## Critical metals recovery from the spent lithium-ion batteries'

## leaching solution using electrodialysis technologies: strategies and

## challenges

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Methode	Target metals	Solution composition	Solvent liquid	pН	
Electrodialysis	Li, Ni, Co, Mn	$C_{Li}=1.00\times10^{-2} \text{ M},$ $C_{Ni}=C_{Co}=C_{Mn}=3.33\times10^{-3} \text{ M}$	EDTA	2	Sta
				3	Sta
			_	1.5	St
Solvent extraction	Li, Ni, Co	$\begin{array}{c} C_{Li}\!\!=\!\!2.8 \text{ g } L^{-1}\!, C_{Ni}\!\!=\!\!0.5 \text{g } L^{-1}\!, \\ C_{Co}\!\!=\!\!14.4 \text{ g } L^{-1} \end{array}$	Cyanex 272+TOA	~7	The The
Solvent extraction	Li, Co, Mn	$C_{Li}=C_{Co}=C_{Mn}=4\times 10^{-3} M$	Cyanex272+PC-88A	4.95	The order
Solvent extraction	Li, Ni, Co	$\begin{split} & C_{Fe}{=}3.6 \text{ g } L^{-1}, \ & C_{Cu}{=}1.8 \text{ g } L^{-1}, \\ & C_{Mn}{=}1.8 \text{ g } L^{-1}, \ & C_{Ni}{=}0.5 \text{ g } L^{-1}, \\ & C_{Co}{=}20.6 \text{ g } L^{-1}, \ & C_{Li}{=}2.5 \text{ g } L^{-1}. \end{split}$	P507	3.5	Reco Reco
Precipitation	Li, Ni, Co, Mn	100% of Li, Ni, Co, and Al in NCA material are leached out.	For Co: NaClO	3	The
			For Ni: NaOH	11	still j
Precipitation	Li, Ni, Co, Mn	LCO, LMO, and LCNM were mixed and leached according to the weight ratio of 1:1:1.	For Mn: KMnO <sub>4</sub>	2	The Mn, ( 97.43
			For Ni: dimethylglyoxime	9	
			For Co: NaOH	11	
			For Li: saturated Na <sub>2</sub> CO <sub>3</sub>	_	

**Table S1.** The performance data of several separation methods in the recovery of leaching solution from spent lithium-ion batteries

Method	Advantage	Disadvantage
	Selective ion removal	• Dependence on electric energy
Electrodialysis	• Low consumption of chemicals	Membrane fouling
	Scalable     Environmentally friendly	High initial cost
		Complicated process
	• Low cost	• High reagent consumption
Precipitation	• Low energy consumption	• Hard to precipitate only one ion
		Generation of solid waste
	High selectivity and recovery	• High price of solvent
Solvent extraction	• Low energy consumption	Consumption of organic solvents
	• Scalable	• Generation of toxic waste liquid

Table S2. Advantages and disadvantages of the treatment of leaching solutions methods