Interfacial modification engineering for efficient and stable MA-free wide-bandgap perovskite

# solar cells by grain regrowth

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### **Materials**

Nickel oxide nanoparticles  $(NiO_x)$  and cesium iodide (CsI, 99.99% purity) were purchased from Advanced Election Technology Co., Ltd. Me-4PACz  $(C_{18}H_{22}NO_3P, >99.0\%$  purity), lead(II) iodide (PbI<sub>2</sub>,  $>99.99\%$  purity), lead(II) bromide (PbBr<sub>2</sub>,  $>99.99\%$  purity), formamidinium iodide  $(HC(NH<sub>2</sub>)<sub>2</sub>I, FAI, \geq 99.5\%$  purity), lead(II) chloride (PbCl<sub>2</sub>, >99.99% purity), and GuSCN were purchased from Xi'an Yuri Solar Co., Ltd. Absolute alcohol was obtained from Sinopharm Chemical Reagent Co., Ltd. Tin dioxide (SnO<sub>2</sub>) was purchased from Dongguan Nanofrontier Microelectronic Equipment Co., Ltd. Dimethyl sulfoxide (DMSO, 99.9%), N,Ndimethylformamide (DMF, 99.9%), and isopropyl alcohol (IPA, ≥99.7% purity) were purchased from Sigma-Aldrich. All materials were commercially sourced and utilized as received.

#### **Solution Preparation**

NiO<sup>x</sup> NPs were dispersed in deionized water at a concentration of 30 mg/mL using ultrasonic treatment for approximately 2 hours. Me-4PACz was dissolved in absolute alcohol at a concentration of 1 mg/mL. The perovskite solution  $(1.3 \text{ M})$  was prepared by dissolving FAI, PbI<sub>2</sub>, PbBr<sub>2</sub>, and CsI in a DMF: DMSO mixed solvent (5:1 volume ratio) with a stoichiometric ratio of  $CS_{0.2}FA_{0.8}Pb(I_{0.8}Br_{0.2})$ 3. Subsequently, 1.5 mol% of PbCl<sub>2</sub> was added to the perovskite solution, and the mixture was stirred at 60 °C for 2 hours. GuSCN was dissolved in isopropyl alcohol (IPA) to form solutions with concentrations of 15 mM, 25 mM, and 35 mM, respectively.

### **Device fabrication**

ITO conductive glass was cleaned sequentially with glass cleaner, deionized water, and isopropyl alcohol under ultrasonic treatment for 15 minutes, respectively, followed by drying with nitrogen gas. Subsequently, the substrates were treated in a UV-ozone cleaner for 30 minutes. A hole transport layer (HTL) was formed by spin-coating a  $NiO<sub>x</sub>$  nanoparticle solution onto the ITO glass at 2000 rpm for 30 seconds and annealing in air at 150 °C for 30 minutes. Solutions of Me-4PACz and GuSCN were prepared in a glovebox. A Me-4PACz precursor solution in absolute alcohol (1 mg/mL) was then spin-coated onto the  $NiO<sub>x</sub>$  substrates at 6000 rpm for 30 seconds, followed by annealing at 100 °C for 10 minutes. A two-step spin-coating process was used to deposit the perovskite film. First, the substrate was covered with the perovskite precursor solution and rotated at 1000 rpm for 15 seconds, followed by a spin-coating step at 6000 rpm for 60 seconds. During the second step, approximately 150 µL of chlorobenzene was dropped onto the center of the spinning substrate 30 seconds before the end of the program. Immediately after spin-coating, the substrates were placed on a hotplate and annealed for 10 minutes at 100 °C. After annealing at 100 °C for 10 minutes, the perovskite active layer was modified with GuSCN and subsequently annealed again at 100 °C for 10 minutes. Finally, C60 was deposited by thermal evaporation, followed by atomic layer deposition (ALD) of  $SnO<sub>2</sub>$  for 30 minutes in the air. The perovskite solar cell (PSC) fabrication was completed by thermal evaporation of the Ag electrode. The effective area of the PSC was determined using an aperture shield with an area of 0.0588 cm². To prepare transparent electrode PSCs, C60 was first deposited on the perovskite layer by thermal evaporation, followed by atomic layer deposition of  $SnO<sub>2</sub>$ . ITO layers with a thickness of 400 nm were deposited by magnetron sputtering under the following conditions: 30 W for 10 minutes and 60 W for 20 minutes. Finally, a fingerlike Ag electrode was prepared by thermal evaporation to complete the transparent electrode PSCs.

