## **Electronic Supporting Information**

Study of mixed matrix membranes with in-situ synthesized zeolite imidazolate frameworks (ZIF-8, ZIF-67) in polyethersulfone polymer for CO<sub>2</sub>/CH<sub>4</sub> separation

Aditya Jonnalagedda, Bhanu Vardhan Reddy Kuncharam\*

Department of Chemical Engineering,

Birla Institute of Technology & Science, Pilani, Pilani Campus

Rajasthan, 333031, India

\*Corresponding author:

Dr. Bhanu Vardhan Reddy Kuncharam,

Associate Professor, Department of Chemical Engineering,

Birla Institute of Technology and Science, Pilani, Pilani, Rajasthan, India

Phone: +91-1596255839

Email: bhanu.vardhan@pilani.bits-pilani.ac.in

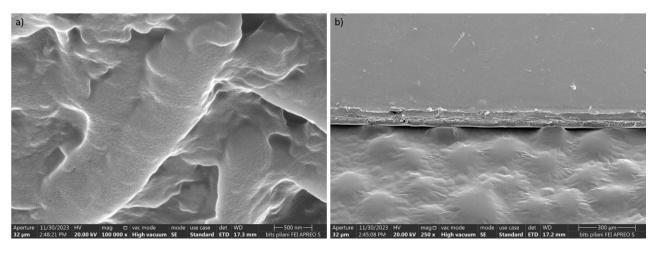
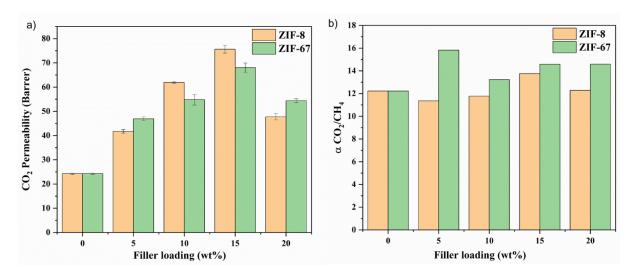
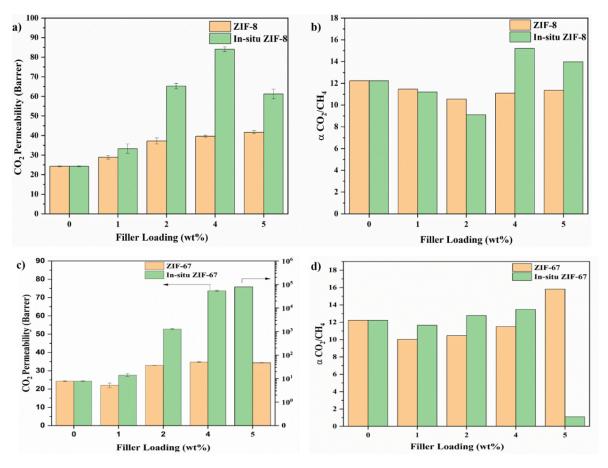


Fig S1. (a, b) Cross-section FESEM images of pure polyether sulfone (PES) membrane



**Fig. S2** Comparison plot of a) CO<sub>2</sub> permeability and b) CO<sub>2</sub>/CH<sub>4</sub> Selectivity for ZIF-8/PES and ZIF-67/PES MMMs at 0.5 bar



**Fig. S3** Comparison plot of (a, c) CO<sub>2</sub> permeability and (b, d) CO<sub>2</sub>/CH<sub>4</sub> Selectivity for in-situ and non-in-situ MMMs at 0.5 bar

## Effect of unreacted species (metal ions and linkers)

We have used gas permeation data of in-situ membranes to study the effect of unreacted metal ions (Zn<sup>+2</sup>/Co<sup>+2</sup>) and imidazole linkers. In Fig.S3, we can see the CO<sub>2</sub> permeability of in-situ ZIF-8 membranes is increased until 4 wt.% ZIF-8 and decreased at 5 wt.% ZIF-8, this decrease is possibly associated with unreacted Zn<sup>+2</sup> ions hindering the CO<sub>2</sub> passage through the membrane, the unreacted Zn<sup>+2</sup> ions in PES polymer functioned as a gas barrier without causing any significant structural defects <sup>2</sup> and also some literature studies show significant decrease in CO<sub>2</sub> permeability at high Zn<sup>+2</sup> ions concentration. The presence of unreacted 2-methyl imidazole linkers should enhance CO<sub>2</sub>/CH<sub>4</sub> selectivity exponentially, but when we observe the in-situ MMMs selectivity it has decreased for 2 wt.% in-situ ZIF-8, 5 wt.% ZIF-8 and 5 wt.% in-situ MMMs despite the increase of CO<sub>2</sub> permeability in some cases. The CO<sub>2</sub> permeability of in-situ ZIF-67 MMMs has not dropped, this might indicate the low presence of Co<sup>+2</sup> ions in PES polymer chains (possibility of higher conversion of ZIF-67). A detailed

experimental investigations need to be conducted in future studies to get the full idea of this phenomenon.

## References

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