

**Synergistic impact of nano-supramolecular coordination polymer based on cadmium, ethyl nicotinate and thiocyanate ligands as efficient catalyst to remove harmful elements from wastewater.**

By

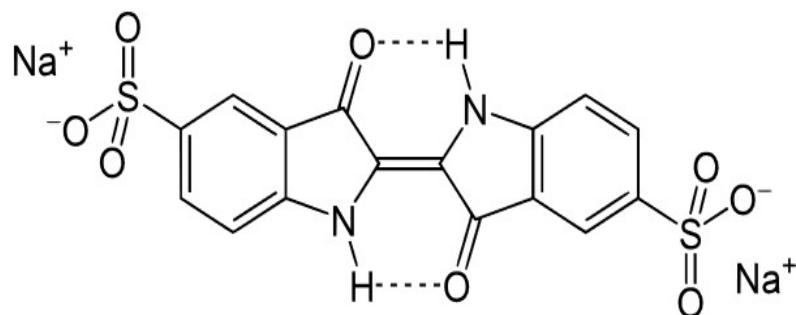
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**Supplementary Materials**



**Scheme S1** Chemical structure of Indigo carmine dye (IC).

**Table S1** The SCP 1's crystal data and structural refinement parameters.

1	
Empirical Formula	C <sub>18</sub> H <sub>18</sub> N <sub>4</sub> O <sub>4</sub> S <sub>2</sub> Cd
Formula Weight g/mol	530.894

Temperature (K)	298
Crystal system	Monoclinic
Space group	P2 <sub>1</sub> /c
Crystal size	0.51×0.32×0.16 mm
a/Å	5.7859 (2)
b/Å	15.7906 (7)
c/Å	12.0818 (5)
α/°	90.00
β/°	95.403 (2)
γ/°	90.00
V/Å <sup>3</sup>	1098.92 (8)
Z	2
μ(Mo-Kα)/m.m <sup>-1</sup>	2.14
Calculated density/ mg.cm <sup>-3</sup>	1.604
Goodness-of-fit on F <sup>2</sup>	0.989
F(000)	532
R indices[I>3σ(I)] R1/wR2	0.0390/ 0.1341
R indices(all data)	0.0390/ 0.1068
Data / restraints / parameters	4744/0/130
CCDC	1527036

**Table S2** Bond lengths (Å) and bond angles (deg.) of the SCP **1**.

Cd1—N4 <sup>i</sup>	2.318 (2)	N4 <sup>i</sup> —Cd1—N4 <sup>ii</sup>	180.00 (10)
Cd1—N4 <sup>ii</sup>	2.318 (2)	N4 <sup>i</sup> —Cd1—N16	89.01 (9)
Cd1—N16	2.398 (2)	N4 <sup>ii</sup> —Cd1—N16	90.99 (9)
Cd1—N16 <sup>iii</sup>	2.398 (2)	N4 <sup>i</sup> —Cd1—N16 <sup>iii</sup>	90.99 (9)
Cd1—S2	2.7291 (8)	N4 <sup>ii</sup> —Cd1—N16 <sup>iii</sup>	89.01 (9)
Cd1—S2 <sup>iii</sup>	2.7291 (8)	N16—Cd1—N16 <sup>iii</sup>	180.00 (13)
S2—C12	1.6312 (7)	N4 <sup>i</sup> —Cd1—S2	93.65 (7)
C3—N16	1.342 (4)	N4 <sup>ii</sup> —Cd1—S2	86.35 (7)
Cd1-Cd1 <sup>i</sup>	5.786(8)	N4 <sup>i</sup> —Cd1—S2 <sup>iii</sup>	86.35 (7)
Cd1-Cd1 <sup>iii</sup>	14.554(4)	N4 <sup>ii</sup> —Cd1—S2 <sup>iii</sup>	93.65 (7)