# **Characterization**

The surface and cross-sectional morphology of the perovskite active layer were analyzed using field emission scanning electron microscopy (SEM) with a Merlin microscope from Zeiss. The surface elemental distributions of perovskites without and with GuSCN modification were measured by X-ray photoelectron spectroscopy (XPS) using an Escalab 250Xi instrument from Thermo Fisher Scientific. Current density-voltage (J-V) curves for all devices were obtained with a solar simulator (Sol3A from Newport) and a digital sourcemeter (2400 from Keithley). Spacecharge limited current (SCLC) measurements were performed to evaluate the defect density of control and target devices, providing further insight into the defect passivation effect. The crystallinity of the perovskite active layer was investigated using X-ray diffraction (XRD) with a MiniFlex600 diffractometer from Rigaku. The absorbance spectra of the perovskite films were measured using a Lambda 365 UV-Vis spectrophotometer from PerkinElmer. The contact angles of the perovskite films and NiO*<sup>x</sup>* were determined using a contact angle test apparatus (SL200KS from American Kono Group). The TRPL spectra (with excitation at 430 nm and emission at 740 nm) were characterized by full-function fluorescence phosphorescence spectroscopy (FLS920, Edinburgh Instruments, EI) with an IRF of  $\approx$  0.4 ns.



**Fig. S1** (a) Perovskite with the post-treatment and without thermal annealing, (b) Perovskite with the post-treatment and thermal annealing.



**Fig. S2** SEM images of perovskite layers treated with GuSCN and thermal annealed with different

times (a) 0.5 min, (b) 1 min, (c) 3 min, (d) 5 min, and (e) 10 min.



**Fig. S3** (a) XRD patterns and (b) the 100-peak half-peak width and the 110-peak half-peak width of perovskite thin films prepared with GuSCN and thermal-annealed at different times.



**Fig. S4** SEM images of perovskite layers treated with different concentrations of GuI (a) 0 mM, (b)

15 mM, (c) 25 mM, and (d) 35 mM.



Fig. S5 SEM images of perovskite layers with different concentrations of Pb(SCN)<sub>2</sub> additive (a) 0

mol%, (b) 1 mol%, (c) 3 mol%, and (d) 5 mol%.



**Fig. S6** The current density-voltage (JV) curves of PSCs post-treated with different concentrations

of GuSCN.



**Fig. S7** SEM images of perovskite films treated with different concentrations of GuSCN and then deposited with C60/SnO<sub>2</sub>.



**Fig. S8** SEM images of perovskite films treated with different concentrations of GuSCN and then put into an ALD machine and heated at 100 ℃ for 30 minutes.

We have measured the dark JV curves of the target (with GuSCN) and control (without GuSCN) devices. The ideality factor and reverse saturation current density were calculated by using the basic equation:

$$
j = j_0 \left( e^{qV/nk_B T} - 1 \right) \tag{1}
$$

Where  $k_{B_1}$ ,  $T$ ,  $n$  and  $j_0$  represent Boltzmann constant, temperature, ideality factor and reverse

saturation current density, respectively. As the value of  $e^{qV/nk}$  is much larger than 1, we can simplify the basic equation to equation 2:

$$
j = j_0 \left( e^{qV/nk_B T} \right) \tag{2}
$$

Then taking the natural logarithm of both sides of the equation gives the following equation:

$$
ln(j) = ln(j_0) + \left(\frac{q}{nk_B T}\right)V\tag{3}
$$

 $q/k_BT_{=38.46}$ 

Finally, the slope gives  $q/nk_BT$  and the intercept gives ln  $(\dot{q}$ <sup>0</sup>.

The n and  $j_0$  of the target device are 1.78 and 1.59 x 10<sup>-10</sup> mA/cm<sup>2</sup> which are much lower than the 2.00 and 4.49 x  $10^{-9}$  mA/cm<sup>2</sup> of the control device, as exhibited in Fig. S9. We characterized the dark current curves of the devices, but limitations in the test software (Oriel IV Test Station. vi) prevented us from capturing currents below  $10^{-8}$  A. These low-current values are insignificant for our analysis since only data points with  $V > 0.5$  V will be used to fit Eq. (3) and calculate the n and  $j_0$ .



**Fig. S9** Dark current to voltage curves of target and control devices.



**Fig. S10** The current-density-voltage curves of PSCs treated with GuSCN which was added into the perovskite precursor solution with a concentration of 1%-5%.



