

## Electronic Supplementary Information (ESI) for:

### Deep-ultraviolet sulfamate halides with halogen-centered secondary building units for enhanced optical anisotropy

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**Table S1.** Crystallographic data of MX(NH<sub>3</sub>SO<sub>3</sub>) (M = Rb, Cs; X = Cl, Br)

	<b>RbCl(NH<sub>3</sub>SO<sub>3</sub>)</b>	<b>CsCl(NH<sub>3</sub>SO<sub>3</sub>)</b>	<b>RbBr(NH<sub>3</sub>SO<sub>3</sub>)</b>	<b>CsBr(NH<sub>3</sub>SO<sub>3</sub>)</b>
Formula weight	218.01	265.45	262.47	309.90
Temp/K	300.0	299.0	299.0	300.0
Crystal system	monoclinic	monoclinic	orthorhombic	orthorhombic
Space group	<i>P2<sub>1</sub>/m</i>	<i>P2<sub>1</sub>/m</i>	<i>Pnma</i>	<i>Pnma</i>
<i>a</i> /Å	4.7178(5)	7.100(2)	13.292(3)	13.9270(9)
<i>b</i> /Å	6.1824(6)	5.5448(17)	5.7921(9)	5.7869(4)
<i>c</i> /Å	9.8402(10)	7.915(3)	7.6673(18)	7.9082(5)
<i>α</i> /°	90	90	90	90
<i>β</i> /°	91.718(3)	102.466(10)	90	90
<i>γ</i> /°	90	90	90	90
<i>V</i> /Å <sup>3</sup>	286.88(5)	304.25(17)	590.3(2)	637.35(7)
<i>Z</i>	2	2	4	4
$\rho_{\text{calc}}$ g/cm <sup>3</sup>	2.524	2.898	2.953	3.230
$\mu$ /mm <sup>-1</sup>	9.362	6.777	15.417	12.320
<i>F</i> (000)	208.0	244.0	488.0	559.1
Crystal size/mm <sup>3</sup>	0.259 × 0.143 × 0.108	0.252 × 0.111 × 0.093	0.140 × 0.115 × 0.096	0.189 × 0.124 × 0.102
2 theta range/°	7.786 to 60.93	5.27 to 60.946	6.13 to 60.964	5.86 to 60.98
Index ranges	-6 ≤ <i>h</i> ≤ 6 -8 ≤ <i>k</i> ≤ 8 -14 ≤ <i>l</i> ≤ 14	-10 ≤ <i>h</i> ≤ 10 -7 ≤ <i>k</i> ≤ 7 -11 ≤ <i>l</i> ≤ 11	-18 ≤ <i>h</i> ≤ 18 -7 ≤ <i>k</i> ≤ 4 -10 ≤ <i>l</i> ≤ 10	-19 ≤ <i>h</i> ≤ 16 -8 ≤ <i>k</i> ≤ 8 -11 ≤ <i>l</i> ≤ 11
Reflns collected	5991	7769	5974	10301
Independent reflns	936 ( <i>R</i> <sub>int</sub> = 0.0382)	1010 ( <i>R</i> <sub>int</sub> = 0.0374)	934 ( <i>R</i> <sub>int</sub> = 0.0538)	1049 ( <i>R</i> <sub>int</sub> = 0.0781)
Data/restraints/param	936/2/47	1010/2/45	934/2/48	1049/2/47
Goof on <i>F</i> <sup>2</sup>	1.106	1.253	1.053	1.228
<i>R</i> <sub>1</sub> <sup>a</sup> / <i>wR</i> <sub>2</sub> <sup>b</sup> [I ≥ 2σ ( <i>I</i> )]	0.0201/0.0401	0.0198/0.0384	0.0323/0.0768	0.0491/0.0847
<i>R</i> <sub>1</sub> <sup>a</sup> / <i>wR</i> <sub>2</sub> <sup>b</sup> [all data]	0.0250/0.0410	0.0234/0.0520	0.0543/0.0823	0.0589/0.0875
Largest diff peak/hole / e Å <sup>-3</sup>	0.44/-0.50	0.63/-0.73	0.77/-0.78	1.33/-1.90

$$^a R_1 = \sum ||F_o| - |F_c|| / \sum |F_o|. \quad ^b wR_2 = [\sum w(F_o^2 - F_c^2)^2 / \sum wF_o^4]^{1/2}$$

**Table S2.** Fractional atomic coordinates ( $\times 10^4$ ), equivalent isotropic displacement parameters ( $\text{\AA}^2 \times 10^3$ ) and bond valence sum (BVS) for the non-H atoms in  $\text{MX}(\text{NH}_3\text{SO}_3)$  (M = Rb, Cs; X = Cl, Br).  $U_{\text{eq}}$  is defined as 1/3 of the trace of the orthogonalized  $U_{ij}$  tensor.

Atom	x	y	z	$U(\text{eq})$	BVS
<b>RbCl(NH<sub>3</sub>SO<sub>3</sub>)</b>					
Rb1	3030.4(5)	2500	8283.0(3)	26.2(1)	0.99
Cl1	7879.7(14)	2500	5742.7(6)	26.50(15)	0.24
S1	8797.5(12)	7500	8155.7(6)	18.04(13)	5.89
O1	7792(3)	5541(2)	8756.6(13)	27.6(3)	1.96
O2	11760(4)	7500	7850.5(19)	30.1(4)	1.81
N1	7173(5)	7500	6523(2)	21.6(4)	0.90
<b>CsCl(NH<sub>3</sub>SO<sub>3</sub>)</b>					
Cs1	6170.5(4)	12500	8266.3(4)	24.94(9)	1.09
Cl1	1333.4(17)	2500	9124.5(16)	28.0(2)	0.40
S1	2645.6(16)	7500	6173.5(14)	20.1(2)	5.88
O1	4134(5)	7500	7719(4)	26.6(7)	2.09
O2	2478(4)	5311(5)	5184(3)	31.7(5)	1.80
N1	547(6)	7500	7037(5)	24.3(8)	0.88
<b>RbBr(NH<sub>3</sub>SO<sub>3</sub>)</b>					
Rb1	4236.7(5)	2500	1122.2(8)	30.5(2)	0.79
Br1	1978.9(5)	2500	3992.3(8)	26.1(2)	0.43
S1	3617.0(11)	7500	4357.0(18)	19.8(3)	5.79
O1	2867(3)	7500	3009(6)	27.5(10)	1.66
O2	3692(2)	5409(5)	5369(4)	28.5(7)	1.64
N1	4750(4)	7500	3125(7)	22.2(10)	1.22
<b>CsBr(NH<sub>3</sub>SO<sub>3</sub>)</b>					
Cs1	1965.5(4)	2500	4013.3(6)	22.59(18)	1.08
Br1	4241.8(7)	2500	1121.9(10)	26.6(2)	0.37
S1	3696.4(14)	7500	4377(2)	17.7(4)	5.91
O1	3790(3)	5407(7)	5344(5)	27.0(9)	1.85
O2	2936(4)	7500	3139(7)	27.9(13)	1.94
N1	4729(5)	7500	3090(8)	19.8(13)	0.97

