1,2,3-Triarylazulenes as Precursors of Azulene-Embedded Polycyclic Aromatic Hydrocarbons

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General

Synthesis: Sensitive reactions were conducted under N₂ using standard Schlenk techniques. Glassware was dried at 120 °C. Solvents for synthesis were treated as follows: 1,4-Dioxane (POCH): distilled over Na/benzophenone, MeNO₂ (POCH) dried over molecular sieves, THF (POCH) purified using mBraun solvent purification system, 1,2-dichloroethane (Acros, 99.8%, Extra Dry) and CH₂Cl₂ (ChemPur) were used without further purification.

1-Bromo-4-(*tert*-butyl)benzene (Sigma Aldrich, 97%), 1-iodo-4-methoxybenzene (Aldrich, 98%), 4-(*tert*-butyl)phenyl)boronic acid (Apollo Scientific, 98%), (4-methoxyphenyl)boronic acid (Apollo Scientific, 97%), *N*-iodosuccinimide (NIS, Alfa Aesar, 97%), B₂(pin)₂ (Alfa Aesar, 98%), FeCl₃ (POCH, pure for analysis), K₂CO₃ (Aldrich, pure for synthesis), Cs₂CO₃ (Sigma Aldrich, 99%) were used as received. Catalysts were purified by recrystallization under N₂: Pd(dppf)Cl₂·CH₂Cl₂ (Aldrich) from CH₂Cl₂/*n*-hexane (1/1) mixture, [Ir(OMe)cod]₂ (Aldrich, 98%) from CH₂Cl₂/MeOH (1/1) mixture.

Following compound were prepared according to the literature methods: 6-*tert*butylazulene,¹ 2-bromo-7-(*tert*-butyl)pyrene,² 2-(7-(*tert*-butyl)pyren-2-yl)-4,4,5,5tetramethyl-1,3,2-dioxaborolane.²

Chromatography: *Column chromatography*: standard glass columns and silica gel 60 (0.040-0.063 mm or 0.063-0.200 mm, Merck Millipore). Solvents for column chromatography were used without additional purification.

SEC (Size exclusion chromatography): Glass columns, BioBeads SX-1 stationary phase and CH_2CI_2 as an eluent.

GPC (Gel permeation chromatography): JAI LaboACE LC-7080 recycling preparative system with JAIGEL-2HR ($20\phi x600$) and JAIGEL-2.5HR ($20\phi x600$) column using CHCl₃ stabilized with EtOH as an eluent.

NMR Spectroscopy: ¹H and ¹³C NMR spectra were recorded with a Bruker Avance 500 MHz, Bruker Avance 600 MHz or JOEL 500 MHz spectrometers. For all the ¹H NMR spectra, the chemical shifts are given in ppm relative to the solvent residual peaks (CDCl₃, ¹H: 7.26 ppm, ¹³C: 77.16 ppm; CD₂Cl₂, ¹H: 5.32 ppm, ¹³C: 53.84 ppm). Coupling constants are given in Hz.

High Resolution Mass Spectrometry: HRMS spectra were recorded as follows: *ESI-HRMS*: Bruker qTOF compact or Shimadzu q-TOF LCMS 9030, *APCI-HRMS*: Bruker qTOF compact.

UV/Vis/NIR Spectroscopy: All measurements of CH_2Cl_2 solutions ($c^{-10^{-5}}$ M) were carried out in 10 mm quartz cuvettes using two-beam JASCO V-770 spectrophotometer.

Electrochemistry: Cyclic voltammetry (CV) and differential pulse voltammetry (DPV) measurements were carried out in CH_2Cl_2 (dried and degassed from mBraun solvent purification system) using a Metrohm Autolab/PGSTAT302N potentiostat / galvanostat in a glass cell under a N_2 atmosphere at room temperature. A platinum disk working electrode, a platinum wire auxiliary electrode and an Ag/AgCl reference electrode were used for all measurements. Tetrabutylammonium hexafluorophosphate (AmBeed) was used as a supporting electrolyte (0.1 M solution). CV: scan rate was 0.050 V s⁻¹. DPV: step size of 0.005

V, a modulation amplitude of 0.01 V, a modulation time of 0.05 s and an interval time of 0.5 s. All potentials were calibrated to the ferrocenium/ferrocene (Fc^+/Fc) redox couple.

X-ray single crystal diffraction: A suitable crystals were selected and measured using Rigaku XtaLAB Synergy-R diffractometer. The crystals were kept at 100 K during data collection. The structures were solved with the olex2.solve³ or SHELXS⁴ structure solution programs and refined with the SHELXL⁵ refinement package using least squares minimization.⁵

Theoretical methods: Gaussian 16 software was used for density functional theory (DFT) and time-dependent density functional theory (TD-DFT) calculations.⁶ B3LYP functional^{7,8} and 6-31g(d) basis set were applied for neutral structure optimization, solvent model SCRF=(Solvent=Dichloromethane) was applied. Optimized ground-state geometries were examined by frequency analysis to possess no negative frequency. B3LYP functionals and 6-31g(d) basis sets were applied for TD-DFT calculations of UV/Vis/NIR spectra and NICS(0). UB3LYP/6-31G(d) level of theory was used for calculations of spin density of possible intermediates. Multiwfn software was used for analysis of the electronic transitions.⁹

Synthesis

Compound 5, 2-(6-(tert-butyl)azulen-2-yl)-4,4,5,5-tetramethyl-1,3,2-dioxaborolane



6-*tert*-butylazulene (1.000 g, 5.43 mmol), 4,4'-dimethyl-2,2'bipirydyl (50 mg, 0.27 mmol), $B_2(pin)_2$ (758 mg, 2.98 mmol) and $[Ir(MeO)(COD)]_2$ (72 mg, 0.11 mmol) were placed in a Schlenk flask and purged with N₂. Next moisture- and oxygen-free THF was added, and the reaction mixture was stirred under N₂ for 24 h at 80 °C. After given time the solvent was removed under reduced pressure and the crude product was purified using silica gel chromatography (hexane/CH₂Cl₂; *v/v*; from 1/0 to 1/1) yielding 1.11 g (3.58 mmol) of product as a deep blue solid. Yield: 66%

¹H NMR (500 MHz, CDCl₃) δ 8.29 (d, *J* = 10.8 Hz, 2H), 7.67 (s, 2H), 7.29 (d, *J* = 10.7 Hz, 2H), 1.45 (s, 9H), 1.40 (s, 12H).

 ^{13}C NMR (126 MHz, CDCl_3) δ 162.9, 139.6, 137.4, 124.7, 121.0, 83.7, 38.9, 32.0, 25.1 (C-B signal not visible).

ESI-HRMS (positive mode) *m*/*z* calcd for C₂₀H₂₇BKO₂⁺ (M+K⁺): 348.1772; found: 348.1793.

Compound 6a, 6-(tert-butyl)-2-(4-(tert-butyl)phenyl)azulene

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2-(6-(*tert*-butyl)azulen-2-yl)-4,4,5,5-tetramethyl-1,3,2-dioxaborolane **5** (350 mg, 1.13 mmol), Pd(dppf)Cl₂·CH₂Cl₂ (20.6 mg, 0.028 mmol), 1-bromo-4-(*tert*-butyl)benzene (288 mg, 1.35 mmol), Cs₂CO₃ (735 mg, 2.26 mmol) were placed in a Schlenk flask and dissolved in 10 mL of 1,4-dioxane and 1 mL of H₂O. Rection mixture was degassed and stirred at 80 °C for 18 h. Solvents were evaporated under reduced pressure and crude mixture was purified using silica gel column (hexane to hexane/CH₂Cl₂, v/v, 1/1) yielding blue solid (235 mg, 0.742 mmol), yield: 66%

¹H NMR (500 MHz, CDCl₃) δ 8.22 (d, J = 10.8 Hz, 2H), 7.90 (d, J = 8.2 Hz, 2H), 7.59 (s, 2H), 7.49 (d, J = 8.2 Hz, 2H), 7.34 (d, J = 10.8 Hz, 2H), 1.46 (s, 9H), 1.38 (s, 9H).

 ^{13}C NMR (126 MHz, CDCl_3) δ 159.9, 151.2, 149.3, 140.2, 134.9, 134.1, 127.3, 126.0, 121.9, 113.9, 38.6, 34.8, 32.1, 31.5.

ESI-HRMS (positive mode) m/z calcd for C₂₄H₂₉⁺ (M+H⁺): 317.2264; found: 317.2279.

Compound 6b, 6-(tert-butyl)-2-(4-methoxyphenyl)azulene



2-(6-(*tert*-butyl)azulen-2-yl)-4,4,5,5-tetramethyl-1,3,2-dioxaborolane **5** (300 mg, 0.967 mmol), 1-iodo-4-methoxybenzene (339 mg, 1.45 mmol), $Pd(dppf)Cl_2 \cdot CH_2Cl_2$ (21.2 mg, 0.029 mmol), Cs_2CO_3 (630 mg, 1.93 mmol), 20 mL of 1,4-dioxane and 2 mL of water were used according to the procedure for **6a**. Reaction was carried out at 80 °C for 21 h, purification: silica gel column (hexane to CH_2Cl_2). Blue solid (70 mg, 0.241 mmol), yield: 25%. ¹H NMR (500 MHz, $CDCl_3$) δ 8.19 (d, J = 10.9 Hz, 2H), 7.92 – 7.88 (m, 2H), 7.53 (s, 2H), 7.33 (d, J = 10.9 Hz, 2H), 7.02 – 6.97 (m, 2H), 3.87 (s, 3H), 1.45 (s, 9H).

 ^{13}C NMR (126 MHz, CDCl_3) δ 159.9, 159.5, 149.1, 140.3, 134.5, 129.7, 128.8, 122.0, 114.5, 113.4, 55.5, 38.6, 32.1.

APCI-HRMS (positive mode) *m*/*z* calcd for C₂₁H₂₃O⁺ (M+H⁺): 291.1743; found: 291.1809.

Compound 6c, 2-(tert-butyl)-7-(6-(tert-butyl)azulen-2-yl)pyrene



2-(6-(*tert*-butyl)azulen-2-yl)-4,4,5,5-tetramethyl-1,3,2-dioxaborolane **5** (227 mg, 0.732 mmol), 2-bromo-7-(*tert*-butyl)pyrene (271 mg, 0.804 mmol), Pd(dppf)Cl₂·CH₂Cl₂ (16.1 mg, 0.022 mmol), Cs₂CO₃ (477 mg, 1.46 mmol), 10 mL of 1,4-dioxane and 1 mL of water were used according to the procedure for **6a**. Reaction was carried out at 80 °C for 18 h, purification: silica gel column (hexane to CH_2Cl_2) and then solid was washed with hexane. Green solid (146 mg, 0.331 mmol), yield: 45%.

¹H NMR (500 MHz, $CDCl_3$) δ 8.72 (s, 2H), 8.32 (d, *J* = 10.8 Hz, 2H), 8.21 (s, 2H), 8.12 (d, *J* = 8.9 Hz, 2H), 8.06 (d, *J* = 8.9 Hz, 2H), 7.90 (s, 2H), 7.39 (d, *J* = 10.8 Hz, 2H), 1.60 (s, 9H), 1.49 (s, 9H).

¹³C NMR (126 MHz, CDCl₃) δ 160.5, 149.6, 149.3, 140.5, 135.4, 134.2, 131.6, 131.3, 128.1, 127.7, 124.6, 124.0, 123.1, 122.5, 122.2, 114.6, 38.7, 35.4, 32.1, 32.1. APCI-HRMS (positive mode) m/z calcd for C₂₄H₃₃⁺ (M+H⁺): 441.2577; found: 441.2641.

Compound 7a, 6-(tert-butyl)-2-(4-(tert-butyl)phenyl)-1,3-diiodoazulene



Azulene **6a** (100 mg, 0.316 mmol) was dissolved in 50 mL of CH_2Cl_2 and the reaction mixture was cooled down to 0 °C. NIS (149 mg, 0.664 mmol) was added and the blue solution was stirred at room temperature for 3 h. Solvent was evaporated under reduced pressure and the crude product was purified using silica gel column (hexane/CH₂Cl₂, v/v, 1/1) yielding **7a** as a green solid (176 mg, 0.310 mmol), yield: 98%

¹H NMR (500 MHz, CDCl₃) δ 8.23 (d, J = 11.2 Hz, 2H), 7.57 (d, J = 11.2 Hz, 2H), 7.54 – 7.50 (m, 2H), 7.47 – 7.44 (m, 2H), 1.48 (s, 9H), 1.42 (s, 9H).

 ^{13}C NMR (126 MHz, CDCl_3) δ 162.8, 154.1, 151.2, 141.0, 138.4, 135.3, 130.4, 124.8, 123.8, 77.4, 39.0, 34.9, 31.9, 31.6.

ESI-HRMS (positive mode) m/z calcd for $C_{24}H_{26}I_2^+$ (M⁺): 568.0118; found: 568.0133.

Compound 7b, 6-(tert-butyl)-1,3-diiodo-2-(4-methoxyphenyl)azulene



Azulene **6b** (70 mg, 0.241 mmol), NIS (136 mg, 0.602 mmol) and 50 mL of CH_2Cl_2 were used according to the procedure for **7a**. Reaction time: 1 h, purification: silica gel column (petroleum ether/ CH_2Cl_2 , v/v, 1/1), green solid, yield: 97% (127 mg, 0.234 mmol).

¹H NMR (500 MHz, CDCl₃) δ 8.19 (d, J = 10.9 Hz, 2H), 7.55 (d, J = 10.9 Hz, 2H), 7.46 – 7.42 (m, 2H), 7.05 – 7.02 (m, 2H), 3.89 (s, 3H), 1.46 (s, 9H).

 ^{13}C NMR (126 MHz, CDCl_3) δ 162.7, 159.7, 153.9, 141.0, 138.3, 132.1, 130.8, 123.8, 113.4, 77.5, 55.4, 39.0, 31.9.

APCI-HRMS (positive mode) m/z calcd for $C_{21}H_{21}I_2O^+$ (M⁺): 542.9676; found: 542.9698.

Compound 7c, 2-(tert-butyl)-7-(6-(tert-butyl)-1,3-diiodoazulen-2-yl)pyrene



Azulene **6c** (110 mg, 0.250 mmol), NIS (124 mg, 0.602 mmol) and 50 mL of CH_2Cl_2 were used according to the procedure for **7a**. Reaction time: 1 h, purification: silica gel column (hexane/CH₂Cl₂, v/v, 1/1), green solid, yield: 84% (146 mg, 0.211 mmol).

¹H NMR (500 MHz, $CDCl_3$) δ 8.32 (d, J = 11.1 Hz, 2H), 8.26 (s, 2H), 8.25 (s, 2H), 8.14 (d, J = 8.9 Hz, 2H), 8.11 (d, J = 8.9 Hz, 2H), 7.64 (d, J = 11.1 Hz, 2H), 1.61 (s, 9H), 1.52 (s, 9H).

 ^{13}C NMR (126 MHz, CDCl_3) δ 163.2, 154.6, 149.5, 141.1, 138.6, 135.8, 131.3, 130.6, 128.0, 127.6, 127.1, 124.4, 124.0, 123.0, 122.5, 77.9, 39.1, 35.4, 32.1, 32.0.

APCI-HRMS (positive mode) m/z calcd for $C_{34}H_{31}I_2^+$ (M+H⁺): 693.0510; found: 693.0499.

Compound 1a, 6-(tert-butyl)-1,2,3-tris(4-(tert-butyl)phenyl)azulene



Diiodoazulene **7a** (100 mg, 0.176 mmol (4-(*tert*-butyl)phenyl)boronic acid (94 mg, 0.528 mmol), $Pd(dppf)Cl_2 \cdot CH_2Cl_2$ (6.4 mg, 0.0087 mmol), Cs_2CO_3 (172 mg, 0.528 mmol) were placed in a Schlenk flask and dissolved in 10 mL of 1,4-dioxane and 1 mL of H₂O. Rection mixture was

degassed and stirred at 80°C for 2 h. Solvents were evaporated under reduced pressure and crude mixture was purified using silica gel column (hexane to hexane/ CH_2Cl_2 , v/v, 1/1) yielding green solid (80 mg, 0.138 mmol), yield: 78%.

¹H NMR (500 MHz, CDCl₃) δ 8.26 (d, *J* = 11.0 Hz, 2H), 7.35 – 7.31 (m, 4H), 7.23 (d, *J* = 11.1 Hz, 2H), 7.21 – 7.18 (m, 4H), 7.10 – 7.06 (m, 2H), 6.97 – 6.94 (m, 2H), 1.41 (s, 9H), 1.35 (s, 18H), 1.25 (s, 9H).

 ^{13}C NMR (126 MHz, CDCl_3) δ 161.4, 149.1, 148.8, 147.3, 136.2, 134.8, 133.9, 133.7, 131.3, 131.1, 128.8, 124.8, 124.3, 121.6, 38.5, 34.6, 34.5, 32.0, 31.6, 31.5.

APCI-HRMS (positive mode) m/z calcd for C₄₄H₅₃⁺⁺ (M⁺⁺): 581.4142; found: 581.4092.

Compound 1b, 6-(tert-butyl)-2-(4-(tert-butyl)phenyl)-1,3-bis(4-methoxyphenyl)azulene



Diiodoazulene **7a** (100 mg, 0.176 mmol), (4-methoxyphenyl)boronic acid (80 mg, 0.528 mmol), $Pd(dppf)Cl_2 \cdot CH_2Cl_2$ (6.4 mg, 0.0087 mmol), Cs_2CO_3 (172 mg, 0.528 mmol), 20 mL of 1,4-dioxane and 2 mL of water were used according to the procedure for **1a**. Reaction carried out at 80 °C for 18 h, purification: silica gel column (hexane to hexane/CH₂Cl₂, v/v, 1/1). Green solid (67 mg, 0.127 mmol), yield: 72%.

¹H NMR (500 MHz, CDCl₃) δ 8.22 (d, J = 11.0 Hz, 2H), 7.23 (d, J = 11.0 Hz, 2H), 7.22 – 7.18 (m, 4H), 7.14 – 7.10 (m, 2H), 6.98 – 6.95 (m, 2H), 6.91 – 6.87 (m, 4H), 3.85 (s, 6H), 1.42 (s, 9H), 1.26 (s, 9H).

 ^{13}C NMR (126 MHz, CDCl_3) δ 161.5, 158.1, 149.2, 147.1, 136.2, 134.6, 133.8, 132.7, 131.0, 129.2, 128.3, 124.6, 121.5, 113.6, 55.4, 38.5, 34.6, 32.0, 31.5.

