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Supplementary Materials

Physicochemical properties and mechanism of action of the new copper(II) pyrazine-based complex with high anticancer activity and selectivity towards cancer cells

Table S1. SC-XRD experimental details for **Cu(L)** compound.

Chemical formula	$2(\text{C}_{14}\text{H}_{15}\text{Cl}_2\text{CuN}_5\text{S}_2)$
M_r	903.74
Crystal system, space group	Orthorhombic, $Pna2_1$
Temperature (K)	101
a, b, c (Å)	14.7703 (5), 9.2488 (3), 26.0606 (8)
V (Å ³)	3560.06 (19)
Z	4
Radiation type	Mo $K\alpha$
μ (mm ⁻¹)	1.77
Crystal size (mm)	$0.59 \times 0.22 \times 0.02$
No. of measured, independent and observed [$I > 2\sigma(I)$] reflections	28753, 9961, 8915
R_{int}	0.057
$(\sin \theta/\lambda)_{\text{max}}$ (Å ⁻¹)	0.748
$R[F^2 > 2\sigma(F^2)], wR(F^2), S$	0.048, 0.103, 1.04
No. of reflections	9961
No. of parameters	436
No. of restraints	1
$\Delta_{\text{max}}, \Delta_{\text{min}}$ (e Å ⁻³)	0.87, -0.91
Absolute structure	Refined as an inversion twin
Absolute structure parameter	0.504 (16)

Table S2. The sequences of primers used in the gene expression evaluation.

Gene symbol	Forward Sequence (5'→3')	Reverse Sequence (5'→3')
<i>CAT</i>	CCGGGACTACACCCAGATGA	TCTTGGCGTTCTCCTGATGC
<i>GPX</i>	CTTCAGGGTGGTATGGCTGT	TGGCCAGACCTTAATGTTCC
<i>GSR</i>	TCAGCTCACCACAACCTCTG	GAGACCAGCCTGACCAACAT
<i>SOD2</i>	CTTCAGGGTGGTATGGCTGT	TGGCCAGACCTTAATGTTCC
<i>NF2L2</i>	AGCTTAGCGTTCATCCGTGT	TCCAATCATCCGTCAAAACA
<i>ATM</i>	GCCGCGTTGATACTACTTTG	GCAGCAGGGTGACAATAAACA
<i>ATR</i>	AATGGTTGGAGAATGCTGGC	ACATCACCCTTGGACCAGAG
<i>OGG1</i>	CCTGTGGGGACCTTATGCTG	TGTGAATCCCCTCTCCCGAT
<i>PARP</i>	CCCCACGACTTTGGGATGAA	AGACTGTAGGCCACCTCGAT
<i>RNA18SN5</i>	GAAACTGCGAATGGCTCATTA	CACAGTTATCCAAGTGGGAGAGG

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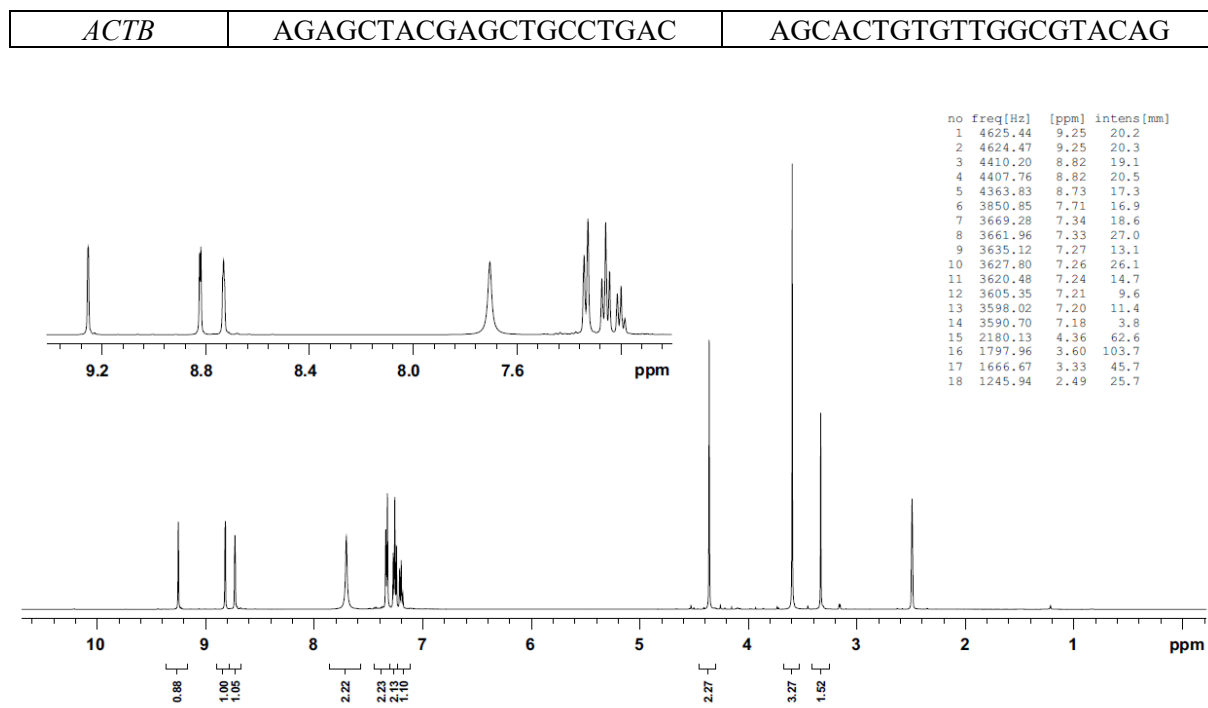