**Table S3.** Selected bond lengths (Å) and angles (°) for RbCl(NH<sub>3</sub>SO<sub>3</sub>).

Rb1-Cl1	3.4388(7)	O1 <sup>6</sup> -Rb1-O1 <sup>3</sup>	162.18(5)
Rb1-Cl1 <sup>3</sup>	3.4336(7)	O1 <sup>6</sup> -Rb1-O1 <sup>4</sup>	74.16(4)
Rb1-O1 <sup>4</sup>	3.1891(14)	O1 <sup>5</sup> -Rb1-O1 <sup>4</sup>	61.38(4)
Rb1-O1 <sup>3</sup>	3.1513(14)	O1 <sup>6</sup> -Rb1-O1 <sup>5</sup>	101.12(4)
Rb1-O1 <sup>5</sup>	3.1513(14)	O1-Rb1-O1 <sup>2</sup>	74.16(4)
Rb1-O1 <sup>6</sup>	2.9556(13)	O1 <sup>2</sup> -Rb1-O1 <sup>4</sup>	44.63(5)
Rb1-O1 <sup>2</sup>	3.1891(14)	O1 <sup>3</sup> -Rb1-O1 <sup>4</sup>	88.52(2)
Rb1-O1	2.9556(13)	O1 <sup>5</sup> -Rb1-O1 <sup>3</sup>	73.27(5)
Rb1-O2 <sup>3</sup>	3.1750(5)	O1 <sup>6</sup> -Rb1-O1	79.02(5)
Rb1-O2 <sup>1</sup>	3.1750(5)	O1 <sup>5</sup> -Rb1-O2 <sup>1</sup>	45.02(4)
S1-O1	1.4345(13)	O1 <sup>3</sup> -Rb1-O2 <sup>3</sup>	45.02(4)
S1-O1 <sup>7</sup>	1.4345(13)	O1-Rb1-O2 <sup>1</sup>	141.27(4)
S1-O2	1.4383(19)	O1 <sup>6</sup> -Rb1-O2 <sup>3</sup>	141.27(4)
S1-N1	1.759(2)	O1 <sup>3</sup> -Rb1-O2 <sup>1</sup>	117.07(4)
Cl1 <sup>3</sup> -Rb1-Cl1	86.705(17)	O1-Rb1-O2 <sup>3</sup>	62.69(4)
O1 <sup>5</sup> -Rb1-Cl1	130.49(2)	O1 <sup>5</sup> -Rb1-O2 <sup>3</sup>	117.07(4)
O1-Rb1-Cl1	66.15(3)	O1 <sup>6</sup> -Rb1-O2 <sup>1</sup>	62.69(4)
O1 <sup>4</sup> -Rb1-Cl1 <sup>3</sup>	124.21(3)	O2 <sup>1</sup> -Rb1-Cl1 <sup>3</sup>	77.04(3)
O1 <sup>6</sup> -Rb1-Cl1 <sup>3</sup>	129.25(3)	O2 <sup>3</sup> -Rb1-Cl1 <sup>3</sup>	77.04(3)
O1 <sup>4</sup> -Rb1-Cl1	139.98(3)	O2 <sup>3</sup> -Rb1-Cl1	91.70(4)
O1 <sup>3</sup> -Rb1-Cl1 <sup>3</sup>	64.31(2)	O2 <sup>1</sup> -Rb1-Cl1	91.70(4)
O1-Rb1-Cl1 <sup>3</sup>	129.25(3)	O2 <sup>1</sup> -Rb1-O1 <sup>4</sup>	74.05(4)
O1 <sup>2</sup> -Rb1-Cl1 <sup>3</sup>	124.21(3)	O2 <sup>3</sup> -Rb1-O1 <sup>2</sup>	74.05(4)
O1 <sup>2</sup> -Rb1-Cl1	139.98(3)	O2 <sup>3</sup> -Rb1-O1 <sup>4</sup>	117.68(4)
O1 <sup>6</sup> -Rb1-Cl1	66.15(3)	O2 <sup>1</sup> -Rb1-O1 <sup>2</sup>	117.68(4)
O1 <sup>5</sup> -Rb1-Cl1 <sup>3</sup>	64.31(2)	O2 <sup>3</sup> -Rb1-O2 <sup>1</sup>	153.61(6)
O1 <sup>3</sup> -Rb1-Cl1	130.49(2)	O1 <sup>8</sup> -S1-O1	115.15(12)
O1-Rb1-O1 <sup>5</sup>	162.18(5)	O1 <sup>8</sup> -S1-O2	114.94(7)
O1 <sup>6</sup> -Rb1-O1 <sup>2</sup>	102.13(3)	O1-S1-O2	114.94(7)
O1 <sup>3</sup> -Rb1-O1 <sup>2</sup>	61.38(4)	O1-S1-N1	103.65(7)
O1-Rb1-O1 <sup>4</sup>	102.13(3)	O1 <sup>8</sup> -S1-N1	103.65(7)
O1-Rb1-O1 <sup>3</sup>	101.12(4)	O2-S1-N1	102.05(11)
O1 <sup>5</sup> -Rb1-O1 <sup>2</sup>	88.52(2)		

<sup>1</sup>-1+x, -1+y, +z; <sup>2</sup>1-x, 1-y, 2-z; <sup>3</sup>-1+x, +y, +z; <sup>4</sup>1-x, -1/2+y, 2-z; <sup>5</sup>-1+x, 1/2-y, +z; <sup>6</sup>+x, 1/2-y, +z; <sup>7</sup>+x, 3/2-y, +z; <sup>8</sup>+x, 3/2-y, +z

**Table S4.** Selected bond lengths (Å) and angles (°) for CsCl(NH<sub>3</sub>SO<sub>3</sub>).

Cs1-O2 <sup>1</sup>	3.314(3)	O2 <sup>6</sup> -Cs1-O2 <sup>5</sup>	84.04(5)
Cs1-Cl1 <sup>2</sup>	3.6759(11)	O2 <sup>1</sup> -Cs1-O2 <sup>7</sup>	84.04(5)
Cs1-Cl1 <sup>3</sup>	3.5808(17)	O2 <sup>1</sup> -Cs1-O2 <sup>5</sup>	64.73(7)
Cs1-Cl1 <sup>4</sup>	3.6759(11)	O2 <sup>7</sup> -Cs1-O2 <sup>7</sup>	64.73(7)
Cs1-Cl1 <sup>5</sup>	3.6418(16)	O2 <sup>6</sup> -Cs1-O2 <sup>1</sup>	42.96(9)
Cs1-O2 <sup>5</sup>	3.534(3)	O1 <sup>4</sup> -Cs1-Cl1 <sup>4</sup>	64.78(4)
Cs1-O2 <sup>6</sup>	3.314(3)	O1 <sup>4</sup> -Cs1-Cl1 <sup>2</sup>	64.78(4)
Cs1-O2 <sup>7</sup>	3.534(3)	O1 <sup>5</sup> -Cs1-Cl1 <sup>2</sup>	154.54(7)
Cs1-O1	3.1135(17)	O1 <sup>4</sup> -Cs1-Cl1 <sup>3</sup>	95.54(6)
Cs1-O1 <sup>5</sup>	3.1135(17)	O1 <sup>4</sup> -Cs1-Cl1 <sup>5</sup>	63.29(6)
Cs1-O1 <sup>4</sup>	3.232(4)	O1 <sup>5</sup> -Cs1-Cl1 <sup>5</sup>	66.28(6)
S1-O2 <sup>7</sup>	1.435(2)	O1-Cs1-Cl1 <sup>3</sup>	116.87(6)
S1-O2	1.435(2)	O1 <sup>5</sup> -Cs1-Cl1 <sup>3</sup>	116.87(6)
S1-O1	1.433(3)	O1 <sup>5</sup> -Cs1-Cl1 <sup>4</sup>	63.88(6)
S1-N1	1.768(4)	O1-Cs1-Cl1 <sup>4</sup>	154.54(7)
Cl1 <sup>5</sup> -Cs1-Cl1 <sup>4</sup>	104.77(2)	O1-Cs1-Cl1 <sup>2</sup>	63.88(6)
Cl1 <sup>4</sup> -Cs1-Cl1 <sup>2</sup>	104.77(2)	O1-Cs1-Cl1 <sup>5</sup>	66.28(6)
Cl1 <sup>3</sup> -Cs1-Cl1 <sup>2</sup>	62.86(2)	O1 <sup>5</sup> -Cs1-O2 <sup>6</sup>	114.81(8)
Cl1 <sup>4</sup> -Cs1-Cl1 <sup>2</sup>	97.91(4)	O1 <sup>5</sup> -Cs1-O2 <sup>7</sup>	92.50(7)
Cl1 <sup>3</sup> -Cs1-Cl1 <sup>5</sup>	158.83(4)	O1-Cs1-O2 <sup>7</sup>	42.12(7)
Cl1 <sup>3</sup> -Cs1-Cl1 <sup>4</sup>	62.86(2)	O1 <sup>5</sup> -Cs1-O2 <sup>5</sup>	42.12(7)
O2 <sup>5</sup> -Cs1-Cl1 <sup>4</sup>	104.54(5)	O1-Cs1-O2 <sup>1</sup>	114.81(8)
O2 <sup>7</sup> -Cs1-Cl1 <sup>2</sup>	104.54(5)	O1 <sup>5</sup> -Cs1-O2 <sup>1</sup>	76.55(8)
O2 <sup>6</sup> -Cs1-Cl1 <sup>5</sup>	127.31(5)	O1-Cs1-O2 <sup>6</sup>	76.55(8)
O2 <sup>6</sup> -Cs1-Cl1 <sup>3</sup>	71.93(5)	O1 <sup>4</sup> -Cs1-O2 <sup>7</sup>	118.80(7)
O2 <sup>5</sup> -Cs1-Cl1 <sup>2</sup>	156.38(4)	O1 <sup>4</sup> -Cs1-O2 <sup>5</sup>	118.80(7)
O2 <sup>7</sup> -Cs1-Cl1 <sup>5</sup>	62.61(5)	O1 <sup>4</sup> -Cs1-O2 <sup>1</sup>	154.57(5)
O2 <sup>7</sup> -Cs1-Cl1 <sup>3</sup>	135.01(5)	O1-Cs1-O2 <sup>5</sup>	92.50(7)
O2 <sup>1</sup> -Cs1-Cl1 <sup>5</sup>	127.31(5)	O1 <sup>4</sup> -Cs1-O2 <sup>6</sup>	154.57(5)
O2 <sup>5</sup> -Cs1-Cl1 <sup>5</sup>	62.61(5)	O1-Cs1-O1 <sup>5</sup>	125.86(11)
O2 <sup>7</sup> -Cs1-Cl1 <sup>4</sup>	156.38(4)	O1 <sup>5</sup> -Cs1-O1 <sup>4</sup>	90.53(7)
O2 <sup>6</sup> -Cs1-Cl1 <sup>4</sup>	123.30(5)	O1-Cs1-O1 <sup>4</sup>	90.53(7)
O2 <sup>1</sup> -Cs1-Cl1 <sup>4</sup>	89.81(5)	O2-S1-O2 <sup>7</sup>	115.5(2)
O2 <sup>6</sup> -Cs1-Cl1 <sup>2</sup>	89.81(5)	O2 <sup>7</sup> -S1-N1	103.50(13)
O2 <sup>1</sup> -Cs1-Cl1 <sup>3</sup>	71.93(5)	O2-S1-N1	103.50(13)
O2 <sup>5</sup> -Cs1-Cl1 <sup>3</sup>	135.01(5)	O1-S1-O2	115.12(12)
O2 <sup>1</sup> -Cs1-Cl1 <sup>2</sup>	123.30(5)	O1-S1-O2 <sup>7</sup>	115.12(13)
O2 <sup>7</sup> -Cs1-O2 <sup>5</sup>	52.34(8)	O1-S1-N1	101.4(2)

