

## **Supporting information:**

### **Highly defective and conductive Cu-doped 1T/2H-MoS<sub>2</sub> nanosheets as high-capacity cathode materials for enhanced magnesium ion storage**

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*Ao Xu, and Yan Liu were the co-first authors in this work.*

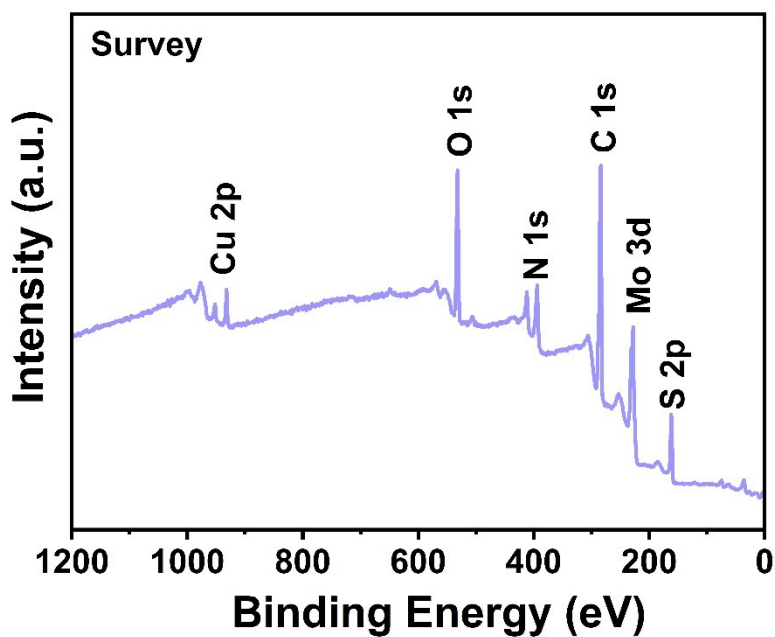


Fig. S1 XPS survey spectrum of Cu-MoS<sub>2</sub>-2.

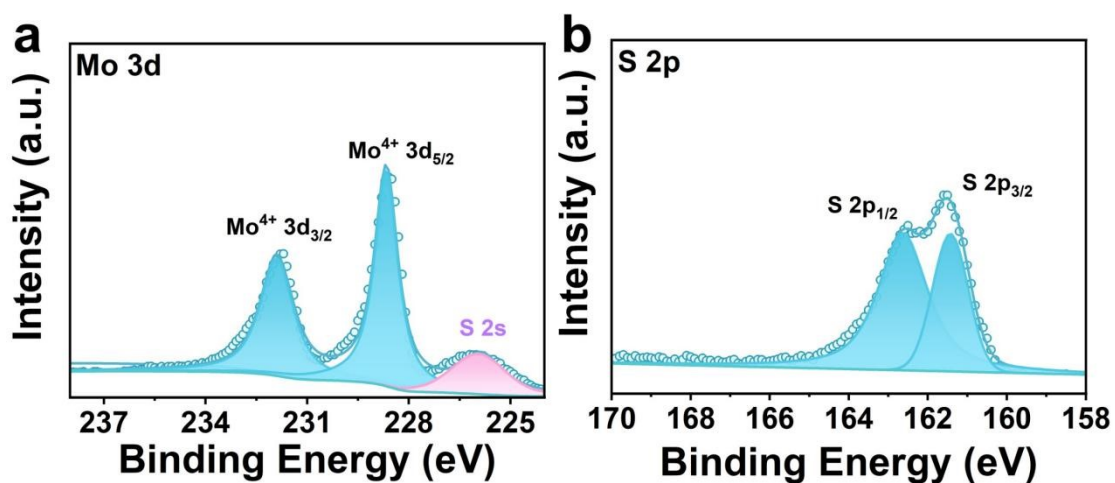


Fig. S2 XPS survey spectrum of MoS<sub>2</sub>.

