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Supporting Information

Sulfur/Carbon Cathode Material Chemistry and Morphology Optimisation for Lithium-Sulfur Batteries

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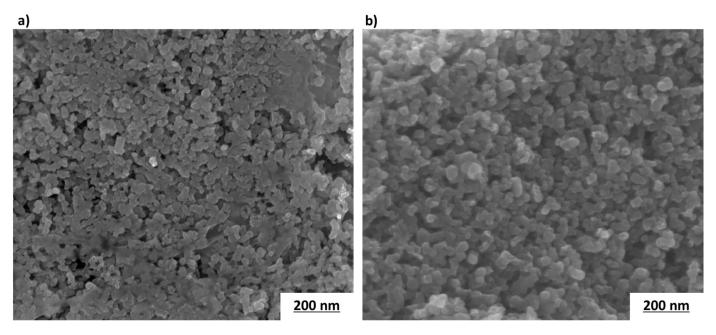


Figure S1. : a) TC10.5 cathode surface SEM image at 200nm; b) TC10.5 S/C composite powder surface SEM at 200nm

	Mass	Atomic	Mass	Atomic
Sample	concentration % (± 40 – 60%)	concentration %(\pm 40 – 60%)	concentration %(± 53 – 60%)	concentration %(± 53 – 60%)

	Sulfur		Carbon	
SC10	35.4	17.0	64.6	83.0
SC30	26.9	12.1	73.1	87.9
SC50	23.1	10.1	76.9	89.9
SC70	34.4	16.4	65.6	83.6
SC90	35.7	17.2	64.3	82.8

 Table ST1: EDX atomic and mass % across different cathode samples

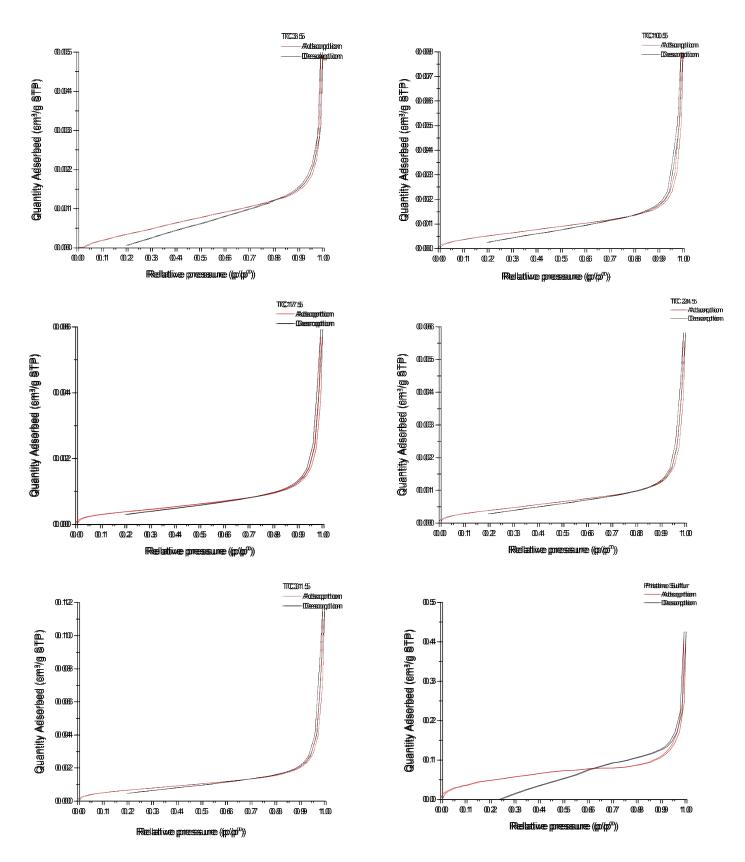


Figure S2: Adsorption and desorption isotherms for all TC sample and pristine sulfur

