

## CONTENTS

- 1. Structural Characterization Information**
- 2. Supporting Figures**
- 3. Supporting Tables**

## 1. Structural Characterization Information

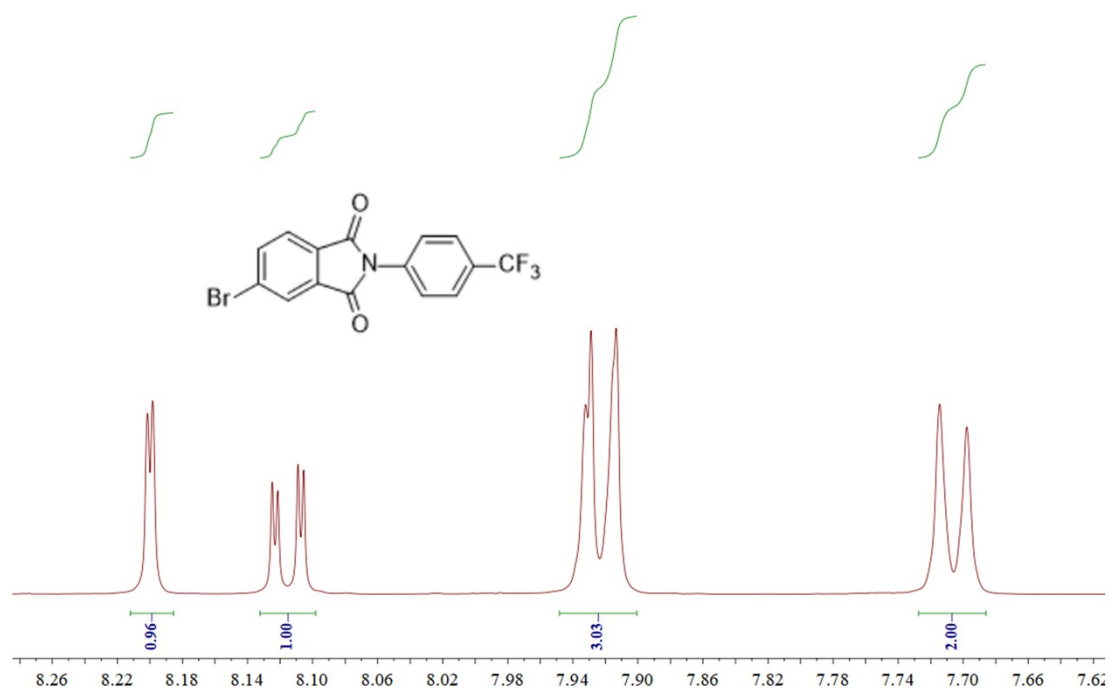


Figure S1. The <sup>1</sup>H NMR spectrum of 3FmBr in DMSO-d<sub>6</sub>.

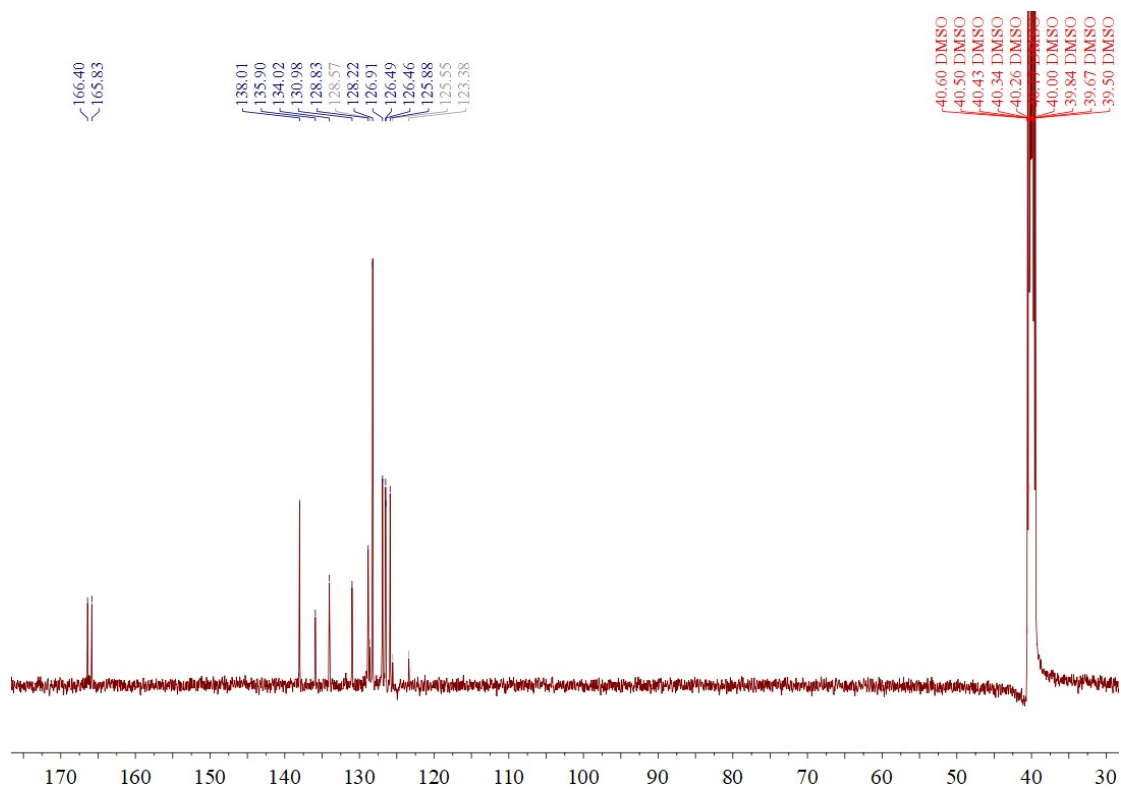
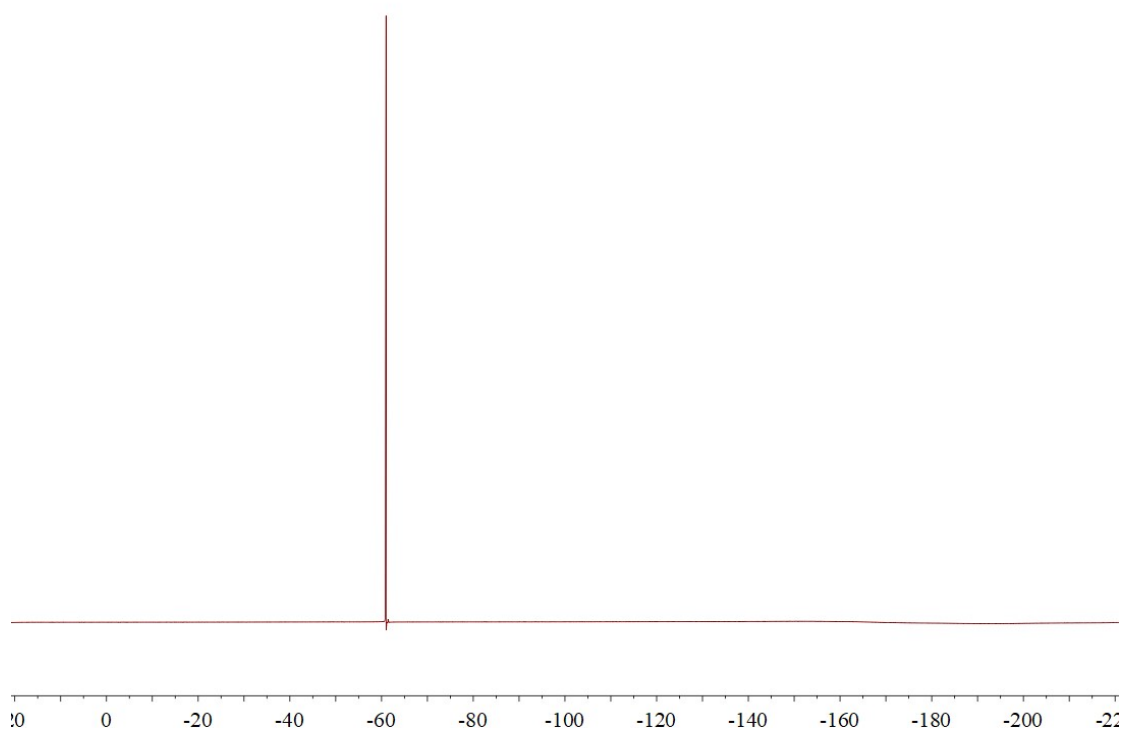
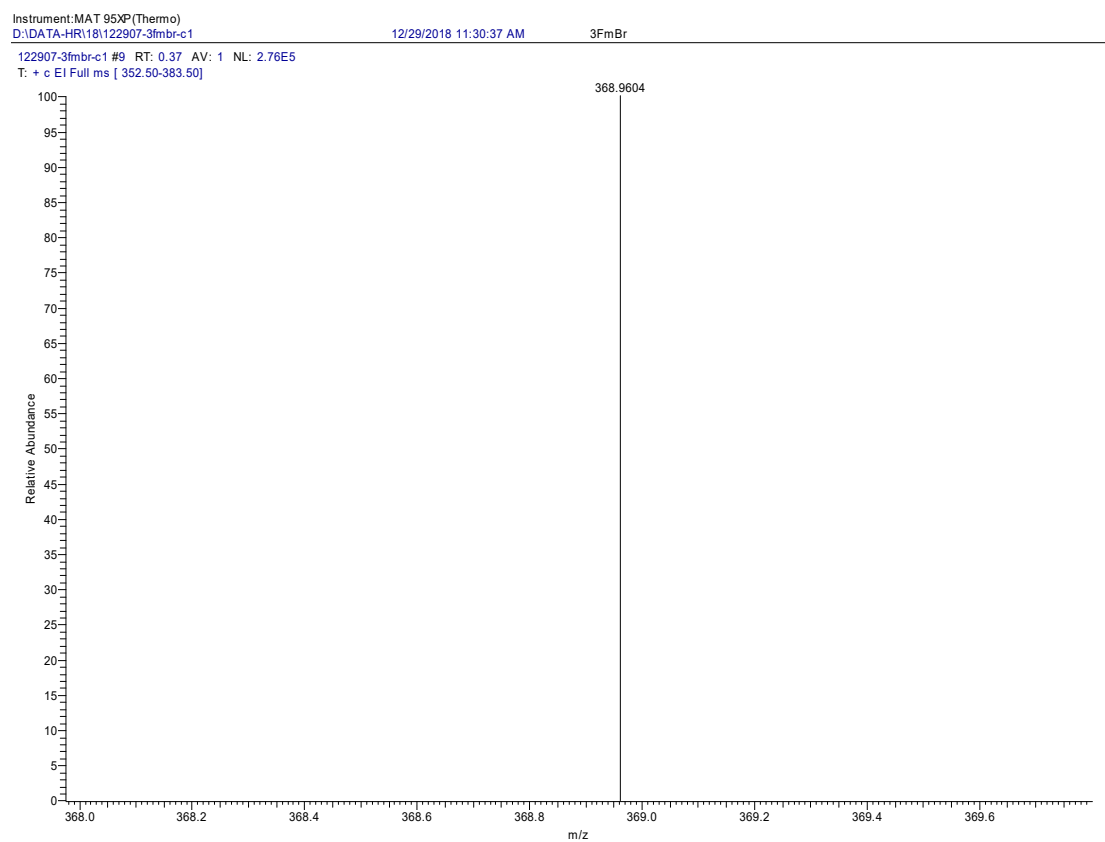


