

Supporting Information

In situ growth of δ -MnO₂/C fibers as binder-free and free-standing cathode for advanced aqueous Zn-ions batteries

Yan Li^a, Fei Zhang^a, Miaomiao Wu^a, Yong Guo^{a*}, Yuanyuan Liang^a, Reyihanguli Ababaikeri^a, Luyang Wang^a, Qiao Liu^a, Xingchao Wang^{ab*}

^aState Key Laboratory of Chemistry and Utilization of Carbon Based Energy Resources; Key Laboratory of Advanced Functional Materials, Autonomous Region; Institute of Applied Chemistry, College of Chemistry, Xinjiang University, Urumqi, 830046, Xinjiang, P. R. China.

^bKey Laboratory of Advanced Energy Materials Chemistry (Ministry of Education), Nankai University, Tianjin 300071, China

E-mail: xju1998@163.com; ichemabc@126.com

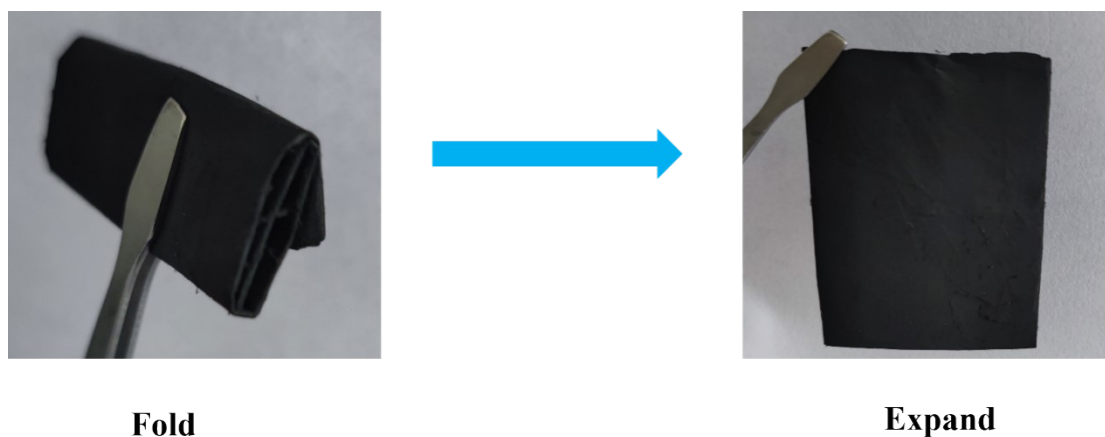


Figure S1. Flexibility of δ -MnO₂@CCFs.

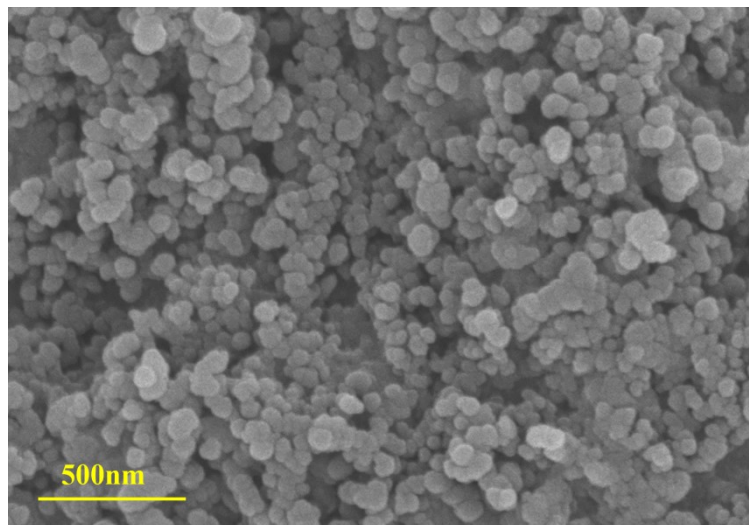


Figure S2. SEM of δ -MnO₂.

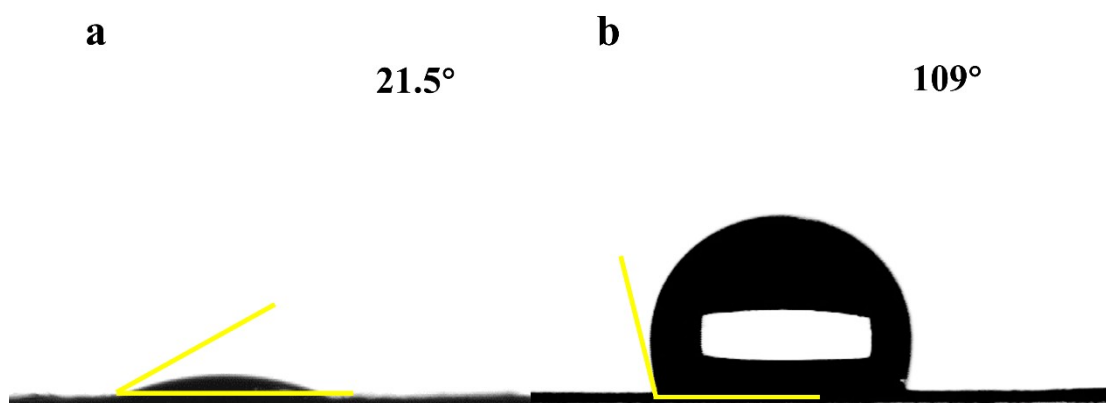


Figure S3 Contact angle of a) δ -MnO₂@CCFs and b) δ -MnO₂.

