

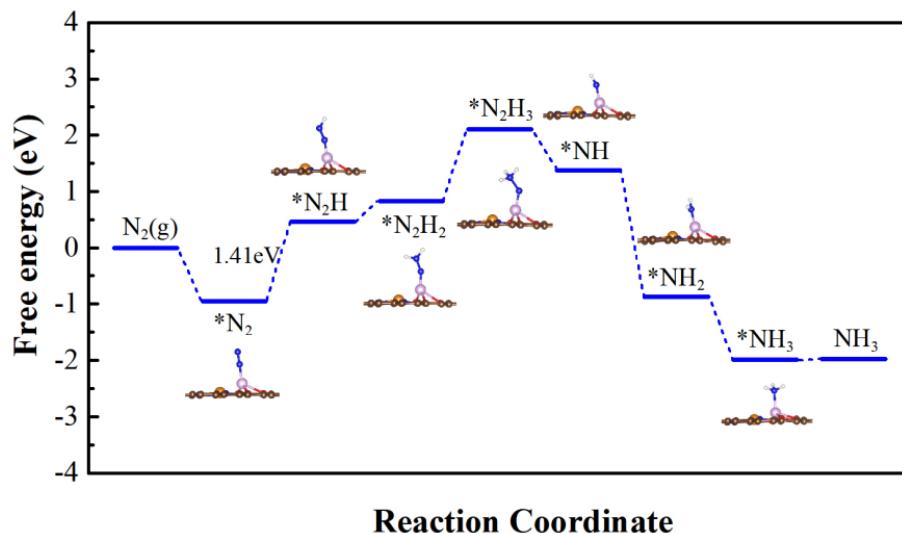
## Theoretical study of a CuCo dual-atom catalyst for nitrogen fixation

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Reaction Coordinate

Fig. S1. Free energy diagram along the reaction path of NRR proceeded on N-CuCo-O starting with  $\text{N}_2$  adsorbed on Co site.

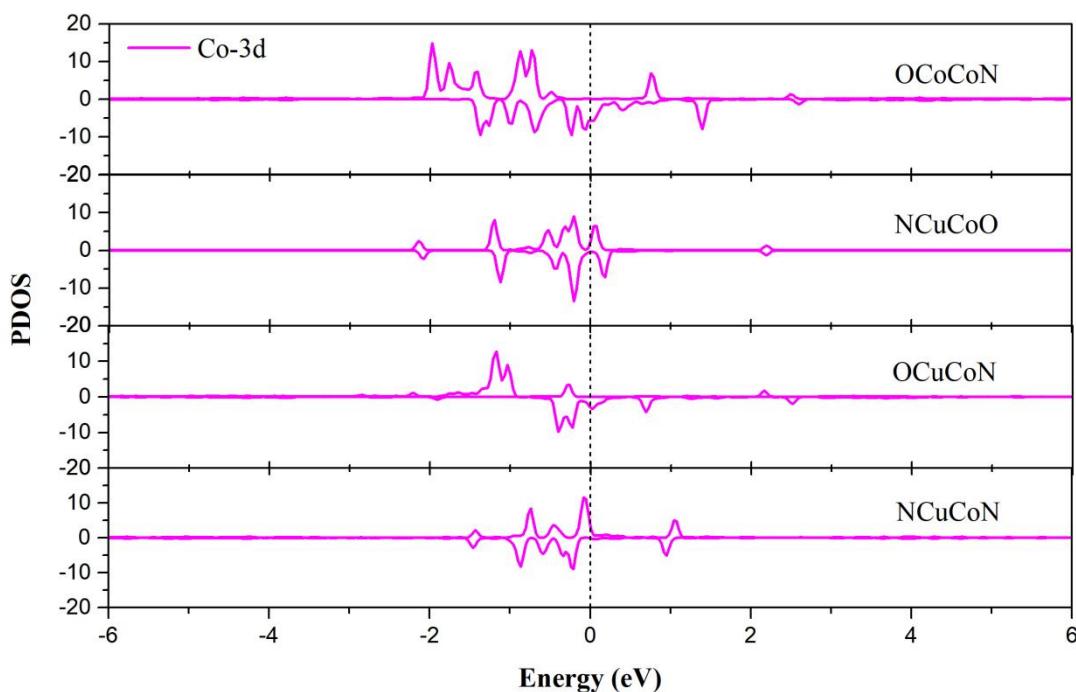


Fig. S2. The projected density of states of Co-3d in different configurations.