APCI-HRMS (positive mode) m/z calcd for $C_{38}H_{41}O_2^+$ (M+H⁺): 529.3101; found: 529.3061.

Compound 1c, 6-(*tert*-butyl)-1,2,3-tris(4-methoxyphenyl)azulene



Diiodoazulene **7b** (50 mg, 0.092 mmol, 1.0 equiv.), boronic acid (42 mg, 0.277 mmol), $Pd(dppf)Cl_2 \cdot CH_2Cl_2$ (2.0 mg, 0.0028 mmol), Cs_2CO_3 (90 mg, 0.277 mmol), 5 mL of 1,4-dioxane

and 1 mL of water were used according to the procedure for **1a**. Reaction carried out at 80 °C for 18 h, purification: silica gel column (CH_2Cl_2). Green solid (43 mg, 085mmol), yield: 92%.

¹H NMR (500 MHz, CDCl₃) δ 8.21 (d, J = 11.0 Hz, 2H), 7.23 (d, J = 11.0 Hz, 2H), 7.22 – 7.19 (m, 4H), 6.99 – 6.95 (m, 2H), 6.92 – 6.89 (m, 4H), 6.69 – 6.65 (m, 2H), 3.85 (s, 6H), 3.75 (s, 3H), 1.42 (s, 9H).

 ^{13}C NMR (126 MHz, CDCl_3) δ 161.4, 158.3, 158.2, 146.7, 136.2, 134.4, 132.7, 132.6, 129.3, 129.2, 128.1, 121.6, 113.7, 113.3, 55.4, 55.2, 38.5, 32.0.

APCI-HRMS (positive mode) m/z calcd for $C_{35}H_{35}O_3^+$ (M+H⁺): 503.2581; found: 503.2547.

Compound 1d, 2-(*tert*-butyl)-7-(6-(*tert*-butyl)-1,3-bis(4-(*tert*-butyl)phenyl)azulen-2-yl)pyrene



Diiodoazulene **7c** (90 mg, 0.130 mmol), 4-*tert*-butylphenylboronic acid (69 mg, 0.390 mmol), $Pd(dppf)Cl_2 \cdot CH_2Cl_2$ (2.9 mg, 0.0039 mmol), Cs_2CO_3 (85 mg, 0.260 mmol), 10 mL of 1,4-dioxane and 1 mL of water were used according to the procedure for **1a**. Reaction at 80 °C for 20 h, purification: silica gel column (hexane/CH₂Cl₂, v/v, 10/1), then GPC (CHCl₃). Green solid (44 mg, 0.062 mmol), yield: 48%.

¹H NMR (500 MHz, CDCl₃) δ 8.40 (d, *J* = 11.0 Hz, 2H), 8.13 (s, 2H), 7.90 (d, *J* = 9.0 Hz, 2H), 7.78 (s, 2H), 7.68 (d, *J* = 9.0 Hz, 2H), 7.31 (d, *J* = 11.0 Hz, 2H), 7.25 (d, J=8.6 Hz, 4H), 7.21 (d, *J* = 8.6 Hz, 4H), 1.55 (s, 9H), 1.46 (s, 9H), 1.29 (s, 18H).

¹³C NMR (126 MHz, CDCl₃) δ 162.0, 149.1, 149.0, 147.3, 136.3, 135.2, 134.5, 133.5, 131.4, 131.2, 130.4, 129.3, 128.4, 127.7, 127.1, 125.0, 123.4, 122.9, 122.0, 121.8, 38.6, 35.3, 34.6, 32.1, 32.0, 31.5.

APCI-HRMS (positive mode) m/z calcd for $C_{54}H_{57}^+$ (M+H⁺): 705.4455; found: 705.4449.

Compound 1e, 7,7'-(6-(*tert*-butyl)-2-(4-(*tert*-butyl)phenyl)azulene-1,3-diyl)bis(2-(*tert*-butyl)pyrene)



Diiodoazulene **7a** (100 mg, 0.176 mmol), 2-(7-(*tert*-butyl)pyren-2-yl)-4,4,5,5-tetramethyl-1,3,2-dioxaborolane (203 mg, 0.528 mmol), Pd(dppf)Cl₂·CH₂Cl₂ (3.9 mg, 0.0053 mmol), Cs₂CO₃ (115 mg, 0.352 mmol), 10 mL of 1,4-dioxane and 1 mL of water were used according to the procedure for **1a.** Reaction at 80°C for 24 h, purification: silica gel column (hexane/CH₂Cl₂, v/v, 10/1), then SEC (BioBeads SX-1, CH₂Cl₂). Blue solid (59 mg, 0.071 mmol), yield: 40%.

¹H NMR (500 MHz, CDCl₃) δ 8.46 (d, J = 11.1 Hz, 2H), 8.23 (s, 4H), 8.15 (s, 4H), 8.04 (d, J = 8.9 Hz, 4H), 7.97 (d, J = 8.9 Hz, 4H), 7.36 (d, J = 11.1 Hz, 2H), 6.96 (d, J = 8.5 Hz, 2H), 6.87 (d, J = 8.6 Hz, 2H), 1.61 (s, 18H), 1.46 (s, 9H), 1.09 (s, 9H).

¹³C NMR (126 MHz, CDCl₃, 300 K) δ 162.0, 149.4, 149.0, 148.4, 137.0, 135.0, 134.4, 133.4, 131.3, 131.2, 131.0, 129.0, 128.3, 127.6, 127.6, 124.6, 123.4, 123.1, 122.5, 122.3, 38.7, 35.4, 34.5, 32.1, 32.0, 31.2.

APCI-HRMS (positive mode) m/z calcd for $C_{64}H_{60}^+$ (M+H⁺): 828.4689; found: 828.4634.

Compound 1f, 7,7',7"-(6-(tert-butyl)azulene-1,2,3-triyl)tris(2-(tert-butyl)pyrene)



Diiodoazulene **7c** (100 mg, 0.144 mmol), 2-(7-(*tert*-butyl)pyren-2-yl)-4,4,5,5-tetramethyl-1,3,2-dioxaborolane (166 mg, 0.433 mmol), Pd(dppf)Cl₂·CH₂Cl₂ (3.2 mg, 0.0043 mmol), Cs₂CO₃ (141 mg, 0.433 mmol), 10 mL of 1,4-dioxane and 1 mL of water were used according to the procedure for **1a**. Reaction at 80 °C for 20 h, purification: silica gel column (hexane to hexane/CH₂Cl₂, v/v, 1/1), then SEC (BioBeads SX-1, CH₂Cl₂). Green solid (63 mg, 0.066 mmol), yield: 46%.

¹H NMR (500 MHz, CDCl₃) δ 8.56 (d, *J* = 11.0 Hz, 2H), 8.18 (s, 4H), 8.17 (s, 4H), 8.01 (s, 2H), 7.95 (d, *J* = 9.0 Hz, 4H), 7.88 (s, 2H), 7.84 (d, *J* = 9.0 Hz, 4H), 7.67 (d, *J* = 9.0 Hz, 2H), 7.41 (d, *J* = 11.0 Hz, 2H), 7.37 (d, *J* = 9.0 Hz, 2H), 1.58 (s, 18H), 1.49 (s, 9H), 1.48 (s, 9H).

 ^{13}C NMR (126 MHz, CDCl_3) δ 162.5, 149.0, 149.0, 147.9, 137.3, 135.4, 134.2, 133.9, 131.1, 131.0, 130.5, 129.5, 128.5, 128.3, 127.7, 127.6, 127.6, 127.1, 123.4, 123.4, 123.1, 122.7, 122.7,

122.3, 122.0, 38.8, 35.4, 35.3, 32.1, 32.0, 32.0. (one aromatic signal missing due to overlapping).

APCI-HRMS (positive mode) m/z calcd for $C_{73}H_{63}^+$ (M+H⁺): 939.4924; found: 939.5025.

Compound 3a, 3,6,11-tri-*tert*-butyl-13*a*-(4-(*tert*-butyl)phenyl)azuleno[1,2-/]phenanthren-14(13*a*H)-one



Starting azulene **1a** (10 mg, 0.017 mmol) and 48 mg of K_2CO_3 (0.344 mmol) were placed in a screw-sealed vial and then 1 mL of dry 1,2-dichloroethane was added. The mixture was bubbled with N_2 and then 22 mg (0.138 mmol) of dry FeCl₃ in 0.4 mL of dry nitromethane was added to the reaction mixture. Vial was tightly closed and reaction mixture was stirred for 16 h at 80 °C. Solvent was removed under reduced pressure and mixture was passed through short silica column (hexane) yielding product as yellow solid (9.7 mg, 0.016 mmol). Yield: 93%

¹H NMR (500 MHz, $CDCl_3$) δ 9.16 (d, J = 8.7 Hz, 1H), 8.77 (d, J = 11.0 Hz, 1H), 8.77 (s, 1H), 8.63 (d, J = 1.7 Hz, 1H), 7.80 (dd, J = 8.7, 2.0 Hz, 1H), 7.72 (dd, J = 8.7, 1.8 Hz, 1H), 7.55 (d, J = 7.0 Hz, 1H), 7.15 – 7.11 (m, 2H), 7.09 – 7.06 (m, 2H), 6.63 (d, J = 10.1 Hz, 1H), 6.46 (d, J = 10.1 Hz, 1H), 6.40 (d, J = 6.9 Hz, 1H), 1.55 (s, 9H), 1.49 (s, 9H), 1.19 (s, 9H), 0.99 (s, 9H).

 ^{13}C NMR (126 MHz, CDCl₃) δ 204.8, 154.5, 152.7, 150.8, 149.4, 148.0, 137.4, 135.7, 132.0, 131.2, 128.7, 128.3, 127.1, 127.0, 126.7, 126.0, 125.9, 125.8, 125.7, 124.2, 123.8, 122.8, 119.7, 118.1, 60.7, 36.3, 35.6, 35.5, 34.4, 31.5, 31.4, 30.1, 29.8.

APCI-HRMS (positive mode) m/z calcd for $C_{44}H_{51}O^+$ (M+H⁺): 595.3934; found: 595.3974.

Compound 3b, 3,11-di-*tert*-butyl-6-methoxy-13*a*-(4-methoxyphenyl)azuleno[1,2-/]phenanthren-14(13*a*H)-one



Starting azulene **1b** (25 mg, 0.047 mmol) and 131 mg of K_2CO_3 (0.946 mmol) were placed in a Schlenk flask and then 5 mL of dry 1,2-dichloroethane was added. The mixture was bubbled with N₂ and then 123 mg (0.757 mmol) of dry FeCl₃ in 1.25 mL of dry nitromethane was added to the reaction mixture. Reaction mixture was stirred at 80 °C for 96 h. Solvent was removed under reduced pressure and mixture was purified using short silica gel column (eluent gradually changed from hexane to CH₂Cl₂) and washed with methanol yielding product as a yellow solid (15 mg, 0.028 mmol). Yield: 58%

¹H NMR (500 MHz, CDCl₃) δ 9.16 (d, *J* = 8.7 Hz, 1H), 8.79 (d, *J* = 9.2 Hz, 1H), 8.51 (d, *J* = 1.7 Hz, 1H), 8.16 (d, *J* = 2.6 Hz, 1H), 7.73 (dd, *J* = 8.7, 1.8 Hz, 1H), 7.50 (d, *J* = 7.0 Hz, 1H), 7.34 (dd, *J* = 9.1, 2.6 Hz, 1H), 7.13 (d, *J* = 8.9 Hz, 2H), 6.63 (dd, *J* = 10.6, 8.7 Hz, 3H), 6.44 – 6.39 (m, 2H), 4.09 (s, 3H), 3.69 (s, 3H), 1.48 (s, 9H), 1.03 (s, 9H).

 ^{13}C NMR (126 MHz, CDCl_3) δ 204.5, 160.7, 158.4, 156.1, 154.6, 150.8, 148.1, 137.3, 137.2, 131.9, 131.1, 130.5, 129.0, 128.8, 128.4, 127.1, 127.0, 126.2, 125.9, 123.6, 122.8, 122.5, 118.3, 116.1, 112.9, 106.6, 55.7, 55.3, 36.4, 35.5, 31.5, 30.1.

APCI-HRMS (positive mode) m/z calcd for $C_{38}H_{39}O_{3}^+$ (M+H⁺): 543.2894; found: 543.2931.

Compound 3c, 11-(*tert*-butyl)-3,6-dimethoxy-13*a*-(4-methoxyphenyl)azuleno[1,2-/]phenanthren-14(13*aH*)-one



Starting azulene **1c** (10 mg, 0.020 mmol) and 44 mg of K_2CO_3 (0.318 mmol) were placed in a Schlenk flask and then 2 mL of dry 1,2-dichloroethane was added. The mixture was bubbled with N₂ and then 65 mg (0.398 mmol) of dry FeCl₃ in 0.5 mL of dry nitromethane was added to the reaction mixture. Reaction mixture was stirred at 80 °C for 48 h. Solvent was removed under reduced pressure and mixture was passed through short silica column (CH₂Cl₂/hexane, v/v, 5/2) yielding product as a yellow solid (3 mg, 0.0058 mmol). Yield: 29%

¹H NMR (500 MHz, CDCl₃) δ 7.28 (d, J = 8.8 Hz, 2H), 7.20 (d, J = 8.9 Hz, 2H), 7.17 (d, J = 9.0 Hz, 2H), 6.95 (d, J = 8.8 Hz, 2H), 6.72 (d, J = 4.7 Hz, 2H), 6.70 (d, J = 4.6 Hz, 2H), 6.53 (d, J = 10.3 Hz, 1H), 6.38 (d, J = 10.4 Hz, 1H), 6.34 (d, J = 6.9 Hz, 1H), 6.20 (d, J = 6.9 Hz, 1H), 3.86 (s, 3H), 3.75 (s, 3H), 3.74 (s, 3H), 0.96 (s, 9H).

¹³C NMR (126 MHz, CDCl₃) δ 204.9, 162.5, 160.2, 159.3, 158.5, 154.5, 141.7, 135.1, 132.4, 131.5, 131.3, 130.9, 130.5, 128.9, 128.1, 126.2, 123.9, 122.6, 122.0, 114.3, 113.6, 113.1, 55.5, 55.4, 55.3, 36.5, 30.0.

APCI-HRMS (positive mode) m/z calcd for $C_{35}H_{35}O_4^+$ (M+H⁺): 519.2523; found: 519.2523.

Compound 4d: 4,9,14-tri-*tert*-butyl-17-(4-(*tert*-butyl)phenyl)azuleno[2,1*k*]dibenzo[*m*,*pqr*]tetraphene and compound 2d: 2,7,12,17-tetra-*tert*-butylazuleno[1,2,3*cd*]tribenzo[*a*,*f*,*lm*]perylene



Starting azulene **1d** (10 mg, 0.014 mmol) and 31 mg of K_2CO_3 (0.227 mmol) were placed in a Schlenk flask and then 2 mL of dry 1,2-dichloroethane was added. The mixture was bubbled with N₂ and then 18 mg (0.113 mmol) of dry FeCl₃ in 0.4 mL of dry nitromethane was added to the reaction mixture. Reaction mixture was stirred for 24 h at room temperature. Solvent was removed under reduced pressure and mixture was passed through short silica column (hexane/CH₂Cl₂, *v*/*v*, 7/3) and purified using GPC (CHCl₃) yielding product **4d** as a green solid (3.0 mg, 0.0043 mmol, yield: 30%) and traces of **2d**. Extension of the reaction time to 7 d followed by the same workup gave **4d** as a green solid (3.0 mg, 0.0043 mmol, yield: 30%) and **2d** as a red solid (3.5 mg, 0.0050 mmol, yield: 36%).

Compound 4d: 4,9,14-tri-tert-butyl-17-(4-(tert-butyl)phenyl)azuleno[2,1k]dibenzo[m,pqr]tetraphene:

¹H NMR (500 MHz, CDCl₃) δ 9.38 (d, J = 10.2 Hz, 1H), 9.20 (d, J = 9.3 Hz, 1H), 9.02 (d, J = 2.0 Hz, 1H), 8.92 (d, J = 8.6 Hz, 1H), 8.53 (s, 1H), 8.24 (d, J = 1.7 Hz, 1H), 8.23 (d, J = 2.7 Hz, 1H), 8.21 (s, 1H), 8.16 (d, J = 1.7 Hz, 1H), 7.90 (d, J = 9.0 Hz, 1H), 7.83 (dd, J = 8.5, 2.0 Hz, 1H), 7.70 (d, J = 8.3 Hz, 2H), 7.60 – 7.51 (m, 4H), 7.38 (dd, J = 11.1, 1.9 Hz, 1H), 1.60 (s, 9H), 1.58 (s, 9H), 1.54 (s, 9H), 1.51 (s, 9H).

 13 C NMR (126 MHz, CDCl₃) δ 160.8, 150.7, 149.4, 146.3, 139.7, 137.7, 135.9, 134.6, 134.5, 131.8, 131.3, 130.9, 130.6, 129.1, 129.0, 128.8, 128.5, 128.3, 127.9, 127.6, 127.3, 127.2, 127.0, 126.6, 126.0, 125.3, 125.3, 124.9, 123.9, 123.4, 122.8, 122.6, 122.5, 121.7, 121.3, 120.9, 38.6, 35.4, 35.2, 35.1, 32.1, 32.0, 31.8, 31.7.

APCI-HRMS (positive mode) m/z calcd for $C_{54}H_{55}^+$ (M+H⁺): 703.4298; found: 703.4299.

Compound 2d: 2,7,12,17-tetra-tert-butylazuleno[1,2,3-cd]tribenzo[a,f,lm]perylene:

¹H NMR (500 MHz, CD_2Cl_2) δ 9.85 (d, J = 11.1 Hz, 2H), 9.24 (d, J = 10.6 Hz, 2H), 9.24 (s, 2H), 9.15 (d, J = 8.6 Hz, 2H), 8.40 (s, 2H), 8.33 (d, J = 9.2 Hz, 2H), 7.97 – 7.92 (m, 4H), 1.68 (s, 9H), 1.65 (s, 9H), 1.59 (s, 18H).

¹³C NMR (151 MHz, CD₂Cl₂) δ 162.3, 150.0, 147.4, 138.3, 135.1, 134.9, 131.7, 130.3, 129.8, 128.8, 127.6, 126.9, 126.6, 126.0, 125.8, 125.4, 124.8, 123.9, 122.7, 122.6, 122.0, 118.3, 39.0, 35.6, 35.5, 32.4, 32.3, 32.1.