<sup>1</sup>1-x, 2-y, 1-z; <sup>2</sup>1-x, 1-y, 2-z; <sup>3</sup>1+x, 1+y, +z; <sup>4</sup>1-x, 2-y, 2-z; <sup>5</sup>+x, 1+y, +z; <sup>6</sup>1-x, 1/2+y, 1-z; <sup>7</sup>+x, 3/2-y, +z; <sup>8</sup>+x, -1+y, +z; <sup>9</sup>-1+x, -1+y, +z

**Table S5.** Selected bond lengths (Å) and angles (°) for RbBr(NH<sub>3</sub>SO<sub>3</sub>).

Br1-Rb1 <sup>1</sup>	3.6460(13)	N1-Rb1-Br1 <sup>7</sup>	78.90(10)
Br1-Rb1 <sup>2</sup>	3.6965(7)	N1-Rb1-Br1 <sup>8</sup>	67.29(8)
Br1-Rb1 <sup>3</sup>	3.6965(7)	N1 <sup>11</sup> -Rb1-Br1 <sup>7</sup>	78.89(10)
Br1-Rb1	3.7213(11)	N1 <sup>10</sup> -Rb1-Br1 <sup>7</sup>	66.15(9)
Rb1-N1 <sup>5</sup>	3.348(3)	N1 <sup>11</sup> -Rb1-Br1	83.86(9)
Rb1-N1	3.348(3)	N1-Rb1-Br1	83.86(9)
Rb1-N1 <sup>4</sup>	3.524(5)	N1 <sup>11</sup> -Rb1-Br1 <sup>9</sup>	67.29(8)
S1-O1	1.436(4)	N1 <sup>11</sup> -Rb1-Br1 <sup>8</sup>	165.77(10)
S1-O2	1.442(3)	N1 <sup>10</sup> -Rb1-Br1 <sup>9</sup>	76.05(6)
S1-O2 <sup>6</sup>	1.442(3)	N1 <sup>10</sup> -Rb1-Br1	148.73(9)
S1-N1	1.778(6)	N1-Rb1-N1 <sup>11</sup>	119.76(16)
Br1 <sup>7</sup> -Rb1-Br1 <sup>8</sup>	115.235(16)	N1 <sup>11</sup> -Rb1-N1 <sup>10</sup>	110.24(9)
Br1 <sup>7</sup> -Rb1-Br1 <sup>9</sup>	115.235(16)	N1-Rb1-N1 <sup>10</sup>	110.24(9)
Br1 <sup>8</sup> -Rb1-Br1	84.765(17)	O1-S1-O2 <sup>12</sup>	115.82(15)
Br1 <sup>7</sup> -Rb1-Br1	145.13(2)	O1-S1-O2	115.82(15)
Br1 <sup>8</sup> -Rb1-Br1 <sup>9</sup>	103.15(3)	O1-S1-N1	101.9(3)
Br1 <sup>9</sup> -Rb1-Br1	84.765(17)	O2-S1-O2 <sup>12</sup>	114.3(3)
N1-Rb1-Br1 <sup>9</sup>	165.77(10)	O2-S1-N1	103.13(16)
N1 <sup>10</sup> -Rb1-Br1 <sup>8</sup>	76.05(6)	O2 <sup>12</sup> -S1-N1	103.13(16)

<sup>1</sup>-1/2+x, +y, 1/2-z; <sup>2</sup>1/2-x, 1-y, 1/2+z; <sup>3</sup>1/2-x, 1-y, 1/2+z; <sup>4</sup>1-x, 1-y, -z; <sup>5</sup>+x, -1+y, +z; <sup>6</sup>+x, 3/2-y, +z; <sup>7</sup>1/2+x, +y, 1/2-z; <sup>8</sup>1/2-x, 1-y, -1/2+z; <sup>9</sup>1/2-x, -y, -1/2+z; <sup>10</sup>1-x, 1-y, -z; <sup>11</sup>+x, -1+y, +z; <sup>12</sup>+x, 3/2-y, +z

**Table S6.** Selected bond lengths (Å) and angles (°) for CsBr(NH<sub>3</sub>SO<sub>3</sub>).

Cs1-Br1 <sup>1</sup>	3.7390(7)	O1-Cs1-O1 <sup>13</sup>	136.59(11)
Cs1-Br1 <sup>2</sup>	3.7390(7)	O1-Cs1-O1 <sup>11</sup>	110.20(8)
Cs1-S1 <sup>3</sup>	3.7809(18)	O1 <sup>12</sup> -Cs1-O2	101.83(12)
Cs1-S1 <sup>4</sup>	3.7769(13)	O1 <sup>12</sup> -Cs1-O2 <sup>10</sup>	44.03(12)
Cs1-S1	3.7770(13)	O1-Cs1-O2	44.03(12)
Cs1-O1 <sup>5</sup>	3.316(4)	O1 <sup>12</sup> -Cs1-O2 <sup>8</sup>	68.95(11)
Cs1-O1	3.224(4)	O1-Cs1-O2 <sup>10</sup>	101.83(12)
Cs1-O1 <sup>6</sup>	3.224(4)	O1-Cs1-O2 <sup>8</sup>	68.95(11)
Cs1-O1 <sup>3</sup>	3.316(4)	O2-Cs1-Br1 <sup>8</sup>	66.11(9)
Cs1-O2 <sup>2</sup>	3.266(6)	O2 <sup>8</sup> -Cs1-Br1 <sup>8</sup>	64.74(5)
Cs1-O2	3.268(3)	O2 <sup>10</sup> -Cs1-Br1 <sup>9</sup>	66.11(9)
Cs1-O2 <sup>4</sup>	3.268(3)	O2-Cs1-Br1 <sup>9</sup>	164.93(10)
S1-O1	1.439(4)	O2 <sup>8</sup> -Cs1-Br1 <sup>9</sup>	64.74(5)
S1-O1 <sup>7</sup>	1.439(4)	O2-Cs1-O1 <sup>11</sup>	67.84(13)
S1-O2	1.442(6)	O2 <sup>10</sup> -Cs1-O1 <sup>13</sup>	67.84(13)
S1-N1	1.761(7)	O2 <sup>10</sup> -Cs1-O1 <sup>11</sup>	105.65(12)
Br1 <sup>8</sup> -Cs1-Br1 <sup>9</sup>	101.40(2)	O2 <sup>8</sup> -Cs1-O1 <sup>11</sup>	152.58(10)
O1 <sup>12</sup> -Cs1-Br1 <sup>9</sup>	78.76(8)	O2-Cs1-O1 <sup>13</sup>	105.65(12)
O1-Cs1-Br1 <sup>8</sup>	78.76(8)	O2 <sup>8</sup> -Cs1-O1 <sup>13</sup>	152.58(10)
O1 <sup>11</sup> -Cs1-Br1 <sup>8</sup>	87.99(7)	O2 <sup>10</sup> -Cs1-O2	124.63(18)
O1 <sup>13</sup> -Cs1-Br1 <sup>8</sup>	122.02(8)	O2 <sup>8</sup> -Cs1-O2 <sup>11</sup>	101.18(10)
O1 <sup>12</sup> -Cs1-Br1 <sup>8</sup>	127.78(7)	O2 <sup>8</sup> -Cs1-O2	101.18(10)
O1 <sup>13</sup> -Cs1-Br1 <sup>9</sup>	87.99(7)	O1 <sup>14</sup> -S1-O1	114.7(4)
O1-Cs1-Br1 <sup>9</sup>	127.78(7)	O1-S1-O2	115.3(2)
O1 <sup>11</sup> -Cs1-Br1 <sup>9</sup>	122.02(8)	O1 <sup>14</sup> -S1-O2	115.3(2)
O1 <sup>13</sup> -Cs1-O1 <sup>11</sup>	42.85(14)	O1 <sup>14</sup> -S1-N1	103.5(2)
O1 <sup>12</sup> -Cs1-O1 <sup>13</sup>	110.20(8)	O1-S1-N1	103.5(2)
O1-Cs1-O1 <sup>12</sup>	62.90(14)	O2-S1-N1	102.0(3)
O1 <sup>12</sup> -Cs1-O1 <sup>11</sup>	136.59(11)		