Figure S1. ¹H NMR spectrum of the L ligand.

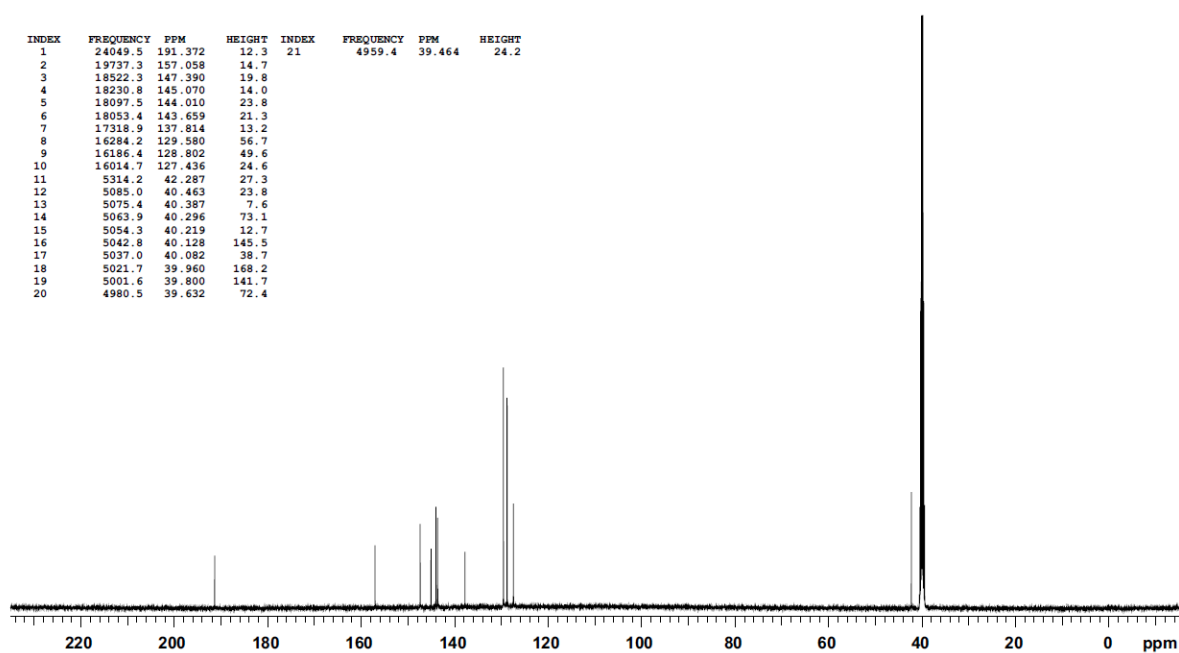


Figure S2. ¹³C NMR spectrum of the L ligand.

