

Supporting Information

Large Optical Anisotropy in Noncentrosymmetric Phosphate with Pseudo 2D Intercalated Layer

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Table S1. Selected bond distances (Å) and angles (°) for (C₄H₇N₂)(H₂PO₄).

Atom to atom	Length (Å)	Atom to atom	Length (Å)
P(1)-O(1)	1.5339(18)	P(1)-O(2)	1.5653(19)
P(1)-O(3)	1.515(2)	P(1)-O(4)	1.5115(18)
N(1)-C(1)	1.463(3)	N(1)-C(2)	1.321(3)
N(1)-C(4)	1.367(4)	N(2)-C(2)	1.310(3)
N(2)-C(3)	1.358(4)	C(3)-C(4)	1.336(4)
Atom to atom to atom	Angle (°)	Atom to atom to atom	Angle (°)
O(1)-P(1)-O(2)	104.90(11)	O(1)-P(1)-O(3)	111.57(11)
O(1)-P(1)-O(4)	110.49(11)	O(2)-P(1)-O(3)	109.12(12)
O(2)-P(1)-O(4)	109.26(11)	O(3)-P(1)-O(4)	111.28(12)
C(1)-N(1)-C(2)	126.5(2)	C(1)-N(1)-C(4)	125.7(2)
C(2)-N(1)-C(4)	107.8(2)	C(2)-N(2)-C(3)	108.7(2)
N(1)-C(2)-N(2)	109.1(2)	C(4)-C(3)-N(2)	107.1(3)
C(3)-C(4)-N(1)	107.3(2)		

Table S2. Selected bond distances (Å) and angles (°) for (C₃H₅N₂)(H₂PO₄).

Atom to atom	Length (Å)	Atom to atom	Length (Å)
P(1)-O(1)	1.5043(15)	P(1)-O(2)	1.5630(17)
P(1)-O(3)	1.5059(19)	P(1)-O(4)	1.5739(18)
N(1)-C(1)	1.321(3)	N(1)-C(2)	1.369(3)
N(2)-C(1)	1.320(3)	N(2)-C(3)	1.376(3)
C(2)-C(3)	1.344(4)		
Atom to atom to atom	Angle (°)	Atom to atom to atom	Angle (°)
O(1)-P(1)-O(2)	109.42(10)	O(1)-P(1)-O(3)	114.90(10)
O(1)-P(1)-O(4)	106.69(9)	O(2)-P(1)-O(3)	107.19(10)
O(2)-P(1)-O(4)	107.35(10)	O(3)-P(1)-O(4)	111.05(11)
C(1)-N(1)-C(2)	108.4(2)	C(1)-N(2)-C(3)	108.4(2)
N(1)-C(1)-N(2)	109.0(2)	N(1)-C(2)-C(3)	107.3(2)
N(2)-C(3)-N(2)	106.8(2)		

Table S3. Atomic Coordinates ($\times 10^4$) and Equivalent Isotropic Displacement Parameters ($\text{\AA}^2 \times 10^3$) for $(\text{C}_4\text{H}_7\text{N}_2)(\text{H}_2\text{PO}_4)$ and $(\text{C}_3\text{H}_5\text{N}_2)(\text{H}_2\text{PO}_4)$. U_{eq} is defined as 1/3 of the trace of the orthogonalised U_{ij} tensor.

