## **Supplementary Materials for**

## **Degree of Disorder Regulated Ion Transport through Amorphous Monolayer Carbon**

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**Supplementary Figure 1.** AFM characterizations of AMC-400 (1a-1f) and AMC-600 (1g-1l), respectively, after transferring onto SiO<sub>2</sub>/Si substrates, showing a uniform thickness of  $\sim$  1 nm.



**Supplementary Figure 2.** *I*<sub>D</sub>/*I*<sub>G</sub> (area ratio) versus the synthesis temperatures for AMC samples. After analyzing multi-Raman spectra (> 50),  $I_D/I_G$  slightly increases with the synthesis temperature, but with a large variation and uncertainty for each data point.



**Supplementary Figure 3.** Full scan of XPS on AMC-400, AMC-500, and AMC-600, respectively, after transferring onto SiO2/Si substrates.



**Supplementary Figure 4.** More spectroscopic characterizations on AMC samples. (a, b) UV-Vis absorbance spectra and derived optical bandgaps of 5.32 eV, 5.38 eV, and 5.44 eV, corresponding to AMC-300, AMC-400, and AMC-500 respectively. (c) Photoluminescence spectra of AMC samples. The excitation wavelength is 532 nm.



**Supplementary Figure 5.** Additional STEM images to show the relatively large nanopores for AMC-400 (a), AMC-450 (b), and AMC-500 (c), respectively.



**Supplementary Figure 6.** Schematic diagram of AMC growth process under varying temperatures. When the precursor is exposed to different temperatures, distinct crack levels can result in different distribution of fragments. For example, at higher temperature the molecules experience more thorough thermal cracking. This feature, combined with the temperature-tuned catalytic property of Cu surface and species mobility, together determine the different level of nanopores, in terms of the size and density, in the as-grown AMC films.



**Supplementary Figure 7.** More transmembrane ionic transport measurements of AMC samples obtained between 400 °C and 500 °C. (a) Optical image of AMC samples on Cu foils (left column), after transferring onto glass pores (middle column), and on the *Trans*/*Cis* chips. (b) The transmembrane current versus bias curves. (c) The transmembrane conductance of AMC samples calculated from *I*-*V* curves in (b).



**Supplementary Figure 8.** Schematic diagram of carbon nanomaterials: AMC (a), graphene (b), carbon nanotube (c), and fullerene (d).

<b>Materials</b>	<b>Nanopore Fabrication</b>	<b>Pore Size</b>	<b>Testing Area</b>	Conductance
Graphene $1-4$	Electron irradiation	$5-23$ nm	$0.04 \mu m^2$	$50-240$ nS
	Electrochemical reaction		$0.8 \mu m^2$	$0.4 - 16$ nS
	$O2$ plasma treatment		$0.2 \text{ mm}^2$	$0.14 - 11$ nS
	$Ga^{\dagger}/He^{\dagger}$ ion bombardment or $H_2/N_2$ plasma treatment	$0.4 - 3 \text{ nm}$	$20 \mu m^2$	$2-8$ nS (0.1 M KCl)
$MoS2$ <sup>5-7</sup>	Electron irradiation or electrochemical reaction	$2-25$ nm	$3-500$ nm <sup>2</sup>	$20-300$ nS
	$Ga+$ ion bombardment	$0.4 - 1.4$ nm	$0.03 \mu m^2$	$0.02 - 3$ nS
	<b>PAN</b> etchant reaction	$14-26$ nm	$0.008 \mu m^2$	$0.9 - 130$ nS
$WS_2$ <sup>8</sup>	Electron irradiation	$2-8$ nm	$0.07 \mu m^2$	$27-108$ nS
AMC (This work)	None	$0-4$ nm	$0.8 \mu m^2$	$0.6 - 23$ nS

**Supplementary Table 1.** Comparison of AMC with other 2D materials reported as ionic transmembrane.

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