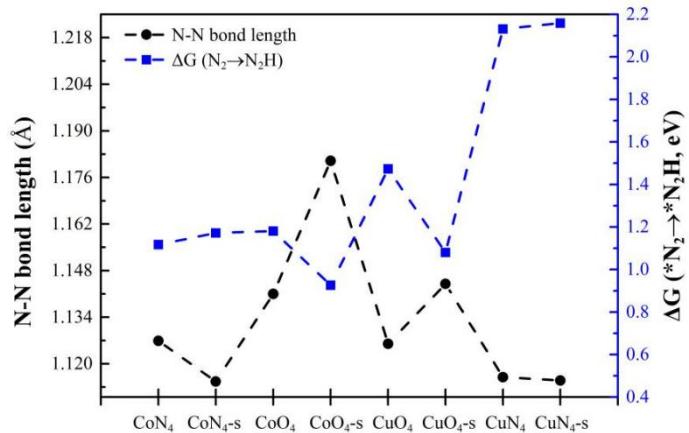


Fig. S3. The relationship between N-N bond length of adsorbed  ${}^*\text{N}_2$  and Gibbs free energy change  $\Delta G$  from  ${}^*\text{N}_2$  to  ${}^*\text{N}_2\text{H}$  on different SACs.

Table SI. calculated energy(E), zero point energy(ZPE) and contribution from entropy(TS) of O-CuCo-N DAC.

O-CuCo-N (side-on)	E	ZPE	TS
${}^*\text{N}_2$	-639.300	0.204	0.106
${}^*\text{NNH-Cu}$	-642.549	0.530	0.084
${}^*\text{NNH-Co}$	-642.289	0.530	0.084
${}^*\text{NNH}$	-646.228	0.835	0.11
${}^*\text{NNH}_2$	-645.308	0.826	0.123
${}^*\text{NH}_2\text{NH}$	-649.771	1.149	0.160
${}^*\text{NH}_2\text{NH}_2$	-654.360	1.319	0.179
${}^*\text{NH}_2\text{NH}_3$	-659.042	1.700	0.228
${}^*\text{NH}_3$	-642.495	1.04	0.10
${}^*\text{NH}_3\text{NH}_3$	-662.789	2.074	0.198
O-CuCo-N (end-on)	E	ZPE	TS
${}^*\text{Co-N}_2$	-639.306	0.196	0.160
${}^*\text{N}_2\text{H}$	-642.318	0.506	0.120
${}^*\text{N}_2\text{H}_2$	-646.008	0.820	0.139
${}^*\text{N}_2\text{H}_3$	-648.798	1.159	0.154
${}^*\text{NH}$	-634.440	0.341	0.089
${}^*\text{NH}_2$	-639.233	0.713	0.076

*NH <sub>3</sub>	-642.492	1.018	0.127
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Table SII calculated energy(E), zero point energy(ZPE) and contribution from entropy(TS) of O-CoCo-O DAC.

O-CoCo-O (side-on)	Energy	ZPE	TS
*N <sub>2</sub>	-631.469	0.203	0.106
*N <sub>2</sub> H	-634.771	0.513	0.098
*NHNH	-638.346	0.889	0.346
*NH <sub>2</sub> N	-637.697	0.829	0.121
*NH <sub>2</sub> NH	-641.992	1.134	0.167
*NH <sub>2</sub> NH <sub>2</sub>	-646.944	1.309	0.261
*NH <sub>3</sub> NH	-646.468	1.280	0.216
*NH <sub>3</sub> NH <sub>2</sub>	-651.269	1.683	0.229
*NH <sub>3</sub> NH <sub>3</sub>	-654.432	2.032	0.323