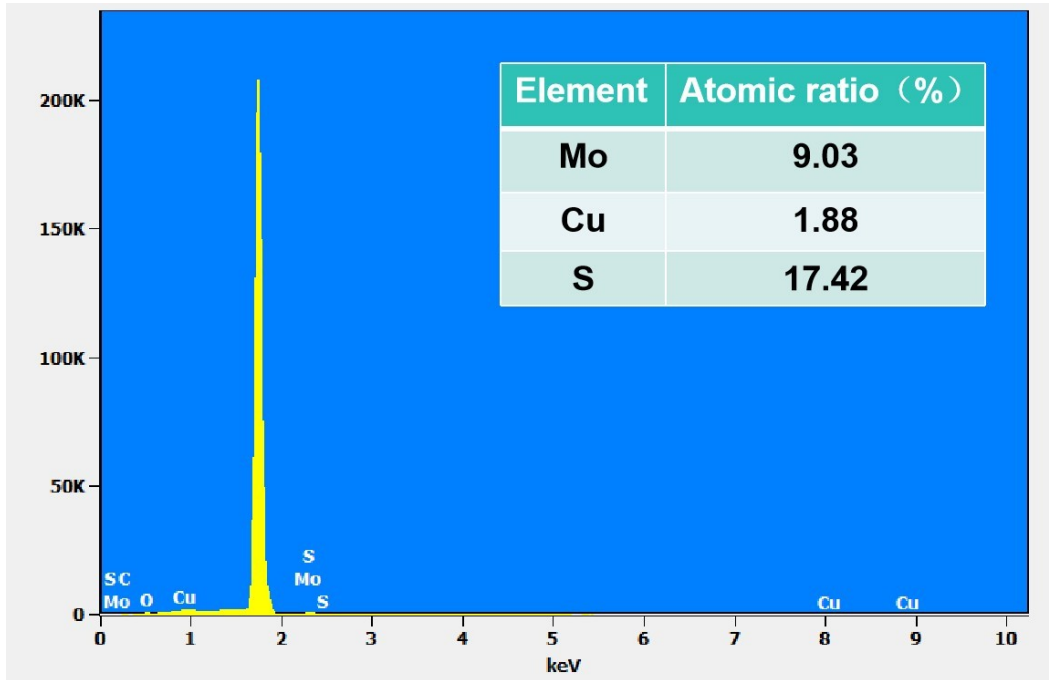


Fig. S3 EDS spectrum of Cu-MoS<sub>2</sub>-2.

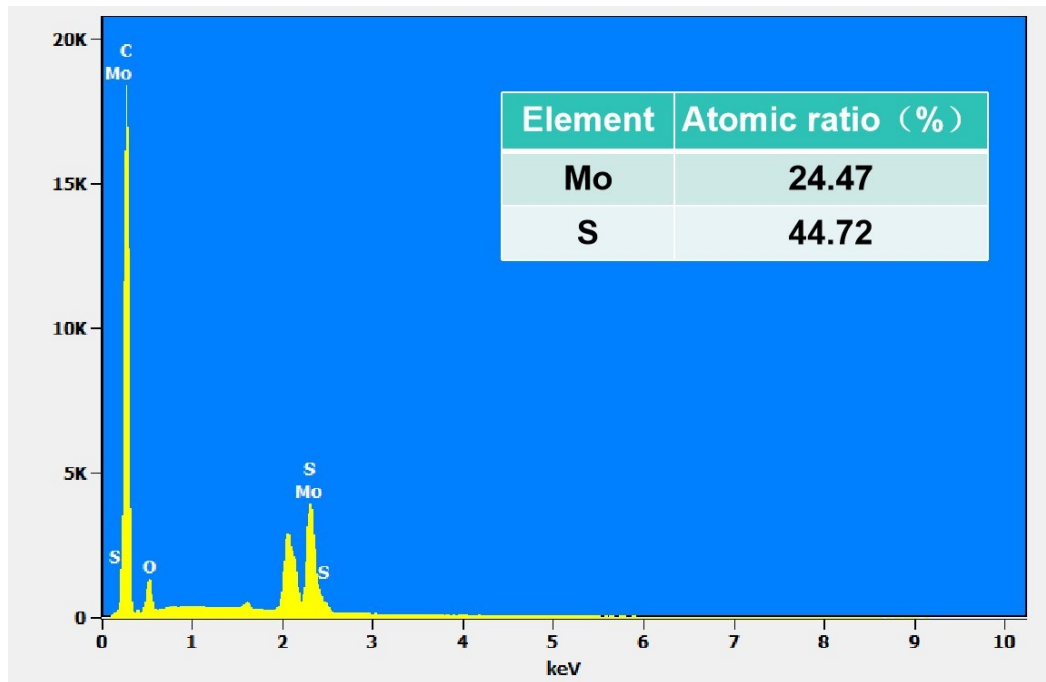


Fig. S4 EDS spectrum of MoS<sub>2</sub>.

