

## Supplementary data

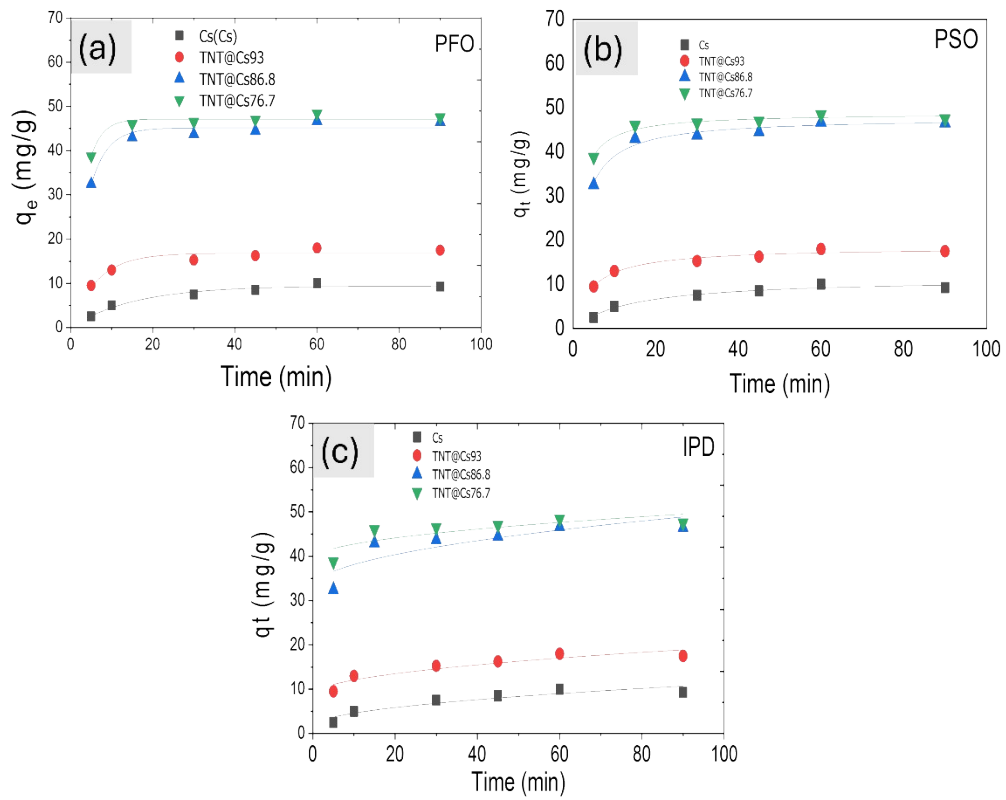
### **Enhanced adsorption of carbon sphere by doping with titania nanotubes for crystal violet removal: Isotherm, kinetics, and thermodynamic studies**

