

## Enhanced removal of emerging contaminants from tap water by developing graphene oxide and nanoplatelet hybrid aerogels

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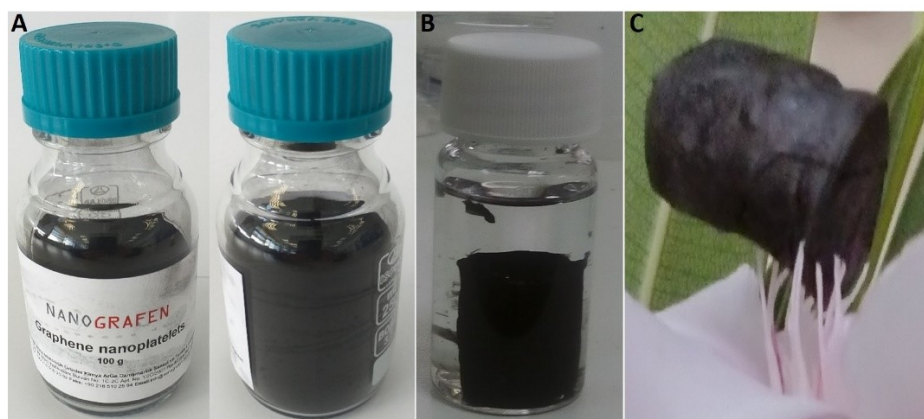
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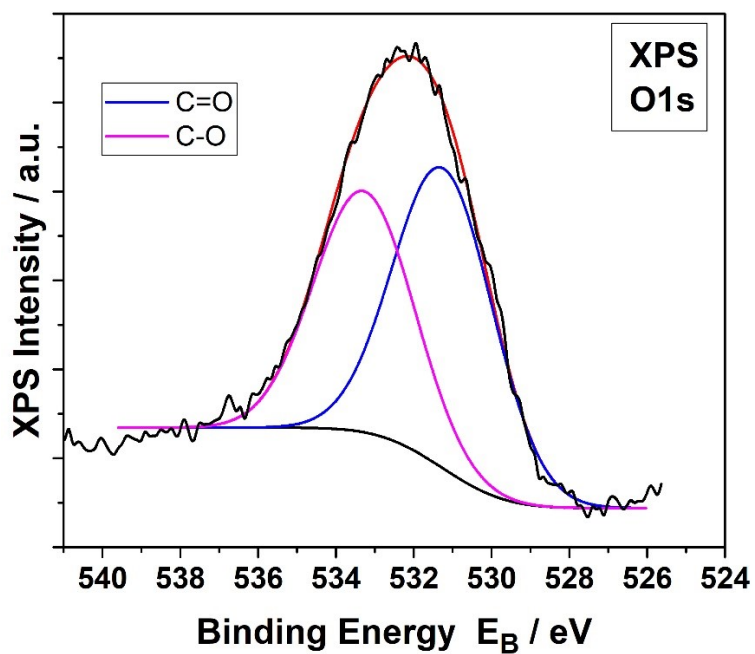
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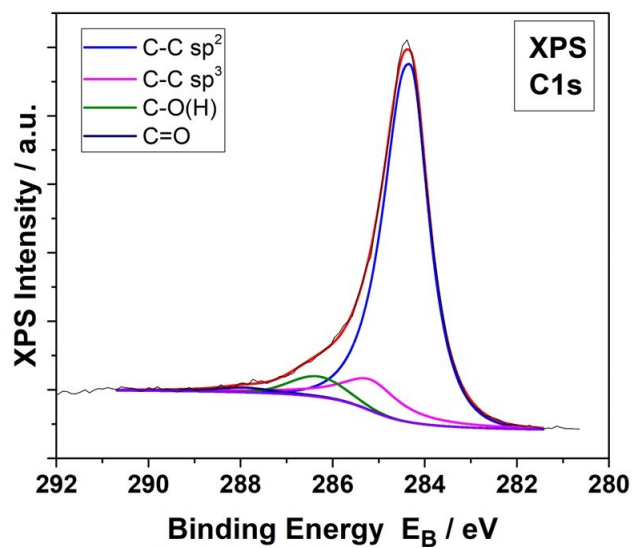
**Figure S1:** Graphene nanoplatelets derived from waste tires (GNP) and Graphene aerogels synthesis. (A) Digital image of the used GNPs powder from waste tires, (B) Graphene (rGO) hydrogel, and, (C) the as-obtained graphene aerogel after the freeze-drying.