$(\text{C}_4\text{H}_7\text{N}_2)(\text{H}_2\text{PO}_4)$				
Atom	x	y	z	$U_{(\text{eq})}$ (\AA^2)
P(1)	2544.2(8)	5214.6(7)	364.8(7)	39.9(2)
O(1)	3885(2)	6136(2)	1053(2)	53.4(5)
O(2)	1546(2)	6120(2)	-725(2)	58.4(6)
O(3)	3275(2)	4047(2)	-355(2)	58.3(6)
O(4)	1419(2)	4734(2)	1396.7(17)	52.1(5)
C(1)	4369(4)	6960(3)	6907(3)	57.7(8)
C(2)	3168(4)	6271(3)	4584(3)	49.6(7)
C(3)	2163(4)	4322(3)	5021(3)	59.9(8)
C(4)	2871(4)	4825(3)	6197(3)	59.9(8)
N(1)	3499(3)	6053(2)	5911(2)	44.5(5)
N(2)	2361(3)	5241(2)	4028(2)	49.1(6)
$(\text{C}_3\text{H}_5\text{N}_2)(\text{H}_2\text{PO}_4)$				
Atom	x	y	z	$U_{(\text{eq})}$ (\AA^2)
P(1)	6728.3(6)	5772.1(3)	7181.4(14)	25.11(16)
O(1)	7530(2)	6532.4(8)	6778(4)	35.5(4)
O(2)	7669(2)	5304.8(9)	9463(4)	32.9(4)
O(3)	6635(2)	5286.9(10)	4556(4)	38.2(4)
O(4)	5000(2)	5936.4(9)	8408(4)	35.5(4)
C(1)	5351(3)	7570.4(14)	1587(5)	33.9(6)
C(2)	5861(3)	8787.1(14)	1852(7)	39.9(6)
C(3)	6782(3)	8432.8(14)	3787(6)	39.8(6)
N(1)	4980(2)	8239.6(11)	490(5)	33.9(5)
N(2)	6441(2)	7668.2(12)	3591(5)	36.4(5)

Table S4. Hydrogen-bonding interactions for (C₄H₇N₂)(H₂PO₄) and (C₃H₅N₂)(H₂PO₄).

(C ₄ H ₇ N ₂)(H ₂ PO ₄)				
D-H...A	<i>d</i> (D-H)	<i>d</i> (H...A)	<i>d</i> (D...A)	∠(DHA)
N(2)-H(2)...O(4)	1.05	1.44	2.491(2)	175
O(1)-H(1)...O(3)	0.86	1.81	2.668(3)	171
O(2)-H(2B)...O(4)	0.82	1.78	2.591(2)	169
C(3)-H(3)...O(1)	0.93	2.56	3.428(4)	157
C(4)-H(4)...O(3)	0.93	2.56	3.455(4)	162
(C ₃ H ₅ N ₂)(H ₂ PO ₄)				
D-H...A	<i>d</i> (D-H)	<i>d</i> (H...A)	<i>d</i> (D...A)	∠(DHA)
N(1)-H(1)...O(1)	0.86	1.86	2.714(3)	169
N(2)-H(2)...O(1)	0.86	1.80	2.657(3)	174
O(2)-H(2B)...O(3)	0.82	1.75	2.553(3)	166
O(4)-H(4)...O(3)	0.82	1.79	2.599(2)	169
C(1)-H(1A)...O(4)	0.93	2.40	3.252(3)	152
C(2)-H(2A)...O(2)	0.93	2.56	3.180(3)	124

Table S5. π - π stacking interactions in $(C_4H_7N_2)(H_2PO_4)$ and $(C_3H_5N_2)(H_2PO_4)$.

$(C_4H_7N_2)(H_2PO_4)$						
Number	Cg(I)->Cg(J)	<i>D</i>	η	<i>d</i>	ϑ	Strong or Weak
1	Cg(1)->Cg(1)	4.627	0	3.704	36.82	weak
2	Cg(1)->Cg(1)	3.679	0	3.537	15.98	strong
$(C_3H_5N_2)(H_2PO_4)$						
Number	Cg(I)->Cg(J)	<i>D</i>	η	<i>d</i>	ϑ	Strong or Weak
1	Cg(1)->Cg(1)	4.720	0	3.197	47.38	weak
2	Cg(1)->Cg(1)	4.720	0	3.197	47.38	weak
3	Cg(1)->Cg(1)	4.720	12.08	2.786	53.82	weak
4	Cg(1)->Cg(1)	4.720	12.08	3.259	46.33	weak

Note: 1. Cg(I) represents the center of mass(I) of the five-membered ring imidazole plane, and the coordinates of Cg(1) are (0.28, 0.53, 0.51) and (0.59, 0.81, 0.23) in $(C_4H_7N_2)(H_2PO_4)$ and $(C_3H_5N_2)(H_2PO_4)$, respectively; 2. *D*: the distance between the centroids of the rings (Å); 3. η : the dihedral angle (°) between plane I and J; 4. *d*: vertical distance between adjacent imidazole planes (Å); 5. ϑ : displacement angle, the angle between the connection of adjacent centroids and the vertical line formed by the corresponding plane (°).

