

## Supplementary Information

### Deciphering the Influence of Pd<sup>II</sup> and Pd<sup>IV</sup> Oxidation State on Non-Standard Chemical Bonds within Bis(*j*N-Heterocyclic Carbene) Complexes: Insights from DFT

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**Table 1S.** NBO atomic charges (Q/e) of the C<sub>j</sub>NHC carbon atom and Pd<sup>i+</sup> ion in isolated moieties[bis(jNHC) and PdXi] and in each complex[1a, 2a, 1n, 2a, 3a, 4a, 3n and 4n].

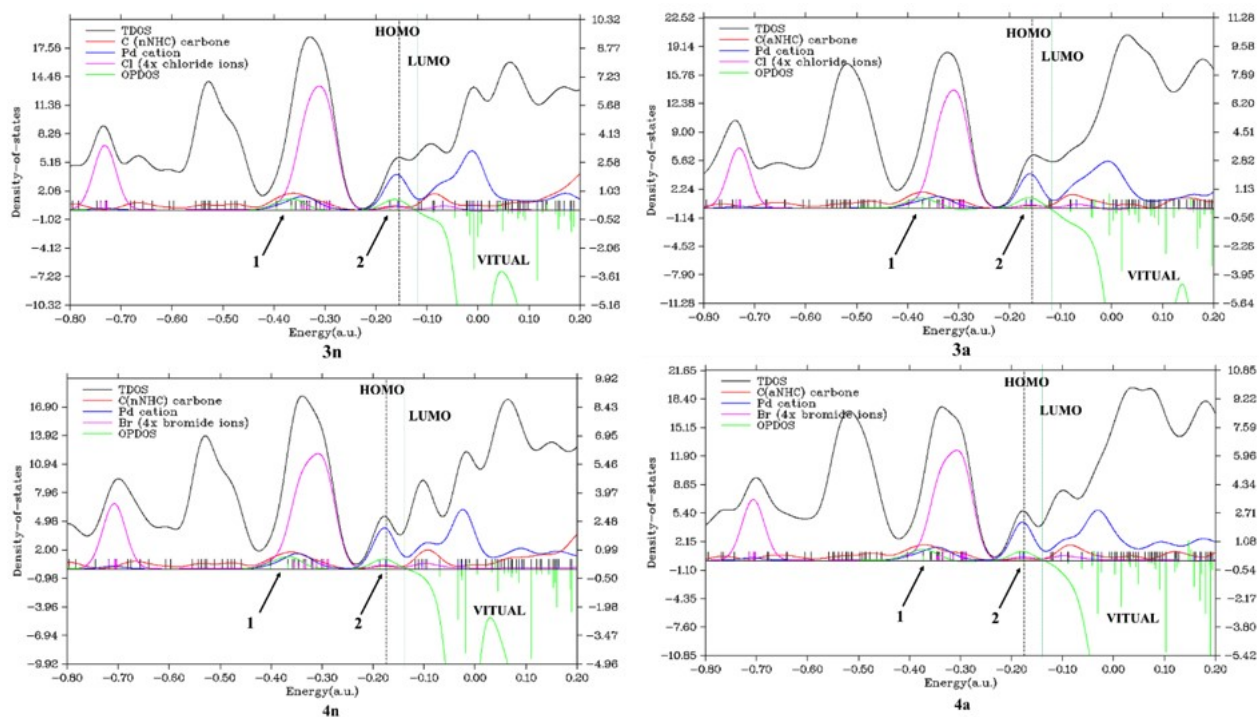
Complexes		in Complex		in PdXi		in Bis(jNHC)	
		Pd <sup>i+</sup>	C <sub>jNHC</sub>	Pd <sup>2+</sup>	Pd <sup>4+</sup>	C <sub>nNHC</sub>	C <sub>aNHC</sub>
X = Cl	<b>1n</b>	-0.01	0.32	0.53	-	0.14	-
	<b>1a</b>	0.04	0.02				
X = Br	<b>2n</b>	-0.18	0.32	0.44	-		
	<b>2a</b>	-0.16	0.02				
X = Cl	<b>3n</b>	-0.08	0.39	-	0.44	-	-0.18
	<b>3a</b>	-0.02	0.10				
X = Br	<b>4n</b>	-0.33	0.38	-	0.18		
	<b>4a</b>	-0.27	0.09				

