

## Supporting Information

# Thermoelectric performance of organic conductors

*Takehiko Mori*

Department of Materials Science and Engineering, Tokyo Institute of Technology, O-  
okayama 2-12-1, Meguro-ku, Tokyo 152-8552, Japan. E-mail: [mori.t@mac.titech.ac.jp](mailto:mori.t@mac.titech.ac.jp)

### Calculation of thermoelectric power

Eqn (2) is numerically integrated.<sup>47,48</sup>

$$K_0 = \tau \iiint v^2 \left( -\frac{df_0}{dE} \right) dk^3 = \frac{\tau}{\hbar^2} \frac{N_x \sum \Delta E^2 \left( -\frac{df_0}{dE} \right) a\pi}{N_y N_z bc} \quad (\text{S1})$$

The second form for  $v_x$  is derived from the following relation.

$$v_x = \frac{1}{\hbar} \frac{\partial E(k)}{\partial k_x} = \frac{N_x a \Delta E}{\hbar \pi} \quad (\text{S2})$$

The Brillouin zone is divided by a mesh of  $N_x \times N_y \times N_z$ , and  $E(k)$  is calculated at the respective points.  $E(k)$  is estimated from the tight-binding energy band considering only the highest occupied molecular orbital (HOMO) or lowest unoccupied molecular orbital (LUMO).<sup>S1,S2</sup> The interval of  $k$  contains  $\Delta k_x = \pi/aN_x$ , where  $\Delta k_x$  appears not only in the denominator coming from eqn (S2) but also in the numerator from the integration in eqn (S1). Accordingly,  $N_y N_z$  appears as the number of the summation, and  $N_x$  comes from the segment of the  $v_x$  calculation. In the actual calculation, the coefficient of eqn (S1) is  $\pi e^3/\hbar^2 = 1.2867 \times 10^{14} \text{ C}^2 \text{ \AA} \text{ J}^{-1} \text{ s}^{-2} \text{ cm}^{-2}$ , where  $(-df_0/dE)$  is proportional to inverse temperature and represented in  $\text{eV}^{-1}$  unit. Eqn (3) is estimated similarly.

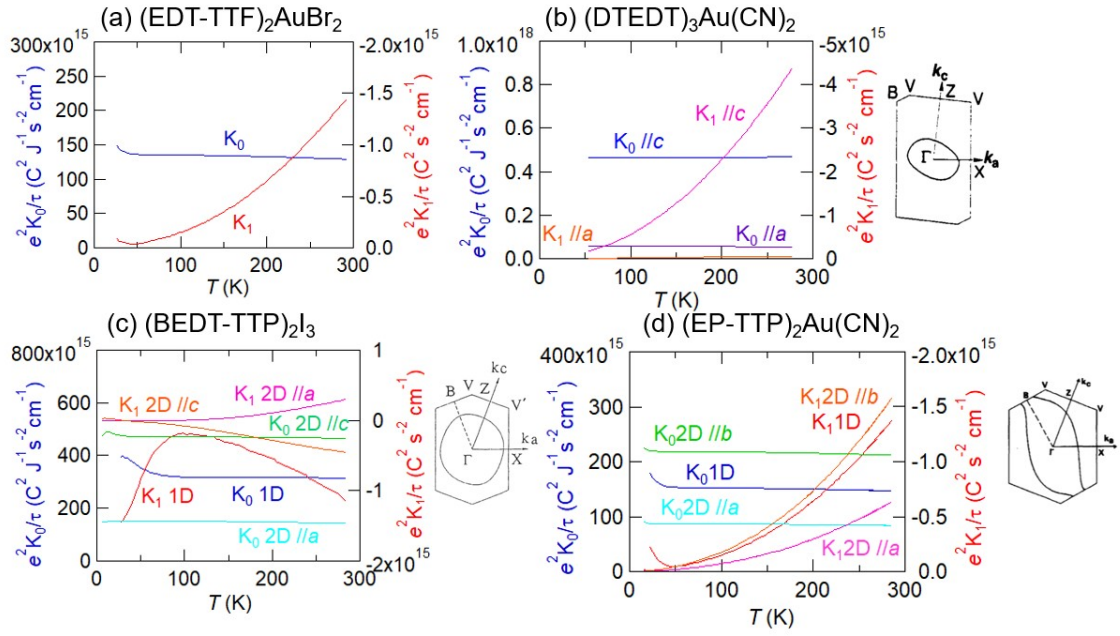
For  $\beta$ -(BEDT-TTF)<sub>2</sub>I<sub>3</sub>,  $\kappa$ -(BEDT-TTF)<sub>2</sub>Cu(NCS)<sub>2</sub>, and  $\alpha$ -(BEDT-TTF)<sub>2</sub>NH<sub>4</sub>Hg(SCN)<sub>4</sub>, transfer integrals that are optimized so as to reproduce the observed thermoelectric power are used.<sup>47,48</sup> The reported transfer integrals are used for other cases.