<sup>1</sup>1/2-x, -y, 1/2+z; <sup>2</sup>1/2-x, 1-y, 1/2+z; <sup>3</sup>1/2-x, 1-y, -1/2+z; <sup>4</sup>+x, -1+y, +z; <sup>5</sup>1/2-x, -1/2+y, -1/2+z; <sup>6</sup>+x, 1/2-y, +z;  
<sup>7</sup>+x, 3/2-y, +z; <sup>8</sup>1/2-x, 1-y, 1/2+z; <sup>9</sup>1/2-x, -y, 1/2+z; <sup>10</sup>+x, -1+y, +z; <sup>11</sup>1/2-x, 1-y, -1/2+z; <sup>12</sup>+x, 1/2-y, +z;  
<sup>13</sup>1/2-x, -1/2+y, -1/2+z; <sup>14</sup>+x, 3/2-y, +z

**Table S7.** Hydrogen bonds for RbCl(NH<sub>3</sub>SO<sub>3</sub>).

D-H...A	<i>d</i> <sub>D-H</sub> (Å)	<i>d</i> <sub>H-A</sub> (Å)	<i>d</i> <sub>D-A</sub> (Å)
N1-H1A-Cl1 <sup>1</sup>	0.848(15)	2.420(16)	3.2049(7)
N1-H1B-Cl1 <sup>2</sup>	0.842(18)	2.72(3)	3.214(2)
N1-H1B-O2 <sup>3</sup>	0.842(18)	2.15(2)	2.903(3)

<sup>1</sup>+x, 1+y, +z; <sup>2</sup>1-x, 1-y, 1-z; <sup>3</sup>-1+x, +y, +z

**Table S8.** Hydrogen bonds for CsCl(NH<sub>3</sub>SO<sub>3</sub>).

D-H...A	<i>d</i> <sub>D-H</sub> (Å)	<i>d</i> <sub>H-A</sub> (Å)	<i>d</i> <sub>D-A</sub> (Å)
N1-H1A-Cl1 <sup>1</sup>	0.826(18)	2.42(2)	3.214(2)
N1-H1B-O2 <sup>2</sup>	0.871(19)	2.234(16)	2.916(4)
N1-H1B-O2 <sup>3</sup>	0.871(19)	2.234(16)	2.916(4)

<sup>1</sup>+x, 1+y, +z; <sup>2</sup>-x, 1-y, 1-z; <sup>3</sup>-x, 1/2+y, 1-z

**Table S9.** Hydrogen bonds for RbBr(NH<sub>3</sub>SO<sub>3</sub>).

D-H...A	<i>d</i> <sub>D-H</sub> (Å)	<i>d</i> <sub>H-A</sub> (Å)	<i>d</i> <sub>D-A</sub> (Å)
N1-H1A-O2 <sup>1</sup>	0.84(2)	2.289(18)	2.908(5)
N1-H1B-Br1 <sup>2</sup>	0.870(18)	3.57(4)	3.914(6)

<sup>1</sup>1-x, 1-y, 1-z; <sup>2</sup>1/2-x, 1-y, -1/2+z

**Table S10.** Hydrogen bonds for CsBr(NH<sub>3</sub>SO<sub>3</sub>).

D-H...A	<i>d</i> <sub>D-H</sub> (Å)	<i>d</i> <sub>H-A</sub> (Å)	<i>d</i> <sub>D-A</sub> (Å)
N1-H1A-Br1 <sup>1</sup>	0.853(19)	2.55(3)	3.355(3)
N1-H1B-O1 <sup>2</sup>	0.84(2)	2.299(18)	2.936(7)
N1-H1B-O1 <sup>3</sup>	0.84(2)	2.299(18)	2.936(7)

<sup>1</sup>+x, 1+y, +z; <sup>2</sup>1-x, 1-y, 1-z; <sup>3</sup>1-x, 1/2+y, 1-z



**Table S11.** Weight and atomic ratios for  $\text{MX}(\text{NH}_3\text{SO}_3)$  (M = Rb, Cs; X = Cl, Br) obtained from SEM-EDX.

Element	$\text{RbCl}(\text{NH}_3\text{SO}_3)$		$\text{CsCl}(\text{NH}_3\text{SO}_3)$		$\text{RbBr}(\text{NH}_3\text{SO}_3)$		$\text{CsBr}(\text{NH}_3\text{SO}_3)$	
	Wt %	Atomic %	Wt %	Atomic %	Wt %	Atomic %	Wt %	Atomic %
Rb	38.24	12.79	/	/	28.81	11.47	/	/
Cs	/	/	39.30	8.43	/	/	37.41	9.87
Cl	15.23	12.28	10.93	8.79	/	/	/	/
Br	/	/	/	/	30.92	13.16	22.56	9.90
O	25.79	46.07	29.17	52.02	20.84	44.32	22.46	49.24
S	11.73	10.46	9.76	8.69	11.80	12.52	9.24	10.11
N	9.01	18.40	10.84	22.07	7.63	18.53	8.33	20.87
<b>Total</b>	100		100		100		100	

**Table S12.** Calculated and experimental residual weight for  $\text{MX}(\text{NH}_3\text{SO}_3)$  (M = Rb, Cs; X = Cl, Br) at 800 °C in TGA.

Compound	Cal. (%)	Exp. (%)
$\text{RbCl}(\text{NH}_3\text{SO}_3)$	61.24	68.41
$\text{CsCl}(\text{NH}_3\text{SO}_3)$	68.30	73.01
$\text{RbBr}(\text{NH}_3\text{SO}_3)$	50.86	58.12
$\text{CsBr}(\text{NH}_3\text{SO}_3)$	58.41	61.97