**Table S1:** XPS data of the starting GNPs powder: % C1s and O1s component concentration derived from the C1s and O1s peak deconvolution respectively and relative atomic ratio C:O:Fe:Si

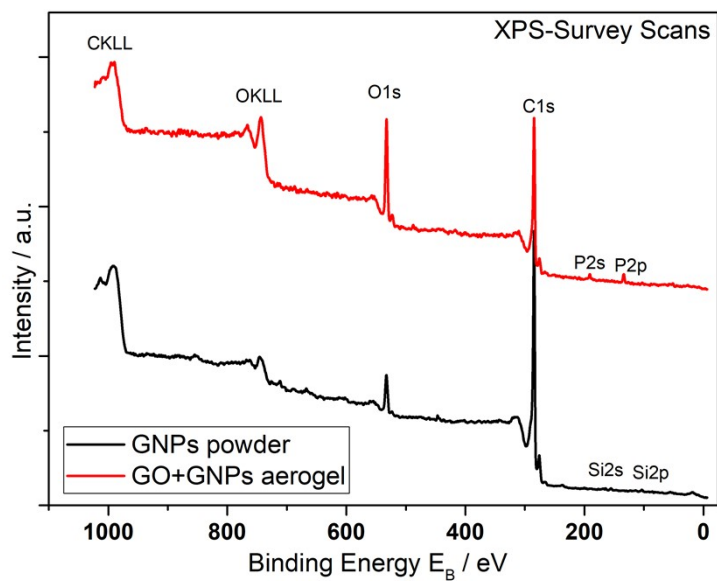
% component of O bonds		% Relative atomic ratio				
C=O	C-O	C-C sp <sup>2</sup>	C-C sp <sup>3</sup>	C-O(H)	C=O	C:O:Fe:Si
43.0	56.0	81.1	12.0	5.1	1.8	93.5%, 5.5%, 0.20%, 0.80%



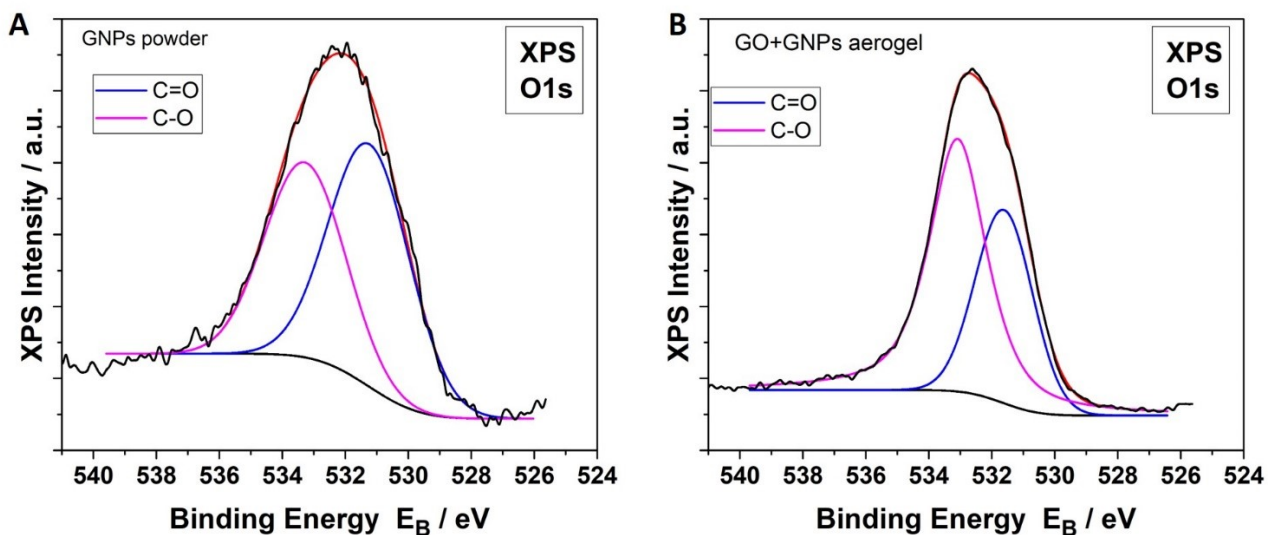
**Figure S2:** Deconvoluted O1s XPS spectrum of the GNPs powder.