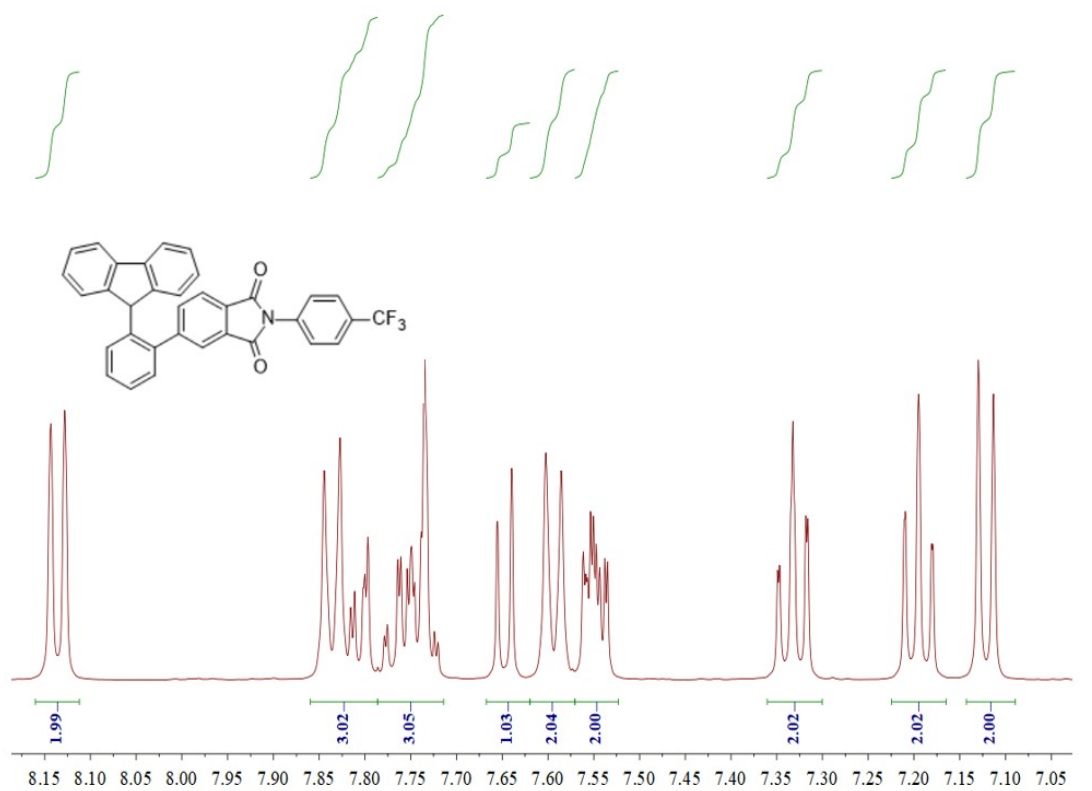
Figure S2. The <sup>13</sup>C NMR spectrum of 3FmBr in DMSO-d<sub>6</sub>.



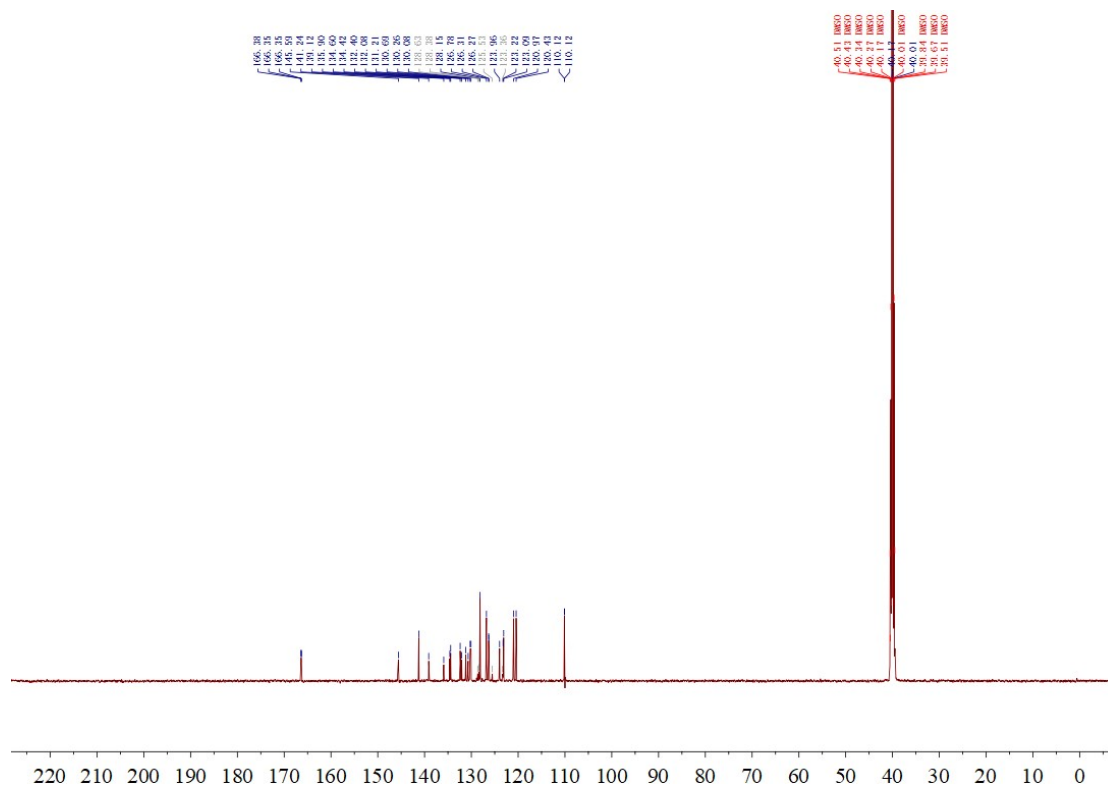
**Figure S3.** The  $^{19}\text{F}$  NMR spectrum of 3FmBr in  $\text{DMSO-d}_6$ .



