

## Supporting Information:

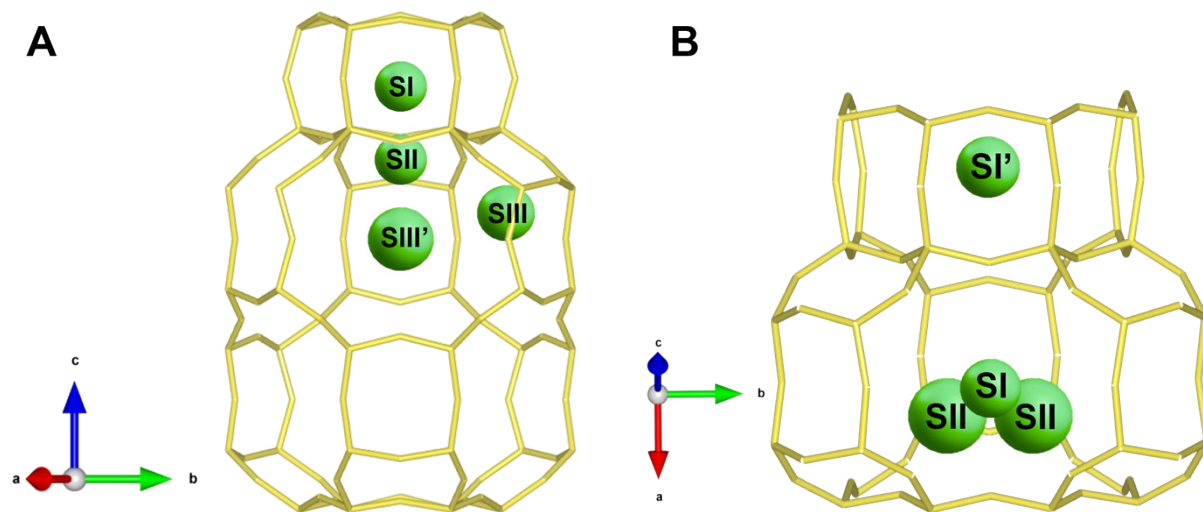
### Synthesis and properties of pure and intergrown CHA/PHI zeolites from inorganic multi-cation colloidal suspensions

Aymeric Magisson,<sup>a</sup> Edwin B. Clatworthy,<sup>\*a</sup> Diógenes Honorato Piva,<sup>a</sup> Sajjad Ghojavand,<sup>a</sup> Oleg I. Lebedev<sup>b</sup> and Svetlana Mintova<sup>\*a</sup>

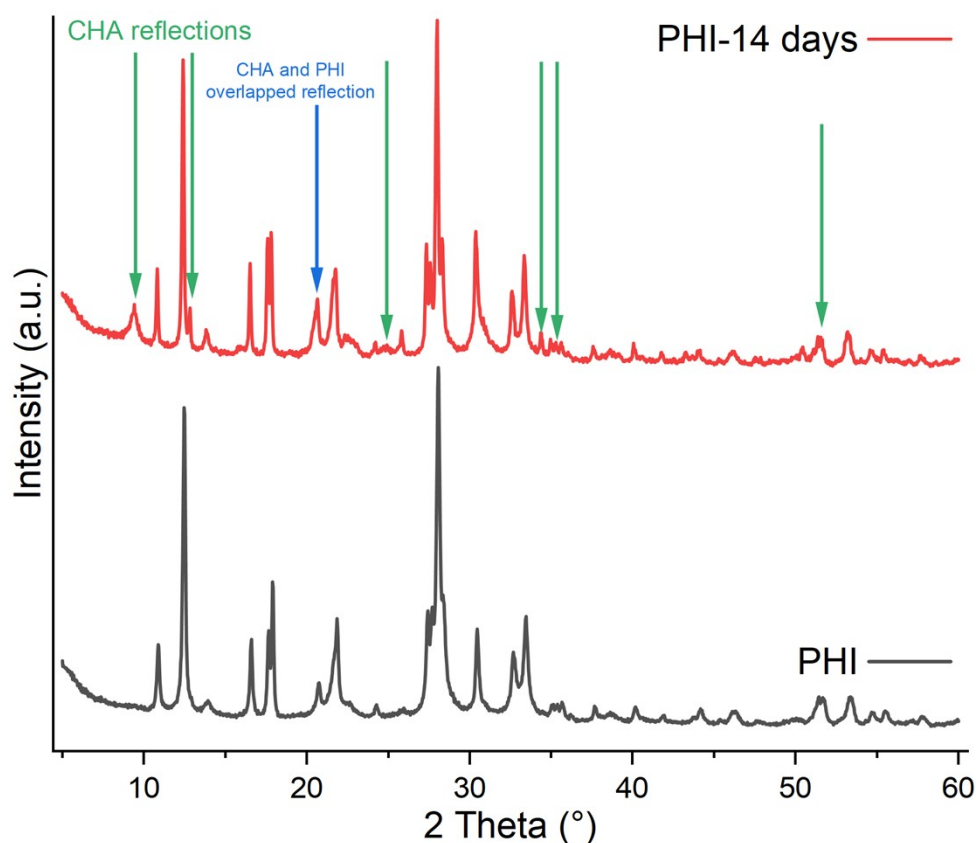
<sup>a</sup>Université de Caen Normandie, ENSICAEN, CNRS, LCS, Laboratoire Catalyse et Spectrochimie, 14000 Caen, France

<sup>b</sup>Université de Caen Normandie, ENSICAEN, CNRS, CRISMAT, Laboratoire de Cristallographie et Sciences des Matériaux, 14000 Caen, France

#### Additional figures and tables



**Figure S1.** Extra-framework cations site location in the chabazite (A) and in the phillipsite (B). The structures are viewed as a projection along  $[\bar{1}00]$ .



