

Supporting information

Cobalt-Zinc Carbides Embedded in N-doped Porous Carbon Nanospheres as Polysulfides Mediator for Efficient Lithium-Sulfur Batteries

Xiangyu Tan ^{1,‡}, Jiajun Du ^{3,‡}, Haitao Wu ³, Xuyu Ma ², Chunbo Zhou ⁴, Fangrong
Zhou ¹, Caijin Fan ⁵, Xueqin Xu ⁴, Shaocong Hou ^{2,*} Xin Cai ^{3,*}

¹ *Power Science Research Institute of Yunnan Power Grid Co., Ltd, Kunming 650214, China*

² *School of Electrical Engineering and Automation, Wuhan University, Wuhan 430072, China*

³ *College of Materials and Energy, South China Agricultural University, Guangzhou 510642, China*

⁴ *Qujing Power Supply Bureau of Yunnan Power Grid Co., Ltd, Qujing 655099, China*

⁵ *State Key Laboratory of HVDC, Electric Power Research Institute, CSG, Guangzhou, 510080, China*

[‡]*Xiangyu Tan and Jiajun Du contributed equally to this work.*

**Corresponding author E-mail address:*

sc.hou@whu.edu.cn (Shaocong Hou)

caixin2015@scau.edu.cn (Xin Cai)

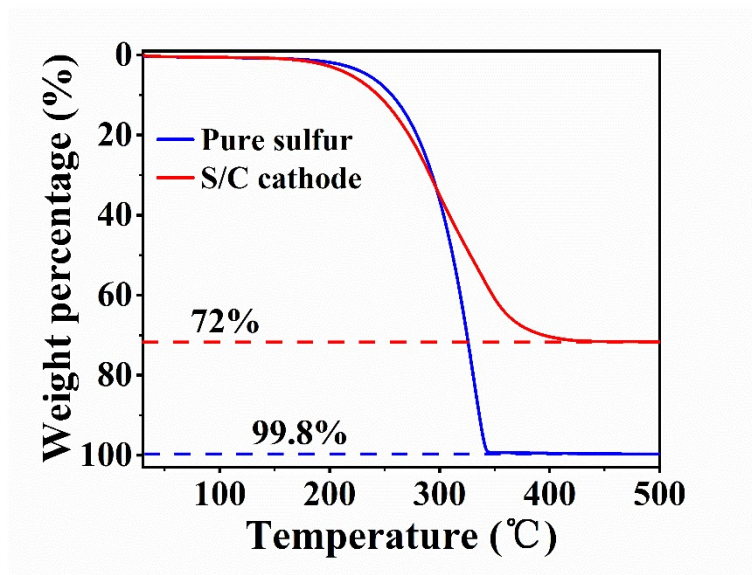


Figure S1 Thermogravimetric curve of the sulfur/carbon (S/C) composite cathode.

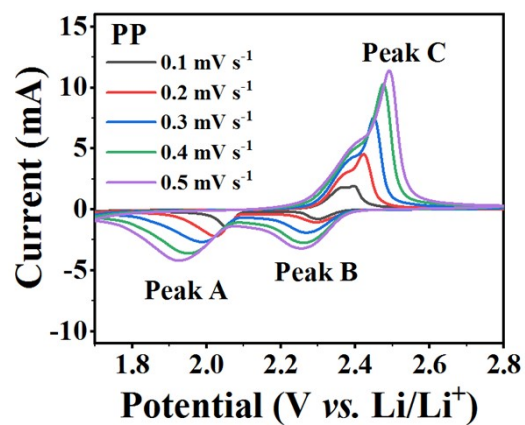


Figure S2 CV curves of the PP battery at different scanning rates.

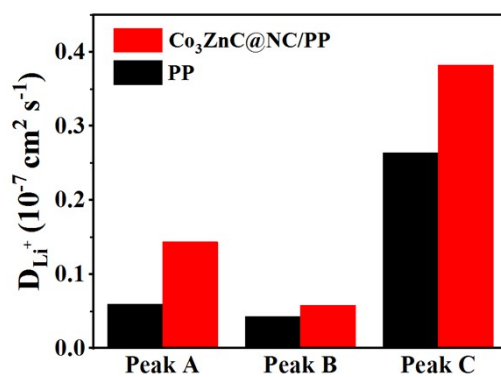


Figure S3 Comparison of the calculated lithium ion diffusion coefficient (D_{Li^+}) by linear fitting of the three peak currents of peaks A, B and C for the $\text{Co}_3\text{ZnC@NC/PP}$ battery and PP battery, respectively.