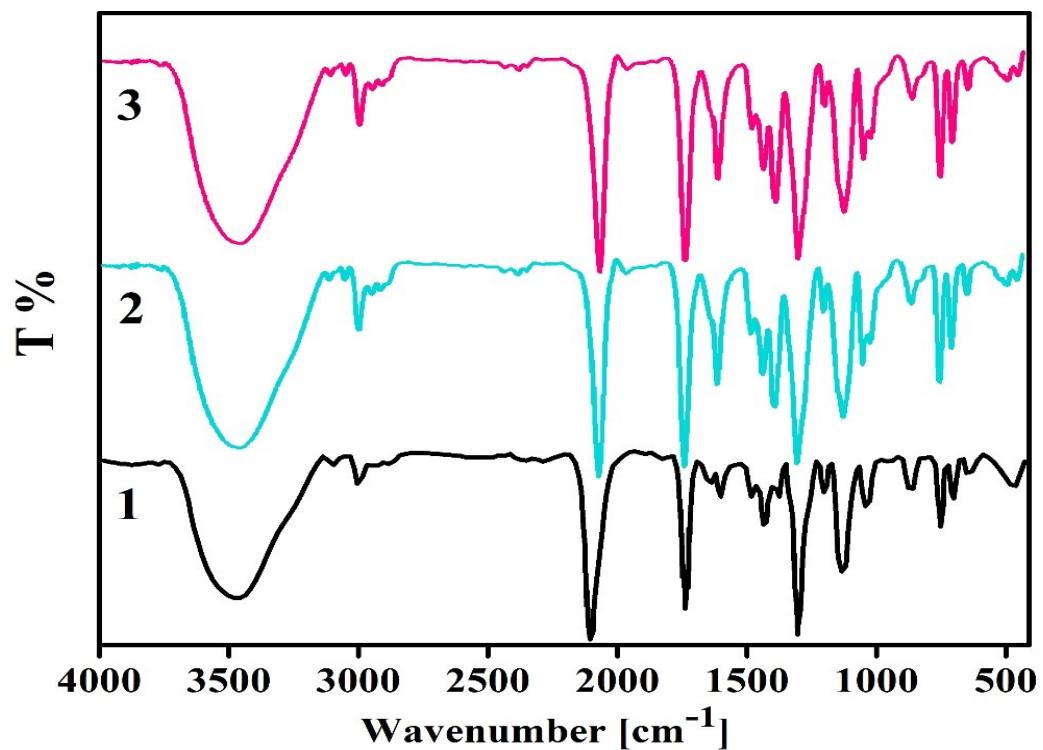
Symmetry codes: (i) -x,y,1/2-z; (ii) -x,-y,1-z; (iii) x,-y,z-1/2;

**Table S3** Hydrogen bond lengths (Å) and bond angles (deg) of the SCP **1**.

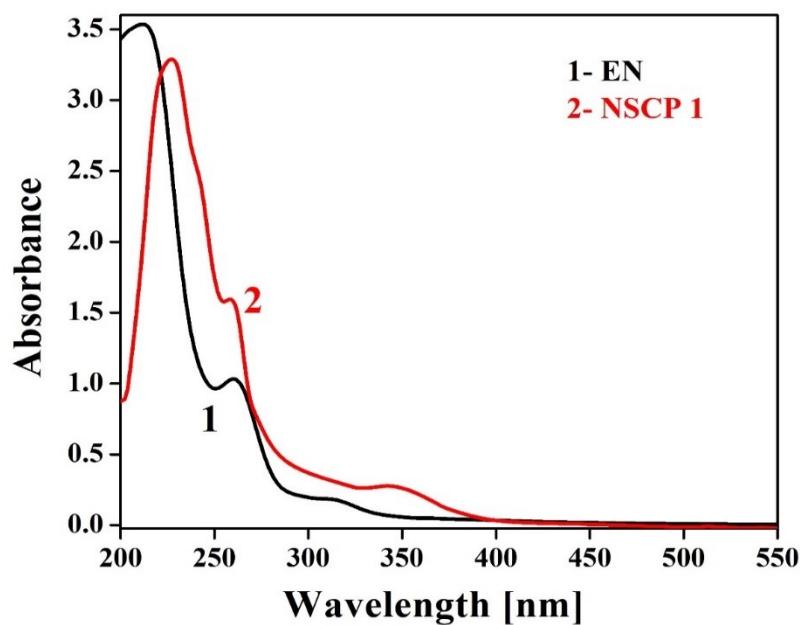
D-H....A	d(D-H)	d(H....A)	d(D....A)	∠(DHA)
C9-H9.....S2	0.960	2.824	3.392	118.76
C6-H6.....N3	0.960	2.844	3.333	112.60
C15-H15A.....N3	0.960	2.947	3.780	145.80
C13-H13.....S2	0.960	3.006	3.875	151.33