**Table 2S.** Palladium Natural Electron Configuration in different complexes.

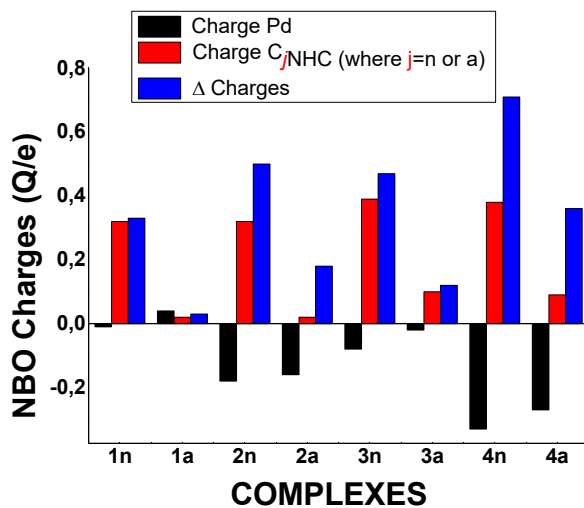
Species	Natural Electron Configuration	Isolated state
1n	[core]5S(0.41)4d(9.02)5p(0.29)5d(0.01)6p(0.21)	[core]4d(8.00)
1a	[core]5S(0.40)4d(9.01)5p(0.28)5d(0.01)6p(0.19)	
2n	[core]5S(0.42)4d(9.07)5p(0.31)5d(0.01)6p(0.23)	
2a	[core]5S(0.41)4d(9.06)5p(0.29)5d(0.01)6p(0.21)	
3n	[core]5S(0.38)4d(8.73)5p(0.32)5d(0.01)6p(0.64)	
3a	[core]5S(0.36)4d(8.71)5p(0.32)5d(0.01)6p(0.62)	
4n	[core]5S(0.42)4d(8.85)5p(0.70)5d(0.01)6p(0.36)	
4a	[core]5S(0.40)4d(8.83)5p(0.68)5d(0.01)6p(0.35)	

**Table 3S.** Natural Hybrid Orbital (NHO) Analysis of Pd-X bond in different complexes.

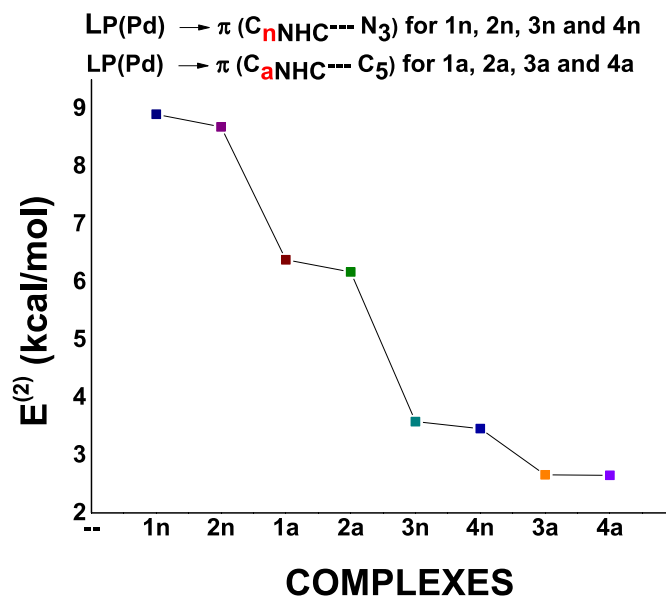
Comp	Occu.	Polarizability		Hybridation	
		Pd	X/X'X <sup>eq</sup> /X <sup>eq</sup> '	Pd	X/X <sup>eq</sup> /X <sup>eq</sup> '
1n	1.90	18.34%	81.66%	s(23.44%),p(52.83%),d( 23.73%)	s(15.35%),p(84.65%)
	1.90	18.34%	81.66%	s(23.44%),p(52.83%),d( 23.73%)	s(15.35%),p(84.65%)
	-	-	-	-	-
	-	-	-	-	-
2n	1.89	21.04%	78.96%	s(23.23%),p(53.33%),d( 23.44%)	s(14.26%),p( 85.74%)
	1.89	21.04%	78.96%	s(23.23%),p(53.33%),d( 23.44%)	s(14.26%),p( 85.74%)
	-	-	-	-	-
	-	-	-	-	-
3n	1.86	27.13%	72.87%	s(15.19%),p(52.71%),d(32.10%)	s(10.71%),p(89.29%)
	1.86	27.13%	72.87%	s(15.19%),p(52.71%),d(32.10%)	s(10.71%),p(89.29%)
	1.83	29.12%	70.88%	s(16.23%),p(49.61%),d(34.16%)	s(9.00%),p(91.00%)
	1.83	28.46%	71.54%	s(15.89%),p(50.24%),d(33.88%)	s(9.40%),p(90.60%)
4n	1.84	31.82%	68.18%	s(14.40%),p(53.15%),d(32.45%)	s(9.28%),p(90.72%)
	1.84	31.82%	68.18%	s(14.40%),p(53.15%),d(32.45%)	s(9.28%),p(90.72%)
	1.81	32.99%	67.01%	s(16.84%),p(49.79%),d(33.36%)	s(7.98%),p(92.02%)
	1.81	31.94%	68.06%	s(16.79%),p(50.19%),d(33.03%)	s(8.34%),p(91.66%)
1a	1.91	17.26%	82.74%	s(23.41%),p(53.70%),d(22.89%)	s(16.25%),p(83.75%)
	1.91	17.26%	82.74%	s(23.41%),p(53.70%),d(22.89%)	s(16.25%),p(83.75%)
	-	-	-	-	-
	-	-	-	-	-
2a	1.90	19.61%	80.39%	s(23.22%),p( 54.12%),d(22.67%)	s(15.15%),p( 84.85%)
	1.90	19.61%	80.39%	s(23.22%),p( 54.12%),d(22.67%)	s(15.15%),p( 84.85%)
	-	-	-	-	-
	-	-	-	-	-
3a	1.87	24.93%	75.07%	s(15.46%),p(54.22%),d(30.32%)	s(12.19%),p(87.81%)
	1.87	24.93%	75.07%	s(15.46%),p(54.22%),d(30.32%)	s(12.19%),p(87.81%)
	1.84	26.73%	73.27%	s(15.99%),p(50.66%),d( 33.36%)	s(10.02%),p(89.98%)
	1.82	29.18%	70.82%	s(16.09%),p(49.10%),d(34.81%)	s(9.08%),p(90.92%)
4a	1.86	29.14%	70.86%	s(14.81%),p(54.64%),d( 30.55%)	s(10.80%),p(89.20%)
	1.86	29.14%	70.86%	s(14.81%),p(54.64%),d( 30.55%)	s(10.80%),p(89.20%)
	1.82	30.38%	69.62%	s(16.86%),p(50.55%),d(32.58%)	s(8.97%),p(91.03%)
	1.80	32.76%	67.24%	s(16.91%),p(49.27%),d(33.83%)	s(8.21%),p(91.79%)