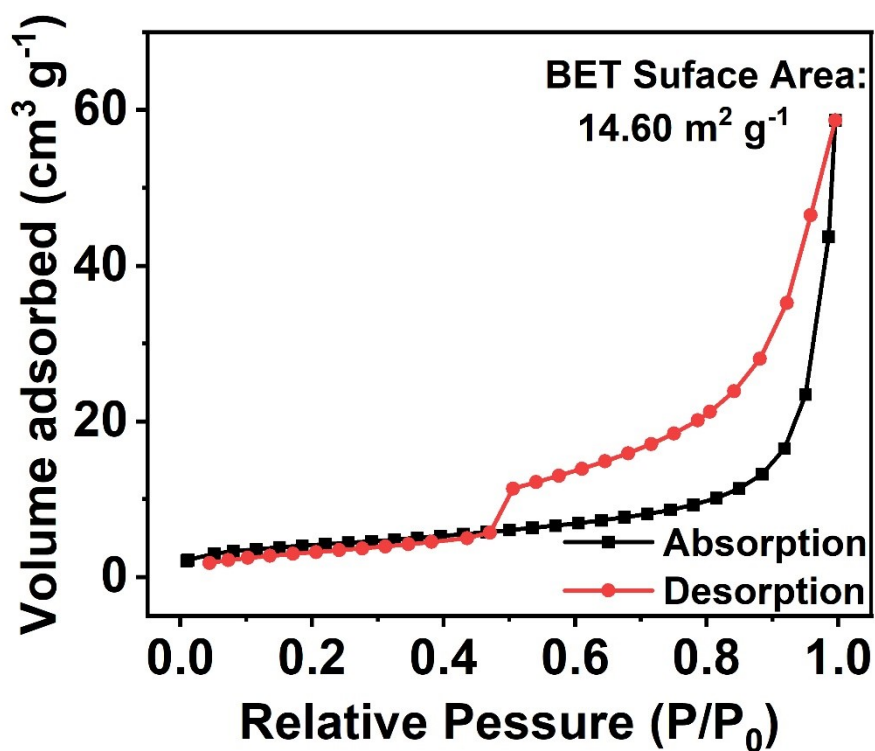


Fig. S5 Nitrogen isothermal adsorption/desorption curves of Cu-MoS<sub>2</sub>-2.

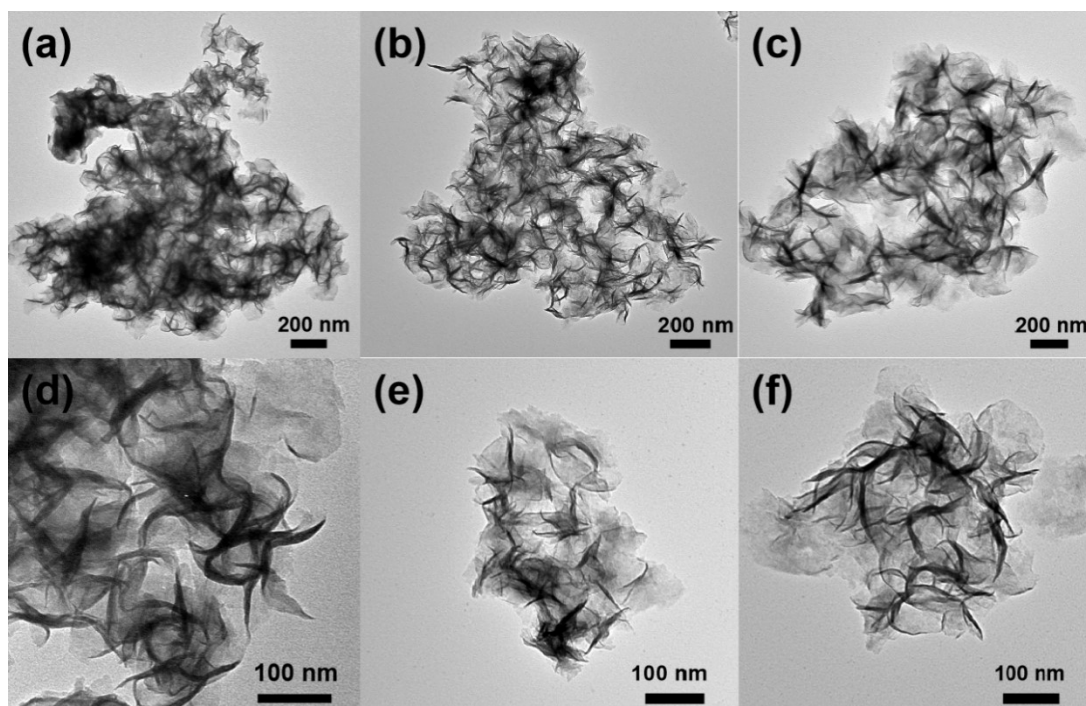


Fig. S6 TEM images of (a-d) MoS<sub>2</sub>, (b-e) Cu-MoS<sub>2</sub>-1, and (c-f) Cu-MoS<sub>2</sub>-3.

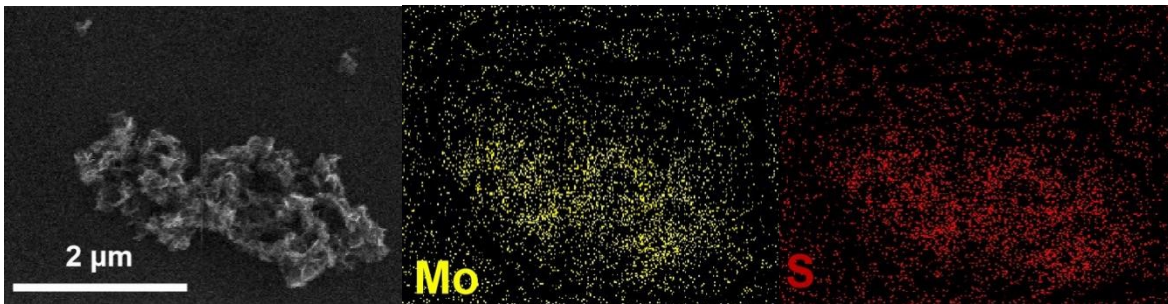


Fig. S7 EDS elemental mappings of MoS<sub>2</sub>.