C12-H12.....S2	0.960	3.0541	3.733	128.97
C15-H15C.....S2	0.960	3.196	4.107	159.12

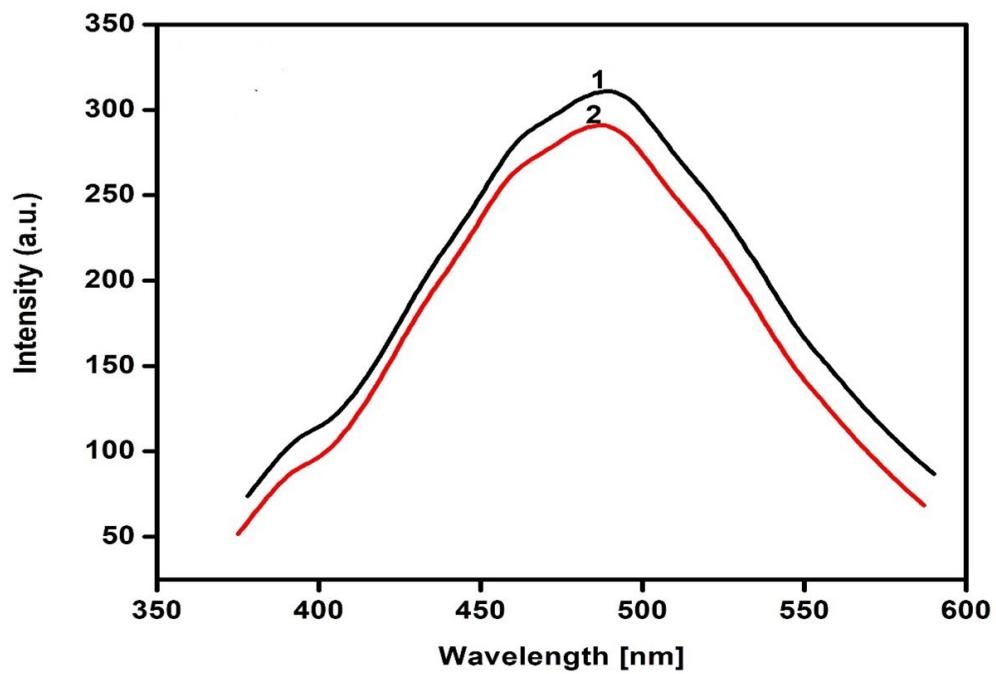
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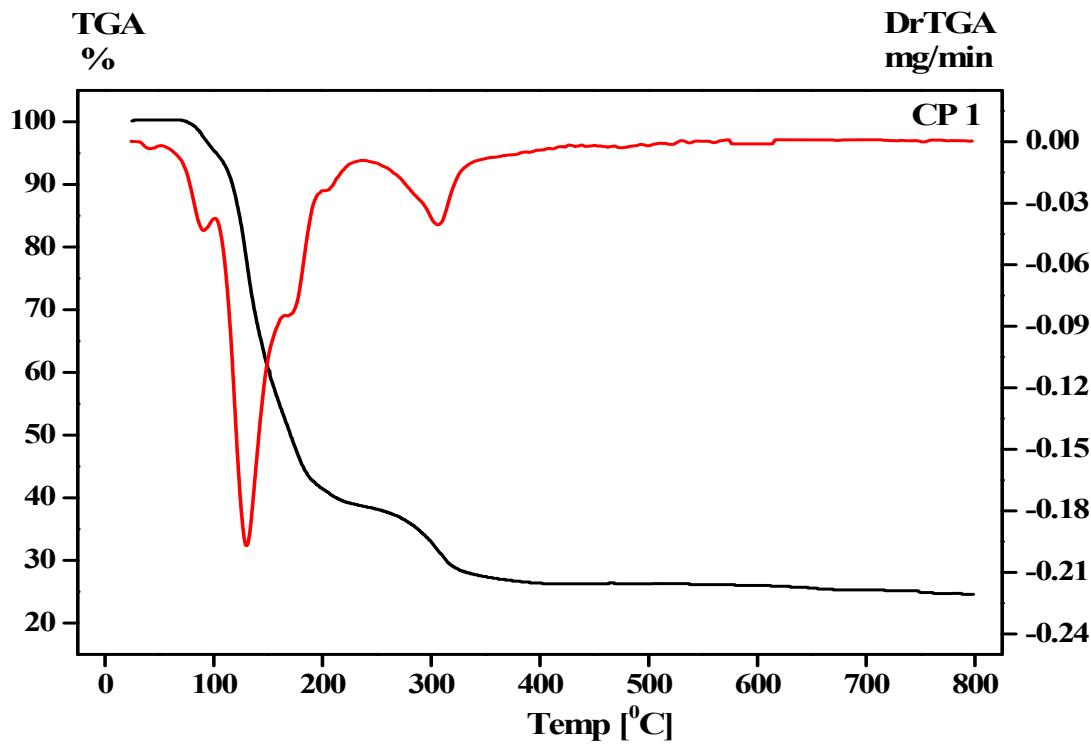
**Figure S 1** IR spectra of SCP1 (1), NSCP1 (2) before, and after catalysis (3).