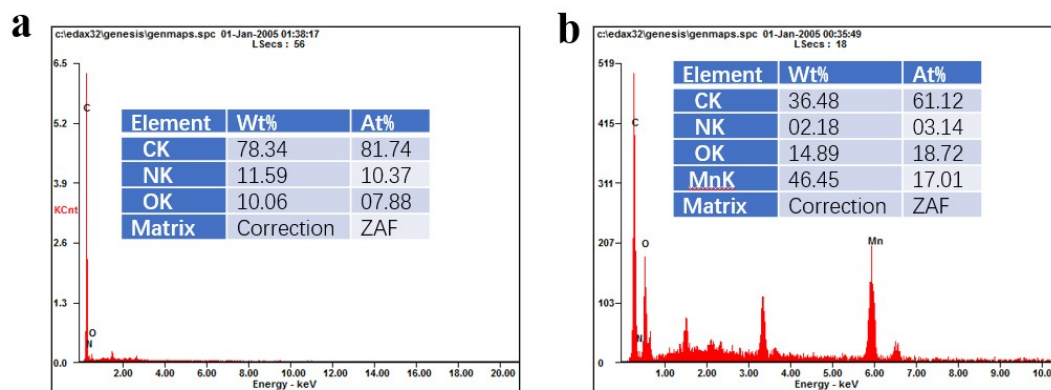


Figure S4. SEM-EDS of a) CCFs and b) δ -MnO₂@CCFs.

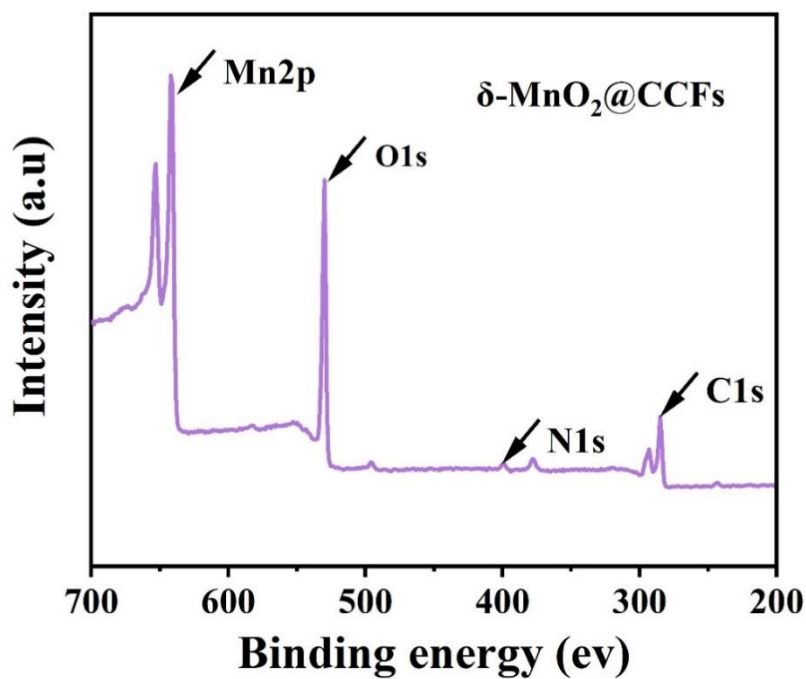


Figure S5. XPS survey spectra of δ -MnO₂@CCFs.