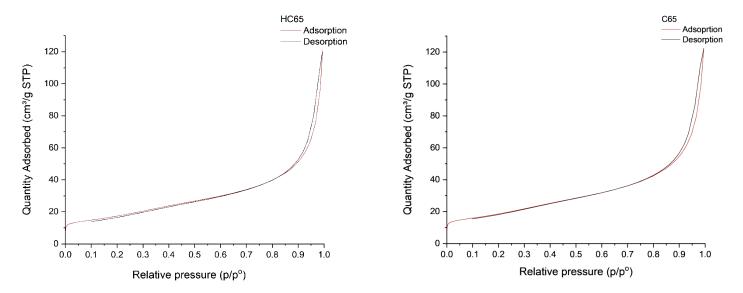


Figure S3: Adsorption and desorption isotherm for heated C65 and pristine C65

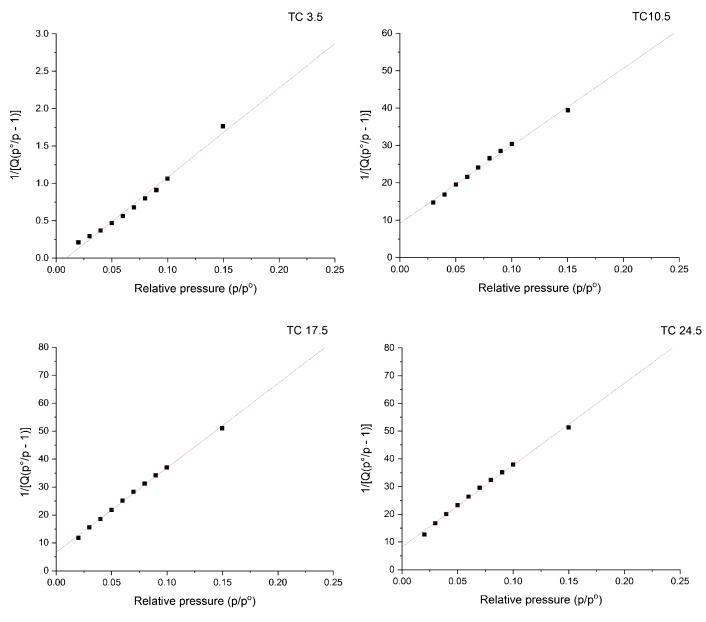


Figure S4: BET surface area plots for TC3.5, 10.5, 17.5 and 24.5

BET surface area plot

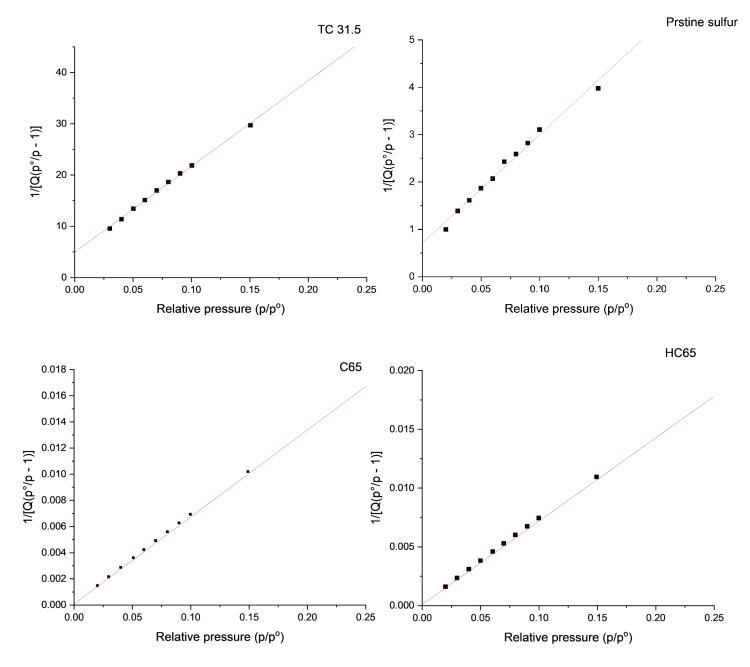


Figure S5: BET surface area plots for TC31.5, pristine sulfur, pristine C65 and HC65