**Figure S2.** Powder XRD pattern of a chabazite/phillipsite zeolite compared to the pattern of a pure phillipsite pattern. The green arrows indicate peaks attributed to the CHA phase and the blue arrow indicates an overlapped CHA and PHI peak.

**Table S1.** Screening of the synthesis parameters yielding pure CHA, PHI and intergrown zeolites.

	Chemical composition								Crystallisation		Zeolitic product	
	Cs <sub>2</sub> O	K <sub>2</sub> O	Na <sub>2</sub> O	BaO	CaO	SiO <sub>2</sub>	Al <sub>2</sub> O <sub>3</sub>	H <sub>2</sub> O	Ageing time (days)	HT time (h)	XRD phase	CHA/PHI
(1) CHA <sup>42</sup>	0.2	1.5	6	-	-	16	0.7	140	7	7	CHA	1
(2) Chabazite <sup>40</sup>	0.15	1.35	6	-	-	16	0.6	120	12	8	CHA	1
(3) CHA-Na,Cs-1.3K,0.2Ba <sup>42</sup>	0.2	1.3	6	0.2	-	16	0.7	140	7	7	CHA	1
(4) CHA-Na,K,Cs-0.2Ca <sup>42</sup>	0.2	1.5	6	-	0.2	16	0.7	140	7	10	CHA	1
(5) CHA-Na,K,Cs-0.2Ba <sup>42</sup>	0.2	1.5	6	0.2	-	16	0.7	140	7	4	CHA	1
(6) CHA-Na,K	-	1.5	6	-	-	16	0.7	140	7	7	CHA	1
(7) CHA-Na,Cs-1.3K,0.2Ca <sup>42</sup>	0.2	1.3	6	-	0.2	16	0.7	140	7	7	CHA & PHI	0.81
(8) CHA-Na,K,Cs-0.50Al	0.2	1.5	6	-	-	16	0.5	140	7	7	CHA & PHI	0.74
(9) CHA-Na,K,Cs-0.25Al	0.2	1.5	6	-	-	16	0.25	140	7	7	CHA & PHI	0.69
(10) CHA-Na,K,Cs-0.5Ba	0.2	1.5	6	0.5	-	16	0.7	140	7	4	CHA & PHI	0.59
(11) CHA-Na,K,Cs-1.0Ba	0.2	1.5	6	1	-	16	0.7	140	7	4	CHA & PHI	0.42
(12) PHI	-	1.5	6	-	0.2	16	0.7	140	7	7	PHI	0
(13) PHI-14 days *	-	1.5	6	-	0.2	16	0.7	140	14	7	CHA & PHI	0.29