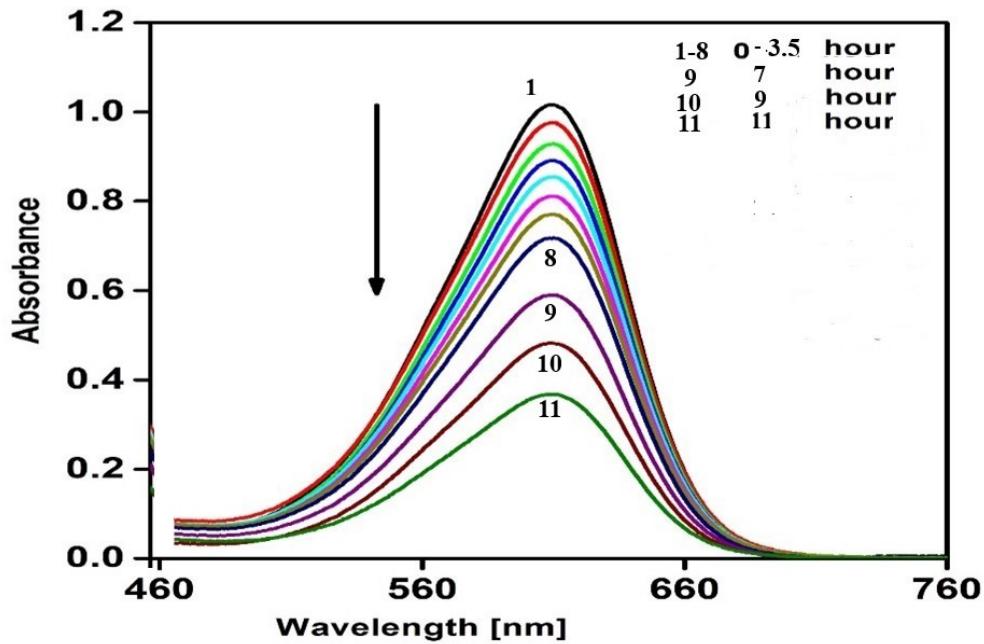
**Figure S2** UV-spectra of EN and NSCP1



