

## Supporting Information

### Design, Synthesis, and Inhibitory Activity of Hydroquinone Ester

#### Derivatives against Mushroom Tyrosinase

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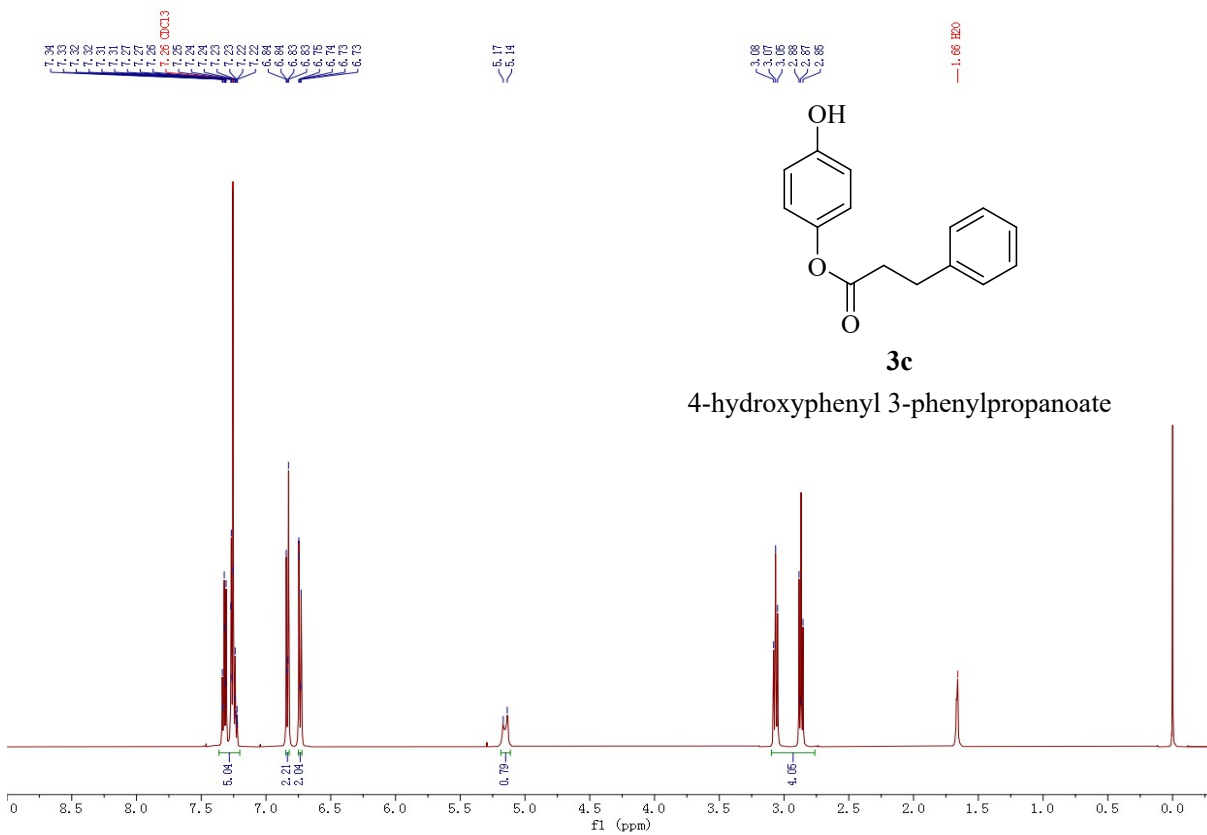
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Table S1. Linear fitting equation, Michaelis constant ( $K_m$ ), maximum reaction rate ( $V_m$ ), and inhibition type for mTyr at varying concentrations of compounds 3a-3e

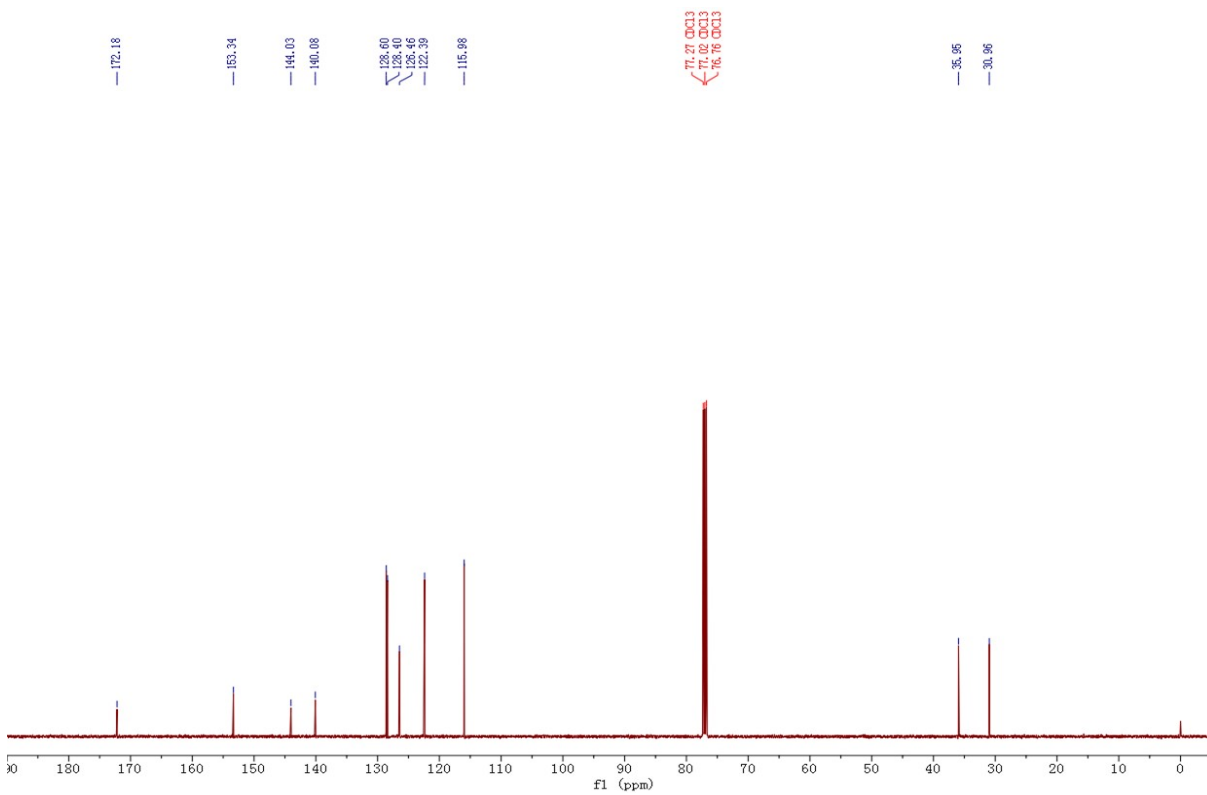
Table S2. Docking energy and bonding condition of compounds 3a-3e with mTyr





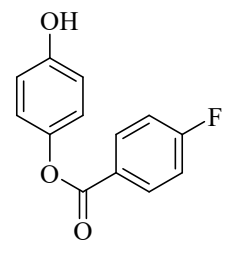
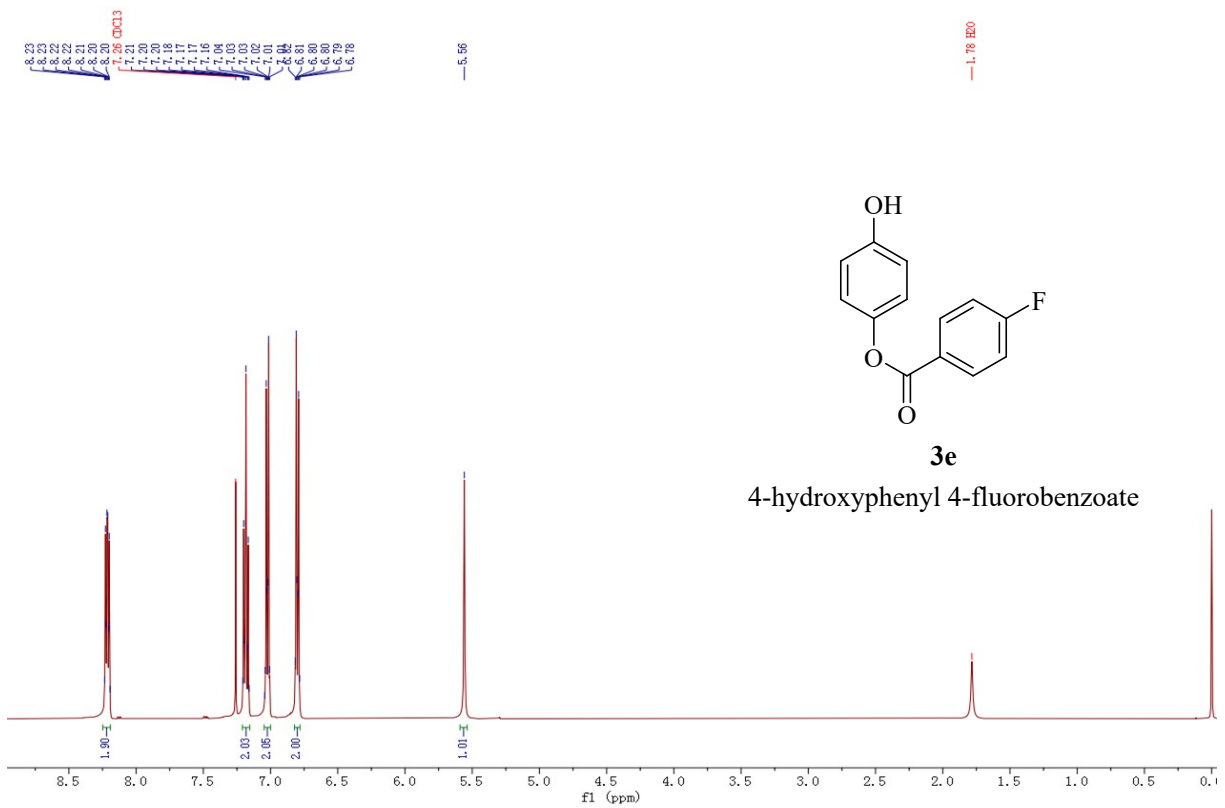