When the Fermi energy crosses more than two energy bands,  $K_0$  and  $K_1$  are summed for these bands.

$$S = \frac{1}{eT} \frac{\sum_i K_{1i}}{\sum_i K_{0i}} = \frac{1}{eT} \frac{\sum_i (K_{1i}/K_{0i}) K_{0i}}{\sum_i K_{0i}} = \frac{\sum_i S_i \sigma_i}{\sum_i \sigma_i} \quad (\text{S3})$$

Here,  $K_{0i}$  represents  $K_0$  for different energy bands  $i$ . Using eqn (5), this is a weighted average of  $S_i$  with respect to  $\sigma_i$ .

**$K_0$  and  $K_1$  of TTF donors**

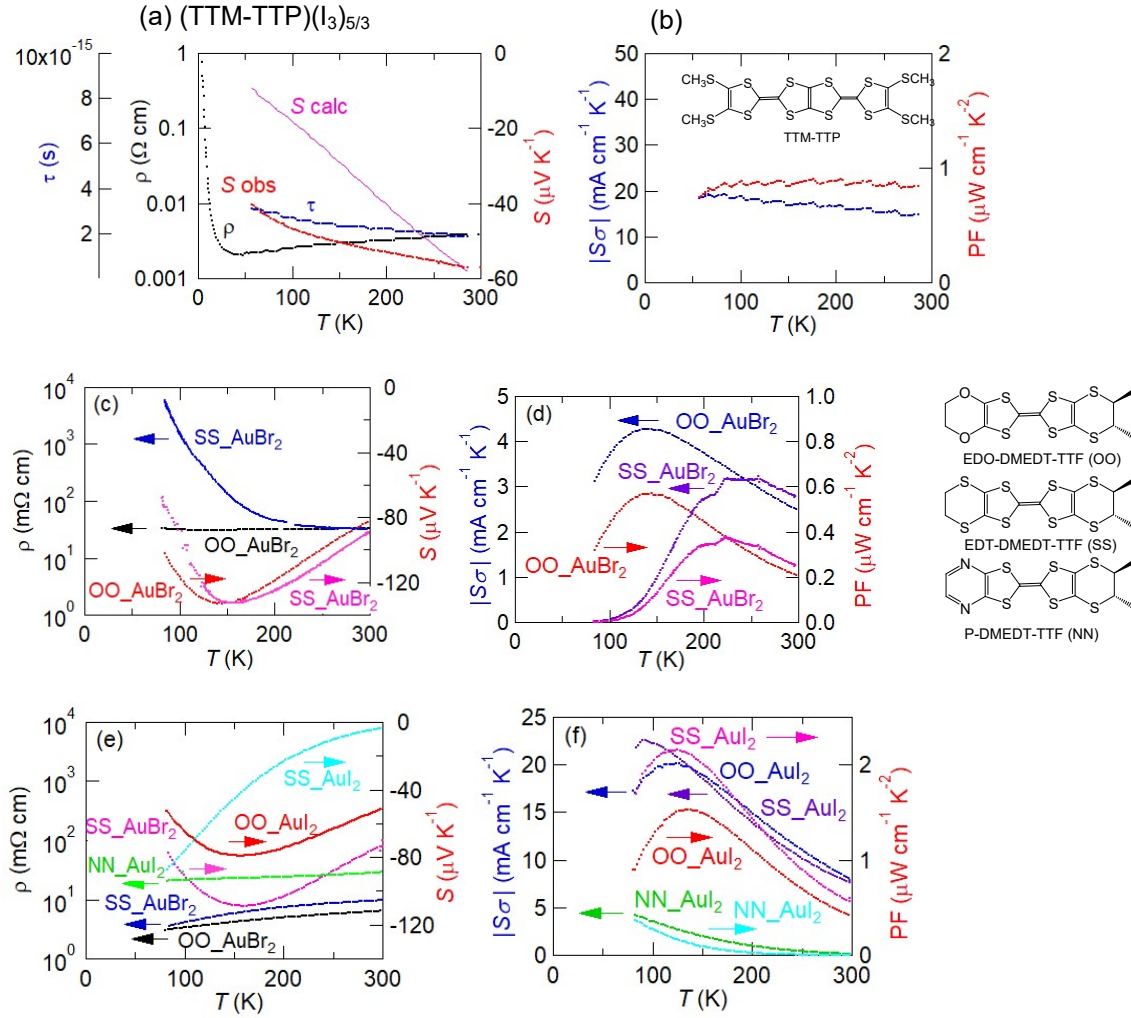


**Fig. S1** Calculated  $K_0$  and  $K_1$  integrals, and the Fermi surface of (a)  $(\text{EDT-TTF})_2\text{AuBr}_2$ , (b)  $(\text{DTEDT})_3\text{Au}(\text{CN})_2$ , (c)  $(\text{BEDT-TTP})_2\text{I}_3$ , and (d)  $(\text{EP-TTP})_2\text{Au}(\text{CN})_2$ .