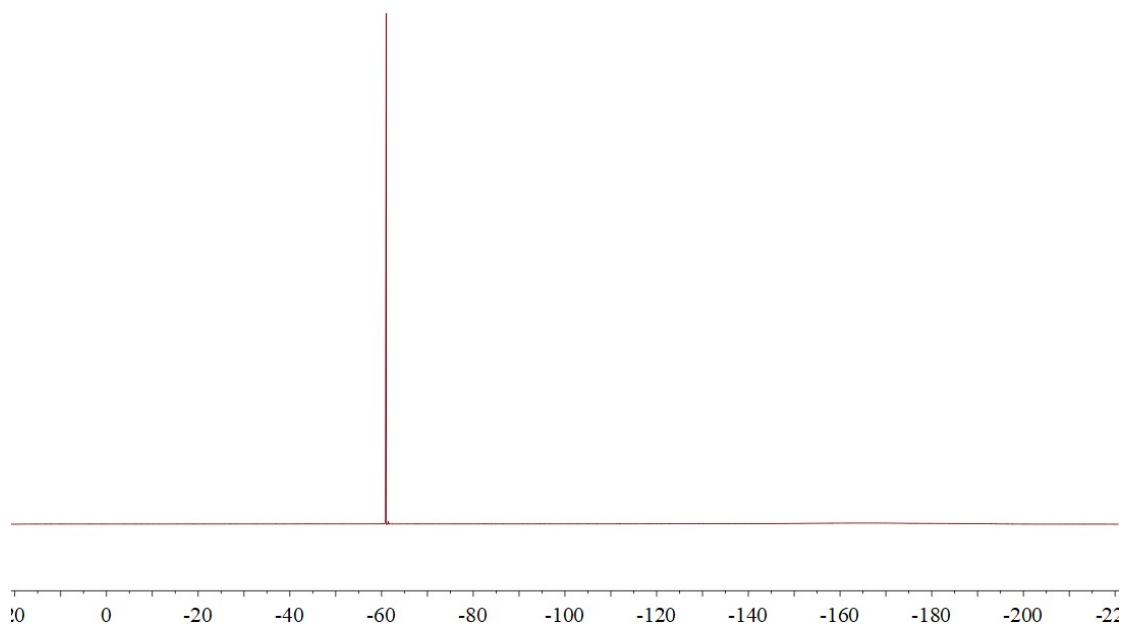
**Figure S4.** The high resolution mass spectra of 3FmBr.



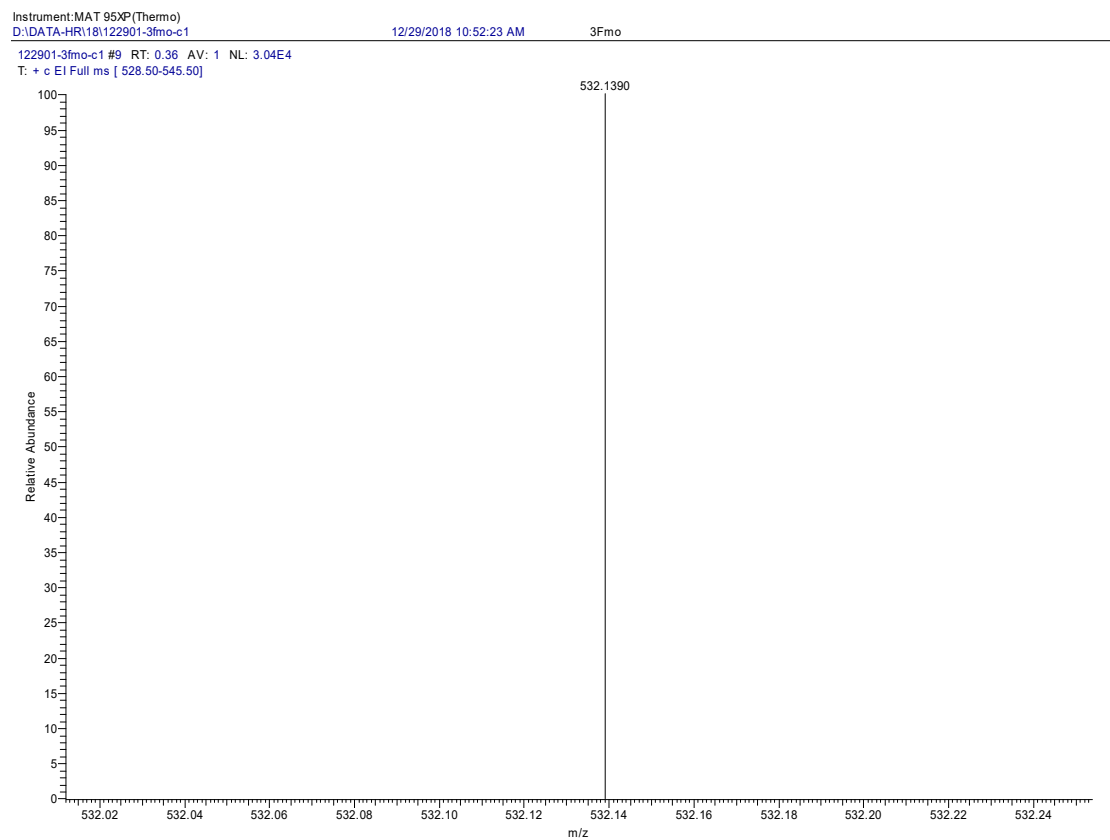
**Figure S5.** The <sup>1</sup>H NMR spectrum of 3Fmo in DMSO-d<sub>6</sub>.



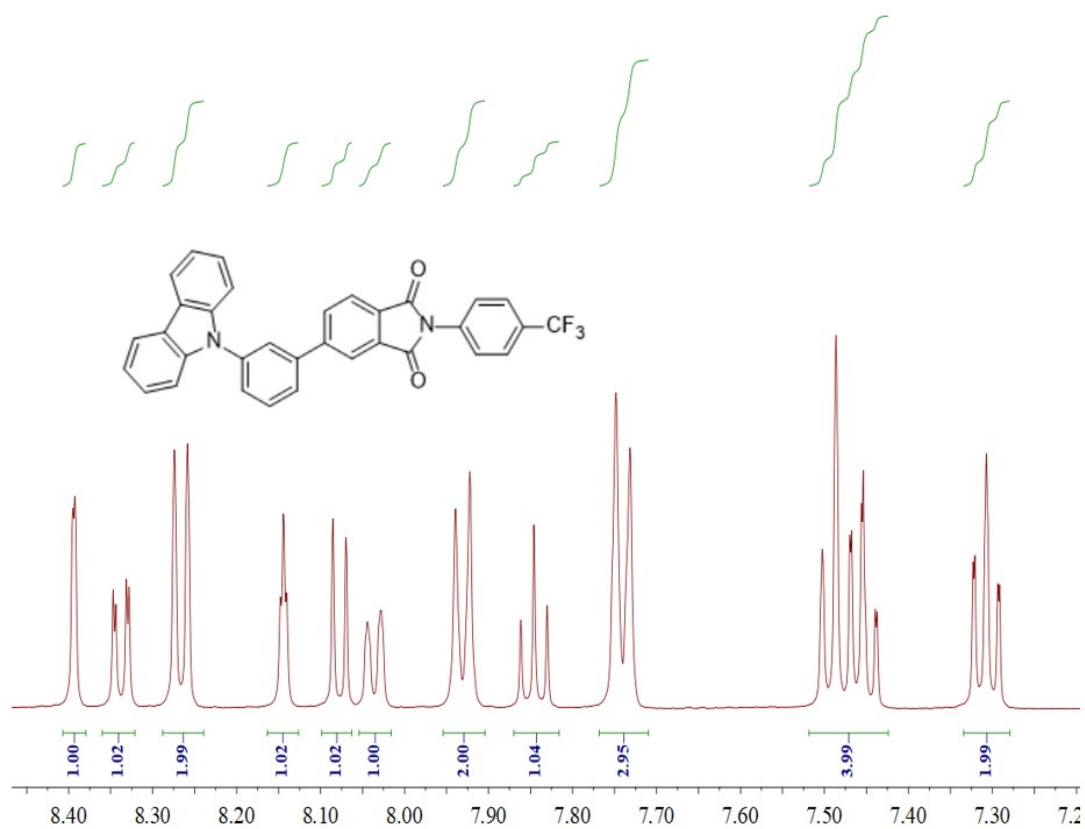
**Figure S6.** The <sup>13</sup>C NMR spectrum of 3Fmo in DMSO-d<sub>6</sub>.