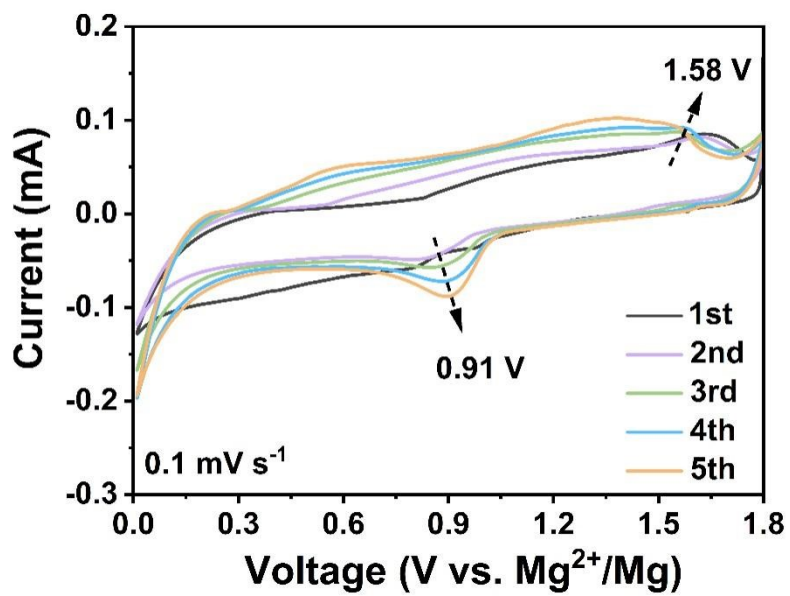
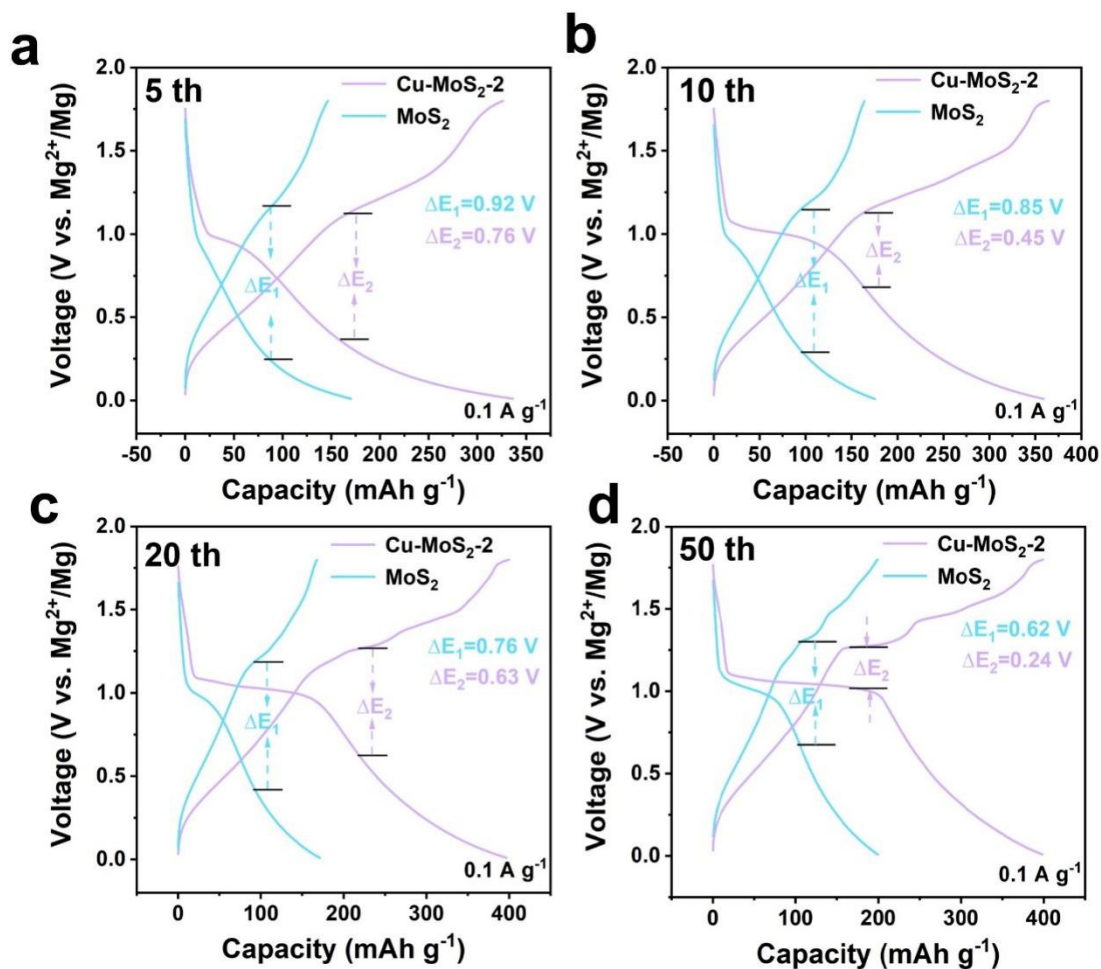
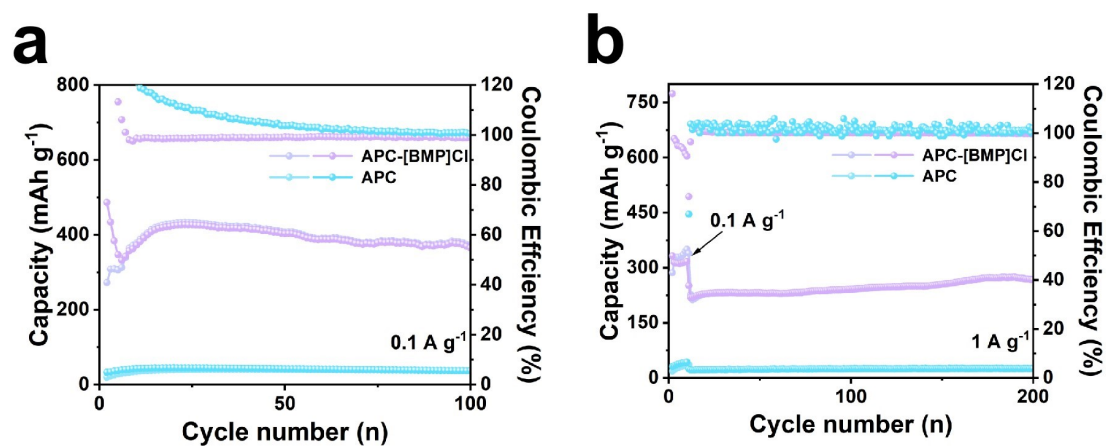


