# STEREOCHEMICAL INSIGHTS IN $\beta$-AMINO-NACYLHYDRAZONES AND THEIR IMPACT ON DPP-4 INHIBITION 

Eduardo Reina, ${ }^{\mathrm{a}, \mathrm{b}}$ Lucas Silva Franco, ${ }^{\mathrm{a}, \mathrm{b}}$ Teiliane Rodrigues Carneiro, ${ }^{\text {a }}$ Eliezer J. Barreiro, ${ }^{\mathrm{a}, \mathrm{b}}$ and Lidia Moreira Lima ${ }^{\mathrm{a}, \mathrm{b}^{*}}$<br>a. Instituto Nacional de Ciência e Tecnologia de Fármacos e Medicamentos (INCT-INOFAR), Laboratório de Avaliação e Síntese de Substâncias Bioativas (LASSBio ${ }^{\circledR}$ ), Universidade Federal do Rio de Janeiro<br>(UFRJ), CCS, Cidade Universitária, Rio de Janeiro-RJ, Brasil.<br>b. Pós-graduação em Farmacologia e Química Medicinal, Instituto de Ciências Biomédicas, Universidade<br>Federal do Rio de Janeiro, Rio de Janeiro-RJ, Brasil

## Table of Contents

Supplementary Information: Molecular Modelling ..... 4
Supplementary Information: Chemistry ..... 10

1. Synthesis of LASSBio-2123 (5), LASSBio-2124 (6), LASSBio-2125 (7), LASSBio-2126 (9), and LASSBio-2127 (13) ..... 11
2. Synthesis of 3 -amino- $N^{\prime}$-(3,4-difluorobenzyl)-4-(2,4,5-trifluorophenyl)butanehydrazide: LASSBio-2127 (13) ..... 16
3. Synthesis of LASSBio-2128 (14) ..... 16
4. Enantiomeric Synthesis of $N$-acylhydrazones (6-R and 6-S) ..... 20
5. Aqueous solubility determination ..... 27
6. Dissociation constant determination ..... 27
7. Chemical stability studies ..... 27
8. In vitro DPP-4 inhibition evaluation ..... 28
Supplementary Information: Analytical Spectra ..... 29
References ..... 79

## Table of Supplementary Figures and Tables

Figure S1. Sitagliptin (1) co-crystallized (yellow) in the active site of DPP4. S1 hydrophobic pocket in green. Key glutamates residues of S2 pocket in light-blue. Residues that form the "oxyanion hole" in white (Tyr659/547). Selected residues on S2 extended pocket in purple. Catalytic triad (Ser630, His740, Asp708) in orange. .. 5
Figure S2. a) Alignment of co-crystallographic structures of gliptins 7-16 approved for the treatment of DMT2 and the apo form with human DPP-4 b) Zoom of the active site. In yellow aminoacids without conformational differences. In purple aminoacids that presented conformational differences (Tyr547, co-crystallographic structure of vildagliptin 6b1o) and Trp629 (apo form structure of DPP-4, 1pfq) from active site of DPP-4. .. 5
Figure S3. Superposition of sitagliptin co-crystallized (grey) and re-docked (purple) with GOLD
5.1 program. RMSD $=0.387 \AA$ A................................................................................................................
Figure S4. Superposition of co-crystallized (R)-sitagliptin (purple) and docked (S)-sitagliptin (green) inside DPP-4 active site. .. 6

Figure S5. Superposition of compounds (10-R) (purple), (11-R) (orange), (12-R) (blue) and R- sitagliptin (1-R) (green) at the DPP-4 active site. 7
Figure S6. Linagliptin (11) interaction with Trp629. .....  8
Figure S7. Structures of LASSBio-294 and LASSBio-785 obtained by X-ray diffraction. ..... 8
Figure S8. Superposition of compounds (13-R) (green) and R-sitagliptin (1-R) (purple) inside the DPP-4 active site. 8
Figure S9. a) ${ }^{1} \mathrm{H}-\mathrm{NMR}$ NOE-diff ( $400 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}$ ) spectrum of 14 . Signal irradiated $\delta_{H} 7.86$( $\mathrm{N}=\mathrm{CH}$ ). b) ${ }^{1} \mathrm{H}-\mathrm{NMR}$ NOE-diff ( $400 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}$ ) spectrum of 14. Signal irradiated $\delta_{H} 3.36$ ( N -$\mathrm{CH}_{3}$ ).19
Figure S10. Chemical shifts differences $\left(\Delta \delta^{R-S}\right)$ between diastereoisomers (R)-MPA-( $R^{*}$ )- $\beta$ -aminoester (23) and (S)-MPA-( $\mathrm{R}^{*}$ )- $\beta$-aminoester (24). Representation of the model thatallowed the determination of absolute stereochemistry of 20-R* as R. .................................... 24

Figure S11. a) Energetic rotation profile of H-N-C-H bond (blue bond for diastereoisomer 23 and orange bond for diastereoisomer 24) obtained by semi-empirical method PM3 in SPARTAN program. b) Favored conformers for diastereoisomers 23 and 24 obtained by semi-empirical method PM3 in SPARTAN program. .24

Table S1. PDB codes of co-crystal structures of gliptins approved for the treatment of DMT2 and the Apo form with human DPP-4, their resolution, and type of interaction with DPP-4....... 4 Table S2. GOLD 5.1 scores for (R)-sitagliptin (1-R) and the E and Z isomers of compound (6-R). 6 Table S3. Score values obtained with GOLD 5.1 for (R)-sitagliptin, (6-R), (6-S), (13-R), (13-S), (14-R) and (14-S). .. 7
Table S4. Physicochemical and drug-likeness properties of sitagliptin (7), LASSBio-1773 (2) and compounds 5-14 calculated by ACD/Percepta Software. .....  9
Table S5. Physical and spectroscopic properties of LASSBio-2123 to LASBio-2128 (5-7, 9, 13- 14). ..... 19
Table S6. ${ }^{1} \mathrm{H}-\mathrm{NMR}$ data for the MPA amides of $\mathbf{2 3}$ and $\mathbf{2 4}$ at 400 MHz , in $\mathrm{CDCl}_{3} ; \delta$ in ppm , J inHz .23

## Supplementary Information: Molecular Modelling

All compounds were constructed and energy-minimized at the HF/3-21G level using Spartan 14 (Wavefunction Inc.). Among the DPP-4 crystallographic structures (Table S1, Figure S1) available in the Protein Data Bank, the one with code $1 \times 70$ (resolution $2.1 \AA$ ) was used for docking runs with the GOLD 5.1 program (CCDC; License key: G/4142006). GOLD was used for addition of hydrogen atoms in the protein structure. The binding site was determined as the amino acids having atoms within 6 Å radius of the co-crystallized ligand. With the aim to obtain the more stable complex (protein-ligand) all scoring functions of GOLD program were explored, GoldScore, ChemScore, ChemPLP and ASP. All the score functions were evaluated with and without water for the re-docking of the co-crystallized sitagliptin (7) to identify the most adequate fitness function for docking studies into DPP-4. After docking, the RMSD between the best result for each fitness function and the crystallographic structure was calculated. The fitness function with the best performance in the re-docking test, which was ChemPLP (without water) (Figure S3 and S4), was used for the semi-rigid docking study of the compounds described in this study. Three consecutive runs were conducted, each of which generated 10 conformations. The highest-scoring conformation of each run was chosen and analyzed.

Table S1. PDB codes of co-crystal structures of gliptins approved for the treatment of DMT2 and the Apo form with human DPP-4, their resolution, and type of interaction with DPP-4.

| Co-crystal Structure | PDB File | Resolution ( $\AA$ ) | Interaction type |
| :---: | :---: | :---: | :---: |
| Sitagliptin (7) | $1 \times 70$ | 2.10 | Reversible |
| Vildagliptin (8) | 6 b 1 o | 1.91 | Covalent Reversible |
| Saxagliptin (9) | 3 BJM | 2.35 | Covalent Reversible |
| Alogliptin (10) | 3 G 0 B | 2.25 | Reversible |
| Linagliptin (11) | 5 t 4 b | 1.76 | Reversible |
| Anagliptin (13) | 3 WQH | 2.85 | Reversible |
| Tenegliptin (14) | 3 VJK | 2.49 | Reversible |
| Trelagliptin (15) | 5 KBY | 2.24 | Reversible |
| Omarigliptin (16) | 4 pnz | 1.90 | Reversible |
| Apo form | 1 pfq | 1.90 |  |



Figure S1. Sitagliptin (1) co-crystallized (yellow) in the active site of DPP4. S1 hydrophobic pocket in green. Key glutamates residues of S2 pocket in light-blue. Residues that form the "oxyanion hole" in white (Tyr659/547). Selected residues on S2 extended pocket in purple. Catalytic triad (Ser630, His740, Asp708) in orange.


Figure S2. a) Alignment of co-crystallographic structures of gliptins 7-16 approved for the treatment of DMT2 and the apo form with human DPP-4 b) Zoom of the active site. In yellow aminoacids without conformational differences. In purple aminoacids that presented conformational differences (Tyr547, co-crystallographic structure of vildagliptin 6b1o) and Trp629 (apo form structure of DPP-4, 1pfq) from active site of DPP-4.



Re-docked Sitagliptin
Co-crystallized Sitagliptin

Figure S3. Superposition of sitagliptin co-crystallized (grey) and re-docked (purple) with GOLD 5.1 program. RMSD $=0.387$ Å.


Figure S4. Superposition of co-crystallized (R)-sitagliptin (purple) and docked (S)-sitagliptin (green) inside DPP-4 active site.

Table S2. GOLD 5.1 scores for (R)-sitagliptin (1-R) and the E and Z isomers of compound (6-R).


(6-R) $E$ isomer

(6-R) $Z$ isomer

| Compound | Score Value |
| :---: | :---: |
| $(R)$-Sitagliptin (1-R) | 88.0 |
| $(6-R) E$ isomer | 86.9 |
| $(6-R) Z$ isomer | 76.3 |



Figure S5. Superposition of compounds (10-R) (purple), (11-R) (orange), (12-R) (blue) and R-sitagliptin (1-R) (green) at the DPP-4 active site.

Table S3. Score values obtained with GOLD 5.1 for (R)-sitagliptin, (6-R), (6-S), (13-R), (13-S), (14-R) and (14-S).

(6-R,S)




Figure S6. Linagliptin (11) interaction with Trp629.


Figure S8. Superposition of compounds (13-R) (green) and R-sitagliptin (1-R) (purple) inside the DPP-4 active site.

Table S4. Physicochemical and drug-likeness properties of sitagliptin (7), LASSBio-1773 (2) and compounds 5-14 calculated by ACD/Percepta Software.



| Compound | $\underset{\mathrm{P}}{\text { cLog }}$ | LogD |  | pKa | PPB | BBB | Bioavailability (F) | $\begin{gathered} \text { Permeation } \\ (\mathrm{CaCo}-2) \\ (\mathrm{cm} / \mathrm{s}) \end{gathered}$ | HIA | Solubility ( $\mathrm{mg} / \mathrm{mL}$ ) |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  |  | 4.6 | 7.4 |  |  |  |  |  |  |  |
| Sitagliptin (1) | $\begin{gathered} 1.17 \\ \left(1.5^{a}\right) \\ \hline \end{gathered}$ | $1.93$ | $0.11$ | $\begin{gathered} \hline 7,2 \\ \left(7,7^{\mathrm{a}}\right) \\ \hline \end{gathered}$ | $\begin{gathered} \hline 68 \% \\ \left(38 \%{ }^{\text {a }}\right. \end{gathered}$ | $\begin{gathered} -2.97 \\ \text { (Penetrant) } \\ \hline \end{gathered}$ | (87\% ${ }^{\text {a }}$ ) 77\% | $\mathrm{Pe}=49 \mathrm{E}^{-6}$ | $\begin{gathered} \hline 100 \% \\ \left(90 \%{ }^{\text {ab }}\right) \\ \hline \end{gathered}$ | $\begin{gathered} 0.09 \\ \text { Insoluble } \end{gathered}$ |
| LASSBio-1773 (2) | 2.79 | 2.36 | 2.36 | $\begin{gathered} \hline 8.7 ;- \\ 3.7 \\ \hline \end{gathered}$ | 98\% | -3.34 (weak) | 98\% | $\mathrm{Pe}=206 \mathrm{E}^{-6}$ | 100\% | $\begin{gathered} 0.003 \\ \text { Insoluble } \end{gathered}$ |
|  | 2.39 | $1.14$ | 0.2 | $\begin{aligned} & 7.0 \\ & 12.9 \end{aligned}$ | 87\% | $-2.97$ <br> (Penetrant) | 82\% | $\mathrm{Pe}=84 \mathrm{E}^{-6}$ | 100\% | $\begin{gathered} 0.02 \\ \text { Insoluble } \end{gathered}$ |
|  | 2.74 | $0.92$ | 0.43 | $\begin{aligned} & 7.1 ; \\ & 12.6 \end{aligned}$ | 90\% | -3.24 (weak) | 89\% | $\mathrm{Pe}=114 \mathrm{E}^{-6}$ | 100\% | $\begin{gathered} 0.008 \\ \text { Highly } \\ \text { insoluble } \end{gathered}$ |
|  | 4.00 | 0.47 | 1.82 | $\begin{aligned} & \text { 7.1; } \\ & \text { 12.7 } \end{aligned}$ | 96\% | $\begin{gathered} -2.70 \\ \text { (Penetrant) } \end{gathered}$ | 48\% | $\mathrm{Pe}=156 \mathrm{E}^{-6}$ | 100\% | $\begin{gathered} 0.0002 \\ \text { Insoluble } \end{gathered}$ |
| 8 | 2.04 | $2.18$ | 0.08 | $\begin{gathered} \hline 7.1 ; \\ 13.6 ; \\ 4.4 ; \end{gathered}$ | 89\% | -3.38 (weak) | 48\% | $\mathrm{Pe}=29 \mathrm{E}^{-6}$ | 100\% | $\begin{gathered} \hline 0.007 \\ \text { Insoluble } \end{gathered}$ |


