

Supplementary materials

High-efficiency Removal of As(III) from Water using Siderite as Iron Source in the Electrocoagulation Process

Haitao Yu^a, Junfeng Li^{ab,*}, Wenying Qu^{ab}, Wenhui Wang^{ab}, Jiankang Wang^{ab}

^a *College of Water Conservancy and Architectural Engineering, Shihezi University, Shihezi,*

Xinjiang, 832000, PR China;

^b *Key Laboratory of Cold and Arid Regions Eco-Hydraulic Engineering of Xinjiang Production &*

Construction Corps, Shihezi, 832000, Xinjiang, PR China.

Submitted to

RSC Advances

April 11, 2024

**Corresponding author*

E-mail: ljfshz@126.com (J.F. Li)

Table S1 Main instruments and equipment

Name of instrument	Model number
Magnetic stirrer	MS-H-S
Electric heating blast drying oven	WGL-30B
Portable multi-parameter water quality analyzer	SL1000
Direct current power supply	MS-155D
Peristaltic pump	NKCP-S10B
Electronic balance	MS204TS
Ultrasonic cleaning device	KH-256DB
Atomic absorption spectrophotometer	AA-6880
Ultraviolet visible spectrophotometer	DR6000

Table S2 Main test drugs

Name of drug	Specifications	Manufacturer of products
NaAsO ₂	Super pure	Sigma-Aldrich Corporation
CH ₄ N ₂ S	Super pure	Tianjin Yongsheng Fine Chemicals Corporation
L-Ascorbic acid	Super pure	Tianjin Yongsheng Fine Chemicals Corporation
HCl	Super pure	Tianjin Yongsheng Fine Chemicals Corporation
H ₂ SO ₄	Analytical pure	Tianjin Yongsheng Fine Chemicals Corporation
NaHB	Super pure	Tianjin Yongsheng Fine Chemicals Corporation
HNO ₃	Analytical pure	Tianjin Yongsheng Fine Chemicals Corporation
Na ₂ SO ₄	Analytical pure	Tianjin Yongsheng Fine Chemicals Corporation
NaOH	Super pure	Tianjin Yongsheng Fine Chemicals Corporation
HONH ₃ Cl	Analytical pure	Tianjin Yongsheng Fine Chemicals Corporation
CH ₃ COONH ₄	Analytical pure	Tianjin Yongsheng Fine Chemicals Corporation
1,10-Phenanthroline	Analytical pure	Tianjin Yongsheng Fine Chemicals Corporation
Fe(NH ₄) ₂ •(SO ₄) ₂ •6H ₂ O	Analytical pure	Tianjin Yongsheng Fine Chemicals Corporation
NaNO ₃	Analytical pure	Tianjin Yongsheng Fine Chemicals Corporation
Na ₃ PO ₄	Analytical pure	Tianjin Yongsheng Fine Chemicals Corporation
K ₂ Cr ₂ O ₇	Analytical pure	Tianjin Yongsheng Fine Chemicals Corporation

Table S3 Experimental design and results of arsenic removal efficiency optimization with response surface

Test Number	A- Intensity of current (mA)	B-pH	C- Rate of circulation (mL/min)	Actual arsenic removal rate (%)
1	-1	-1	0	73.43
2	1	-1	0	78.03
3	-1	1	0	63.55
4	1	1	0	73.02
5	-1	0	-1	61.27
6	1	0	-1	75.33
7	-1	0	1	80.88
8	1	0	1	70.59
9	0	-1	-1	80.34
10	0	1	-1	58.52
11	0	-1	1	71.71
12	0	1	1	73.23
13	0	0	0	98.62
14	0	0	0	99.84
15	0	0	0	99.85
16	0	0	0	98.93
17	0	0	0	96.49

Table S4 Regression analysis results of the actual arsenic removal rate model and regression coefficient

Source of information	Sum of squared deviations	Degree of freedom	Mean square	F	<i>P</i>	Significance of significance
model	3132.47	9	348.05	71.38	<0.0001	**
A- Intensity of current	39.78	1	39.78	8.16	0.0245	*
B-pH	154.79	1	154.79	31.75	0.0008	**
C- Rate of circulation	54.86	1	54.86	11.25	0.0122	*
AB	5.93	1	5.93	1.22	0.3066	
AC	148.23	1	148.23	30.4	0.0009	**
BC	136.19	1	136.19	27.93	0.0011	**
A ²	693.68	1	693.68	142.26	<0.0001	**
B ²	813.87	1	813.87	166.91	<0.0001	**
C ²	812.7	1	812.7	166.67	<0.0001	**
Residual error	34.13	7	4.88			
Loss of fit term	26.58	3	8.86	4.69	0.0848	ns
Pure error	7.55	4	1.89			
Sum	3166.61	16				

R²=0.9892 Adj R²=0.9754 Pre R²=0.8620

Note: *P* < 0.01 is highly significant and is represented by “***”; *P* < 0.05 is significant and is represented by “**”; *P* > 0.05 is not significant and is represented by “ns”.