**Table S13.** Assigned vibration peaks for  $\text{MX}(\text{NH}_3\text{SO}_3)$  (M = Rb, Cs; X = Cl, Br).

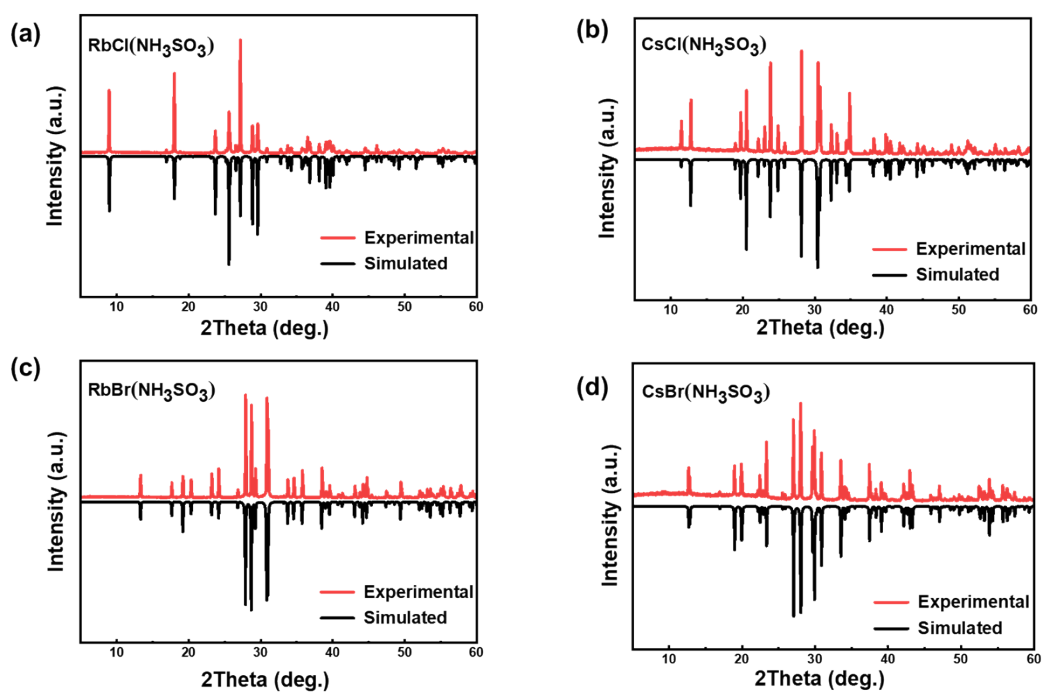
Functional group	Vibration type	$\text{RbCl}(\text{NH}_3\text{SO}_3)$	$\text{CsCl}(\text{NH}_3\text{SO}_3)$	$\text{RbBr}(\text{NH}_3\text{SO}_3)$	$\text{CsBr}(\text{NH}_3\text{SO}_3)$
		Wavenumber ( $\text{cm}^{-1}$ )			
NH <sub>3</sub>	stretching	3166	3161	3202	3207
NH <sub>3</sub>	Wagging	2436	2432		
NH <sub>3</sub>	bending	1425	1424	1410	1414
SO <sub>3</sub>	Antisymmetric and	1265	1273	1260	1249
		1060	1063	1067	1061
	symmetric stretching	978	987	990	1002
		691	704	694	680

**Table S14.** Comparison of the non- $\pi$ -conjugated optical materials derivatives in the short-wave UV region.

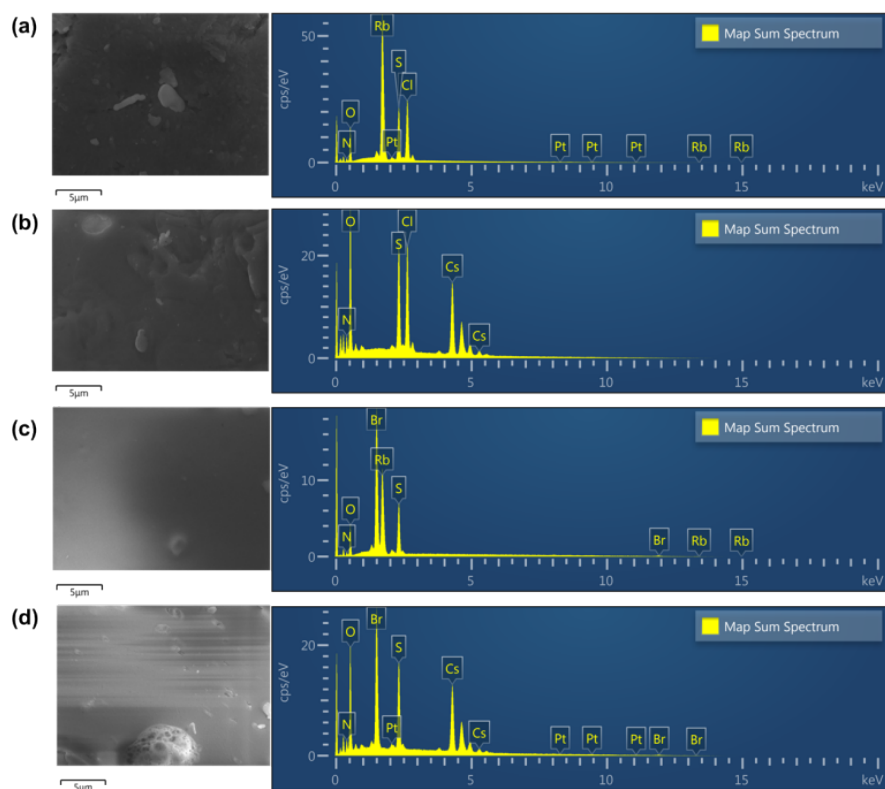
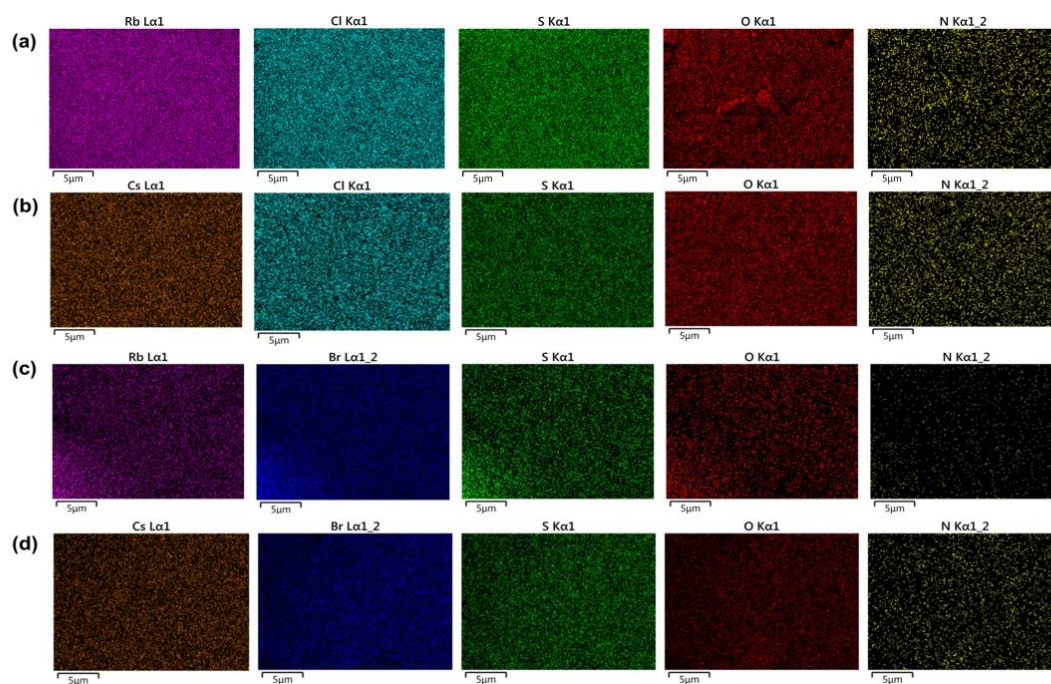
	<b>Compound</b>	<b>Space group</b>	<b>Cut-off edge (nm)</b>	<b>Birefringence</b>	
1	RbCl(NH <sub>3</sub> SO <sub>3</sub> )	<i>P2<sub>1</sub>/m</i>	185	0.069 @ 546 nm <sup>a</sup>	This work
2	CsCl(NH <sub>3</sub> SO <sub>3</sub> )	<i>P2<sub>1</sub>/m</i>	185	0.073 @ 546 nm <sup>a</sup>	This work
3	RbBr(NH <sub>3</sub> SO <sub>3</sub> )	<i>Pnma</i>	193	0.072 @ 546 nm <sup>a</sup>	This work
4	CsBr(NH <sub>3</sub> SO <sub>3</sub> )	<i>Pnma</i>	195	0.075 @ 546 nm <sup>a</sup>	This work
<b>Phosphates</b>					
5	NaBaAl(PO <sub>4</sub> ) <sub>2</sub>	<i>P2<sub>1</sub>/c</i>	<190	0.005 @ 589 nm <sup>a</sup>	[1]
6	Na <sub>2</sub> Ca <sub>17</sub> Al <sub>2</sub> (PO <sub>4</sub> ) <sub>14</sub>	<i>R3c</i>	179	0.012 @ 1064 nm <sup>b</sup>	[2]
7	Mg <sub>2</sub> PO <sub>4</sub> Cl	<i>Pna2<sub>1</sub></i>	<190	0.046 @ 1064 nm <sup>b</sup>	[3]
8	K <sub>2</sub> SrP <sub>4</sub> O <sub>12</sub>	$\bar{I}4$	<200	0.016 @ 1064 nm <sup>b</sup>	[4]
9	LiRb <sub>2</sub> PO <sub>4</sub>	<i>Cmc2<sub>1</sub></i>	<170	0.009 @ 1064 nm <sup>b</sup>	[5]
10	LiCs <sub>2</sub> PO <sub>4</sub>	<i>Cmc2<sub>1</sub></i>	174	0.010 @ 1064 nm <sup>b</sup>	[6]
11	KMg(H <sub>2</sub> O)PO <sub>4</sub>	<i>Pmn2<sub>1</sub></i>	<200	0.017 @ 1064 nm <sup>b</sup>	[7]
12	KH <sub>2</sub> PO <sub>4</sub>	$\bar{I}4_2d$	180	0.035 @ 1064 nm <sup>a</sup>	[8]
13	NaBa <sub>2</sub> Al(P <sub>2</sub> O <sub>7</sub> ) <sub>2</sub>	<i>P1</i>	<190	0.007 @ 589 nm <sup>a</sup>	[1]
14	K <sub>4</sub> Mg <sub>4</sub> (P <sub>2</sub> O <sub>7</sub> ) <sub>3</sub>	<i>Pc</i>	170	0.0108 @ 1064 nm <sup>a</sup>	[9]
15	RbNaMgP <sub>2</sub> O <sub>7</sub>	<i>Pna2<sub>1</sub>/Cmc2<sub>1</sub></i>	185	0.035 @ 1064 nm <sup>b</sup>	[10]
16	$\beta$ -Cd(PO <sub>3</sub> ) <sub>2</sub>	<i>P2<sub>1</sub>2<sub>1</sub>2<sub>1</sub></i>	<190	0.059 @ 1064 nm <sup>b</sup>	[11]
17	La(PO <sub>3</sub> ) <sub>3</sub>	<i>C222<sub>1</sub></i>	<190	0.040 @ 1064 nm <sup>b</sup>	[11]
18	CsLa(PO <sub>3</sub> ) <sub>4</sub>	<i>P2<sub>1</sub></i>	167	0.006 @ 1064 nm <sup>b</sup>	[12]
19	RbBa <sub>2</sub> (PO <sub>3</sub> ) <sub>5</sub>	<i>Pc</i>	163	0.009 @ 1064 nm <sup>b</sup>	[13]
20	KLa(PO <sub>3</sub> ) <sub>4</sub>	<i>P2<sub>1</sub></i>	163	0.0084 @ 1064 nm <sup>b</sup>	[14]
21	Ba <sub>3</sub> P <sub>3</sub> O <sub>10</sub> Cl	<i>Pca2<sub>1</sub></i>	180	0.028 @ 1064 nm <sup>b</sup>	[15]
22	Ba <sub>3</sub> P <sub>3</sub> O <sub>10</sub> Br	<i>P2<sub>1</sub>2<sub>1</sub>2<sub>1</sub></i>	<200	0.023 @ 1064 nm <sup>b</sup>	[15]
23	LiHgPO <sub>4</sub>	<i>P4<sub>2</sub>/m</i>	<300	0.0068 @ 1064 nm <sup>b</sup>	[16]
24	K <sub>2</sub> TeP <sub>2</sub> O <sub>8</sub>	<i>P2<sub>1</sub>2<sub>1</sub>2<sub>1</sub></i>	270	0.05 @ 1064 nm <sup>b</sup>	[18]
25	CsMgPO <sub>4</sub> ·6H <sub>2</sub> O	<i>P6<sub>3</sub>/mmc</i>	258	0.006 @ 1064 nm <sup>b</sup>	[18]
26	RbMgPO <sub>4</sub> ·6H <sub>2</sub> O	<i>Pmn2<sub>1</sub></i>	288	0.005 @ 1064 nm <sup>b</sup>	[18]
<b>Sulfates</b>					
27	Li <sub>8</sub> NaRb <sub>3</sub> (SO <sub>4</sub> ) <sub>6</sub> ·2H <sub>2</sub> O	<i>C2</i>	<190	0.021 @ 1064 nm <sup>a</sup>	[19]
28	NH <sub>4</sub> NaLi <sub>2</sub> (SO <sub>4</sub> ) <sub>2</sub>	<i>C2</i>	<186	0.008 @ 1064 nm <sup>b</sup>	[20]
29	La(NH <sub>4</sub> )(SO <sub>4</sub> ) <sub>2</sub>	<i>Pmn2<sub>1</sub></i>	<190	0.03 @ 1064 nm <sup>b</sup>	[21]
30	LiKSO <sub>4</sub>	<i>P6<sub>3</sub></i>	160	0.03 @ 1064 nm <sup>b</sup>	[22]
31	LiN <sub>2</sub> H <sub>5</sub> SO <sub>4</sub>	<i>Pna2<sub>1</sub></i>	<190	0.0126 @ 1064 nm <sup>b</sup>	[23]
32	K <sub>2</sub> Zn <sub>3</sub> (SO <sub>4</sub> )(HSO <sub>4</sub> ) <sub>2</sub> F <sub>4</sub>	<i>Cmc2<sub>1</sub></i>	<200	0.01268 @ 546 nm <sup>a</sup>	[24]
33	KZnSO <sub>4</sub> F	<i>Pna2<sub>1</sub></i>	<190	0.022 @ 550 nm <sup>b</sup>	[25]
34	KMgSO <sub>4</sub> F	<i>Pna2<sub>1</sub></i>	170	0.012 @ 550 nm <sup>b</sup>	[26]
35	KSO <sub>3</sub> CF <sub>3</sub>	<i>P2<sub>1</sub></i>	156	0.017 @ 1064 nm <sup>b</sup>	[27]
36	$\alpha$ -RbSO <sub>3</sub> CF <sub>3</sub>	<i>Cm</i>	<200	0.035 @ 1064 nm <sup>b</sup>	[28]