Table S6. Optical properties for selected phase-matching UV NLO phosphates.

Compound	Space Group	Δn at 1064 nm	Reference
KH ₂ PO ₄ (KDP)	<i>I</i> 42 <i>d</i>	0.034 ^[b]	36
KTiOPO ₄ (KTP)	<i>Pna</i> 2 ₁	0.0921 ^[b]	35
LAP	<i>P</i> 2 ₁	0.075 ^[b]	87
Ba ₃ P ₃ O ₁₀ Cl	<i>Pca</i> 2 ₁	0.028 ^[a]	16
Ba ₃ P ₃ O ₁₀ Br	<i>P</i> 2 ₁ 2 ₁ 2 ₁	0.023 ^[a]	16
RbMgPO ₄ ·6H ₂ O	<i>Pmn</i> 2 ₁	0.005 ^[a]	17
CsMgPO ₄ ·6H ₂ O	<i>P</i> 6 ₃ / <i>mmc</i>	0.006 ^[a]	17
NH ₄ MgPO ₄ ·6H ₂ O	<i>Pmn</i> 2 ₁	0.0063 ^[a]	18
KMgPO ₄ ·6H ₂ O	<i>Pmn</i> 2 ₁	0.01 ^[a]	18
LiCs ₂ PO ₄	<i>Cmc</i> 2 ₁	0.01 ^[a]	85
RbNaMgP ₂ O ₇ (LTP)	<i>Pna</i> 2 ₁	0.031 at 532 nm ^[a]	21
RbNaMgP ₂ O ₇ (HTP)	<i>Ccm</i> 2 ₁	0.035 at 532 nm ^[a]	21
NaNH ₄ PO ₃ F·H ₂ O	<i>Pc</i>	0.053 at 589.3 nm ^[b]	84
RbTiOPO ₄	<i>Pna</i> 2 ₁	0.0884 ^[b]	36
Na ₃ TaP ₂ O ₉	<i>P</i> 2 ₁ 2 ₁ 2 ₁	0.1101 (static) ^[a]	37
K ₂ ZnMoP ₂ O ₁₀	<i>P</i> 2 ₁ 2 ₁ 2 ₁	0.0534 at 450.2 nm ^[b]	38
Na ₁₂ (NbO) ₃ (PO ₄) ₇	<i>Pna</i> 2 ₁	0.03 (static) ^[a]	39
Rb ₃ PbBi(P ₂ O ₇) ₂	<i>P</i> 2 ₁ 2 ₁ 2 ₁	0.031 ^[b]	40
Cs ₃ PbBi(P ₂ O ₇) ₂	<i>P</i> 2 ₁ 2 ₁ 2 ₁	0.02 ^[b]	40
Rb ₃ BaBi(P ₂ O ₇) ₂	<i>P</i> 2 ₁	0.025 ^[a]	41
Cs ₃ BaBi(P ₂ O ₇) ₂	<i>P</i> 2 ₁ 2 ₁ 2 ₁	0.025 ^[a]	41
K ₂ Sb(P ₂ O ₇)F	<i>P</i> 4 <i>bm</i>	0.157 at 546 nm ^[b]	42
Rb ₂ Sb(P ₂ O ₇)F	<i>P</i> 4 <i>bm</i>	0.15 at 546 nm ^[b]	43
Sn ₂ PO ₄ Cl	<i>Pna</i> 2 ₁	0.162 at 546 nm ^[b]	44
β -Cd(PO ₃) ₂	<i>P</i> 2 ₁ 2 ₁ 2 ₁	0.059 ^[a]	1
(NH ₄) ₃ (H ₃ O)Zn ₄ (PO ₄) ₄	<i>P</i> 6 ₃	0.032 ^[a]	45
LiHgPO ₄	<i>P</i> 42 ₁ <i>m</i>	0.068 ^[b]	46
[C(NH ₂) ₃] ₆ (PO ₄) ₂ ·3H ₂ O	<i>Cc</i>	0.078 at 546 nm ^[b]	55
4HPP	<i>P</i> 2 ₁ 2 ₁ 2 ₁	0.25 ^[a]	50
2APP	<i>P</i> 2 ₁	0.225 ^[b]	51
[C(NH ₂) ₃] ₂ PO ₃ F	<i>Cm</i>	0.039 at 532 nm ^[a]	52
[C(NH ₂) ₃] ₃ PO ₄ ·2H ₂ O	<i>Pna</i> 2 ₁	0.055 at 546 nm ^[b]	49
(C ₃ H ₇ N ₆) ₆ (H ₂ PO ₄) ₄ (HPO ₄)·4H ₂ O	<i>P</i> 2 ₁	0.220 ^[a]	53
(C ₃ H ₅ N ₂)(H ₂ PO ₄)	<i>Pna</i> 2 ₁	0.15 at 546 nm ^[b]	this work