(5) <sup>1</sup>H NMR for compound 3c



(6) <sup>13</sup>C NMR for compound 3c

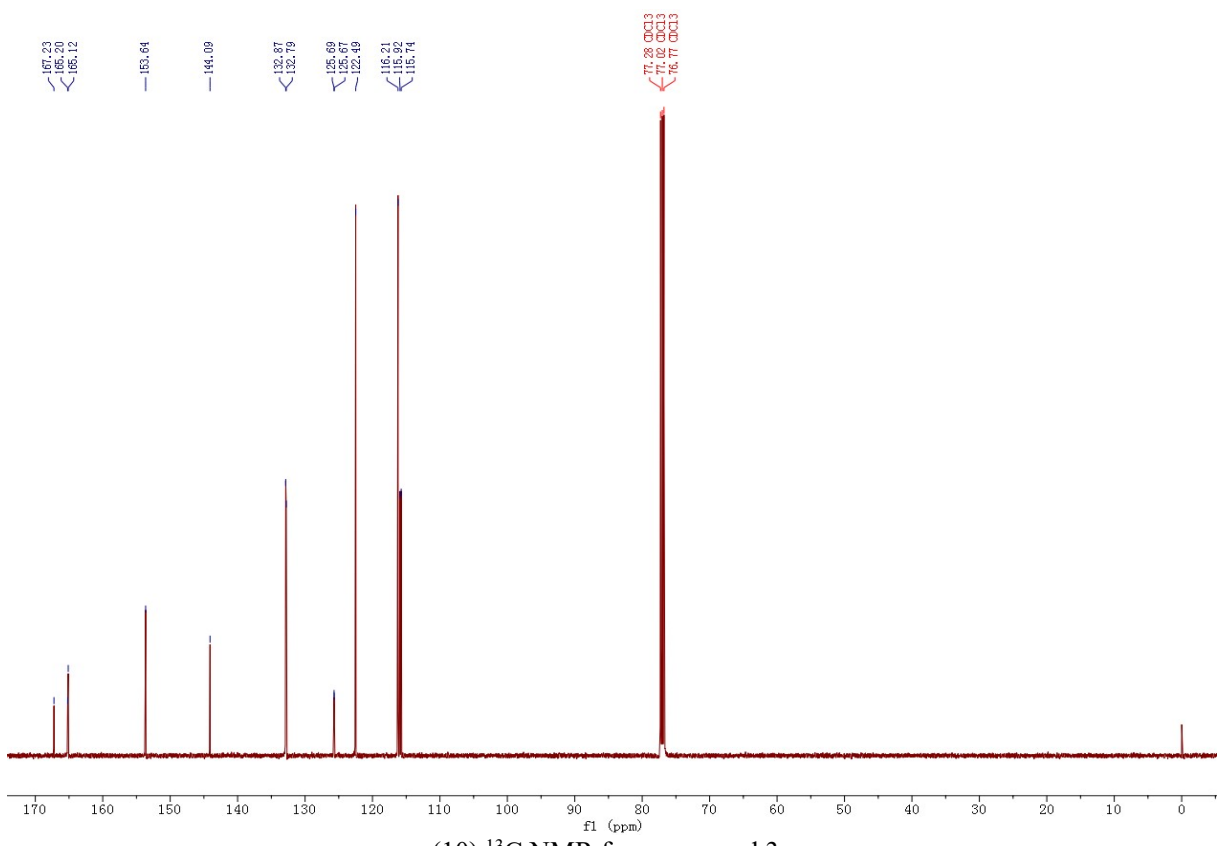




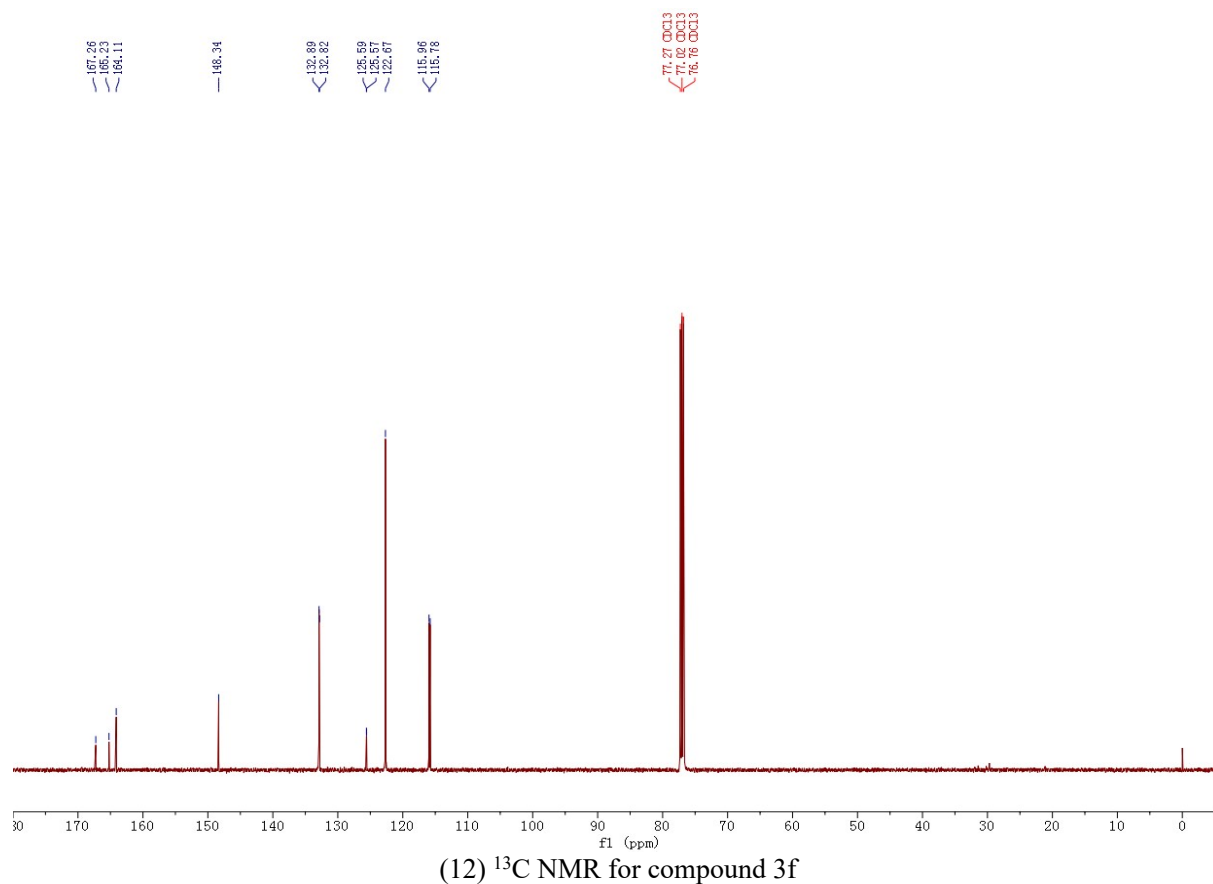
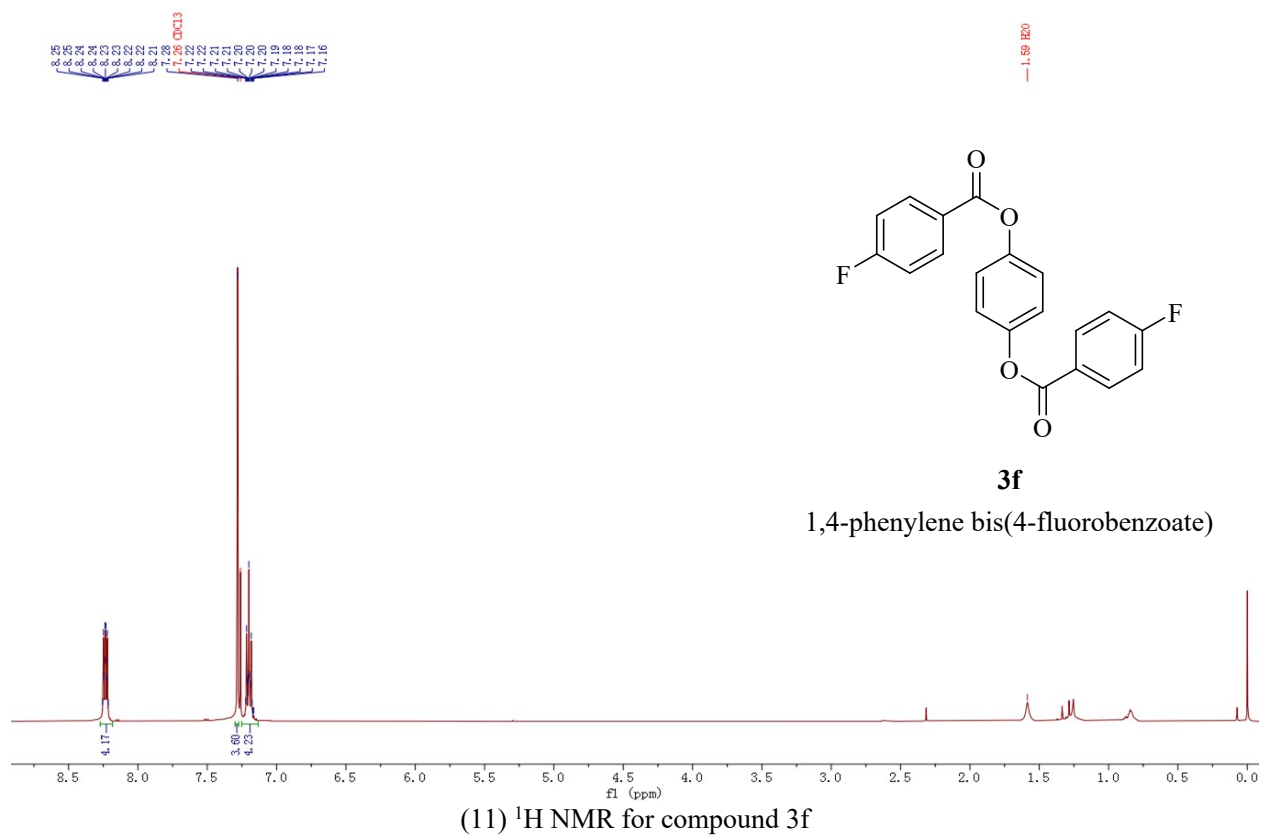
**3e**

4-hydroxyphenyl 4-fluorobenzoate

(9) <sup>1</sup>H NMR for compound 3e



(10) <sup>13</sup>C NMR for compound 3e





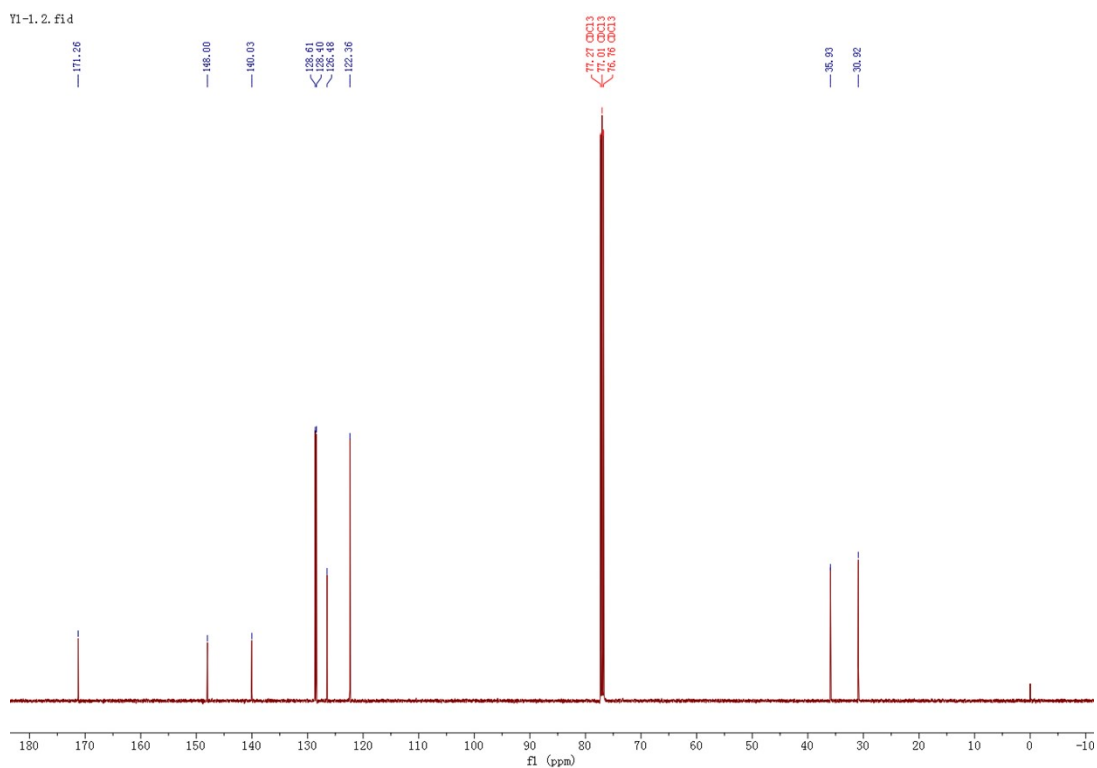
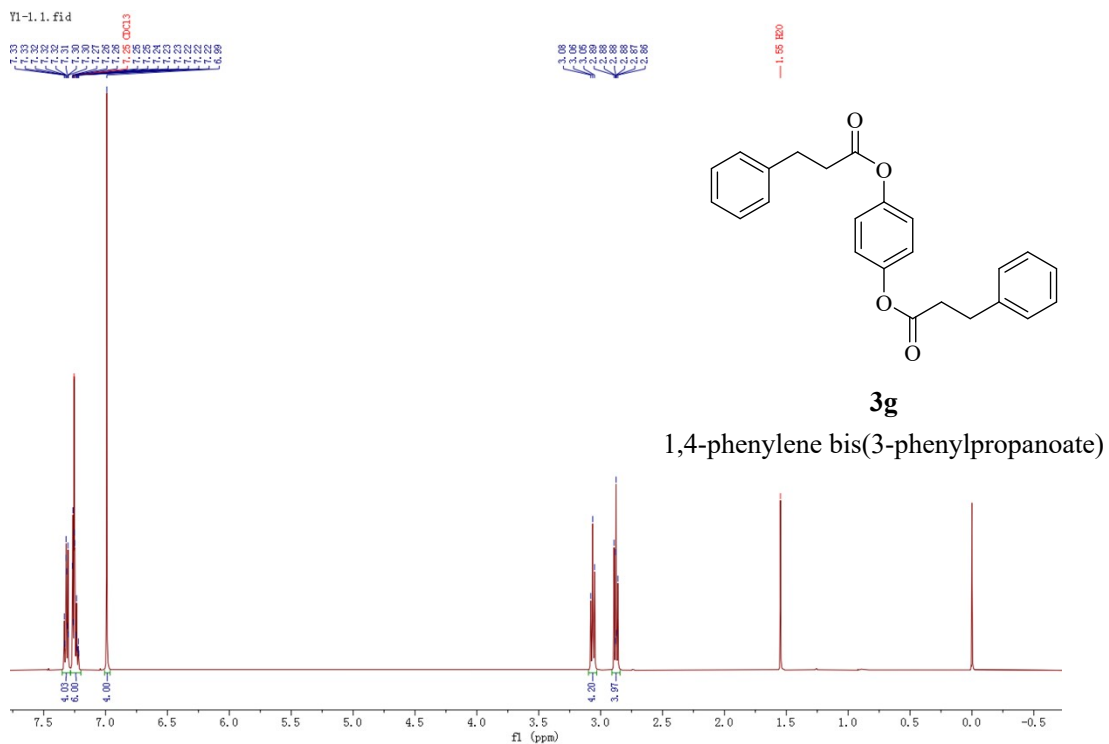
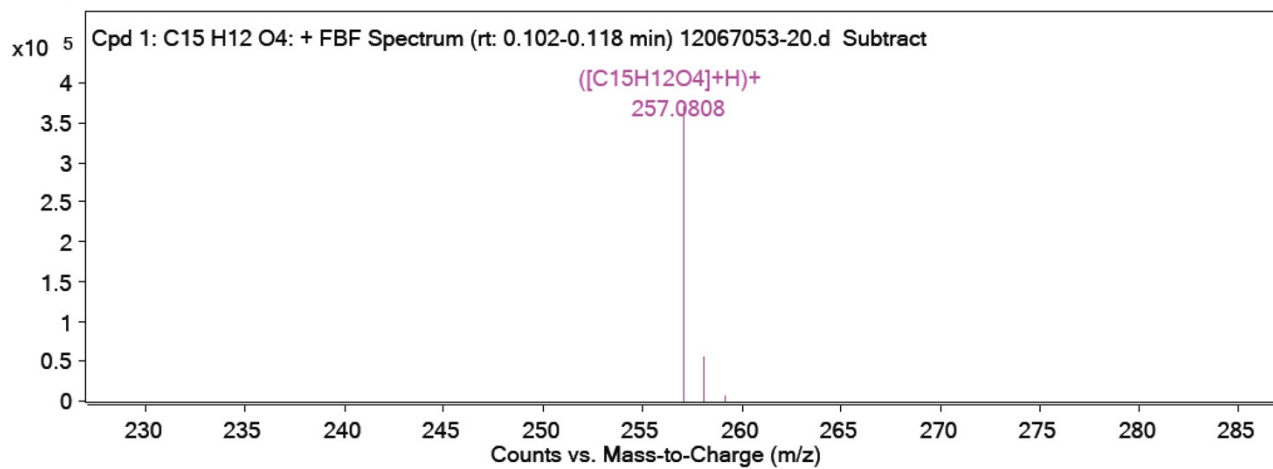
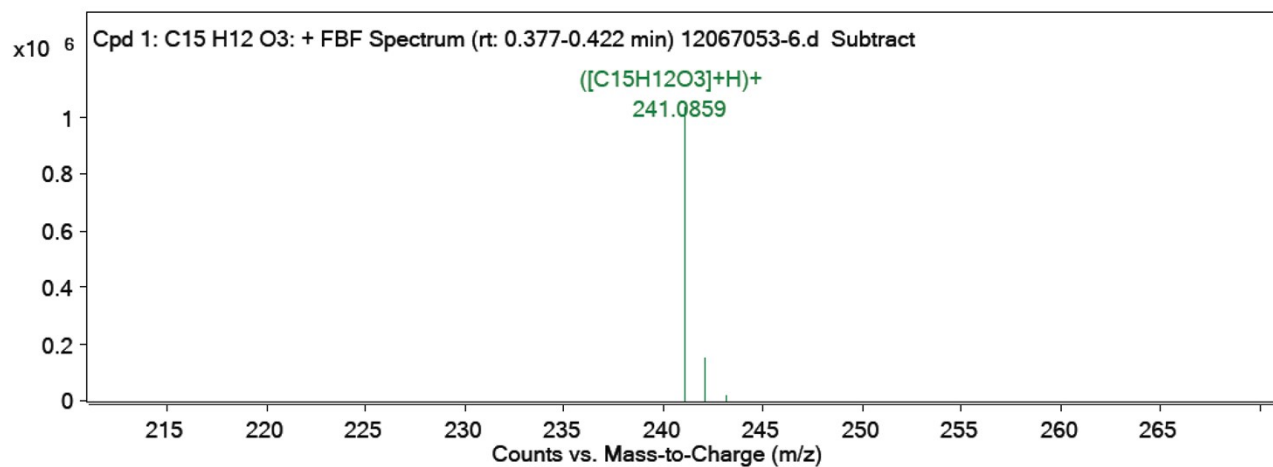