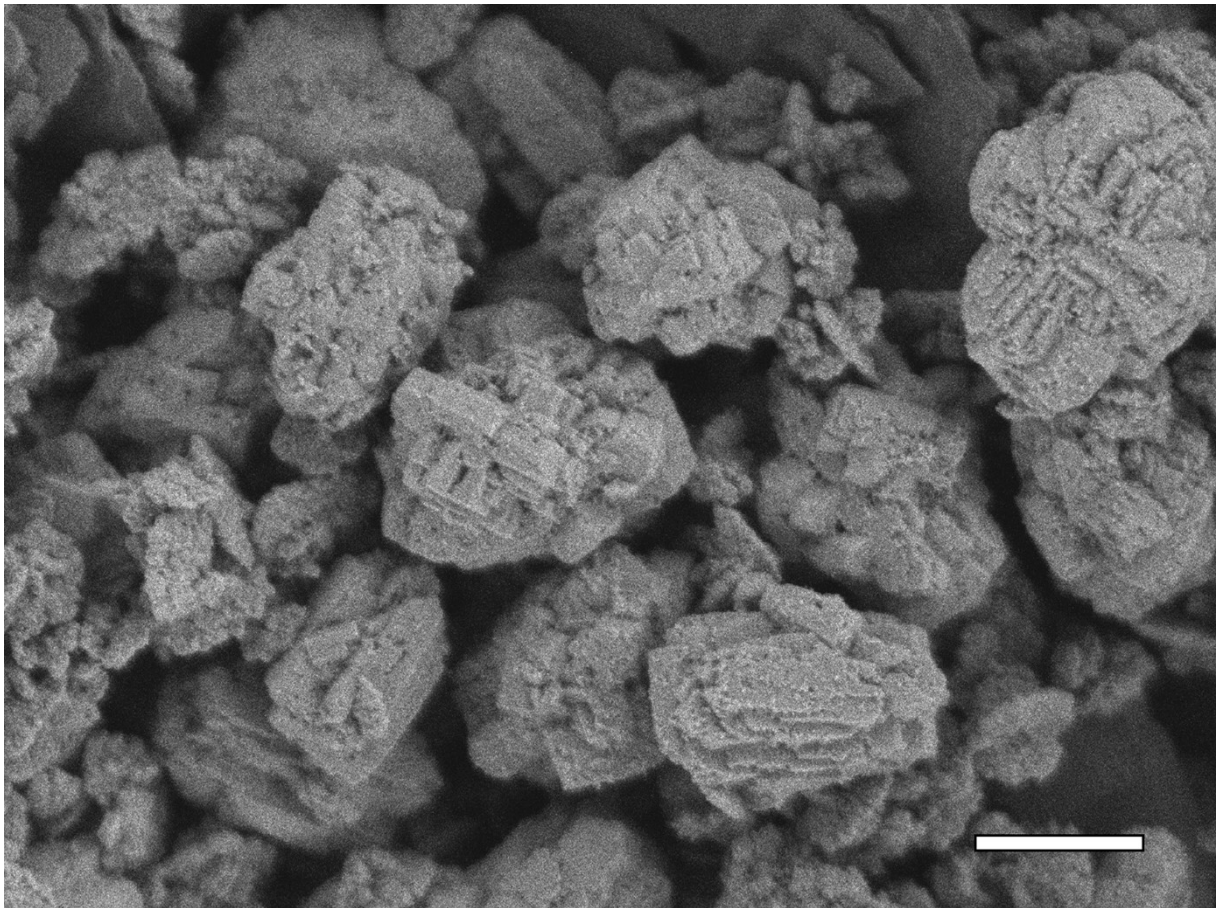
\* The suffix "14 days" refers to the ageing time that was prolonged for fourteen days instead of seven

**Table S2A.** Scherrer crystalline domain size calculation of the PHI-14 days sample. The Miller indices were chosen to average out over crystal orientation and not overlap with those of chabazite.

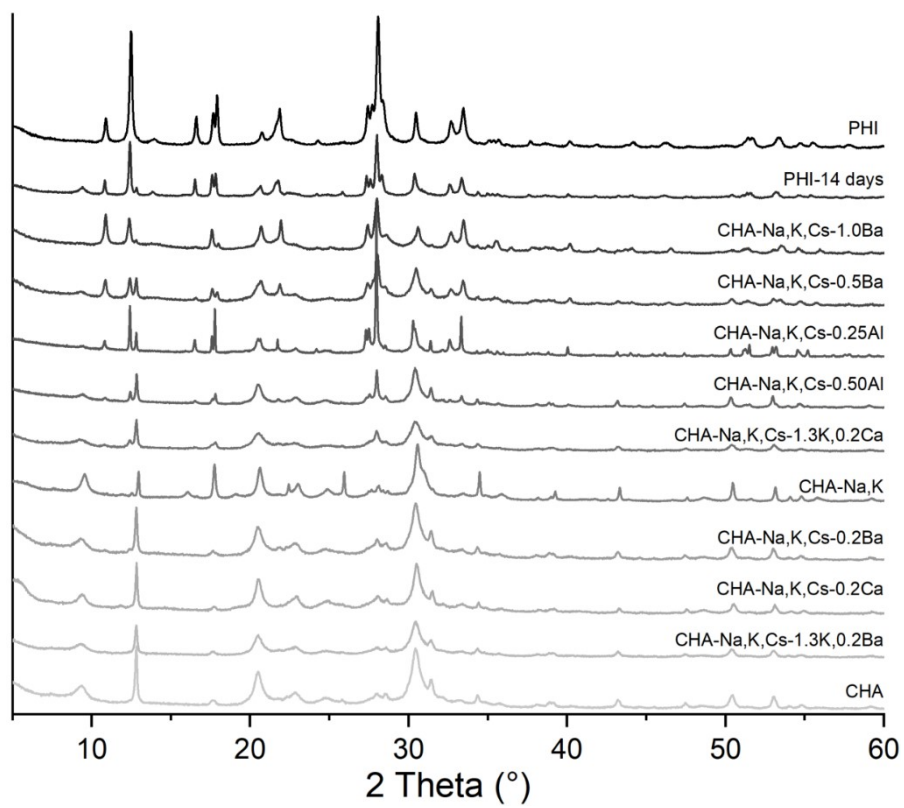
h,k,l	B obs. [° 2 $\theta$ ]	B std. [° 2 $\theta$ ]	Peak pos. [° 2 $\theta$ ]	B struct. [° 2 $\theta$ ]	Crystallite size [Å]
100	0.121	0.003	10.806	0.118	676
020	0.148	0.003	12.392	0.145	551
120	0.108	0.003	16.505	0.105	765
22-1	0.144	0.003	21.855	0.141	574
14-2	0.18	0.003	32.566	0.177	468
34-2	0.108	0.003	37.57	0.105	799
					average: 64±13 nm

