

Highly efficient CuNi-ZrO₂ nanaocomposite for selective hydrogenation of levulinic acid to γ -valerolactone

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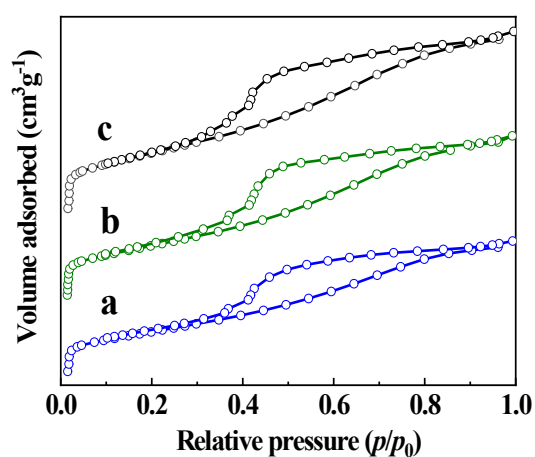


Figure S1. Adsorption-desorption isotherms of Cu_{0.5}-ZrO₂ (a), Ni_{0.5}-ZrO₂ (b) and Cu_{0.05}Ni_{0.45}-ZrO₂ (c) catalysts.

Table S1. Textural properties of the Cu_{0.5}-ZrO₂, Ni_{0.5}-ZrO₂ and Cu_{0.05}Ni_{0.45}-ZrO₂ catalysts.

Catalyst	Specific surface area (m ² g ⁻¹)	Pore volume (cm ³ g ⁻¹)	Average pore size (nm)
Cu _{0.5} /ZrO ₂	85	0.13	3.6
Ni _{0.5} /ZrO ₂	94	0.16	3.5
Cu _{0.05} Ni _{0.45} /ZrO ₂	101	0.18	3.7

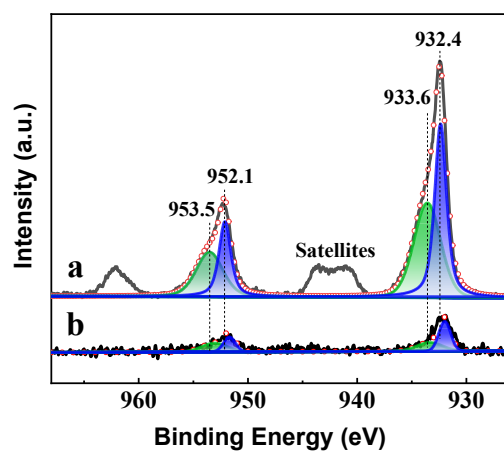


Figure S2. XPS spectra of Cu_{2p} for the Cu_{0.5}/ZrO₂ (a) and Cu_{0.05}-Ni_{0.45}/ZrO₂ (b) catalysts.

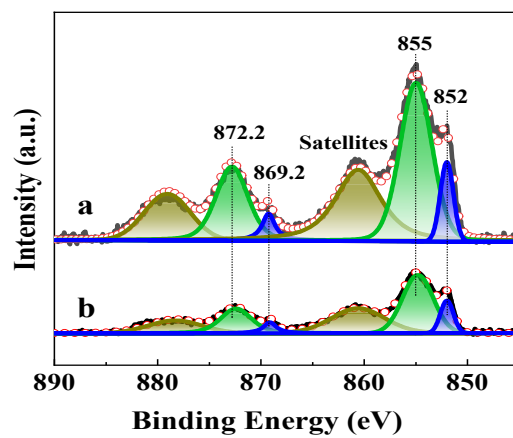


Figure S3. XPS spectra of Ni_{2p} for the Cu_{0.5}/ZrO₂ (a) and Cu_{0.05}-Ni_{0.45}/ZrO₂ (b) catalysts.

Table S2. Catalytic performance for the hydrogenation of LA to GVL on different catalysts.

Entry	Catalyst	T/°C	P _{H₂} /Mpa	t/h	LA Conv.	GVL Sel.	Ref.
1	Ni/SiO ₂	250	1	1	99.9	87	1
2	Ni/TiO ₂	270	1	3	99.9	99.1	2
3	Cu/ZrO ₂	200	3.5	5	100	100	3
4	Cu/Al ₂ O ₃	200	3.5	5	100	100	3
5	Cu-Ni/SiO ₂	250	1	8	99.1	99.2	4
6	Ni-Cu/Al ₂ O ₃	220	3	6	100	>99	5
7	CuAgZrO ₂	200	3	4	100	>99	6
8	Ni-Cu/ZrO ₂	200	3	1.5	>99.9	>99.9	This work

References

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