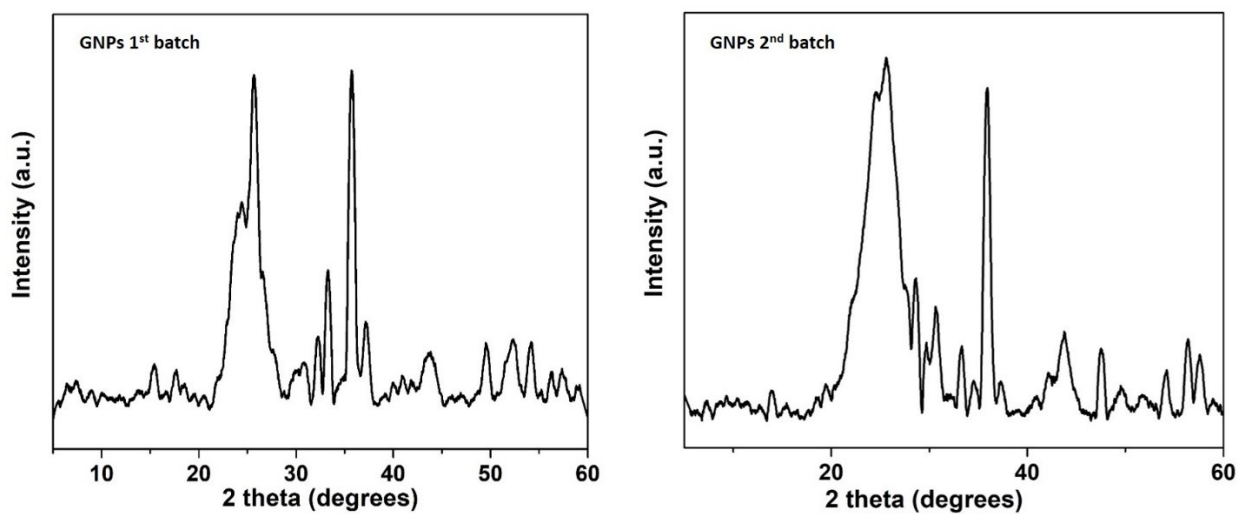
**Figure S3:** Deconvoluted C1s XPS spectrum of the GNPs powder.



**Figure S4:** XPS Survey Scans from GNPs powder and GO+GNPs powder.



**Figure S5:** Deconvoluted O1s XPS spectra from (A) the GNPs powder, and, (B) the GO+GNPs aerogel.



**Figure S6:** XRD patterns of the two examined products (1<sup>st</sup> batch and 2<sup>nd</sup> batch of GNPs derived from waste tires).

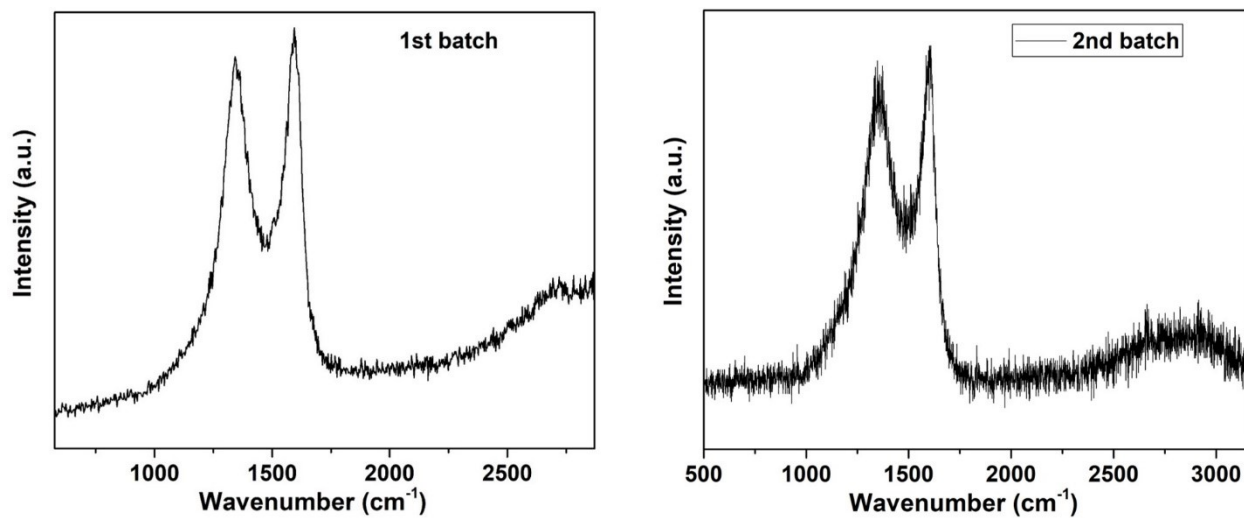


Figure S7: Raman spectra of the examined products (GNPs from the 1<sup>st</sup> and 2<sup>nd</sup> batches).