**Table S2B.** Scherrer crystalline domain size calculation of the PHI sample.

h,k,l	B obs. [° 2 $\theta$ ]	B std. [° 2 $\theta$ ]	Peak pos. [° 2 $\theta$ ]	B struct. [° 2 $\theta$ ]	Crystallite size [Å]
100	0.144	0.003	10.879	0.141	566
020	0.18	0.003	12.476	0.177	452
120	0.144	0.003	16.581	0.141	569
22-1	0.144	0.003	21.855	0.141	574
14-2	0.252	0.003	32.632	0.249	332
34-2	0.18	0.003	37.65	0.177	474
					average: 49±10 nm



**Figure S3.** SEM images of PHI-14 days. Scale bar is 500 nm.



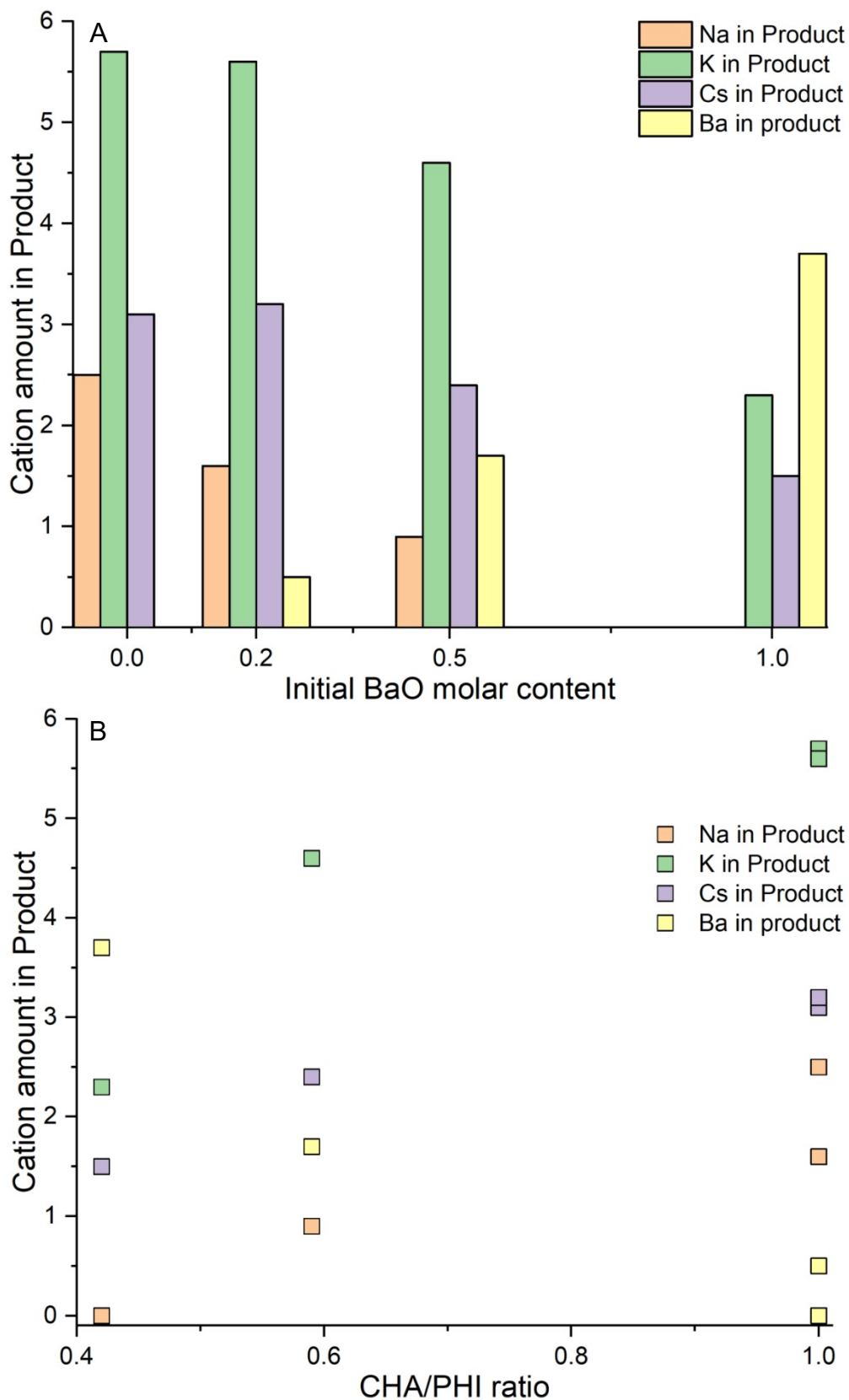
**Figure S4.** Powder XRD patterns of all zeolite samples.