**Fig. S11** SEM images of perovskite films treated with GuSCN which was added into the perovskite precursor solution with a concentration of 1%-5%.



**Fig. S12** The JV curves of PSCs with GuSCN modify the buried interface of perovskite.



**Fig. S13** SEM images of perovskite films on the NiO*<sup>x</sup>* substrates modified with different concentrations of GuSCN.



**Fig. S14** UV–Vis absorption spectra of perovskite films post-treated with different concentrations

of GuSCN.



**Fig. S15** Steady-state PL spectra of perovskite layer on glass with or without GuSCN with back

incidence light.



**Fig. S16** (a) The complete XPS spectra and (b) XPS spectra of C 1s for control and GuSCN-treated perovskite films.



**Fig. S17** EDS images of the control perovskite and perovskite treated with 25 mM GuSCN.



**Fig. S18** The J-V curves of the semitransparent WBG-PSCs.



**Fig. S19** Transmittance spectrum of ITO prepared in glass substrate by sputter.



**Fig. S20** The photos of (a) the semitransparent perovskite solar cells, (b) the semitransparent perovskite solar cells covered with a light mask, (c) the 2.25 cm<sup>2</sup> covered semitransparent perovskite solar cell and silicon-based solar cell.



Fig. S21 Transmittance spectrum of the semi-transparent WBG PSC. The AVT is calculated from 700 nm to 1100 nm.

	$V_{OC}$	$J_{\rm SC}$	FF.	<b>PCE</b>
	(V)	(mA/cm <sup>2</sup> )	(% )	$(\%)$
0 mM GuSCN	1.15	19.92	77.22	17.67
15 mM GuSCN	1.18	20.59	79.81	19.46
25 mM GuSCN	1.20	21.26	82.04	20.92
35 mM GuSCN	1.21	21.16	66.27	16.93

**Table S1.** Photoelectric performance parameters of PSCs in GuSCN with different concentrations.



**Table S2.** Photoelectric performance parameters of PSCs based on WBG perovskites.



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	$V_{OC}$	$J_{SC}$	FF	<b>PCE</b>
	(V)	(mA/cm <sup>2</sup> )	$(\%)$	$(\%)$
Target	$1.212 \pm 0.01$	$20.62 \pm 1.08$	$78.90 \pm 1.67$	19.70±0.87
Control	$1.146 \pm 0.02$	$20.20 \pm 0.31$	$75.30 \pm 1.85$	$17.43 \pm 0.53$

**Table S3.** The average data of 30 PSCs were summarized.

**Table S4.** Photovoltaic performance parameters of PSCs treated with GuSCN. The GuSCN with





	$V_{OC}$	$J_{SC}$	FF	<b>PCE</b>
	(V)	(mA/cm <sup>2</sup> )	$(\%)$	$(\%)$
$0 \text{ mM}$ GuSCN	1.15	19.92	77.22	17.67
15 mM GuSCN	1.10	21.38	73.96	17.44
25 mM GuSCN	1.12	21.08	72.45	17.12
35 mM GuSCN	1.12	21.09	71.80	16.93

**Table S5.** Photoelectric performance parameters of PSCs with GuSCN modifying the buried interface of perovskite.

**Table S6.** Fitted lifetime results of perovskite and passivated perovskite with GuSCN, where τ<sub>1</sub> and

$\tau_2$ refer to fast and slow decay lifetimes respectively, and $\tau_{\text{average}}$ is intensity weighted.
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	$I(\%)$	Br $(\% )$	Pb $(\% )$	S(%)	I/Br
Control	58.17	17.27	24.55	$\theta$	3.37
Target	59.36	15.17	24.87	0.60	3.92

**Table S7.** Atomic ratio of the elements derived from EDS.

**Table S8.** Photovoltaic parameters of the semitransparent WBG-PSCs.

	$V_{OC}$	$J_{SC}$	FF	<b>PCE</b>
	(V)	(mA/cm <sup>2</sup> )	$(\%)$	$(\%)$
<b>ST-PSC RS</b>	1.18	21.90	72.25	18.61
<b>ST-PSC FS</b>	1.17	22.01	71.53	18.46

**Table S9.** 4-T tandem solar cell.

	$V_{OC}$	$J_{SC}$	FF	<b>PCE</b>
	(V)	(mA/cm <sup>2</sup> )	$(\%)$	$(\%)$
<b>PVK</b>	1.18	21.90	72.25	18.61
Original Si	0.69	40.84	73.75	20.60
Filtered Si	0.66	17.30	75.25	8.55
4-T tandem				27.16