| pKa 3 |  |  |  | 11.9 |  |  |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | 1.00 | $4.53$ | $1.59$ | $\begin{gathered} 7.1 ; \\ 14.3 ; \\ 3.9 ; \\ 11.5 \end{gathered}$ | 58\% | -3.33 (weak) | 17\% | $\mathrm{Pe}=8 \mathrm{E}^{-6}$ | 70\% | $\begin{gathered} 0.36 \\ \text { Soluble } \end{gathered}$ |
| 10 pKa 3 <br> pKa 4 | 0.52 | -3.0 | $1.68$ | $\begin{gathered} 7.6 ; \\ 13.5 ; \\ 0.5 ; \\ 9.22 \end{gathered}$ | 47\% | -4.02 (Nonpenetrant) | 14\% | $P \mathrm{e}=1 \mathrm{E}^{-6}$ | 50\% | $\begin{gathered} 1.35 \\ \text { Soluble } \end{gathered}$ |
|  | 1.29 | $4.07$ | $1.33$ | $\begin{gathered} 7.5 ; \\ 12.0 \\ 5.1 ; \\ 14.9 \end{gathered}$ | 58\% | -3.47 (weak) | 15\% | $P \mathrm{e}=5 \mathrm{E}^{-6}$ | 43\% | $\begin{gathered} 0.15 \\ \text { Soluble } \end{gathered}$ |
| 12 <br> pKa 3 | 1.95 | $3.15$ | $0.55$ | $\begin{gathered} 7.4 ; \\ 12.4 ; \\ 5.2 \end{gathered}$ | 49\% | $-2.82$ <br> (Penetrant) | 19\% | $\mathrm{Pe}=35 \mathrm{E}^{-6}$ | 91\% | 0.1 <br> Soluble |
| 13 | 1.66 | $0.49$ | 0.88 | $\begin{gathered} 8.0 ; \\ 0.5 \\ \hline \end{gathered}$ | 77\% | -3.88 (Nonpenetrant) | 63\% | $\mathrm{Pe}=8 \mathrm{E}^{-6}$ | 87\% | 0.28 <br> Soluble |
| 14 | 2.59 | $1.26$ | 0.97 | 9.1 | 92\% | -4.33 (Nonpenetrant) | 98\% | $\mathrm{Pe}=48 \mathrm{E}^{-6}$ | 100\% | $\begin{gathered} 0.69 \\ \text { Soluble } \end{gathered}$ |

${ }^{\text {a) }}$ Experimental data from literature report ${ }^{1}$. b) $\operatorname{In}$ dogs ${ }^{12}$
PPB=Protein plasma binding; $\mathrm{BBB}=\mathrm{Blood}$-brain barrier; CaCo-2=Human epithelial colorectal adenocarcinoma cells; HIA=Human intestinal absorption.

## Supplementary Information: Chemistry

## General Remarks

Unless otherwise noted, all the materials were obtained from commercially available sources and were used without purification. Reactions were routinely monitored by thin-layer chromatography (TLC) in silica gel (KieselGel 60 F254 Merck); the products were visualized with UV lamps ( 254 nm \& 365 nm ), or using the following color reagents: iodine, 2,4dinitrophenylhydrazine or 4-dimethylaminobenzaldehyde. Purifications by column chromatography were carried out in silica gel 230-400 Mesh (Merck). Melting points (m.p.) were determined in a melting point system MP70 Mettler Toledo and the values obtained were not corrected. ${ }^{1} \mathrm{H}$ - and ${ }^{13} \mathrm{C}$-Nuclear Magnetic Resonance (NMR) spectra were determined in deuterated solutions using a Varian model MR400 spectrometer with 400 MHz for ${ }^{1} \mathrm{H}$ - and 100 MHz for ${ }^{13} \mathrm{C}$ (LAMAR, Laboratório Multiusuários de Análises por RMN (LAMAR). Núcleo de Pesquisas de Produtos Naturais, Universidade Federal do Rio de Janeiro; UFRJ, Brazil), or Varian Mercury- 300 spectrometer with 300 MHz for ${ }^{1} \mathrm{H}$ and 75 MHz for ${ }^{13} \mathrm{C}$ (IMA, Laboratório de Ressonância Magnética Nuclear (RMN) de Alta Resolução. Instituto de Macromoléculas, Universidade Federal do Rio de Janeiro; UFRJ, Brazil) or Bruker AC-200 spectrometer with 200 MHz for ${ }^{1} \mathrm{H}$ and 50 MHz for ${ }^{13} \mathrm{C}$ (LABRMN, Laboratório de Ressonância Magnética Nuclear (LABRMN). Instituto de Química (IQ). Universidade Federal do Rio de Janeiro (UFRJ), Brazil). The chemical shifts are given in parts per million ( $\delta$ ) from solvent residual peaks and the coupling constant values (J) are given in Hertz. Signal multiplicities are represented as s
(singlet), d (doublet), t (triplet), q (quartet) and m (multiplet). Deuterated solvents like $\mathrm{CDCl}_{3}$, DMSO- $d_{6}$ and $C D_{3} O D$ were used for the determinations. High performance liquid chromatography (HPLC) for purity determinations were conducted using Shimadzu LC-20AD equipped with a SIL-20A autosampler, with a Shimadzu CBM-20A interface and a Shimadzu SPD-M20A detector (Diode Array) at substance-specific wavelengths. The column used was Kromasil 100-5 C18 ( $4.6 \mathrm{~mm} \times 250 \mathrm{~mm}$ ) and the solvent system for HPLC purity analyses was conducted in mobile phase $\mathrm{MeCN}-\mathrm{H}_{2} \mathrm{O}-\mathrm{NH}_{4} \mathrm{OH}(5.0 \mathrm{~N})$ 60:40:0.01 ( $\mathrm{pH}=9$ ), with $1 \mathrm{~mL} / \mathrm{min}$ flow, sample concentration of $1 \mathrm{mg} / \mathrm{mL}$, injection volume of $20 \mu \mathrm{~L}$, and run time $=15 \mathrm{~min}$. Data were acquired by software "LC solution" version 4.0. Mass spectra were obtained in electrospray ionization (EI) mode on Bruker AmaZon SL ion-trap MS. Spectra were analyzed using Compass 4.0 software. Infrared (IR) spectra were obtained in a Thermo Nicolet's iS 10 FT-IR spectrometer, equipped with smart iTR ATR for direct measurements. All described products showed MS spectra, IR spectra and ${ }^{1} \mathrm{H}$ - and ${ }^{13} \mathrm{C}-\mathrm{NMR}$ spectra according to the assigned structures and obtained a purity ranging from 95.0 \% to 99.8 \% determined by HPLC. Optical rotation was measured with a Perkin-Elmer polarimeter model 341 using a sodium lamp (589 nm ) at $20^{\circ} \mathrm{C}$ (Central Analítica - Núcleo de Pesquisas de Produtos Naturais, Universidade Federal do Rio de Janeiro; UFRJ, Brazil).

## 1. Synthesis of LASSBio-2123 (5), LASSBio-2124 (6), LASSBio-2125 (7), LASSBio-2126 (9), and LASSBio-2127 (13)

## 1.1. (5-(1-hydroxy-2-(2,4,5-trifluorophenyl)ethylidene)-2,2-dimethyl-1,3-dioxane-4,6-dione)

(17)


Adapted from Sun et al (2011) ${ }^{3}$. In a 25 mL two necked round-bottom flask equipped with a reflux condenser, a magnetic stirrer and under argon atmosphere, 2 g ( 10.52 mmol ) of 2,4,5trifluorophenylacetic acid (15) was dissolved in 15 mL of dry THF. Next, 1.1 equivalents of CDI were added. The reaction was heated to $50^{\circ} \mathrm{C}$ and then 1.1 equivalents of Meldrum's acid (16) were added, keeping the stirring and temperature for 4 hours. Completion of the reaction was verified by TLC. Next, the reaction mixture was concentrated under reduced pressure and then, the residue was dissolved in DCM followed by a partition with cold water. pH was adjusted to 2 with $\mathrm{HCl} 10 \%$ and the aqueous phase was separated. The organic phase was washed with HCl 0.1 N and brine, respectively. The organic phase was then dried with anhydrous $\mathrm{Na}_{2} \mathrm{SO}_{4}$, filtered, and concentrated under vacuum to yield the Meldrum's adduct (17).

Appearance: white pale solid; Yield: 92\%; TLC conditions: DCM/MeOH 98:2 (Rf: 0.75); mp: 102 $-103^{\circ} \mathrm{C}$ (lit. 101.5-103.5$\left.{ }^{\circ} \mathrm{C}\right)^{4}$; NMR- ${ }^{1} \mathrm{H}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta(\mathrm{ppm}): 7.18-7.12(1 \mathrm{H}, \mathrm{m}) ; 6.99-6.93$ (1H, m); $4.44(2 \mathrm{H}, \mathrm{s}) ; 1.76(6 \mathrm{H}, \mathrm{s})$.

### 1.2. Synthesis of methyl 3-oxo-4-(2,4,5-trifluorophenyl)butanoate (18)



Adapted from Gore et al (2010) ${ }^{4}$. In a 100 mL round-bottom flask equipped with a reflux condenser, a magnetic stirrer and under argon atmosphere, 3.0 g ( 9.48 mmol ) of Meldrum's adduct 17 was dissolved in 20 mL of dry MeOH . The resulting solution was stirrer at $60^{\circ} \mathrm{C}$ overnight. After verifying total consumption of starting material (monitoring by TLC), the reaction mixture was concentrated under reduce pressure and the residue was treated with $\mathrm{Na}_{2} \mathrm{CO}_{3} 10 \%$ aqueous solution, achieving $\mathrm{pH}=8$, then it was extracted with DCM ( $3 \times 40 \mathrm{~mL}$ ). The organic phase was dried with anhydrous $\mathrm{Na}_{2} \mathrm{SO}_{4}$, filtered, and concentrated under vacuum to yield compound 18.
Appearance: yellow oil; Yield: 75\%; TLC conditions: Hexane/ $\mathrm{CHCl}_{3} 3: 7$ (Rf: 0.6); MS (ESI): $[\mathrm{M}+\mathrm{H}]^{+}, 247,23 ; \mathrm{NMR}^{-1} \mathrm{H}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta(\mathrm{ppm}): 7.08-7.02(1 \mathrm{H}, \mathrm{m}) ;$ 6.99-6.92 (1H, m); 3.85 $(2 \mathrm{H}, \mathrm{s}) ; 3.76(3 \mathrm{H}, \mathrm{s}) ; 3.55(2 \mathrm{H}, \mathrm{s}) ; \mathrm{NMR}^{-13} \mathrm{C}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta(\mathrm{ppm}): 198.0$ (C8); 167.3 (C14); 158.6/153.7 (C4); 152.1/147.1 (C2); 149.2/144.4 (C1); 119.3 (C3); 116.9 (C5); 105.6 (C6); 52.6 (C15); 48.5 (C7); 42.1 (C3).

### 1.3.Synthesis of (Z)-methyl 3-amino-4-(2,4,5-trifluorophenyl)but-2-enoate (19)



Adapted from Kubryk et al (2006) ${ }^{5}$ and Gore et al (2010) ${ }^{4}$. In a 100 mL round-bottom flask equipped with a reflux condenser, a magnetic stirrer and under argon atmosphere, $1,8 \mathrm{~g}(7.31$ mmol) of $\beta$-ketoester 18 and 10 equivalents of $\mathrm{NH}_{4} \mathrm{OAc}$ were dissolved in 25 mL of dry MeOH . The resulting mixture was stirred and heated at $60^{\circ} \mathrm{C}$ overnight. After end of the reaction, it was concentrated under reduce pressure until obtaining a pale-yellow oil, which was dissolved in EtOAc and stirred at $25-30^{\circ} \mathrm{C}$ for 15 min . The white precipitate was filtered, and the solvent distilled until $1 / 4$ of initial volume. Next, hexane was added to the residue and stirred to 30 $35^{\circ} \mathrm{C}$ for 1 h . Finally, the mixture was filtered, dried over anhydrous $\mathrm{Na}_{2} \mathrm{SO}_{4}$, and concentrated under vacuum to yield compound 19.
Appearance: white pale solid; Yield: 90\%; TLC conditions: Hexane/ $\mathrm{CHCl}_{3} 3: 7$ (Rf: 0.5); MS (ESI): $[\mathrm{M}+\mathrm{H}]^{+}, 246,05 ; \mathrm{NMR}^{-1} \mathrm{H}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta(\mathrm{ppm})$ : 7.15-7.02 (1H, m); 7.02-6.89 (1H, m); 4.57 (1H, s); $3.65(3 \mathrm{H}, \mathrm{s}) ; 3.41(2 \mathrm{H}, \mathrm{s}) ; \mathrm{NMR}^{-13} \mathrm{C}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta(\mathrm{ppm}): 170.4$ (C14); 159.4 (C8); 158.5/153.5 (C4); 151.9/146.9 (C2); 149.4/144.4 (C1); 120.0 (C3); 118.5 (C5); 105.7 (C6); 85.0 (C13); 50.4 (C15); 34.7 (C7).