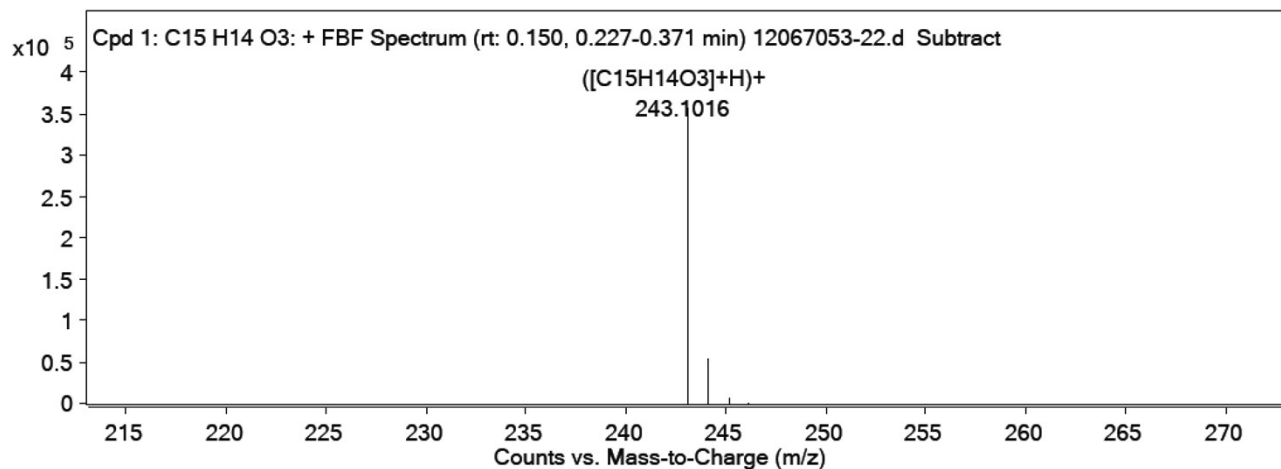
Figure S1. <sup>1</sup>H NMR and <sup>13</sup>C NMR for compound 3a-3g



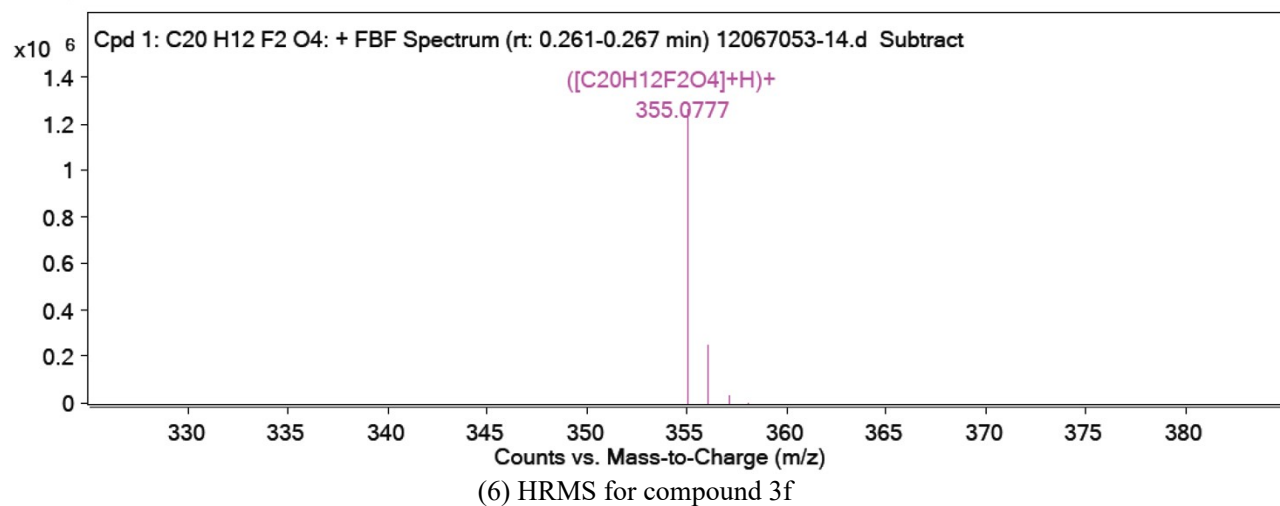
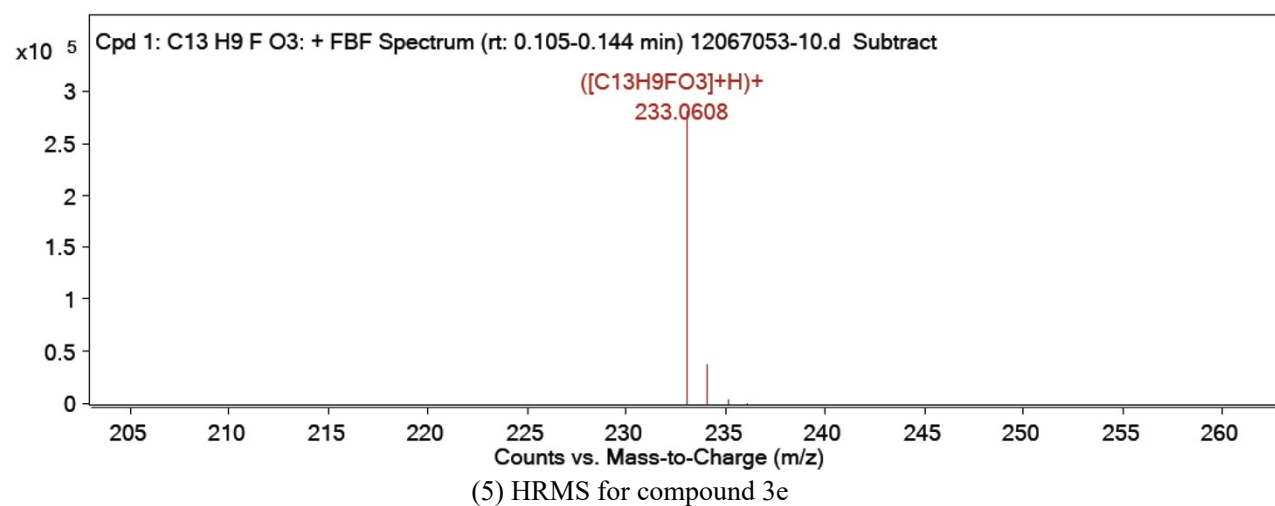
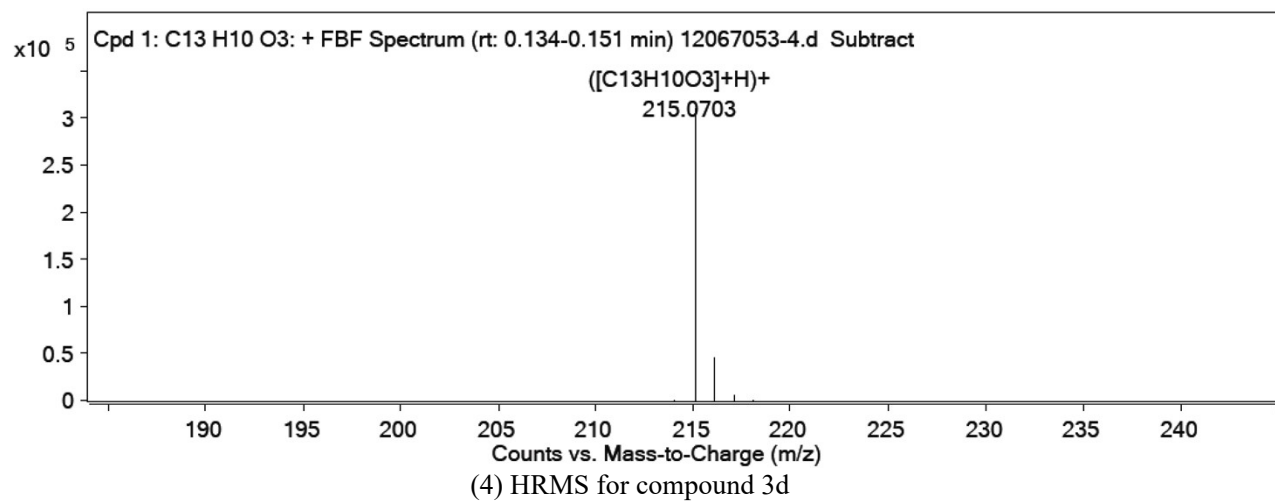
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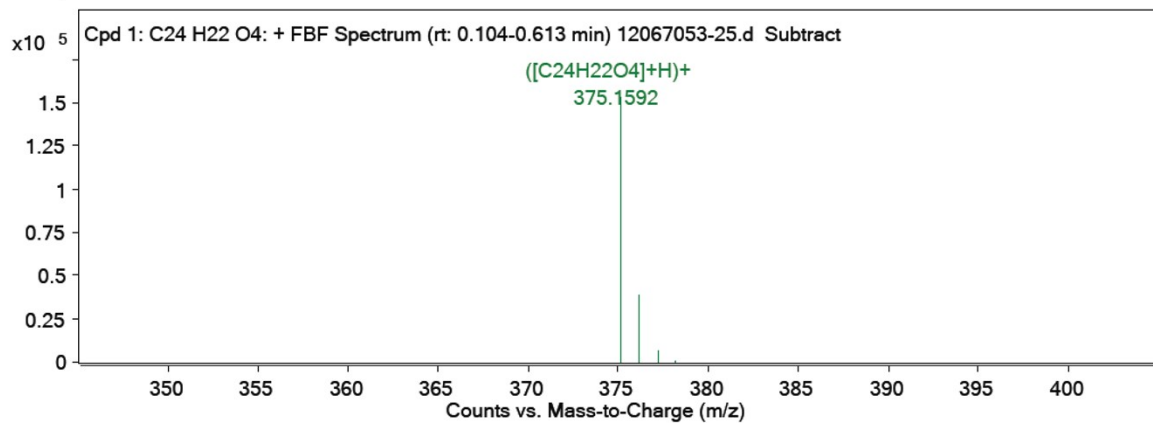


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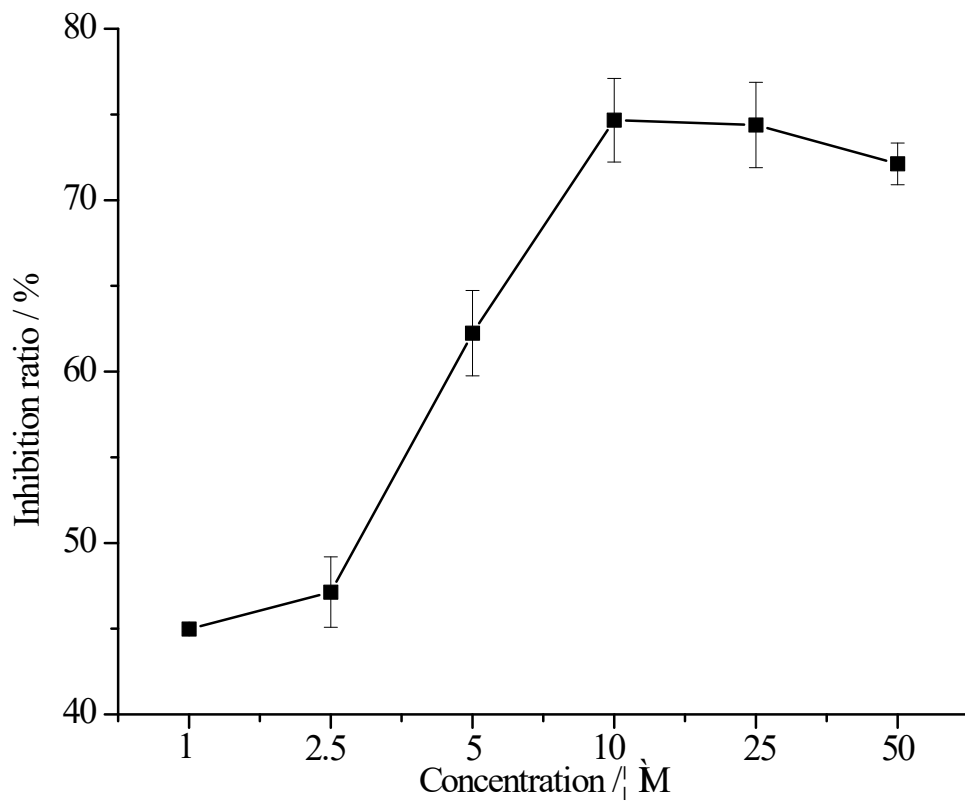
(3) HRMS for compound 3c