Notes:

[a] Cal. Birefringence

[b] Exp. Birefringence

LAP: (H₂N)₂CNH(CH₂)₃CH(NH₃)COO·H₂PO₄·H₂O4HPP: (C₅H₆ON)(H₂PO₄)2APP: (C₄H₆N₃)(H₂PO₃)

Table S7. The linear and nonlinear optical properties of $(\text{C}_3\text{H}_5\text{N}_2)(\text{H}_2\text{PO}_4)$ were calculated by real-space atom cutting method.

	Total	$[\text{C}_3\text{H}_5\text{N}_2]^+$	$[\text{H}_2\text{PO}_4]^-$
n_x	1.51	1.42	1.22
n_y	1.578	1.46	1.22
n_z	1.50	1.40	1.20
$(\text{C}_3\text{H}_5\text{N}_2)(\text{H}_2\text{PO}_4)$ $\Delta n @ 546 \text{ nm}$	0.078	0.057 (78.08%) ^a	0.016 (21.92%)
d_{15}	-0.045	0.034	-0.053
d_{24}	0.19	0.11	0.25
d_{33}	-0.49	-0.35 (55.56%) ^b	-0.28 (44.44%)

Notes:

d values in pm/V.

^aThe percentage in the table represents the contribution of each component to Δn .

^bThe percentage in the table represents the contribution of each component to d_{33} .

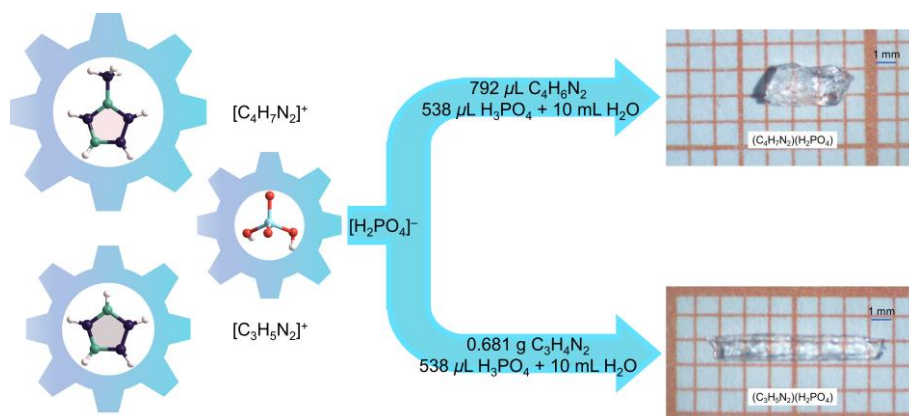


Figure S1. Unpolished photos of $(C_4H_7N_2)(H_2PO_4)$ crystal (top) and $(C_3H_5N_2)(H_2PO_4)$ crystal (bottom).