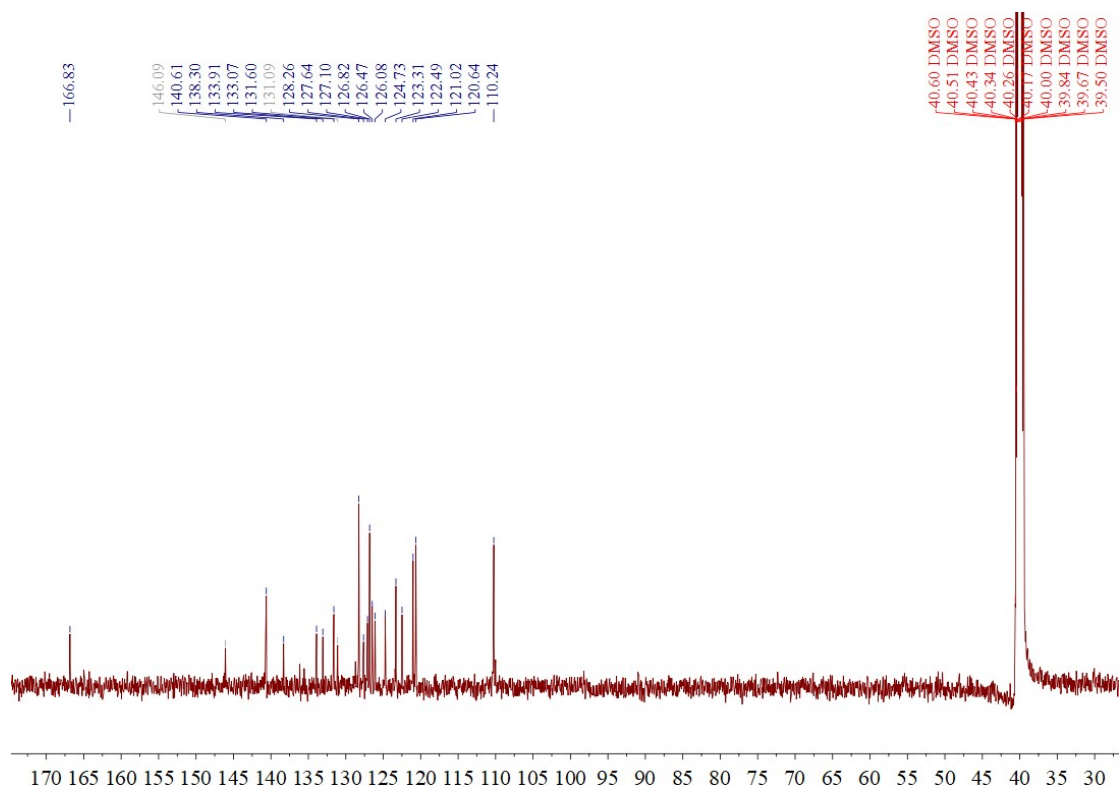
**Figure S7.** The  $^{19}\text{F}$  NMR spectrum of 3Fmo in  $\text{DMSO-d}_6$ .



**Figure S8.** The high resolution mass spectra of 3Fmo.



**Figure S9.** The  $^1\text{H}$  NMR spectrum of 3Fmm in  $\text{DMSO-d}_6$ .



**Figure S10.** The  $^{13}\text{C}$  NMR spectrum of 3Fmm in  $\text{DMSO-d}_6$ .

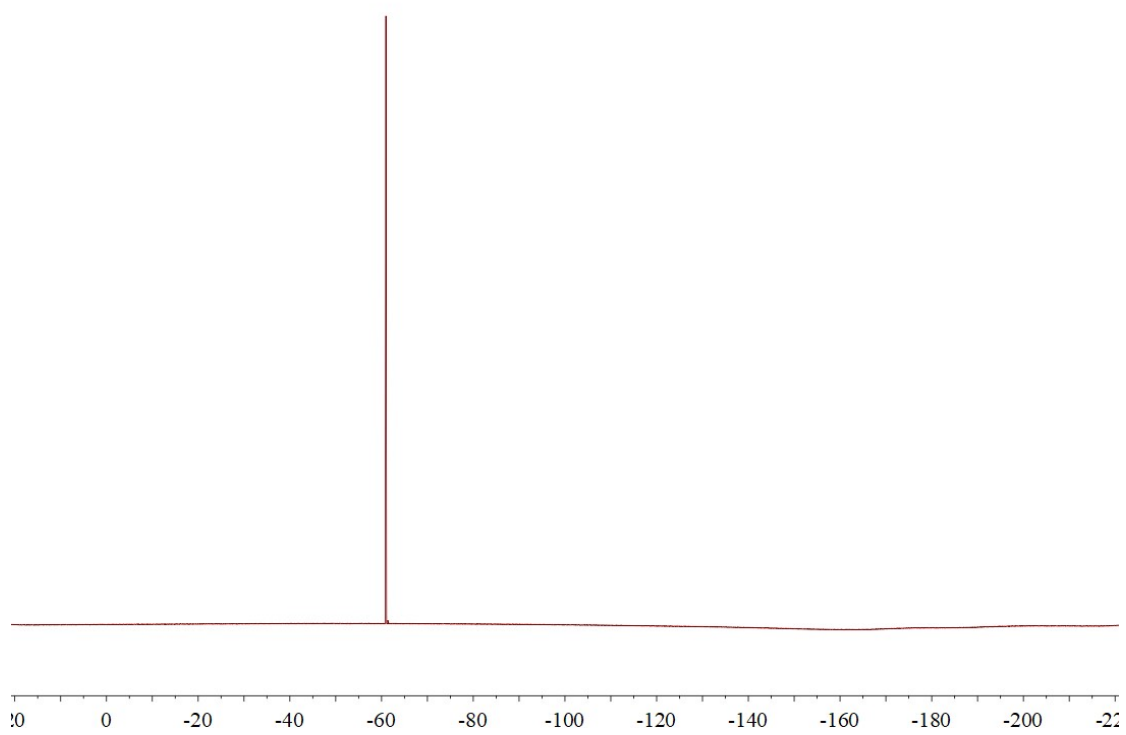


Figure S11. The  $^{19}\text{F}$  NMR spectrum of 3Fmm in  $\text{DMSO-d}_6$ .

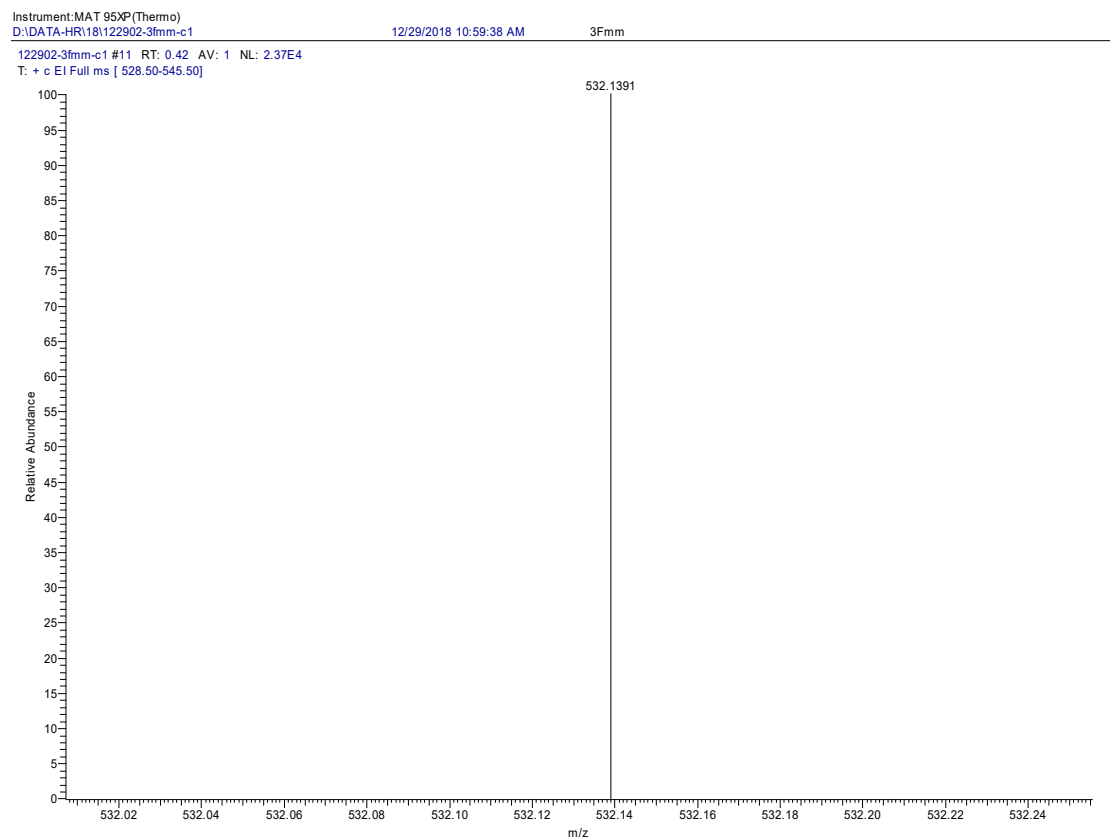


Figure S12. The high resolution mass spectra of 3Fmm.

