

Supporting information

Enabling flexible Zn/MnO₂ secondary batteries by fumed silica-doped hydrogel electrolyte

Wenlong Xiong^{a,c}, Qiyuan Xie^a, Haoran Zhang^a, Chenjie Zhu^{b,*}, Md Asraful Alam^a,

Lele Wang^a, Jingliang Xu^{a,c,*}

^a State Key Laboratory of Biobased Transportation Fuel Technology, School of Chemical Engineering, Zhengzhou University, Zhengzhou 450001, PR China

^b College of Biotechnology and Pharmaceutical Engineering, Nanjing Tech University, Nanjing 211816, PR China

^c Henan Center for Outstanding Overseas Scientists

*Corresponding authors: State Key Laboratory of Biobased Transportation Fuel Technology, School of Chemical Engineering, Zhengzhou University, Zhengzhou 450001, PR China.

E-mail address: zhucj@njtech.edu.cn (Chenjie Zhu); xujl@zzu.edu.cn (Jingliang Xu)

Phone: +8613083608578; +8613580316368

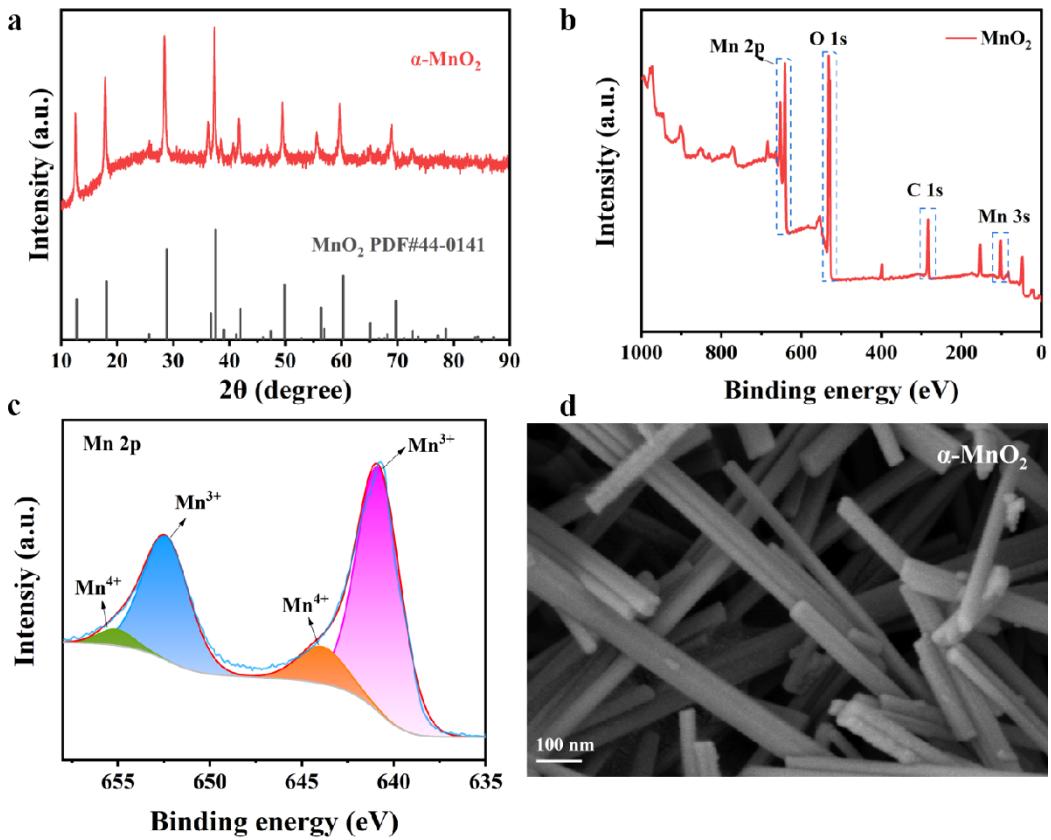


Fig. S1 (a) XRD pattern of α -MnO₂. (b) XPS spectrum of α -MnO₂. (c) Mn 2p XPS spectrum. (d) SEM image of α -MnO₂.