37	$\beta$ -RbSO <sub>3</sub> CF <sub>3</sub>	<i>P2<sub>1</sub></i>	<200	0.034 @ 1064 nm <sup>b</sup>	[28]
38	CsSO <sub>3</sub> CF <sub>3</sub>	<i>P2<sub>1</sub></i>	<200	0.027 @ 1064 nm <sup>b</sup>	[29]
39	KYSO <sub>4</sub> F <sub>2</sub>	<i>P2<sub>1</sub>/m</i>	<190	0.015 @ 546.1 nm <sup>a</sup>	[30]
40	RbYSO <sub>4</sub> F <sub>2</sub>	<i>P2<sub>1</sub>/m</i>	<190	0.02 @ 546.1 nm <sup>a</sup>	[30]
41	(NH <sub>4</sub> ) <sub>2</sub> BeS <sub>2</sub> O <sub>8</sub>	$\bar{I}A2d$	<200	0.019 @ 546.1 nm <sup>a</sup>	[31]
42	K <sub>2</sub> BeS <sub>2</sub> O <sub>8</sub>	$\bar{I}A2d$	<200	0.024 @ 546.1 nm <sup>a</sup>	[31]
43	Rb <sub>2</sub> BeS <sub>2</sub> O <sub>8</sub>	$\bar{I}A2d$	<200	0.024 @ 546.1 nm <sup>a</sup>	[31]
44	Cs <sub>2</sub> BeS <sub>2</sub> O <sub>8</sub>	$\bar{I}A2d$	<200	0.013 @ 546.1 nm <sup>a</sup>	[31]
45	Li <sub>2</sub> SO <sub>4</sub> ·H <sub>2</sub> O	<i>P2<sub>1</sub></i>	240	0.023 @ 1064 nm <sup>b</sup>	[32]
46	K <sub>2</sub> Bi <sub>2</sub> (SO <sub>4</sub> ) <sub>2</sub> Cl <sub>4</sub>	<i>P2<sub>1</sub>2<sub>1</sub>2<sub>1</sub></i>	278	0.056 @ 1064 nm <sup>b</sup>	[33]
47	(NH <sub>4</sub> ) <sub>2</sub> Bi <sub>2</sub> (SO <sub>4</sub> ) <sub>2</sub> Cl <sub>4</sub>	<i>P2<sub>1</sub>2<sub>1</sub>2<sub>1</sub></i>	273	0.055 @ 1064 nm <sup>b</sup>	[33]
48	Rb <sub>2</sub> Bi <sub>2</sub> (SO <sub>4</sub> ) <sub>2</sub> Cl <sub>4</sub>	<i>P2<sub>1</sub>2<sub>1</sub>2<sub>1</sub></i>	276	0.047 @ 1064 nm <sup>b</sup>	[33]
49	Rb <sub>3</sub> In(SO <sub>4</sub> ) <sub>3</sub>	<i>R3c</i>	215	0.019 @ 1064 nm <sup>b</sup>	[34]
50	Sb <sub>6</sub> O <sub>7</sub> (SO <sub>4</sub> ) <sub>2</sub>	<i>Ccc2</i>	286	0.052 @ 1064 nm <sup>b</sup>	[35]
51	CsY(SO <sub>4</sub> ) <sub>2</sub> ·4H <sub>2</sub> O	<i>P2<sub>1</sub>/c</i>	200	0.045 @ 546 nm <sup>b</sup>	[36]
52	NaRbY <sub>2</sub> (SO <sub>4</sub> ) <sub>4</sub>	<i>C2/c</i>	200	0.045 @ 550 nm <sup>a</sup>	[37]
53	NaBi(SO <sub>4</sub> ) <sub>2</sub> ·H <sub>2</sub> O	<i>P3<sub>2</sub>2<sub>1</sub></i>	280	0.068 @ 546 nm <sup>a</sup>	[38]
54	ZrF <sub>2</sub> (SO <sub>4</sub> )	<i>Pca2<sub>1</sub></i>	206	0.052 @ 1064 nm <sup>a</sup>	[39]
55	(NH <sub>4</sub> ) <sub>2</sub> S <sub>2</sub> O <sub>3</sub>	<i>C2</i>	238	0.077 @ 546 nm <sup>a</sup>	[40]
<b>Sulfamates</b>					
56	Pb(NH <sub>2</sub> SO <sub>3</sub> ) <sub>2</sub> ·H <sub>2</sub> O	<i>P2<sub>1</sub>/c</i>	226	0.032 @ 1064 nm <sup>b</sup>	[41]
57	Pb(NH <sub>2</sub> SO <sub>3</sub> ) <sub>2</sub>	<i>P2<sub>1</sub>/c</i>	208	0.023 @ 1064 nm <sup>b</sup>	[41]
58	Ca(NH <sub>2</sub> SO <sub>3</sub> ) <sub>2</sub> ·4H <sub>2</sub> O	<i>C2/c</i>	<200	0.025 @ 1064 nm <sup>b</sup>	[41]
59	Ca(NH <sub>2</sub> SO <sub>3</sub> ) <sub>2</sub> ·H <sub>2</sub> O	<i>P2<sub>1</sub>2<sub>1</sub>2<sub>1</sub></i>	<200	0.033 @ 1064 nm <sup>b</sup>	[41]
60	Cs <sub>2</sub> Mg(NH <sub>2</sub> SO <sub>3</sub> ) <sub>4</sub> ·4H <sub>2</sub> O	<i>Cm</i>	<180	0.054 @ 546.1 nm <sup>a</sup>	[42]
61	K <sub>2</sub> Ca(NH <sub>2</sub> SO <sub>3</sub> ) <sub>4</sub>	$P\bar{1}$	<200	0.035 @ 523 nm <sup>b</sup>	[42]
62	Rb <sub>2</sub> Ca(NH <sub>2</sub> SO <sub>3</sub> ) <sub>4</sub>	$P\bar{1}$	<200	0.036 @ 523 nm <sup>b</sup>	[42]
63	Cd(NH <sub>2</sub> SO <sub>3</sub> ) <sub>2</sub> ·2H <sub>2</sub> O	$P\bar{1}$	212	0.052 @ 1064 nm <sup>b</sup>	[43]
64	Cd(NH <sub>2</sub> SO <sub>3</sub> ) <sub>2</sub>	<i>P2<sub>1</sub>2<sub>1</sub>2<sub>1</sub></i>	210	0.037 @ 1064 nm <sup>b</sup>	[43]
65	Ba(SO <sub>3</sub> NH <sub>2</sub> ) <sub>2</sub>	<i>Pna2<sub>1</sub></i>	<190	0.028 @ 546.1 nm <sup>a</sup>	[44]
66	Sr(SO <sub>3</sub> NH <sub>2</sub> ) <sub>2</sub>	<i>Pc</i>	<190	0.056 @ 589.3 nm <sup>a</sup>	[44]
<b>Others</b>					
67	NH <sub>3</sub> BH <sub>3</sub>	<i>I4mm</i>	<190	0.056 @ 550 nm <sup>a</sup>	[45]
68	KN(SO <sub>2</sub> F) <sub>2</sub>	<i>Pbca</i>	182	0.061 @ 546 nm <sup>b</sup>	[46]
69	(NH <sub>4</sub> ) <sub>2</sub> PO <sub>3</sub> F	<i>Pna2<sub>1</sub></i>	176	0.035 @ 532 nm <sup>b</sup>	[47]
70	NaNH <sub>4</sub> PO <sub>3</sub> F·H <sub>2</sub> O	<i>Pn</i>	176	0.053 @ 532 nm <sup>b</sup>	[48]
71	Na <sub>2</sub> PO <sub>3</sub> F	<i>P2<sub>1</sub>2<sub>1</sub>2<sub>1</sub></i>	<190	0.036 @ 532 nm <sup>b</sup>	[49]
72	KHPO <sub>3</sub> F	<i>P2<sub>1</sub></i>	<190	0.028 @ 532 nm <sup>b</sup>	[50]
73	NaPO <sub>3</sub> NH <sub>3</sub>	<i>P6<sub>3</sub></i>	<190	0.062 @ 546.1 nm <sup>a</sup>	[51]
74	Ba(SO <sub>3</sub> CH <sub>3</sub> ) <sub>2</sub>	<i>Cmc2<sub>1</sub></i>	159	0.04 @ 589.3 nm <sup>a</sup>	[52]
75	Na <sub>2</sub> S <sub>3</sub> O <sub>6</sub>	<i>Fdd2</i>	212	0.056 @ 546 nm <sup>a</sup>	[53]
76	K <sub>2</sub> S <sub>4</sub> O <sub>6</sub>	<i>Cc</i>	298	0.066 @ 1064 nm <sup>b</sup>	[54]