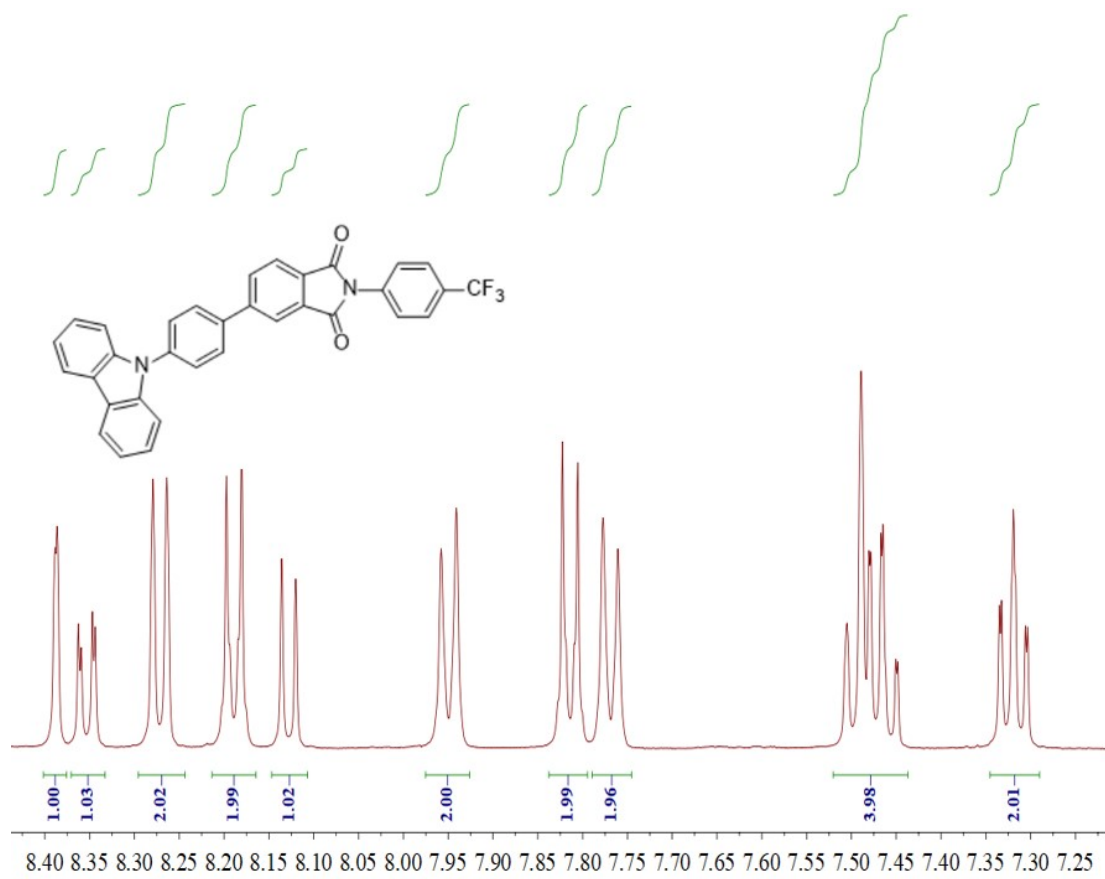


Figure S13. The <sup>1</sup>H NMR spectrum of 3Fmp in DMSO-d<sub>6</sub>.

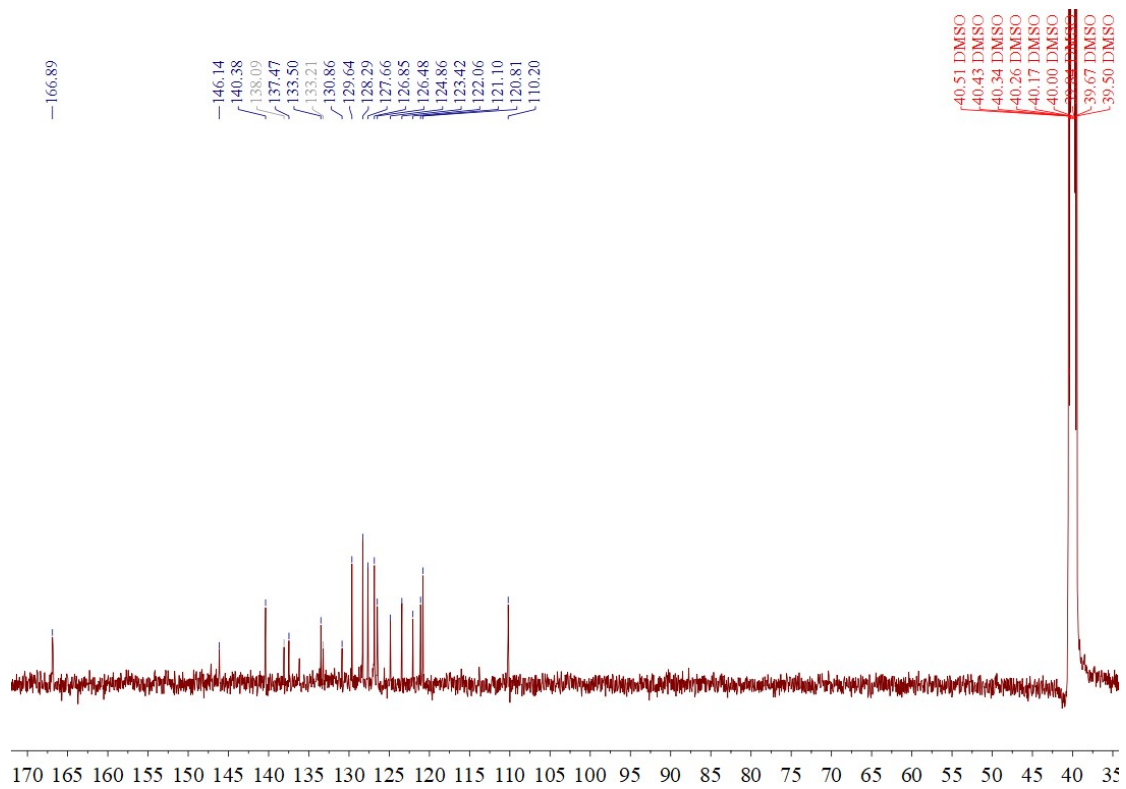
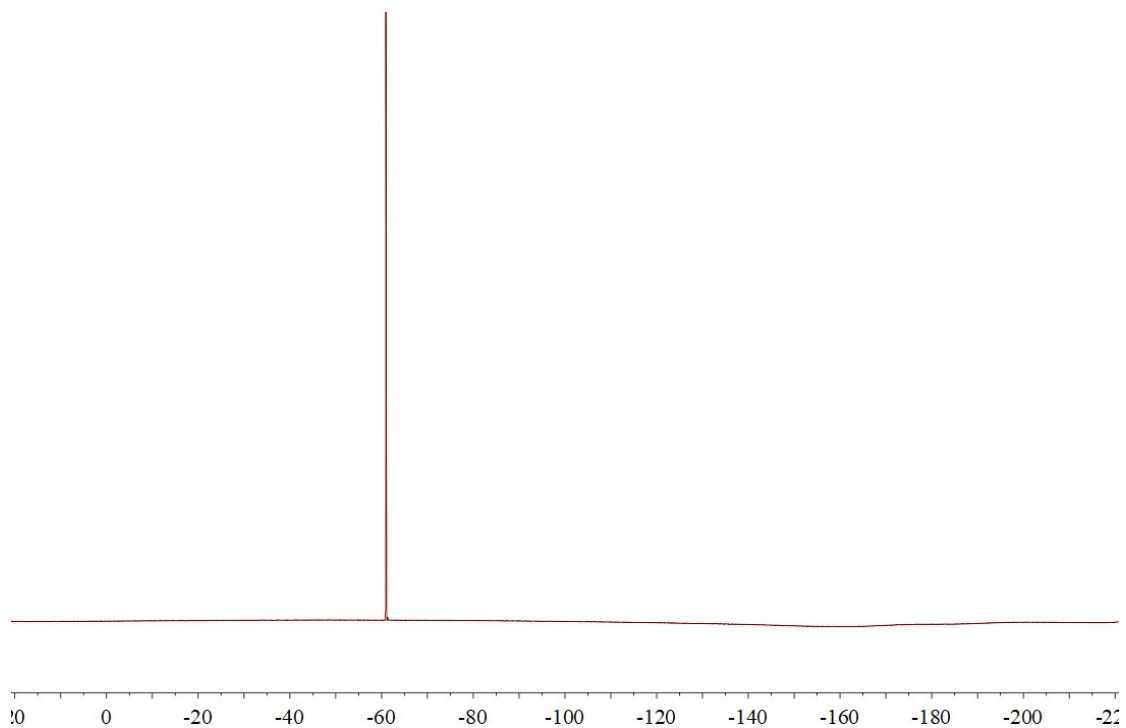
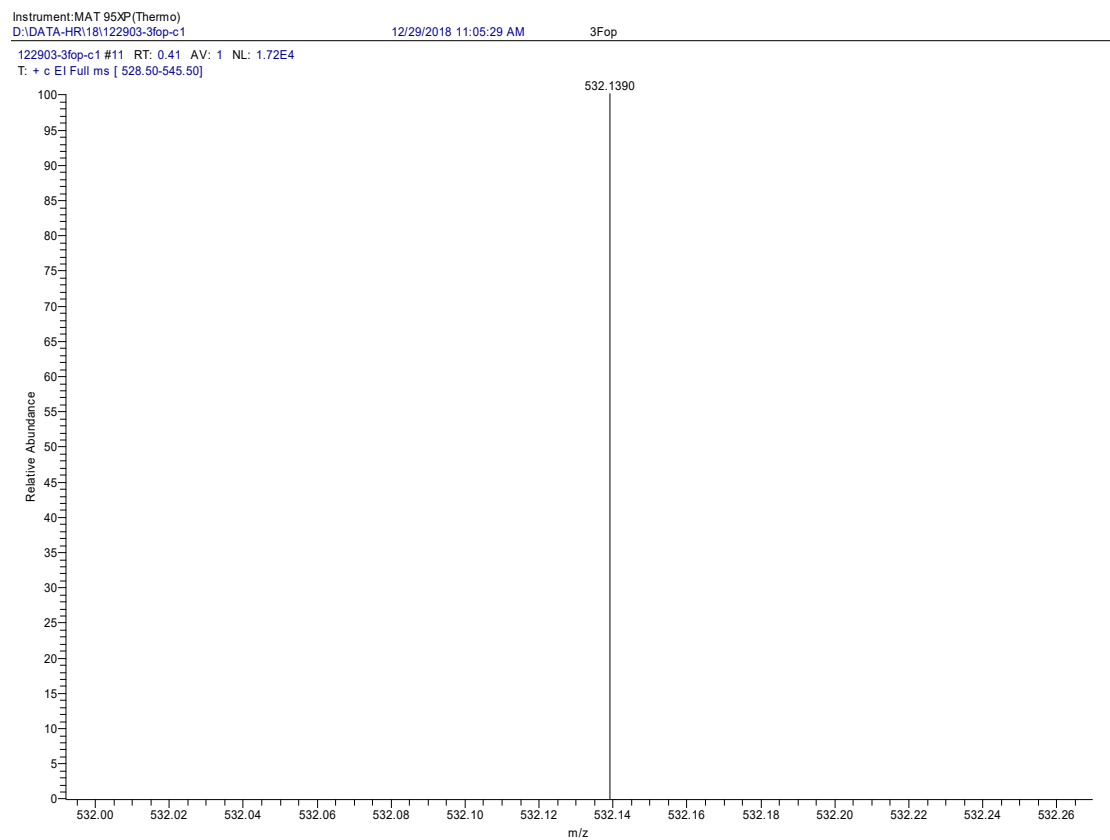