### 1.4. Synthesis of methyl 3-amino-4-(2,4,5-trifluorophenyl)butanoate (20)



Adapted from Gore et al (2010) ${ }^{4}$. In a 100 mL round-bottom flask equipped with a magnetic stirrer and under argon atmosphere, $1,6 \mathrm{~g}(6.53 \mathrm{mmol})$ of $\beta$-enamine ester 19 was dissolved in 20 mL of dry MeOH , followed by addition of AcOH until $\mathrm{pH} 4-5$. The resulting solution was stirred at $0^{\circ} \mathrm{C}$ for 15 min . Subsequently, 3 equivalents of $\mathrm{NaCNBH}_{3}$ were added and the reaction was stirred overnight. After completion of the reaction (monitored by TLC) the solvent was distillated, and the residue was treated with $\mathrm{HCl} 10 \%$ until pH 2 . The resulting suspension was stirred for 15 min and extracted with DCM ( $3 \times 20 \mathrm{~mL}$ ). The organic phase was separated, and the aqueous phase was basified with $\mathrm{NaOH} 10 \%$ until $\mathrm{pH} 8-9$. Then, the product was isolated with EtOAc ( $3 \times 50 \mathrm{~mL}$ ). The organic phase was dried over anhydrous $\mathrm{Na}_{2} \mathrm{SO}_{4}$, filtrated, and concentrated to yield 20.
Appearance: dark yellow oil; Yield: 75\%; TLC conditions: $\mathrm{DCM} / \mathrm{MeOH} / \mathrm{NH}_{4} \mathrm{OH}$ 95:5:0.025 (Rf: 0.3 ); MS (ESI): $[\mathrm{M}+\mathrm{H}]^{+}, 247,98 ; \mathrm{NMR}^{1} \mathrm{H}(400 \mathrm{MHz}, \mathrm{CDCl} 3) \delta(\mathrm{ppm}): 7.11-6.98(1 \mathrm{H}, \mathrm{m}) ; 6.97-$ $6.84(1 \mathrm{H}, \mathrm{m}) ; 3.68(3 \mathrm{H}, \mathrm{s}) ; 3.44(1 \mathrm{H}, \mathrm{bs}) ; 2.72-2.30(4 \mathrm{H}, \mathrm{m}) ; \mathrm{NMR}^{-13} \mathrm{C}(100 \mathrm{MHz}, \mathrm{CDCl} 3) \delta(\mathrm{ppm})$ : 172.6 (C14); 158.9/153.9 (C4); 151.4/146.5 (C2); 149.1/144.4 (C1); 122.0 (C3); 119.1 (C5); 105.7 (C6.); 51.8 (C15); 48.6 (C8); 41.5 (C13); 36.4 (C7).

### 1.5. Synthesis of 3-amino-4-(2,4,5-trifluorophenyl)butanehydrazide (21)



Adapted from Kümmerle et al. (2012) ${ }^{6}$. In a 25 mL round-bottom flask equipped with a magnetic stirrer and reflux condenser, $700 \mathrm{mg}(2.83 \mathrm{mmol})$ of $\beta$-aminoester $\mathbf{2 0}$ was dissolved in 15 mL of EtOH and heated at $70^{\circ} \mathrm{C}$. Next, 10 equivalents of hydrazinium hydroxide (about $100 \% \mathrm{~N}_{2} \mathrm{H}_{5} \mathrm{OH}$ ) were slowly added to the reaction mixture, which was constantly stirred at $70^{\circ} \mathrm{C}$ overnight. After completion of the reaction (verified by TLC) the solvent was completely distilled and the residue dissolved in brine solution ( 40 mL ), followed by exhaustive extraction with DCM ( $6 \times 50 \mathrm{~mL}$ ). Subsequently, the organic phase was dried with anhydrous $\mathrm{Na}_{2} \mathrm{SO}_{4}$, filtered, and concentrated under vacuum to yield compound 21.

Appearance: yellow oil; Yield: 90\%; TLC conditions: DCM/MeOH/NH $\mathrm{NOH}_{4}$ 90:10:0.05 (Rf: 0.45); HPLC purity: 96\%; MS (ESI): [M+H] ${ }^{+}$, 248,79; NMR- ${ }^{1} \mathrm{H}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta(\mathrm{ppm}): 8.27(1 \mathrm{H}, \mathrm{s})$; 7.06-7.00 (1H, m); 6.97-6.90 (1H, m); $3.39(1 \mathrm{H}, \mathrm{m}) ; 2.75(1 \mathrm{H}, \mathrm{dd}, \mathrm{J}=6.0,14.0) ; 2.63(1 \mathrm{H}, \mathrm{dd}$, J=8.0, 13.6); $2.39(1 \mathrm{H}, \mathrm{dd}, \mathrm{J}=3.2,15.6), 2.16(1 \mathrm{H}, \mathrm{dd}, \mathrm{J}=9.2,15.2) ; \mathrm{NMR}^{13} \mathrm{C}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta$ (ppm): 172.1 (C14); 158.0/154.6 (C4); 150.6/147.3 (C2); 148.2/145.1 (C1); 121.6 (C5); 119.1 (C3); 105.7 (C6); 49.1 (C8); 41.3 (C7); 37.5 (C13).
1.6.General Synthetic Procedure for the Target Compounds LASSBio-2123-2126 (5-7, 9)


Adapted from Kümmerle et al. (2012) ${ }^{6}$. In a 25 mL round-bottom flask equipped with a magnetic stirrer, 1 mmol of hydrazide 21 was dissolved in $5-10 \mathrm{~mL}$ of EtOH. After 15 min of stirring 1.1 equivalents of the corresponding aldehyde was added. The reaction mixture was stirred at room temperature until TLC analysis showed complete consumption of the hydrazide. Subsequently, the solvent was completely distillated. The target products were purified as described next:

### 1.6.1. (E)-3-amino-N'-(3,4-dimethoxybenzylidene)-4-(2,4,5-trifluorophenyl) butanehydrazide: LASSBio-2123 (5)


(5)

After completion of the reaction (9h), the residue was concentrated and partitioned with water and EtOAc. The organic phase was washed with brine ( $3 \times 30 \mathrm{~mL}$ ), dried with anhydrous $\mathrm{Na}_{2} \mathrm{SO}_{4}$, filtered, and concentrated under vacuum. The product was purified by column chromatography (Hexane/EtOAc 1:1 $\rightarrow$ EtOAc/MeOH 9:1) yielding compound 5.
Appearance: yellow oil; Yield: 75\%; TLC conditions: DCM/MeOH/NH $\mathrm{NOH}_{4}$ 90:10:0.05 (Rf: 0.65); Purity: 95.7\%; MS (ESI): [M+H] ${ }^{+}$, 396.01; NMR- ${ }^{1} \mathrm{H}\left(300 \mathrm{MHz}, \mathrm{DMSO}-\mathrm{d}_{6}\right) \delta(\mathrm{ppm}): 8.06(0.9 \mathrm{H}, \mathrm{s}$, $\left.\mathrm{H}-13^{\prime}\right) / 7.87(1 \mathrm{H}, \mathrm{s}, \mathrm{H}-13)$; 7.51-7.40 (2H, m, H-6, H-3); $7.27\left(0.9 \mathrm{H}, \mathrm{d}, \mathrm{J}=4.0, \mathrm{H}-19^{\prime}\right) / 7.21(1 \mathrm{H}, \mathrm{d}$, J=4.0, H-19); 7.15 ( $\left.0.9 \mathrm{H}, \mathrm{dd}, \mathrm{J}=2.4,11.2, \mathrm{H}-15^{\prime}\right) / 7.10(1 \mathrm{H}, \mathrm{dd}, \mathrm{J}=2.4,11.2, \mathrm{H}-15) ; 7.00(1 \mathrm{H}, \mathrm{d}$, J=6.8, H-16)/6.97 (0.9H, d, J=6.8, H-16'); 3.79 ( $12 \mathrm{H}, \mathrm{s}, \mathrm{H}-27, \mathrm{H}-28, \mathrm{H}-27$ ', H-28'); 3.42-3.27 (m, H-8); 2.76-2.56 (8H, m, H-7a, H-7a', H-7b, H-7b', H-9a', H-9b); 2.27 (1H, dd, J=6.0, 19.2, H9a)/2.16 (1H, dd, J=11.2, 19.2, H-9b'); NMR- ${ }^{13}$ C ( $75 \mathrm{MHz}, \mathrm{DMSO}^{2}$ ) $\delta(\mathrm{ppm}): 171.7 / 165.7$ (C10); 157.4/154.3 (C-4); 150.6 (C-17); 149.3/147.4 (C-2); 149.0 (C-18); 146.1/142.6 (C-13); 146.0/144.2 (C-1); 127.0 (C-14); 123.4 (C-5); 121.7/120.7 (C-15); 119.5 (C-3); 111.5 (C-16); 108.5/108.1 (C-19); 105.5 (C-6); 55.5 (C-27, C-28); 49.1/48.5 (C-8), 42.2/40.4 (C-9); 36.0 (C-7).

### 1.6.2. (E)-3-amino-N'-(3,4-difluorobenzylidene)-4-(2,4,5-trifluorophenyl) butanehydrazide: LASSBio-2124 (6)


(6)

After completion of the reaction (8h), the residue was concentrated and partitioned with water and EtOAc. The organic phase was washed with brine ( $3 \times 30 \mathrm{~mL}$ ), dried with anhydrous $\mathrm{Na}_{2} \mathrm{SO}_{4}$, filtered, and concentrated under vacuum. The product was purified by column chromatography (Hexane/EtOAc 1:1 $\rightarrow$ EtOAc/MeOH 9:1) yielding compound 6.

Appearance: yellow oil; Yield: 70\%; TLC conditions: DCM/MeOH/ $\mathrm{NH}_{4} \mathrm{OH} 90: 10: 0.05$ (Rf: 0.7); Purity: 99.4\%; MS (ESI): $[\mathrm{M}+\mathrm{H}]^{+}, 372.15$; NMR-1 ${ }^{\mathrm{H}}\left(300 \mathrm{MHz}, \mathrm{DMSO}_{6}\right) \delta(\mathrm{ppm}): 8.13(0.7 \mathrm{H}, \mathrm{s}$, $\mathrm{H}-13) / 7.91$ ( $1 \mathrm{H}, \mathrm{s}, \mathrm{H}-13$ ); 7.73-7.61 (1.8H, m, H-3); 7.53-7.41 (7H, m, H-6, H-15, H-16, H-19); 3.35 (overlapping with water $\mathrm{H}-8$ ); 2.77/2.61 ( $5.8 \mathrm{H}, \mathrm{m}, \mathrm{H}-7 \mathrm{a}, \mathrm{H}-7 \mathrm{~b}, \mathrm{H}-9 \mathrm{a}$ ); 2.32/2.14 (1.72H, m, H-9b);
NMR- ${ }^{13}$ C ( $75 \mathrm{MHz}, \mathrm{DMSO}_{\mathrm{d}}$ ) $\delta(\mathrm{ppm}): 173.3 / 167.5$ (C-10); 157.5/154.6 (C-4); 151.9/148.6 (C18); 151.4/148.2 (C-17); 148.3/146.0 (C-2); 147.3/144.1 (C-1); 143.7/140.2 (C-13); 132.2 (C-14); 124.0 (C-15); 123.3 (C-5); 119.5 (C-19); 118.0 (C-16); 115.0 (C-3); 105.5 (C-6); 49.0/48.7 (C-8); 42.0 (C-7), 36,0 (C-9).

### 1.6.3. (E)-3-amino-N'-(3,4-dichlorobenzylidene)-4-(2,4,5-trifluorophenyl) butanehydrazide: LASSBio-2125 (7)


(7)

After completion of the reaction (8h), the residue was concentrated and partitioned with water and EtOAc. The organic phase was washed with brine ( $3 \times 30 \mathrm{~mL}$ ), dried with anhydrous $\mathrm{Na}_{2} \mathrm{SO}_{4}$, filtered, and concentrated under vacuum. The product was purified by column chromatography (Hexane/ EtOAc 1:1 $\rightarrow$ EtOAc /MeOH 9:1) yielding compound 7.

Appearance: yellow oil; Yield: 70\%; TLC conditions: $\mathrm{DCM} / \mathrm{MeOH} / \mathrm{NH}_{4} \mathrm{OH} 90: 10: 0.05$ (Rf: 0.75); Purity: 99.7\%; MS (ESI): [M+H] ${ }^{+}$, 404,09; NMR- ${ }^{1} \mathrm{H}\left(300 \mathrm{MHz}, \mathrm{DMSO}^{-d_{6}}\right.$ ) $\delta(\mathrm{ppm}): 8.18(0.7 \mathrm{H}, \mathrm{s}$, H-13')/7.97 (1H, s, H-13); 7.93 (0.7H, s, H-19')/7.87 (1H, d, J= 4, H-19); 7.70-7.68 (2.4 H, m, H15', H-16, H-16'); 7.60 ( $1 \mathrm{H}, \mathrm{dd}, \mathrm{J}=12,4, \mathrm{H}-15$ ); 7.50-7.42 (3.7H, m, H-3, H-6); 3.40 (m, H-8); 2.82-2.63 (5.9H, m, H-7a, H-7a', H-7b, H-7b', H-9a', H-9b); 2.38-2.20 (1H, dd, J=8, 12, H9a)/2.20 (0.8H, dd, J=12, 20, H-9b'); NMR- ${ }^{13}$ C ( $75 \mathrm{MHz}, \mathrm{DMSO}_{6}$ ) $\delta(\mathrm{ppm}): 173.3 / 167.6$ (C-10); 157.4/154.2 (C-4); 149.3/146.0 (C-2); 147.2/144.0 (C-1); 143.2/139.9 (C-13); 135.1 (C-14); 132.0 (C-18); 131.8 (C-17); 131.0 (C-15); 128.1 (C-19); 126.4 (C-16); 123.3 (C-5); 119.4 (C-3); 105.5 (C-6); 49.0/48.6 (C-8); 42.1 (C-7), 35.9 (C-9).