**Table S1** Relaxation time ( $\tau$ ) and calculated  $K_0$  and  $K_1$  integrals.

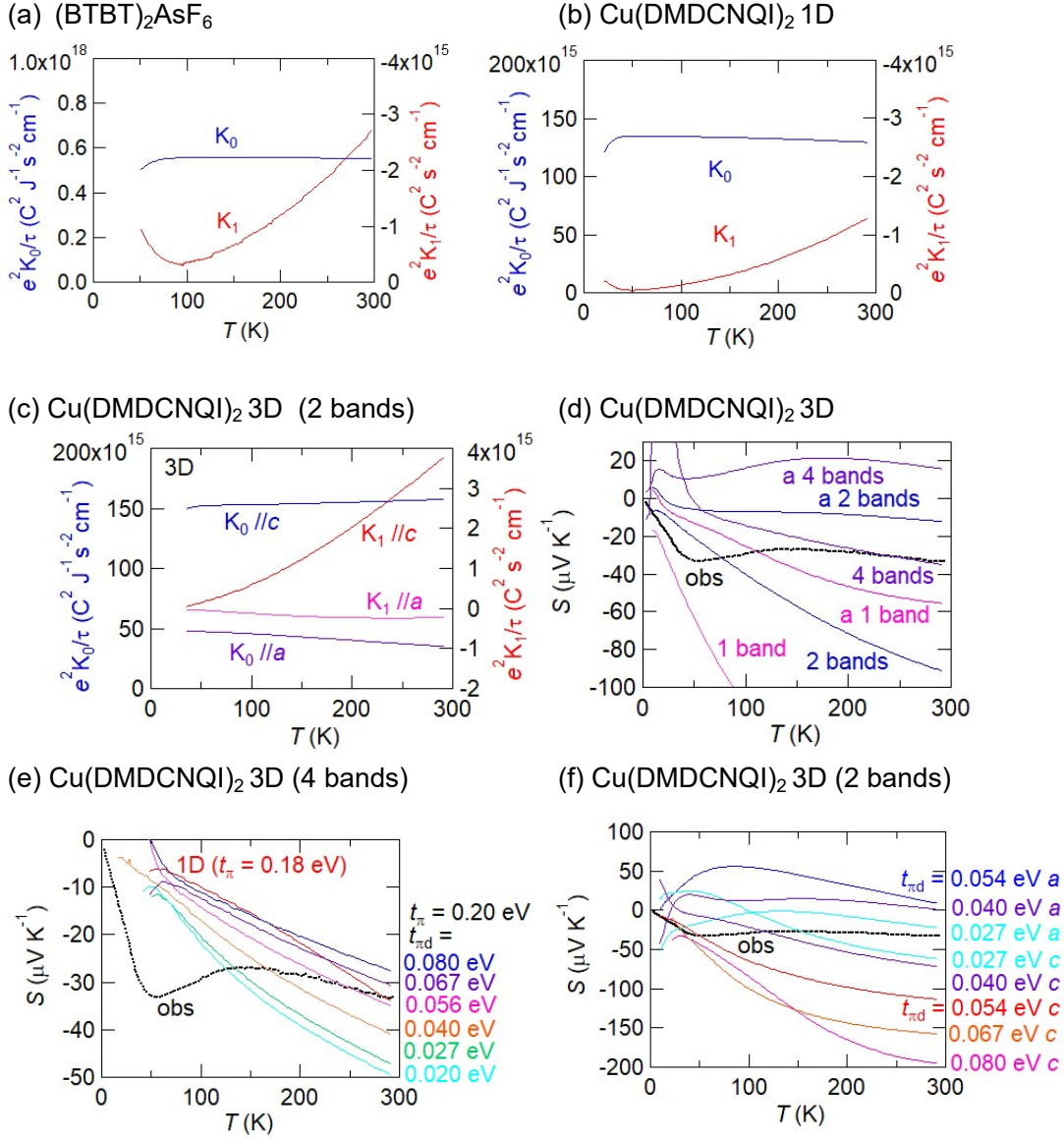
Compound	axis	$\tau_{rt}$ $\times 10^{-15}$ (s)	$\tau_{max}$ $\times 10^{-15}$ (s)	axis	$e^2K_0/\tau$ (rt) $\times 10^{15}$ ( $C^2 J^{-1} s^{-2} cm^{-1}$ )	$e^2K_1/\tau$ (rt) $\times 10^{15}$ ( $C^2 s^{-2} cm^{-1}$ )
(TTF)(TCNQ)	stack	7	70	stack	58	0.49
(TMTSF) <sub>2</sub> PF <sub>6</sub>	stack	1.6	600	stack	310	-2.7
$\beta$ -(BEDT-TTF) <sub>2</sub> I <sub>3</sub>	<i>a</i>	0.3	100	<i>a/b</i>	120/210	-1.2/-1.5
$\kappa$ -(BEDT-TTF) <sub>2</sub> Cu(NCS) <sub>2</sub>	<i>c</i>	0.7	130	<i>c/b</i>	14/8	-0.73/0.62
$\alpha$ -(BEDT-TTF) <sub>2</sub> KHg(SCN) <sub>4</sub>	<i>c</i>	0.1	1.8	<i>c/a</i>	55/560	-0.2/2.1
$\alpha$ -(BEDT-TTF) <sub>2</sub> NH <sub>4</sub> Hg(SCN) <sub>4</sub>	<i>c</i>	1.2	700	<i>c/a</i>	19/210	-0.17/1.8
$\alpha$ -(BEDT-TTF) <sub>2</sub> I <sub>3</sub>	<i>a</i>	0.3	1.5	<i>a/b</i>	56/47	-1.6/0.70
$\theta$ -(BEDT-TTF) <sub>2</sub> CsCo(SCN) <sub>4</sub>	<i>c</i>	3	7	<i>c/a</i>	12/82	-0.12/-0.46
$\beta$ -(BEDT-TTF) <sub>2</sub> PF <sub>6</sub>	<i>c</i>	0.6	0.72	<i>c/a</i>	14/71	-0.21/0.47
$\beta''$ -(BEDT-TTF) <sub>4</sub> Pd(CN) <sub>4</sub> H <sub>2</sub> O	<i>a</i>	0.4		<i>a/c</i>	210/400	-0.74/-2.2