Fig. S8 CV curves of MoS<sub>2</sub> for the first five cycles.



**Fig. S9** (a-d) Discharge/charge profiles for different cycles at  $0.1 \text{ A g}^{-1}$  of  $\text{Cu-MoS}_2\text{-2}$  and  $\text{MoS}_2$ .



**Fig. S10** Cyclic performances of  $\text{Cu-MoS}_2\text{-2}$  in APC and APC-[BMP]Cl electrolytes: (a)  $0.1 \text{ A g}^{-1}$ , and (b)  $1.0 \text{ A g}^{-1}$ .

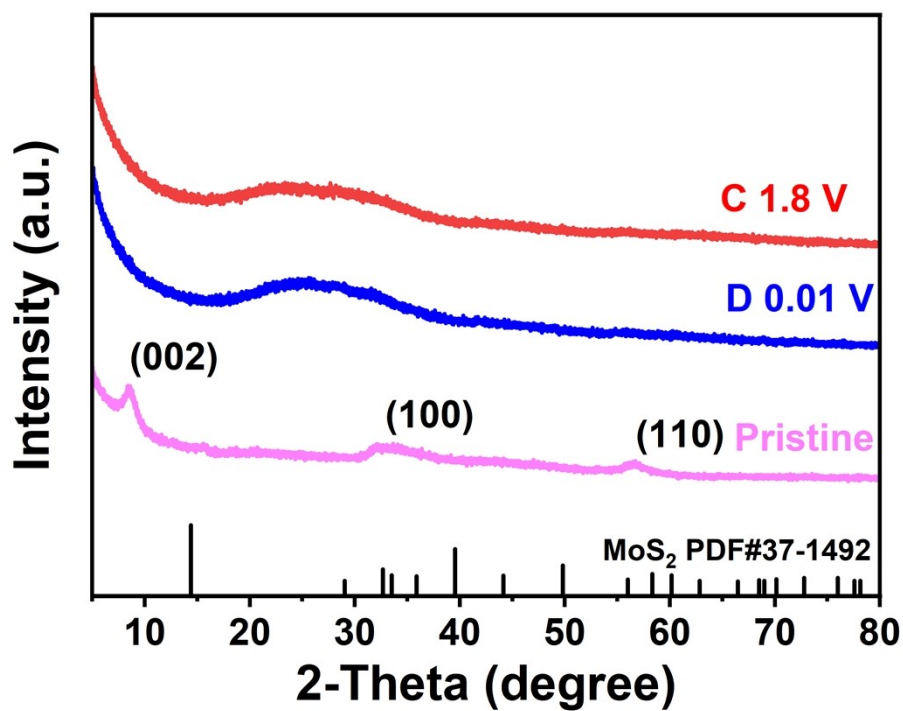


Fig. S11 *Ex-situ* XRD patterns of Cu-MoS<sub>2</sub>-2 at different discharge/charge states.