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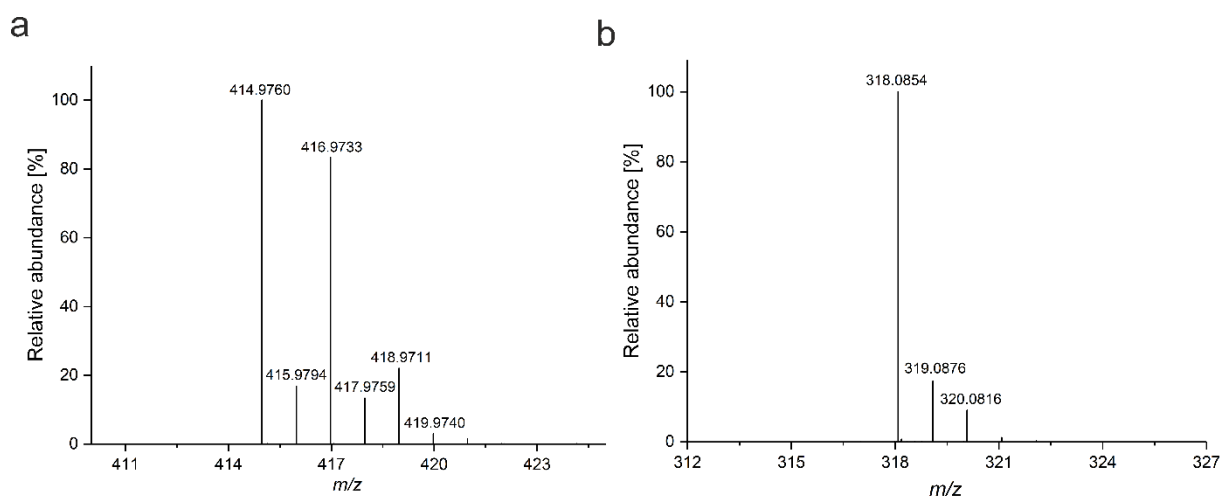


Figure S3. HR-ESI mass spectra obtained for the **Cu(L)** complex (a) and the ligand **L** (b). The experimental m/z value at 414.9760 (a) corresponds to $[M-Cl]^+$ (calculated m/z for $C_{14}H_{15}N_5S_2CuCl$: 414.9753). The experimental m/z value at 318.0854 (b) corresponds to $[M+H]^+$ (calculated m/z for $C_{14}H_{16}N_5S_2$: 318.0847).

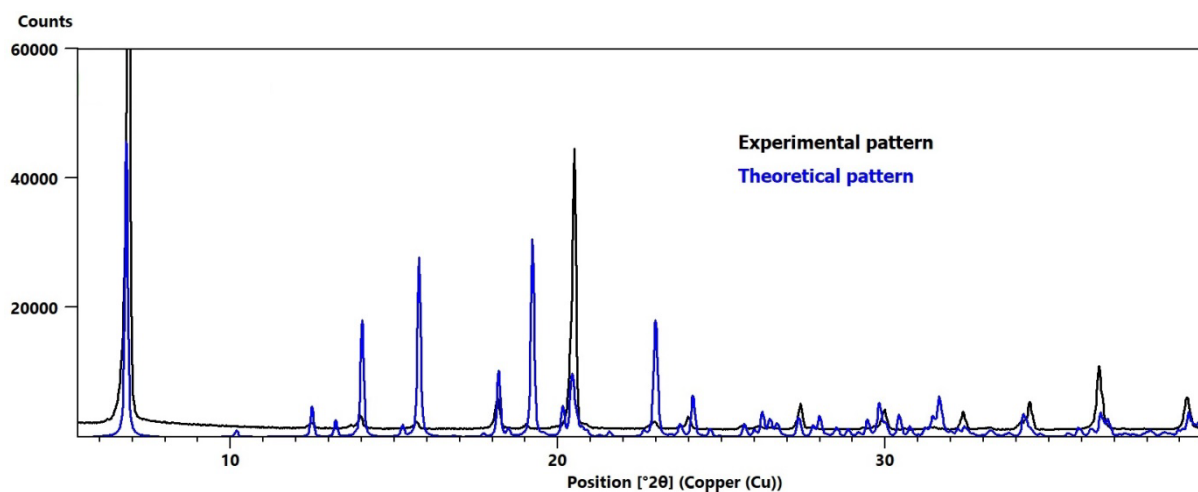


Figure S4. Theoretical and experimental PXRD patterns of the **Cu(L)** compound.

