

## **Pressure induced emission enhancement (PIEE) in solid-state 2,3,4,5-Tetraphenylthiophene: a QM/MM study**

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Table S 1 Geometry parameters of S<sub>0</sub> states for TPT in solid phase. Dihedral angles ( $\theta_1$ ,  $\theta_2$ ,  $\theta_3$ , and  $\theta_4$ ) are shown in Fig. 1

<b>Pressure/GP a</b>	<b><math>\theta_1</math></b>	<b><math>\theta_2</math></b>	<b><math>\theta_3</math></b>	<b><math>\theta_4</math></b>
<b>0</b>	146.07	113.19	66.84	38.21
<b>0.93</b>	146.56	114.24	66.71	37.72
<b>2.12</b>	146.91	115.29	66.07	36.23
<b>3.41</b>	146.81	116.29	66.49	35.14
<b>5.06</b>	146.71	117.93	66.56	32.23
<b>6.63</b>	146.73	119.50	66.74	30.26
<b>7.46</b>	145.89	119.65	67.16	28.87
<b>8.5</b>	146.81	121.30	67.17	28.04

Table S 2 Geometry parameters of S<sub>1</sub> states for TPT in solid phase. Dihedral angles ( $\theta_1$ ,  $\theta_2$ ,  $\theta_3$ , and  $\theta_4$ ) are shown in Fig. 1.

<b>Pressure/GPa</b>	<b><math>\theta_1</math></b>	<b><math>\theta_2</math></b>	<b><math>\theta_3</math></b>	<b><math>\theta_4</math></b>
<b>0</b>	158.93	122.18	58.30	24.08
<b>0.93</b>	158.44	123.06	58.53	24.76
<b>2.12</b>	158.53	123.81	58.05	23.82
<b>3.41</b>	158.62	124.64	58.55	22.84
<b>5.06</b>	158.97	125.61	58.79	20.39
<b>6.63</b>	159.56	126.88	58.89	18.49
<b>7.46</b>	159.72	126.79	59.08	17.19
<b>8.5</b>	161.72	128.37	58.85	16.07

Table S 3 Energy levels of HOMO, LUMO, the corresponding energy gaps ( $\Delta E_{gap}$ ) and  $S_1$  energy level of TPT in solid state at different pressures.

Pressure/GPa	HOMO/eV	LUMO/eV	$\Delta E_{gap}$ /eV	$S_1$ /eV
<b>0</b>	-5.41289	-1.23458	4.17831	3.0649
<b>0.93</b>	-5.38241	-1.23023	4.15218	3.0572
<b>2.12</b>	-5.36282	-1.22995	4.13286	3.0516
<b>3.41</b>	-5.34214	-1.22669	4.11545	3.0456
<b>5.06</b>	-5.29316	-1.22832	4.06484	2.9962
<b>6.63</b>	-5.26622	-1.23376	4.03245	2.9667
<b>7.46</b>	-5.22976	-1.22125	4.00851	2.9308
<b>8.50</b>	-5.21316	-1.22533	3.98783	2.8977

Table S 4 The calculated  $k_r$ ,  $k_{ic}$  and  $\Phi_F$  at different pressure.

Pressure/GP	<b>0</b>	<b>0.93</b>	<b>2.12</b>	<b>3.41</b>	<b>5.06</b>	<b>6.63</b>	<b>7.46</b>	<b>8.50</b>
<b>a</b>								
$k_r(10^8 \text{ s}^{-1})$	2.384	2.348	2.350	2.386	2.260	2.169	2.111	2.100
$k_{ic}(10^8 \text{ s}^{-1})$	5.803	2.242	2.060	2.488	2.979	3.208	5.224	7.884
$\Phi_F(\%)$	29.11	51.16	53.29	48.96	43.14	40.34	28.78	21.64

Table S 5 Adiabatic excitation energy ( $\nu_{fi}$ ) and the oscillator strength ( $f$ ) at different pressure.

Pressure/GP	<b>0</b>	<b>0.93</b>	<b>2.12</b>	<b>3.41</b>	<b>5.06</b>	<b>6.63</b>	<b>7.46</b>	<b>8.50</b>
<b>a</b>								
$\nu_{fi}/\text{a.u.}$	0.1260	0.1262	0.1262	0.1261	0.1252	0.1243	0.1237	0.1234
$f$	0.5887	0.5882	0.5815	0.5754	0.5489	0.5332	0.5100	0.5025

Table S 6 Contributions of low-frequency modes ( $\lambda_{LF}$ ) and high frequency modes ( $\lambda_{HF}$ ) to the total  $\lambda_e$  for TPT in solid state at different pressures.