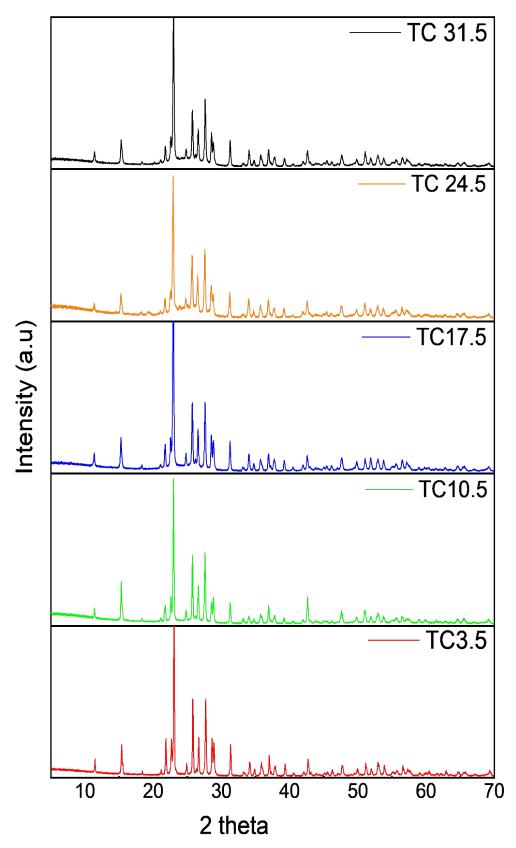


Figure S6: XRD pattern of TC cathode films

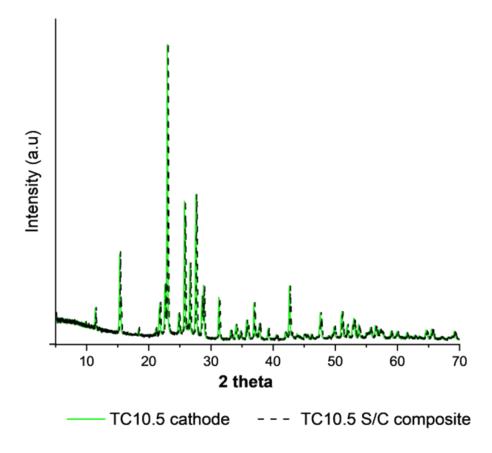


Figure S7: XRD pattern of TC10.5 cathode film with TC10.5 S/C composite

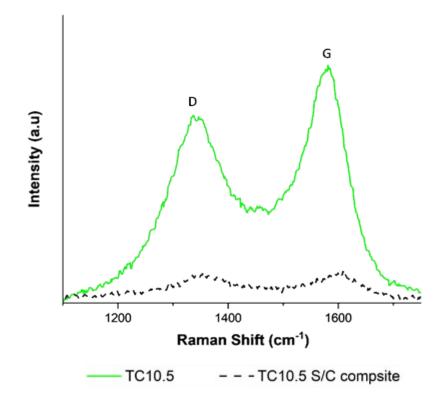


Figure S8: Raman D and G peaks pattern of TC10.5 cathode film with TC10.5 S/C composite