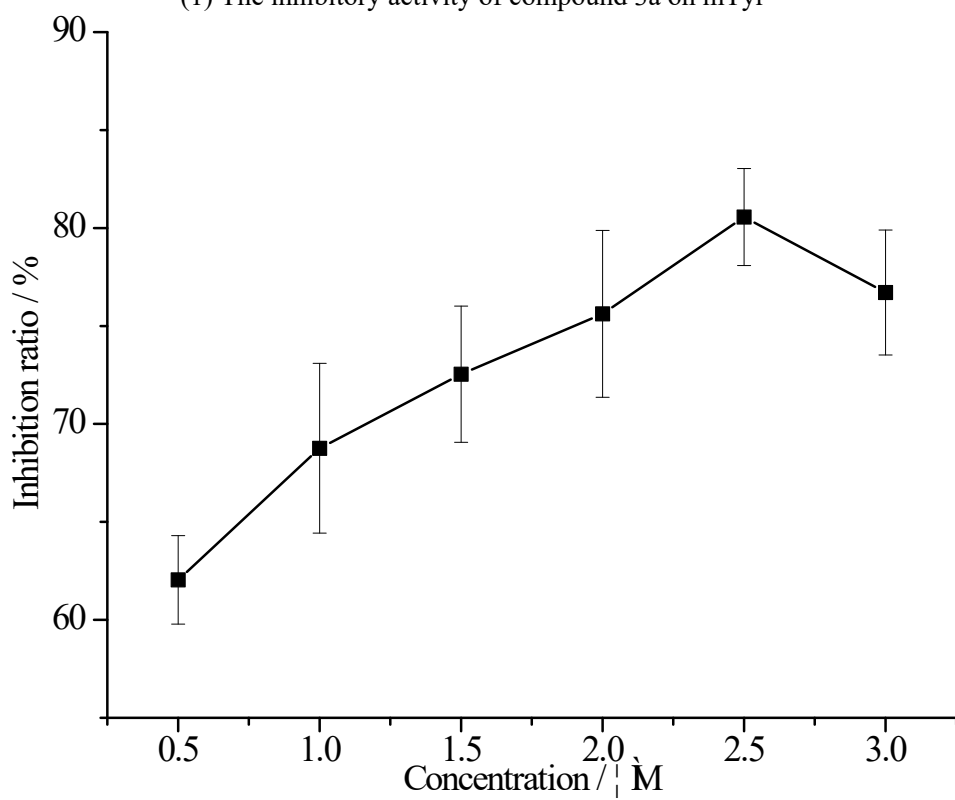


(7) HRMS for compound 3g

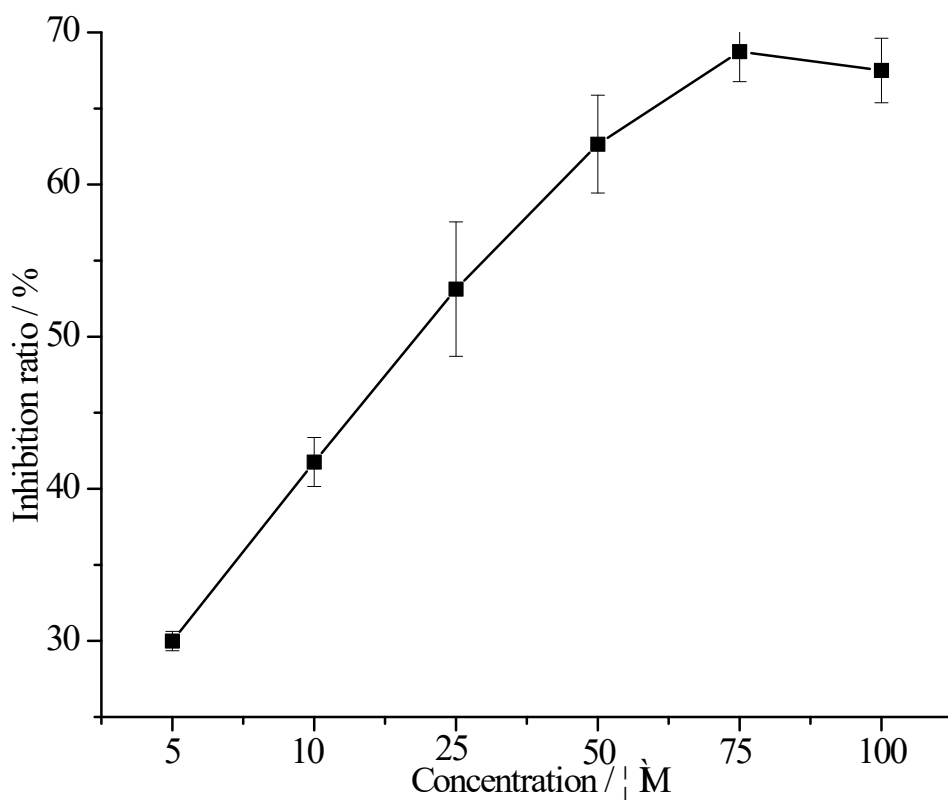
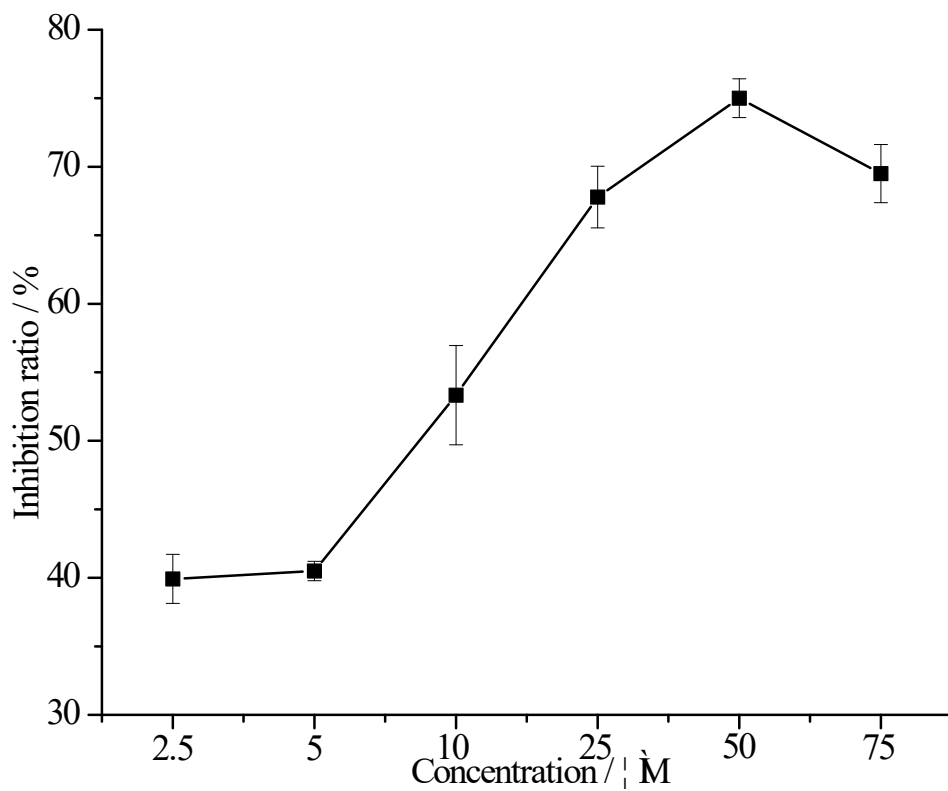
Figure S2. HRMS for compounds 3a-3g

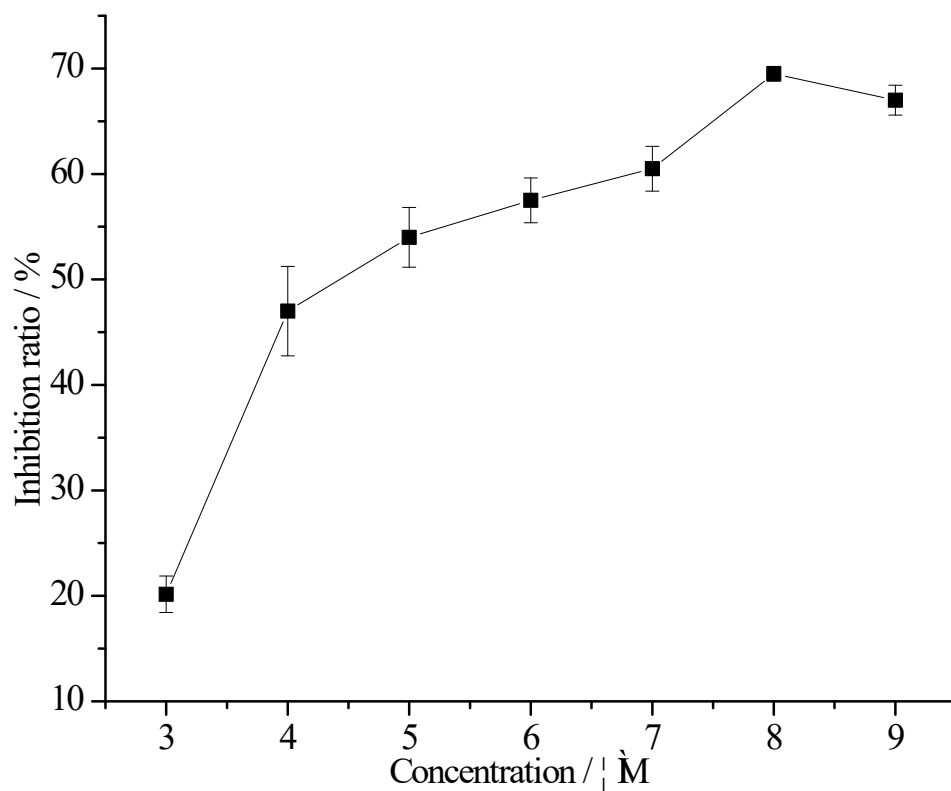


(1) The inhibitory activity of compound 3a on mTyr

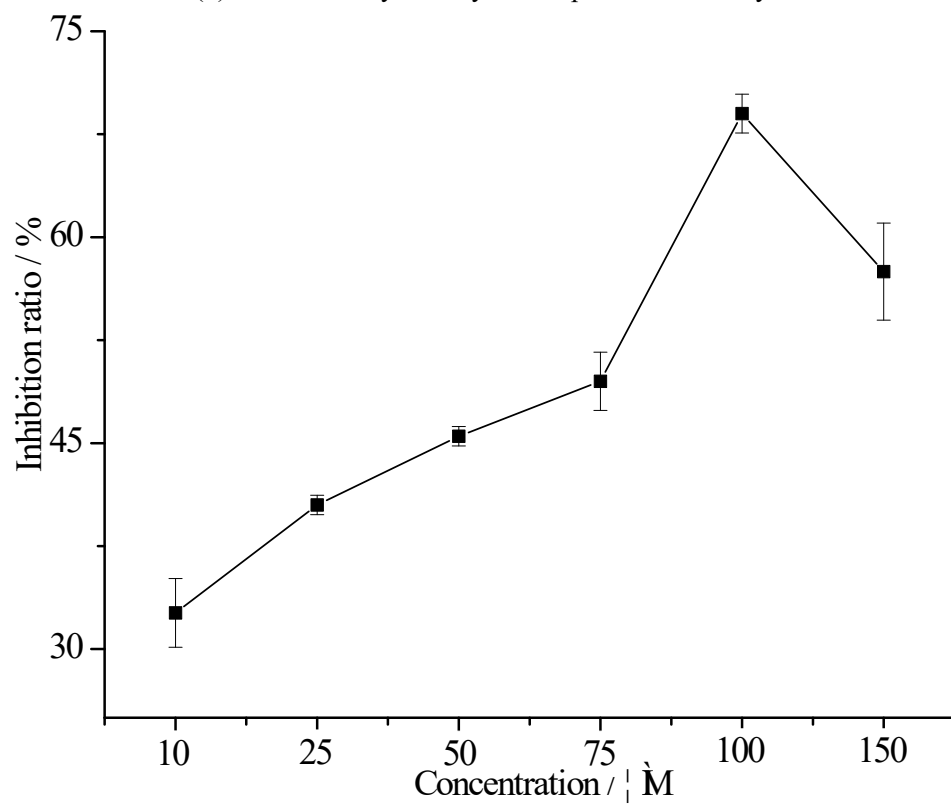


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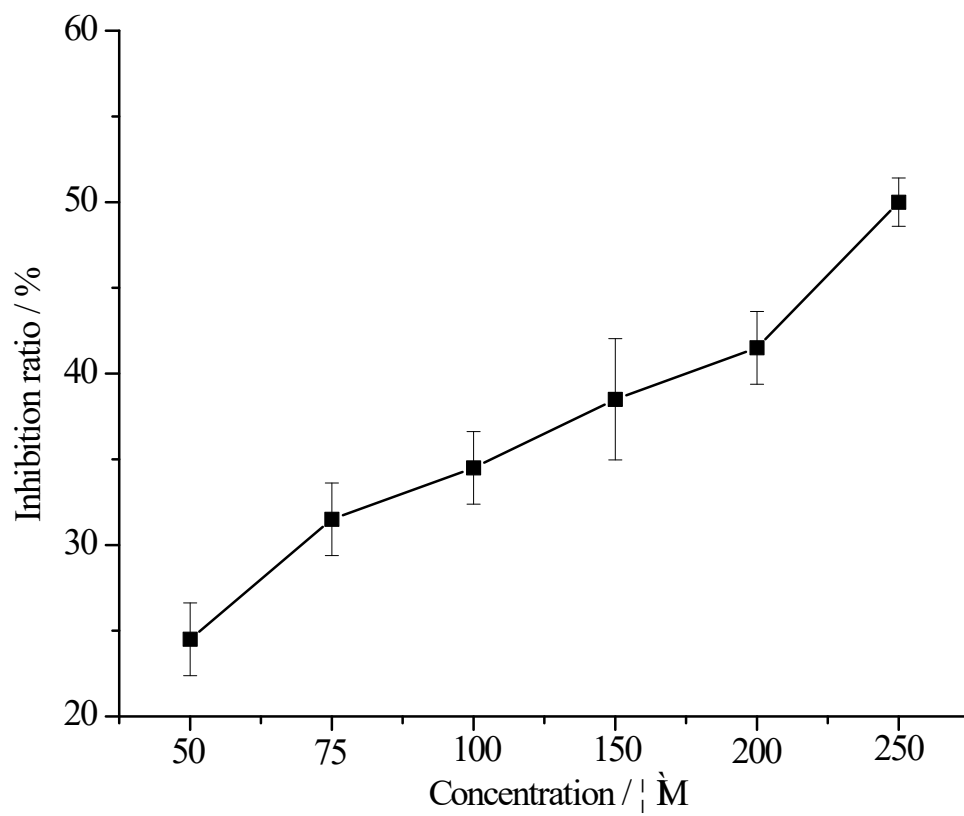