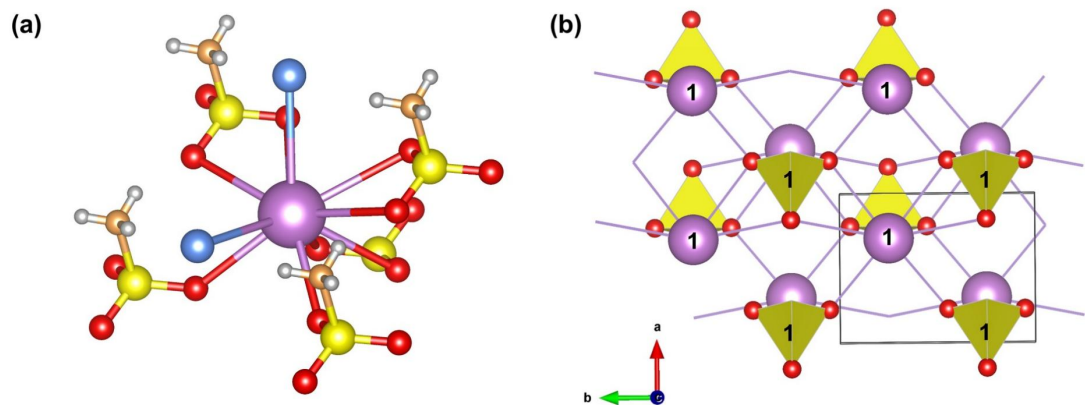
<sup>a</sup>Measured birefringence, <sup>b</sup>calculated birefringence.



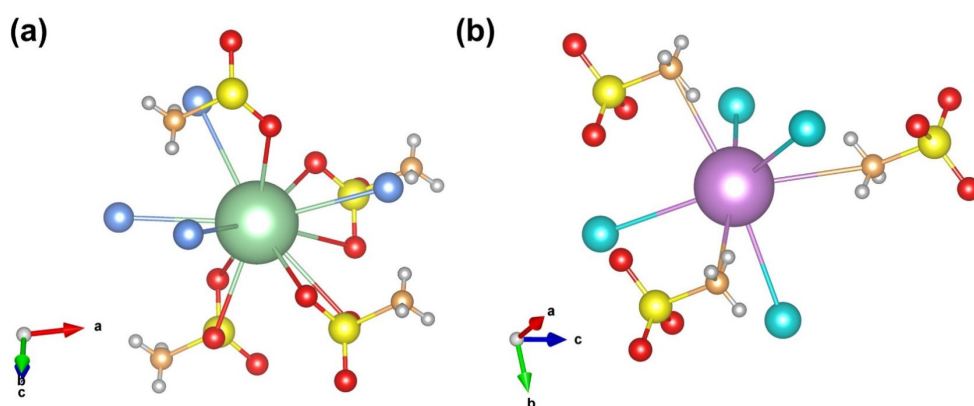
**Figure S1.** Experimental and simulated PXRD patterns for MX(NH<sub>3</sub>SO<sub>3</sub>) (M = Rb, Cs; X = Cl, Br).