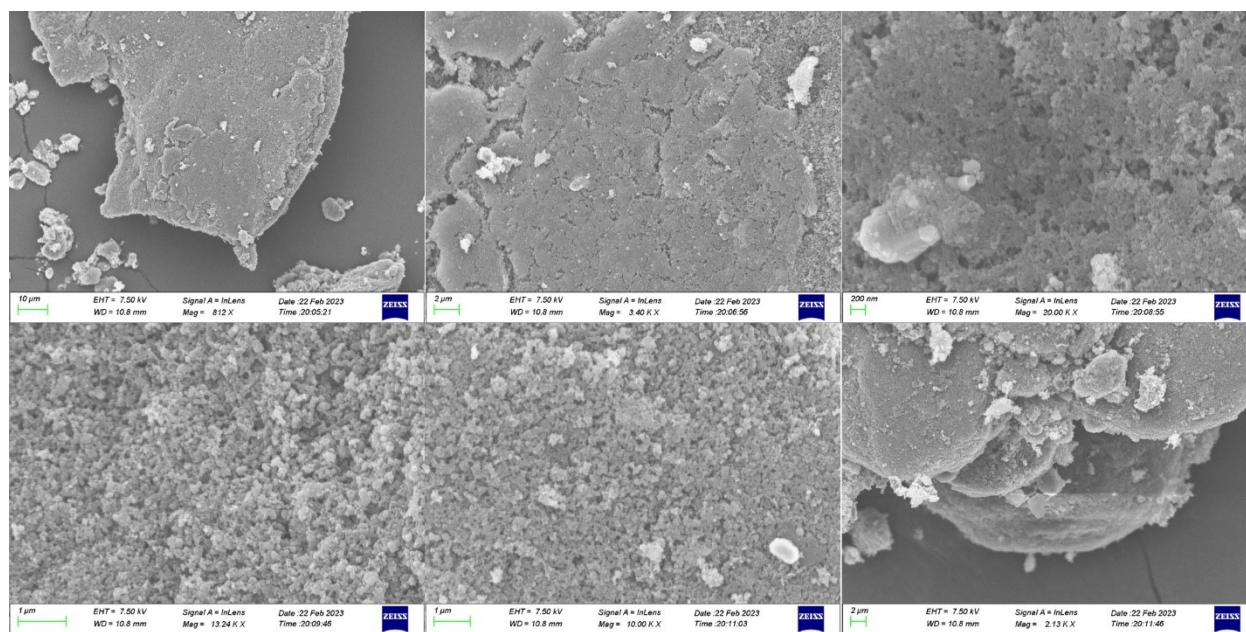
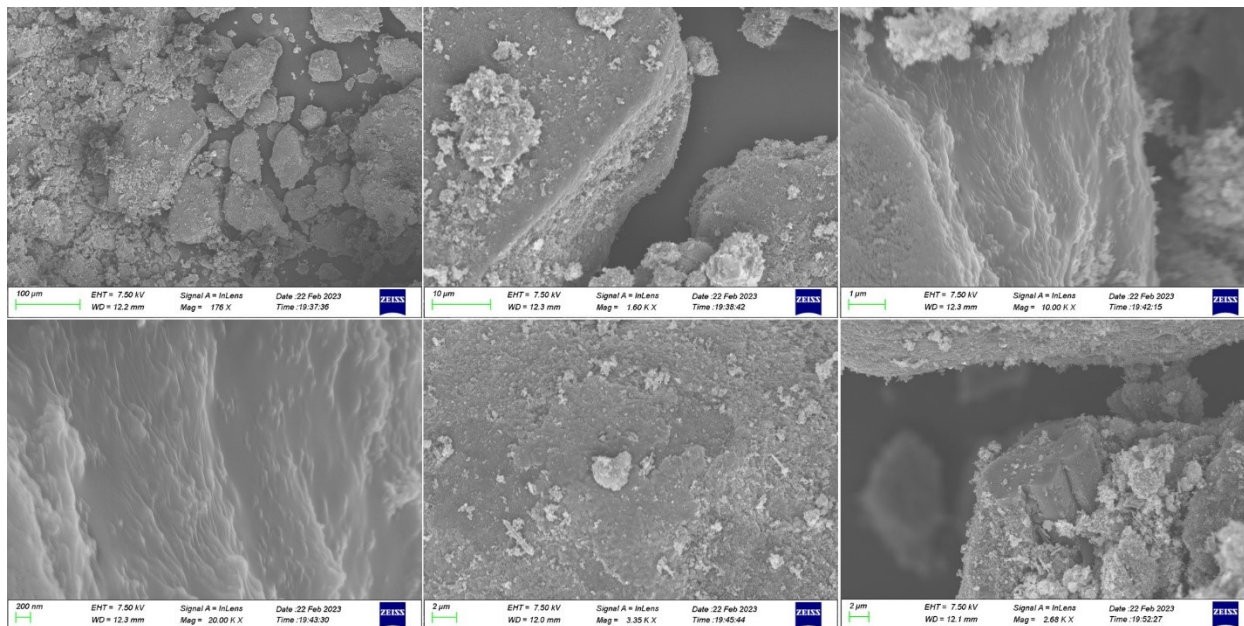
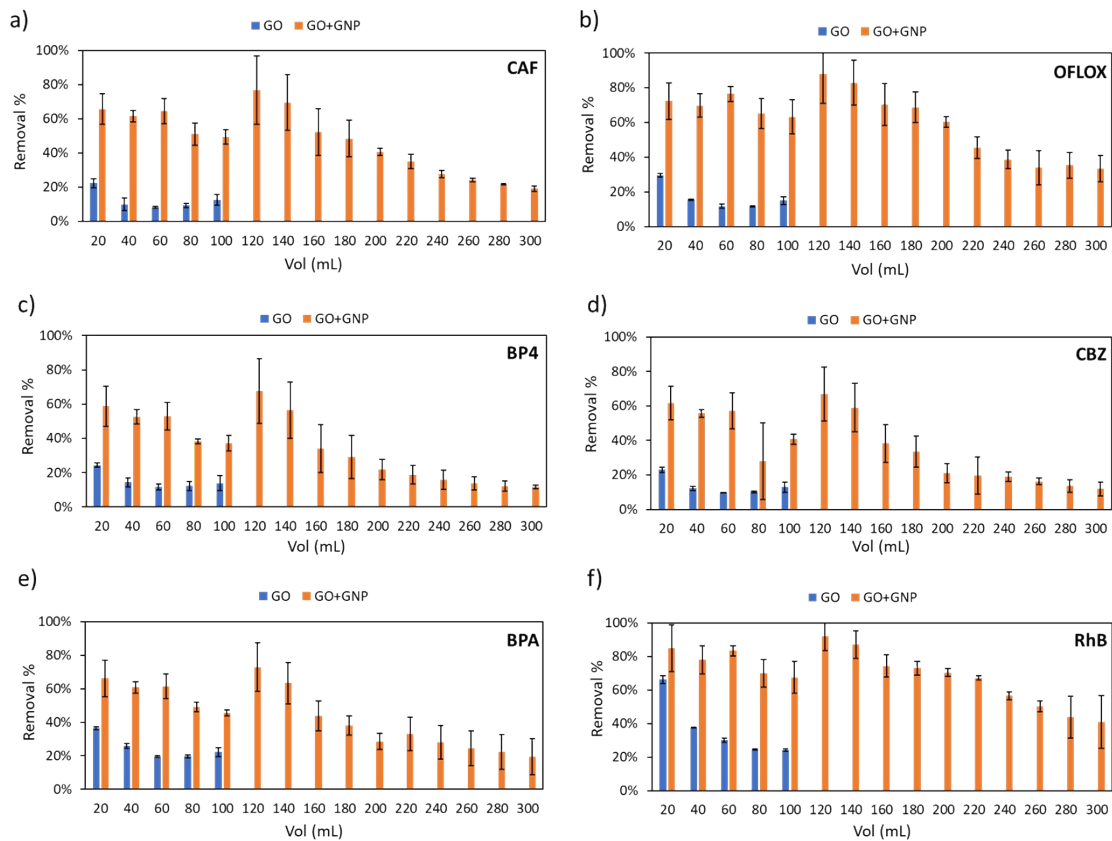


Figure S8: SEM images of the GNPs from 1<sup>st</sup> batch.



**Figure S9:** SEM images of the GNPs from 2<sup>nd</sup> batch.





**Figure S10.** Removal of a) CAF, b) OFLOX, c) BP4, d) CBZ, e) BPA, f) RhB obtained by aerogels of GO (blue) and GO/GNP (orange) (0.5 mg/L each in tap water,  $V_{\text{tot}} = 300$  mL, flow rate = 2.5 mL/min).