(5) The inhibitory activity of compound 3e on mTyr



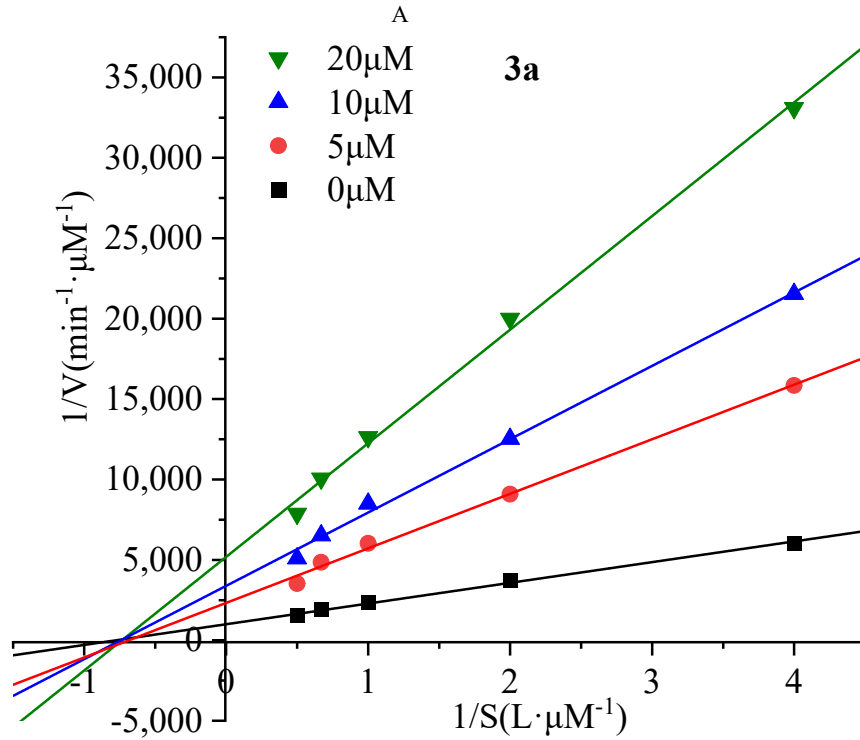
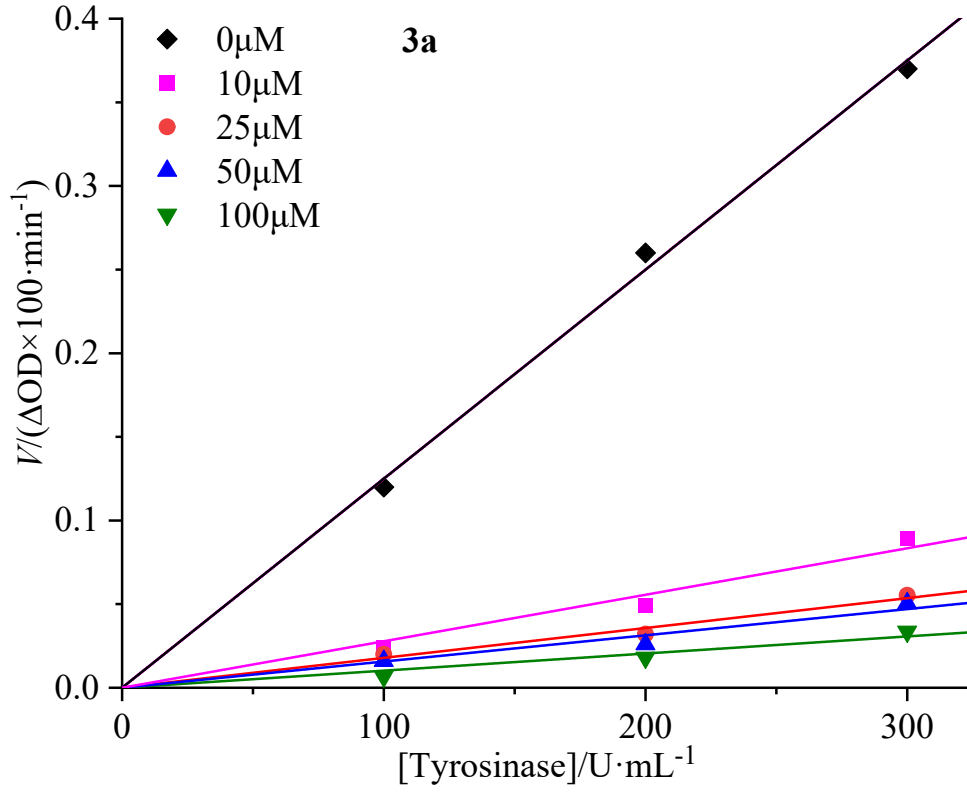
(6) The inhibitory activity of compound 3f on mTyr



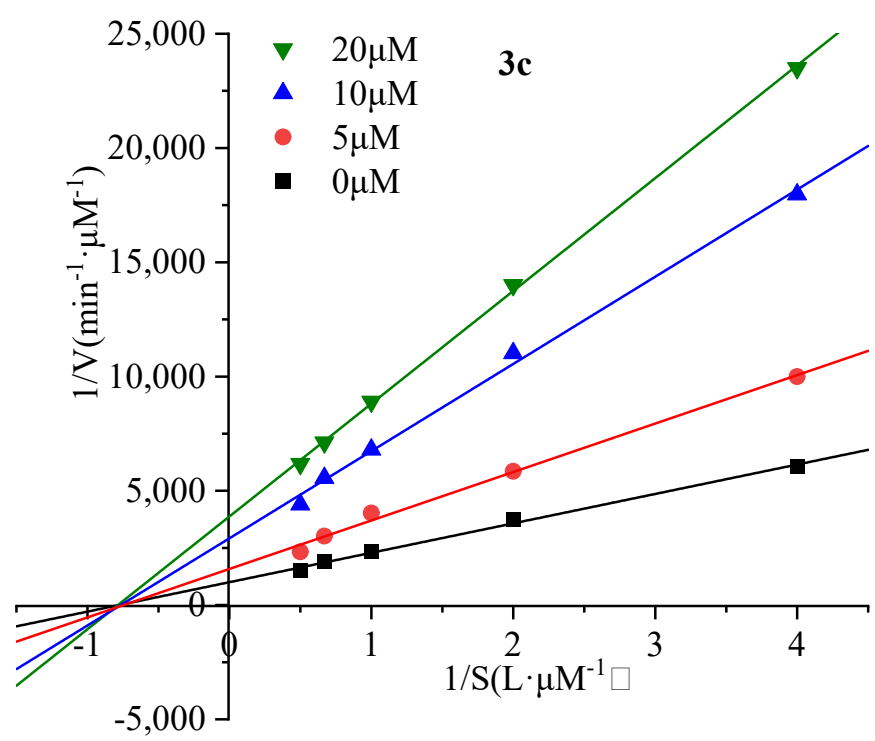
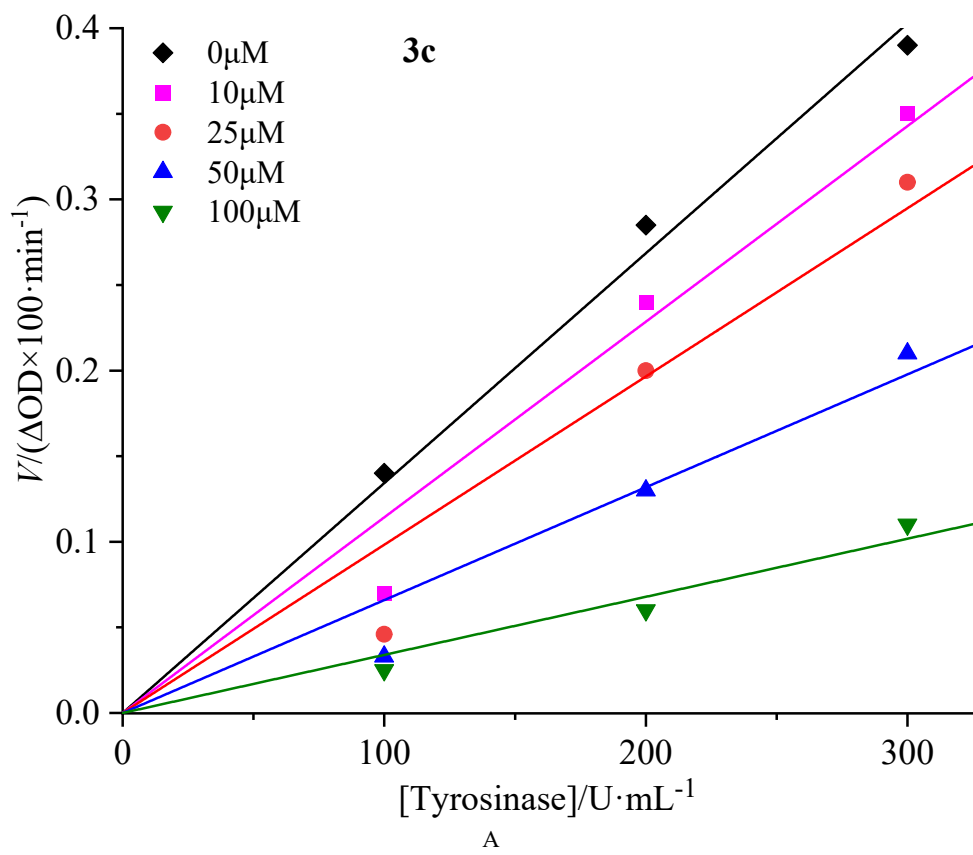
(7) The inhibitory activity of compound 3g on mTyr

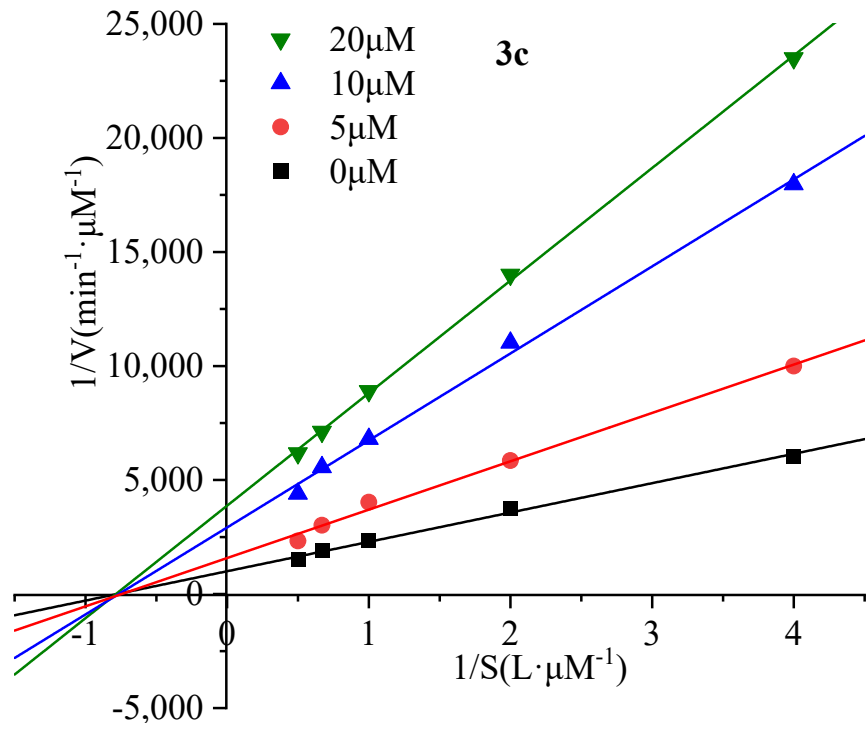
Figure S3. The inhibitory activity of compounds 3a-3g on mTyr





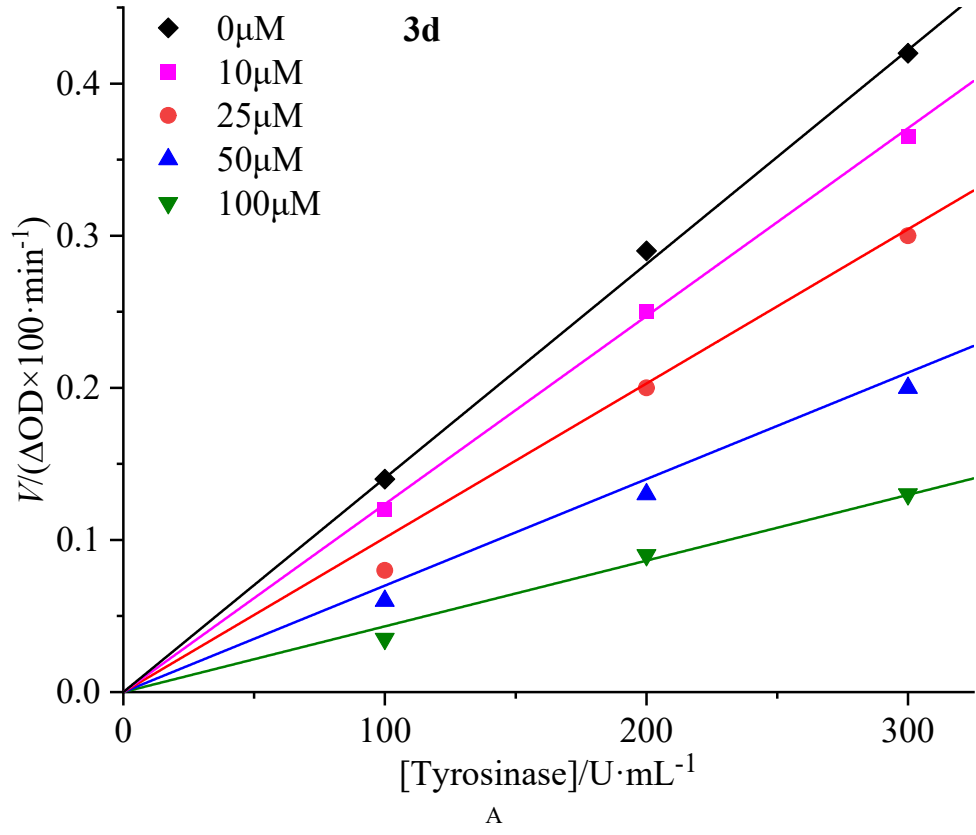
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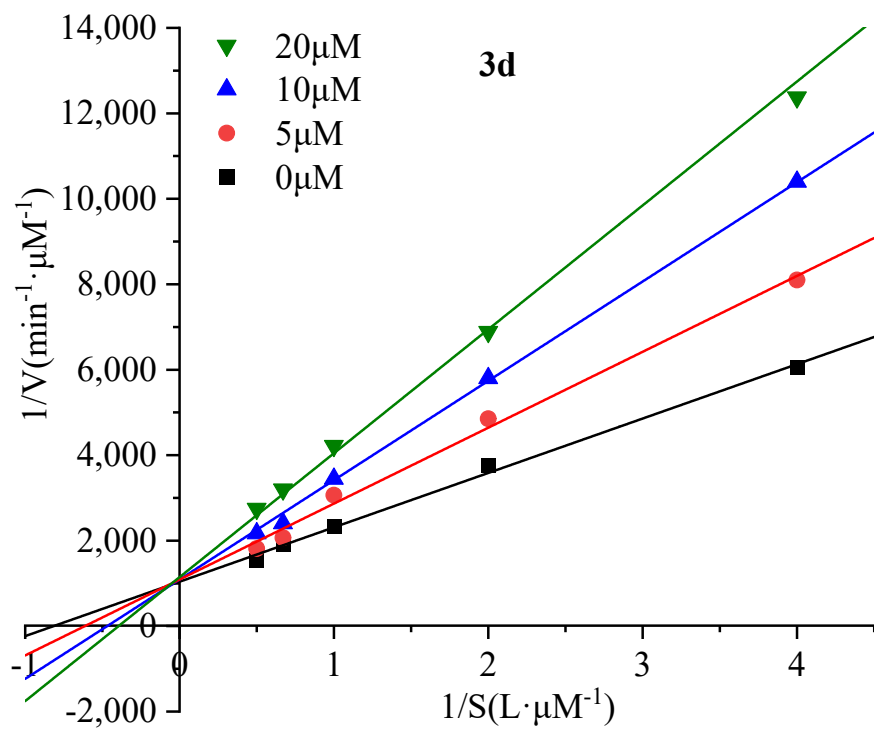
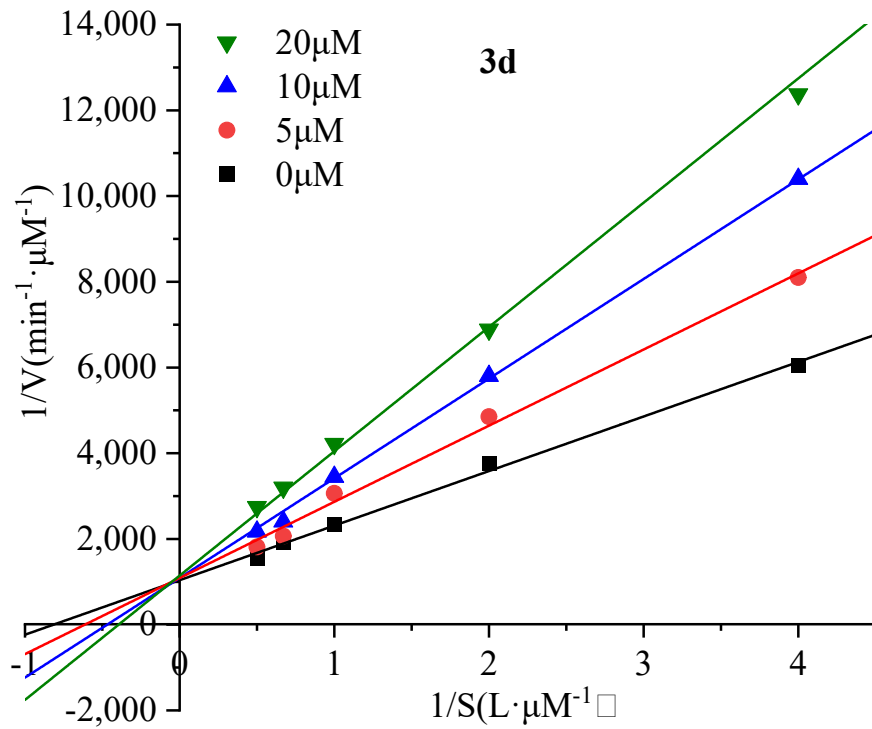