APCI-HRMS (positive mode) m/z calcd for $C_{54}H_{53}^+$ (M+H⁺): 701.4142; found: 701.4146.

Compound 4e, 2,7,13-tri-*tert*-butyl-10-(7-(*tert*-butyl)pyren-2-yl)azuleno[1,2k]dibenzo[m,pgr]tetraphene



Starting azulene **1e** (35 mg, 0.042 mmol) and 117 mg of K_2CO_3 (0.844 mmol) were placed in a Schlenk flask and then 7 mL of dry 1,2-dichloroethane was added. The mixture was chilled in an ice bath and bubbled with N₂. Then 110 mg (0.675 mmol) of dry FeCl₃ in 1.75 mL of dry nitromethane was added to the reaction mixture. Reaction mixture was stirred for 18 h at room temperature. Solvent was removed under reduced pressure and mixture was passed through short silica column (hexane/CH₂Cl₂, v/v, 1/1) and purified using GPC (CHCl₃) yielding product as green solid (11 mg, 0.013 mmol). Yield: 32%

¹H NMR (500 MHz, CDCl₃) δ 9.72 – 9.68 (m, 2H), 9.20 (d, *J* = 9.3 Hz, 1H), 9.01 (d, *J* = 1.4 Hz, 1H), 8.46 (s, 2H), 8.35 – 8.25 (m, 7H), 8.18 – 8.13 (m, 5H), 7.74 (d, *J* = 8.6 Hz, 2H), 7.39 (dd, *J* = 11.1, 1.5 Hz, 1H), 7.15 (d, *J* = 8.7 Hz, 1H), 1.65 (s, 9H), 1.64 (s, 9H), 1.53 (s, 9H), 1.37 (s, 9H).

 13 C NMR (126 MHz, CDCl₃) δ 160.9, 149.4, 149.1, 149.1, 140.3, 138.9, 136.3, 135.8, 134.8, 134.4, 131.8, 131.7, 131.5, 131.5, 131.3, 130.6, 130.3, 128.7, 128.2, 128.1, 128.0, 127.8, 127.7, 127.7, 127.1, 126.9, 126.5, 125.3, 124.2, 124.1, 123.5, 123.4, 123.4, 123.2, 122.8, 122.7, 122.3, 122.0, 121.9, 120.5, 120.4, 38.6, 35.5, 35.4, 35.3, 32.2, 32.1, 32.0, 31.5.

APCI-HRMS (positive mode) m/z calcd for $C_{64}H_{59}^+$ (M+H⁺): 827.4611; found: 827.4675.

Compound 2f, 2,9,16,23-tetra-*tert*-butylazuleno[1,2,3-*hi*]dinaphtho[8,1,2-*pqr*:2',1',8'*yza*]pyreno[1,2,3-*uv*]hexacene



Starting azulene **1f** (12 mg, 0.013 mmol) and 35 mg of K_2CO_3 (0.252 mmol) were placed in a screw-sealed vial and then 2 mL of dry 1,2-dichloroethane was added. The mixture was bubbled with N_2 and the 33 mg (0.201 mmol) of dry FeCl₃ in 0.5 mL of dry nitromethane was added to the reaction mixture. Vial was tightly closed, and reaction mixture was stirred for 16

h at 80 °C. Solvent was removed under reduced pressure and mixture was passed through short silica column (hexane/CH₂Cl₂, v/v, 1/1) and purified using preparative TLC (hexane/CH₂Cl₂, v/v, 5/1) yielding product as a brown solid (5.7 mg, 0.006 mmol). Yield: 46%

¹H NMR (500 MHz, CDCl₃) δ 9.93 (d, *J* = 11.1 Hz, 2H), 9.69 (s, 2H), 8.72 (d, *J* = 9.5 Hz, 2H), 8.59 (d, *J* = 9.5 Hz, 2H), 8.39 (d, *J* = 8.9 Hz, 2H), 8.30 (dd, *J* = 9.1, 1.7 Hz, 4H), 8.26 (s, 2H), 8.22 (d, *J* = 8.9 Hz, 2H), 8.06 (d, *J* = 9.4 Hz, 2H), 8.02 – 7.98 (m, 4H), 1.69 (s, 9H), 1.66 (s, 18H), 1.62 (s, 9H).

¹³C NMR (151 MHz, CD₂Cl₂) δ 162.4, 150.1, 149.6, 141.6, 135.5, 135.2, 131.9, 131.7, 131.4, 131.2, 130.8, 130.4, 129.1, 129.0, 128.7, 128.2, 128.1, 127.5, 127.3, 127.1, 125.1, 124.8, 123.9, 123.4, 123.3, 123.2, 123.1, 122.8, 122.7, 121.0, 117.2, 39.0, 35.6, 35.5, 32.2, 32.1, 32.0. APCI-HRMS (positive mode) m/z calcd for C₇₄H₆₁⁺ (M+H⁺): 949.4768; found: 949.4730.

X-ray crystallography

Table S1	Detail of X-ra	v crystallogra	nhv experime	nt of 1a .
Table J1.		y ci ystanogra	рну слренине	ni or 1 a.

Crystal data CCDC 2371234	
Chemical formula	C ₄₄ H ₅₂
M _r	580.85
Crystal system, space group	Monoclinic, P2 ₁ /c
Temperature (K)	100
a, b, c (Å)	11.333 (2), 14.004 (3), 22.087 (5)
β (°)	96.66 (3)
V (Å ³)	3481.7 (13)
Ζ	4
Radiation type	Cu <i>K</i> α
μ (mm⁻¹)	0.46
Crystal size (mm)	$0.15 \times 0.14 \times 0.08$
Data collection	
Diffractometer	XtaLAB Synergy R, DW system, HyPix-Arc 150
Absorption correction	-
No. of measured, independent and observed $[I > 2\sigma(I)]$ reflections	63206, 7210, 6176
R _{int}	0.024
$(\sin \theta / \lambda)_{max} (Å^{-1})$	0.628
Refinement	
$R[F^2 > 2\sigma(F^2)], wR(F^2), S$	0.040, 0.107, 1.03
No. of reflections	7210
No. of parameters	409
H-atom treatment	H-atom parameters constrained
$\Delta \rho_{max}$, $\Delta \rho_{min}$ (e Å ⁻³)	0.19, -0.21

Crystal data CCDC 23712	33
Chemical formula	C ₄₄ H ₅₀ O
<i>M</i> _r	594.84
Crystal system, space group	Orthorhombic, P2 ₁ 2 ₁ 2 ₁
Temperature (K)	100
a, b, c (Å)	10.365 (2), 30.398 (6), 33.663 (6)
V (Å ³)	10606 (3)
Ζ	12
Radiation type	Μο Κα
μ (mm ⁻¹)	0.06
Crystal size (mm)	$0.33 \times 0.25 \times 0.12$
Data collection	
Diffractometer	Xcalibur, Ruby, Gemini ultra
Absorption correction	-
No. of measured, independent and observed [<i>I</i> > 2σ(<i>I</i>)] reflections	37324, 21232, 10374
R _{int}	0.087
(sin θ/λ) _{max} (Å ⁻¹)	0.679
Refinement	
$R[F^2 > 2\sigma(F^2)], wR(F^2), S$	0.091, 0.204, 1.00
No. of reflections	21232
No. of parameters	1252
H-atom treatment	H-atom parameters constrained
$\Delta \rho_{max}$, $\Delta \rho_{min}$ (e Å ⁻³)	0.38, -0.33
Absolute structure	Flack x determined using 2308 quotients [(I+)-(I-)]/[(I+)+(I-)] (Parsons, Flack and Wagner, Acta Cryst. B69 (2013) 249-259).
Absolute structure parameter	-0.9 (10)

Crystal data CCDC 237	1235
Chemical formula	$C_{64}H_{58} \cdot 1.5(C_6H_4Cl_2)$
M _r	1047.59
Crystal system, space group	Triclinic, P1
Temperature (K)	100
a, b, c (Å)	10.287 (2), 15.743 (3), 18.493 (4)
α, β, γ (°)	112.04 (3), 98.31 (3), 90.23 (3)
<i>V</i> (Å ³)	2741.4 (11)
Ζ	2
Radiation type	Cu <i>K</i> α
μ (mm⁻¹)	1.85
Crystal size (mm)	$0.27 \times 0.20 \times 0.12$
Data collection	
Diffractometer	XtaLAB Synergy R, DW system, HyPix-Arc 150
Absorption correction	Multi-scan <i>CrysAlis PRO</i> 1.171.43.105a (Rigaku Oxford Diffraction, 2024) Empirical absorption correction using spherical harmonics, implemented in SCALE3 ABSPACK scaling algorithm.
T _{min} , T _{max}	0.846, 1.000
No. of measured, independent and observed $[I > 2\sigma(I)]$ reflections	68595, 11179, 10382
R _{int}	0.021
$(\sin \theta / \lambda)_{max} (Å^{-1})$	0.629
Refinement	
$R[F^2 > 2\sigma(F^2)], wR(F^2), S$	0.042, 0.117, 1.03
No. of reflections	11179
No. of parameters	739
H-atom treatment	H atoms treated by a mixture of independent and constrained refinement
$\Delta \rho_{max}$, $\Delta \rho_{min}$ (e Å ⁻³)	0.68, -0.53

Table S3. Detail of X-ray crystallography experiment of **4e**.



Figure S1. UV/Vis spectrum of compound 3a (CH₂Cl₂, c^{\sim} 10⁻⁶ M).



Figure S2. UV/Vis spectrum of compound 3b (CH₂Cl₂, c^{\sim} 10⁻⁶ M).



Figure S3. UV/Vis spectrum of compound 3c (CH₂Cl₂, c^{\sim} 10⁻⁶ M).



Figure S4. UV/Vis spectrum of compound 4d (CH₂Cl₂, c^{\sim} 10⁻⁶ M).



Figure S5. UV/Vis spectrum of compound 2d (CH₂Cl₂, c^{\sim} 10⁻⁶ M).



Figure S6. UV/Vis/NIR spectrum of compound 4e (CH₂Cl₂, c^{\sim} 10⁻⁶ M).



Figure S7. UV/Vis/NIR spectrum of compound **2f** (CH₂Cl₂, c^{\sim} 10⁻⁶ M).

Electrochemistry



Figure S8. CV (black) and DPV (red) of compound **3a** ($[NBu_4][PF_6]$ in CH_2Cl_2 (0.1 M) as a supporting electrolyte, 50 mV/s scanning rate).



Figure S9. CV (black) and DPV (red) of compound **3b** ($[NBu_4][PF_6]$ in CH_2Cl_2 (0.1 M) as a supporting electrolyte, 50 mV/s scanning rate).







Figure S11. CV (black) and DPV (red) of compound **4d** ($[NBu_4][PF_6]$ in CH_2Cl_2 (0.1 M) as a supporting electrolyte, 50 mV/s scanning rate).



Figure S12. CV (black) and DPV (red) of compound **2d** ($[NBu_4][PF_6]$ in CH_2Cl_2 (0.1 M) as a supporting electrolyte, 50 mV/s scanning rate).



Figure S13. CV (black) and DPV (red) of compound **4e** ($[NBu_4][PF_6]$ in CH_2Cl_2 (0.1 M) as a supporting electrolyte, 50 mV/s scanning rate).



Figure S14. CV (black) and DPV (red) of compound **2f** ($[NEt_4][PF_6]$ in CH_2Cl_2 (0.1 M) as a supporting electrolyte, 50 mV/s scanning rate).

	3a	3b	3c	2d	4d	4e	2f
E _{HOMO} (DFT) ^a	-5.45 eV	–5.42 eV	–5.30 eV	–4.67 eV	-4.88 eV	–4.95 eV	–4.75 eV
E _{LUMO} (DFT) ^a	–2.28 eV	–2.23 eV	–2.20 eV	–2.18 eV	–2.12 eV	–2.14 eV	–2.25 eV
E _g (DFT)	3.17 eV	3.19 eV	3.10 eV	2.49 eV	2.76 eV	2.81 eV	2.50 eV
E _{HOMO} (EXP) ^b	-5.71 eV	–5.63 eV	-5.83 eV	-4.90 eV	–5.10 eV	–5.15 eV	–5.09 eV
E _{LUMO} (EXP) ^b	-2.84 eV	-2.81 eV	–2.82 eV	–2.99 eV	–2.88 eV	–2.90 eV	-3.20 eV
E _g (EXP)	2.92 eV	2.87 eV	2.82 eV	1.91 eV	2.22 eV	2.25 eV	1.89 eV
E _{ox} ^c	0.95 V	0.91 V	0.83 V	0.10 V	0.30 V	0.35 V	0.29 V
<i>E</i> _{red} ^c	-1.97 V	-1.96 V	-1.99 V	-1.81 V	-1.92 V	-1.90 V	-1.60 V

Table S4. Experimental and DFT-calculated HOMO and LUMO levels.

^{*a*}B3LYP/6-31G(d) level of theory; ^{*b*}Calculated according to the known procedure using the experimentally determined redox potentials ($E_{LUMO} = -(E_{red} + 4.8 \text{ eV})$ and $E_{HOMO} = -(E_{ox} + 4.8 \text{ eV})$) and the energy level of Fc⁺/Fc with respect to the vacuum level (-4.8 eV);¹⁰ ^{*c*}Potentials versus Fc/Fc⁺ in CH₂Cl₂, 25°C, [NBu₄][PF₆] (0.1 M) as a supporting electrolyte.

DFT calculations

Spin densities of radical cations



Figure S15. DFT-calculated spin densities of radical cations (1a)⁻⁺, (1d)⁻⁺, (1e)⁻⁺ and (1f)⁻⁺, UB3LYP/6-31G(d).



Figure S16. DFT-calculated energy difference between two possible isomers of **2f**: chiral (*P*,*P* or *M*,*M*) and *meso* (*P*,*M*) form, B3LYP/6-31G(d).



Figure S17. HOMO and LUMO plots of compound 3a, B3LYP/6-31G(d).



Figure S18. HOMO and LUMO plots of compound 3b, B3LYP/6-31G(d).



Figure S19. HOMO and LUMO plots of compound 3c, B3LYP/6-31G(d).

Figure S20. HOMO and LUMO plots of compound 4d, B3LYP/6-31G(d).

HOMO: -4.67 eV

LUMO: - 2.18 eV

Figure S21. HOMO and LUMO plots of compound 2d, B3LYP/6-31G(d).

Figure S22. HOMO and LUMO plots of compound 4e, B3LYP/6-31G(d).

Figure S23. HOMO and LUMO plots of compound 2f, B3LYP/6-31G(d).

Figure S24. NICS(0) values of compounds 4d, 2d, 4e and 2f, B3LYP/6-31G(d).

Analysis of electronic transitions

Table S5. Electronic transition contributing to UV/Vis/NIR spectrum of compound **4d**, B3LYP/6-31G(d).

Transition	Energy	Wavelength	Oscillator	Contributions
			strength	
$S_0 \rightarrow S_1$	2.01 eV	617 nm	f= 0.02360	H -> L 92.9%, H-1 -> L 5.2%
$S_0 \rightarrow S_2$	2.63 eV	471 nm	f= 0.30900	H -> L+1 90.4%
S₀→S₃	2.69 eV	460 nm	f= 0.01710	H-1 -> L 86.2%
$S_0 \rightarrow S_4$	2.90 eV	427 nm	f= 0.50590	H-2 -> L 68.7%, H -> L+2 26.0%
S₀→S₅	3.14 eV	395 nm	f= 1.29340	H-1 -> L+1 78.3%, H -> L+2 10.8%, H-2 -> L 6.1%
S ₀ →S ₆	3.27 eV	379 nm	f= 0.11610	H-2 -> L+1 83.9%, H -> L+3 6.5%
S ₀ →S ₇	3.31 eV	375 nm	f= 0.64400	H -> L+2 59.1%, H-2 -> L 18.2%, H-1 -> L+1 16.1%
S₀→S ₈	3.59 eV	345 nm	f= 0.56990	H-1 -> L+2 95.1%
S ₀ →S ₉	3.70 eV	335 nm	f= 0.35450	H -> L+3 88.0%, H-2 -> L+1 6.2%
$S_0 \rightarrow S_{10}$	3.72 eV	334 nm	f= 0.11760	H-2 -> L+2 77.5%, H-3 -> L 13.8%

Table S6. Electronic transition contributing to UV/Vis/NIR spectrum of compound **2d**, B3LYP/6-31G(d).

Transition	Energy	Wavelength	Oscillator	Contributions
			strength	
S₀→S₁	1.79 eV	692 nm	f= 0.01780	H -> L 95.1%
S₀→S₂	2.30 eV	540 nm	f= 0.68030	H -> L+1 95.3%
S₀→S₃	2.66 eV	465 nm	f= 0.01810	H-1 -> L 89.8%
$S_0 \rightarrow S_4$	2.74 eV	453 nm	f= 0.06190	H -> L+2 64.1%, H-2 -> L 28.2%
S₀→S₅	2.97 eV	418 nm	f= 1.33470	H-1 -> L+1 60.5%, H-2 -> L 31.9%
$S_0 \rightarrow S_6$	3.06 eV	405 nm	f= 0.20060	H-2 -> L+1 73.2%, H -> L+3 9.6%
S ₀ →S ₇	3.13 eV	396 nm	f= 0.55430	H-1 -> L+1 30.0%, H-2 -> L 29.6%, H -> L+2 23.7%,
				H-2 -> L+1 10.4%
$S_0 \rightarrow S_8$	3.43 eV	361 nm	f= 0.52820	H -> L+3 80.2%, H-2 -> L+1 10.5%
S ₀ →S ₉	3.49 eV	355 nm	f= 0.50040	H-1 -> L+2 96.5%
$S_0 \rightarrow S_{10}$	3.59 eV	345 nm	f= 0.00920	H-2 -> L+2 85.3%

Table S7. Electronic transition contributing to UV/Vis/NIR spectrum of compound **4e**, B3LYP/6-31G(d).