Sample	Phase	GOF	wR (%)	reduced X <sup>2</sup>	a (Å)	b (Å)	c (Å)	beta (°)	Volume (Å <sup>3</sup> )
CHA	CHA	3.42	9.683	11.67	13.835607	-	15.184503	-	2517.258
CHA-Na,Cs-1.3K,0.2Ba	CHA	3.39	9.847	11.46	13.800795	-	15.163165	-	2501.086
CHA-Na,K,Cs-0.2Ca	CHA	3.76	8.906	14.14	13.779192	-	15.086180	-	2480.604
CHA-Na,K,Cs-0.2Ba	CHA	3.92	8.717	15.40	13.803514	-	15.118306	-	2494.670
CHA-Na,K	CHA	4.44	11.506	19.70	13.823164	-	15.082600	-	2495.869
CHA-Na,Cs-1.3K,0.2Ca	CHA	3.05	8.807	9.28	13.777796	-	15.074650	-	2478.206
	PHI				9.943584	14.187387	8.678807	124.8448	1004.827
CHA-Na,K,Cs-0.50Al	CHA	3.14	8.691	9.85	13.799675	-	15.086954	-	2488.112
	PHI				9.957733	14.202910	8.702450	124.8743	1009.740
CHA-Na,K,Cs-0.25Al	CHA	4.51	12.294	20.32	13.830740	-	15.083669	-	2498.783
	PHI				9.985948	14.261571	8.739134	125.2619	1016.232
CHA-Na,K,Cs-0.5Ba	CHA	2.89	6.596	8.38	13.841154	-	15.052657	-	2497.402
	PHI				9.886988	14.214983	8.705141	124.9366	1002.967
CHA-Na,K,Cs-1.0Ba	CHA	2.56	5.887	6.56	13.829992	-	15.052693	-	2493.381
	PHI				9.870288	14.210071	8.727031	124.8026	1005.081
PHI-14 days	CHA	3.29	11.004	10.84	13.822081	-	15.069711	-	2493.345
	PHI				9.927806	14.204983	8.713394	124.9560	1007.115
PHI	PHI	3.08	7.252	9.5	9.916595	14.202499	8.717479	125.0690	1004.884

**Table S3.** Le Bail refined unit cell parameters of the chabazite and phillipsite zeolites.

**Table S4.** Chemical composition of the chabazite and phillipsite nanozeolites determined by ICP-MS. \*Due to the co-existence of  $\text{Ca}(\text{OH})_2$  precipitates the amount of  $\text{Ca}^{2+}$  cations present in the CHA unit cell was estimated (see *Advanced Sustainable Systems* **2024**, 8 (1), 2300326).

ICP-MS Chemical composition									
	Cs	K	Na	Ba	Ca	Si	Al	Si/Al	CHA/PHI
CHA	3.1	5.7	2.5	-	-	24.7	11.3	2.2	1.00
Chabazite <sup>60</sup>	6.6	3.6	2.4	-	-	23.4	12.6	1.9	1.00
CHA-Na,Cs-1.3K,0.2Ba	3.7	5.2	1.5	0.3	-	24.9	11.1	2.2	1.00
CHA-Na,Cs-1.3K,0.2Ca	3.3	5.2	1.7	-	0.4	25.0	11.0	2.3	1.00
CHA-Na,K,Cs-0.2Ca	2.7	6.0	1.8 *	-	0.6 *	24.3	11.7	2.1	1.00
CHA-Na,K,Cs-0.2Ba	3.2	5.6	1.6	0.5	-	24.6	11.4	2.2	1.00
CHA-Na,K	-	7.7	3.5	-	-	24.5	11.2	2.2	1.00
CHA-Na,K,Cs-0.50Al	3.1	5.6	2.4	-	-	23.8	11.1	2.1	0.74
CHA-Na,K,Cs-0.25Al	2.3	5.6	3.1	-	-	24.0	11.0	2.2	0.69
CHA-Na,K,Cs-0.5Ba	2.3	4.4	0.9	1.6	-	23.5	10.8	2.2	0.59
CHA-Na,K,Cs-1.0Ba	1.4	2.1	-	3.4	-	23.2	10.3	2.3	0.42
PHI-14 days	-	6.1	3.2	-	0.7	22.6	10.7	2.1	0.29
PHI	-	5.8	3.3	-	0.7	21.6	10.5	2.1	0.00