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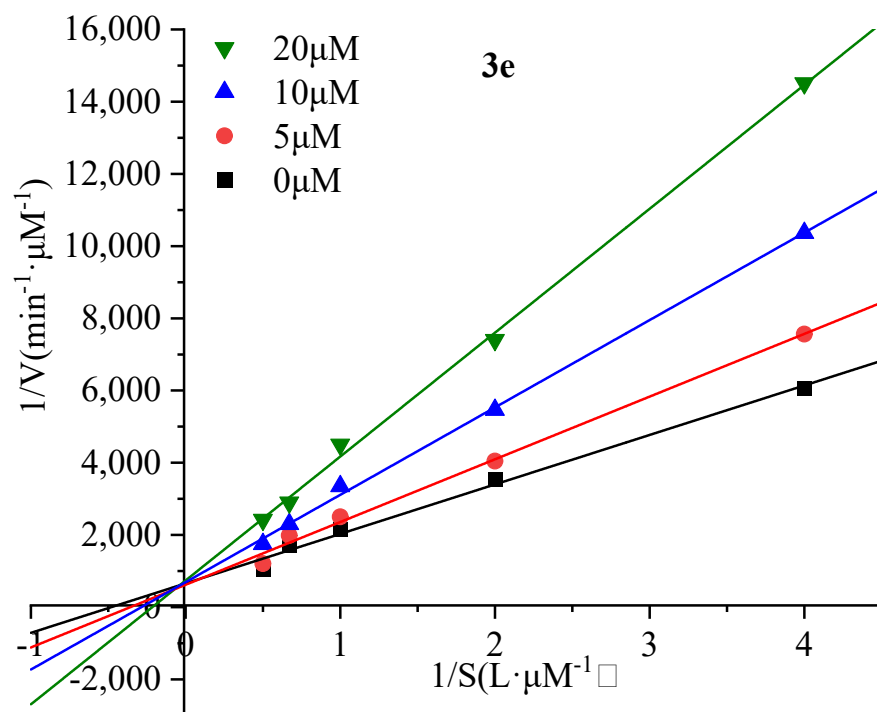
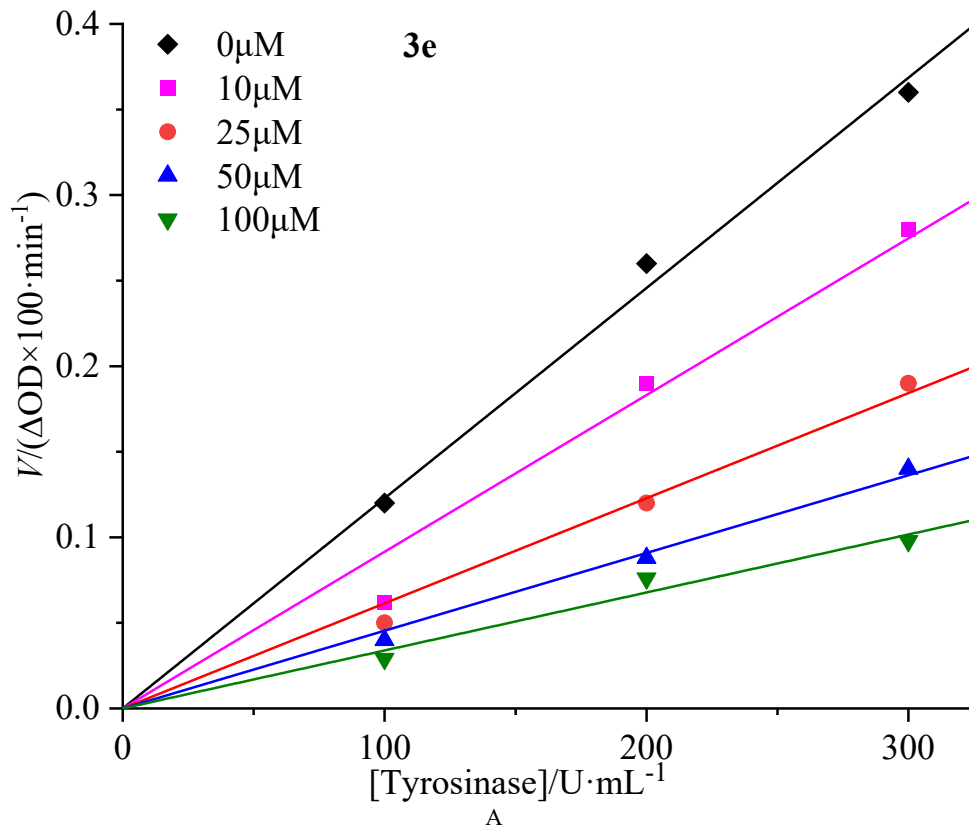
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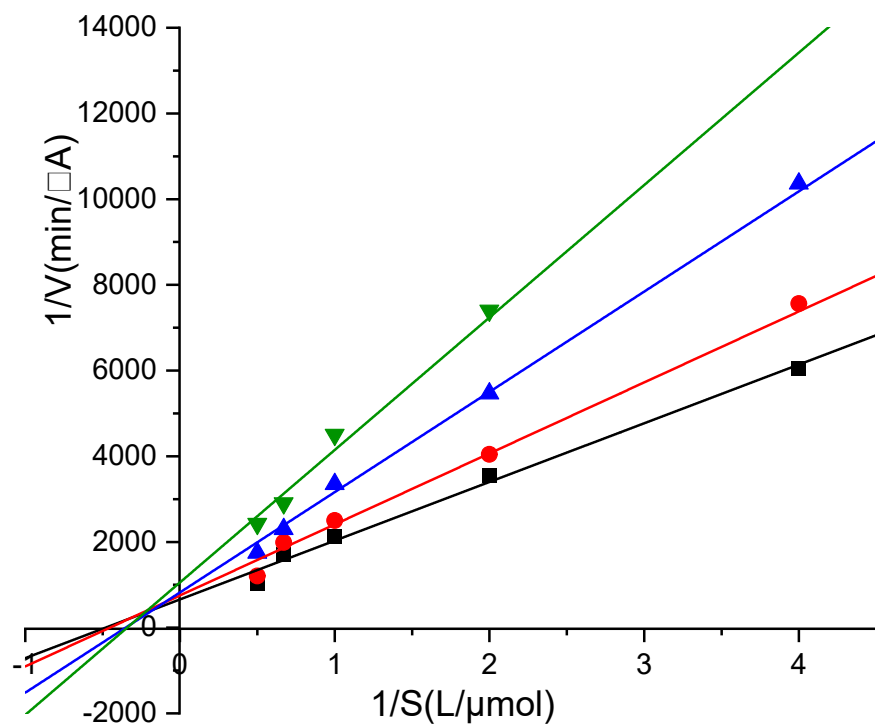


A



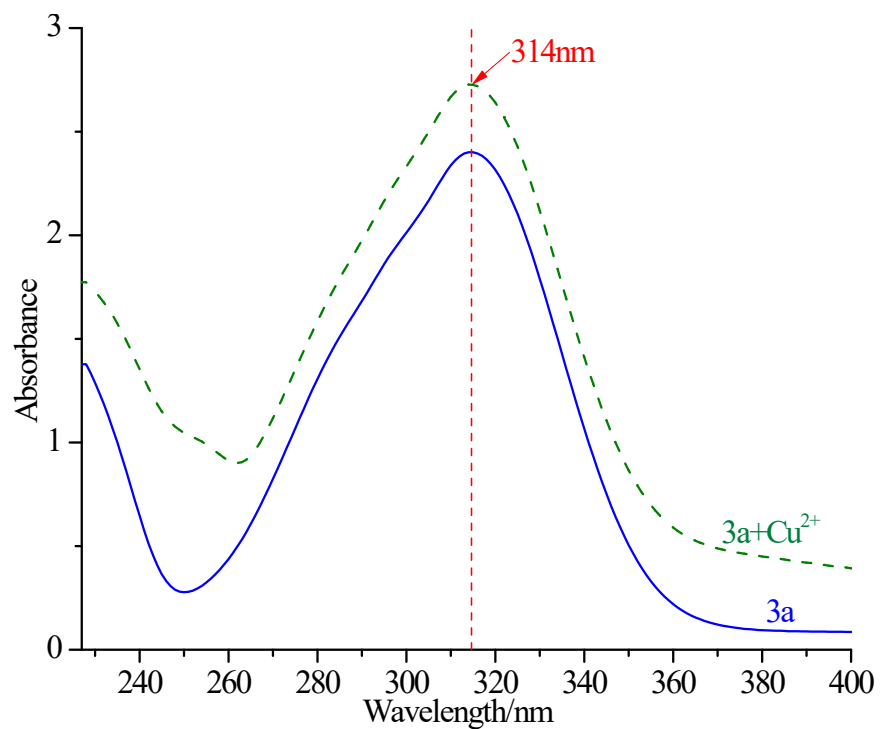
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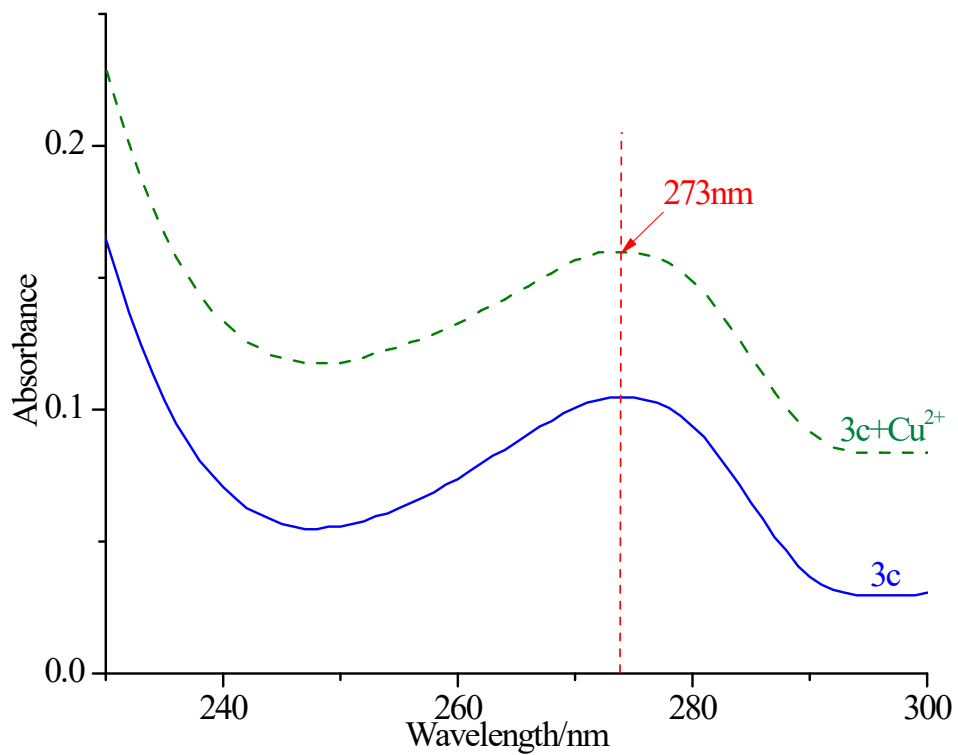


B  
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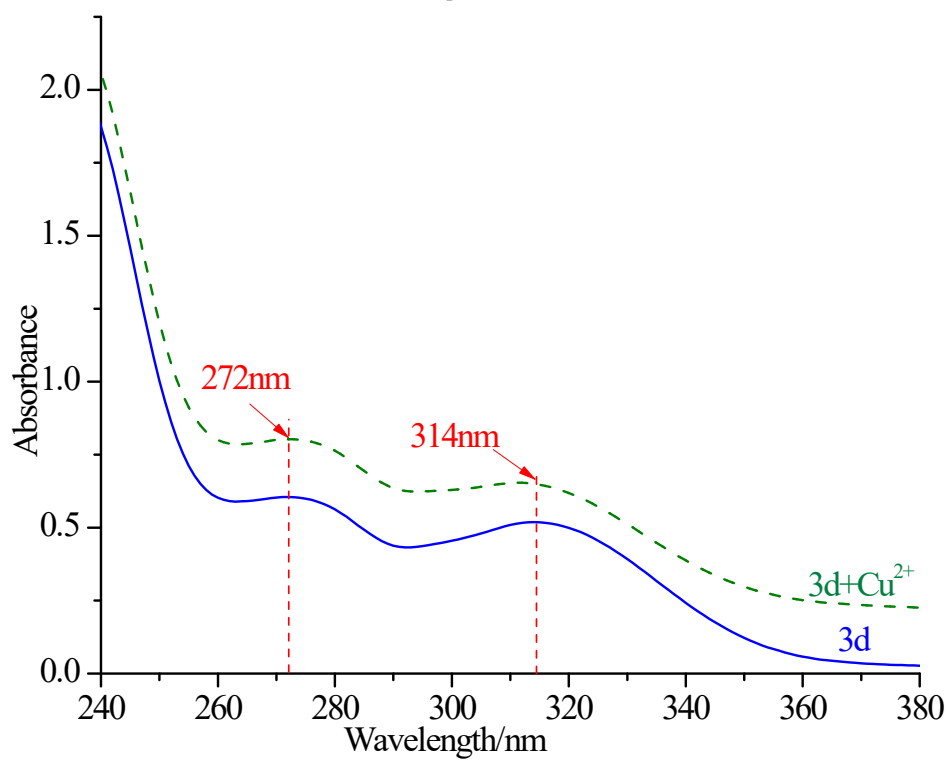
Figure S4 Inhibition reversibility (A) and inhibition type (B) of compound 3a, 3c-3e on mTyr.