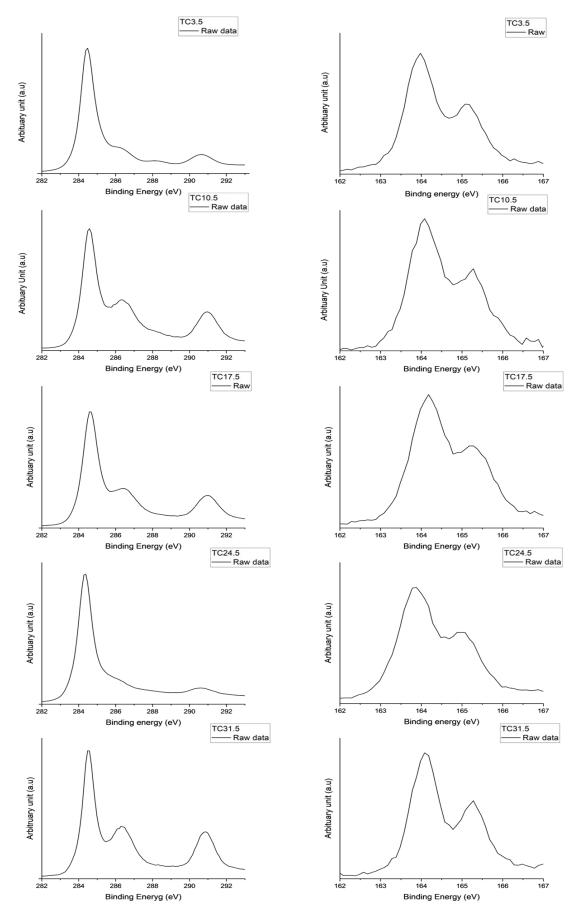


Figure S9a: C1s and S2p raw spectra for TC3.5, 10.5, 17.5, 24.5 and 31.5

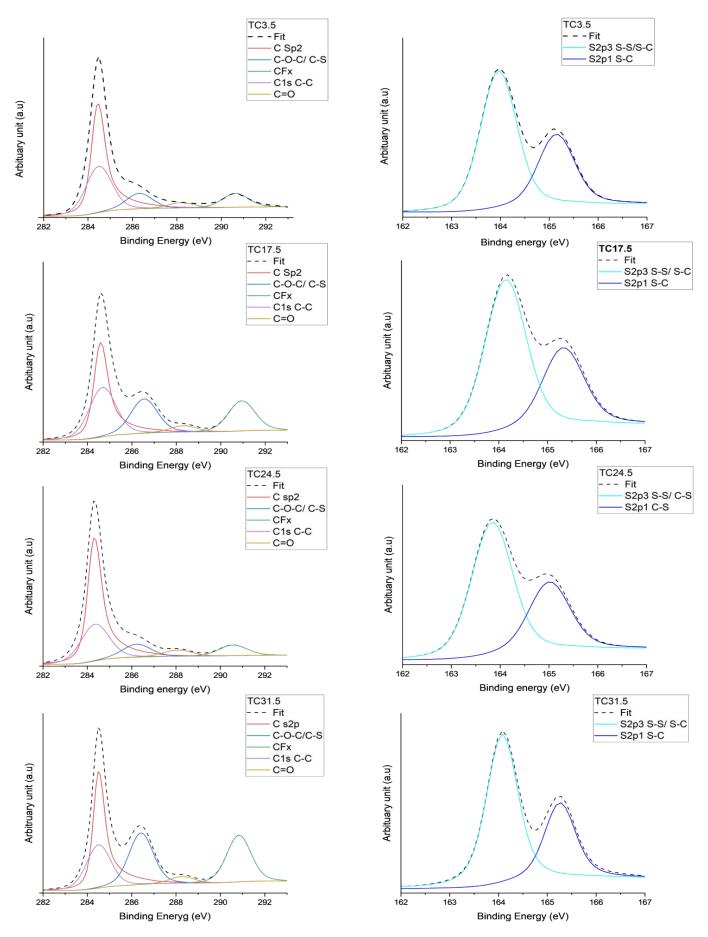


Figure S9b: C1s and S2p fitted spectra for TC3.5, 10.5, 17.5, 24.5 and 31.5

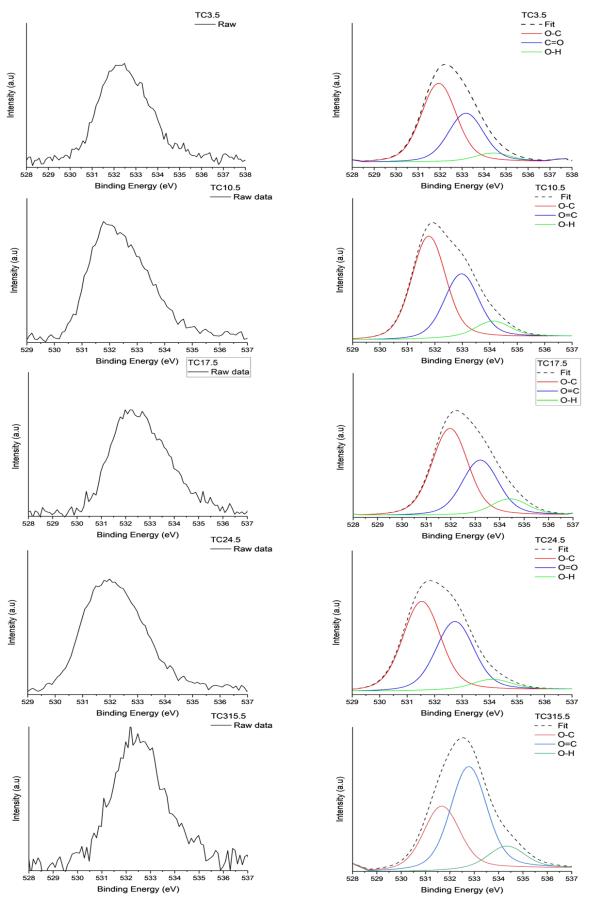


Figure S10: O1s raw and fitted spectra for all TC samples

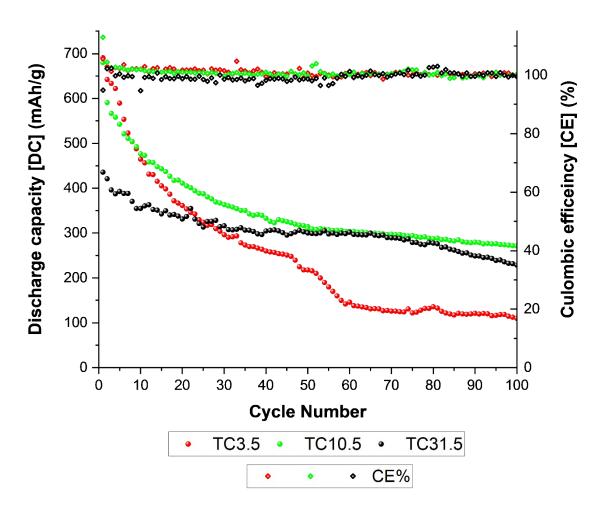


Figure S11: Discharge capacity and coulombic efficiency after 100 cycles

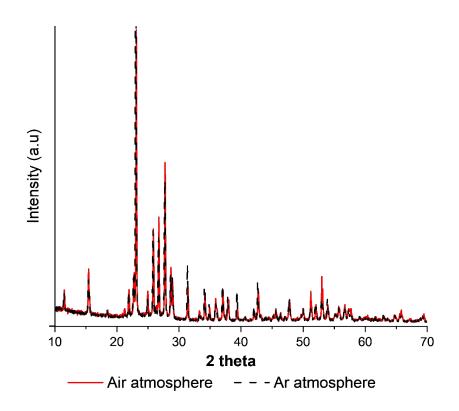


Figure S12: XRD pattern of TC composite milled in Ar and air atmosphere

Sample	$\sigma = (1/R_{eff})^*(d/A) (S/m)$	
TC3.5	1.77E-03	
TC10.5	2.15E-03	
TC17.5	3.00E-03	
TC24.5	3.08E-03	
TC31.5	4.34E-03	
HC65	5.10E-01	
Pristine C65	1.08E-01	

 Table ST2: Electronic conductivities of All TC cathode samples. C65 carbon black and heated C65 carbon black