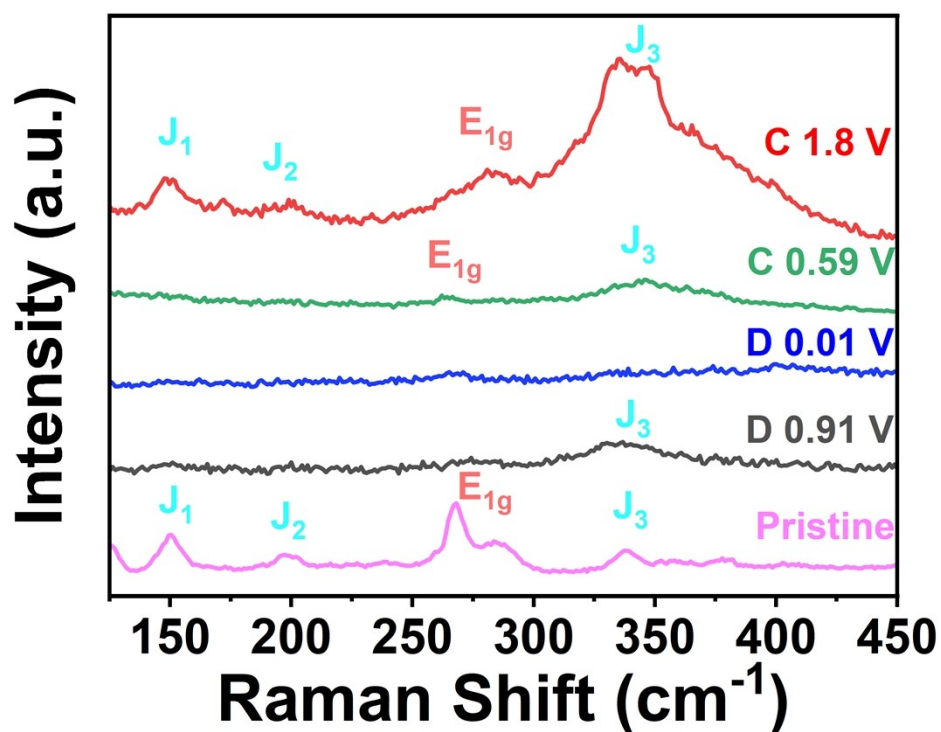


Fig. S12 *Ex-situ* Raman spectra of Cu-MoS<sub>2</sub>-2 at different discharge/charge states.

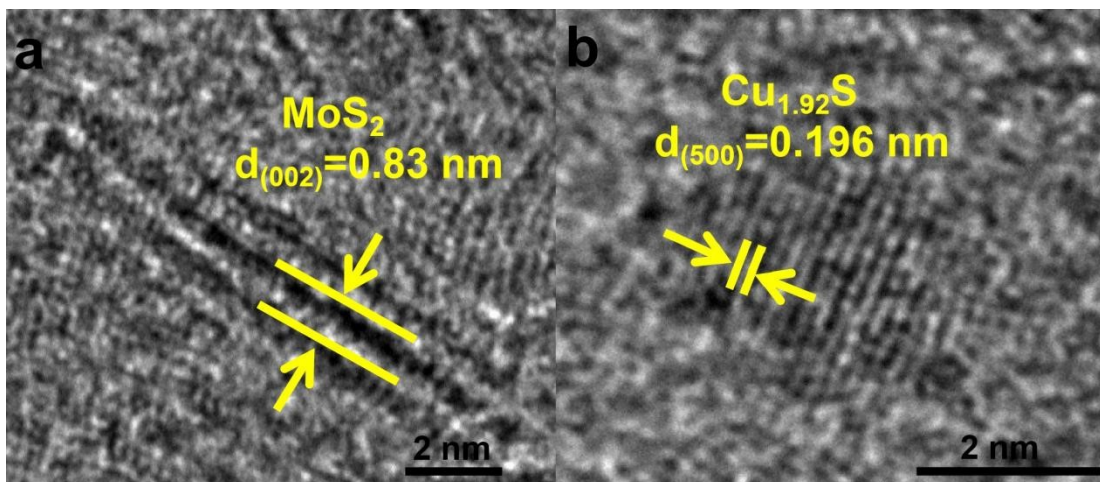


Fig. S13 (a) and (b) HRTEM images of Cu-MoS<sub>2</sub>-2 after charging to 1.8 V.