Figure S14. The <sup>13</sup>C NMR spectrum of 3Fmp in DMSO-d<sub>6</sub>.



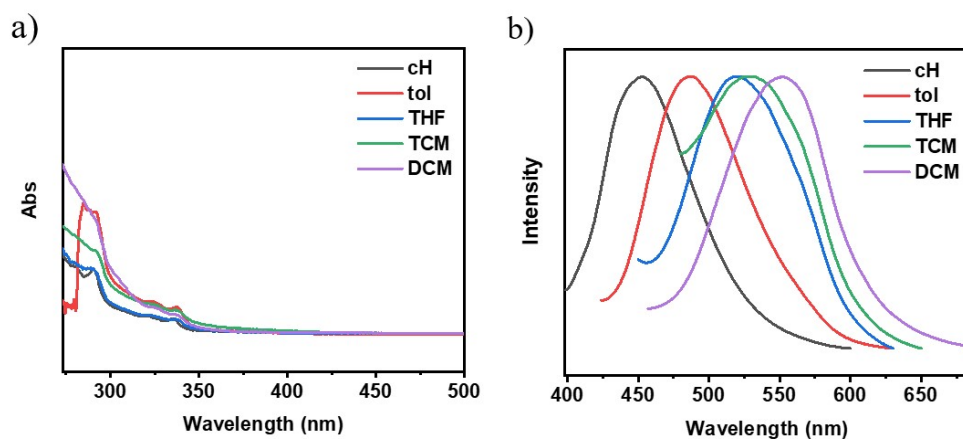


**Figure S15.** The  $^{19}\text{F}$  NMR spectrum of 3Fmp in  $\text{DMSO-d}_6$ .

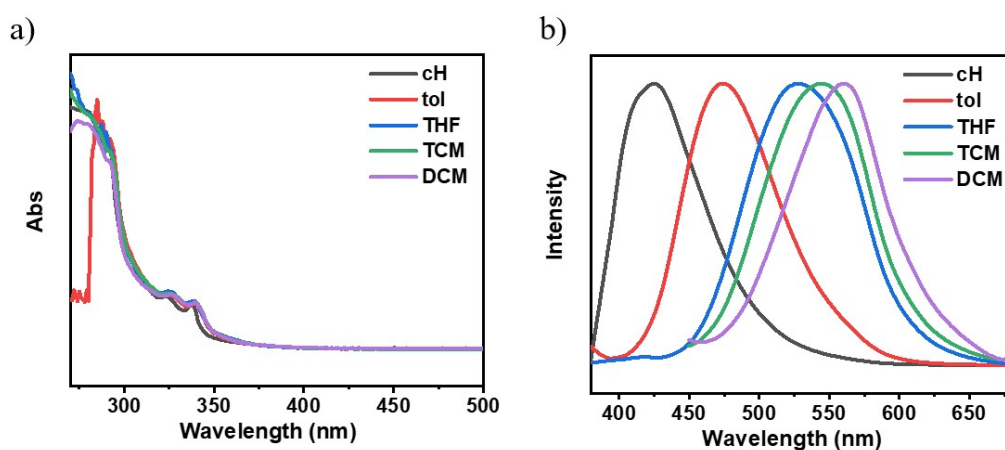


**Figure S16.** The high resolution mass spectra of 3Fmps.

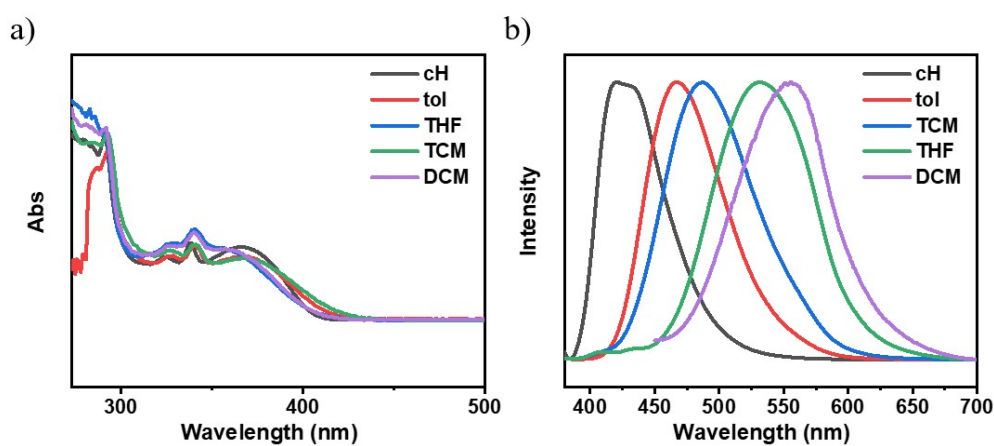
## 2. Supporting Figures



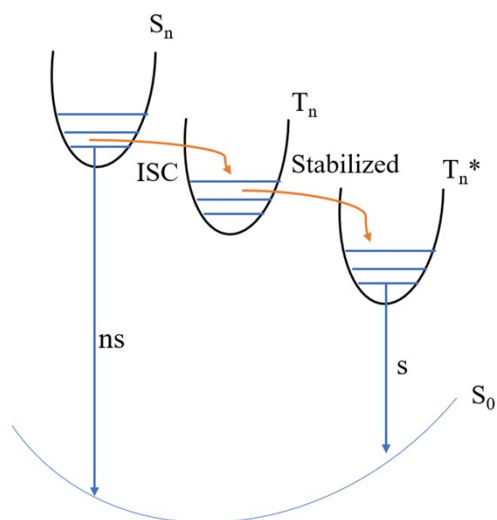
**Figure S17.** The UV-visible absorption spectrum (a) and fluorescence spectrum (b) of 3Fmo in dilute solutions ( $c=10^{-5}$  M) of varying polarity.



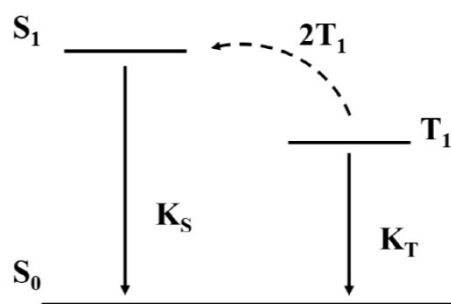
**Figure S18.** The UV-visible absorption spectrum (a) and fluorescence spectrum (b) of 3Fmm in dilute solutions ( $c=10^{-5}$  M) of varying polarity.



