5)
b / Å	35.477(5)
c / Å	8.5649(5)
$\alpha$ / °	90
$\beta$ / °	90
$\gamma$ / °	90
Volume / Å <sup>3</sup>	2602.4(4)
Z	8
$\rho_{\text{calc}}$ / cm <sup>3</sup>	2.678
$\mu$ / mm <sup>-1</sup>	12.357
$F(000)$	1878.0
Radiation	Mo K $\alpha$ ( $\lambda = 0.71073$ )
2 $\theta$ range for data collection / °	4.592 to 52.594
Index ranges	-10 ≤ h ≤ 10, -44 ≤ k ≤ 44, -10 ≤ l ≤ 10
Reflections collected	6692
Independent reflections	1418 [ $R_{\text{int}} = 0.0299$ , $R_{\text{sigma}} = 0.0253$ ]
Data/restraints/parameters	1418/3/86
Goodness-of-fit on $F^2$	1.081
Final $R$ indexes [ $I \geq 2\sigma(I)$ ]	$R_1 = 0.0248$ , $wR_2 = 0.0576$
Final $R$ indexes [all data]	$R_1 = 0.0281$ , $wR_2 = 0.0584$
Largest diff. peak/hole / e Å <sup>-3</sup>	1.84/-1.49
Flack parameter	-0.012(12)

<sup>a</sup> $R_1 = \sum ||F_o| - |F_c|| / \sum |F_o|$ . <sup>b</sup> $wR_2(F^2) = [\sum w(F_o^2 - F_c^2)^2 / \sum wF_o^4]^{1/2}$ .

**Table S2.** Bond lengths for  $(4\text{BrPEA})_3\text{BiI}_6$ .

Atom–Atom	Lengths / Å	Atom–Atom	lengths / Å
Bi1–I1	3.1640 (8)	C3–H3	0.9300
Bi1–I1 <sup>i</sup>	3.1639 (8)	C3–C4	1.377 (12)
Bi1–I2A <sup>ii</sup>	3.0404 (15)	C4–C5	1.491 (17)
Bi1–I2A <sup>iii</sup>	3.0404 (15)	C5–H5A	0.9700
Bi1–I2A <sup>i</sup>	3.0680 (16)	C5–H5A <sup>iv</sup>	0.9700
Bi1–I2A	3.0680 (16)	C5–H5B	0.9700
Bi1–I2B <sup>i</sup>	3.109 (2)	C5–H5B <sup>iv</sup>	0.9700
Bi1–I2B <sup>iii</sup>	3.106 (3)	C5–C6	1.494 (17)
Bi1–I2B <sup>ii</sup>	3.106 (3)	N1–H1A	0.8900
Bi1–I2B	3.109 (2)	N1–H1B	0.8900
Br1–C1	1.892 (12)	N1–H1C	0.8900
C1–C2	1.373 (11)	N1–C6	1.47 (2)
C1–C2 <sup>iv</sup>	1.373 (11)	C6–H6A	0.9700
C2–H2	0.9300	C6–H6B	0.9700
C2–C3	1.383 (12)		

Symmetry codes: (i)  $-x+1, -y+1, z$ ; (ii)  $x+1/2, y, z+1/2$ ; (iii)  $-x+1/2, -y+1, z+1/2$ ; (iv)  $-x, y, z$ ; (v)  $x-1/2, y, z-1/2$ .