Fig. S1a shows the XRD pattern of the prepared α -MnO₂. The diffraction peaks appeared at 13, 18, 29, 37, 42, 50, 56, 60, and 69° are corresponding to the (110), (200), (310), (211), (301), (411), (600), (521), and (541) crystal planes in the α -MnO₂ standard card (PDF# 44-0141). As shown in Fig. S1b, Mn, O, and C coexist on the surface of the prepared α -MnO₂. For Mn element, peaks near 641.1 and 652.6 eV are attributed to Mn³⁺, while peaks near 643.8 and 655.2 eV can be attributed to Mn⁴⁺ (Fig. S1c).^{1,2} The SEM image shows that the morphology of the prepared MnO₂ is mostly nanowires and a few are nanorods (Fig. S1d), which is also consistent with the structural characteristics of the reported tunnel type α -MnO₂.³⁻⁵ The α -MnO₂ was synthesized according to the XRD, XPS, and SEM analyses.

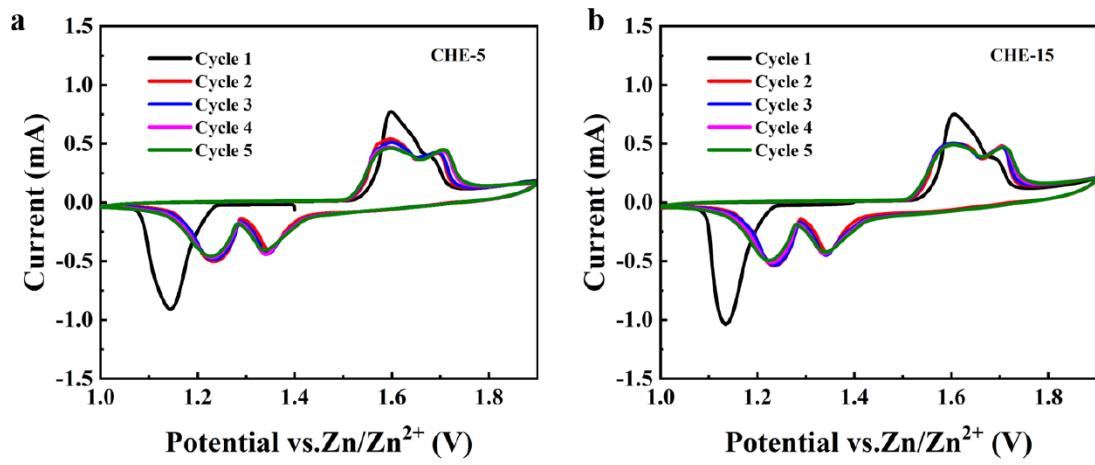


Fig. S2 Cyclic voltammograms of batteries using (a) CHE-5 and (b) CHE-15 from the 1st to 5th scan respectively

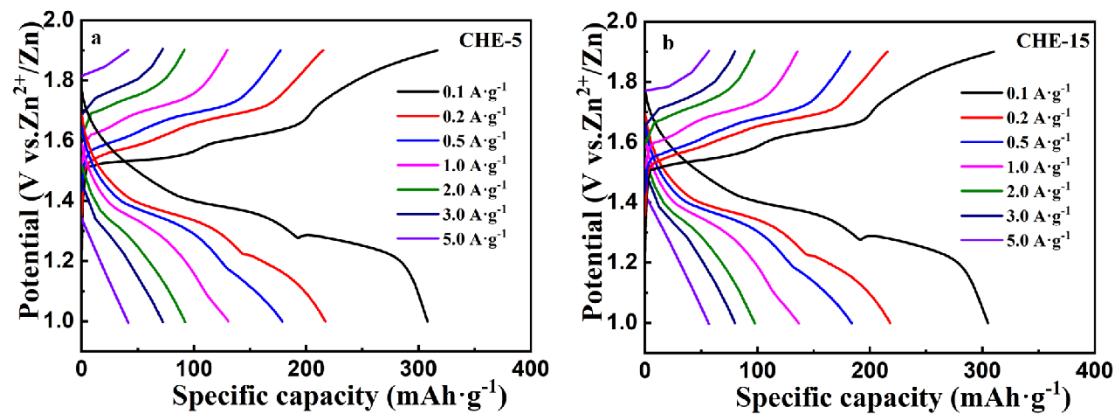


Fig. S3 Charge/discharge profiles of the batteries using (a) CHE-5 and (b) CHE-15