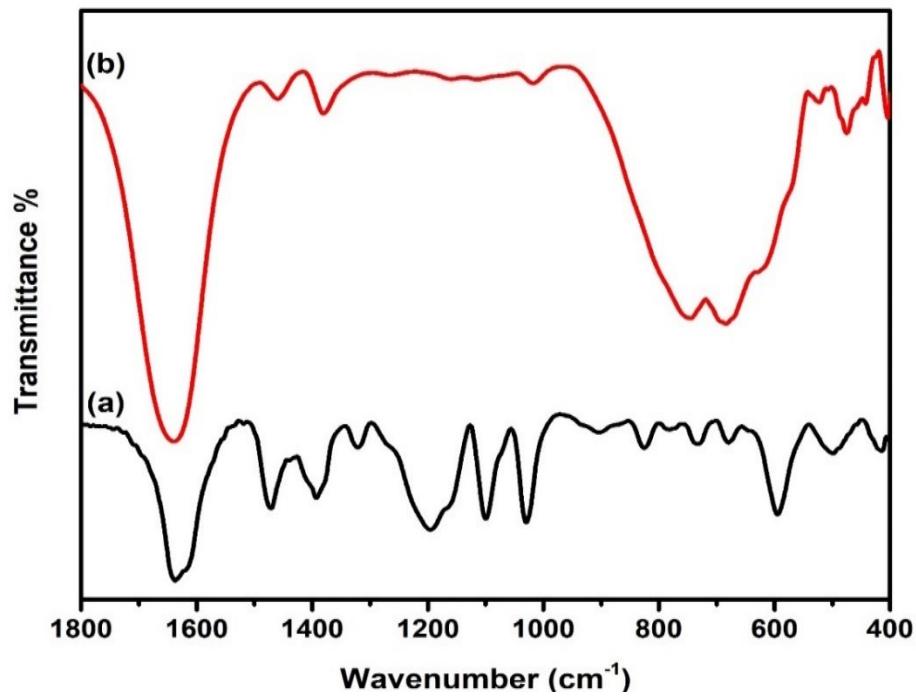
**Figure S3** Emission-spectra of (1) SCP1 and (2) NSCP1



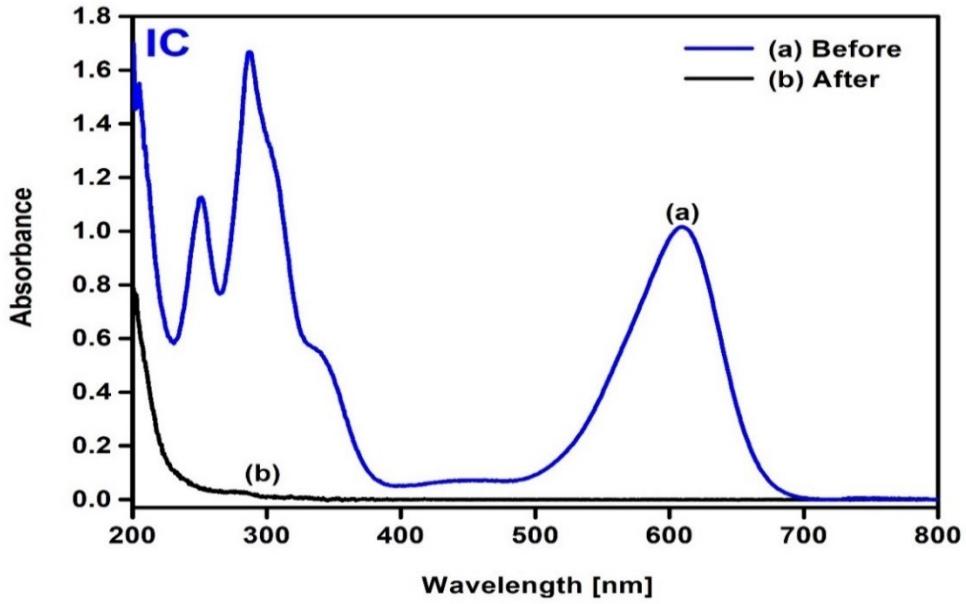
**Figure S 4** TGA and DTG thermal analysis curves of the NSCP **1**.