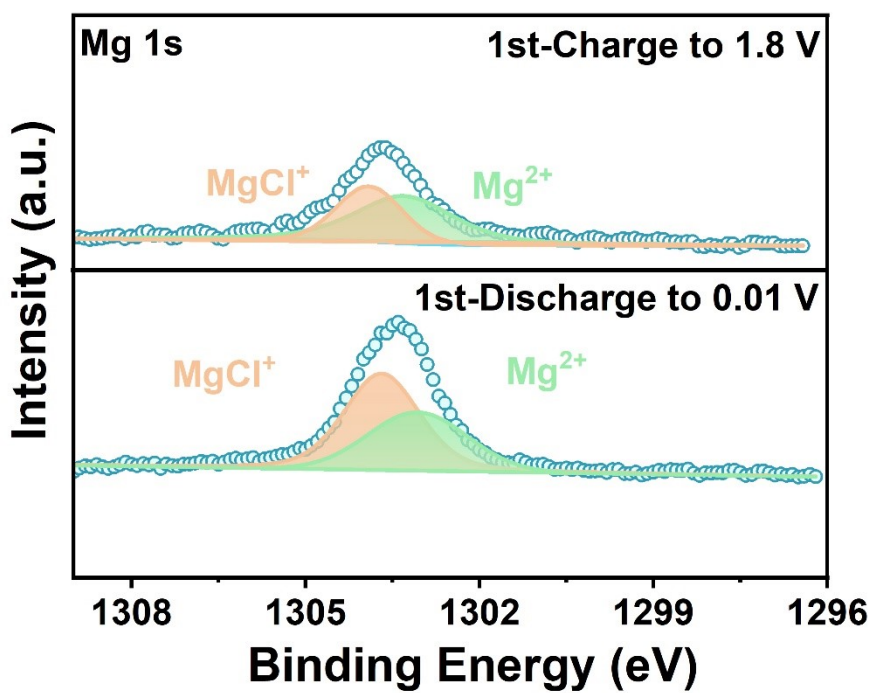


Fig. S14 *Ex-situ* XPS spectra of Mg 1s.



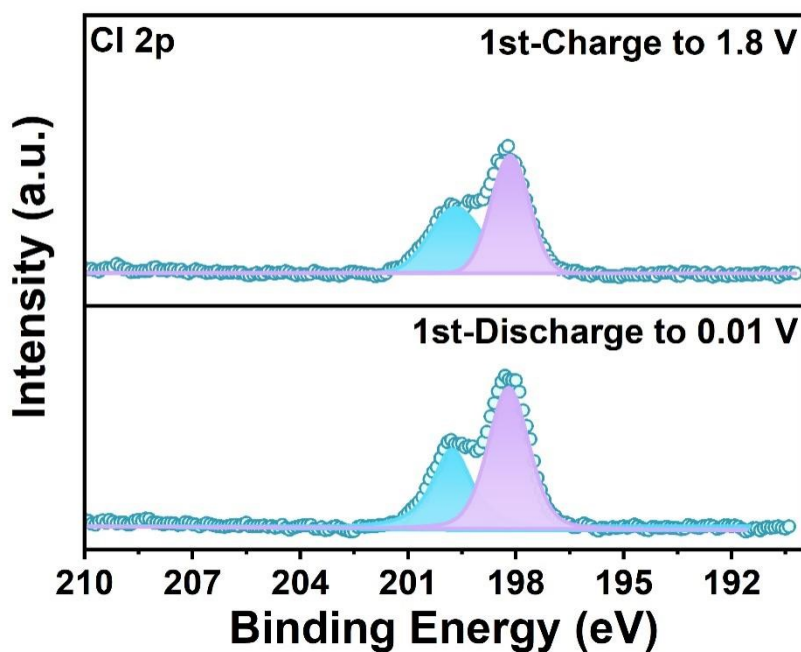


Fig. S15 *Ex-situ* XPS spectra of Cl 2p.