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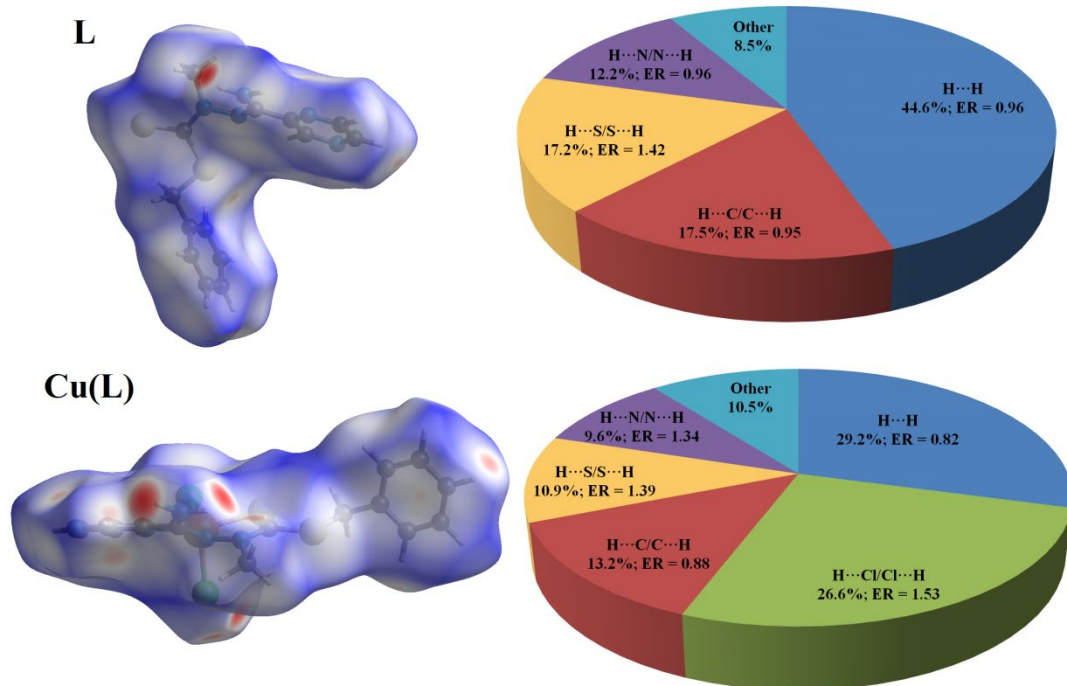


Figure S5. Hirschfeld Surfaces (left) mapped over d_{norm} and pie charts representing intermolecular contacts (right) predominant in the studied compounds, **L** and **Cu(L)** with their percentage involvement and Enrichment Ratios.

