

Supplementary Information

Manganese Tetraphenylporphyrin and Carbon Nanocoils Interface based Electrochemical Sensing of Tyrosine

Syeda Aqsa Batool Bukhari¹, Abeera Aziz¹, Habib Nasir^{1*}, Sharif Ullah¹, Tehmina Akhtar¹, Sadia Iram^{1,2}, Effat Sitara^{1,3}, Shehla Mushtaq^{1,4} and Syed Abdul Moiz⁵

¹Department of Chemistry, School of Natural Sciences, National University of Sciences and Technology H-12 Islamabad (44000) Pakistan

²Department of Chemistry, Rawalpindi women university Rawalpindi Pakistan

³Department of Chemistry, Karakoram International University, Gilgit, Pakistan.

⁴Department of Chemistry, University of Management and Technology, Sialkot, Pakistan

⁵Department of Electrical Engineering, Umm Al-Qura University, Saudi Arabia

*Corresponding Author's Email: habibnasir@sns.nust.edu.pk

Figures

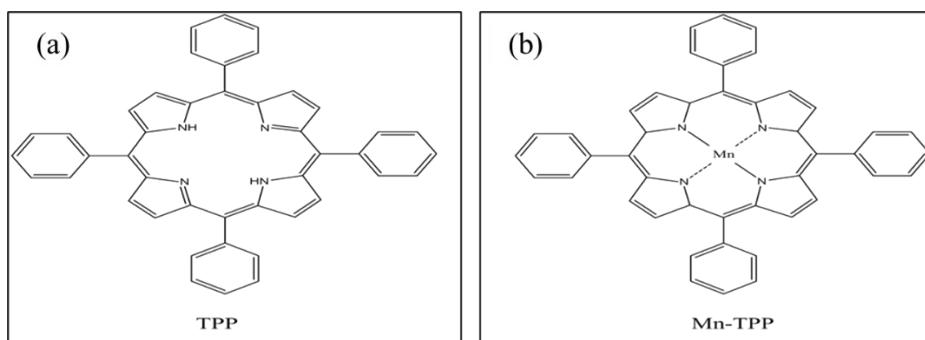


Figure S1. Chemical structure of (a) tetraphenylporphyrin and (b) Manganese tetraphenylporphyrin.

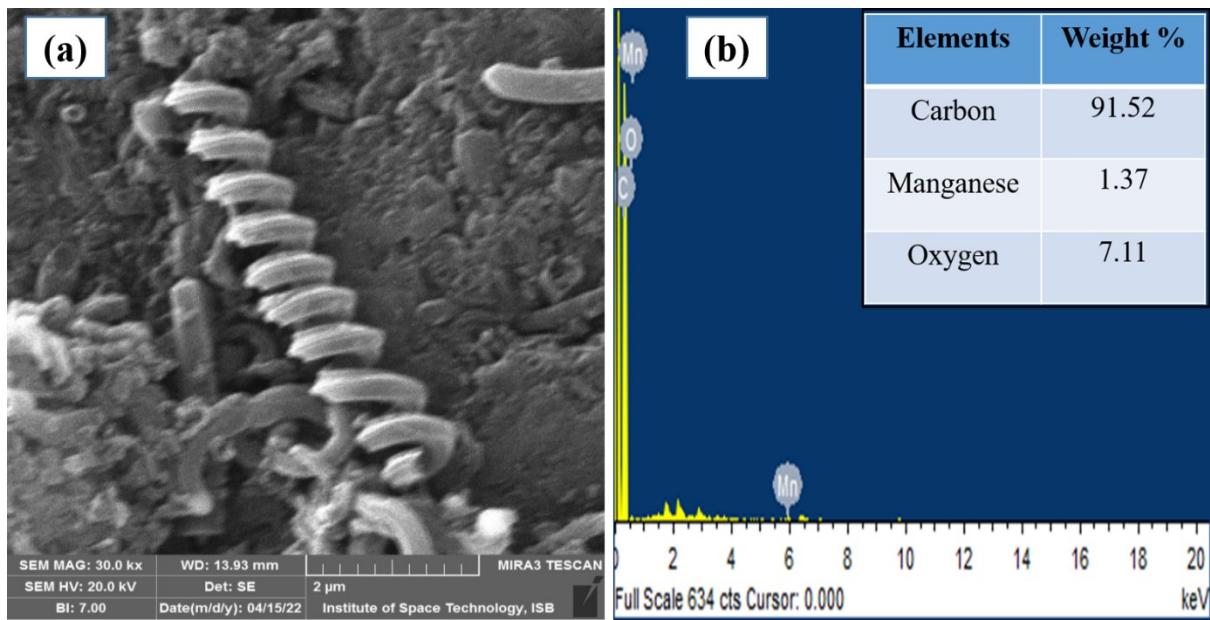


Figure S2. Analysis of CNC/MnTPP nanocomposite (a) SEM and (b) EDS.