### 1.6.4. (E)-N'-((1H-imidazol-4-yl)methylene)-3-amino-4-(2,4,5-trifluorophenyl) butanehydrazide: LASSBio-2126 (9)


(9)

After completion of the reaction (12h), the residue was concentrated and directly subjected to column chromatography in reverse phase (C-18) starting with MeCN $100 \%$, followed by $\mathrm{MeCN} / \mathrm{H}_{2} \mathrm{O}(\mathrm{pH} 3-4)$ 95:5 $\rightarrow \mathrm{MeCN} / \mathrm{H}_{2} \mathrm{O}(\mathrm{pH} 3-4) 40: 60$ yielding compound 9.
Appearance: white solid; Yield: 42\%; TLC conditions: DCM/MeOH/ $\mathrm{NH}_{4} \mathrm{OH}$ 90:10:0.05 (Rf: 0.45 ); Purity: 95.2\%; mp: $145-147^{\circ} \mathrm{C}$; MS (ESI): $[\mathrm{M}+\mathrm{H}]^{+}, 325,99 ;$ NMR- ${ }^{1} \mathrm{H}\left(300 \mathrm{MHz}, \mathrm{DMSO}-\mathrm{d}_{6}\right)$ $\delta(\mathrm{ppm}): 11.1$ (2H, bs, H-imidazole); 8.10 ( $\left.0,9 \mathrm{H}, \mathrm{s}, \mathrm{H}-13^{\prime}\right) / 7.91$ (1H, s, H-13); 7.62 (1H, s, H-22); 7.48-7.32 (5 H, m, H-20', H-3, H-6)/7.20 (1H, s, H-20); 3.46-3.31 (m, H-8); 2.28 ( $1 \mathrm{H}, \mathrm{dd}, \mathrm{J}=3.0$, 6.0, H-9a')/2.17 (0.8H, dd, J=9.0, 9.0, H- 9b'); NMR- ${ }^{13} \mathrm{C}$ ( $75 \mathrm{MHz}, \mathrm{DMSO}_{6}$ ) $\delta(\mathrm{ppm})$ : 172.4/166.8 (C-10); 157.4/154.2 (C-4); 149.2/145.9 (C-2); 147.3/143.9 (C-1); 140.0/136.8 (C13); 135.4 (C-14); 123.2 (C-5); 131.8 (C-22), 120.6/121.0 (C-20); 119.4 (C-3); $105.3(\mathrm{C}-6) ;$ 48.9/48.2 (C-8), 42.9/41.0 (C-9), 35.4 (C-7)

## 2. Synthesis of 3-amino- $N^{\prime}$-(3,4-difluorobenzyl)-4-(2,4,5-trifluorophenyl)butanehydrazide: LASSBio-2127 (13)



Adapted from Gore et al (2010) ${ }^{4}$. In a 25 mL round-bottom flask equipped with a magnetic stirrer and under argon atmosphere, $100 \mathrm{mg}(0.270 \mathrm{mmol})$ of N -acylhydrazone 6 (LASSBio2124) was dissolved in 7 mL of MeOH , then, 12 equivalents of $\mathrm{NaBH}_{3} \mathrm{CN}$ were added and the solution was stirred for 15 min . Subsequently, AcOH was added until pH 3-4 and the reaction mixture was stirred until total consumption of the starting material (monitored by TLC). Next, the solvent was distillated, and the residue was treated with $\mathrm{HCl} 10 \%$ until pH 2 . The resulting suspension was stirred for 15 min and extracted with DCM ( $3 \times 20 \mathrm{~mL}$ ). The organic phase was separated, and the aqueous phase was basified with $\mathrm{NaOH} 10 \%$ until pH 8-9. Then, the product was isolated with EtOAc ( $3 \times 50 \mathrm{~mL}$ ). The organic phase was dried over anhydrous $\mathrm{Na}_{2} \mathrm{SO}_{4}$, filtrated and concentrated. The resulting oil was purified by column chromatography (Hexane/AcOEt 1:1 $\rightarrow$ AcOEt/MeOH 9:1) yielding compound 13.
Appearance: transparent oil; Yield: 60\%; Purity: 95.7\%; MS (ESI): [M+H] ${ }^{+}, 374.18 ;$ NMR- ${ }^{1} \mathrm{H}$ ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta(\mathrm{ppm}): 7.19-7.01$ ( $4 \mathrm{H}, \mathrm{m}, \mathrm{H}-6, \mathrm{H}-15, \mathrm{H}-16, \mathrm{H}-19$ ); 6.93-6.83 (1H, m, H-3); $3.90(2 \mathrm{H}, \mathrm{s}, \mathrm{H}-13) ; 3.70(1 \mathrm{H}, \mathrm{m}, \mathrm{H}-8) ; 2.96-2.84(2 \mathrm{H}, \mathrm{m}, \mathrm{H}-7) ; 2.47-2.38(2 \mathrm{H}, \mathrm{m}, \mathrm{H}-9) ;$ NMR- ${ }^{13} \mathrm{C}$ ( $75 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta(\mathrm{ppm}): 170.1$ (C-10); 158.1/154.7 (C-4); 152.2/148.4 (C-18); 151.0/148.6 (C-

## 3. Synthesis of LASSBio-2128 (14)

### 3.1.Synthesis of $N, 3$-bis(1,3-dioxoisoindolin-2-yl)-4-(2,4,5-trifluorophenyl)butanamide (22a)



Adapted from Rodrigues et al. (2016) ${ }^{7}$. In a 10 mL round-bottom flask, equipped with a magnetic stirrer, 130 mg ( 0.526 mmol ) of hydrazide 21 were added and temperature was adjusted to $80^{\circ} \mathrm{C}$. Next, four equivalents of phthalic anhydride were added without any added solvent. The reaction mixture was stirred at $130^{\circ} \mathrm{C}$ for 4 h (monitored by TLC). Subsequently, the reaction was cooled, $\mathrm{Na}_{2} \mathrm{CO}_{3}(10 \%)$ solution was added, and the suspension was sonicated for 15 min . Then, the solid was filtered and washed with water and hexane to yield 22a.
Appearance: white solid; Yield: 86\%; mp: 223-2250 ; MS (ESI): [M+Na] ${ }^{+}, 530.20 ; \mathrm{NMR}^{\mathbf{1}} \mathrm{H}$ ( $300 \mathrm{MHz}, \mathrm{DMSO}_{\mathrm{d}}$ ) $\delta(\mathrm{ppm}): 8.18$ ( $1 \mathrm{H}, \mathrm{bs}, \mathrm{H}-17$ ); 7.80-7.82 ( $8 \mathrm{H}, \mathrm{m}, \mathrm{H}-25-32$ ); 7.49-7.30 ( 2 H , m, H-6, H-3); $4.80(1 \mathrm{H}, \mathrm{m}, \mathrm{H}-8)$; 3.35 ( 2 H , overlapped with water signal, $\mathrm{H}-7 \mathrm{a}, \mathrm{H}-7 \mathrm{~b}$ ); 3.09-3.03 (2H, m, H-9a, H-9b).

### 3.2.Synthesis of $\quad N, 3-b i s(1,3-d i o x o i s o i n d o l i n-2-y l)-N$-methyl-4-(2,4,5-trifluorophenyl) butanamide (22b)



Adapted from Rodrigues et al. (2016) ${ }^{7}$. In a 10 mL round-bottom flask, equipped with a magnetic stirrer, 230 mg ( 0.454 mmol ) of phthalimide 22a and 3 equivalents of $\mathrm{K}_{2} \mathrm{CO}_{3}$ were suspended in 10 mL of acetone. The suspension was thoroughly mixed under vigorous stirring for 5 min , and 2 equivalents of methyl iodide were subsequently added. The reaction was heated at $50^{\circ} \mathrm{C}$ and maintained under stirring overnight. At the end of the reaction (monitored by TLC), the solvent was evaporated under reduced pressure, and the residual solid was suspended in 2 mL of ethanol and poured into cold water. The solid was collected through filtration and dried under vacuum for 16h to yield 22b.
Appearance: white solid; Yield: 60\%; Melting Point: 216-218 ${ }^{\circ}$; NMR- ${ }^{1} \mathrm{H}$ ( 300 MHz, DMSO-d ${ }_{6}$ ) $\delta(\mathrm{ppm}): 7.77-8.04(8 \mathrm{H}, \mathrm{m}, \mathrm{H}-25-32) ; 7.28-7.44(2 \mathrm{H}, \mathrm{m}, \mathrm{H}-3, \mathrm{H}-6) ; 4.70(1 \mathrm{H}, \mathrm{m}, \mathrm{H}-8) ; 3.19-3.32$
( $2 \mathrm{H}, \mathrm{m}, \mathrm{H}-7 \mathrm{a}, \mathrm{H}-7 \mathrm{~b}$ ); $3.10(3 \mathrm{H}, \mathrm{s}, \mathrm{H}-17) ; 2.76(1 \mathrm{H}, \mathrm{dd}, \mathrm{J}=4.0,16.0, \mathrm{H}-9 \mathrm{a}) ; 2.95$ (1H, dd, J=8.0, 16.0, H-9b).

### 3.3.Synthesis of 3-amino-N-methyl-4-(2,4,5-trifluorophenyl)butanehydrazide (22)




Adapted from Khan et al (1995) ${ }^{8}$ and Rodrigues et al. (2016) ${ }^{7}$. In a 10 mL round-bottom flask, equipped with a magnetic stirrer and reflux condenser, $126 \mathrm{mg}(0.161 \mathrm{mmol})$ of phthalimide 22b was dissolved in 7 mL of MeOH , followed by the addition of 10 equivalents of hydrazine hydroxide. The reaction mixture was stirred at $70^{\circ} \mathrm{C}$ for 8 h . After completion of the reaction (monitored by TLC) the solvent was completely distilled and the residue was treated with 10 mL of a saturated $\mathrm{Na}_{2} \mathrm{CO}_{3}$ solution, then it was partitioned with $\mathrm{DCM}(5 \times 12 \mathrm{~mL})$. The organic extracts were washed with water and brine, subsequently dried over sodium sulfate, filtrated and concentrated under reduced pressure to yield 22. Appearance: yellow oil; Yield: 92\%; MS (ESI): $[\mathrm{M}+\mathrm{H}]^{+}, 262.25$.

### 3.4.Synthesis of 3-amino- $N^{\prime}$-(3,4-difluorobenzyl)- $N$-methyl-4-(2,4,5-trifluorophenyl) butanehydrazide: LASSBio-2128 (14)


(22)

$\mathrm{rt}, 12 \mathrm{~h}$
$89 \%$

(14)

Adapted from Kümmerle et al. (2012) ${ }^{6}$. In a 25 mL round-bottom flask equipped with a magnetic stirrer, 1 mmol of 3-amino- $N$-methyl-4-(2,4,5-trifluorophenyl)butanehydrazide 22 was dissolved in 5 mL of EtOH . After 15 min of stirring 1.1 equivalents of 3,4difluorobenzaldehyde was added. The reaction mixture was stirred at room temperature until TLC analysis showed complete consumption of the hydrazide (12h). Subsequently, the solvent was completely distillated. The product was, partitioned with water and EtOAc. The organic phase was washed with brine ( $3 \times 30 \mathrm{~mL}$ ), dried with anhydrous $\mathrm{Na}_{2} \mathrm{SO}_{4}$, filtered, and concentrated under vacuum. The product was purified by column chromatography (Hexane/AcOEt 9:1 $\rightarrow$ AcOEt/MeOH 9:1) yielding compounds 14.

Appearance: yellow oil; Yield: 89\%; Purity: 98.3\%; MS (ESI): [M+H] ${ }^{+}$, 385.12; NMR- ${ }^{\mathbf{H}} \mathbf{H}$ (300 $\left.\mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}\right) \delta(\mathrm{ppm}): 7.87(1 \mathrm{H}, \mathrm{s}, \mathrm{H}-13) ; 7.62-7.56(1 \mathrm{H}, \mathrm{m}, \mathrm{H}-19) ; 7.49-7.43(1 \mathrm{H}, \mathrm{m}, \mathrm{H}-15)$; 7.35-7.26 (2H, m, H-6, H-16); 7,19-7.10 (1H, m, H-3); 3.79 ( $1 \mathrm{H}, \mathrm{m}, \mathrm{H}-8$ ); 3.34 ( $3 \mathrm{H}, \mathrm{s}, \mathrm{H}-27$ ); 3.23 (1H, m, H-9a); 3.06-2.97 (3H, m, H-9b, H-7a, H-7b); NMR- ${ }^{13}$ C ( $75 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}$ ) $\delta(\mathrm{ppm}): 173.7$ (C-10); 159.4/156.2 (C-4); 154.3/150.8 (C-18); 153.7/150.2 (C-17); 152.2/148.9 (C-2);
149.7/146.4 (C-1); 140.3 (C-13); 133.8 (C-14); 125.4 (C-15); 121.9 (C-5); 120.6 (C-16); 118.6 (C6); 116.0 (C-19); 106.7 (C-3); 50.0 (C-8); 38.0 (C-9); 34.5 (C-7); 28.2 (C-27).
a)


b)


Figure S9. a) ${ }^{1} \mathrm{H}-\mathrm{NMR}$ NOE-diff ( $400 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}$ ) spectrum of 14 . Signal irradiated $\delta_{H} 7.86$ ( $\mathrm{N}=\mathrm{CH}$ ). b) ${ }^{1} \mathrm{H}-\mathrm{NMR}$ NOE-diff $\left(400 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}\right.$ ) spectrum of 14. Signal irradiated $\delta_{H} 3.36(\mathrm{~N}-$ $\mathrm{CH}_{3}$ ).