Transition	Energy	Wavelength	Oscillator strength	Contributions
$S_0 \rightarrow S_1$	2.05 eV	605 nm	f= 0.03730	H -> L 95.0%
$S_0 \rightarrow S_2$	2.58 eV	480 nm	f= 0.21780	H-1 -> L 92.3%
S₀→S₃	2.77 eV	448 nm	f= 0.04220	H -> L+1 87.7%
S ₀ →S ₄	2.86 eV	434 nm	f= 0.00620	H-2 -> L 97.1%
S₀→S₅	2.89 eV	429 nm	f= 0.08580	H -> L+2 68.5%, H-3 -> L 22.4%
S ₀ →S ₆	2.94 eV	422 nm	f= 0.00550	H -> L+3 40.5%, H-3 -> L 27.3%, H -> L+2 22.7%
S ₀ →S ₇	S ₀ → S ₇ 3.11 eV 399 nm f=		f= 1.22720	H-1 -> L+1 65.4%, H -> L+3 17.9%, H-1 -> L+2
				5.5%
S₀→S ₈	3.24 eV	383 nm	f= 0.10450	H-1 -> L+2 54.7%, H-1 -> L+3 27.6%, H-1 -> L+1
				9.5%
S ₀ →S ₉	3.26 eV	381 nm	f= 1.10830	H-3 -> L 42.1%, H -> L+3 28.9%, H-1 -> L+1 15.0%
$S_0 \rightarrow S_{10}$	3.35 eV	371 nm	f= 0.00180	H-1 -> L+3 57.3%, H-1 -> L+2 35.4%

Transition	Energy	Wavelength	Oscillator strength	Contributions
$S_0 \rightarrow S_1$	1.83 eV	677 nm	f= 0.04930	H -> L 91.4%, H-1 -> L 5.5%
S ₀ →S ₂	2.21 eV	560 nm	f= 0.38040	H -> L+1 94.9%
S₀→S₃	2.40 eV	516 nm	f= 0.13420	H-1 -> L 90.2%, H -> L 6.1%
S ₀ →S ₄	2.51 eV	494 nm	f= 0.07640	H-2 -> L 63.8%, H -> L+2 28.0%
S₀→S₅	2.63 eV	472 nm	f= 0.34180	H-1 -> L+1 76.2%, H-2 -> L 13.7%, H -> L+2 5.4%
S₀→S ₆	2.65 eV	469 nm	f= 0.00850	H -> L+3 80.1%, H-2 -> L+1 11.0%
S ₀ →S ₇	2.68 eV	462 nm	f= 0.81920	H-2 -> L+1 81.0%, H -> L+3 9.6
S₀→S ₈	2.72 eV	456 nm	f= 0.58530	H -> L+2 52.2%, H-2 -> L 17.2%, H-1 -> L+1 14.5%, H-3 -> L 7.7%
S ₀ →S ₉	2.98 eV	416 nm	f= 0.25920	H -> L+4 40.2%, H-3 -> L 37.8%, H-1 -> L+2 13.6%
$S_0 \rightarrow S_{10}$	3.04 eV	408 nm	f= 0.01830	H-3 -> L+1 38.1%, H-1 -> L+3 23.4%, H -> L+5 20.4%, H-2 -> L+2 7.5%

Table S8. Electronic transition contributing to UV/Vis/NIR spectrum of compound **2f**, B3LYP/6-31G(d).

NMR Spectra

Compound 5, 2-(6-(tert-butyl)azulen-2-yl)-4,4,5,5-tetramethyl-1,3,2-dioxaborolane

Compound 6a, 6-(tert-butyl)-2-(4-(tert-butyl)phenyl)azulene

Compound 6b, 6-(tert-butyl)-2-(4-methoxyphenyl)azulene

Compound 6c, 2-(tert-butyl)-7-(6-(tert-butyl)azulen-2-yl)pyrene

Compound 7a, 6-(tert-butyl)-2-(4-(tert-butyl)phenyl)-1,3-diiodoazulene

Compound 7b, 6-(tert-butyl)-1,3-diiodo-2-(4-methoxyphenyl)azulene

Compound 7c, 2-(tert-butyl)-7-(6-(tert-butyl)-1,3-diiodoazulen-2-yl)pyrene

Compound 1b, 6-(tert-butyl)-2-(4-(tert-butyl)phenyl)-1,3-bis(4-methoxyphenyl)azulene

Compound 1c, 6-(tert-butyl)-1,2,3-tris(4-methoxyphenyl)azulene

Compound 1d, 2-(*tert*-butyl)-7-(6-(*tert*-butyl)-1,3-bis(4-(*tert*-butyl)phenyl)azulen-2-yl)pyrene

Compound 1e, 7,7'-(6-(*tert*-butyl)-2-(4-(*tert*-butyl)phenyl)azulene-1,3-diyl)bis(2-(*tert*-butyl)pyrene)

Compound 1f, 7,7',7"-(6-(tert-butyl)azulene-1,2,3-triyl)tris(2-(tert-butyl)pyrene)

Compound 3a, 3,6,11-tri-*tert*-butyl-13*a*-(4-(*tert*-butyl)phenyl)azuleno[1,2-/]phenanthren-14(13*a*H)-one

Compound 3b, 3,11-di-*tert*-butyl-6-methoxy-13*a*-(4-methoxyphenyl)azuleno[1,2-/]phenanthren-14(13*a*H)-one

Compound 3c, 11-(*tert*-butyl)-3,6-dimethoxy-13*a*-(4-methoxyphenyl)azuleno[1,2-/]phenanthren-14(13*aH*)-one

Compound 4d, 4,9,14-tri-*tert*-butyl-17-(4-(*tert*-butyl)phenyl)azuleno[2,1k]dibenzo[m,pqr]tetraphene

Compound 2d, 2,7,12,17-tetra-*tert*-butylazuleno[1,2,3-*cd*]tribenzo[*a*,*f*,*lm*]perylene

Compound 4e, 2,7,13-tri-*tert*-butyl-10-(7-(*tert*-butyl)pyren-2-yl)azuleno[1,2k]dibenzo[m,pqr]tetraphene

Compound 2f, 2,9,16,23-tetra-*tert*-butylazuleno[1,2,3-*hi*]dinaphtho[8,1,2-*pqr*:2',1',8'*yza*]pyreno[1,2,3-*uv*]hexacene

Cartesian coordinates of DFT-optimized structures

Compound 3a

Energy: -4681847.344 kJ/mol

С	0.86938	-1.16403	-0.46896	Н	2.01187	-0.39003	-3.36540
С	1.68630	-0.09963	-1.18210	Н	-5.01775	-0.84086	0.67919
С	0.61401	0.93592	-1.60320	Н	-2.79681	-4.46977	1.13381
С	-0.66447	0.53062	-0.99994	Н	-0.82208	-3.38451	0.26427
С	-0.53549	-0.73623	-0.44285	Н	-1.14378	3.00472	-2.00874
С	1.47230	-2.23743	0.12155	Н	-3.22988	4.27566	-1.85173
С	2.83764	-2.64733	-0.04427	Н	-5.08490	1.00880	0.20195
с	3.66373	-2.46156	-1.12765	н	4.02163	1.17274	-1.89611
С	3.21937	-1.75989	-2.31741	Н	5.53246	2.50265	-0.51749
с	2.30865	-0.76271	-2.38700	н	3.07841	1.70501	2.92472
C	-1.71279	-1.42481	0.04055	н	1.57753	0.37960	1.54324
с	-1.89828	1.26421	-0.99400	н	6.85292	-3.05685	0.13221
C	2.68455	0.65763	-0.28527	н	5.35497	-2.85174	1.05260
с	-2.94907	-0.70962	0.11130	н	5,96894	-1.51801	0.06437
c	-4.08477	-1.38270	0.62003	н	6,89817	-3,20805	-2.26764
c	-4.06848	-2.71228	1.02213	Н	5,98131	-1.71642	-2.53075
c	-2.85057	-3.41577	0.87912	Н	5,45714	-3.23654	-3.28354
c	-1.71591	-2.79260	0.40350	Н	5.91744	-5.13702	-1.02297
c	-2.00917	2,56649	-1,52684	н	4,36613	-5,06250	-1.87988
c	-3.19255	3,27685	-1,43216	Н	4,40833	-4.97729	-0.11100
c	-4.32860	2,72870	-0.79865	Н	-6.53374	-5.16176	1.05025
c	-4.21838	1,43660	-0.28722	Н	-4.83516	-5.35175	0.59279
c	-3.03684	0.67120	-0.37345	Н	-5.89064	-4.29129	-0.35611
c	3.80908	1,28475	-0.83752	Н	-5.86275	-4,48651	3,40760
c	4.66893	2.04189	-0.04515	Н	-4.75299	-3.12829	3.67729
c	4.44540	2,22034	1,33146	Н	-4.14732	-4.65145	3.00487
c	3.30875	1,60260	1,86985	н	-7.38632	-3.07785	2.07338
c	2.44341	0.84214	1.07870	Н	-6.84106	-2.14301	0.67645
0	0.83710	1.88567	-2.34256	н	-6.36438	-1.65595	2.31743
c	5,06310	-3,12890	-1.11600	н	-7.73820	3,24942	-1.26995
C	5.85288	-2.60684	0.10911	н	-6.92382	1.71506	-0.93104
C	5.88820	-2.79733	-2.37698	н	-6.57173	2.58723	-2.43190
c	4.92625	-4.66717	-1.02437	Н	-6.94821	4.20481	0.95344
C	-5.30713	-3.43519	1.57715	н	-5.22189	4.22286	1.36446
C	-5.65720	-4.63152	0.65930	н	-6.11382	2.69277	1.34115
C	-4.99347	-3.95639	3.00065	н	-6.55080	5.40804	-1.16455
с	-6.54031	-2.51647	1.66194	н	-5.37243	4.85959	-2.36294
С	-5.65632	3.49839	-0.65939	н	-4.82255	5.52739	-0.81200
С	-6.78720	2.71142	-1.36437	н	6.18033	5.11015	2.19480
С	-6.00299	3.66027	0.84022	н	4.49360	4.97803	1.65823
C	-5.58647	4.90303	-1.28902	н	5.81114	4.51998	0.56703
с	5.42431	3.06013	2.17150	н	5,73346	3.74292	4.20739
С	5.47952	4.50129	1.61053	Н	4.97820	2.14617	4.12009
С	5.00809	3.13708	3.65270	н	4.02407	3.60352	3.77630
С	6.83607	2.43057	2.09934	Н	7.54736	3.02283	2.68800
н	0.90025	-2.84362	0.81641	Н	7.20982	2.38597	1.07096
н	3.20384	-3.29543	0.74911	Н	6.83060	1.41003	2.49966
н	3.65546	-2.07913	-3.26033				

Compound 3b

Ene	Energy: -4457175.222 kJ/mol										
С	-1.10195	-1.00235	0.20012	с	3.73256	-2.24874	-0.74392				
С	-1.70041	0.22819	0.86171	С	3.41808	-3.55355	-1.10143				
С	-0.44925	1.04234	1.27549	С	2.08983	-4.01343	-0.97858				
С	0.73061	0.36752	0.71986	С	1.10743	-3.15068	-0.55202				
С	0.36148	-0.86824	0.19828	С	2.45152	2.10341	1.22419				
С	-1.89906	-1.95029	-0.37508	С	3.75745	2.55312	1.14609				
С	-3.32084	-2.07457	-0.22548	С	4.77279	1.76256	0.56593				
С	-4.10759	-1.69508	0.83652	С	4.41173	0.50351	0.08932				
С	-3.54874	-1.05781	2.01399	С	3.09626	-0.00026	0.16029				
С	-2.45913	-0.25869	2.07236	С	-3.51745	1.96995	0.41698				
С	1.38153	-1.79620	-0.23069	С	-4.20686	2.86399	-0.40816				
С	2.08816	0.83314	0.72727	С	-3.88482	2.94061	-1.76801				
С	-2.51511	1.13373	-0.08091	С	-2.86944	2.11871	-2.27918				
С	2.74137	-1.35048	-0.28752	С	-2.19534	1.23584	-1.44406				

0	-0.49140	2.04605	1.97658	Н	-1.41547	0.60855	-1.86525
С	-5.61109	-2.07401	0.81745	Н	-7.33559	-1.69313	-0.46622
С	-6.26759	-1.44501	-0.43581	Н	-5.81392	-1.80901	-1.36304
С	-6.37072	-1.54803	2.05323	Н	-6.17061	-0.35347	-0.42145
С	-5.77875	-3.61127	0.77097	Н	-7.44035	-1.75545	1.93672
0	4.31193	-4.47269	-1.54897	Н	-6.25115	-0.46564	2.17511
С	5.67957	-4.08913	-1.68322	Н	-6.04622	-2.03494	2.97959
С	6.23367	2.23852	0.44858	Н	-6.84293	-3.87722	0.76573
С	7.15813	1.26508	1.21849	Н	-5.31691	-4.08261	1.64644
С	6.64874	2.26774	-1.04203	Н	-5.32175	-4.04485	-0.12449
С	6.43647	3.65117	1.02962	Н	6.19765	-4.96910	-2.06611
0	-4.49162	3.77029	-2.66789	Н	6.10935	-3.80443	-0.71560
С	-5.52259	4.63213	-2.19738	Н	5.79187	-3.26059	-2.39235
н	-1.45061	-2.67949	-1.04202	Н	8.20198	1.59208	1.13950
н	-3.79813	-2.66183	-1.00708	Н	7.09754	0.24549	0.82355
Н	-4.05201	-1.25403	2.95709	Н	6.89253	1.23027	2.28137
Н	-2.10739	0.07889	3.04532	Н	7.68829	2.60354	-1.13880
Н	4.76214	-1.92538	-0.79342	Н	6.01472	2.95657	-1.61206
Н	1.86909	-5.05037	-1.21026	Н	6.57355	1.27911	-1.50709
Н	0.10809	-3.53921	-0.41662	Н	7.48705	3.94320	0.92265
Н	1.68249	2.72571	1.66527	Н	6.18876	3.69401	2.09628
Н	3.98848	3.53752	1.53684	Н	5.83043	4.39941	0.50636
Н	5.18427	-0.10927	-0.35928	Н	-5.85926	5.19846	-3.06721
Н	-3.78047	1.92406	1.46896	Н	-5.14860	5.32596	-1.43434
Н	-4.98315	3.48814	0.01952	н	-6.36526	4.06247	-1.78619
Н	-2.62371	2.18688	-3.33483				

Compound 3c

Energy: -4348167.184 kJ/mol

С	0.56973	-0.82746	-0.38637	С	-6.61581	4.32064	-0.90061
С	1.21697	0.46467	-0.83264	0	4.13616	3.20398	3.26358
С	-0.01635	1.36655	-1.09848	С	5.18398	4.11518	2.94888
С	-1.23107	0.65121	-0.65091	Н	0.73046	-2.77327	0.39791
С	-0.87405	-0.62678	-0.27002	н	3.09699	-2.96527	0.33211
С	1.27112	-1.95201	-0.06798	Н	3.49322	-0.70250	-3.19454
С	2.67391	-2.17027	-0.27841	Н	1.66860	0.76446	-2.98907
С	3.49225	-1.61215	-1.23046	Н	-4.10770	-2.16288	2.69243
С	2.99220	-0.68241	-2.23016	Н	-2.70674	-4.92862	-0.29448
С	1.96827	0.19167	-2.11348	н	-1.18360	-3.18797	-1.18427
С	-1.78181	-1.67646	0.24204	Н	-1.92335	3.21336	-0.15588
С	-2.56915	1.25194	-0.77295	н	-4.12236	4.26791	-0.35848
С	2.05822	1.15960	0.25511	Н	-5.78906	0.53237	-1.69693
С	-2.62787	-1.41928	1.33074	н	3.32834	2.19847	-1.13669
С	-3.47811	-2.39901	1.84266	н	4.58852	3.42799	0.56393
С	-3.50638	-3.67041	1.25250	Н	2.23814	1.56348	3.64795
С	-2.67067	-3.94228	0.15762	н	0.96978	0.32535	1.92528
С	-1.81702	-2.96247	-0.33170	Н	6.69525	-2.20270	0.00285
С	-2.76447	2.61242	-0.48296	Н	5.16008	-2.43158	0.85223
С	-4.01669	3.21708	-0.60198	н	5.63813	-0.81212	0.32160
С	-5.11344	2.46202	-1.03534	Н	6.82910	-1.63411	-2.32060
С	-4.93467	1.10395	-1.34696	Н	5.73531	-0.24525	-2.21926
С	-3.68667	0.51372	-1.21658	Н	5.43931	-1.53709	-3.40054
С	3.07767	2.05034	-0.09111	н	6.04672	-3.94552	-1.75177
С	3.80005	2.75352	0.87766	Н	4.52163	-3.79018	-2.64411
С	3.49671	2.57627	2.23236	н	4.50408	-4.21611	-0.92470
С	2.46734	1.69476	2.59452	Н	-5.70914	-5.41956	2.91254
С	1.75925	1.00456	1.61791	Н	-5.89812	-3.68153	2.55003
0	0.04358	2.46544	-1.63209	Н	-4.61906	-4.22973	3.67623
С	4.96238	-2.09837	-1.32028	Н	-7.67656	4.48695	-1.09391
С	5.65005	-1.87338	0.04840	Н	-6.02024	4.97524	-1.54831
С	5.77852	-1.32837	-2.37991	н	-6.39631	4.55101	0.14899
С	5.00724	-3.60266	-1.67925	Н	5.54561	4.49678	3.90506
0	-4.29980	-4.69749	1.66262	Н	4.81943	4.95063	2.33843
С	-5.17688	-4.47885	2.76560	Н	6.00649	3.61482	2.42280
0	-6.37803	2.94595	-1.19165	н	-2.61685	-0.43741	1.79427
Н	-3.57002	-0.53330	-1.47540				