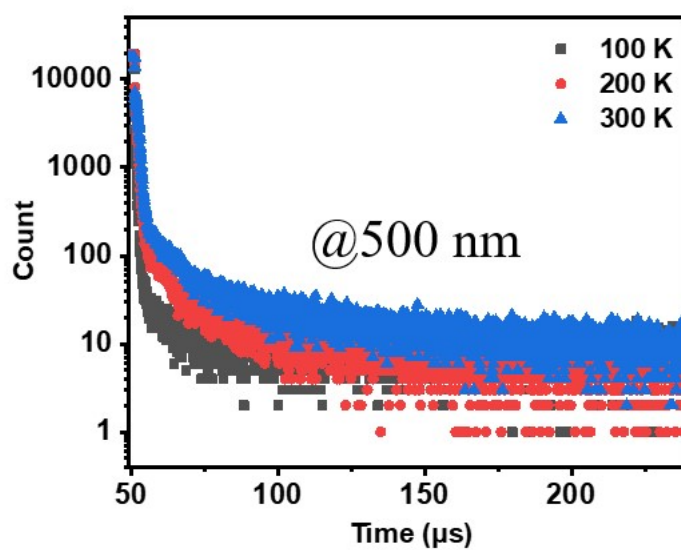
**Figure S19.** The UV-visible absorption spectrum (a) and fluorescence spectrum (b) of 3Fmp in dilute solutions ( $c=10^{-5}$  M) of varying polarity.



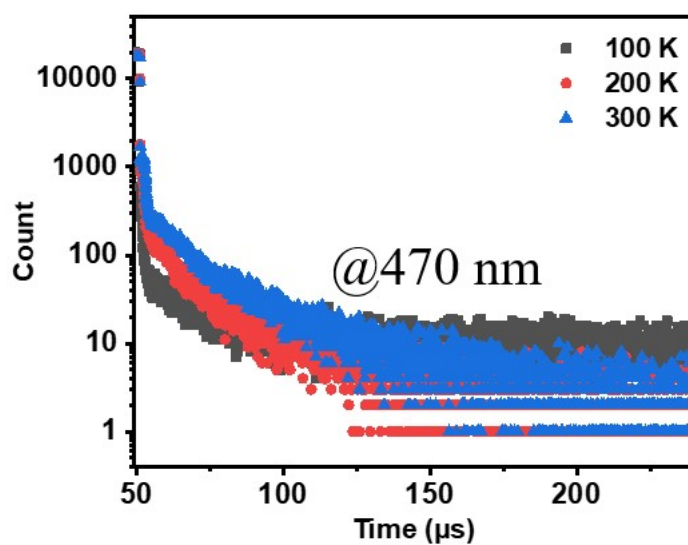
**Figure S20.** Schematic diagram of the multi-molecular phosphorescence emission process.



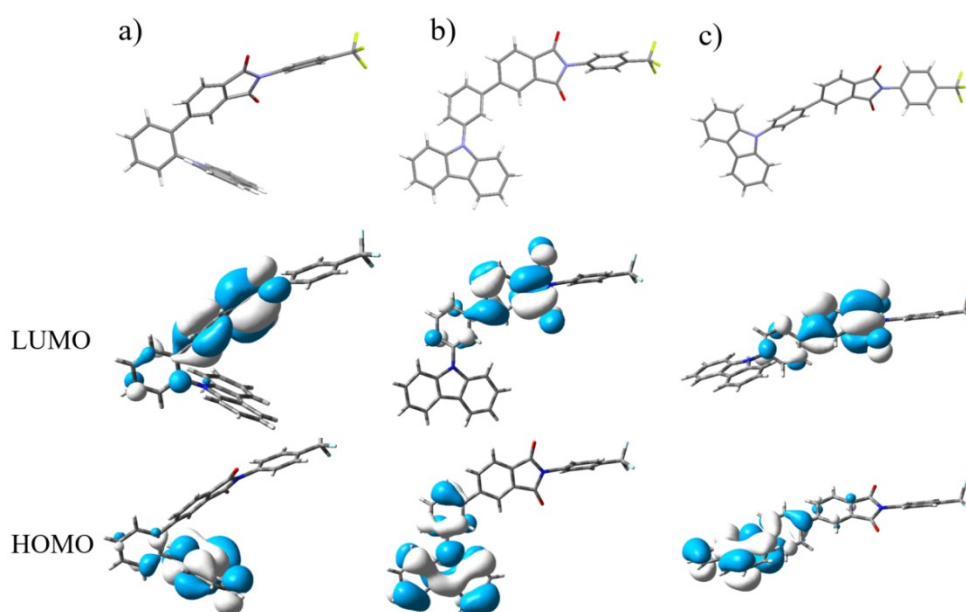
**Figure S21.** Schematic diagram of the TTA emission mechanism.



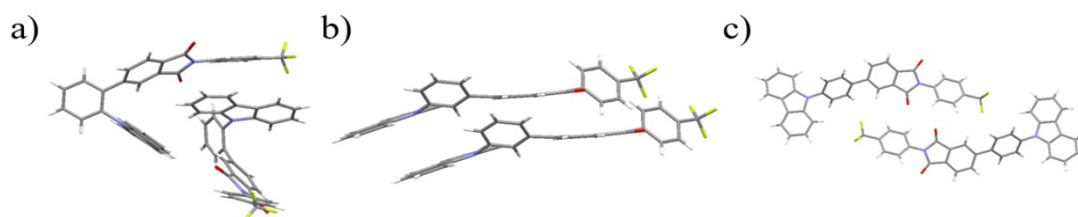
**Figure S22.** Temperature-dependent lifetimes of the excited state associated with the 500 nm emission peak in 3Fmo powder.



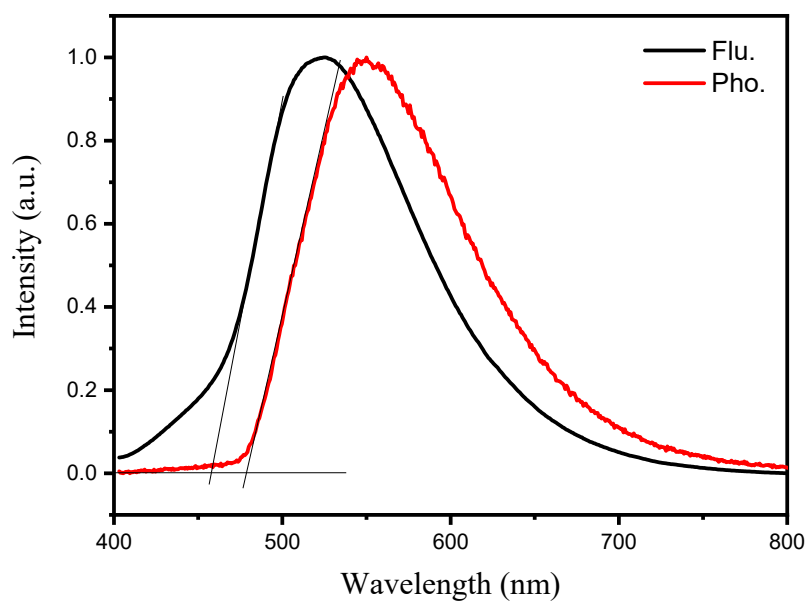
**Figure S23.** Temperature-dependent lifetimes of the excited state associated with the 470 nm emission peak in 3Fmp powder.



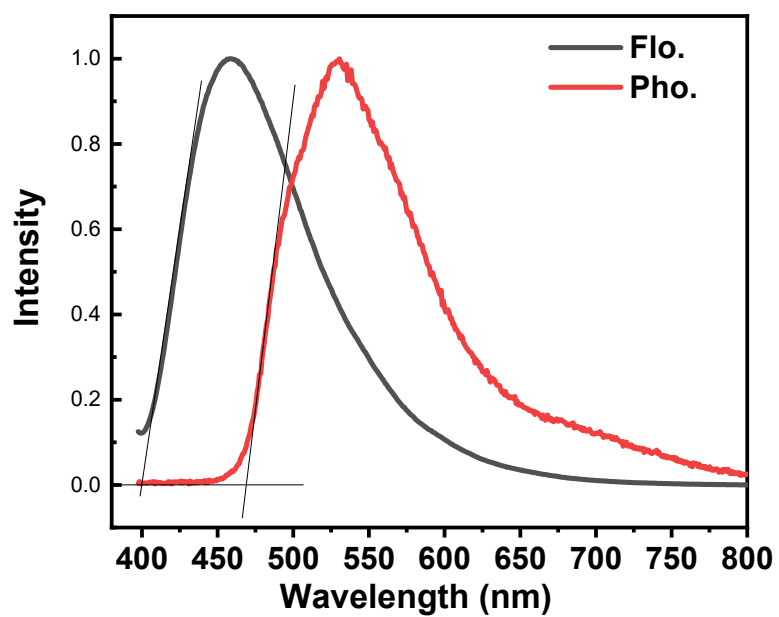
**Figure S24.** The single-crystal structures and distributions of electron density for the compounds are identified as: a) 3Fmo; b) 3Fmm; c) 3Fmp.