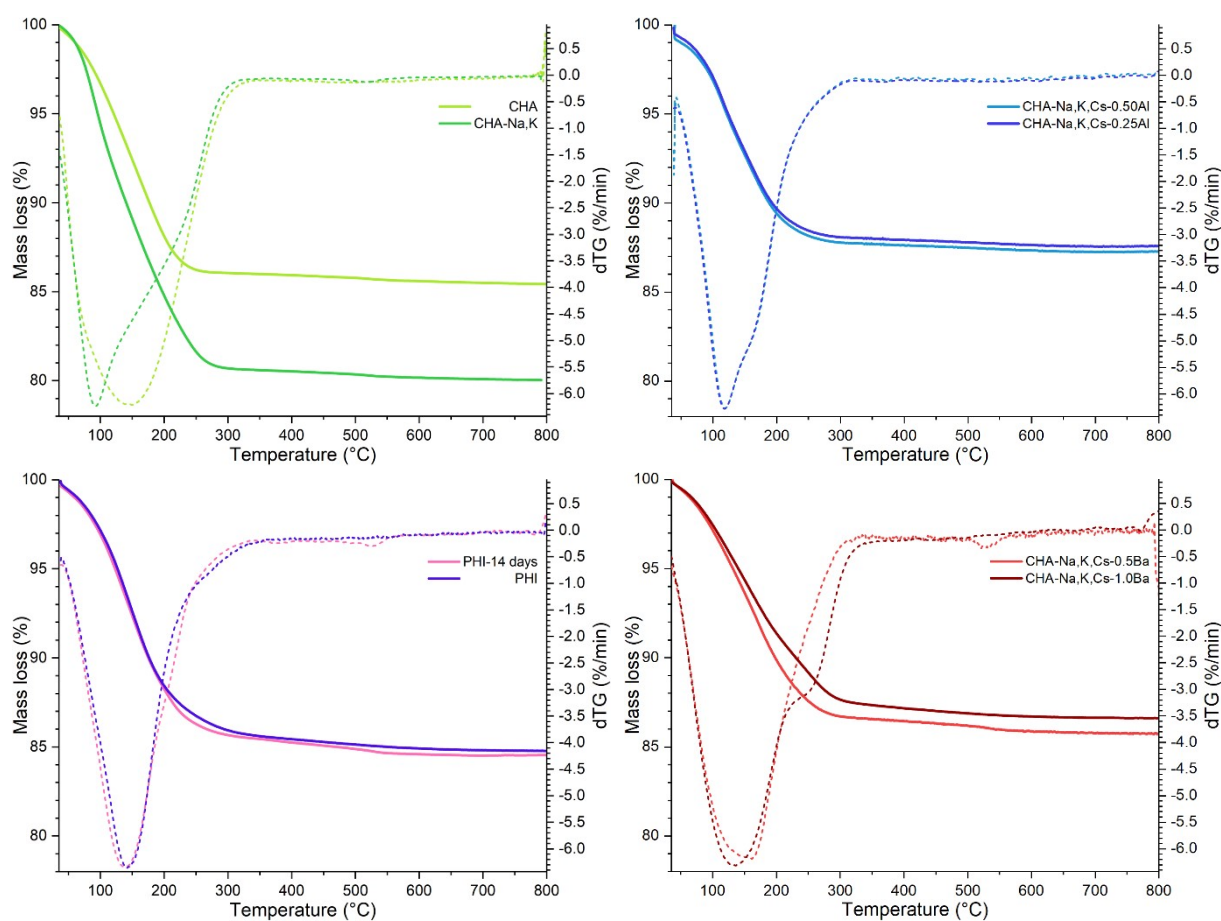


**Figure S5.** Plot of Cs, K, Na, Ba amount over the initial BaO molar content (A) or the CHA/PHI ratio (B), for samples CHA, CHA-Na,K,Cs-0.2Ba, CHA-Na,K,Cs-0.5Ba and CHA-Na,K,Cs-1.0Ba.

**Table S5.** Detailed molar ratio and coordinates of samples presented in the ternary phase diagram.

	CHA:PHI ratio	molar ratio			%		
		Al + K	Ca or Ba	Cs	Al + K	Ca or Ba	Cs
CHA	1.00	2.20	0.00	0.20	0.92	0.00	0.08
Chabazite <sup>60</sup>	1.00	1.95	0.00	0.15	0.93	0.00	0.07
CHA-Na,Cs-1.3K,0.2Ba	1.00	2.00	0.20	0.20	0.83	0.08	0.08
CHA-Na,Cs-1.3K,0.2Ca	0.81	2.00	0.20	0.20	0.83	0.08	0.09
CHA-Na,K,Cs-0.2Ca	1.00	2.20	0.20	0.20	0.85	0.08	0.08
CHA-Na,K,Cs-0.2Ba	1.00	2.20	0.20	0.20	0.85	0.08	0.08
CHA-Na,K	1.00	2.20	0.00	0.00	1.00	0.00	0.00
CHA-Na,K,Cs-0.50Al	0.74	2.00	0.00	0.20	0.91	0.00	0.09
CHA-Na,K,Cs-0.25Al	0.69	1.76	0.00	0.20	0.90	0.00	0.10
CHA-Na,K,Cs-0.5Ba	0.59	2.20	0.50	0.20	0.76	0.17	0.07
CHA-Na,K,Cs-1.0Ba	0.42	2.20	1.00	0.20	0.65	0.29	0.06
PHI*	0.00	2.20	0.20	0.00	0.92	0.08	0.00

\*PHI-14 days is not included as its precursor chemical composition is the same as PHI