Compound 2d

Energy: -5488784.837 kJ/mol

С	2.51166	-3.58953	0.67258	С	-2.59457	2.43504	-0.45061
С	0.20689	0.00148	0.23855	С	-3.92511	2.41478	-0.73640
С	-0.17850	-5.92802	2.15623	С	-4.65806	1.18564	-0.76590
С	1.90074	-4.72194	1.19555	С	-6.02543	-1.24322	-0.99950
С	0.52663	-4.73591	1.48216	С	-6.03306	1.14830	-1.03930
С	-0.22688	3.62185	1.02077	С	-6.74636	-0.05444	-1.14969
С	2.39207	1.17218	-0.06520	С	-8.25798	-0.02575	-1.44235
С	0.49195	4.78484	1.29025	С	-8.50865	0.69222	-2.79002
С	-0.19932	-3.58458	1.16130	С	-8.86577	-1.43848	-1.53060
С	1.86975	4.76659	0.99741	С	-8.98330	0.74173	-0.31118
С	-0.45570	-1.25459	0.23522	н	-2.03650	-3.41393	-0.40103
С	1.78679	-2.42690	0.35216	н	-4.41444	-3.41096	-0.86951
С	0.36379	-2.44611	0.55116	н	-6.52137	-2.20240	-1.10319
Ċ	3,69564	0.74554	-0.45543	н	-6.54453	2.09737	-1.17690
c	3,69916	-0.75156	-0.42465	Н	-4,44295	3,33770	-0.98578
c	0.34702	2.45933	0.45558	н	-2.06583	3, 37552	-0.52008
c	-0 46484	1 25272	0 18907	н	-9 93999	-1 36313	-1 73301
c	8.14744	-0.06182	-1.96627	 H	-8.74340	-1.99468	-0.59425
c	6 04483	-1 25202	-1 26872	н	-8 41758	-2 02622	-2 33975
c	-2 53524	-0 01737	-0 2/9//	н	-8 01231	0 16501	-3 61297
c	1 62600	0.01/5/	0.24044	и Ц	-0 58320	0.10501	-3 00688
c	1 76725	2 13020	0.12144	и Ц	-8 138/0	1 72272	-2 77921
c	1 94242	1 26750	0.25505	и Ц	-0.13049 0.63401	1 77270	-2.77921 0 22202
c	1 95205	-1.20/30	-0.07913	п Ц	-0.02491	0.77270	-0.22203
c	-1.0000	1.24207	-0.12555	п Ц	-10.00142	0.77829	-0.50900
c	2.40594	-1.10625	-0.02008	п С	-0.05197	0.24097	0.00022
C C	4.75991	-1.50080	-0.02/05	C	-1.05111	7,00440	2.10055
C	6.042/5	1.23156	-1.318/5	C	0.02/92	7.23094	0.93375
C	0.0/353	-0.00824	-1.48412	C .	0.55407	6.3/335	3.24807
C	2.48/85	3.62247	0.52333	н	0.44061	5.55047	3.96329
C	-0.14648	6.04142	1.90850	н	1.62528	6.55/04	3.11433
C	4.74533	1.54883	-0.88957	н	-2.21519	5.65425	1.2/065
C	8.21975	-0.80583	-3.32344	н	-2.05490	6.78716	2.61823
Н	7.85437	-1.83501	-3.25517	н	-1.84270	5.05518	2.90033
Н	9.25877	-0.84465	-3.67101	C	0.77985	-7.11045	2.39578
Н	7.62653	-0.28699	-4.08509	C	-1.33675	-6.42432	1.25828
С	8.77976	1.33011	-2.16473	C	-0.74854	-5.48378	3.52468
Н	8.26764	1.91140	-2.93944	Н	-0.96338	-6.74917	0.28011
Н	9.82042	1.20724	-2.48346	Н	-2.08682	-5.64411	1.09052
Н	8.78664	1.91622	-1.23907	Н	-1.84363	-7.27594	1.72821
С	9.00095	-0.81465	-0.91475	Н	0.05211	-5.13788	4.18858
Н	8.97336	-0.29931	0.05219	Н	-1.25801	-6.32275	4.01435
Н	10.04502	-0.85827	-1.24610	Н	-1.47328	-4.66946	3.41975
Н	8.65952	-1.84236	-0.75830	Н	1.20975	-7.48297	1.45885
Н	6.64742	-2.12671	-1.49280	н	0.23265	-7.93780	2.86131
Н	4.55406	-2.63268	-0.81697	Н	1.60304	-6.84034	3.06684
Н	6.63065	2.10549	-1.57034	Н	2.52148	-5.58015	1.42738
Н	4.53160	2.61268	-0.92163	Н	-1.26821	3.58147	1.30550
Н	3.59071	-3.59077	0.57530	Н	-1.24377	-3.54479	1.44304
Н	3.56631	3.62672	0.42067	н	2.48130	5.64275	1.19360
С	-2.57330	-2.47610	-0.36414	Н	-0.47327	7.03197	-0.02067
С	-3.93527	-0.02946	-0.56855	Н	1.08335	7.43375	0.72300
С	-3.90406	-2.47587	-0.65205	Н	-0.40755	8.14134	1.36339
С	-4.64616	-1.25490	-0.72407	н	0.11449	7.27409	3.69356

Compound 4d Energy: -5491837.596 kJ/mol

С	-2.59742	3.13779	-0.67734	С	-3.62827	0.06369	-0.21082
С	0.05841	-0.04048	0.22351	С	0.30219	-2.51983	0.31742
С	-2.30919	6.53456	1.15976	С	0.89762	-1.19546	0.09475
С	-2.57633	4.51403	-0.43186	С	-7.93570	-1.45216	-1.49719
С	-2.34669	5.02687	0.85168	С	-6.11828	0.09491	-0.69389
С	1.07575	-3.61429	0.77387	С	2.85194	0.30604	-0.11492
С	-1.92790	-1.51986	-0.03347	С	-1.38555	-0.22138	0.16204
С	0.54548	-4.87612	1.02808	С	-1.11015	-2.68886	0.20485
С	-2.14471	4.09133	1.88362	С	2.02238	1.42528	0.20243
С	-0.84371	-5.03302	0.84651	С	2.28173	-0.99941	-0.21797
С	0.64610	1.23392	0.30215	С	-2.44288	0.74769	0.10279
С	-2.38788	2.21059	0.35253	С	-4.90008	0.64907	-0.33458
С	-2.16830	2.72035	1.64521	С	-5.54563	-2.27835	-1.17664
С	-3.31491	-1.37875	-0.34652	С	-6.45301	-1.22181	-1.10356

С	-1.64519	-3.96791	0.47613	н	2.71176	-3.00415	-0.96609
С	1.39471	-6.05950	1.52492	н	10.09309	2.63374	-0.92218
С	-4.18095	-2.34731	-0.82858	н	8.82005	2.87795	0.28043
С	-8.30764	-0.51884	-2.67679	н	8.52170	3.26793	-1.42630
н	-8.18706	0.53939	-2.42613	н	8.36388	1.38733	-3.16704
н	-9.35521	-0.67602	-2.95934	н	9.97350	0.85964	-2.63767
н	-7.68460	-0.73079	-3.55328	Н	8.63517	-0.29758	-2.69277
С	-8.23771	-2.89830	-1.93728	Н	9.08054	-0.87018	-0.19298
н	-7.66617	-3.18813	-2.82596	Н	10.41136	0.29713	-0.19557
н	-9.30021	-2.98024	-2.19059	Н	9.11383	0.43146	1.00775
н	-8.03301	-3.62345	-1.14191	С	2.88282	-5.69416	1.68126
С	-8.84209	-1.13758	-0.28025	С	1.28705	-7.22280	0.50981
н	-8.60419	-1.79515	0.56373	С	0.86875	-6.53441	2.90088
н	-9.89296	-1.29710	-0.54882	н	0.93809	-5.73197	3.64443
н	-8.74095	-0.10266	0.06034	Н	-0.17683	-6.85585	2.84968
н	-6.93946	0.80487	-0.68413	Н	3.32607	-5.36842	0.73331
н	-4.93096	1.71427	-0.11375	Н	3.44196	-6.57228	2.02341
н	-5.93409	-3.22085	-1.54235	Н	3.03325	-4.89910	2.42042
н	-3.74478	-3.32721	-0.99679	С	-2.55771	7.39488	-0.09368
н	-2.76906	2.77904	-1.68887	С	-0.92097	6.90679	1.73350
н	-2.71808	-4.11354	0.46313	С	-3.40045	6.87521	2.20281
С	2.62335	2.72478	0.34889	н	-0.12667	6.68551	1.01102
С	4.24623	0.51658	-0.34179	н	-0.70025	6.35737	2.65468
С	3.96133	2.90811	0.18008	н	-0.88045	7.97802	1.96543
С	4.82136	1.81180	-0.18129	н	-4.39816	6.63091	1.81997
С	3.14229	-2.04079	-0.72427	н	-3.37959	7.94635	2.43761
С	4.46824	-1.83568	-0.97116	н	-3.25594	6.32605	3.13916
С	5.08556	-0.56169	-0.74484	н	-1.79703	7.22365	-0.86386
С	6.19967	1.98521	-0.38849	Н	-2.52184	8.45622	0.17643
С	6.45517	-0.33376	-0.94986	Н	-3.54167	7.19878	-0.53474
С	7.03844	0.92859	-0.76956	Н	-2.73966	5.18553	-1.26764
С	8.55102	1.11119	-0.99973	Н	2.12141	-3.44024	0.98238
С	8.89760	0.74006	-2.46152	Н	-1.96996	4.43592	2.89927
С	9.01352	2.55936	-0.75018	Н	-1.31589	-5.98975	1.05121
С	9.33192	0.18419	-0.03768	Н	1.66776	-6.92162	-0.47308
Н	1.97798	3.56286	0.60033	Н	0.25194	-7.55688	0.38135
Н	4.40273	3.89438	0.30110	Н	1.87490	-8.08268	0.85345
н	6.60644	2.98089	-0.24775	Н	1.46132	-7.38410	3.26141
н	7.06798	-1.17569	-1.26130	Н	0.01685	2.10268	0.43942
н	5.07520	-2.64388	-1.37217				
н	-2.01197	2.02930	2.46966				

Compound 4e

Energy: -6497620.397 kJ/mol

С	-3.69843	1.26140	0.94037	н	-8.06383	-2.75967	-2.63085
н	-3.57005	1.90851	1.62392	С	0.76525	-3.26884	2.02965
С	0.19491	-0.74607	0.85399	С	-0.02916	0.65123	0.53359
С	-6.47066	-0.73871	-0.58652	С	2.42415	1.03735	0.39493
С	-4.98394	0.73717	0.71914	С	-0.55879	-2.83212	1.85926
С	-5.17790	-0.19300	-0.33406	н	-1.27612	-3.39454	2.12712
С	3.89406	-0.89398	-0.01562	С	1.77648	-2.45879	1.52546
С	1.06641	1.53458	0.52966	н	2.67578	-2.75247	1.61229
С	4.98353	-0.00621	-0.23392	С	-1.24729	1.39224	0.38508
С	-4.07209	-0.60791	-1.12356	С	-9.00655	-1.91559	-1.03684
С	4.78485	1.40175	-0.13411	С	7.37930	0.39228	-0.72220
С	-0.83122	-1.59567	1.30713	С	-1.78499	3.80343	-0.00498
Н	-1.73464	-1.31155	1.23145	Н	-2.66556	3.53671	-0.24174
С	-2.60587	0.84786	0.17327	С	8.86374	-1.49488	-1.22243
С	-2.80382	-0.09862	-0.83712	С	0.74466	5.37087	0.86886
н	-2.05762	-0.40245	-1.34059	Н	1.34028	5.99479	1.26683
С	0.54752	2.87144	0.52701	С	-0.42874	5.91736	0.38295
С	-0.91502	2.74792	0.33455	С	-7.56595	-0.40379	0.24657
С	2.63415	-0.37273	0.42060	С	3.50893	1.88846	0.11219
С	1.53402	-1.22539	0.89297	Н	3.36280	2.82681	0.08931
С	8.63960	-0.12905	-1.01451	С	-5.53822	-1.99804	-2.48039
н	9.37635	0.46772	-1.07488	Н	-5.66342	-2.57657	-3.22352
С	-4.30384	-1.52550	-2.20813	С	-8.80748	-1.02007	0.01307
Н	-3.57113	-1.80475	-2.74451	Н	-9.53524	-0.81967	0.59001
С	7.77009	-2.35220	-1.13368	С	-6.66770	-1.64044	-1.66444
н	7.89342	-3.27904	-1.30159	С	6.48577	-1.87966	-0.80120
С	-0.56585	7.45099	0.27597	С	7.14577	1.80752	-0.57390
С	-1.55781	5.16046	-0.04718	Н	7.87375	2.41216	-0.65716
Н	-2.26444	5.67316	-0.42173	С	6.27863	-0.49938	-0.58015
С	5.90929	2.28836	-0.31860	С	-7.34326	0.53619	1.31075
Н	5.77727	3.22730	-0.25930	Н	-8.06863	0.77881	1.87412
С	4.12322	-2.28263	-0.34072	С	10.29127	-1.98959	-1.50712
н	3.39689	-2.89295	-0.29052	С	5.34669	-2.74406	-0.71660
С	-7.93124	-2.18715	-1.88435	Н	5.44828	-3.66456	-0.92867

С	1.20023	4.03361	0.86838	Н	-0.06038	7.50757	-1.74236
н	2.10063	3.91853	1.14847	н	-0.89244	8.79415	-1.28029
с	-1.74368	7.92177	1.14354	н	-1.63936	7.39699	-1.50645
н	-2.56482	7.47464	0.85003	С	-0.11404	-5.28707	3.30823
н	-1.84943	8.89180	1.05234	н	-0.60172	-4.70835	3.93087
н	-1.56787	7.69916	2.08159	н	0.16118	-6.10601	3.77078
с	0.68424	8.21451	0.73879	н	-0.69505	-5.51772	2.55344
н	0.86946	7.99988	1.67689	С	10.90485	-1.17514	-2.66529
н	0.53062	9.17824	0.64842	н	11.78030	-1.54819	-2.89912
н	1.45017	7.95266	0.18638	н	11.01121	-0.24138	-2.38745
С	-6.11821	1.08683	1.53110	н	10.31243	-1.21853	-3.44478
н	-6.01030	1.71519	2.23546	С	11.13963	-1.80157	-0.24362
С	-10.35329	-2.60422	-1.30270	н	10.77060	-2.34721	0.48189
С	10.31946	-3.47069	-1.89546	н	11.12875	-0.85755	0.01962
н	9.76588	-3.61022	-2.69200	н	12.06178	-2.07941	-0.42485
н	9.96869	-4.00916	-1.15552	С	-11.32407	-2.43662	-0.14030
н	11.24180	-3.74073	-2.08727	н	-11.56287	-1.49068	-0.04799
С	1.12719	-4.55035	2.79124	н	-10.90065	-2.74891	0.68647
С	1.89407	-5.49720	1.85604	н	-12.13298	-2.96254	-0.31190
н	1.34110	-5.70186	1.07318	С	-10.97196	-1.98838	-2.57382
н	2.10525	-6.32698	2.33269	н	-11.82799	-2.42529	-2.76525
н	2.72520	-5.06690	1.56528	н	-10.36232	-2.11872	-3.33002
С	2.01359	-4.18664	3.99169	Н	-11.11865	-1.02948	-2.43414
н	2.83110	-3.74974	3.67372	С	-10.13489	-4.11377	-1.51593
н	2.24851	-5.00144	4.48298	Н	-9.70441	-4.49672	-0.72315
н	1.52660	-3.57680	4.58446	н	-9.56174	-4.25358	-2.29846
С	-0.81179	7,82100	-1,19693	н	-10,99942	-4.55154	-1.66177

Compound 2f (chiral)

Energy	: -75039	62.382 kJ/mol					
с	3.57994	2.65426	-0.92695	С	7.25108	2.14639	-1.23688
С	0.00096	0.46080	-0.00124	С	-10.27631	-1.41179	-0.44540
С	6.24870	0.18445	-0.13110	С	-5.24387	-1.64591	-1.03434
с	4.83401	2.04445	-0.91180	С	-1.51983	5.16212	0.13803
с	4.95099	0.74391	-0.33191	С	0.81479	9.21164	-1.28700
с	-3.78109	0.06047	-0.04492	н	1.84143	8.83816	-1.34871
С	-1.12449	2.66733	0.28597	Н	0.85828	10.30695	-1.29755
С	-4.94538	0.79281	0.33618	Н	0.28227	8.88474	-2.18744
С	3.77798	0.02297	0.04524	С	-1.30483	9.39139	0.01379
С	-4.81335	2.09085	0.91915	Н	-1.90371	9.12563	-0.86438
С	1.24122	-0.21319	-0.19590	Н	-1.18114	10.47962	0.00824
С	2.41639	1.99027	-0.49891	Н	-1.87345	9.13085	0.91330
С	2.48057	0.57595	-0.22912	С	6.02030	2.71033	-1.38060
с	-0.71118	4.02815	0.14999	С	10.25648	-1.51942	0.45469
С	0.75367	4.01951	-0.14289	C	-10.28299	-2.76030	-1.19006
с	-2.47688	0.59870	0.22894	н	-9.72173	-3.52934	-0.64725
С	-1.24648	-0.20251	0.19068	Н	-9.85911	-2.67287	-2.19691
с	-8.67191	0.37822	0.39992	н	-11.31411	-3.11567	-1.29518
с	3.97751	-1.19349	0.78994	С	0.85650	9.21507	1.24487
С	-7.71357	-1.51291	-0.72449	Н	0.35313	8.89200	2.16322
с	0.08627	8.72746	-0.00820	н	0.90105	10.31041	1.24976
С	1.27019	6.50801	-0.03551	Н	1.88468	8.84175	1.27496
С	-5.99183	2.76814	1.39183	C	-10.93020	-1.62384	0.94103
С	-3.99496	-1.15115	-0.79325	Н	-11.94784	-2.01578	0.82412
С	7.69212	-1.59420	0.72766	Н	-10.99586	-0.68979	1.50880
С	-0.01278	-2.31295	-0.00763	Н	-10.35647	-2.34137	1.53903
С	0.00743	1.87793	0.00145	C	-11.12353	-0.40677	-1.26205
С	-2.39697	2.01188	0.50197	Н	-10.68780	-0.24285	-2.25444
С	1.16048	-1.61818	-0.44401	Н	-11.19837	0.56513	-0.76313
С	-1.17903	-1.60924	0.43168	Н	-12.14163	-0.79161	-1.39717
С	1.15035	2.65975	-0.28135	C	10.91104	-1.73610	-0.93070
С	8.84022	-0.94310	0.26633	Н	10.98849	-0.80174	-1.49648
С	-7.40690	0.94535	0.59823	Н	10.33068	-2.44601	-1.53138
С	1.57513	5.15066	-0.12993	Н	11.92396	-2.13965	-0.81226
С	-8.85358	-0.85099	-0.25899	C	11.11271	-0.52501	1.27490
С	-1.20884	6.52544	0.03755	Н	12.12628	-0.92107	1.41155
С	0.03044	7.17738	-0.00237	Н	10.67664	-0.35809	2.26664
С	7.41442	0.87062	-0.58921	Н	11.19915	0.44688	0.77783
С	-3.55325	2.68716	0.93376	C	10.24713	-2.86922	1.19702
С	5.22055	-1.70212	1.03189	Н	9.67905	-3.63135	0.65158
С	8.67304	0.28946	-0.38981	Н	9.82185	-2.77903	2.20304
С	6.40337	-1.05989	0.53938	н	11.27419	-3.23567	1.30393
С	-6.41870	-0.99273	-0.53752	н	2.14900	7.14512	-0.02476
С	-7.22870	2.21775	1.24894	Н	2.64006	4.95046	-0.18114
С	-6.24943	0.24800	0.13607	н	-2.08606	7.16025	0.02706