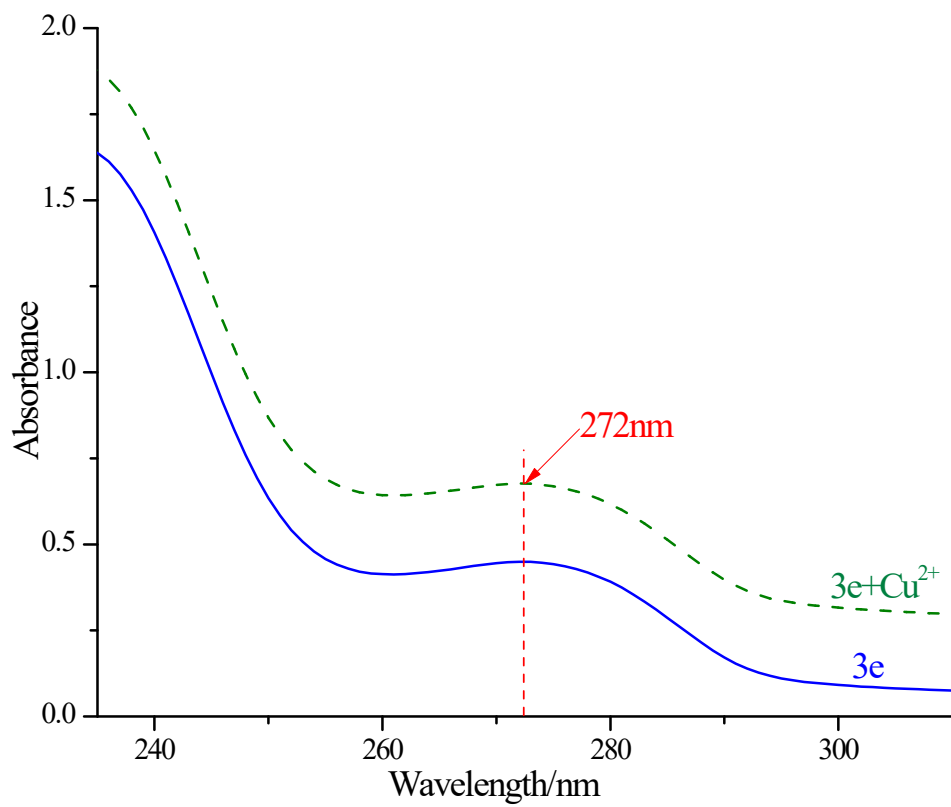
Compound 3a



Compound 3c



Compound 3d



Compound 3e

Figure S5. UV Spectra of compound 3a, 3c-3e before and after interaction with mTyr



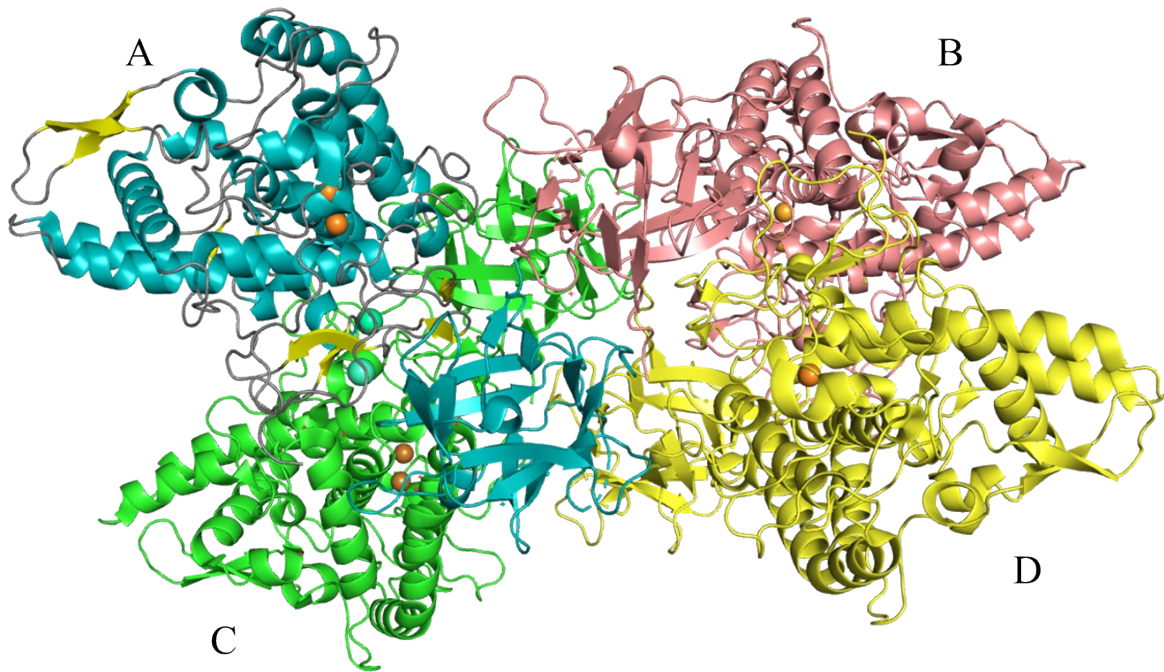
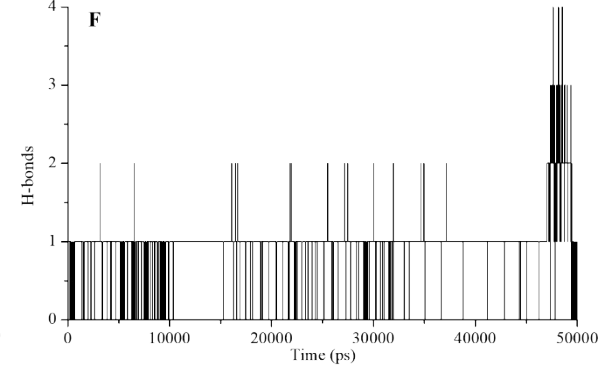
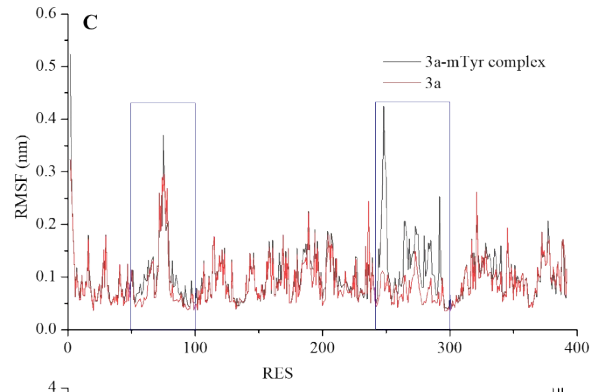
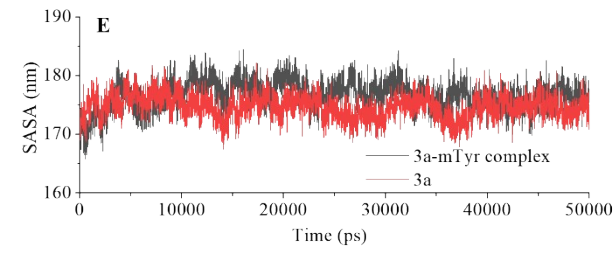
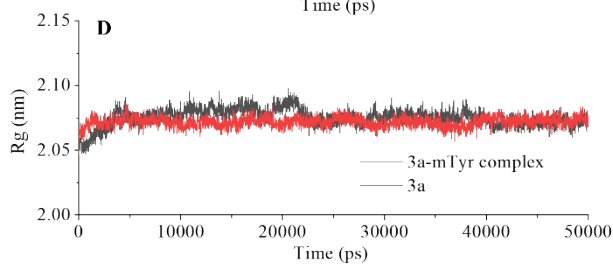
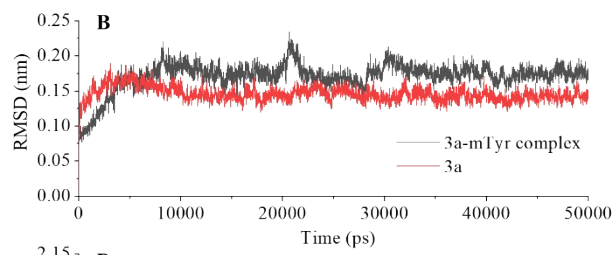
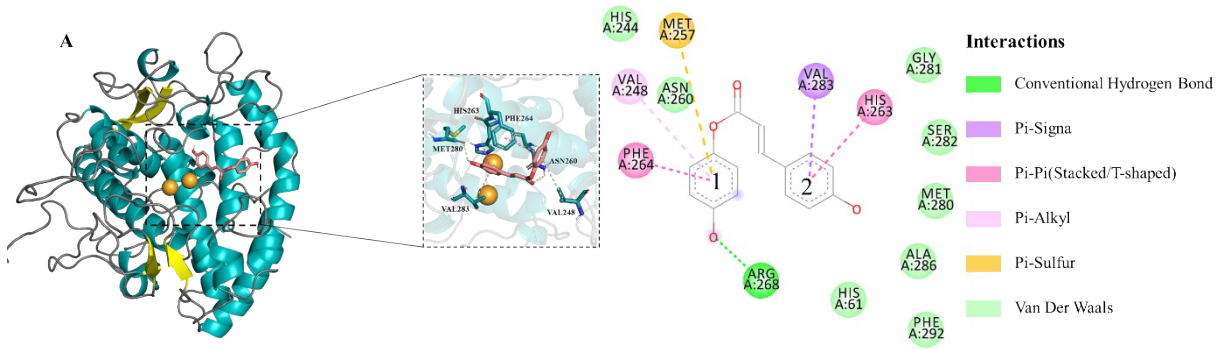
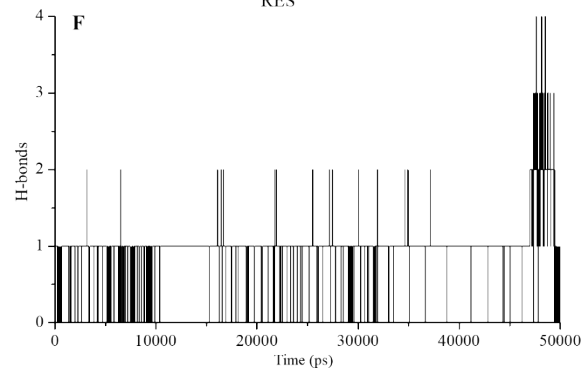
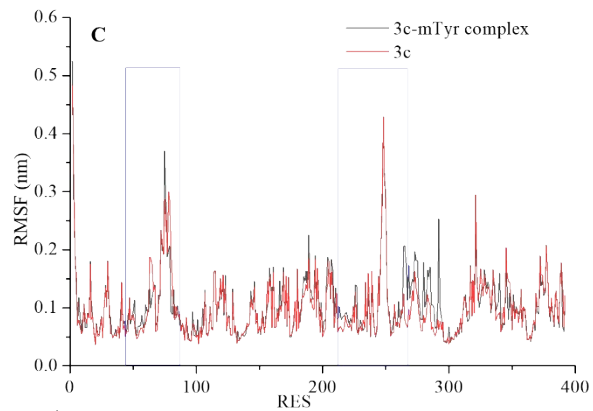
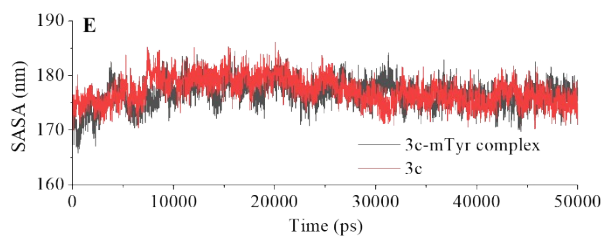
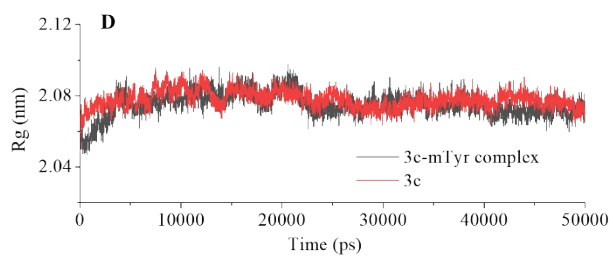
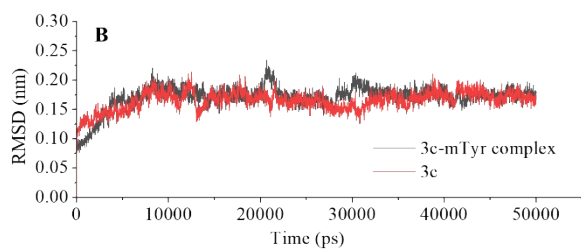
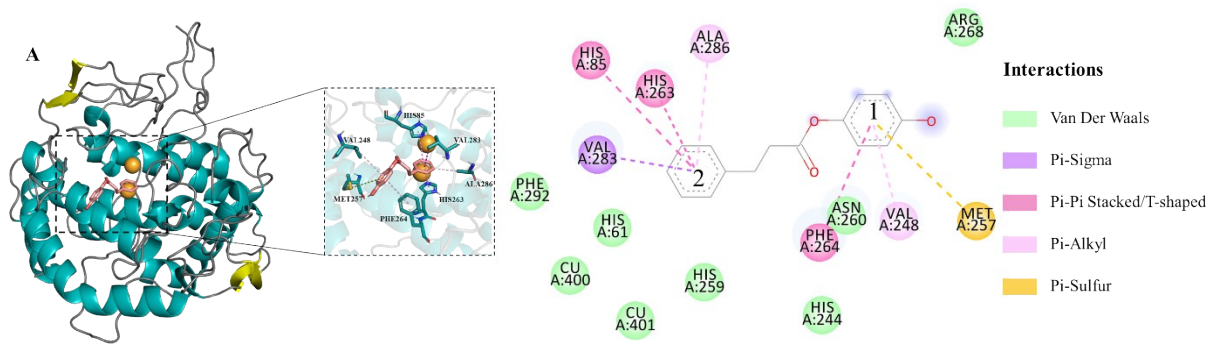