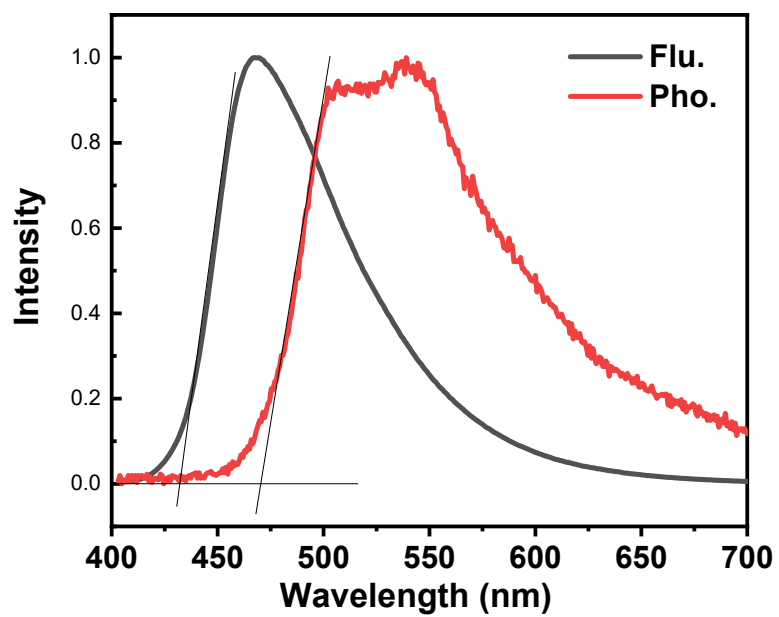
**Figure S25.** Interaction of molecules in the crystal: a) 3Fmo; b) 3Fmm; c) 3Fmp.



**Figure S26.** The fluorescence and phosphorescence spectra of 3Fmo.



**Figure S27.** The fluorescence and phosphorescence spectra of 3Fmm.



**Figure S28.** The fluorescence and phosphorescence spectra of 3Fmp.

### 3. Supporting Tables

**Table S1.** Photophysical properties of the compound powder

	<b>3Fmo</b>	<b>3Fmm</b>	<b>3Fmp</b>
$\lambda_{max}^{em}$ (nm)	494	456	468
$\lambda_{max}^{abs}$ (nm)	429	378	421
$\lambda_{max}^{ex}$ (nm)	486	385	454
$\tau$ ( $\mu$ s)	2.25	59.6	11.7

**Table S2.** Crystal data and structure refinement for 3Fmo

<b>Identification code</b>	<b>3Fmo</b>
<b>Empirical formula</b>	C <sub>33</sub> H <sub>19</sub> F <sub>3</sub> N <sub>2</sub> O <sub>2</sub>
<b>Formula weight</b>	532.50
<b>Temperature/K</b>	292.9(3)
<b>Crystal system</b>	monoclinic
<b>Space group</b>	P2 <sub>1</sub> /n
<b>a/Å</b>	14.1951(3)
<b>b/Å</b>	10.5832(2)
<b>c/Å</b>	17.5467(3)
<b><math>\alpha</math>/°</b>	90
<b><math>\beta</math>/°</b>	98.318(2)
<b><math>\gamma</math>/°</b>	90
<b>Volume/Å<sup>3</sup></b>	2608.30(9)
<b>Z</b>	4
<b><math>\rho_{calc}</math>/cm<sup>3</sup></b>	1.356
<b><math>\mu</math>/mm<sup>-1</sup></b>	0.832
<b>F(000)</b>	1096.0
<b>Crystal size/mm<sup>3</sup></b>	0.400 × 0.300 × 0.060
<b>Radiation</b>	CuK $\alpha$ ( $\lambda$ = 1.54184)
<b>2<math>\theta</math> range for data collection/°</b>	7.502 to 149.844
<b>Index ranges</b>	-14 ≤ h ≤ 17, -12 ≤ k ≤ 8, -19 ≤ l ≤ 21
<b>Reflections collected</b>	8076
<b>Independent reflections</b>	5132 [R <sub>int</sub> = 0.0251, R <sub>sigma</sub> = 0.0339]
<b>Data/restraints/parameters</b>	5132/0/361
<b>Goodness-of-fit on F<sup>2</sup></b>	1.048
<b>Final R indexes [I ≥ 2<math>\sigma</math> (I)]</b>	R <sub>1</sub> = 0.0643, wR <sub>2</sub> = 0.1798

<b>Final R indexes [all data]</b>	$R_1 = 0.0772$ , $wR_2 = 0.1967$
<b>Largest diff. peak/hole / e Å<sup>-3</sup></b>	0.58/-0.42

**Table S3.** Crystal data and structure refinement for 3Fmm.

<b>Identification code</b>	<b>3Fmm</b>
<b>Empirical formula</b>	$C_{33}H_{19}F_3N_2O_2$
<b>Formula weight</b>	532.50
<b>Temperature/K</b>	150.00(10)
<b>Crystal system</b>	monoclinic
<b>Space group</b>	$P2_1/c$
<b>a/Å</b>	26.4492(10)
<b>b/Å</b>	16.7231(6)
<b>c/Å</b>	5.3699(2)
<b><math>\alpha/^\circ</math></b>	90
<b><math>\beta/^\circ</math></b>	94.146(4)
<b><math>\gamma/^\circ</math></b>	90
<b>Volume/Å<sup>3</sup></b>	2368.96(15)
<b>Z</b>	4
<b><math>\rho_{\text{calc}}/\text{cm}^3</math></b>	1.493
<b><math>\mu/\text{mm}^{-1}</math></b>	0.916
<b>F(000)</b>	1096.0
<b>Crystal size/mm<sup>3</sup></b>	? × ? × ?
<b>Radiation</b>	CuK $\alpha$ ( $\lambda = 1.54184$ )
<b>2<math>\theta</math> range for data collection/<math>^\circ</math></b>	17.386 to 148.082
<b>Index ranges</b>	$-32 \leq h \leq 27$ , $-17 \leq k \leq 20$ , $-4 \leq l \leq 6$
<b>Reflections collected</b>	7230
<b>Independent reflections</b>	3890 [ $R_{\text{int}} = 0.0801$ , $R_{\text{sigma}} = 0.0622$ ]
<b>Data/restraints/parameters</b>	3890/0/361
<b>Goodness-of-fit on F<sup>2</sup></b>	1.070
<b>Final R indexes [<math>I \geq 2\sigma(I)</math>]</b>	$R_1 = 0.0901$ , $wR_2 = 0.2456$
<b>Final R indexes [all data]</b>	$R_1 = 0.1138$ , $wR_2 = 0.2670$
<b>Largest diff. peak/hole / e Å<sup>-3</sup></b>	0.40/-0.40

**Table S4.** Crystal data and structure refinement for 3Fmp.

<b>Identification code</b>	<b>3Fmp</b>
<b>Empirical formula</b>	$C_{33}H_{19}F_3N_2O_2$
<b>Formula weight</b>	532.50
<b>Temperature/K</b>	150.00(10)
<b>Crystal system</b>	monoclinic
<b>Space group</b>	$P2_1/c$



<b>a/Å</b>	9.9603(2)
<b>b/Å</b>	7.8995(2)
<b>c/Å</b>	32.1373(6)
<b><math>\alpha</math>/°</b>	90
<b><math>\beta</math>/°</b>	93.108(2)
<b><math>\gamma</math>/°</b>	90
<b>Volume/Å<sup>3</sup></b>	2524.89(9)
<b>Z</b>	4
<b><math>\rho_{\text{calc}}/\text{cm}^3</math></b>	1.401
<b><math>\mu/\text{mm}^{-1}</math></b>	0.860
<b>F(000)</b>	1096.0
<b>Crystal size/mm<sup>3</sup></b>	1.800 × 0.090 × 0.050
<b>Radiation</b>	CuK $\alpha$ ( $\lambda$ = 1.54184)
<b>2<math>\Theta</math> range for data collection/°</b>	8.892 to 134.224
<b>Index ranges</b>	-11 ≤ h ≤ 11, -6 ≤ k ≤ 9, -37 ≤ l ≤ 33
<b>Reflections collected</b>	7581
<b>Independent reflections</b>	4312 [ $R_{\text{int}}$ = 0.0396, $R_{\text{sigma}}$ = 0.0406]
<b>Data/restraints/parameters</b>	4312/0/361
<b>Goodness-of-fit on F<sup>2</sup></b>	1.046
<b>Final R indexes [<math>I \geq 2\sigma(I)</math>]</b>	$R_1$ = 0.0418, $wR_2$ = 0.1035
<b>Final R indexes [all data]</b>	$R_1$ = 0.0499, $wR_2$ = 0.1099

**Table S5.** Translation of electronic energy levels obtained from the single-crystal studies of three compounds.

<b>Structur</b>	<b>HOMO</b>	<b>LUMO</b>	<b><math>\Delta E_g</math></b>	<b>Dipole</b>	<b>T<sub>1</sub></b>	<b>S<sub>1</sub></b>	<b><math>\Delta E_{ST}</math></b>
<b>e</b>	<b>(eV)</b>	<b>(eV)</b>	<b>(eV)</b>		<b>(eV)</b>	<b>(eV)</b>	<b>(eV)</b>
<b>3Fmo</b>	-5.85	-2.76	3.09	7.77	2.54	2.57	0.03
<b>3Fmm</b>	-5.81	-2.65	3.16	5.93	2.70	2.78	0.08
<b>3Fmp</b>	-5.84	-2.92	2.92	6.07	2.41	2.59	0.18