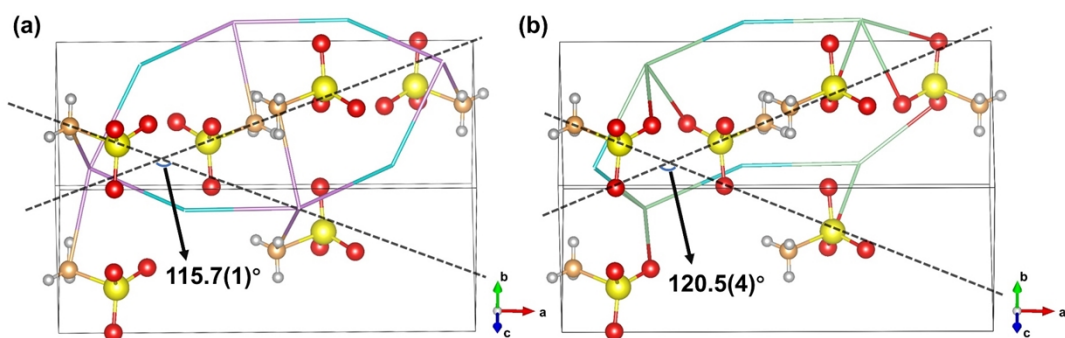
**Figure S2.** SEM-EDX for MX(NH<sub>3</sub>SO<sub>3</sub>) (M = Rb, Cs; X = Cl, Br).



**Figure S3.** (a) The asymmetric unit and (b) pseudo-doubled layer along the  $ab$  planes for  $\text{RbCl}(\text{NH}_3\text{SO}_3)$ .



**Figure S4.** (a) The asymmetric unit for  $\text{CsCl}(\text{NH}_3\text{SO}_3)$  and (b) the asymmetric unit for  $\text{RbBr}(\text{NH}_3\text{SO}_3)$ .



**Figure S5.** The intersection angle of the  $\text{NH}_3\text{SO}_3$  tetrahedra in (a)  $\text{RbBr}(\text{NH}_3\text{SO}_3)$  and (b)  $\text{CsBr}(\text{NH}_3\text{SO}_3)$ .

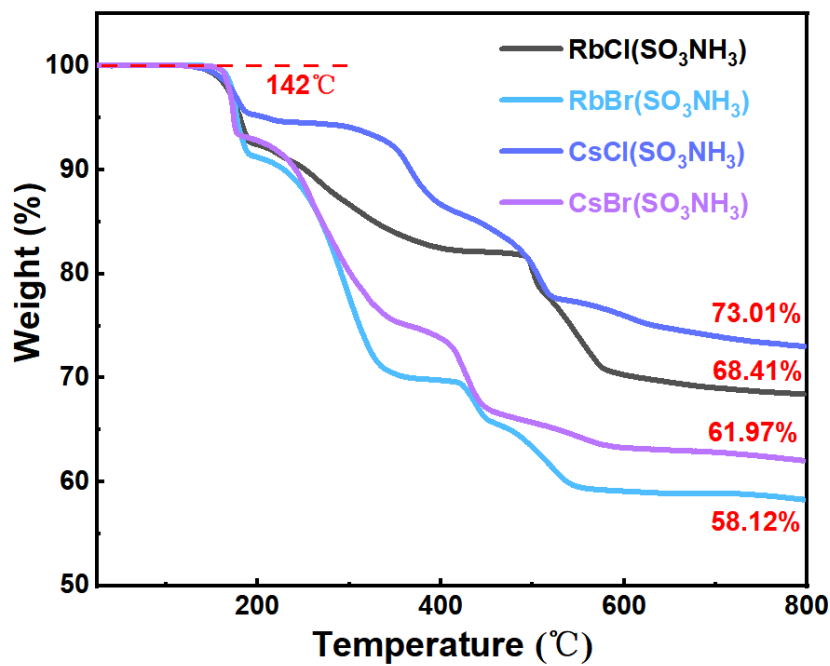


Figure S6. TGA diagrams for MX(NH<sub>3</sub>SO<sub>3</sub>) (M = Rb, Cs; X = Cl, Br).

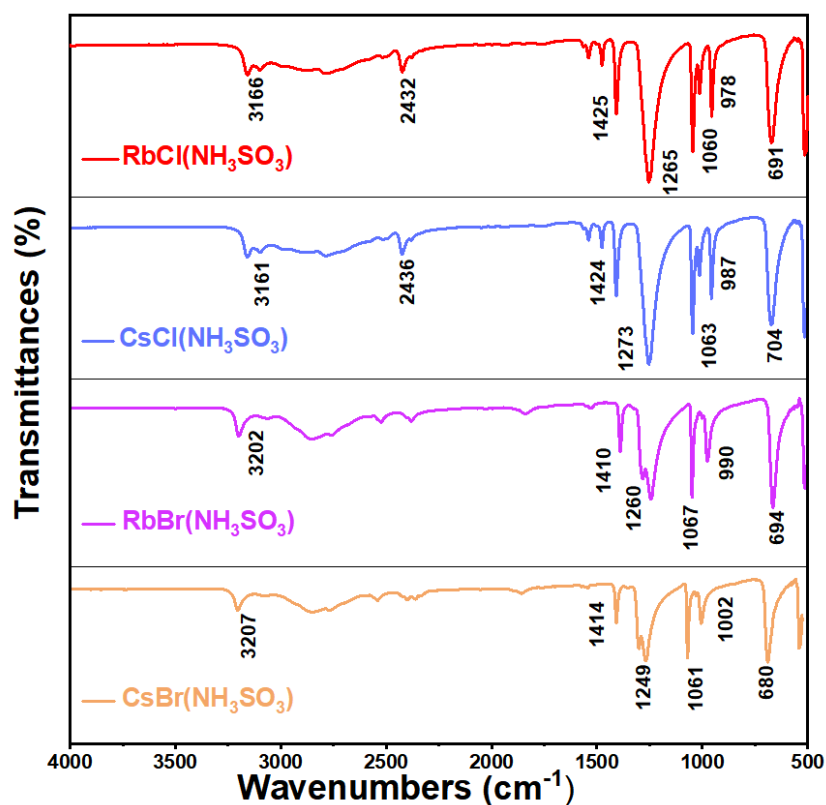
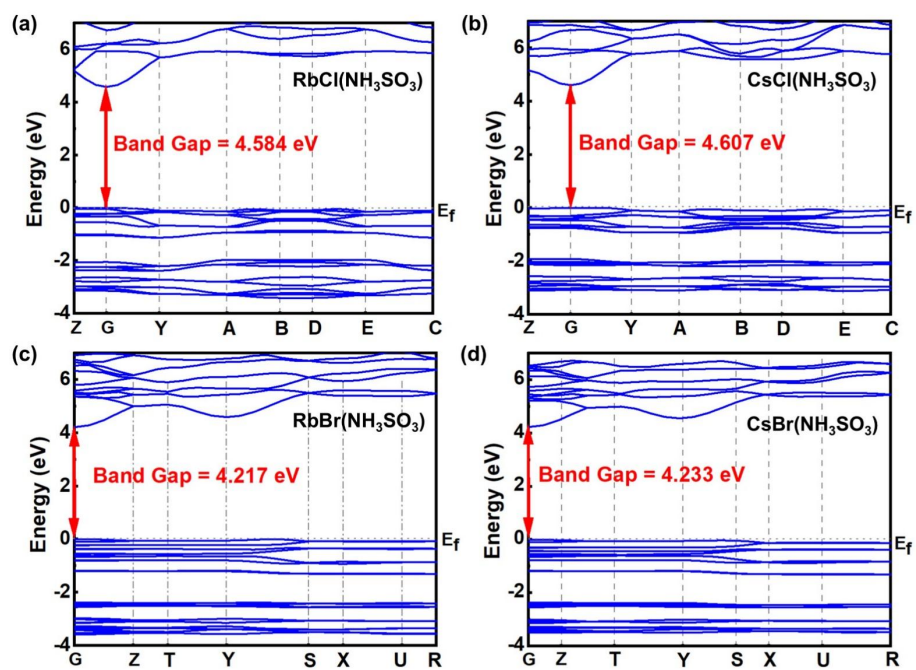


Figure S7. IR spectra for MX(NH<sub>3</sub>SO<sub>3</sub>) (M = Rb, Cs; X = Cl, Br).



**Figure S8.** Band structures of compounds for MX(NH<sub>3</sub>SO<sub>3</sub>) (M = Rb, Cs; X = Cl, Br).



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