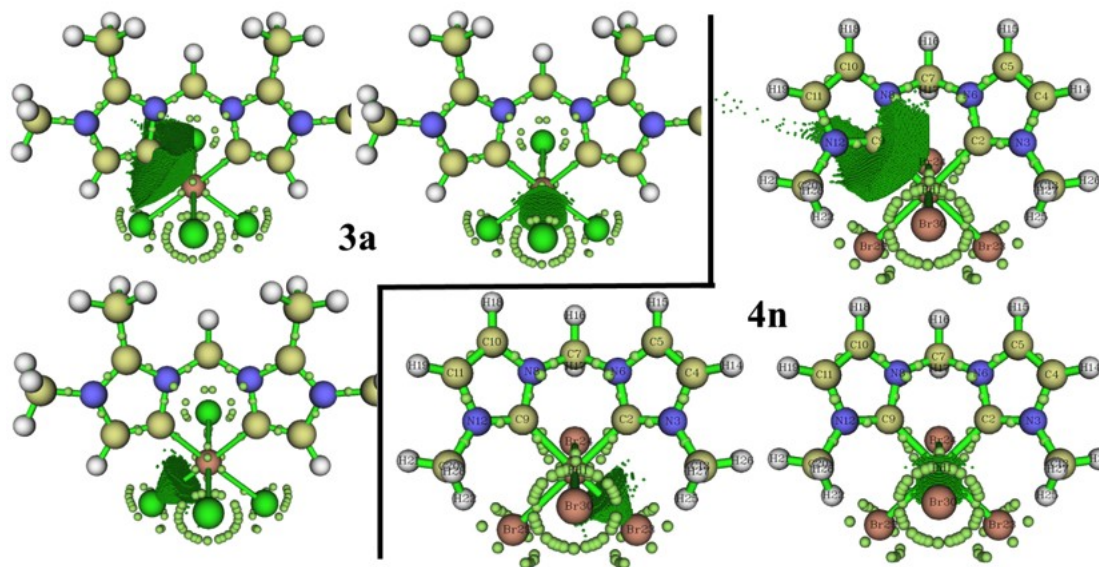
**Figure 1S.** Total (black), partial [red ( $C_{jNHC}$ ), blue ( $Pd^{4+}$ ), and magenta ( $Cl \vee Br$ ), and overlap [green, between  $C_{jNHC}$  and  $Pd^{4+}$  and  $X(X = Cl \vee Br)$  basis functions] density-of-states map of  $La[Ln]Pd^{IV}X_4$  complexes.



**Figure 2S.** Natural Population Analysis (NPA) charges on the  $C_{jNHC}$  atom (bis( $jNHC$ )) and  $Pd^{4+}$  ( $PdXi$ ) isolated moieties and their difference in 1-4a(n) complexes.



**Figure 3S.** The second-order perturbation energy stabilization ( $E^{(2)}$ ) for each donor NBO (i) and acceptor NBO(j) from the lone pair Pd to  $C_{nNHC} \cdots N$  or  $C_{aNHC} \cdots N$  bond in all complexes.



**Figure 4S.** Local region basin analysis on  $C_{jNHC}$ -Pd and Pd-X in only **3a** and **4n** bis(aNHC)[bis(nNHC)]Pd<sup>IV</sup>X<sub>4</sub> complexes.