(EDT-TTF) <sub>2</sub> AuBr <sub>2</sub>	stack	1.2	2.6	stack	130	-1.4
(DTEDT) <sub>3</sub> Au(CN) <sub>2</sub>	<i>c</i>	3.2	48	<i>c/a</i>	470/54	-4.4/0.054
(BEDT-TTP) <sub>2</sub> I <sub>3</sub>	<i>c</i>	2.2	75	<i>c/a</i>	470/140	-0.45/0.30
(EP-TTP) <sub>2</sub> Au(CN) <sub>2</sub>	stack	7.2	100	<i>a/c</i>	210/84	-1.6/-0.63
(TTM-TTP)(I <sub>3</sub> ) <sub>3/5</sub>	stack	1.8	3.0	stack	140	2.3
(BTBT) <sub>2</sub> AsF <sub>6</sub>	stack	12	43	stack	550	-2.7
Cu(DMDCNQI) <sub>2</sub>	stack	8	1600	stack	130	1.3

## TTM-TTP and $\tau$ -phase conductors



**Fig. S2** (a) Resistivity,  $\tau$ , and thermoelectric power, and (b)  $S\sigma$  and PF of (TTMTTP)(I<sub>3</sub>)<sub>5/3</sub>. (c) Resistivity,  $\tau$ , and thermoelectric power, and (d)  $S\sigma$  and PF of  $\tau$ -(EDO-DMEDT-TTF)<sub>2</sub>AuBr<sub>2</sub> and  $\tau$ -(EDT-DMEDT-TTF)<sub>2</sub>AuBr<sub>2</sub>. (e) Resistivity,  $\tau$ , and thermoelectric power, and (f)  $S\sigma$  and PF of  $\tau$ -(EDO-DMEDT-TTF)<sub>2</sub>AuI<sub>2</sub>,  $\tau$ -(EDT-DMEDT-TTF)<sub>2</sub>AuI<sub>2</sub>, and  $\tau$ -(P-DMEDT-TTF)<sub>2</sub>AuI<sub>2</sub>. EDO-DMEDT-TTF: ethylenedioxy-*S,S*-dimethylethylenedithio-TTF EDT-DMEDT-TTF: ethylenedithio-*S,S*-dimethylethylenedithio-TTF, and P-DMEDT-TTF: pyrazino-*S,S*-dimethylethylenedithio-TTF.