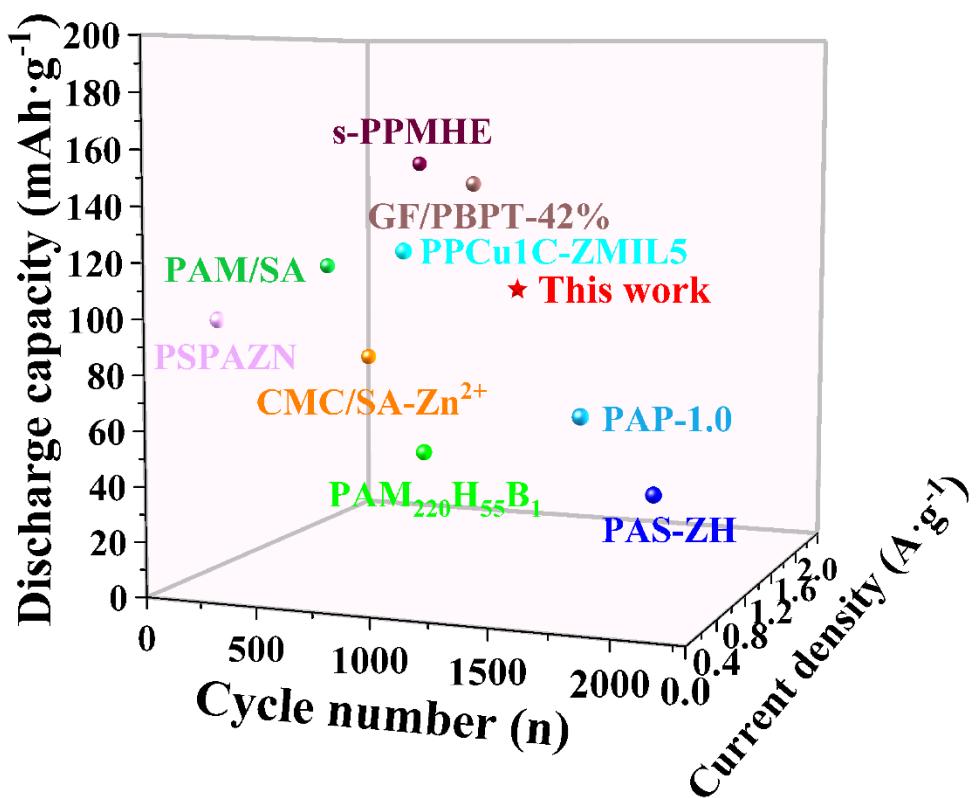


Fig. S4 Comparison of the cycle performance of Zn//MnO₂ battery in this work with recently reported hydrogel electrolyte-based Zn//MnO₂ batteries

Table S1 Comparison of mechanical properties of hydrogel electrolytes

Hydrogel Electrolyte	Tensile stress (MPa)	Tensile strain (%)	References
CSC	0.0542	160	6
Gel/SA-acetate	0.3	500	7
PZD-20%	0.082	690	8
HE-RS	0.02	273	9
PNMA-SA	0.838	200	10
PZDHE	0.796	450	11
PAZPM	0.1	320	12
CHE-10	2.55	465	This work

Table S2 Comparison of the cycle performance of Zn//MnO₂ battery in this work with recently reported hydrogel electrolyte-based Zn//MnO₂ batteries

Hydrogel Electrolyte	Current density (A·g ⁻¹)	Discharge capacity (mAh·g ⁻¹)	Cycle number (n)	References
GF/PBPT-42%	1	145	1000	13
PAM/SA	1	112	300	14
PAM ₂₂₀ H ₅₅ B ₁	0.5	50	1000	15
s-PPMHE	1.525	150	500	16
PAS-ZH	0.5	40	2000	17
PAP-1.0	0.2	71	1800	18
CMC/SA-Zn ²⁺	1	77	500	19
PPCu1C-ZMIL5	0.308	124.6	1000	20
PSPAZn	0.2	98.6	220	21
CHE-10	1.5	100.5	1000	This work