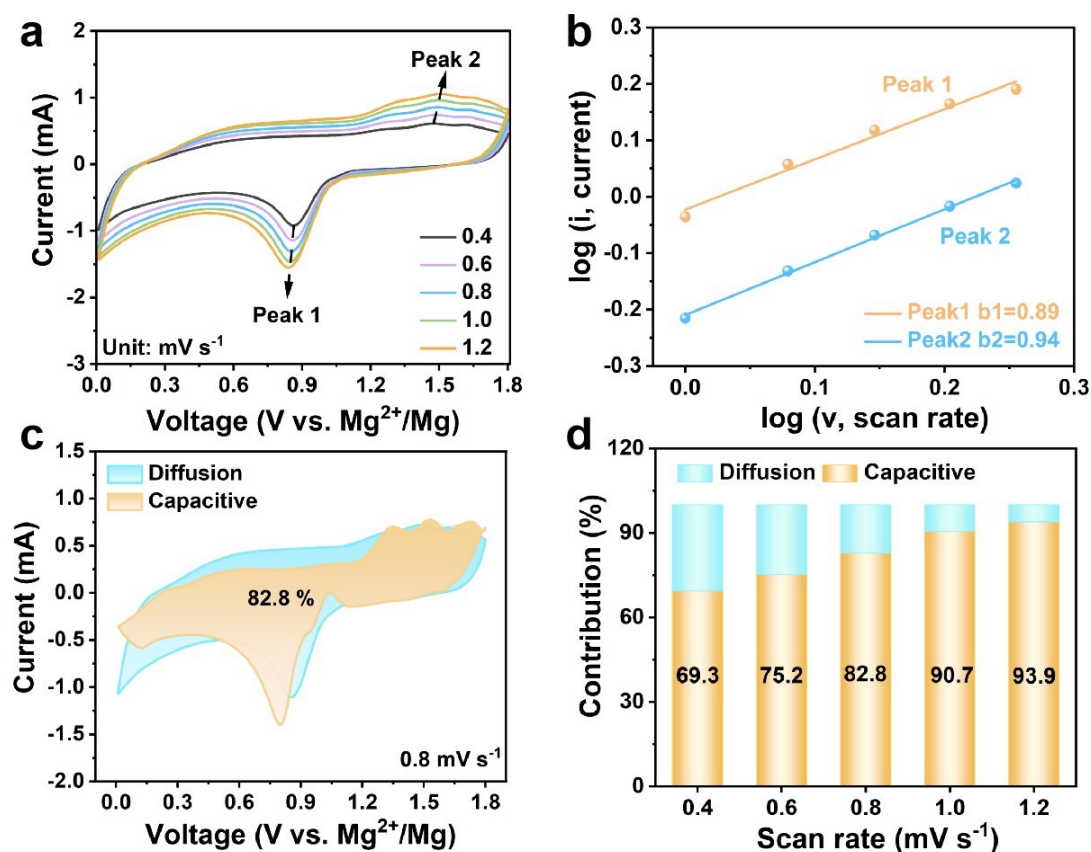


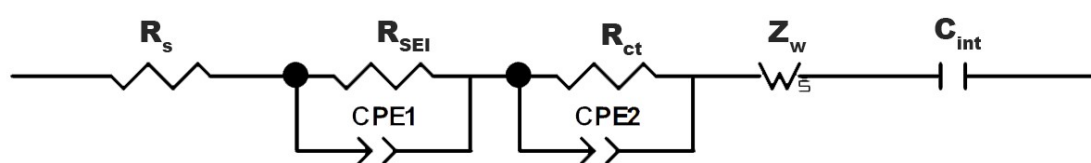
Fig. S16 Kinetic analysis of MoS<sub>2</sub>: (a) CV curves at different scan rates, (b)  $b$  values obtained from  $\log i$  versus  $\log v$ , (c) percentage of capacitive contributions at 0.8 mV s<sup>-1</sup>, (d) percentage of capacitive contributions at different scan rates.

**Table S1** Electric conductivity data at different pressures of MoS<sub>2</sub> and Cu-MoS<sub>2</sub>-2 via the four probe method.

Electrode materials	Height (mm)	Pressure (Mpa)	Temperature (°C)	Humidness (%RH)	Electric conductivity (S m <sup>-1</sup> )
<b>MoS<sub>2</sub></b>	0.94	18.3	25	50	0.0077
	0.97	16.07	25	50	0.0070
	0.98	14.07	25	50	0.0049
	1	12.07	25	50	0.0042
	1.04	10.17	25	50	0.0037
	1.07	8	25	50	0.0031
	1.1	6.11	25	50	0.0026
	1.12	4.3	25	50	0.0022
	1.14	2.24	25	50	0.0018
	<b>Cu-MoS<sub>2</sub>-2</b>	0.88	18.25	25	50
0.96		16.12	25	50	1.3736
0.97		14.28	25	50	1.1990
0.98		12.45	25	50	1.0330
1		10.24	25	50	0.4545
1.03		8.33	25	50	0.1852
1.16		6.06	25	50	0.0980
1.21		4.22	25	50	0.0390
	1.29	2.34	25	50	0.0128

**Table S2** Reaction impedance derived from EIS spectra.

Electrode materials	$R_s$ ( $\Omega$ )	$R_{SEI}$ ( $\Omega$ )	$R_{ct}$ ( $\Omega$ )
MoS <sub>2</sub>	26.68	18.93	28.17
Cu-MoS <sub>2</sub> -1	19.38	15.91	26.66
Cu-MoS <sub>2</sub> -2	18.36	11.17	26.38
Cu-MoS <sub>2</sub> -3	19.21	12.41	25.66



**Fig. S17** Diagram of equivalent circuit.