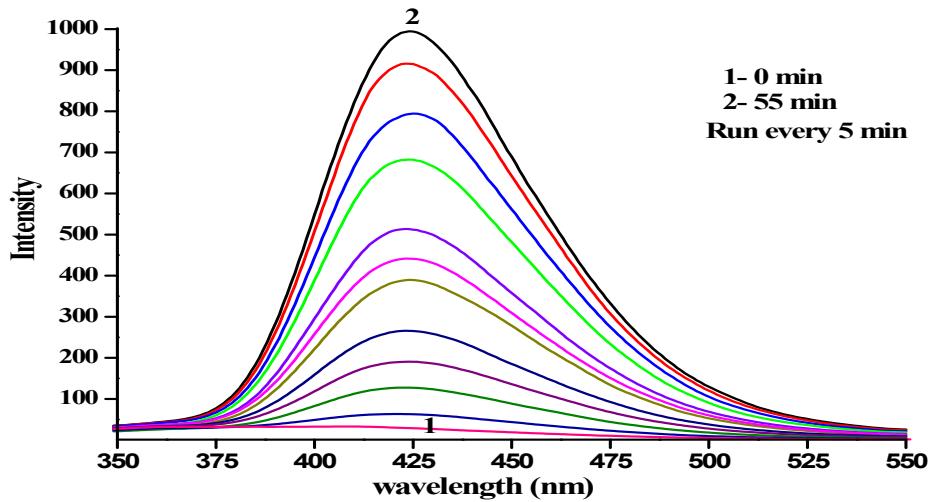
**Figure S 5** Control spectra of degradation of the IC dye solution ( $7.0 \times 10^{-5}$  M) under UV-light irradiation.



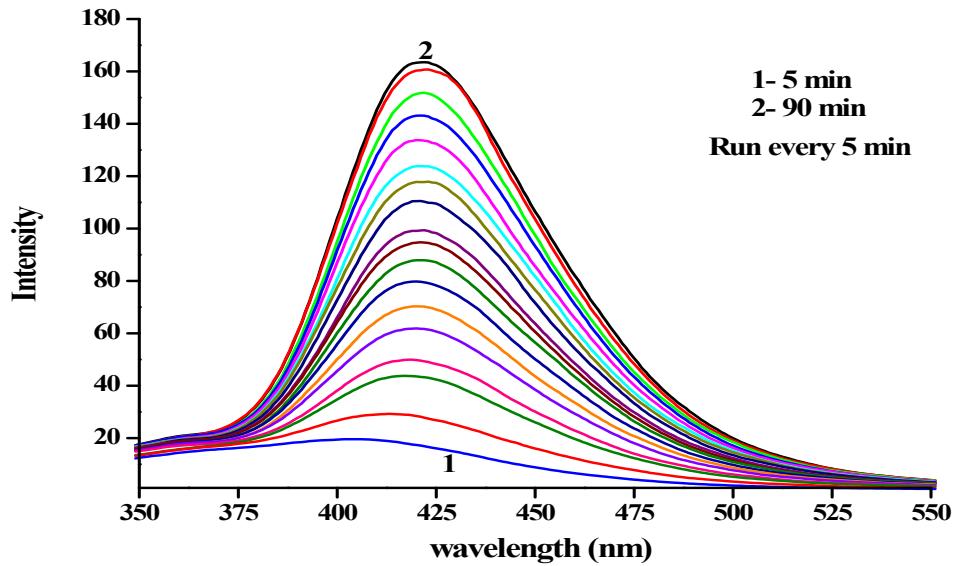
**Figure S6** Indigo carmine dye (IC) infrared spectra (a) prior to and (b) after degradation.



**Figure S7** The electronic absorption spectra of Indigo carmine dye (IC) (a) before and (b) after degradation.



**Figure S8** SCP1 (25 mg), NaTA (0.05 M), and H<sub>2</sub>O<sub>2</sub> (0.1 M) photoluminescence spectra as a function of time.



**Figure S9** Photoluminescence spectra of the SCP1 (25 mg), indigo carmin (7.0x10<sup>-5</sup> M), NaTA (0.05 M) and H<sub>2</sub>O<sub>2</sub> (0.1M) as function of time