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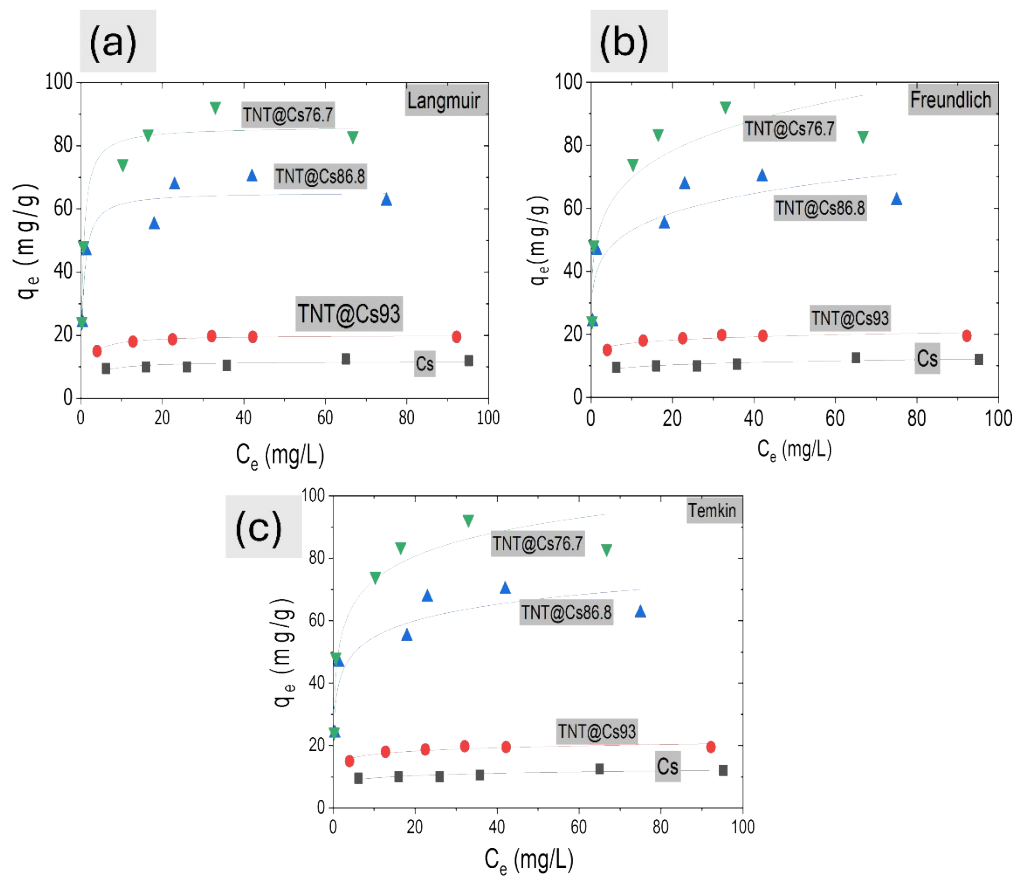
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**Figure S1:** The non-linear kinetics fitting curves for the (a) PFO, (b) PSO, and (c) IPD models.

**Table S1:** The parameters estimated from the non-linear adsorption kinetic models for the adsorption of CV dye.

Adsorbent	Pseudo first order			Pseudo second order			Intraparticle diffusion		
	$q_{e,cal}$	$K_1$	$R^2$	$q_{e,cal}$	$k_2$	$R^2$	$C$	$k_p$	$R^2$
	(mg/g)	(1/min)		(mg/g)	(mgg <sup>-1</sup> /min)		(mg/g)	(mgg <sup>-1</sup> /min)	
Cs	9.42	0.064	0.957	11.38	0.006	0.958	1.52	0.965	0.834
TNTs@Cs <sub>93</sub>	16.82	0.156	0.892	18.42	0.012	0.959	8.65	1.08	0.833
TNTs@Cs <sub>86.8</sub>	45.15	0.250	0.929	47.73	0.009	0.962	32.78	1.70	0.648
TNTs@Cs <sub>76.7</sub>	47.15	0.343	0.941	48.74	0.017	0.957	39.36	1.068	0.557



**Figure S2:** The non-linear isotherm curves for the (a) Langmuir, (b) Freundlich, and (c) Temkin models.

**Table S2:** The parameters estimated from the non-linear adsorption isotherm models for the adsorption of CV dye.

Adsorbent	Freundlich model			Langmuir model			Temkin model		
	$K_f$ (mg/g)	$n$	$R^2$	$K_L$ (L/mg)	$q_m$ (mg/g)	$R^2$	$b_t$	$A_T$ (L/g)	$R^2$
Cs	7.494	9.434	0.757	0.513	11.764	0.449	1.092	652.89	0.732
TNTs@Cs <sub>93</sub>	14.27	12.43	0.750	0.737	20.063	0.971	1.518	8130.14	0.790
TNTs@Cs <sub>86.8</sub>	38.08	6.96	0.746	1.612	65.345	0.892	7.565	139.140	0.805
TNTs@Cs <sub>76.7</sub>	47.32	5.94	0.820	1.773	86.036	0.9513	11.148	69.568	0.897