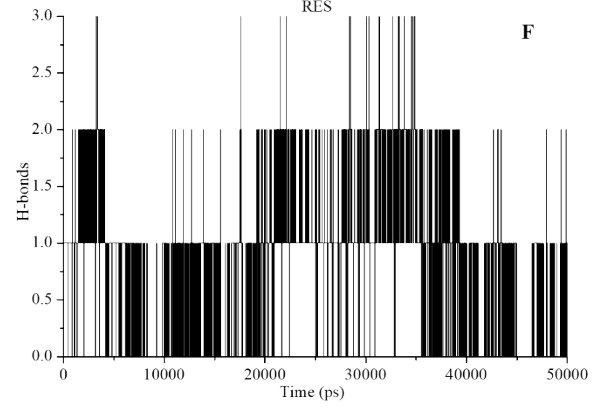
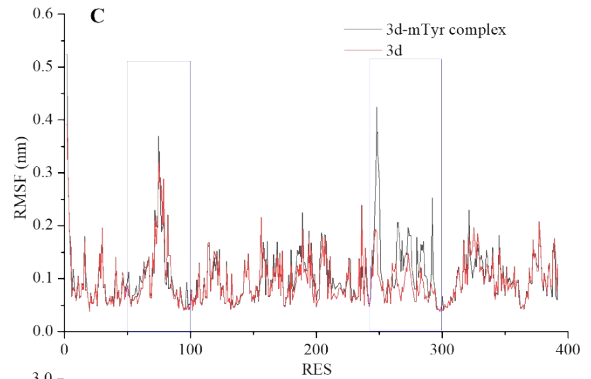
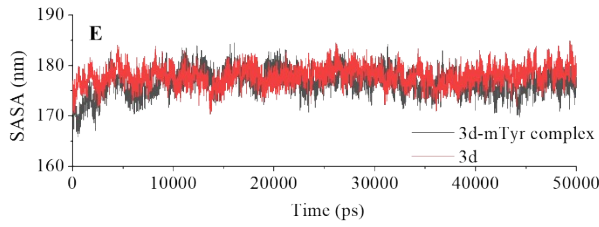
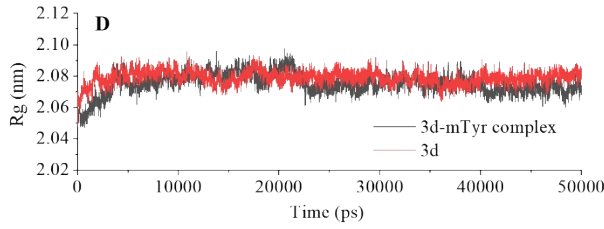
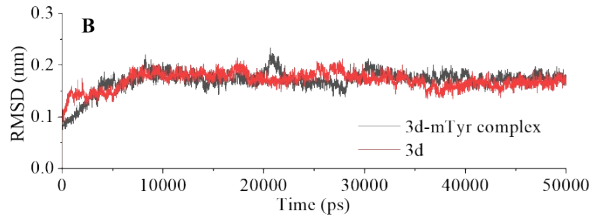
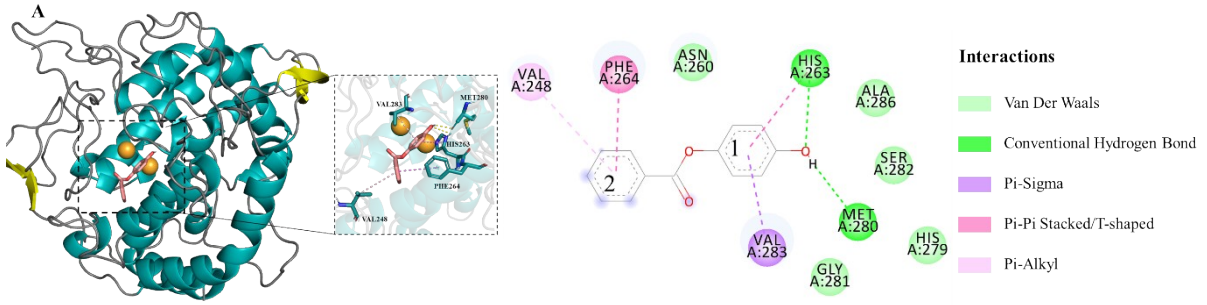
Figure S6. The crystal structure of mTyr (PDB ID: 2Y9X)



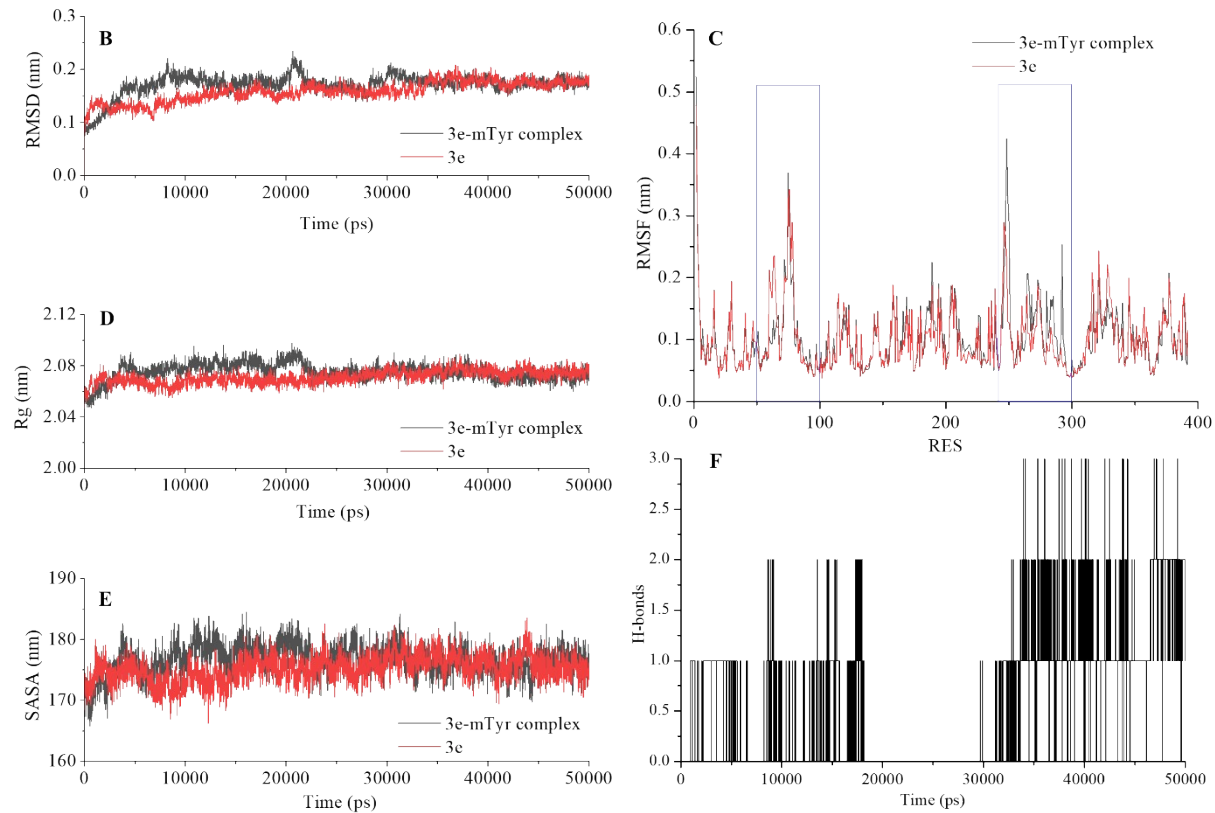
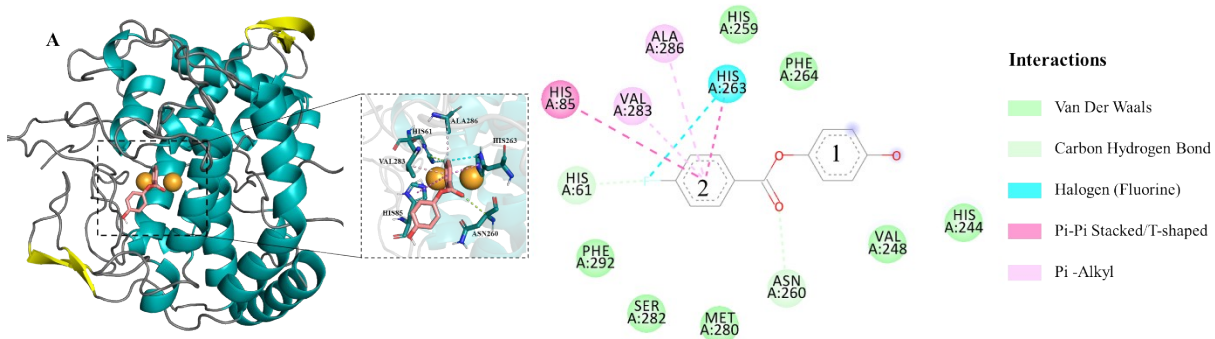
Compound 3a



Compound 3c



Compound 3d



Compound 3e

Figure S7. Docking model for compound 3a, 3c-3e with mTyrosinase (A) and Molecular dynamics results of compound 3a, 3c-3e and (3a, 3c-3e) -mTyrosinase complex with: B. RMSD, C. RMSF, D. Rg, E. SASA, and F. H-bonds

Table S1. Linear fitting equation, Michaelis constant ( $K_m$ ), maximum reaction rate ( $V_m$ ), and inhibition type for mTyr at varying concentrations of compounds 3a-3e

Compound	Concentration ( $\mu\text{M}$ )	Linear fitting equation	$R^2$	$K_m$ ( $\mu\text{M}$ )	$V_m$ ( $\times 10^3, \text{min}^{-1}$ )	Inhibition type
3a	0	$Y=1000.25+1287.41X$	0.9991	1.2871	0.9998	Noncompetitive
	5	$Y=2340.91+3012.40X$	0.9958	1.2870	0.4272	
	10	$Y=3516.19+4525.44X$	0.9963	1.2867	0.2843	
	20	$Y=5446.87+7064.94X$	0.9965	1.2869	0.1836	
	0	$Y=333.72+1645.85X$	0.9902	4.9318	2.9965	
3b	1	$Y=1320.62+2456.41X$	0.9949	2.6173	0.7572	Mix-noncompetitive
	2	$Y=2889.49+3833.45X$	0.9964	1.3267	0.3461	
	4	$Y=4662.26+5183.79X$	0.9982	1.1119	0.2145	
	0	$Y=1000.25+1287.41X$	0.9991	1.2871	0.9998	
3c	5	$Y=1580.45+2034.21X$	0.9990	1.2870	0.6327	Noncompetitive
	10	$Y=2920.18+3759.70X$	0.9991	1.2873	0.3424	
	20	$Y=3875.12+4950.54X$	0.9999	1.2774	0.2581	
	0	$Y=1084.97+1272.89X$	0.9961	1.1732	0.9217	
3d	5	$Y=1085.12+1776.46X$	0.9985	1.6371	0.9216	Competitive
	10	$Y=1086.06+2324.89X$	0.9996	2.1407	0.9208	
	20	$Y=1085.95+2900.15X$	0.9967	2.6706	0.9209	
	0	$Y=679.35+1370.31X$	0.9904	2.0171	1.472	
3e	5	$Y=678.55+1736.97X$	0.9943	2.5598	1.474	Competitive
	10	$Y=680.49+2419.72X$	0.9983	3.5558	1.470	
	20	$Y=681.12+3431.14X$	0.9983	5.0375	1.468	
	0	$Y=679.35+1370.31X$	0.9904	2.0171	1.472	

Table S2 Docking energy and bonding condition of compounds 3a-3e with mTyr

Compounds	Docking energy (kcal·mol <sup>-1</sup> )	Bonding condition			
		Hydrogen Bonds	Hydrophobic	Miscellaneous (Pi-Sulfur)	Halogen (Fluorine)
3a	-7.6	Ph1O←Arg 268 (6.66 Å, Conventional Hydrogen Bond)	Ph1←Phe264 (6.68Å, Pi-Pi Stacked/T-shaped); Ph1←Val248 (5.16Å, Pi-alkyl); Ph2←His263 (5.57Å, Pi-Pi Stacked/T-shaped); Ph2←Val283 (5.11Å, Pi-Sigma).	Ph1←Met257 (6.50 Å)	
3b	-7.2	Ph1OH←Met280(5.01 Å, Conventional Hydrogen Bond)	Ph1←His263 (5.32Å, Pi-Pi Stacked); Ph1←Val283 (4.89Å, Pi-Sigma); Ph2←Val248 (5.31 Å, Pi-Sigma).	Ph2←Met257 (6.57 Å)	
3c	-6.7		Ph1←Val248 (5.57Å, Pi-Alkyl); Ph1←Phe264 (6.74Å, Pi-Pi Stacked/T-shaped); Ph2←Val283 (4.26Å, Pi-Sigma); Ph2←His85/His263 (5.47/5.48Å, Pi-Pi Stacked/T-shaped); Ph2←Ala 286 (6.53Å, Pi-Alkyl).	Ph1←Met257(7.58 Å)	
3d	-7.0	Ph1OH←Met280(4.88 Å, Conventional Hydrogen Bond); Ph1O←His263 (4.06 Å, Conventional Hydrogen Bond)	Ph1←His 263 (5.52Å, Pi-Pi Stacked/T-shaped); Ph1←Val283 (4.82Å, Pi-Sigma); Ph2←Val248 (6.29Å, Pi-Alkyl); Ph2←Phe264 (6.62Å, Pi-Pi Stacked/T-shaped).		
3e	-6.7	Ph2F←His61 (5.50 Å, Carbon Hydrogen Bond)	Ph2←Val283 (4.42Å, Pi-Sigma); Ph2←Val283/Ala286 (4.63/6.95Å, Pi-Alkyl); Ph2←His85/His263 (7.42/4.57Å, Pi-Pi Stacked/T-shaped).		Ph2F←His 263 (5.41 Å Halogen (Fluorine))