н	-2.58608	4.97121	0.19626	С	1.00317	-5.88162	-0.62694
н	3.50909	3.66730	-1.30438	С	-1.06206	-5.87116	0.59173
н	8.13865	2.66325	-1.59361	С	-0.03169	-6.60247	-0.02051
н	5.91288	3.68794	-1.84428	С	-0.07359	-8.14234	-0.00222
н	3.11397	-1.69773	1.20430	С	-1.36590	-8.63076	-0.69929
н	5.32765	-2.60961	1.62128	С	1.13018	-8.77152	-0.72896
н	7.77155	-2.54163	1.25002	С	-0.06464	-8.63811	1.46371
н	9.54289	0.82747	-0.75759	Н	2.98754	-1.84454	-1.59994
н	-3.47103	3.69817	1.31429	н	2.86635	-4.27733	-1.78797
н	-5.87312	3.74338	1.85777	н	1.81833	-6.40191	-1.11840
н	-8.11009	2.74309	1.60859	Н	-1.88099	-6.39312	1.07975
н	-3.13765	-1.66316	-1.21104	н	-2.91164	-4.25685	1.76622
н	-5.36168	-2.55037	-1.62637	Н	-3.00666	-1.82240	1.59001
н	-9.53526	0.92448	0.77088	Н	1.05242	-9.86370	-0.68853
н	-7.80409	-2.45809	-1.24914	н	2.08137	-8.49033	-0.26292
С	2.16604	-2.37624	-1.13520	Н	1.16672	-8.48136	-1.78517
С	-0.01950	-3.74522	-0.01126	н	-1.39648	-8.30025	-1.74400
С	2.10026	-3.73453	-1.23989	Н	-1.41111	-9.72650	-0.68803
С	1.03909	-4.47515	-0.62227	Н	-2.26587	-8.25505	-0.20135
С	-2.19187	-2.36081	1.12118	Н	-0.92302	-8.25908	2.02817
С	-2.14081	-3.71950	1.21946	н	-0.10285	-9.73380	1.49320
С	-1.08735	-4.46887	0.59677	н	0.84669	-8.31588	1.98066

Compound 2f (*meso*) Energy: -7503971.074 kJ/mol

С	-3.63059	2.68370	0.00481	С	-0.79227	8.38039	3.94648
С	-0.00171	0.42156	0.33359	н	-1.81943	8.00587	3.90043
С	-5.89635	0.11359	-1.46127	н	-0.83193	9.41507	4.30658
С	-4.79621	2.08088	-0.46936	н	-0.24838	7.78384	4.68775
С	-4.75396	0.70330	-0.83956	С	1.30776	8.97381	2.73865
С	3.56567	-0.02555	-0.62470	н	1.91310	8.46056	3.49393
С	1.15943	2.61005	0.56266	н	1.18375	10.01047	3.06963
С	4.74751	0.74758	-0.84132	н	1.87004	8.99395	1.79848
С	-3.56416	-0.05797	-0.62314	С	-6.01402	2.82238	-0.66660
с	4.77514	2.12623	-0.47302	С	-9.44196	-1.68539	-3.33416
С	-1.25085	-0.24074	0.44115	С	9.26563	-3.06075	-3.76373
с	-2.45179	1.96055	0.24594	н	8.47247	-3.16185	-4.51321
С	-2.44848	0.53418	0.04747	н	9.02012	-3.70902	-2.91486
с	0.72874	3.90418	0.97889	н	10.19241	-3.43752	-4.21063
с	-0.76940	3.89521	0.98046	С	-0.87134	9.18792	1.54778
C	2.44320	0.55590	0.04455	н	-0.38099	9.17661	0.56764
c	1.25405	-0.22999	0.43935	н	-0.91683	10.22762	1.89226
c	8.20723	0.33062	-2.23106	н	-1.89932	8.83774	1.41401
c	-3.52948	-1.38207	-1.19346	С	9.84069	-0.76898	-4.58122
c	7,01124	-1.72725	-2,53261	н	10,74547	-1.17889	-5.04617
c	-0.08307	8.33058	2,56983	н	10.03858	0.27935	-4.33442
c	-1.26897	6.22401	1.87598	н	9.03570	-0.79334	-5.32483
c	5,98542	2.88005	-0.67081	С	10.62711	-1.54732	-2.31213
c	3.54477	-1.35033	-1.19391	Н	10.39098	-2.13523	-1.41759
c	-6.99083	-1.79269	-2.53528	н	10.84818	-0.52353	-1,99264
c	0.01093	-2.17048	1.31911	н	11,53851	-1.96140	-2.76017
c	-0.00781	1.84385	0.38036	C	-10,60898	-1.64980	-2.31852
c	2.43147	1,98296	0.24105	н	-10.84095	-0.62866	-1.99827
c	-1.24671	-1.54266	1.01770	н	-10.36807	-2.23622	-1,42428
c	1.26212	-1.53152	1.01614	Н	-11.51549	-2.07269	-2.76824
c	-1.18540	2,60134	0.56707	С	-9.82752	-0.86093	-4.58567
c	-8,18303	-1.07732	-2.68718	н	-10.72749	-1.27945	-5.05230
c	7.09016	0.93256	-1.63746	Н	-9.02135	-0.87626	-5.32824
c	-1.58188	4.94244	1,42458	н	-10.03635	0.18505	-4.33790
c	8.19604	-0.99989	-2.68447	C	-9.23038	-3.14771	-3.77002
c	1.21257	6.24283	1.87132	Н	-8.97916	-3.79437	-2.92161
c	-0.02785	6.86316	2.06964	н	-8.43547	-3.23991	-4.51879
c	-7.09795	0.86499	-1.63666	Н	-10.15282	-3.53338	-4.21831
c	3,60342	2.71774	-0.00043	н	-2.14428	6,81342	2,13078
c	-4.61174	-1.93656	-1.81017	Н	-2.64521	4.72714	1.44338
c	-8,20835	0.25228	-2,23195	н	2,08728	6.83289	2,11522
c	-5.85159	-1.22678	-1.93410	н	2,59557	4,74889	1,43343
c	5.86568	-1.17232	-1,93299	н	-3,63693	3,75925	0.13299
c	7.09584	2.30748	-1.21060	н	-8.03331	2.81050	-1.34284
c	5.89627	0.16901	-1.46179	н	-6.03280	3.86814	-0.36956
c	-7.11820	2,23908	-1.20788	н	-2.60511	-1.94405	-1.14926
c	9,46185	-1.59588	-3.32925	н	-4,53881	-2,93403	-2.23702
c	4.63310	-1.89454	-1.80923	н	-6.91695	-2.81611	-2.88736
с	1.53032	4.95479	1.41855	н	-9.11280	0.84450	-2.34458

Н	3.59882	3.79350	0.12598	С	0.81293	-7.32274	5.23452
н	5.99316	3.92637	-0.37528	С	-1.33182	-7.99593	4.13844
н	8.00505	2.88816	-1.34587	С	0.83217	-8.40291	2.95297
н	2.62601	-1.92151	-1.14999	н	-3.39452	-1.68939	1.33045
н	4.57055	-2.89311	-2.23516	Н	-3.36148	-3.88243	2.38171
н	9.10562	0.93198	-2.34382	н	-2.12369	-5.83869	3.15154
н	6.94818	-2.75184	-2.88339	н	2.17690	-5.82174	3.15229
С	-2.44923	-2.19996	1.45695	н	3.40195	-3.84977	2.37993
С	0.01682	-3.45316	1.96242	н	3.41178	-1.65718	1.32841
С	-2.43189	-3.42219	2.05590	Н	-1.24360	-8.98496	4.60165
С	-1.20052	-4.10932	2.30511	н	-1.90212	-8.11477	3.21014
С	2.47190	-2.17755	1.45520	н	-1.91240	-7.36292	4.81907
С	2.46769	-3.39916	2.05406	н	0.29241	-6.63934	5.91536
С	1.24260	-4.09915	2.30445	н	0.86010	-8.30857	5.71278
С	-1.17133	-5.37713	2.91348	н	1.83949	-6.96091	5.11489
С	1.22158	-5.36299	2.91121	н	1.85869	-8.07304	2.76177
С	0.02663	-6.03126	3.21826	н	0.88103	-9.39690	3.41384
С	0.07379	-7.42173	3.87862	н	0.32473	-8.50125	1.98630

Compound (1a)^{.+}

Energy	Energy: –4486762.026 kJ/mol										
c	0.34709	2.25004	-0.06412	н	0.69915	7.61479	-0.21318				
c	1,03968	0.95782	-0.03796	н	0.39112	7,28791	1.51009				
C	0.07353	0.00692	0.00080	н	-0.38727	8.64892	0.70594				
c	-1.25077	0.66823	-0.03812	Н	-1.12992	7.13401	-1.96126				
C	-1.09025	2.01404	-0.09646	н	-2.03343	8.35022	-1.04765				
Ċ	-2.14244	2,97884	-0.29173	н	-2.78388	6.80652	-1.42437				
С	-2.04791	4.31398	-0.26543	н	-1.95191	6.44953	2.25422				
C	-0.90543	5.18821	0.00793	н	-3.27367	6.41101	1.07692				
C	0.37089	4.76019	0.09544	н	-2.52249	7.94231	1.49473				
С	0.95416	3,44179	-0.00544	н	1.26594	-1.56209	-1.87989				
C	0.30793	-1.44433	0.05210	н	1.64017	-3.96773	-1.78664				
С	-1.27046	6.70313	0.18418	н	-0.25591	-4.08793	2.10178				
C	-0.06956	7,60798	0.56825	н	-0.60969	-1.69206	1.99038				
С	-1.83753	7.27492	-1.13586	н	-0.61511	-6.34119	1.60219				
С	-2.31369	6.88114	1.31361	н	0.96002	-6.06805	2.37508				
C	0.93333	-2.11135	-1.00249	н	0.66678	-7.54315	1.45405				
С	1.14339	-3.49332	-0.94330	н	-0.81782	-6.23273	-0.97482				
С	0.73024	-4.25547	0.16639	н	0.39009	-7.52764	-0.98092				
С	0.09609	-3.56955	1.21456	н	0.64091	-6.08585	-1.96165				
С	-0.11200	-2.18558	1.15861	н	3.03770	-5.56776	0.91518				
С	0.97757	-5.77346	0.19552	н	2.92133	-5.69278	-0.84378				
С	0.46724	-6.46510	1.47992	н	2.69985	-7.12934	0.15152				
С	0.25881	-6.43935	-0.99770	н	-3.13678	0.60930	1.90844				
С	2.49271	-6.05413	0.09741	н	-5.26334	-0.56283	1.89132				
С	-2.54305	-0.03381	-0.06503	н	-4.38077	-2.05777	-2.07401				
С	2.48683	0.71354	-0.03255	н	-2.25238	-0.86769	-2.03474				
С	-3.40963	0.03004	1.02880	н	2.55027	-0.02189	1.99568				
С	-4.63343	-0.65055	1.01067	н	4.94095	-0.43444	2.00592				
С	-5.02519	-1.41817	-0.09831	н	5.18416	0.98687	-2.07537				
С	-4.13605	-1.47369	-1.18950	н	2.77722	1.38945	-2.06253				
С	-2.91394	-0.79325	-1.17475	н	-7.49273	-0.96532	1.28581				
С	3.12456	0.20423	1.09986	н	-6.71133	-2.38493	2.01187				
С	4.50528	-0.03058	1.09641	н	-8.16447	-2.57624	1.03160				
С	5.28984	0.24010	-0.03666	Н	6.93009	-1.54449	1.49641				
С	4.62974	0.75859	-1.16771	Н	7.21673	0.10735	2.08150				
С	3.25056	0.99255	-1.16722	н	8.46337	-0.73896	1.16496				
С	-6.35803	-2.18515	-0.14995	Н	-7.38567	-0.61123	-1.27731				
С	-7.22607	-2.01524	1.11736	Н	-8.16934	-2.19552	-1.38517				
С	6.80722	-0.01159	-0.07616	Н	-6.70267	-1.87522	-2.30650				
С	7.38126	-0.57829	1.24209	Н	-5.45955	-4.07369	0.50611				
С	-7.19768	-1.68922	-1.34711	Н	-5.55971	-3.92092	-1.25162				
С	-6.08037	-3.69501	-0.31457	н	-7.01361	-4.27026	-0.31981				
С	7.13496	-1.02835	-1.19096	Н	6.60041	-1.97265	-1.03316				
С	7.54636	1.31323	-0.36323	Н	8.20770	-1.25279	-1.21838				
н	-3.12406	2.56732	-0.52268	Н	6.86020	-0.65721	-2.18441				
н	-2.98381	4.82845	-0.47701	Н	7.30984	2.06911	0.39507				
н	1.15504	5.48907	0.28902	Н	7.28115	1.73299	-1.33973				
н	2.04320	3.45958	0.03724	Н	8.63301	1.16796	-0.36185				

Compound (1d) ^{.+}
Energy: -5494388.608 kJ/mol

C	3 66992	0 39030	0 08958	н	8 62454	1 10779	-1 03784
c	2 41775	0.0000	0.00000	и Ц	0.02404	0.96140	2 10159
Ċ	2.41/75	0.92722	0.07291	п	8.40721	-0.00149	2.19130
C	1.40478	-0.15590	0.01449	п	0.04904	0.75728	1.51652
C	2.05192	-1.35051	-0.05526	н	9.91928	-0.44652	1.3/192
C	3.49031	-1.05212	-0.01665	н	-0.11/59	1.48867	-1.52164
C	4.46081	-1.96999	-0.10124	н	2.0/955	-3.28988	1.82994
C	5.89671	-1.79644	-0.10103	Н	1.29991	2.37104	2.09121
С	6.67460	-0.70083	0.02820	Н	2.89007	2.71972	-1.89264
С	6.15481	0.65704	0.20829	Н	-4.61648	2.79233	-2.32714
С	4.90044	1.12742	0.22905	Н	-2.17788	2.58982	-2.33486
С	-0.06590	0.01406	0.05003	Н	-2.56225	-2.14358	2.46786
С	8.23938	-0.80411	0.00510	Н	-5.00378	-1.99129	2.42363
С	8.78124	-2.24436	-0.19561	Н	-7.05333	-1.01184	1.50623
с	8.81791	0.03688	-1.15718	н	-6.80246	2.07720	-1.49025
С	8.83180	-0.30813	1.34575	н	-8.82297	1.55663	1.98228
с	-0.85096	-0.72317	0.94929	н	-9.02987	-0.19975	2.02686
c	-2.25072	-0.63599	0.94559	н	-10.33341	0.81426	1,43111
c	-2.88408	0.26373	0.06381	н	-8.87222	2.77881	-0.34884
c	_2 107/1	1 053/8	-0 81164	н	-10 33963	1 86221	-0 70780
c	-0 70900	0 90/21	-0 82003	н	-8 98521	1 80262	-1 82795
c	1 /1000	-2 67492	-0 13220	н Ц	-8 91/1/	-1 56/15	-0 137/5
c	1.41909	-2.07492	-0.13222	п	-0.91414	-1.50415	-0.13/43
c	2.15/50	2.5/102	0.10209	п 11	-0.03/21	-0.74705	-1.70194
C	1.51190	-3.56322	0.94273	H	-10.30869	-0.65201	-0.72199
C	0.85593	-4./9962	0.90663	C	1.38202	6.68598	0.132/3
C	0.08884	-5.18588	-0.20433	C	0.70415	7.21622	1.415/8
C	0.032//	-4.29109	-1.29082	C	0.4/119	7.06828	-1.05447
C	0.68621	-3.05463	-1.25706	C	2.73361	7.41934	-0.00896
С	1.55888	2.97037	1.22108	С	-0.67347	-6.52111	-0.26369
С	1.30666	4.34901	1.23812	C	-0.08573	-7.39606	-1.39007
С	1.62434	5.16641	0.13966	С	-2.16991	-6.26164	-0.54862
С	2.19725	4.54092	-0.98364	C	-0.60051	-7.33941	1.04550
С	2.44559	3.16574	-1.00524	Н	0.43011	-7.62097	1.29048
С	-4.29805	0.36706	0.05448	Н	-1.17496	-8.26989	0.96253
С	-4.93540	1.27930	-0.81011	Н	-1.01255	-6.77926	1.89319
С	-4.14868	2.07606	-1.65532	Н	-0.18624	-6.92423	-2.37381
С	-2.75507	1.96081	-1.66057	Н	-0.59417	-8.36581	-1.44334
с	-3.03279	-1.43410	1.79062	н	0.98125	-7.58740	-1.22503
С	-4.42628	-1.34831	1.76333	Н	-2.33314	-5.79671	-1.52687
С	-5.07242	-0.44986	0.90319	н	-2.60658	-5.59924	0.20834
С	-6.47389	-0.35090	0.86476	н	-2.74147	-7.19722	-0.54299
с	-7.13923	0,56031	0.02172	н	-0.27748	6,75426	1,57279
c	-6.34129	1.36586	-0.81002	н	0.54630	8.30006	1.35983
c	-8.67788	0.61911	0.00428	н	1,31729	7.02715	2,30456
c	-9 24283	0 70069	1 44113	н	3 41525	7 15192	0 80729
c	-9 24466	1 83/86	-0 76368	н	2 59786	8 50696	0.00723
c	_0 21351	-0 65762	-0.67581	н Ц	2,33700	7 17754	-0.05263
L L	- J.21551 A 16946	2 01515	0.07501		0 19152	6 52212	1 00677
а	4.10040	-3.01313	-0.20250	п	-0.40452 0 02007	6 82202	-1.000//
n U	C.5/009	-2./0445 1 /2522	-0.22333 0 3E035	п	0.2520/	0.05/02	1 05401
п 11	0.91323	1.42533	6.20225	н	0.25284	0.14209	-1.05491
п 	4./945/	2.198/2	0.39384	н	86969.0	4./5899	2.13830
н	8.44949	-2.6/1/5	-1.14894	н	2.4661/	5.1204/	-1.86403
н	8.46683	-2.91113	0.615/6	н 	-0.53694	-4.5389/	-2.18389
н	9.87815	-2.25572	-0.20733	Н	0.95579	-5.44072	1.77787
Н	8.39162	-0.27498	-2.11783	Н	-0.36892	-1.39982	1.65154
Н	9.90625	-0.07917	-1.22436	Н	0.59848	-2.38324	-2.10835