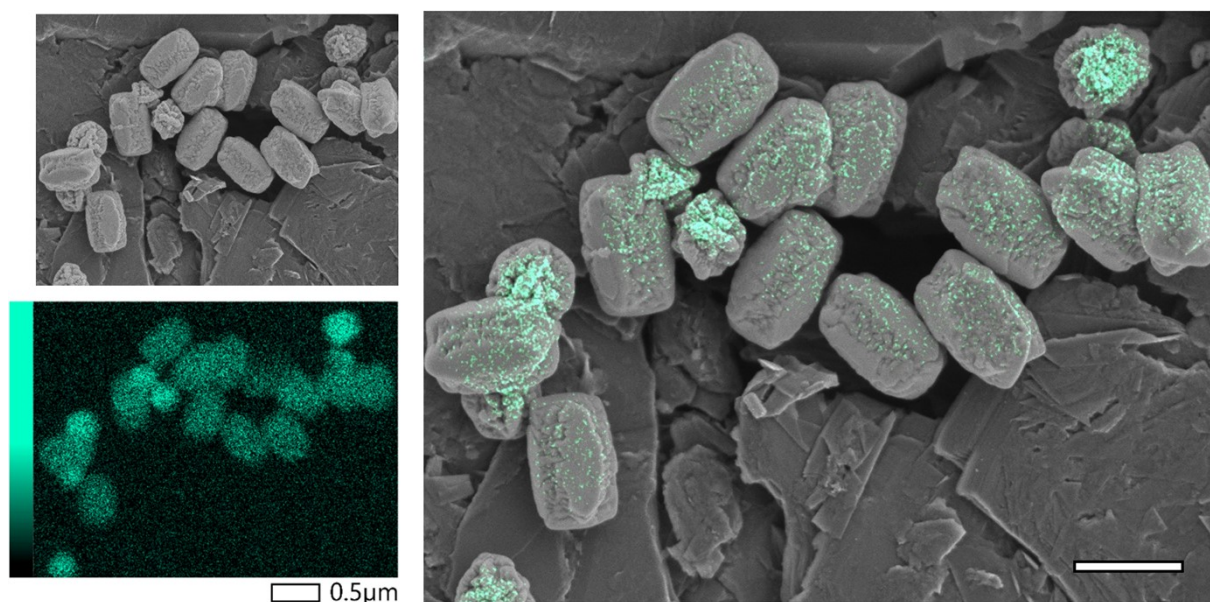
**Figure S6.** Thermogravimetric (solid) and differential thermogravimetric (dotted) curves of all zeolite samples.



**Table S6.** Mass loss at 800 °C of all nanosized zeolites calculated from thermogravimetric analysis.

	Mass loss at 800 °C
CHA*	14.5 %
CHA-Na,Cs-1.3K,0.2Ba*	14.3 %
CHA-Na,Cs-1.3K,0.2Ca*	14.7 %
CHA-Na,K,Cs-0.2Ca*	14.3 %
CHA-Na,K,Cs-0.2Ba*	14.7 %
CHA-Na,K	19.9 %
CHA-Na,K,Cs-0.50Al	12.8 %
CHA-Na,K,Cs-0.25Al	12.4 %
CHA-Na,K,Cs-0.5Ba	14.3 %
CHA-Na,K,Cs-1.0Ba	13.4 %
PHI-14 days	15.4 %
PHI	15.2 %

\* Samples described in *Advanced Sustainable Systems* **2024**, 8 (1), 2300326.