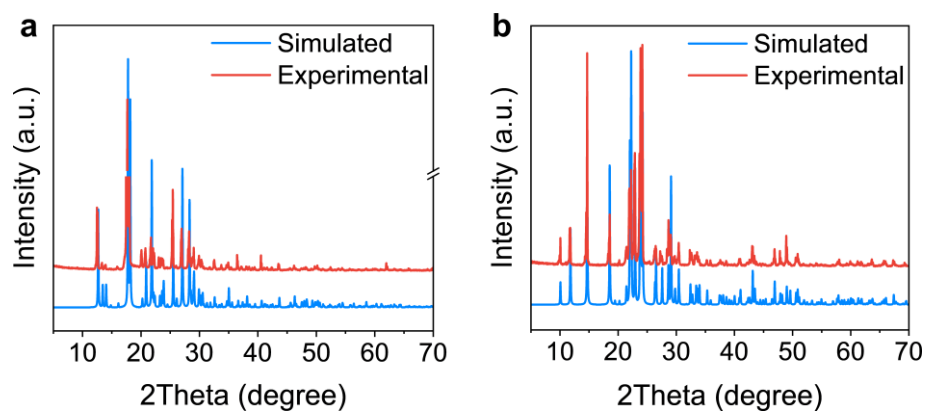


Figure S2. Simulated and experimental powder X-ray diffraction patterns of $(C_4H_7N_2)(H_2PO_4)$ (a) and $(C_3H_5N_2)(H_2PO_4)$ (b).

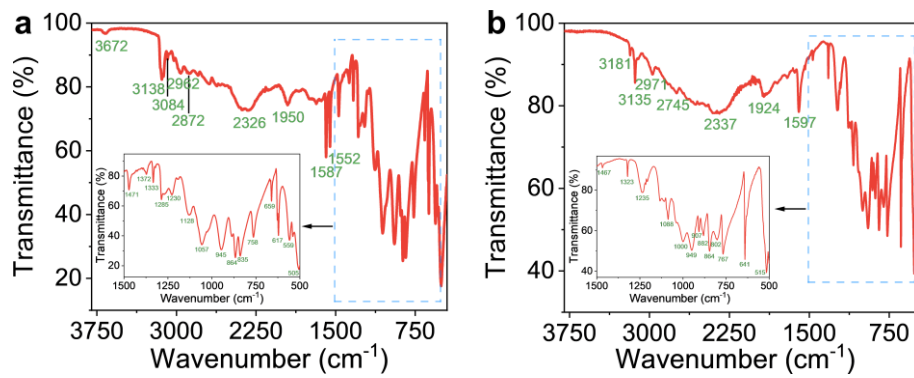


Figure S3. IR spectra of (C₄H₇N₂)(H₂PO₄) (a) and (C₃H₅N₂)(H₂PO₄) (b).

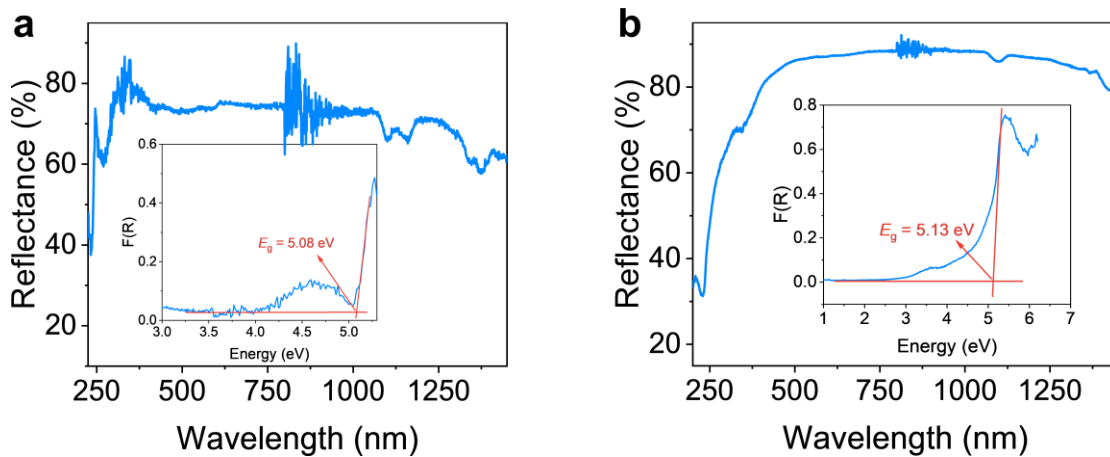


Figure S4. UV-Vis-NIR diffuse reflectance spectra of compound $(C_4H_7N_2)(H_2PO_4)$ (a) and $(C_3H_5N_2)(H_2PO_4)$ (b). The experimental optical band gaps are calculated by the Kubelka-Munk formula, $F(R) = (1 - R)^2/(2R)$, where R is the reflectance.