Compound (1e)^{.+}

Energy: -6501990.421 kJ/mol

С	0.77480	2.85916	-0.12605	С	-1.16295	5.35908	0.14805
С	1.20203	1.45316	-0.16484	С	0.15175	5.99476	0.04391
С	0.06764	0.71238	-0.10128	С	1.30777	5.32171	-0.12714
С	-1.10053	1.62293	-0.02928	С	1.60066	3.90984	-0.20667
С	-0.67764	2.91125	-0.00749	С	-0.00263	-0.75638	-0.02385
С	-1.52324	4.06995	0.12478	С	0.13299	7.55938	0.16266

С	-0.44068	7.98271	1.53647	Н	0.12634	7.52611	2.35636
С	1.53141	8.22542	0.06139	Н	-1.49097	7.69933	1.65895
С	-0.72043	8.18109	-0.96709	Н	-0.39257	9.07043	1.66663
С	-0.65556	-1.48873	-1.01759	Н	2.20128	7.88256	0.85865
С	-0.80146	-2.87703	-0.90370	Н	1.45772	9.31575	0.15757
С	-0.29697	-3.57579	0.20477	Н	2.00589	8.02393	-0.90598
С	0.37511	-2.82430	1.18754	Н	-1.77677	7.90265	-0.89829
С	0.51608	-1.43723	1.07929	Н	-0.35516	7.86908	-1.95239
С	-2.50154	1.16838	0.01773	Н	-0.68320	9.27651	-0.93146
с	2.57400	0.91924	-0.18195	н	-1.08022	-0.98301	-1.88192
С	-3.35333	1.38863	-1.07239	Н	-1.33580	-3.38634	-1.70072
С	-4.65873	0.87423	-1.08212	Н	0.79136	-3.30980	2.06747
С	-5.11165	0.11329	0.01912	Н	1.02325	-0.89391	1.87314
С	-4.26031	-0.10515	1.12521	Н	-2.98536	1.94821	-1.93067
С	-2.96582	0.43385	1.11493	Н	-2.29799	0.25212	1.95509
С	3.00006	0.05357	-1.19725	Н	2.30813	-0.21729	-1.99278
С	4.29787	-0.48238	-1.19483	Н	3.13738	1.92407	1.63862
С	5.19345	-0.12537	-0.16240	Н	-6.32769	-2.01944	3.03825
С	4.77772	0.75807	0.85781	Н	-4.07926	-1.05746	3.06074
С	3.46832	1.26032	0.84147	Н	-5.19283	1.66751	-3.02535
С	-6.41515	-0.44362	0.00853	Н	-7.44758	0.71848	-3.03304
С	-6.86284	-1.22022	1.09723	Н	-9.19576	-0.62460	-1.95173
С	-6.00523	-1.42271	2.18786	Н	-8.46540	-2.37792	1.91771
С	-4.72141	-0.87211	2.20243	Н	-2.23995	-4.88200	1.65710
С	-5.51920	1.08312	-2.16783	Н	-0.73347	-5.02396	2.56979
С	-6.80675	0.54021	-2.17230	Н	-1.43285	-6.45089	1.80950
С	-7.26587	-0.23056	-1.09462	Н	-2.25011	-5.37193	-0.88105
С	-8.55090	-0.80100	-1.09257	Н	-1.33333	-6.85282	-0.60593
С	-9.02029	-1.59192	-0.02486	Н	-0.71700	-5.64525	-1.73426
С	-8.15336	-1.78083	1.06469	Н	7.65758	0.89456	2.67110
С	-0.47006	-5.09413	0.38044	Н	5.38462	1.79623	2.65890
С	-1.26123	-5.37631	1.67671	Н	4.05153	-1.66289	-2.99300
С	-1.23548	-5.77400	-0.77709	Н	6.31991	-2.57883	-2.96291
С	6.50841	-0.65499	-0.14768	Н	8.53658	-2.72872	-1.92917
С	7.40617	-0.29218	0.87674	Н	9.37989	-0.51694	1.67369
С	6.98038	0.59715	1.87345	Н	11.46653	-0.39553	-0.70925
С	5.68214	1.11254	1.86686	Н	11.05709	-1.45033	-2.06611
С	4.72654	-1.36767	-2.19253	Н	12.47553	-1.80411	-1.07043
С	6.02239	-1.89114	-2.17432	Н	11.40211	-1.28384	1.67241
С	6.92512	-1.54362	-1.15880	Н	12.20597	-2.79418	1.28549
С	8.23350	-2.06086	-1.12631	Н	10.60221	-2.82691	2.03143
С	9.14515	-1.73320	-0.10442	Н	10.34893	-3.86245	-1.65240
С	8.70686	-0.82832	0.87906	Н	9.93080	-4.40061	-0.01648
С	10.57597	-2.30411	-0.10195	Н	11.62273	-4.19349	-0.49447
С	11.44185	-1.44084	-1.03961	Н	-9.80228	-3.88108	-1.35299
С	11.22725	-2.29965	1.30122	Н	-10.45751	-2.51792	-2.26754
С	10.61616	-3.76983	-0.59449	Н	-11.54679	-3.57839	-1.38024
С	-10.42974	-2.20835	-0.08210	Н	-10.07721	-3.93994	1.22693
С	-10.56452	-3.09293	-1.34141	Н	-11.78113	-3.52323	1.03971
С	-10.77422	-3.09887	1.13347	Н	-10.75242	-2.53056	2.07056
С	-11.48215	-1.08086	-0.13785	Н	-11.37473	-0.45853	-1.03306
С	0.91492	-5.76923	0.47273	Н	-11.39905	-0.42044	0.73349
н	-2.58751	3.86852	0.23761	Н	-12.49921	-1.48986	-0.15062
н	-1.99898	6.04640	0.26792	Н	1.48789	-5.42341	1.34004
Н	2.23461	5.88681	-0.20143	Н	1.51466	-5.56258	-0.42155
Н	2.66552	3.71557	-0.33301	Н	0.81853	-6.85729	0.56611

Compound (1f)^{.+}

Ene	nergy: –7509594.912 kJ/mol									
С	0.58603	3.92141	-0.20988	С	-0.61361	-1.91114	-1.01729			
С	1.07931	2.54071	-0.30406	С	0.08229	-2.54163	0.03881			
С	-0.01396	1.74216	-0.20027	С	0.73256	-1.76242	1.01992			
С	-1.22227	2.59017	-0.06936	С	0.67813	-0.36538	0.93447			
С	-0.86049	3.89706	-0.02413	С	-2.59946	2.06936	-0.00095			
С	-1.75422	5.01228	0.16238	С	2.47576	2.06920	-0.35414			
С	-1.45030	6.31414	0.23347	С	-3.49709	2.31011	-1.04994			
С	-0.16542	7.01056	0.14581	С	-4.78435	1.75212	-1.04129			
С	1.00955	6.40213	-0.11262	С	-5.17133	0.92146	0.03380			
С	1.35705	5.01224	-0.29192	С	-4.27247	0.67602	1.09601			
С	0.01276	0.26996	-0.12271	С	-2.99814	1.26198	1.07034			
С	-0.23913	8.55773	0.39690	С	2.90244	1.16714	-1.33780			
С	-0.78993	8.83763	1.81616	С	4.20156	0.63501	-1.31900			
С	1.13078	9.28406	0.32124	С	5.10481	1.04693	-0.31461			
С	-1.14520	9.24095	-0.65298	С	4.69691	1.98402	0.65941			
С	-0.63650	-0.50869	-1.08784	С	3.38227	2.47103	0.63666			

С	-6.45667	0.32326	0.04068	Н	-0.79858	9.02625	-1.67040
С	-6.83877	-0.52274	1.10211	Н	-1.14345	10.33016	-0.52509
С	-5.93402	-0.75370	2.14804	Н	-1.16624	-0.01816	-1.90219
С	-4.66839	-0.16163	2.14599	Н	1.17287	0.23768	1.69359
С	-5.69111	1.98836	-2.08296	н	-3.18095	2,92636	-1.89002
C	-6.96080	1.40503	-2.06956	н	-2.29701	1.06428	1.87928
c	-7.35492	0.56531	-1.01807	н	2.20407	0.84702	-2.10910
c	-8.62114	-0.04556	-0.99952	н	3.05353	3,15973	1,41322
c	-9 02579	-0 90117	0.04176	н	-6 20436	-1 40562	2 97582
c	-8 11224	-1 12220	1 08693	н	-3 98895	-0 37139	2 96938
c	0 13087	-3.95603	0 11550	н Ц	-5 41608	2 62745	_2.90958
c	0.13007	4 59510	1 16427	н ц	7 62020	1 60705	2.91921
c	1 47497	2 70254	1.10457	п	-7.03930	0 15262	1 02/07
ć	1.4/40/	-3.79554	2.12/1/	п	-9.502//	0.15205	-1.62467
C	1.42397	-2.40008	2.05/1/	н	-8.3/266	-1.//32/	1.91/5/
C	-1.262/8	-2.70403	-1.9/2/2	н	2.02569	-4.25630	2.94298
C	-1.21/40	-4.09892	-1.89331	н	1.93812	-1.81619	2.81/50
C	-0.52083	-4.73906	-0.85765	н	-1.81328	-2.24184	-2.78931
C	-0.45592	-6.14112	-0.76738	Н	-1.73277	-4.68335	-2.65221
С	0.24427	-6.79661	0.26557	Н	-0.96449	-6.73147	-1.52766
С	0.88062	-5.98980	1.22339	Н	1.43380	-6.44154	2.04278
С	6.41683	0.51209	-0.27572	Н	7.59232	2.22438	2.43615
С	7.32237	0.93178	0.71933	Н	5.31870	3.12382	2.39277
С	6.90828	1.88275	1.66257	Н	3.93745	-0.65112	-3.04092
С	5.61010	2.39702	1.63771	Н	6.19705	-1.58379	-2.96290
С	4.61966	-0.30975	-2.26525	Н	8.40661	-1.70711	-1.91765
С	5.91066	-0.84361	-2.21895	Н	9.30231	0.74388	1.51391
С	6.82047	-0.44366	-1.22935	Н	11.40901	0.76315	-0.49010
С	8.12120	-0.97779	-1.16387	Н	11.08509	-0.17022	-1.95483
С	9.03938	-0.59421	-0.16879	н	12.42962	-0.62335	-0.89696
C	8.61767	0.38547	0.74921	н	11.14920	-0.31189	1.80228
С	10,46312	-1.17963	-0.11092	н	11,96157	-1.80305	1,36866
c	11.39899	-0.25061	-0.90735	н	10.29808	-1.85790	1.97117
c	10,99202	-1.29171	1.33887	н	10.36614	-2.60463	-1.79448
c	10 55237	-2 60067	-0 71510	н	9 83230	-3 27901	-0 24264
c	0 29326	-8 33//2	0.71010	н	11 55220	-3 02878	-0 57/35
c	0.25520	-8 97191	-0 07005	н Ц	1 05520	-8 46450	-1 11064
c	1 10499	-0.07101	1 50010		0.26969	-0.40450	1 97469
c	1.10466	-8.90032	1.50018	п	1 02740	-0.01347	-1.8/408
c	-1.13903	-0.09024	0.43134	п	1.02/40	-9.90462	-0.93031
Ċ	-10.41661	-1.56340	0.00247	н	2.15054	-8.5/28/	1.46/14
C	-10.55926	-2.40452	-1.28532	н	1.11099	-9.99690	1.48/8/
C	-10.69663	-2.50834	1.19318	н	0.68004	-8.59222	2.46263
C	-11.50686	-0.47086	0.01921	н	-1.76032	-8.63649	-0.43314
н	-2.80735	4.76085	0.27851	Н	-1.64201	-8.50852	1.32562
Н	-2.31373	6.95941	0.38842	Н	-1.13059	-9.99022	0.50192
Н	1.91040	7.00947	-0.17169	Н	-9.77439	-3.16777	-1.34666
Н	2.42178	4.87478	-0.47939	Н	-10.49420	-1.79227	-2.19133
Н	-0.18181	8.33701	2.57906	Н	-11.52724	-2.91836	-1.31675
Н	-1.82397	8.49974	1.93936	Н	-9.97457	-3.33236	1.23225
Н	-0.78165	9.91120	2.03897	Н	-11.69407	-2.95767	1.11494
н	1.83509	8.89812	1.06759	Н	-10.65944	-1.97488	2.15008
Н	1.02020	10.35785	0.51681	Н	-11.44915	0.18548	-0.85601
Н	1.58483	9.18782	-0.67174	Н	-11.41816	0.16110	0.91080
н	-2.18928	8.92080	-0.58023	Н	-12.50980	-0.91358	0.02265

Compound (4a)^{.+} Energy: -4483663.045 kJ/mol

с	0.25974	2.25085	-0.22222	С	-2.05839	-3.79778	-0.06536
с	-0.61735	1.23607	0.01393	С	-0.68245	-3.78089	0.16625
с	0.13035	-0.04648	0.13070	С	0.02077	-2.57844	0.26458
С	1.46372	0.21862	0.07177	С	-2.80481	-5.13700	-0.19787
с	1.58903	1.67340	-0.07367	С	-2.22165	-5.93936	-1.38163
С	2.74064	2.35738	-0.00980	С	-2.63722	-5.95526	1.10112
С	2.92245	3.79213	0.08129	С	-4.32269	-4.99162	-0.45101
с	2.09345	4.81269	-0.22606	С	2.59093	-0.71687	0.08047
С	0.76949	4.61782	-0.81990	С	-2.09694	1.23785	0.05306
С	-0.03370	3.54521	-0.78783	С	3.07203	-1.25601	-1.11312
С	-0.61504	-1.33431	0.14028	С	4.15951	-2.13523	-1.10007
С	2.50198	6.30467	-0.00038	С	4.80430	-2.49405	0.10059
С	1.40575	7.05502	0.79429	С	4.31244	-1.92904	1.28933
С	3.81300	6.48373	0.80883	С	3.22316	-1.04949	1.28062
С	2.72143	7.00796	-1.35842	С	-2.78075	0.00250	-0.01900
С	-2.02573	-1.30249	-0.02837	С	-4.19910	0.05180	0.01223
С	-2.69912	-2.54860	-0.15552	С	-4.94680	1.24355	0.14234

С	-4.22574	2.42599	0.28779	Н	-3.18838	-6.90139	1.04632
С	-2.82978	2.41852	0.26085	Н	-3.01349	-5.39966	1.96831
С	5.99558	-3.46807	0.07446	Н	-4.52669	-4.44913	-1.38154
С	7.12409	-2.88773	-0.80519	Н	-4.82019	-4.46252	0.37020
С	-6.48506	1.19772	0.16614	Н	-4.80427	-5.97300	-0.53907
С	-7.01075	0.52210	-1.11977	Н	2.60055	-0.99999	-2.05950
С	6.60266	-3.75029	1.46740	Н	4.49773	-2.53876	-2.05217
С	5.54193	-4.82348	-0.50933	Н	4.75997	-2.15517	2.25318
С	-6.95876	0.39193	1.39452	Н	2.86853	-0.62895	2.21943
С	-7.14865	2.59170	0.24658	Н	-4.76736	-0.87078	-0.03404
н	3.67038	1.80201	0.10879	Н	-4.71792	3.38206	0.44059
н	3.90557	4.01627	0.48955	Н	-2.32671	3.36461	0.43907
Н	0.37374	5.47334	-1.36448	Н	7.44686	-1.90692	-0.43628
Н	-0.97196	3.62203	-1.33307	Н	6.81384	-2.76099	-1.84813
н	1.19935	6.55530	1.74804	Н	8.00000	-3.54705	-0.81021
Н	0.46307	7.12598	0.24147	Н	-6.68820	-0.52126	-1.20433
Н	1.71391	8.08318	1.01871	Н	-6.65841	1.04939	-2.01424
Н	3.74950	5.99716	1.78911	Н	-8.10683	0.51878	-1.14461
Н	4.02436	7.54493	0.98912	Н	6.97927	-2.83379	1.93644
н	4.67973	6.07891	0.27376	Н	7.44716	-4.44636	1.39665
Н	1.80420	7.06326	-1.95348	Н	5.86844	-4.20355	2.14360
Н	3.47577	6.48325	-1.95632	Н	5.19489	-4.73345	-1.54455
н	3.06922	8.03816	-1.21680	Н	4.71957	-5.25017	0.07720
н	-3.76659	-2.54814	-0.33432	Н	6.36371	-5.54911	-0.50870
н	-0.12050	-4.70744	0.26243	Н	-6.61858	-0.64900	1.36350
н	1.08593	-2.66182	0.43788	Н	-8.05316	0.37061	1.45500
н	-1.16522	-6.18867	-1.23340	Н	-6.58177	0.83292	2.32495
н	-2.29788	-5.37289	-2.31739	Н	-6.86308	3.22304	-0.60298
н	-2.75742	-6.88577	-1.52002	Н	-6.87643	3.11700	1.16942
Н	-1.58985	-6.20673	1.30089	Н	-8.24223	2.50995	0.23650