Table S5. Physical and spectroscopic properties of LASSBio-2123 to LASBio-2128 (5-7, 9, 13-14).

| Compound | R | Appearance | Overall <br> Yield (\%) | Purity <br> $(\%)$ | Melting <br> point ${ }^{\mathrm{b}}\left({ }^{\circ} \mathrm{C}\right)$ | MS-ESI ${ }^{\boldsymbol{c}}$ <br> $[\mathrm{M}+\mathrm{H}]^{+}$ <br> $(\mathrm{m} / \mathrm{z})$ | $\boldsymbol{\delta}_{\mathrm{H}}$ iminic | $\boldsymbol{\delta}_{\mathbf{C}}$ iminic |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| LASSBio-2123 (5) | $\mathrm{OCH}_{3}$ | Transparent Oil | 29.3 | 95.7 | - | 396.01 | $8.06 / 7.87$ | $146.1 / 142.6$ |
| LASSBio-2124 (6) | F | Transparent Oil | 31.4 | 99.4 | - | 372.15 | $8.13 / 7.91$ | $143.7 / 140.2$ |
| LASSBio-2125 (7) | Cl | Transparent Oil | 29.3 | 99.7 | - | 404.09 | $8.18 / 7.97$ | $143.2 / 139.9$ |
| LASSBio-2126 (9) | - | White Solid | 17.6 | 95.2 | $145-147$ | 325.99 | $8.10 / 7.91$ | $140.0 / 136.8$ |
| LASSBio-2127 (13) | - | Light yellow oil | 18.2 | 95.7 | - | 374.18 | 3.90 | 54.6 |
| LASSBio-2128(14) | - | Light yellow <br> solid | 18.4 | 98.3 | $81-83$ | 386.18 | 7.87 | 140.3 |

${ }^{\text {a }}$ Relative purity determined by HPLC. ${ }^{\text {b }}$ Melting point determined on an apparatus Melt Point System MP70, Mettler Toledo (values obtained were not corrected). ${ }^{\text {c }}$ Direct injection.

## 4. Enantiomeric Synthesis of $N$-acylhydrazones (6-R and 6-S)

### 4.1. Resolution of the $\beta$-aminoester intermediate ( $20-\mathrm{R}$ and $20-S$ )



Adapted from Gore et al (2010) ${ }^{4}$. In a 50 mL round-bottom flask equipped with a magnetic stirrer, 1.62 mmol of $\beta$-Aminoester (20) was dissolved in 5 mL of isopropanol (IPA). Then, 2 equivalents of $(R)$-mandelic acid dissolved in 6 mL of IPA were added to the mixture reaction. The resulting solution was stirred at $25-30^{\circ} \mathrm{C}$ for 5 h . At the end of the reaction, the white precipitate formed was filtered and washed with 12 mL of cold IPA. The supernatant was concentrated under reduce pressure, dissolved in 40 mL of water, and treated with $\mathrm{Na}_{2} \mathrm{CO}_{3}$ (10\%) until pH 9. Subsequently, it was extracted with EtOAc ( $3 \times 40 \mathrm{~mL}$ ), dried with anhydrous $\mathrm{Na}_{2} \mathrm{SO}_{4}$, filtered, and concentrated under vacuum to yield a transparent oil 20-(S).
The solid obtained from the filtration, was dried in a desiccator for 24 and weighted (Rend. $89 \%$ ), next, it was subjected to recrystallization in a mixture IPA/water ( $18 \mathrm{~mL} / 1 \mathrm{~mL}$ ) at 25 $30^{\circ} \mathrm{C}$. The mixture was heated at $75-80^{\circ} \mathrm{C}$ for 1 h , and it was cooled gradually until room temperature. The white precipitate was filtered, washed with 5 mL of cold IPA, dried in a desiccator for 24 h and weight (Rend. $66 \%$ ). Subsequently, the solid was dissolved in 30 mL of water and treated with $\mathrm{Na}_{2} \mathrm{CO}_{3}(10 \%)$ until pH 9. This solution was extracted with EtOAc ( $3 \times 40$ mL ), dried with anhydrous $\mathrm{Na}_{2} \mathrm{SO}_{4}$, filtered, and concentrated under vacuum to yield a transparent oil 20-(R).

### 4.1.1. (R)-methyl 3-amino-4-(2,4,5-trifluorophenyl)butanoate (20-(R))



Appearance: transparent oil; Yield: 66\%; TLC conditions: Same as racemate; MS (ESI): Same as racemate; $\mathbf{N M R}^{-1} \mathbf{H}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta(\mathrm{ppm})$ : Same as racemate; $\mathbf{N M R} \mathbf{- ~}^{13} \mathbf{C}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta$ (ppm): Same as racemate; Optical Rotation ([ $\alpha]_{\mathrm{D}}{ }^{25}$ ): - 6.83 (c 0.3 in MeOH); ee > 99\%.

### 4.1.2. (S)-methyl 3-amino-4-(2,4,5-trifluorophenyl)butanoate (20-(S))



Appearance: transparent oil; Yield: 89\%; TLC conditions: Same as racemate; MS (ESI): Same as racemate; $\mathbf{N M R} \mathbf{- 1}^{1} \mathbf{H}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta(\mathrm{ppm})$ : Same as racemate; $\mathbf{N M R}^{13} \mathbf{C}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta$ (ppm): Same as racemate; Optical Rotation ( $[\alpha]_{\mathrm{D}}{ }^{25}$ ): + 9.75 (c 0.3 in MeOH); ee: $82 \%$.

### 4.2. Determination of absolute stereochemistry of $\boldsymbol{\beta}$-amino ester (20-R) and (20-S)

4.2.1. Tasnadi and colleagues (2010) ${ }^{9}$ method evaluating carboxylic $\beta$-aminoacids (20a) and (20b)


To establish a methodology useful to determine the absolute stereochemistry of target compounds, we selected $\beta$-aminoester intermediate ( $\mathbf{2 0} \boldsymbol{-} \boldsymbol{R}, \boldsymbol{S}$ ) as a template. This selection was made considering we could easily convert esters (20-R,S) in the carboxylic $\beta$-aminoacids (20a and 20b), which absolute stereochemistry has been previously reported Tasnadi and colleagues (2010) ${ }^{9}$. Therefore, once synthesized the ( $R$ )- and (S)- $\beta$ aminoacid (20a and 20b), it would be feasible to compare their specific rotation values with those previously described by Tasnadi and co-workers (2010).
In a 10 mL round-bottom flask equipped with a reflux condenser and a magnetic stirrer, 25 mg ( 0.0929 mmol ) of the corresponding $\beta$-aminoester's 20-( $\boldsymbol{R}^{*}$ ) and 20-( $\left.\boldsymbol{S}^{*}\right)$ were added to 2.5 mL of $18 \%$ aqueous HCl . Then, the reaction mixture was stirred at reflux overnight. When TLC analysis showed total consumption of the starting material, the solvent was azeotropically distilled under reduced pressure and the residue dried for 24h to give compounds 20a and 20b.
Once obtained, the carboxylic ( $R$ )- and (S)- $\beta$-aminoacids were characterized by ${ }^{1} \mathrm{H}$ - and ${ }^{13} \mathrm{C}-\mathrm{NMR}$, and the specific rotation of the $\beta$-amino acids 20a and 20b was measured. The $\left(R^{*}\right)$-1-carboxy-3-(2,4,5-trifluorophenyl)propan-2-aminium chloride 20a has shown $[\alpha]_{D}{ }^{25}=-7.42$ (c 0.3 in $\mathrm{H}_{2} \mathrm{O}$ ); while compound ( $S^{*}$ )-1-carboxy-3-(2,4,5-trifluorophenyl)propan-2-aminium chloride 20b has displayed $[\alpha]_{D}{ }^{25}=+8.87$ (c 0.3 in $\mathrm{H}_{2} \mathrm{O}$ ). Surprisingly, these results contrasted to those reported by Tasnadi and coworkers $(2010)^{9}$, who have reported a specific rotation of $[\alpha]_{D}{ }^{25}=+10.6$ (c 0.3 in $\mathrm{H}_{2} \mathrm{O}$ ) to (R)-1-carboxy-3-(2,4,5-trifluorophenyl)propan-2-aminium chloride (20a) and $[\alpha]_{D}{ }^{25}=-9.8$ (c 0.3 in $\mathrm{H}_{2} \mathrm{O}$ ) to (S)-1-carboxy-3-(2,4,5-trifluorophenyl)propan-2-aminium chloride (20b).
4.2.1.1. (R)-1-carboxy-3-(2,4,5-trifluorophenyl)propan-2-aminium chloride (20a)


Chemical Formula: $\mathrm{C}_{10} \mathrm{H}_{11} \mathrm{ClF}_{3} \mathrm{NO}_{2}$; Appearance: white solid; Yield: 90\%; Purity: 95.2\%; NMR-1 H ( $400 \mathrm{MHz}, \mathrm{D}_{2} \mathrm{O}$ ) $\delta(\mathrm{ppm}): 7.30-7.25(1 \mathrm{H}, \mathrm{m}, \mathrm{H}-3) ; 7.19-7.14(1 \mathrm{H}, \mathrm{m}, \mathrm{H}-6) ; 3.95-3.90(1 \mathrm{H}, \mathrm{m}, \mathrm{H}-8)$; 3.08 ( $1 \mathrm{H}, \mathrm{dd}, J=8.0,12.0, \mathrm{H}-13 \mathrm{a}$ ); $3.03(1 \mathrm{H}, \mathrm{dd}, J=4.0,12.0, \mathrm{H}-13 \mathrm{~b}) ; 2.80(1 \mathrm{H}, \mathrm{dd}, J=8.0,12.0$,
$\mathrm{H}-7 \mathrm{a}) ; 2.72$ (1H, dd, J = 8.0, 16.0, H-7b); NMR- ${ }^{13} \mathrm{C}\left(100 \mathrm{MHz}, \mathrm{D}_{2} \mathrm{O}\right) \boldsymbol{\delta}$ (ppm): 173.9 (C-14);
157.4/155.5 (C-4); 150.4/148.5 (C-2); 147.6/145.7 (d, C-1); 119.2 (C-3); 105.9 (C-6); 48.1 (C-8), 35.4 (C-7); 30.9 (C-13); Optical Rotation ( $[\alpha]_{\mathrm{D}}{ }^{25}$ ): - 7.42 (c 0.3 in $\mathrm{H}_{2} \mathrm{O}$ ).

### 4.2.1.2. (S)-1-carboxy-3-(2,4,5-trifluorophenyl)propan-2-aminium chloride (20b)



Chemical Formula: $\mathrm{C}_{10} \mathrm{H}_{11} \mathrm{ClF}_{3} \mathrm{NO}_{2}$; Appearance: white solid; Yield: 90\%; Purity: 95.2\%; NMR-1 $\mathbf{H}$ (400 MHz, $\left.\mathrm{D}_{2} \mathrm{O}\right) \delta(\mathrm{ppm}): 7.30-7.25(1 \mathrm{H}, \mathrm{m}, \mathrm{H}-3) ; 7.19-7.14(1 \mathrm{H}, \mathrm{m}, \mathrm{H}-6) ; 3.95-3.90(1 \mathrm{H}, \mathrm{m}, \mathrm{H}-8)$; 3.08 (1H, dd, J = 8.0, 12.0, H-13a); 3.03 ( $1 \mathrm{H}, \mathrm{dd}, J=8.0,12.0, \mathrm{H}-13 \mathrm{~b}$ ); $2.80(1 \mathrm{H}, \mathrm{dd}, J=4.0,16.0$, $\mathrm{H}-7 \mathrm{a}) ; 2.72$ ( $1 \mathrm{H}, \mathrm{dd}, J=8.0,16.0, \mathrm{H}-7 \mathrm{~b}) ;$ NMR- ${ }^{13} \mathrm{C}\left(100 \mathrm{MHz}, \mathrm{D}_{2} \mathrm{O}\right) \delta(\mathrm{ppm}): 173.9$ (C-14);
157.4/155.5 (C-4); 150.4/148.5 (C-2); 147.6/145.7 (d, C-1); 119.2 (C-3); 105.9 (C-6); 48.1 (C-8), 35.4 (C-7); 30.9 (C-13); Optical Rotation ([ $\alpha]_{\mathrm{D}}{ }^{25}$ ): + 8.87 (c 0.3 in MeOH).
4.2.2. Mosher's method by chiral derivatization to obtain (23) and (24)


Intrigued by those controversial data regarding the specific rotation, we decided to study another methodology to determine the absolute stereochemistry. The approach adopted was the so-called modified Mosher's method, which is based on the chemical shift differences obtained in the NMR spectra of the diastereoisomeric products (usually esters and/or amides) obtained by reaction of the target substrate with chiral derivatizing reagents ${ }^{10,11}$. Accordingly, the methoxyphenylacetic acid (MPA) amides 23 and 24 were obtained separately through $N, N^{\prime}$-dicyclohexylcarbodiimide (DCC) coupling between ( $R$ )-MPA (25) and (S)-MPA (26) with the ( $R$ )- $\beta$-amino ester (20- $\mathbf{R}^{*}$ ). In a 10 mL round-bottom flask equipped with a magnetic stirrer, $20 \mathrm{mg}(0.0810 \mathrm{mmol})$ of the $\beta$ aminoester's (20-R) were treated with (R)-MPA (25) and (S)-MPA (26) acids separately in the presence of DCC ( 0.05 mmol ) in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(10 \mathrm{~mL})$. The reactions were monitored by TLC, purified by CC on silica gel (hexane 100\% - EtOAc 100\%), to yield amides 23 and 24.