Pressure/GPa	<b>0</b>	<b>0.93</b>	<b>2.12</b>	<b>3.41</b>	<b>5.06</b>	<b>6.63</b>	<b>7.46</b>	<b>8.50</b>
$\lambda_{LF}/\lambda_e$	0.50	0.51	0.52	0.54	0.60	0.63	0.69	0.72
$\lambda_{HF}/\lambda_e$	0.50	0.49	0.48	0.47	0.48	0.48	0.46	0.44

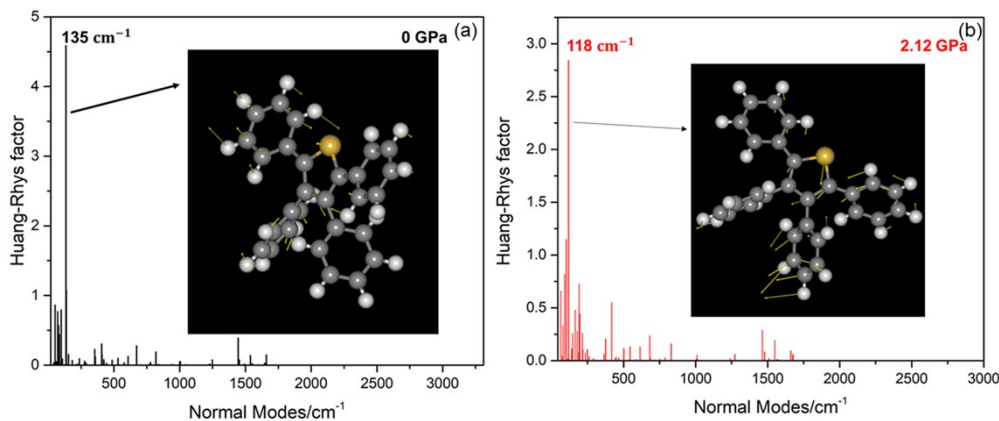


Fig. S 1 Calculated HR factors versus the normal mode frequencies at 0 GPa (a) and 2.12 GPa(b). Representative vibration modes are shown in the insets.

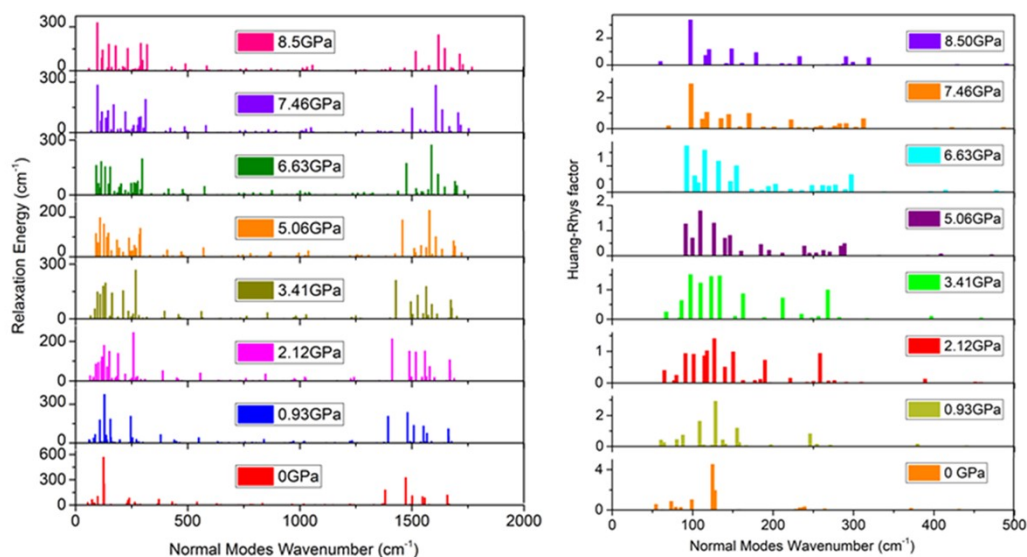


Fig. S 2 Calculated Huang-Rhys factor  $HR_k$  (right) and reorganization energies  $\lambda_{j,g}$  (left) versus the normal mode frequencies for TPT molecules in solid state at different pressure.

Table S7 Reorganization energy  $\lambda_e$  contributions (meV) from the bond lengths, bond angles, and dihedral angles for TPT in the solid phase at different pressure.

Pressure/GPa	Bond length	Bond angle	Dihedral angle	Total
0	235.03	7.56	150.89	384.48
0.93	220.92	2.81	150.18	373.91
2.12	215.26	2.55	156.00	373.81
3.41	211.07	7.70	162.34	381.11
5.06	203.50	18.10	157.04	378.64
6.63	199.60	23.56	159.99	383.15
7.46	195.63	36.11	176.47	408.21
8.50	194.53	50.30	205.92	450.75