**Table S3.** Bond angles for  $(4\text{BrPEA})_3\text{BiI}_6$ .

Atom–Atom–Atom	Angle / °	Atom–Atom–Atom	Angle / °
I1 <sup>i</sup> –Bi1–I1	175.99 (5)	C2 <sup>iv</sup> –C1–C2	122.1 (11)
I2A–Bi1–I1	91.59 (2)	C1–C2–H2	120.8
I2A–Bi1–I1 <sup>i</sup>	91.59 (2)	C1–C2–C3	118.4 (9)
I2A <sup>iii</sup> –Bi1–I1 <sup>i</sup>	88.782 (16)	C3–C2–H2	120.8
I2A <sup>i</sup> –Bi1–I1 <sup>i</sup>	91.59 (2)	C2–C3–H3	119.6
I2A <sup>iii</sup> –Bi1–I1	88.782 (16)	C4–C3–C2	120.9 (9)
I2A <sup>ii</sup> –Bi1–I1	88.782 (16)	C4–C3–H3	119.6
I2A <sup>ii</sup> –Bi1–I1 <sup>i</sup>	88.782 (16)	C3 <sup>iv</sup> –C4–C3	119.3 (12)
I2A <sup>i</sup> –Bi1–I1	91.59 (2)	C3 <sup>iv</sup> –C4–C5	120.3 (6)
I2A <sup>ii</sup> –Bi1–I2A <sup>i</sup>	89.934 (9)	C3–C4–C5	120.3 (6)
I2A <sup>ii</sup> –Bi1–I2A <sup>iii</sup>	105.06 (8)	C4–C5–H5A <sup>iv</sup>	108.5 (4)
I2A <sup>i</sup> –Bi1–I2A	75.07 (8)	C4–C5–H5A	108.5
I2A <sup>iii</sup> –Bi1–I2A <sup>i</sup>	165.01 (8)	C4–C5–H5B	108.5
I2A <sup>ii</sup> –Bi1–I2A	165.01 (8)	C4–C5–H5B <sup>iv</sup>	108.5 (2)
I2A <sup>iii</sup> –Bi1–I2A	89.934 (9)	C4–C5–C6	114.9 (11)
I2A <sup>i</sup> –Bi1–I2B <sup>ii</sup>	110.47 (5)	H5A–C5–H5A <sup>iv</sup>	91.6
I2A <sup>ii</sup> –Bi1–I2B <sup>iii</sup>	84.52 (10)	H5A–C5–H5B	107.5
I2A <sup>ii</sup> –Bi1–I2B <sup>i</sup>	69.48 (5)	H5A <sup>iv</sup> –C5–H5B <sup>iv</sup>	107.5
I2A–Bi1–I2B <sup>i</sup>	95.53 (10)	H5A–C5–H5B <sup>iv</sup>	17.4
I2A–Bi1–I2B <sup>iii</sup>	110.47 (5)	H5B–C5–H5A <sup>iv</sup>	17.4
I2A–Bi1–I2B <sup>ii</sup>	174.45 (10)	H5B–C5–H5B <sup>iv</sup>	122.3
I2A <sup>iii</sup> –Bi1–I2B <sup>ii</sup>	84.52 (10)	C6–C5–H5A <sup>iv</sup>	121.7 (14)
I2A <sup>i</sup> –Bi1–I2B <sup>i</sup>	20.46 (5)	C6–C5–H5A	108.5
I2A <sup>iii</sup> –Bi1–I2B <sup>iii</sup>	20.54 (5)	C6–C5–H5B	108.5

I2A <sup>ii</sup> –Bi1–I2B <sup>ii</sup>	20.54 (5)	C6–C5–H5B <sup>iv</sup>	93.7 (13)
I2A <sup>i</sup> –Bi1–I2B <sup>iii</sup>	174.45 (10)	H1A–N1–H1B	109.5
I2A <sup>iii</sup> –Bi1–I2B <sup>i</sup>	174.53 (10)	H1A–N1–H1C	109.5
I2B <sup>ii</sup> –Bi1–I1	88.30 (2)	H1B–N1–H1C	109.5
I2B–Bi1–I1	91.061 (14)	C6–N1–H1A	109.5
I2B <sup>iii</sup> –Bi1–I1	88.30 (2)	C6–N1–H1B	109.5
I2B <sup>i</sup> –Bi1–I1	91.061 (14)	C6–N1–H1C	109.5
I2B <sup>ii</sup> –Bi1–I2B	154.00 (14)	C5–C6–H6A	108.4
I2B <sup>iii</sup> –Bi1–I2B	90.02 (2)	C5–C6–H6B	108.4
Bi1 <sup>v</sup> –I2A–Bi1	165.01 (8)	N1–C6–C5	115.4 (19)
Bi1 <sup>v</sup> –I2B–Bi1	154.00 (13)	N1–C6–H6A	108.4
C2–C1–Br1	119.0 (6)	N1–C6–H6B	108.4
C2 <sup>iv</sup> –C1–Br1	119.0 (6)	H6A–C6–H6B	107.5

Symmetry codes: (i)  $-x+1, -y+1, z$ ; (ii)  $x+1/2, y, z+1/2$ ; (iii)  $-x+1/2, -y+1, z+1/2$ ; (iv)  $-x, y, z$ ; (v)  $x-1/2, y, z-1/2$ .

**Table S4.** N–H $\cdots$ I hydrogen bond lengths extracted from the crystal structure of (4FPEA)<sub>3</sub>BiI<sub>6</sub> and (4BrPEA)<sub>3</sub>BiI<sub>6</sub>.

(4FPEA) <sub>3</sub> BiI <sub>6</sub>	(4BrPEA) <sub>3</sub> BiI <sub>6</sub>
2.738 Å	2.753 Å
2.77 Å	2.827 Å
2.807 Å	3.065 Å
2.842 Å	3.092 Å
2.904 Å	3.162 Å
2.928 Å	
2.945 Å	
2.992 Å	
2.993 Å	
3.087 Å	