## BTBT and DMDCNQI



**Fig. S3** Calculated  $K_0$  and  $K_1$  integrals of (a)  $(\text{BTBT})_2\text{AsF}_6$ , (b)  $\text{Cu}(\text{DMDCNQI})_2$  based on the one-dimensional band, and (c) the three-dimensional two bands. (d) Calculated  $S$  along the  $c$  and  $a$  axes using the upper one, two, and four bands of  $\text{Cu}(\text{DMDCNQI})_2$  with  $t_c: 0.2$ ,  $t_{\pi d}: 0.054$ , and  $\Delta: 0.3$  eV.<sup>76</sup>  $t_{\pi d}$  dependence of  $S$  calculated for (e) upper four bands, and (f) two bands.

Due to the body centered symmetry, a unit of  $\text{Cu}(\text{DMDCNQI})_2$  contains four DMDCNQI molecules and two Cu atoms.<sup>75</sup> Considering the LUMO of DMDCNQI and a single 3d orbital of Cu, the energy bands consist of six bands (Fig. 7(e)). The resulting energy bands are identical to Ref. 77 using the same parameters,  $t_c: 0.2$ ,  $t_{\pi d}: 0.054$ , and  $\Delta: 0.3$  eV.

Owing to the 3D character, the upper two bands afford large  $S$ , while  $S$  of the next two is small (Table S2). When all four bands are included, the resulting  $S$  is small (Fig. 7(f)) because  $S$  of a multiband is a weighted average of  $S$  with respect to  $K_0$  following eqn (S3). Since the conductivity of a 1D band is larger than that of a 3D band ( $K_0$  in Table S2),  $S$  is small even when a 3D band with large  $S$  participates in the conduction. Even increasing  $t_{\pi d}$ , the overall  $S$  decreases (Fig. S3(e)) because  $S$  of the 1D band decreases (Table S2). By contrast,  $S$  of the 3D band increases with increasing  $t_{\pi d}$  (Fig. S3(f)). At the same time,  $S$  along the  $a$  axis shifts to the positive direction (Fig. S3(d) and (f)).

Even in the 3D band,  $S$  decreases with lowering the temperature. The observed increasing temperature dependence of  $S$  is not reproduced even including the Cu d orbital. The observed  $S$  is reproducible when the contribution of the 1D band is lost between 200 and 100 K potentially due to the 1D instability.

**Table S2**  $S$ ,  $K_0$ , and  $K_1$  of each band in Cu(DMDCNQI)<sub>2</sub>.

$t_{\pi d}$ (eV)	$S_c$ (rt) ( $\mu\text{V K}^{-1}$ )			$e^2 K_0/\tau$ (rt) $\times 10^{15}$ ( $\text{C}^2 \text{J}^{-1} \text{s}^{-2} \text{cm}^{-1}$ )			$e^2 K_1/\tau$ (rt) $\times 10^{15}$ ( $\text{C}^2 \text{s}^{-2} \text{cm}^{-1}$ )		
	0.27	0.54	0.80	0.27	0.54	0.80	0.27	0.54	0.80
1	-141	-156	-185	21	14	8.7	0.85	0.66	0.47
2	-100	-71	-56	33	37	38	0.97	0.75	0.62
3	-20	-13	-9	66	73	78	0.39	0.28	0.21
4	-20	-13	-9	66	73	78	0.39	0.28	0.21
sum	-47	-35	-28	186	198	202	2.6	1.97	1.51

## References

- S1 T. Mori, A. Kobayashi, Y. Sasaki, H. Kobayashi, G. Saito, H. Inokuchi, The intermolecular interaction of tetrathiafulvalene and bis(ethylenedithio)tetrathiafulvalene in organic metals. Calculation of orbital overlaps and models of energy-band structures, *Bull. Chem. Soc. Jpn.* 1984, **57**, 627.
- S2 The software is available from <http://indigo1026.la.coocan.jp/lib/program.html> (accessed 1 July 2024).