Once obtained and characterized by ${ }^{1} \mathrm{H},{ }^{13} \mathrm{C}, \mathrm{HMBC}$ and HSQC NMR experiments, the absolute stereochemistry of intermediate $\left(\mathbf{2 0}-\mathbf{R}^{*}\right)$ has been determined stablishing the differences in the chemical shifts between the diastereoisomers ( $R$ )-MPA-( $R^{*}$ )- $\beta$-amino ester (23) and (S)-MPA-( $R^{*}$ )- $\beta$-amino ester (24), ( $\left.\Delta \delta^{R-S}\right)$, where it was found negative $\Delta \delta$ values for hydrogens $\mathrm{H}-3, \mathrm{H}-6, \mathrm{H}-7$ a and $\mathrm{H}-7 \mathrm{~b}$, and positive $\Delta \delta$ values for hydrogens $\mathrm{H}-8$, $\mathrm{H}-13 \mathrm{a}, \mathrm{H}-13 \mathrm{~b}$ and $\mathrm{H}-17$ (Table S 6 and Figures S 10 and S 11 ). Thus, it was possible to observe that the shielding effects of the aromatic ring fits with an $R$-configuration in the asymmetric centre and not with an $S$-configuration. These results allowed to propose an absolute configuration of $R$ for the asymmetric carbon 8.

Table S6. ${ }^{1} \mathrm{H}-\mathrm{NMR}$ data for the MPA amides of $\mathbf{2 3}$ and $\mathbf{2 4}$ at 400 MHz , in $\mathrm{CDCl}_{3} ; \delta$ in ppm , J in Hz .


| C \# | (R)-MPA-( $\mathbf{R}^{*}$ )-6aminoester (23) $\delta(H)$ | (S)-MPA-( $R^{*}$ )-6aminoester (24) $\delta(H)$ | $\Delta \delta^{R-S}$ |
| :---: | :---: | :---: | :---: |
| 1 | - | - | - |
| 2 | - | - | - |
| 3 | 6.88 | 7.00 | 0.12 |
| 4 | - | - | - |
| 5 | - | - | - |
| 6 | 6.77 | 6.89 | 0.12 |
| 7a | 2.84 | 2.93 | ${ }^{-}$ |
| 7b | 2.84 | 2.93 | ${ }^{-}$ |
| 8 | 4.47 | 4.38 | 0.09 |
| 9 | - | - | - |
| 10 | - | - | - |
| 11 | - | - | - |
| 12 | - | - | - |
| 13a | 2.64 | 2.56 | 0.08 |
| 13b | 2.57 | 2.54 | 0.03 |
| 14 | - | - | - |
| 15 | - | - | - |
| 16 | - | - | - |
| 17 | 3.71 | 3.62 | 0.09 |



Figure S10. Chemical shifts differences $\left(\Delta \delta^{R-S}\right)$ between diastereoisomers (R)-MPA-( $\left.R^{*}\right)$ - $\beta$ aminoester (23) and (S)-MPA-( $\mathrm{R}^{*}$ )- $\beta$-aminoester (24). Representation of the model that allowed the determination of absolute stereochemistry of $\mathbf{2 0}-\mathbf{R}^{*}$ as $\mathbf{R}$.

It is important to mention, that a suitable spatial disposition of the more stable conformer of the Mosher's amides is mandatory for the correct interpretation of the results obtained experimentally from this methodology. Hence, a conformational analysis of both MPA-amides 23 and 24 using a semiempirical model (PM3) in SPARTAN program was carried out. The in silico approach has supported the NMR results obtained here and, in consequence, the proposed absolute stereochemistry of the (R)methyl 3-amino-4-(2,4,5-trifluorophenyl)butanoate (20-R).



Figure S11. a) Energetic rotation profile of $\mathrm{H}-\mathrm{N}-\mathrm{C}-\mathrm{H}$ bond (blue bond for diastereoisomer 23 and orange bond for diastereoisomer 24) obtained by semi-empirical method PM3 in SPARTAN program. b) Favored conformers for diastereoisomers 23 and 24 obtained by semi-empirical method PM3 in SPARTAN program.
4.2.2.1. (R)-methyl 3-((R)-2-methoxy-2-phenylacetamido)-4-(2,4,5-trifluorophenyl)butanoate
(23)


Chemical Formula: $\mathrm{C}_{20} \mathrm{H}_{20} \mathrm{~F}_{3} \mathrm{~N}_{5} \mathrm{O}_{4}$; Appearance: white solid; Yield: 91\%; HPLC Purity: 95.2\%; NMR- ${ }^{1} \mathrm{H}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta(\mathrm{ppm}): 7.31-7.28(3 \mathrm{H}, \mathrm{m}, \mathrm{H}-28, \mathrm{H}-27, \mathrm{H}-26) ; 7.24-7.22(2 \mathrm{H}, \mathrm{m}, \mathrm{H}-4$, $\mathrm{H}-25)$; 6.90-6.85 (1H, m, H-6); 6.80-6.74 (1H, m, H-3); $4.53(1 \mathrm{H}, \mathrm{s}, \mathrm{H}-20) ; 4.51-4.44$ ( $1 \mathrm{H}, \mathrm{m}, \mathrm{H}-1$ ); 3.71 (3H, s, H-15); 3.33 (3H, s, H-29); 2.84 ( $2 \mathrm{H}, \mathrm{d}, \mathrm{J}=4.0, \mathrm{H}-10$ ) 2.64 ( $1 \mathrm{H}, \mathrm{dd}, \mathrm{J}=4.0,16.0, \mathrm{H}-12 \mathrm{a}$ ); 2.57 (1H, dd, J=4.0, 12.0, H-12b); NMR- ${ }^{13} \mathrm{C}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta(\mathrm{ppm}): 171.8$ (C-13); 170.4 (C18); 136.9 (C-22); 128.5, (C-24, C-25); 126.7 (C-26, C-27, C-28); 120.9 (C-5); 119.0 (C-6); 105.4 (C-3), 83.7 (C-20); 57.4 (C-29); 52.0 (C-15); 45.8 (C-11); 37.9 (C-12); 32.7 (C-10).

### 4.2.2.2. (R)-methyl 3-((S)-2-methoxy-2-phenylacetamido)-4-(2,4,5-trifluorophenyl)butanoate

(24)


Chemical Formula: $\mathrm{C}_{20} \mathrm{H}_{20} \mathrm{~F}_{3} \mathrm{~N}_{5} \mathrm{O}_{4}$; Appearance: white solid; Yield: 89\%; HPLC Purity: 95.2\%; NMR- ${ }^{1} \mathrm{H}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta(\mathrm{ppm}): 7.33-7.31(4 \mathrm{H}, \mathrm{m}, \mathrm{H}-28, \mathrm{H}-27, \mathrm{H}-26, \mathrm{H}-24) ; 7.24-7.22(1 \mathrm{H}, \mathrm{d}$, J=8.0, H-25); 7.03-6.97 (1H, m, H-6); 6.92-6.86 (1H, m, H-3); $4.52(1 \mathrm{H}, \mathrm{s}, \mathrm{H}-20) ; 4.43-4.34(1 \mathrm{H}$, m, H-11); 3.62 (3H, s, H-15); 3.32 (3H, s, H-29); 2.93 (2H, m, H-7a, H-10b) 2.56 (1H, dd, J=8.0, 16.0, $\mathrm{H}-12 \mathrm{a}$ ), 2.54 ( $1 \mathrm{H}, \mathrm{dd}, \mathrm{J}=4.0,12.0, \mathrm{H}-12 \mathrm{~b}$ ); $\mathrm{NMR}^{-13}{ }^{13}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta(\mathrm{ppm}): 171.7$ (C13); 170.4 (C-18); 137.0 (C-22); 128.6 (C-24, C-25); 126.9 (C-26, C-27, C-28); 121.2 (C-5); 119.2 (C-6); 105.5 (C-3); 83.8 (C-20); 57.4 (C-29); 51.9 (C-15); 46.5 (C-11); 37.5 (C-12); 32.6 (C-10).

### 4.3.Synthesis of $(S)$ and ( $R$ )-hydrazide intermediates (21-R, 21-S)



Compounds were obtained according to Kümmerle et al (2012) ${ }^{6}$ by the synthetic procedure previously reported on item 1.5 (compound 21).

### 4.3.1. ( $R$ )-3-amino-4-(2,4,5-trifluorophenyl)butanehydrazide (21-R)



Chemical Formula: $\mathrm{C}_{10} \mathrm{H}_{12} \mathrm{~F}_{3} \mathrm{~N}_{3} \mathrm{O}$; Appearance: transparent oil; Yield: 90\%; MS (ESI): same as racemate; $\mathbf{N M R -}{ }^{\mathbf{1}} \mathbf{H}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta(\mathrm{ppm})$ : same as racemate; $\mathbf{N M R}^{-13} \mathbf{C}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta$ (ppm): same as racemate; Optical Rotation ([ $\alpha]_{\mathrm{D}}{ }^{25}$ ): - 6.21 (c 0.3 in $\mathrm{H}_{2} \mathrm{O}$ ).

### 4.3.2. (S)-3-amino-4-(2,4,5-trifluorophenyl)butanehydrazide (21-S)



Chemical Formula: $\mathrm{C}_{10} \mathrm{H}_{12} \mathrm{~F}_{3} \mathrm{~N}_{3} \mathrm{O}$; Appearance: transparent oil; Yield: 90\%; MS (ESI): same as racemate; $\mathbf{N M R - 1} \mathbf{H}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta(\mathrm{ppm})$ : same as racemate.; $\mathbf{N M R -}{ }^{13} \mathbf{C}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta$ (ppm): same as racemate; Optical Rotation ([ $\alpha]_{\mathrm{D}}{ }^{25}$ ): + 9.73 (c 0.3 in MeOH).
4.4. General Synthetic Procedure for the (R)- (S)- $N$-acylhydrazones: LASSBio-2129 (6-(R)) and LASSBio-2130 (6-(S)) (KUMMERLE et al, 2012)



Compounds were obtained according to Kümmerle et al (2012) ${ }^{6}$ by the synthetic procedure previously reported on item 1.6 (compounds 5-7, 9).

### 4.4.1. ( $R, E$ )-3-amino-N'-(3,4-difluorobenzylidene)-4-(2,4,5-trifluorophenyl)butanehydrazide:

 LASSBio-2129 (6-(R))

Appearance: same as racemate; Yield: 75\%; HPLC purity: 98.8\%; MS (ESI): same as racemate; NMR- ${ }^{1} \mathbf{H}\left(300 \mathrm{MHz}, \mathrm{DMSO}-\mathrm{d}_{6}\right) \delta(\mathrm{ppm}):$ same as racemate; $\mathrm{NMR}^{13} \mathrm{C}\left(75 \mathrm{MHz}, \mathrm{DMSO}-\mathrm{d}_{6}\right) \delta$ (ppm): same as racemate; Optical Rotation ([ $\boldsymbol{\alpha}]_{\mathrm{D}}{ }^{\mathbf{2 5}}$ ): - 21.18 (c 0.3 in MeOH ); ee > 99.9 \%.

### 4.4.2. (S,E)-3-amino-N'-(3,4-difluorobenzylidene)-4-(2,4,5-trifluorophenyl)butanehydrazide: LASSBio-2130 (6-(S))


(6-S)
Appearance: same as racemate; Yield: 73\%; HPLC Purity: 98.2\%; MS (ESI): same as racemate;
 (ppm): same as racemate; Optical Rotation ([ $\boldsymbol{\alpha}]_{\mathrm{D}}{ }^{25}$ ): + 25.86 (c 0.3 in MeOH ).

## 5. Aqueous solubility determination

Solubility was measured by spectrophotometric assay. The solubility was determined by constructing a calibration curve as follows: 1 mg of analyte was dissolved in MeOH and transferred to a 20 mL volumetric glass (stock solution). Six solutions of different concentrations ( $200 \mu \mathrm{M}-0.5 \mu \mathrm{M}$ ) were prepared by dilution of the stock solution. Data were obtained in triplicate and the mean values were used for the graphs. The values of the correlation coefficient ( $\mathrm{R}^{2}$ ) were between 0.9972 and 0.9999 . Saturated aqueous solutions of the analytes were prepared at $0.01 \mathrm{mg} / \mathrm{ml}$ and kept at $37^{\circ} \mathrm{C}$ for 4 hours with stirring. The supernatant was filtered through 0.45 mm filters and transferred to a quartz cuvette ( 10 mm ) for spectral acquisition. Adapted method from Schneider et al. (2009) ${ }^{12}$.