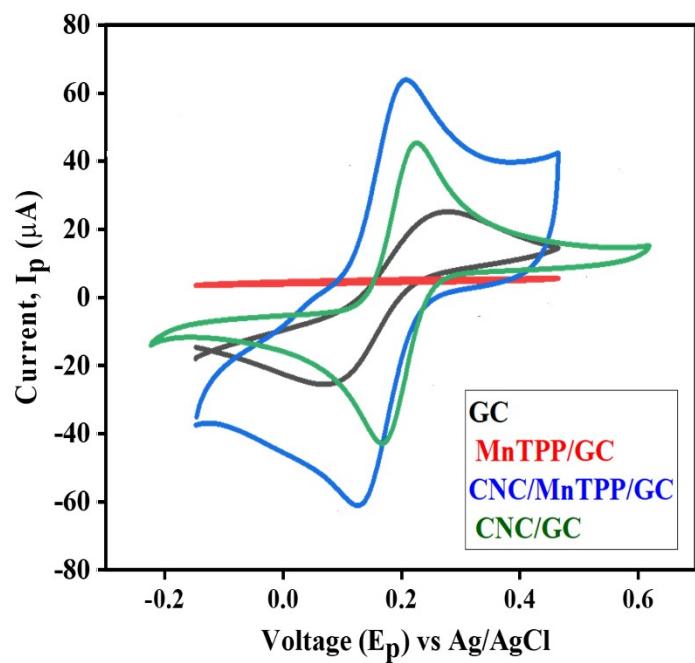


Figure S3. Cyclic voltammograms obtained in 5 mM potassium ferricyanide solution (0.1 M KCl).

Schemes

Calculation:

Slope of $\frac{E_{pa}}{\log v} = 0.06$ (From calibration curve of Figure 9 d)

$$\text{Slope of } \frac{E_{pa}}{\log v} = \frac{2.303 R \times T}{n\alpha F}$$

As we know that, Eq. 1

By substituting the values of R, T, F which are the parameters with usual meanings such as gas constant, absolute temperature, Faraday's constant, respectively and Slope in Eq.1, the value obtained for **nα equals to 1.002**.

While the value of α can be obtained from another equation (Eq. 2)

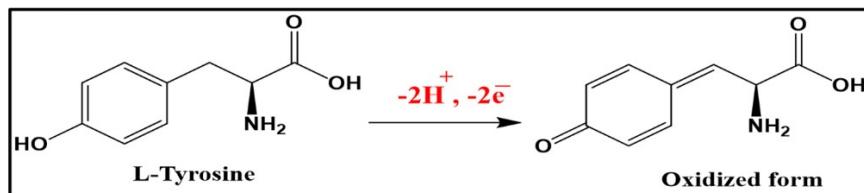
$$\alpha = \frac{47.7}{E_p - E_{p_{1/2}}} \quad Eq. 2$$

Whereas $E_p = E_{pa} = 800$ mV ; $E_{p_{1/2}} = 715$ mV, so by placing these values in Eq. 2 the value of **α equals to 0.56**

After solving Eq. 3

$$n\alpha = 1.002 \quad Eq. 3$$

$$n = \frac{1.002}{0.56} = 1.78 \approx 2$$



Scheme S1. Probable mechanism of tyrosine detection.

Tables

Table S1. Selectivity study data for CNC/MnTPP/GC towards tyrosine at pH 5.

Interfering agents	Current (μA)	Tolerance (%)
Tyrosine	9.67	--
Tyrosine + Dopamine	10.06	-4.03 %
Tyrosine + Ascorbic acid	9.68	-0.10 %
Tyrosine + Uric acid	8.91	+7.85 %

Tyrosine + Glucose	9.19	+4.96 %
Tyrosine + Glycine	9.63	+0.41 %

Table S2. Comparison of the Performance of CNC/MnTPP/GC Sensor with Existing Literature.

Sr. #	Electrode	Linear range	LoD	Method	Reference
1	CB-GO/CP ^a	0.1–2 mM	100 μM	--	¹
2	GR/ZnO/GSPE ^b	1.0×10^{-6} to 8.0×10^{-4} M	3.4×10^{-7} M	SQV	²
3	CoTPyPRu(bipy) ₂ -Ba ^c	1 to 24 μM	0.5 μM	SQV	³
4	Hemin/PAMAM/MWCNT ^d	0.1 μM to 28.8 μM	0.01 μM	Chronoamperometry	⁴
5	TCPP-Sa/CPE ^e	3 to 51 μM	0.7 μM	SQV	⁵
6	CNC/MnTPP/GC	0.05 to 100 μM	0.021 μM	DPV	Present work

^a CB-GO/CP: carbon black/graphene oxide/hydrophilic carbon paper

^b GR/ZnO/GSPE: Graphene oxide/ZnO nanorods/graphite screen-printed electrodes.

^c CoTPyPRu(bipy)₂-Ba:μ-*meso*-tetra(4-pyridyl)porphyrinatecobalt(II)tetrakis[bis(bipyridine)]chlorido ruthenium(II)- Cameroonian smectite clay form Bagba hill.

^d Hemin/PAMAM/MWCNT: iron protoporphyrin IX/poly(amidoamine)/multi-walled carbon nanotube.

^e TCPP-Sa/CPE: Meso-tetra(4-carboxyl phenyl) porphyrin immobilized Cameroonian smectite clay from Sabga hill deposited on carbon paste electrode.

References

1. Q. Bao, G. Li, Z. Yang, P. Pan, J. Liu, R. Tian, Q. Guo, J. Wei, W. Hu and W. Cheng, *Microchemical Journal*, 2023, **185**, 108238.
2. H. Beitollahi and F. Garkani Nejad, *Electroanalysis*, 2016, **28**, 2237-2244.
3. J. C. Kemmegne-Mbouguen, H. E. Toma, K. Araki, V. R. L. Constantino, E. Ngameni and L. Angnes, *Microchimica Acta*, 2016, **183**, 3243-3253.
4. Q. Ma, S. Ai, H. Yin, Q. Chen and T. Tang, *Electrochimica Acta*, 2010, **55**, 6687-6694.
5. J. C. Kemmegne-Mbouguen and E. Ngameni, *Analytical Methods*, 2017, **9**, 4157-4166.