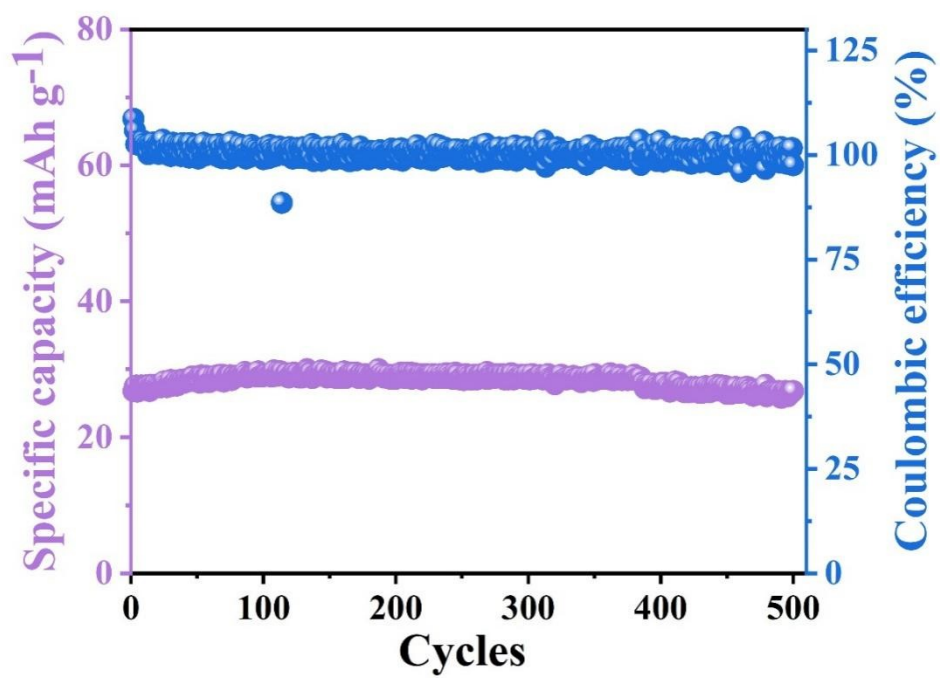


Figure S6. Cycling performance of CCFs at 1.0 A g⁻¹.

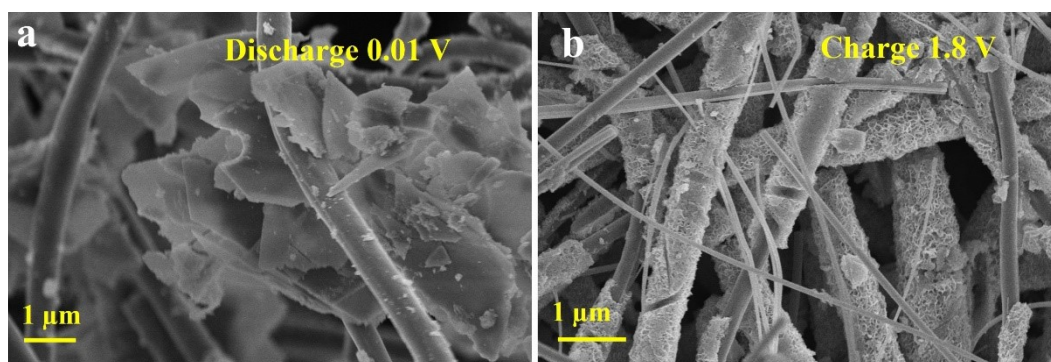


Figure S7. a) The SEM of discharge to 0.01 V and b) charge to 1.80 V of δ -MnO₂@CCFs.

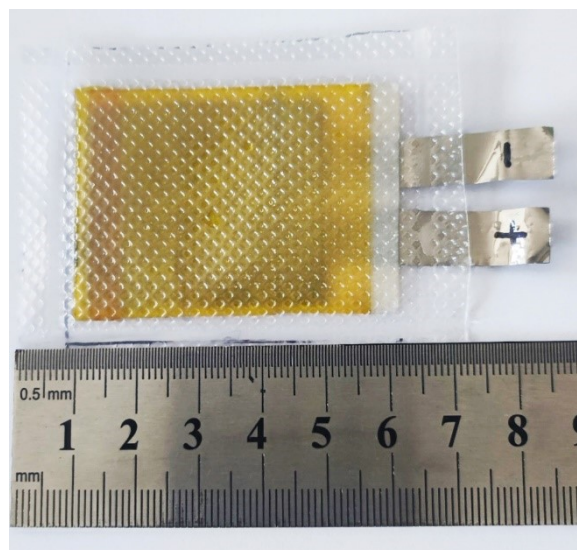


Figure S8 The photograph of a 3 cm × 5 cm flexible soft-packaged batteries.

Table S1 Comparison of electrochemical performance of δ -MnO₂@CCFs and reported MnO₂ electrode for ZIBs

Materials	Current density (A g ⁻¹)	Capacity (mAh g ⁻¹)
δ -MnO ₂ @CCFs	0.5	352 at 100th cycle (this work)
	3	132.3 at 7000th cycle (this work)
S-MnO ₂	1	122.9 at 1000th cycle ¹
MG10	1	80 at 1000th cycle ²
CN ₆₀₀ /MnO ₂	0.1	249 at 60th cycle ³
MnO ₂ nanospheres	0.3	358 at 100th cycle ⁴
MnO ₂ -NHCS	2	100 at 2000th cycle ⁵
MnO ₂ @PEDOT	2	213 at 2000th cycle ⁶
MGS	3	145 at 3000th cycle ⁷
S-MnO ₂	0.2	320 at 100th cycle ⁸
MnO ₂ @NC	2	90 at 2500th cycle ⁹

References

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