References

1. M. Karbak, M. Baazizi, S. Sayah, C. Lambert, Y. Tison, H. Martinez, T. Chafik and F. Ghamouss, *J. Mater. Chem. A*, 2023, 11, 2634–2640.
2. Y. Liu, K. Xiang, W. Zhou, W. Deng, H. Zhu and H. Chen, *Small*, 2023, 20, 2308741.
3. X. Xiao, L. Zhang, W. Xin, M. Yang, Y. Geng, M. Niu, H. Zhang and Z. Zhu, *Small*, 2024, 20, 2309271.
4. X. Tian, X. Cheng, S. Liao, J. Chen, P. Lv and Q. Wei, *ACS Appl. Mater. Interfaces*, 2023, 15, 52415–52426.
5. Y. Li, Y. Li, Q. Liu, Y. Liu, T. Wang, M. Cui, Y. Ding, H. Li and G. Yu, *Angew. Chem.-Int. Edit.*, 2024, 63, e202318444.
6. Y. Hu, Z. Wang, Y. Li, P. Liu, X. Liu, G. Liang, D. Zhang, X. Fan, Z. Lu and W. Wang, *Chem. Eng. J.*, 2024, 479, 147762.
7. C. Tian, J. Wang, R. Sun, T. Ali, H. Wang, B. Xie, Y. Zhong and Y. Hu, *Angew. Chem.-Int. Edit.*, 2023, 62, e202310970.
8. J. Zhang, C. Lin, L. Zeng, H. Lin, L. He, F. Xiao, L. Luo, P. Xiong, X. Yang, Q. Chen and Q. Qian, *Small*, 2024, 20, 2312116.
9. H. Tian, M. Yao, Y. Guo, Z. Wang, D. Xu, W. Pan and Q. Zhang, *Adv. Energy Mater.*, 2024, 2403683.
10. W. Zeng, S. Zhang, J. Lan, Y. Lv, G. Zhu, H. Huang, W. Lv and Y. Zhu, *ACS Nano*, 2024, 18, 26391–26400.

11. M. Shi, J. Zhang, G. Tang, B. Wang, S. Wang, X. Ren, G. Li, W. Chen, C. Liu and C. Shen, *Nano Res.*, 2024, 17, 5278–5287.
12. K. Zhu, X. Niu, W. Xie, H. Yang, W. Jiang, M. Ma and W. Yang, *Energy Environ. Sci.*, 2024, 17, 4126–4136.
13. X. Liu, W. Wei, Y. Yang, Y. Li, Y. Li, S. Xu, Y. Dong and R. He, *Chem. Eng. J.*, 2022, 437, 135409.
14. Z. Xiang, Y. Li, X. Cheng, C. Yang, K. Wang, Q. Zhang and L. Wang, *Chem. Eng. J.*, 2024, 490, 151524.
15. X. Wan, Q. Xie, H. Song, C. Li and J. Wang, *J. Membr. Sci.*, 2022, 655, 120606.
16. M. Sun, G. Ji and J. Zheng, *Chem. Eng. J.*, 2023, 463, 142535.
17. G. Liu, Y. Peng, M. Yu, L. Zhao, Y. Fu, J. Zhang, H. Zeng, R. Liu, S. Niu, Y. Meng and F. Ran, *J. Power Sources*, 2024, 621, 235280.
18. G. Ji, M. Sun, M. Li, R. Hu and J. Zheng, *J. Power Sources*, 2024, 612, 234823.
19. Y. Yang, L. Shi, Y. Wu, Z. Chen, X. Zhu, L. Du and Y. Wang, *Chem. Eng. J.*, 2024, 484, 149780.
20. J. Hu, Y. Qu, F. Shi, J. Wang, X. He, S. Liao and L. Duan, *Adv. Funct. Mater.*, 2022, 32, 2209463.
21. F. Yang, H. Hua, P. Lai, P. Lin, J. Yang, M. Zhang, Y. Yang and J. Zhao, *ACS Appl. Energ. Mater.*, 2022, 5, 10872–10882.