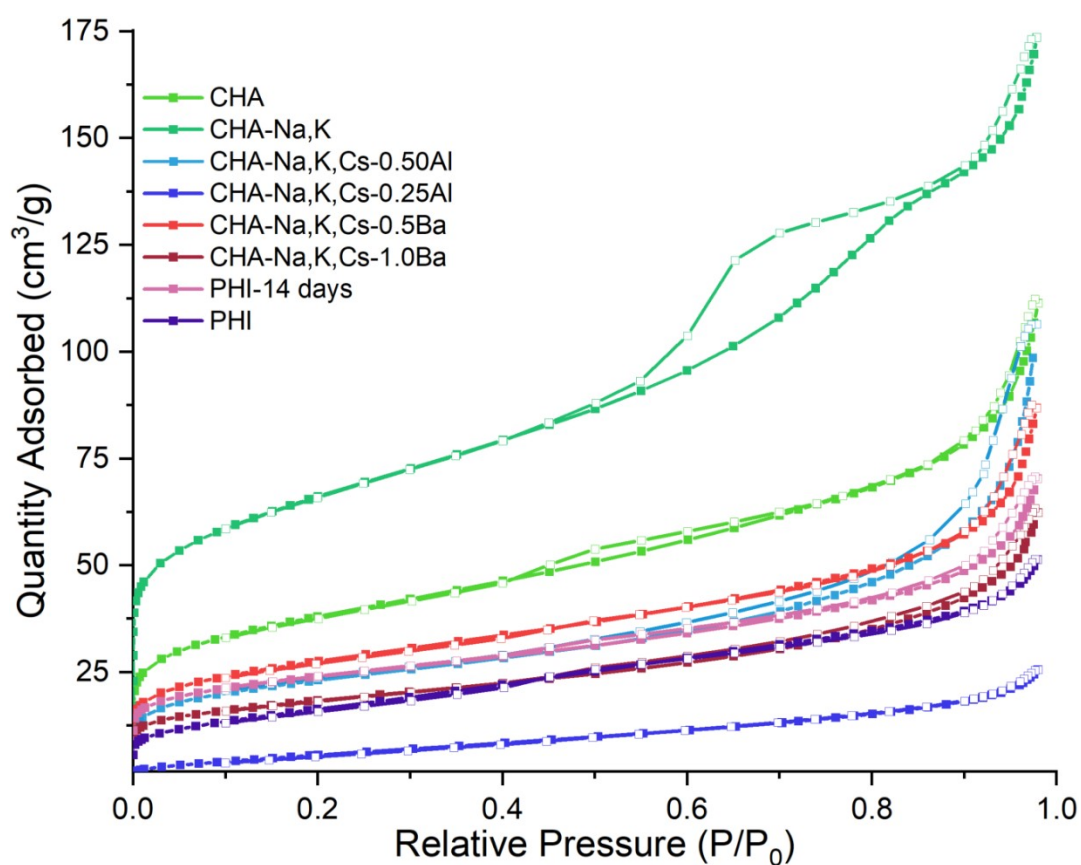


**Figure S7.** SEM picture and EDS mapping of caesium in the sample CHA-Na,K,Cs-0.25Al (left), overlay of the SEM and EDS pictures highlighting the greater amount of caesium in chabazite crystals (right). Scale bar is 500 nm.

**Table S7.** BET specific surface areas and microporous volumes of the synthesised nanozeolites from N<sub>2</sub> adsorption isotherms recorded at -196°C.

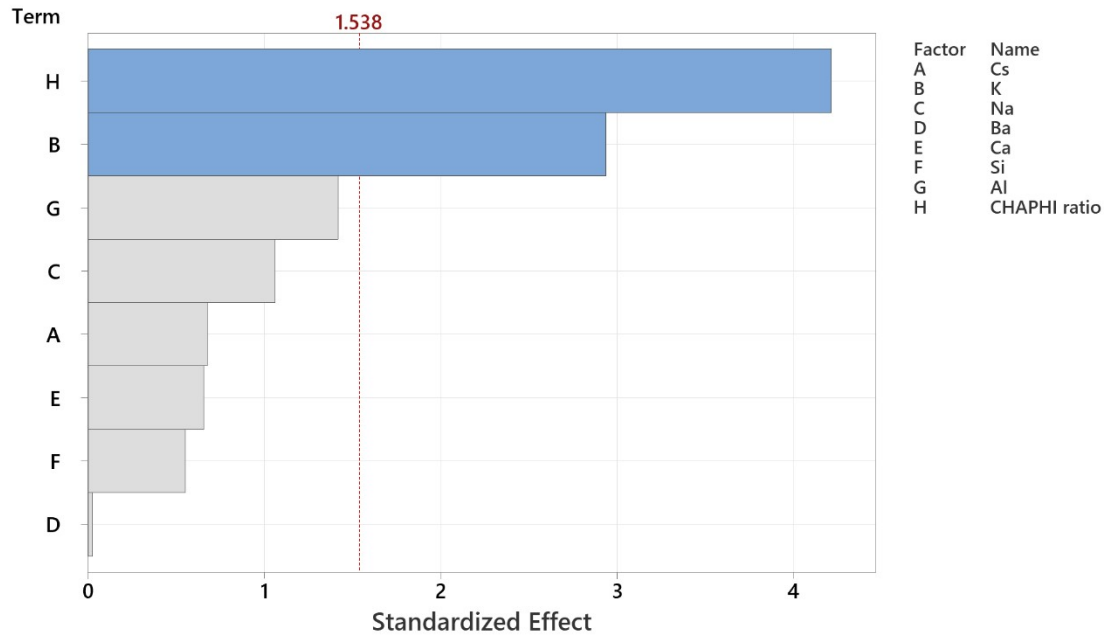
	BET specific surface areas (m <sup>2</sup> /g)	Microporous volumes (cm <sup>3</sup> /g)
CHA*	136	0.04
CHA-Na,Cs-1.3K,0.2Ba*	155	0.04
CHA-Na,Cs-1.3K,0.2Ca*	172	0.04
CHA-Na,K,Cs-0.2Ca*	290	0.07
CHA-Na,K,Cs-0.2Ba*	169	0.05
CHA-Na,K	233	0.06
CHA-Na,K,Cs-0.50Al	83	0.02
CHA-Na,K,Cs-0.25Al	24	0.00
CHA-Na,K,Cs-0.5Ba	98	0.03
CHA-Na,K,Cs-1.0Ba	66	0.02
PHI-14 days	86	0.03
PHI	60	0.00

\* Samples described in *Advanced Sustainable Systems* **2024**, 8 (1), 2300326.



**Figure S8.** N<sub>2</sub> adsorption (full symbols) and desorption (empty symbols) isotherms of all zeolite samples measured at -196 °C.

**Pareto Chart of the Standardized Effects**  
(response is CO<sub>2</sub>;  $\alpha = 0.15$ )



*A gray bar represents a term not in the model.*

**Coded Coefficients**

Term	Coef	SE Coef	T-Value	P-Value	VIF
Constant	2.267	0.252	8.99	0.000	
K	1.275	0.434	2.93	0.013	1.13
CHAPHI ratio	1.201	0.285	4.21	0.001	1.13

**Figure S9.** Pareto chart of the standardised effects of elements from ICP-MS and CHA/PHI ratio from RiR method on the CO<sub>2</sub> adsorption capacity at 90 kPa. Based on a screening design model, parameters H and B are statistically significant at the 15% threshold (CHA/PHI ratio uncertainty) using the terms of the current model, computed by Minitab.