Compound (4d)^{.+}

Energy: -5491253.190 kJ/mol

С	3.66338	0.14264	0.21096	С	-8.55526	1.10417	-0.21521
С	2.45365	0.74680	0.07692	С	-9.06720	1.02172	1.23925
С	1.40596	-0.32323	-0.01364	С	-8.99932	2.47345	-0.77799
С	2.01162	-1.57315	0.20298	С	-9.25233	0.01699	-1.06173
С	3.46446	-1.30045	0.14383	н	4.17010	-3.21816	-0.33076
С	4.43296	-2.18150	-0.14677	н	6.27311	-2.80706	-0.90993
С	5.83475	-1.93615	-0.42701	н	6.90913	1.00414	0.87036
С	6.61700	-0.86032	-0.19677	н	4.76523	1.76978	1.10488
С	6.13687	0.35555	0.46103	Н	7.98898	-2.15187	-2.36679
С	4.88503	0.81193	0.60097	н	8.48819	-2.99341	-0.87954
С	0.01094	-0.20730	-0.18396	н	9.62798	-1.96961	-1.74274
С	8.12657	-0.83435	-0.60601	н	7.78747	0.44572	-2.35563
С	8.57580	-2.05881	-1.44557	н	9.47353	0.40339	-1.82030
С	8.43412	0.41165	-1.47100	н	8.29795	1.34792	-0.91964
С	9.02014	-0.81628	0.65500	н	8.80346	-1.67339	1.30331
С	-0.81713	-1.33877	0.04242	н	8.88703	0.09320	1.24957
С	-2.25752	-1.08583	0.14344	н	10.08210	-0.86468	0.38621
С	-2.80844	0.15537	-0.27230	н	0.05092	1.83091	-0.91048
С	-1.95694	1.18553	-0.69070	н	2.85126	-4.07212	0.98200
С	-0.57903	1.00078	-0.60422	н	1.30189	2.09505	2.13001
С	1.24769	-2.73804	0.35375	н	3.21697	2.66241	-1.68345
С	2.27963	2.20262	0.20736	н	-4.23558	3.57387	-1.53143
С	1.81061	-3.99159	0.68396	н	-1.82248	3.20308	-1.47045
С	1.06890	-5.17308	0.67456	Н	-2.85640	-2.88374	1.22140
С	-0.26973	-5.15535	0.30942	н	-5.22325	-2.50139	1.18528
С	-0.83733	-3.88657	0.06290	н	-7.14676	-1.09397	0.62531
С	-0.15469	-2.64125	0.17634	Н	-6.49525	2.80427	-1.08431
С	1.65692	2.74577	1.33329	Н	-8.56834	1.76244	1.87550
С	1.46993	4.12907	1.44561	Н	-8.89406	0.03604	1.68483
С	1.90343	5.00923	0.44018	Н	-10.14588	1.21084	1.29000
С	2.53920	4.44331	-0.68225	Н	-8.57660	3.30373	-0.20030
С	2.72658	3.06232	-0.79837	Н	-10.09023	2.58021	-0.74242
С	-4.21425	0.40339	-0.26068	Н	-8.69779	2.59642	-1.82476
С	-4.74049	1.63000	-0.72356	Н	-9.08429	-0.98992	-0.66431
С	-3.86070	2.61782	-1.17378	Н	-8.88718	0.02789	-2.09556
С	-2.48457	2.40305	-1.14413	Н	-10.33719	0.17272	-1.08944
С	-3.19343	-1.98321	0.71506	С	1.70370	6.53261	0.52648
С	-4.57351	-1.75059	0.74121	С	1.00111	6.99841	1.82168
С	-5.10029	-0.56815	0.23133	С	0.83653	7.00754	-0.65956
С	-6.48514	-0.32228	0.23547	С	3.07401	7.24184	0.47049
С	-7.03265	0.88325	-0.24051	С	-1.11874	-6.43015	0.17326
С	-6.13141	1.84888	-0.71484	С	-0.35846	-7.72559	0.53858

С	-1.59758	-6.57763	-1.28706	Н	0.88369	8.08868	1.83716
С	-2.34734	-6.34665	1.10513	Н	1.57616	6.72369	2.71363
Н	-2.04228	-6.19844	2.14778	Н	3.72773	6.90135	1.28226
Н	-2.94160	-7.26680	1.05884	Н	2.96103	8.32811	0.56640
Н	-3.01690	-5.52222	0.83747	Н	3.59852	7.05740	-0.47341
Н	0.50874	-7.88275	-0.11339	Н	-0.13506	6.49920	-0.66632
Н	-1.00373	-8.60593	0.43214	Н	1.31487	6.81460	-1.62610
Н	-0.00643	-7.70694	1.57662	Н	0.64841	8.08619	-0.60482
Н	-0.74790	-6.61382	-1.97920	Н	0.97160	4.49036	2.34099
Н	-2.23964	-5.74660	-1.59895	Н	2.89799	5.07475	-1.49229
Н	-2.17705	-7.49845	-1.42187	Н	-1.86366	-3.87503	-0.29641
Н	-0.00189	6.56586	1.91487	Н	1.58531	-6.09447	0.92650

Compound (4e)^{.+}

Energy: -6498847.177 kJ/mol c -0.89485 2.58075 0.11295

С	-0.89485	2.58075	0.11295	С	11.32341	-1.52647	1.61488
С	-1.21310	1.27588	-0.09529	С	11.11338	-2.76647	-0.52878
С	0.06072	0.49733	-0.22307	С	11.59553	-0.32774	-0.58232
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c	0 67812	5 18683	-0 34870	н	-2 46441	5 52365	0 87726
c	-0 54857	5 70244	-0 12107	н	-2 75013	3 25851	0 98634
c	-1 6/912	4 93356	0.12107	н	0 18008	7 //931	-2 16063
c	-1 82515	3 60688	0.52940	н	1 1/3/3	8 029/2	-0 61381
c	0 27//7	-0 87336	-0 11509	н	-0 07702	8 96927	-1 /6231
c	-0 80171	7 10213	-0.45508	и Ц	-2 00665	6 68473	-2 27550
c	0.03171	7 9/573	-0.4558	и Ц	-2.00000	8 30511	-1 65583
c	0.25101	7.34373	-1.21333	п	-2.55495	6.00004	-1.05565
c	-2.14343	7.27041	-1.55525	п	-3.04044 0 20560	7 02240	1 5152
c	-1.14/20	7.90000	0.04047	п	-0.26509	7.92249	1.52152
Ċ	1.60004	-1.37249	-0.34114	н	-2.02497	7.62456	1.39185
C	1.71943	-2.79240	-0.34452	н	-1.32334	9.04673	0.63306
C	0.6/21/	-3.70186	-0.6060/	н	2.6/466	-3.22681	-0.06048
C	-0.57701	-3.1514/	-0.85140	н	-1.44311	-3.76449	-1.08331
C	-0.//026	-1.//286	-0./584/	н	-1.//838	-1.41832	-0.94501
C	2.45373	0.86885	0.21065	н	3.21942	2.64045	1.18581
С	-2.59783	0.78570	-0.02267	Н	-2.34517	-0.09720	1.92498
С	3.47275	1.67419	0.75566	Н	-3.13634	1.57862	-1.95045
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С	-3.03863	0.08217	1.10500	Н	9.60586	0.79605	1.24934
С	-4.35303	-0.40141	1.17976	Н	8.59834	-2.78935	-0.91267
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С	5.99813	-2.30643	-1.01076	Н	0.60571	-5.40091	1.56223
С	4.64454	-1.95392	-0.96222	Н	-7.67891	0.49419	-2.85519
С	5.77742	2.08969	1.40320	Н	-5.38531	1.33521	-2.96382
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c	8.32288	-1.86320	-0.41464	Н	-11.39994	-0.49660	1.56743
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c	-6.57137	-0.65221	0.16615	н	-12.56493	-2.27105	-0.73538
c	-7.46099	-0.41880	-0.90291	н	-11,13879	-2.28575	-1.77200
c	-7 01059	0 30121	-2 01877	н	-10 11821	-4 05045	1 41865
c	-5 70030	0.30121	-2 08144	н	-10 10939	-4 17938	-0 34376
c	-4 80425	-1 11888	2 29486	н	-11 6/271	-4 1/853	0.54570
c	-6 11508	_1 50017	2.25400	и Ц	10 75715	-2 27212	2 18422
c	-7 00017	-1.35917	1 20151	и Ц	11 2/212	-2.27512	2.10422
c	-8 22010	-1 85350	1 33/91	и Ц	12 37807	-1 87705	1 63/64
c	-0.33019 _0 3/073	-1 62691	0 281/5	п	10 60511	-1.02203	-0 02/2/
c	- 9.240/2	-1.00001	-0 820143	п	10.00011	-3.00234	-0.03434
c	-0.///20	-0.9121/	-U.025// 0.27616	п	10 010012	2.20222	1 2014
c	-10,0/018 11 20122	-2.10/02	0.3/010	п	11 53110	-2./5/00	-1.50243
c	-11.50123	-1.09194	1.01004	п	11 22071	0.0402/	-0.09040
c	-11.55495	-1.059/2	-0.0202	н	11,229/1	-0.20/23 0 E7042	-1.00090
c	-10.00100	-2./2580	0.10504	н	12.00123	-0.3/942	-0.03538
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Н	-1.13023	-5.91310	-0.27559
Н	-0.06755	-7.13373	-0.97359

Compound (4f)^{.+} Energy: -7506432.872 kJ/mol

С	1.24139	-3.41144	0.78143	С	-0.90039	9.35166	0.62262
c	1 46928	-2 02745	0 50636	C	-3 19565	8 77482	-0 18929
ĉ	0.21051	1 20250	0.10747	c c	10 10212	0.51774	2 40705
C	0.21051	-1.39259	0.42/4/	C	-10.18313	-0.51//4	-2.40/85
C	-0.82558	-2.35630	0.65231	C	-11.16439	-0.30477	-1.22987
С	-0.20202	-3.64112	0.83240	С	-10.28698	0.69788	-3.34783
С	-0.80130	-4.89870	0.89026	С	-10.59647	-1.78106	-3.20118
Ċ	-0 24108	-6 16677	1 07407	н	-1 87580	-4 92025	0 75571
c c	0.24100	6.50067	1.07407		1.07500	4.92029	0.75571
C	1.08956	-0.5886/	1.25579	н	-0.98217	-0.95989	1.08461
С	2.19687	-5.72737	1.18549	Н	3.16953	-6.18772	1.30743
С	2.25746	-4.35316	0.94200	н	3.26413	-3.94632	0.87843
C	-0.12228	0.01594	0.27479	н	1.32116	-8.62791	-0.62310
c	1 30//2	-8 102/6	1 50951	н	-0 28579	-8 75081	0 11129
c c	1.30442	0.10240	1.50551		0.20575	0.75001	0.11120
C	0.78293	-8.90480	0.29018	н	0.93648	-9.9/596	0.46113
С	2.78342	-8.48152	1.72252	Н	3.39842	-8.27254	0.84017
С	0.52512	-8.52028	2.78253	н	2.85157	-9.55677	1.91558
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c	-1 84674	1 78504	0 13008		-0 55186	-8 3/8/8	2 60132
c	-1.04074	1.70554	0.43098		-0.55180	-0.04040	2.09132
C	-0.81946	2.77054	0.33584	н	0.88115	-7.97042	3.6608/
С	0.53769	2.37321	0.14409	Н	0.67272	-9.58931	2.97087
С	0.85438	1.00564	0.18044	н	1.89732	0.72360	0.13848
C	-2,20359	-1.97989	0.56891	н	-3,10514	-3.71864	1,48199
ĉ	2 20202	1 45050	0 22277	 U	2 20421	1 10112	2 42606
C	2.02020	-1.43930	0.323/7	п	5.50431	-1.49445	2.42090
C	-3.28487	-2.814/5	0.91512	Н	2.64990	-1.31/81	-1.82181
С	-4.59981	-2.50873	0.57249	Н	-5.46165	1.57756	-2.74053
С	-4.85102	-1.35346	-0.24612	Н	-3.25022	1.26352	-1.75451
C	-3.80337	-0.43598	-0.51697	н	-5.51367	-4.21076	1.57100
c	2 50746	0 64417	0 09696	 L	7 77046	2 77502	0 72060
C	-2.30740	-0.04417	0.00000	п	-7.77940	-3.//362	0.72000
C	3.66808	-1.26182	1.42914	Н	-9.29881	-2.45184	-0.68998
С	4.96631	-0.74784	1.27553	Н	-7.85111	0.95876	-2.87087
С	5.43649	-0.43547	-0.03574	Н	2.01432	5.45410	-0.05397
c	4 58826	-0 64522	-1 16627	н	2 57426	3 05433	-0 15912
c	2, 20752	1 15404	0.06407		2.01607	1 51104	1 04702
C	5.29/52	-1.15404	-0.96407	п	-2.91007	1.51164	1.04/92
C	-6.15169	-1.14241	-0.77982	Н	-4.42468	3.86921	1.31730
С	-6.39883	-0.05918	-1.66656	Н	-3.73805	6.24472	1.14326
С	-5.30289	0.78600	-2.01283	н	0.35955	7.22583	0.29768
C	-4.06317	0.61263	-1.45602	н	6.72068	0.40091	-3.64011
c	E 70210	2 25719	0 02562	 L	4 45027	0 10112	2 22206
C	-3.70318	-3.33718	0.92505	п	4.45057	-0.49112	-3.33390
C	-6.95/34	-3.11868	0.44899	Н	5.48329	-0./5935	3.38978
С	-7.22587	-2.01652	-0.43792	Н	7.75049	0.13735	3.06902
С	-8.49393	-1.77852	-0.97119	н	9.52073	0.95646	1.57125
C	-8.75976	-0.70893	-1.85565	н	8.87199	1,12475	-2.67642
c	7 60994	0 12050	2 19002	 L	11 00021	0 56606	2 60905
C	-7.09884	0.12059	-2.18995	п	11.09921	0.50000	-2.09805
C	-1.12285	4.15576	0.46927	н	11.67780	-0.28705	-1.25876
С	-0.10419	5.13878	0.31601	Н	12.59308	1.08697	-1.90756
С	1.23645	4.70419	0.06348	н	10.18150	2.99378	-2.46935
C	1.54769	3,37314	-0.00138	н	11.69502	3.46164	-1.68348
c	2 14420	2 22076	0.92669		10 14505	2 76612	0.07640
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C	-3.42995	3.56874	0.99833	Н	11.72334	0.89550	1.02273
С	-2.44436	4.57871	0.77770	Н	11.09512	2.55044	1.17980
С	-2.72068	5.95211	0.89869	Н	12.58113	2.24607	0.27482
C	-1.73711	6.93587	0.72036	н	-1.90425	8.45263	3,03983
c	0 42652	6 60195	0 42257	 L	2 64042	8 00267	2 40211
C	-0.43033	0.30183	0.43237	п	-3.34943	8.09207	2.49211
C	6.75049	0.08177	-0.21681	Н	-2.92846	9.74429	2.38615
С	7.22950	0.39098	-1.52361	Н	-0.10312	9.17851	1.35448
С	6.35394	0.16681	-2.64372	Н	-1.21604	10.39518	0.72448
C	5,09632	-0.32692	-2.47511	н	-0.47941	9.23357	-0.38243
c	F 8440F	0 52052	2 20184		4 10502	0 10024	0.05105
c	3.04403	-0.32300	2.37104	п 	-4.10202	0.10034	-0.00
C	/.10025	-0.02586	2.21310	Н	-2.83346	8.60318	-1.20923
С	7.60118	0.29408	0.90428	Н	-3.47454	9.83112	-0.10204
с	8.89427	0.80500	0.69889	н	-10.89968	0.58824	-0.65251
c	9,38020	1,11373	-0.57678	н	-11,17611	-1.15765	-0.54305
c	9.50020	2.223/3	1 67151		12 10204	0 17202	1 (00(2)
C C	0.52009	0.09/14	-1.0/151	н	-12.18394	-0.1/292	-1.00303
C	10.79607	1.67196	-0.81402	Н	-10.03199	1.63361	-2.83725
С	11.58364	0.70011	-1.72519	Н	-11.31561	0.79273	-3.71097
с	10.69352	3.05378	-1.50322	н	-9.63834	0.59446	-4.22517
с	11,58557	1,84722	0.49704	н	-10,58905	-2.68062	-2,57662
c	_2 10005	Q 10107	0 85166	 u	_0 00101	_1 05250	_1 0/702
c	-2.10020	0.4242/	0.03400	п	-2.32101	-1.2000	-4.04/23
L	-2.655/3	8.68833	2.2//95	Н	-11.61181	-1.66132	-3.59603

References

- 1 T. D. Lash, J. A. El-Beck and G. M. Ferrence, Syntheses and reactivity of meso-unsubstituted azuliporphyrins derived from 6-tert-butyl- and 6-phenylazulene, *Journal of Organic Chemistry*, 2007, **72**, 8402–8415.
- A. C. Bédard, A. Vlassova, A. C. Hernandez-Perez, A. Bessette, G. S. Hanan, M. A. Heuft and S. K. Collins, Synthesis, crystal structure and photophysical properties of pyrene-helicene hybrids, *Chemistry A European Journal*, 2013, **19**, 16295–16302.
- L. J. Bourhis, O. V. Dolomanov, R. J. Gildea, J. A. K. Howard and H. Puschmann, The anatomy of a comprehensive constrained, restrained refinement program for the modern computing environment Olex2 dissected, *Acta Crystallogr A*, 2015, **71**, 59–75.
- 4 G. M. Sheldrick, A short history of SHELX, *Acta Crystallographica Section A*, 2008, **64**, 112–122.
- 5 G. M. Sheldrick, Crystal structure refinement with SHELXL, *Acta Crystallographica Section C*, 2015, **71**, 3–8.
- M. J. Frisch, G. W. Trucks, H. B. Schlegel, G. E. Scuseria, M. A. Robb, J. R. Cheeseman, G. Scalmani, V. Barone, G. A. Petersson, H. Nakatsuji, X. Li, M. Caricato, A. V Marenich, J. Bloino, B. G. Janesko, R. Gomperts, B. Mennucci, H. P. Hratchian, J. V Ortiz, A. F. Izmaylov, J. L. Sonnenberg, D. Williams-Young, F. Ding, F. Lipparini, F. Egidi, J. Goings, B. Peng, A. Petrone, T. Henderson, D. Ranasinghe, V. G. Zakrzewski, J. Gao, N. Rega, G. Zheng, W. Liang, M. Hada, M. Ehara, K. Toyota, R. Fukuda, J. Hasegawa, M. Ishida, T. Nakajima, Y. Honda, O. Kitao, H. Nakai, T. Vreven, K. Throssell, J. A. Montgomery Jr., J. E. Peralta, F. Ogliaro, M. J. Bearpark, J. J. Heyd, E. N. Brothers, K. N. Kudin, V. N. Staroverov, T. A. Keith, R. Kobayashi, J. Normand, K. Raghavachari, A. P. Rendell, J. C. Burant, S. S. Iyengar, J. Tomasi, M. Cossi, J. M. Millam, M. Klene, C. Adamo, R. Cammi, J. W. Ochterski, R. L. Martin, K. Morokuma, O. Farkas, J. B. Foresman and D. J. Fox, 2016.
- 7 A. D. Becke, Density-functional thermochemistry. III. The role of exact exchange, *J Chem Phys*, 1993, **98**, 5648–5652.
- 8 P. J. Stephens, F. J. Devlin, C. F. Chabalowski and M. J. Frisch, Ab Initio Calculation of Vibrational Absorption and Circular Dichroism Spectra Using Density Functional Force Fields, *J Phys Chem*, 1994, **98**, 11623–11627.
- 9 T. Lu and F. Chen, Multiwfn: A multifunctional wavefunction analyzer, *J Comput Chem*, 2012, **33**, 580–592.
- I. Seguy, P. Jolinat, P. Destruel, R. Mamy, H. Allouchi, C. Courseille, M. Cotrait and H. Bock, Crystal and Electronic Structure of a Fluorescent Columnar Liquid Crystalline Electron Transport Material, *ChemPhysChem*, 2001, 2, 448–452.