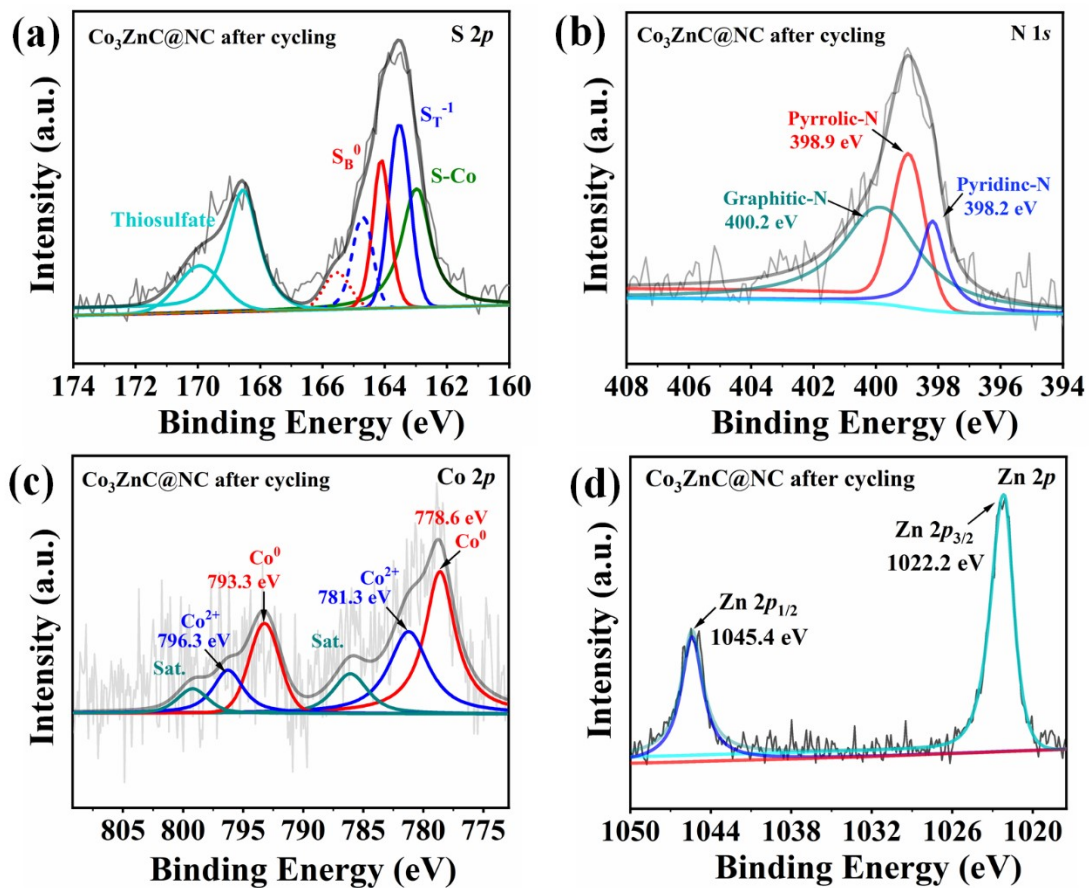


Figure S4 XPS spectra of Co₃ZnC@NC after battery cycling for 50 charge/discharge cycles at 0.5 C. (a) S 2p; (b) N 1s; (c) Co 2p; (d) Zn 2p.

Table S1 Comparison of the electrochemical performances of Co₃ZnC@NC interlayer-based Li-S batteries with recently reported composite analogues.

Catalyst material	Sulfur loading (mg cm ⁻²)	Initial capacity (mAh g ⁻¹)@ Current density	Cyclic capacity (mAh g ⁻¹) @current/cycles	Rate capability (mAh g ⁻¹)	Ref.
Co ₃ ZnC@NC	1.2~1.3	1659.8@0.1 C 1047.3@1 C	695.8@1 C/250	734 @5 C	This work
	6.2 (E/S: 6 μL mg ⁻¹)	1249.2@0.2 C	805.4@0.2 C/50 (~5 mA h cm ⁻²)	/	
		1008.7@0.5 C	542.8@0.5 C/50	/	
VN@NC	1.2~1.3	1594.8@0.1 C	712.1@1 C/500	766.5 @5 C	<i>J. Power Sources</i> 2023 , 566, 2329223
	4.5 (E/S:8 μL mg ⁻¹)	934@0.2 C	811@0.2 C/70	/	
CoFe ₂ O ₄ /N-doped carbon nanofibers	6.3	940@0.2 C	690@0.2 C/450	/	<i>Chem. Eng. J.</i> 2024 , 481, 148374
ZnS@CoS ₂ nanocages	/	1443@0.1 C	507@2 C/650	688@3 C	<i>J. Energy Storage</i> 2024 , 75, 109505
VN QDs/NC	3.2 (E/S:8.8 μL mg ⁻¹)	1214@0.2 C	1085@0.2 C/200 726@2 C/1000	752@4 C	<i>Chem. Eng. J.</i> 2023 , 459, 141526
ZnCo-LDH/rGO	1.2	1365.4@0.2 C	837.7 @1 C/500	796.5@5 C	<i>J. Alloy. Compd.</i> , 2023 , 968, 172059
FeNi@NC	1.43	1378.8@0.1 C	455@1 C/500	874@2 C	<i>Chem. Eng. J.</i> 2023 , 474, 145751
	4.99	6.25 mA h cm ⁻² @0.1 C	4.28 mA h cm ⁻² @0.1 C/100	/	
Ni-CoSe ₂ @NC	2.4	930.3@0.2 C	399.2@0.5 C/400	242.8@3 C	<i>RSC Adv.</i> , 2024 , 14, 15358–15364
Zn-Co SA@DNC	1.5	1344@0.2 C	732@1 C/800 cycles	596@6 C	<i>J. Mater. Chem. A</i> , 2023 , 11, 12025–12033
	5.4 (E/S:9 μL mg ⁻¹)	/	4.3 mA h cm ⁻² @0.1 C/100	/	