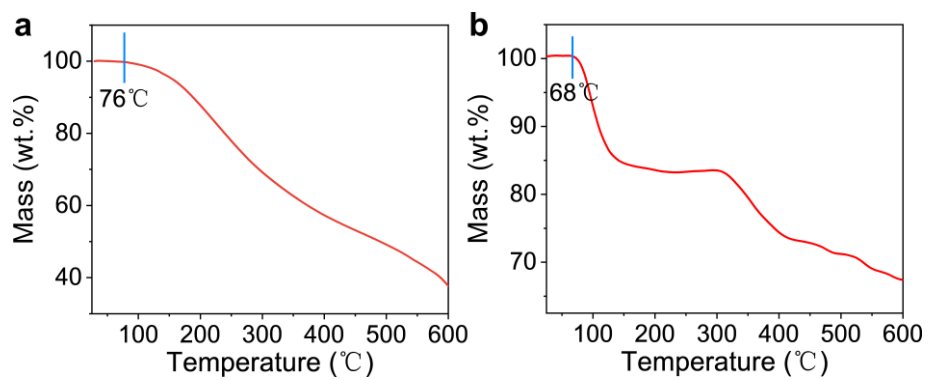


Figure S5. Thermogravimetry (TG) curve of $(C_4H_7N_2)(H_2PO_4)$ (a) and $(C_3H_5N_2)(H_2PO_4)$ (b) under a N_2 atmosphere.

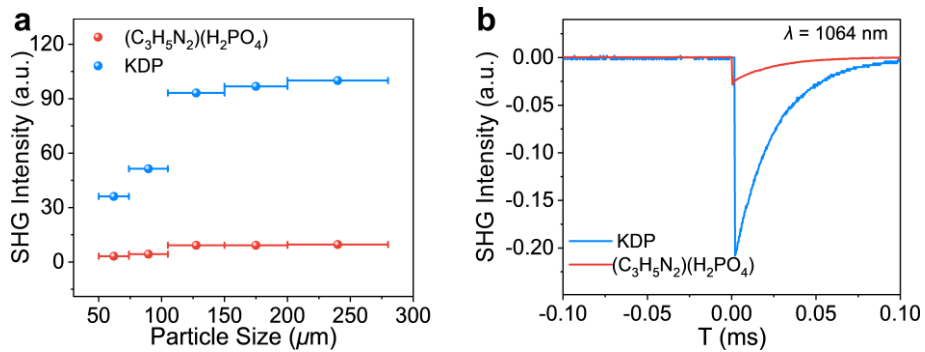


Figure S6. (a) Phase-matching curves of $(\text{C}_3\text{H}_5\text{N}_2)(\text{H}_2\text{PO}_4)$ with 1064 nm laser radiation. (b) Oscilloscope traces of the SHG signals for powders of $(\text{C}_3\text{H}_5\text{N}_2)(\text{H}_2\text{PO}_4)$ (105–150 μm) with 1064 nm laser radiation.

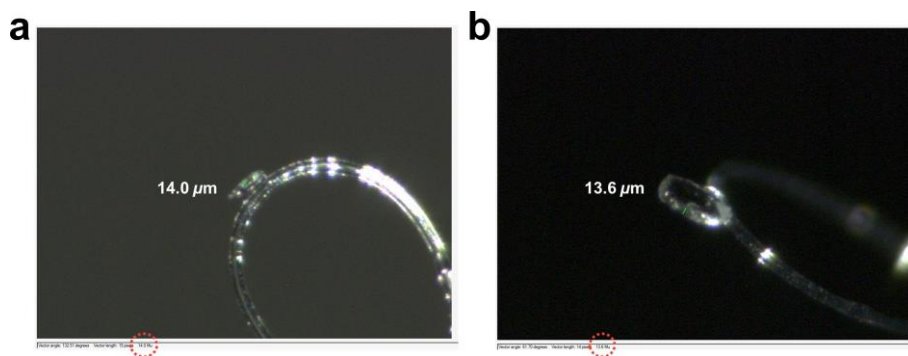


Figure S7. Photograph of the crystal size of $(C_4H_7N_2)(H_2PO_4)$ (a) and $(C_3H_5N_2)(H_2PO_4)$ (b).