Supporting Information

Insight into the effect of cerium dioxide nanoparticle modified cobalt phosphide as an efficient electrocatalyst for high-performance lithium-sulfur battery

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Fig

ure S1 SEM images of the (a) Co(OH)₂, (b) Co(OH)₂/CeO₂-5, (c) Co(OH)₂/CeO₂-10, (d) Co(OH)₂/CeO₂-15.



Figure S2 SEM images of the (a) CoP/CeO_2-0 , (b) CoP/CeO_2-5 .



Figure S3 N₂ adsorption/desorption curves and pore size distributi on

curves.



Figure S4 The digital image of Li_2S_6 solutions before and after adding CoP/CeO₂-0, CoP/CeO₂-5, CoP/CeO₂-10 and CoP/CeO₂-15.



Figure S5 CV curves of Li-S batteries with CoP/CeO₂-0 (a), CoP/CeO₂-5 (b)

and CoP/CeO₂-15 (c) modified separators at 0.1 mV s⁻¹.



Figure S6 CV curves of Li-S batteries with CoP/CeO₂-0 (a), CoP/CeO₂-5 (b) and CoP/CeO₂-15 (c) modified separators at various scan rates. The corresponding linear fitting of the peak currents with the square root of the scan rate of Li-S batteries with CoP/CeO₂-0 (d), CoP/CeO₂-5 (e) and CoP/CeO₂-15 (f) modified separators.



Figure S7 Li₂S nucleation and dissolution curves of the battery using

 CoP/CeO_2-0 (a, d), CoP/CeO_2-5 (b, e) and CoP/CeO_2-15 (c, f).



Figure S8 Discharge-charge curves of Li-S batteries with CoP/CeO₂-0

(a), CoP/CeO₂-5 (b) and CoP/CeO₂-15 (c) modified separators at various

rates.

Table S1 The electrochemical performance of reported Li-S batteries

Material	Sulfur loading (mg cm ⁻²)	Initial capacities (mAh g ⁻¹) at various rate (C)	Cycle number	Capacity decay (per cycle)	Ref.
CMK-3@CeO ₂	2.8	1248 mAh g ⁻¹ at 0.5 C 1000 mAh g ⁻¹ at 1C	200 800	0.106% 0.039%	[1]
P-CeO ₂	1.0	620 mAh g ⁻¹ at 1C	500	0.100%	[2]
CC@CeO ₂ @Li ₂ S ₆	1.0	1311 mAh g ⁻¹ at 0.2 C	100	0.050%	[3]
CeO₂@G	1.2	610 mAh g ⁻¹ at 2 C	1000	0.024%	[4]
CeO₂@CNTs/S	1.0	1437.6 mAh g ⁻¹ at 0.1 C	300	0.170%	[5]
Li ₂ S-CGA/CeO ₂	2.0	591 mAh g ⁻¹ at 1 C	200	0.050%	[6]
CeO _{2-x} @CNF	1.0	1017 mAh g ⁻¹ at 2 C	800	0.046%	[7]
CoP-Co ₃ S ₄	1.1	1516.9 mAh g ⁻¹ at 0.2 C	100	0.51%	[8]
h-O-CoP-NCG	4.2	834 mAh g ⁻¹ at 0.5 C	300	0.048%	[9]
	2.0	1214.5 mAh g ⁻¹ at 0.2 C	500	0.0972%	This work
	4.0	1041.6 mAh g ⁻¹ at 0.2 C	100	0.01%	

containing CeO₂ and/or CoP.

Supplemental References

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