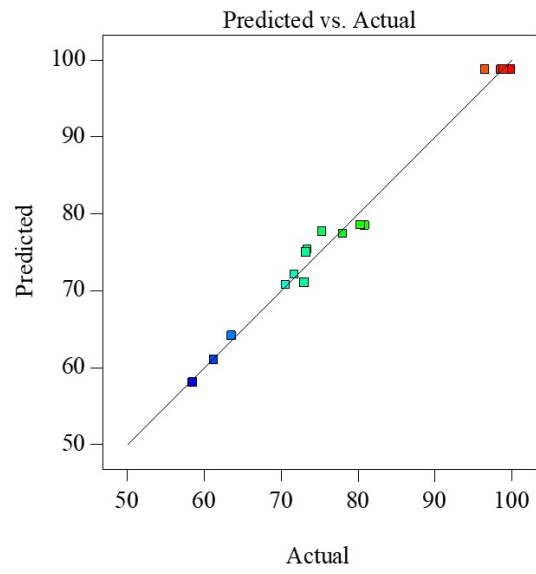


Figure S1. Fitting curve of actual and predicted value of Arsenic removal rate

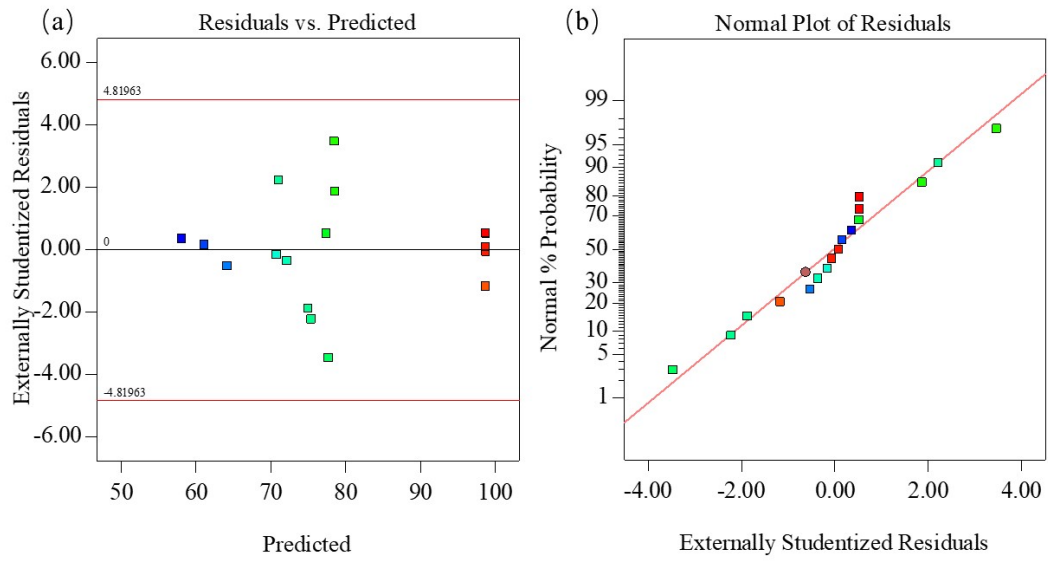


Figure S2. (a) based on the predicted As removal rate residual diagram, (b) Arsenic removal residual diagram

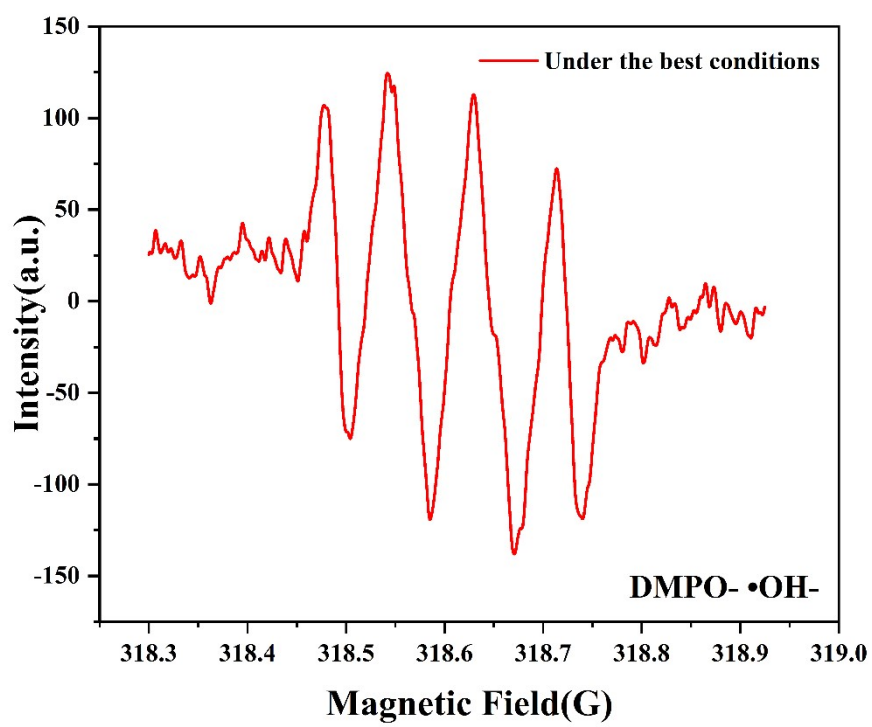


Figure S3. The ESR spectrum of the simulation system ($\bullet\text{OH}$).