## 6. Dissociation constant determination

Phosphate buffer solutions were prepared using $\mathrm{Na}_{2} \mathrm{HPO}_{4}(10 \mathrm{mM})$ and $\mathrm{NaH}_{2} \mathrm{PO}_{4}(10 \mathrm{mM})$. These buffers were first filtered through a $0.45 \mu \mathrm{~m}$ membrane filter and degassed in an ultrasonic bath before use. The mobile phases consisted of mixtures of acetonitrile-phosphate buffer 20:80 (v/v) and 30:70 (v/v) adjusted by the addition of $\mathrm{H}_{3} \mathrm{PO}_{4}(10 \mathrm{mM})$ or $\mathrm{NaOH}(10$ mM ). The ionic strength of the diluted buffers used was close to 1 . The final pH of the mobile phases was measured before and after each chromatographic run at $25^{\circ} \mathrm{C}$. The column was equilibrated by rinsing with the mobile phase at the defined pH for 15 min at a flow rate of 1.0 $\mathrm{mL} / \mathrm{min}$. Solutions containing the analytes were prepared in $\mathrm{MeCN} / \mathrm{H}_{2} \mathrm{O} 1: 1$ to a final concentration of $0.1 \mathrm{mg} / \mathrm{mL}$. A $20 \mu \mathrm{~L}$ injection of each compound was made in triplicate into the chromatographic system. The chromatographic separations were performed at a flow rate of $1 \mathrm{ml} / \mathrm{min}$. The dead time value ( t 0 ) was measured by injecting Uracile ( $0.1 \mathrm{mg} / \mathrm{ml}$ ), which was determined for each mobile phase composition and pH studied. For each mobile phase composition, the retention factor $k$ was calculated according to $k=(r t-t 0) / t 0$, where $r t$ and t0 are the retention time of the analyte and the dead time, respectively. Observed retention factors were plotted against mobile phase pH using a non-linear least squares program (Prism 3.0). Computer generated plots of k vs. pH were obtained and the pH at the inflection point (V50) was used as a valuable index of pKa. Sitagliptin was used as a positive control. Adapted method from Wiczling et al. (2004) ${ }^{13}$.

## 7. Chemical stability studies

Two microlitres ( 0.01 mmol ) of a concentrated solution of the compound of interest ( 40 mM stock solution solubilised in DMSO) and $248 \mu \mathrm{~L}$ acid ( 0.2 M potassium chloride and 0.2 M HCl ; pH 2.0) or neutral (phosphate dibasic, pH 7.4 ) buffer were added to a 2 mL Eppendorf microtube. After vortexing, the mixture was placed in a water bath at $37^{\circ} \mathrm{C}$ with vigorous stirring for $0,30,60,120$ and 240 min . After each reaction, $248 \mu \mathrm{~L}$ of basic buffer (phosphate
buffer, pH 8.4) was added to neutralise the pH of the medium in experiments using acidic buffer. The compound was extracted with 1 mL of acetonitrile, followed by vigorous vortexing and freezing of the aqueous phase $\left(-10^{\circ} \mathrm{C}\right)$. The organic phase was separated, filtered and analysed by HPLC-PDA (acetonitrile/water 1:1, $0.05 \%$ TFA). Adapted method from Konsoula et al. (2008) ${ }^{14}$.

## 8. In vitro DPP-4 inhibition evaluation

A fluorogenic assay was used to measure the activity of DPP-4 using a DPP-4 Inhibitor Screening Test Assay Kit (item number 700210) from Cayman Chemical (Miami, USA). Inhibition of human recombinant DPP-4 was measured using the chromogenic substrate H-Gly-Pro-AMC, which is cleaved by DPP-4 to release the fluorescent AMC leaving group, according to the following protocol: $30 \mu \mathrm{~L}$ buffer ( 20 mM Tris-HCl, pH 8, $100 \mathrm{mM} \mathrm{NaCl}, 1 \mathrm{mM}$ EDTA), 10 $\mu \mathrm{L}$ enzyme and $10 \mu \mathrm{~L}$ DMSO were added to three wells ( $100 \%$ activity wells). $40 \mu \mathrm{~L}$ buffer and $10 \mu \mathrm{~L}$ DMSO were added to three plates (background wells). $30 \mu \mathrm{~L}$ buffer, $10 \mu \mathrm{~L}$ enzyme and $10 \mu \mathrm{~L}$ inhibitor (compounds and sitagliptin as positive control) were added to 90 wells. The reaction was initiated by adding $50 \mu \mathrm{~L}$ of substrate solution ( 5 mM ) to all wells used, followed by incubation at $37^{\circ} \mathrm{C}$ for 30 minutes. DPP-4 activity, resulting in the formation of fluorescent aminomethylcoumarin, was monitored by excitation at 360 nm and emission at 465 nm . The inhibitors (compounds and sitagliptin) dissolved in DMSO were tested in five concentrations $(10 \mu \mathrm{M}, 1 \mu \mathrm{M}, 0.1 \mu \mathrm{M}, 0.01 \mu \mathrm{M}, 0.001 \mu \mathrm{M}$ and $0.0001 \mu \mathrm{M})$ and in triplicate. The percentage of inhibition was obtained according to the following formula $\%$ inhibition $=[$ (initial activityinhibitor)/(initial activity)] x 100.

## Supplementary Information: Analytical Spectra

(5-(1-hydroxy-2-(2,4,5-trifluorophenyl)ethylidene)-2,2-dimethyl-1,3-dioxane-4,6-dione) (17)


Methyl 3-oxo-4-(2,4,5-trifluorophenyl)butanoate (18)


H-15

(Z)-methyl 3-amino-4-(2,4,5-trifluorophenyl)but-2-enoate (19)



H-7
H-13
H-6


${ }^{1} \mathrm{H}-\mathrm{NMR}$ spectrum of $\mathbf{1 9}\left(400 \mathrm{MHz} / \mathrm{CDCl}_{3} / \mathrm{TMS} / 25^{\circ} \mathrm{C}\right)$


3-amino-4-(2,4,5-trifluorophenyl)butanoate (20)

8.1.3-amino-4-(2,4,5-trifluorophenyl)butanehydrazide (21)

(E)-3-amino- $\mathrm{N}^{\prime}$-(3,4-dimethoxybenzylidene)-4-(2,4,5-trifluorophenyl)butanehydrazide:




Expansion of ${ }^{1} \mathrm{H}-\mathrm{NMR}$ spectrum of $5\left(400 \mathrm{MHz} / \mathrm{DMSO}-\mathrm{d}_{6} / \mathrm{TMS} / 25^{\circ} \mathrm{C}\right)$



Expansion of ${ }^{1} \mathrm{H}-\mathrm{NMR}$ spectrum of $\mathbf{5}$ ( $\left.400 \mathrm{MHz} / \mathrm{DMSO}-\mathrm{d}_{6} / \mathrm{TMS} / \mathbf{9 0}{ }^{\circ} \mathbf{C}\right)$


Expansion of ${ }^{1} \mathrm{H}-\mathrm{NMR}$ spectrum of $5\left(400 \mathrm{MHz} / \mathrm{DMSO}-\mathrm{d}_{6} / \mathrm{TMS} / \mathbf{9 0}^{\circ} \mathbf{C}\right)$





## APT of $5\left(100 \mathrm{MHz} / \mathrm{DMSO}-\mathrm{d}_{6} / \mathrm{TMS} / 25^{\circ} \mathrm{C}\right)$



Expansion of APT of $\mathbf{5}\left(100 \mathrm{MHz} /\right.$ DMSO $\left.-\mathrm{d}_{6} / \mathrm{TMS} / 25^{\circ} \mathrm{C}\right)$



Expansion of APT of $\mathbf{5}\left(100 \mathrm{MHz} / \mathrm{DMSO}-\mathrm{d}_{6} / \mathrm{TMS} / 25^{\circ} \mathrm{C}\right)$


HSQC of 5 ( DMSO $^{-d_{6} / T M S / 25^{\circ}}$ )


Expansion of HSQC of $5\left(\mathrm{DMSO}-\mathrm{d}_{6} / \mathrm{TMS} / 25^{\circ} \mathrm{C}\right)$


Expansion of HSQC of 5 (DMSO-d $\left.\mathrm{d}_{6} / \mathrm{TMS} / 25^{\circ} \mathrm{C}\right)$


Mass spectrum of 5 in positive mode (ESI, direct injection)


| Detector A 254nm |
| :--- |
| Peak\# Ret. Time Area Height Height $\%$ <br> 1 1,486 130111 18387 2,571 <br> 2 1,910 67947 7533 1,053 <br> 3 2,113 31535 4197 0,587 <br> 4 4,305 18229542 670704 93,788 <br> 5 5,876 594223 14308 2,0337 <br> Total  19053358 715130 100,000 |

HPLC Purity: Kromasil Column C18 [4,6 mm x 250 mm ]; detector SPD-M20A [Diode Array]; flux: $1 \mathrm{~mL} / \mathrm{min}$; injection volume: $20 \mu \mathrm{~L}$, mobile phase: $6: 4 \mathrm{MeCN} / \mathrm{H}_{2} \mathrm{O}(\mathrm{pH}=$ 9).
(E)-3-amino-N'-(3,4-difluorobenzylidene)-4-(2,4,5-trifluorophenyl) butanehydrazide: LASSBio-2124 (6)



Expansion of ${ }^{1} \mathrm{H}-\mathrm{NMR}$ spectrum of $\mathbf{6}\left(300 \mathrm{MHz} / \mathrm{DMSO}-\mathrm{d}_{6} / \mathrm{TMS} / 25^{\circ} \mathrm{C}\right)$


Expansion of ${ }^{1} \mathrm{H}-\mathrm{NMR}$ spectrum of $6\left(300 \mathrm{MHz} / \mathrm{DMSO}-\mathrm{d}_{6} / \mathrm{TMS} / 25^{\circ} \mathrm{C}\right)$


Expansion of ${ }^{1} \mathrm{H}-\mathrm{NMR}$ spectrum of $5\left(400 \mathrm{MHz} / \mathrm{DMSO}-\mathrm{d}_{6} / \mathrm{TMS} / \mathbf{9 0}{ }^{\circ} \mathbf{C}\right)$


Expansion of ${ }^{1} \mathrm{H}-\mathrm{NMR}$ spectrum of $\mathbf{5}\left(400 \mathrm{MHz} / \mathrm{DMSO}-\mathrm{d}_{6} / \mathrm{TMS} / \mathbf{9 0}{ }^{\circ} \mathbf{C}\right)$



${ }^{13} \mathrm{C}-\mathrm{NMR}$ of $\mathbf{6}\left(100 \mathrm{MHz} / \mathrm{CDCl}_{3} / \mathrm{TMS} / 25^{\circ} \mathrm{C}\right)$






Expansion of ${ }^{13} \mathrm{C}-\mathrm{NMR}$ of $\mathbf{6}\left(100 \mathrm{MHz} / \mathrm{CDCl}_{3} / \mathrm{TMS} / 25^{\circ} \mathrm{C}\right)$



Expansion of HSQC of $6\left(\mathrm{DMSO}-\mathrm{d}_{6} / \mathrm{TMS} / 25^{\circ} \mathrm{C}\right)$


PDA Ch1 281nm 4nm

| Peak\# | Ret. Time | Area | Height | Area $\%$ | Height $\%$ |
| ---: | ---: | ---: | ---: | ---: | ---: |
| 1 | 2.436 | 41020256 | 2448472 | 99.355 | 99.568 |
| 2 | 4.075 | 4113 | 301 | 0.010 | 0.012 |
| 3 | 4.512 | 1987 | 184 | 0.005 | 0.007 |

HPLC Purity: Kromasil Column C18 [4,6 mm x 250 mm ]; detector SPD-M20A [Diode Array]; flux: $1 \mathrm{~mL} / \mathrm{min}$; injection volume: $20 \mu \mathrm{~L}$, mobile phase: $6: 4 \mathrm{MeCN} / \mathrm{H}_{2} \mathrm{O}(\mathrm{pH}=$ 9).
(E)-3-amino- $\mathrm{N}^{\prime}$-(3,4-dichlorobenzylidene)-4-(2,4,5-trifluorophenyl)
butanehydrazide: LASSBio-2125 (7)



Expansion of ${ }^{1} \mathrm{H}-\mathrm{NMR}$ spectrum of $6\left(300 \mathrm{MHz} / \mathrm{DMSO}-\mathrm{d}_{6} / \mathrm{TMS} / 25^{\circ} \mathrm{C}\right)$


Expansion of ${ }^{1} \mathrm{H}-\mathrm{NMR}$ spectrum of $6\left(300 \mathrm{MHz} / \mathrm{DMSO}-\mathrm{d}_{6} / \mathrm{TMS} / 25^{\circ} \mathrm{C}\right)$


Expansion of ${ }^{1} \mathrm{H}-\mathrm{NMR}$ spectrum of $7\left(400 \mathrm{MHz} / \mathrm{DMSO}-\mathrm{d}_{6} / \mathrm{TMS} / \mathbf{9 0}{ }^{\circ} \mathrm{C}\right)$


Expansion of ${ }^{1} \mathrm{H}-\mathrm{NMR}$ spectrum of 7 ( $\left.400 \mathrm{MHz} / \mathrm{DMSO}-\mathrm{d}_{6} / \mathrm{TMS} / \mathbf{9 0}^{\circ} \mathrm{C}\right)$






Expansion of HSQC of $7\left(\mathrm{DMSO}-\mathrm{d}_{6} / \mathrm{TMS} / 25^{\circ} \mathrm{C}\right)$



PDA Ch1 281nm 4nm

| Peak\# | Ret. Time | Area | Height | Area \% | Height \% |
| ---: | ---: | ---: | ---: | ---: | ---: |
| 1 | 2.436 | 41020256 | 2448472 | 99.355 | 99.568 |
| 2 | 4.075 | 4113 | 301 | 0.010 | 0.012 |
| 3 | 4.512 | 1987 | 184 | 0.005 | 0.007 |

HPLC Purity: Kromasil Column C18 [4,6 mm x 250 mm ]; detector SPD-M20A [Diode Array]; flux: $1 \mathrm{~mL} / \mathrm{min}$; injection volume: $20 \mu \mathrm{~L}$, mobile phase: $6: 4 \mathrm{MeCN} / \mathrm{H}_{2} \mathrm{O}(\mathrm{pH}=$ 9).
(E)-N'-((1H-imidazol-4-yl)methylene)-3-amino-4-(2,4,5-trifluorophenyl)butanehydrazide: LASSBio-2126 (9)