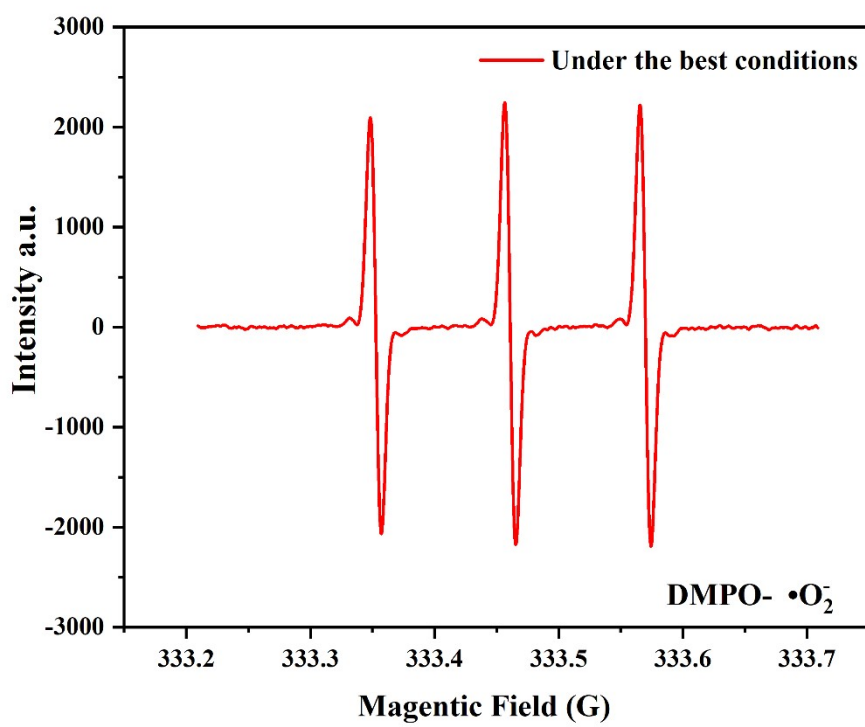


Figure S4. The ESR spectrum of the simulation system ($\cdot\text{O}_2^-$).

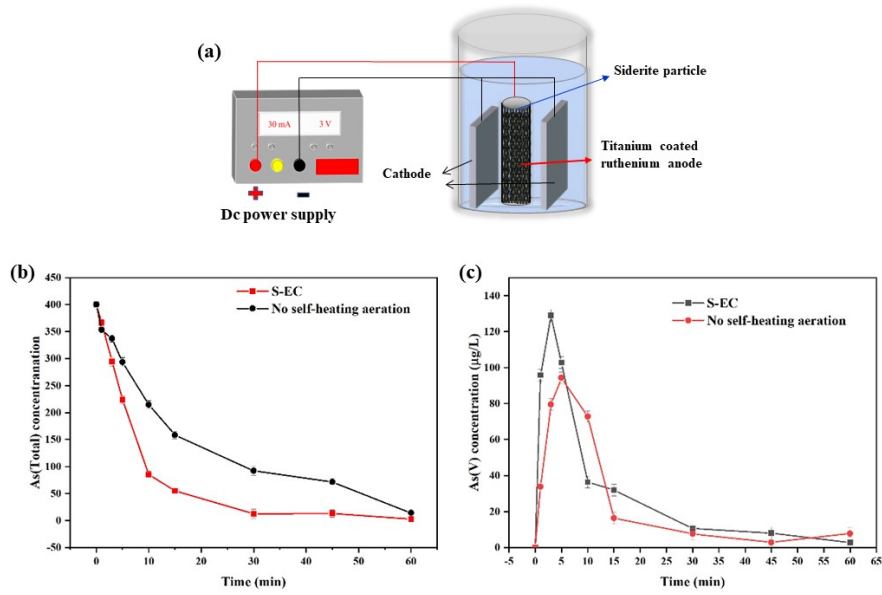


Fig. S5. (a). Schematic representation of an electrocoagulation reaction (no peristaltic pump) apparatus. Comparison of the electrocoagulation system filled with siderite anode and the electrocoagulation system with a siderite-filled anode but no overflow conditions (no peristaltic pump): (b). Arsenic removal efficiency; (c). Change in As(V) concentration.