Expansion of ${ }^{1} \mathrm{H}-\mathrm{NMR}$ spectrum of $9\left(400 \mathrm{MHz} / \mathrm{DMSO}-\mathrm{d}_{6} / \mathrm{TMS} / 25^{\circ} \mathrm{C}\right)$


Expansion of ${ }^{1} \mathrm{H}-\mathrm{NMR}$ spectrum of $9\left(400 \mathrm{MHz} / \mathrm{DMSO}-\mathrm{d}_{6} / \mathrm{TMS} / 25^{\circ} \mathrm{C}\right)$




Expansion of APT of $9\left(100 \mathrm{MHz} / \mathrm{DMSO}-\mathrm{d}_{6} / \mathrm{TMS} / 25^{\circ} \mathrm{C}\right)$


Expansion of APT of $9\left(100 \mathrm{MHz} / \mathrm{DMSO}-\mathrm{d}_{6} / \mathrm{TMS} / 25^{\circ} \mathrm{C}\right)$



Expansion of HSQC of $9\left(\mathrm{DMSO}-\mathrm{d}_{6} / \mathrm{TMS} / 25^{\circ} \mathrm{C}\right)$


Expansion of HSQC of $9\left(\mathrm{DMSO}-\mathrm{d}_{6} / \mathrm{TMS} / 25^{\circ} \mathrm{C}\right)$

mAU PDMAli
PDACh1 254 nm 4 nm

| Peak\# | Ret. Time | Area | Height | Area $\%$ | Height $\%$ |
| ---: | ---: | ---: | ---: | ---: | ---: |
| 1 | 2.231 | 9084646 | 824921 | 95.231 | 97.022 |
| 2 | 3.628 | 454240 | 25234 | 4.762 | 2.968 |
| 3 | 5.013 | 716 | 84 | 0.008 | 0.010 |
| Total |  | 9539601 | 850239 | 100.000 | 100.000 |

HPLC Purity: Kromasil Column C18 [4,6 mm x 250 mm ]; detector SPD-M20A [Diode Array]; flux: $1 \mathrm{~mL} / \mathrm{min}$; injection volume: $20 \mu \mathrm{~L}$, mobile phase: $6: 4 \mathrm{MeCN} / \mathrm{H}_{2} \mathrm{O}(\mathrm{pH}=$ 9).

## 3-amino- $\mathrm{N}^{\prime}$-(3,4-difluorobenzyl)-4-(2,4,5-trifluorophenyl)butanehydrazide:

 LASSBio-2127 (13)

Expansion of ${ }^{1} \mathrm{H}-\mathrm{NMR}$ spectrum of $\mathbf{1 3}\left(400 \mathrm{MHz} / \mathrm{CDCl}_{3} / \mathrm{TMS} / 25^{\circ} \mathrm{C}\right)$


Expansion of ${ }^{1} \mathrm{H}-\mathrm{NMR}$ spectrum of $\mathbf{1 3}\left(400 \mathrm{MHz} / \mathrm{CDCl}_{3} / \mathrm{TMS} / 25^{\circ} \mathrm{C}\right)$



Expansion of APT of $\mathbf{1 3}\left(100 \mathrm{MHz} / \mathrm{CDCl}_{3} / \mathrm{TMS} / 25^{\circ} \mathrm{C}\right)$




mV


| Detector A 254nm |
| :--- |
| Peak\# Ret. Time Area Height Area\% Height\% Area/Height |
| 1 |

HPLC Purity: Kromasil Column C18 [4,6 mm x 250 mm ]; detector SPD-M20A [Diode Array]; flux: $1 \mathrm{~mL} / \mathrm{min}$; injection volume: $20 \mu \mathrm{~L}$, mobile phase: 6:4 MeCN/ $\mathrm{H}_{2} \mathrm{O}(\mathrm{pH}=$ 9).

## N,3-bis(1,3-dioxoisoindolin-2-yl)-4-(2,4,5-trifluorophenyl)butanamide (22a)


${ }^{1} \mathrm{H}-\mathrm{NMR}$ spectrum of $\mathbf{2 2 a}\left(400 \mathrm{MHz} / \mathrm{CDCl}_{3} / \mathrm{TMS} / 25^{\circ} \mathrm{C}\right)$


## N,3-bis(1,3-dioxoisoindolin-2-yl)-N-methyl-4-(2,4,5-trifluorophenyl)butanamide (22b)


${ }^{1} \mathrm{H}-\mathrm{NMR}$ spectrum of $\mathbf{2 2 b}\left(400 \mathrm{MHz} / \mathrm{CDCl}_{3} / \mathrm{TMS} / 25^{\circ} \mathrm{C}\right)$

## 3-amino-N-methyl-4-(2,4,5-trifluorophenyl)butanehydrazide (22)



## 3-amino- $N^{\prime}$-(3,4-difluorobenzyl)-N-methyl-4-(2,4,5-trifluorophenyl)butanehydrazide: LASSBio-2128 (14)



Expansion of ${ }^{1} \mathrm{H}-\mathrm{NMR}$ spectrum of $\mathbf{1 4}\left(400 \mathrm{MHz} / \mathrm{CD}_{3} \mathrm{OD} / \mathrm{TMS} / 25^{\circ} \mathrm{C}\right)$




Expansion of APT of $\mathbf{1 4}\left(100 \mathrm{MHz} / \mathrm{CD}_{3} \mathrm{OD} / \mathrm{TMS} / 25^{\circ} \mathrm{C}\right)$

a)
b)
a) ${ }^{1} \mathrm{H}-\mathrm{NMR}$ NOE-diff $\left(400 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}\right)$ spectrum of 14 . Signal irradiated $\delta H 7.86(\mathrm{~N}=\mathrm{CH})$.
b) ${ }^{1} \mathrm{H}-\mathrm{NMR}$ NOE-diff $\left(400 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}\right)$ spectrum of 14 . Signal irradiated $\delta H 3.36\left(\mathrm{~N}-\mathrm{CH}_{3}\right)$.



Detector A 266nm

| Peak\# | Ret. Time | Area | Height | Conc. | Area\% |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 1 | 1,275 | 1213 | 196 | 0,000 | 0,003 |
| 2 | 1,441 | 39974 | 7253 | 0,000 | 0,111 |
| 3 | 1,755 | 65529 | 5738 | 0,000 | 0,181 |
| 4 | 2,216 | 101399 | 5849 | 0,000 | 0,281 |
| 5 | 3,182 | 39385 | 1402 | 0,000 | 0,109 |
| 6 | 3,871 | 29020 | 1003 | 0,000 | 0,080 |
| 7 | 4,390 | 38983 | 1483 | 0,000 | 0,108 |
| 8 | 4,751 | 19263 | 915 | 0,000 | 0,053 |
| 9 | 5,117 | 6927 | 573 | 0,000 | 0,019 |
| 10 | 5,838 | 42894 | 1366 | 0,000 | 0,119 |
| 11 | 6,151 | 30310 | 1285 | 0,000 | 0,084 |
| 12 | 7,617 | 97735 | 2363 | 0,000 | 0,271 |
| 13 | 10,456 | 35498195 | 532302 | 0,000 | 98,305 |
| 14 | 18,716 | 99490 | 1085 | 0,000 | 0,276 |
| Total |  | 36110318 | 562814 |  | 100,000 |

HPLC Purity: Kromasil Column C18 [4,6 mm x 250 mm ]; detector SPD-M20A [Diode Array]; flux: $1 \mathrm{~mL} / \mathrm{min}$; injection volume: $20 \mu \mathrm{~L}$, mobile phase: $6: 4 \mathrm{MeCN} / \mathrm{H}_{2} \mathrm{O}(\mathrm{pH}=$ 9).

(S)-methyl 3-amino-4-(2,4,5-trifluorophenyl)butanoate (20-(S))

(R)-1-carboxy-3-(2,4,5-trifluorophenyl)propan-2-aminium chloride (20a)

${ }^{1} \mathrm{H}-\mathrm{NMR}$ spectrum of $\mathbf{2 0 a}\left(400 \mathrm{MHz} / \mathrm{CDCl}_{3} / \mathrm{TMS} / 25^{\circ} \mathrm{C}\right)$



| 210 | 190 | 170 | 150 | 130 | ${ }_{\mathrm{f} 1}^{110}(\mathrm{ppm})$ | 90 | 80 | 70 | 60 | 50 | 40 | 30 | 20 | 10 | 0 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |

${ }^{13} \mathrm{C}-\mathrm{NMR}$ and DEPT of $\mathbf{2 0 a}\left(100 \mathrm{MHz} / \mathrm{CDCl}_{3} / \mathrm{TMS} / 25^{\circ} \mathrm{C}\right)$

## (S)-1-carboxy-3-(2,4,5-trifluorophenyl)propan-2-aminium chloride (20b)


${ }^{13} \mathrm{C}$-NMR and DEPT of $\mathbf{2 0 b}\left(100 \mathrm{MHz} / \mathrm{CDCl}_{3} / \mathrm{TMS} / 25^{\circ} \mathrm{C}\right)$
(R)-methyl 3-((R)-2-methoxy-2-phenylacetamido)-4-(2,4,5-trifluorophenyl)butanoate (23)

${ }^{1} \mathrm{H}-\mathrm{NMR}$ spectrum of $23\left(400 \mathrm{MHz} / \mathrm{CDCl}_{3} / \mathrm{TMS} / 25^{\circ} \mathrm{C}\right)$




nis in eis
$\stackrel{\rightharpoonup}{\sim} \underset{\sim}{\sim}$

${ }^{13} \mathrm{C}-\mathrm{NMR}$ spectrum of $\mathbf{2 3}\left(100 \mathrm{MHz} / \mathrm{CDCl}_{3} / \mathrm{TMS} / 25^{\circ} \mathrm{C}\right)$


HSQC of $23\left(\mathrm{CDCl}_{3} / \mathrm{TMS} / 25^{\circ} \mathrm{C}\right)$

（R）－methyl 3－（（S）－2－methoxy－2－phenylacetamido）－4－（2，4，5－trifluorophenyl）butanoate（24）

${ }^{1} \mathrm{H}-\mathrm{NMR}$ spectrum of $\mathbf{2 4}\left(400 \mathrm{MHz} / \mathrm{CDCl}_{3} / \mathrm{TMS} / 25^{\circ} \mathrm{C}\right)$

| ®๐ั | जn ${ }_{\text {¢ }}$ |  |  |  | － |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 지국 |  | －${ }_{\sim}^{\sim}$ |  | がNべか | へin |
| $\bigcirc$ | － | － |  |  |  |


${ }^{13} \mathrm{C}-\mathrm{NMR}$ spectrum of $24\left(100 \mathrm{MHz} / \mathrm{CDCl}_{3} / \mathrm{TMS} / 25^{\circ} \mathrm{C}\right)$

( $R, S, E$ )-3-amino- $N^{\prime}$-(3,4-difluorobenzylidene)-4-(2,4,5-trifluorophenyl)butanehydrazide: LASSBio-2124 (6)

( $R, E$ )-3-amino- $N^{\prime}$-(3,4-difluorobenzylidene)-4-(2,4,5-trifluorophenyl)butanehydrazide: LASSBio-2129 (6-(R))

${ }^{1} \mathrm{H}-\mathrm{NMR}$ spectrum of $\mathbf{6 - ( \boldsymbol { R } )}\left(400 \mathrm{MHz} / \mathrm{CD}_{3} \mathrm{OD} / \mathrm{TMS} / 25^{\circ} \mathrm{C}\right)$
(S,E)-3-amino- $N^{\prime}$-(3,4-difluorobenzylidene)-4-(2,4,5-trifluorophenyl)butanehydrazide: LASSBio-2130 (6-(S))


## References

T. Zerilli and E. Y. Pyon, Clin Ther, 2007, 29, 2614-2634.

2 European Medicines Agency. Assessment Report For Januvia. London, 02 June 2009. Doc. Ref: EMEA/363653/2009 Available in: http://www.ema.europa.eu/docs/en_GB/document_library/EPAR_-_Assessment_Report__Variation/human/000722/WC500039129.pdf. Access in 02/05/2018.

3 P. Sun, Y. Chen, and G. YU, Process for preparing R-beta-amino phenylbutyric acid derivatives, US patent US2011/0130587 A1, 2011.

4 G. V. Govind et al, Sitagliptin synthesis, WO 2010131025 A1, 2010.
5 M. Kubryk and K. B. Hansen, Tetrahedron Asymmetry, 2006, 17, 205-209.
6 A. E. Kümmerle et al., J Med Chem, 2012, 55, 7525-7545.
7 D. A. Rodrigues et al., J Med Chem, 2016, 59, 655-670.
8 M. N. Khan, J Org Chem, 1995, 60, 4536-4541.
9 G. Tasnádi, E. Forró and F. Fülöp, Org. Biomol. Chem., 2010, 8, 793-799.
10 J. M. Seco, E. Quiñoá and R. Riguera, Chem Rev, 2004, 104, 17-118.
11 J. M. Seco, E. Quiñoá and R. Riguera, Tetrahedron Asymmetry, 2001, 12, 2915-2925.

12 P. Schneider, S. S. Hosseiny, M. Szczotka, V. Jordan and K. Schlitter, Phytochem Lett, 2009, 2, 85-87.
P. Wiczling, M. J. Markuszewski and R. Kaliszan, Anal Chem, 2004, 76, 3069-3077.