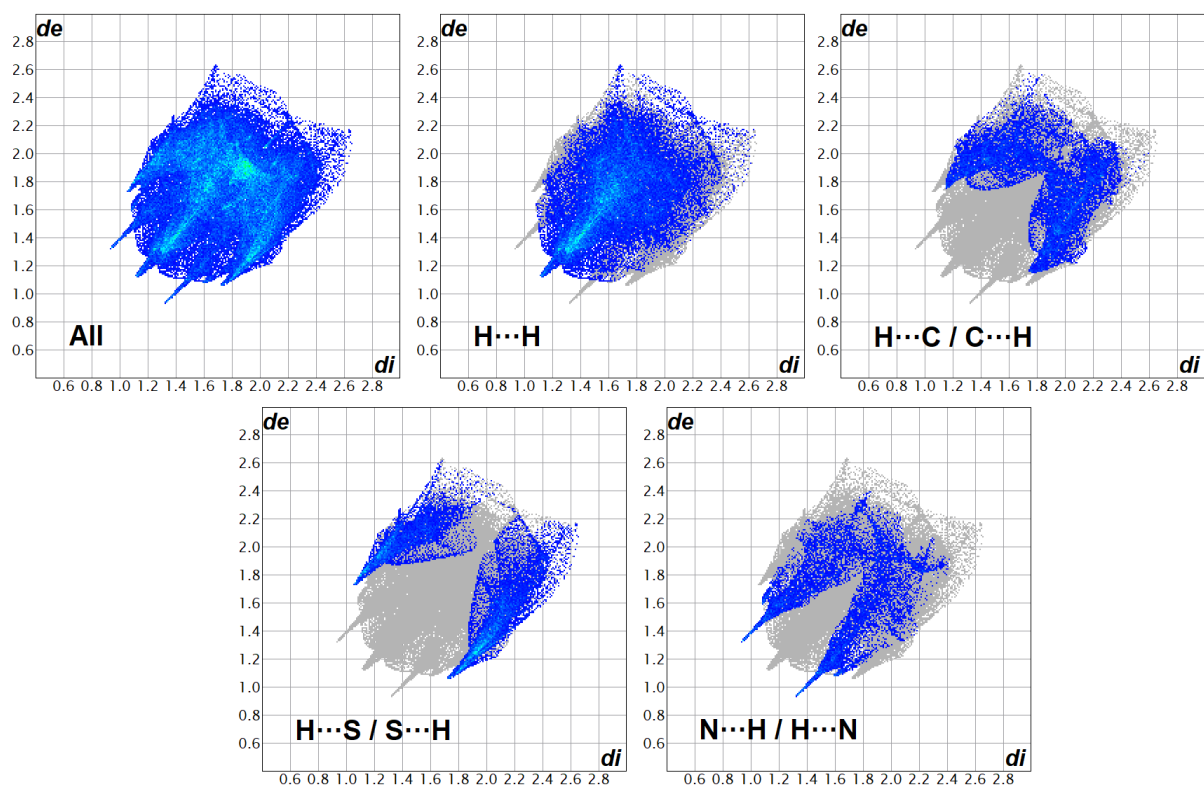


Figure S6. 2D Fingerprint Plots and the predominant contacts in **L** compound. d_e and d_i are the distances to the nearest atomic exterior and interior to the surface.

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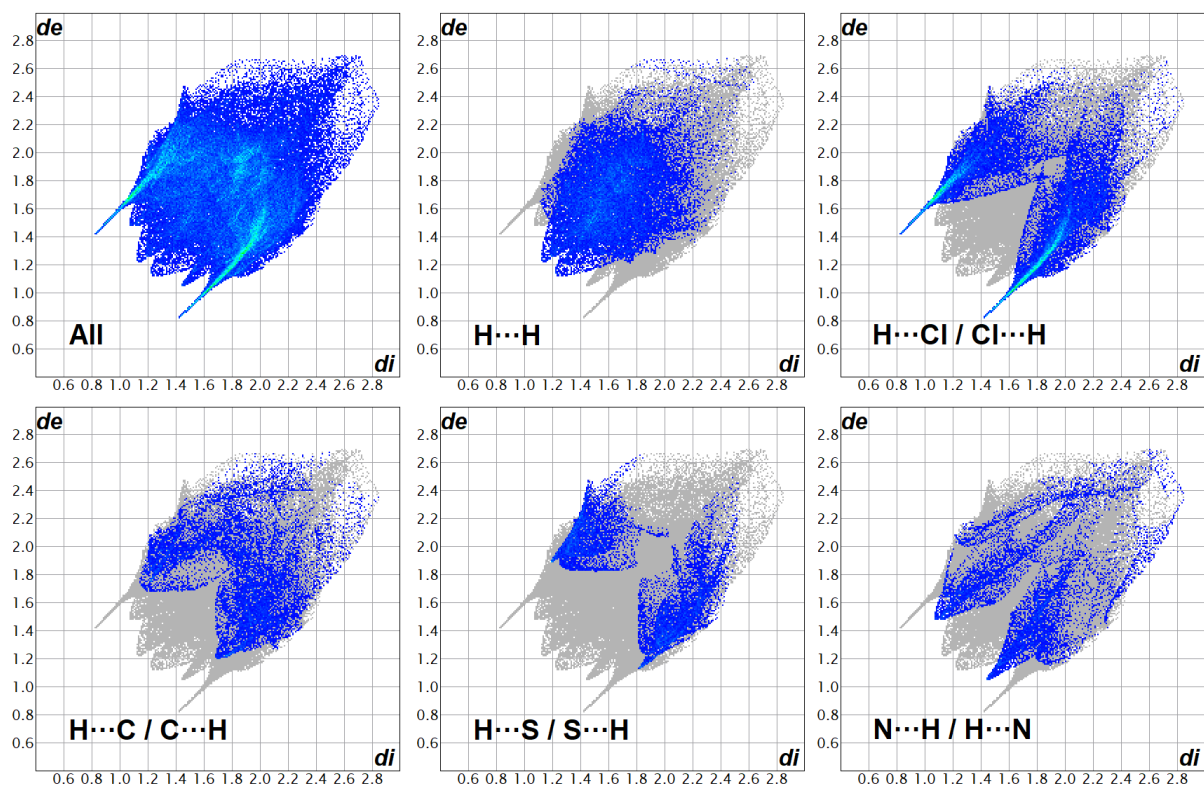


Figure S7. 2D Fingerprint Plots and the predominant contacts in Cu(L) compound. de and di are the distances to the nearest atomic exterior and interior to the surface.

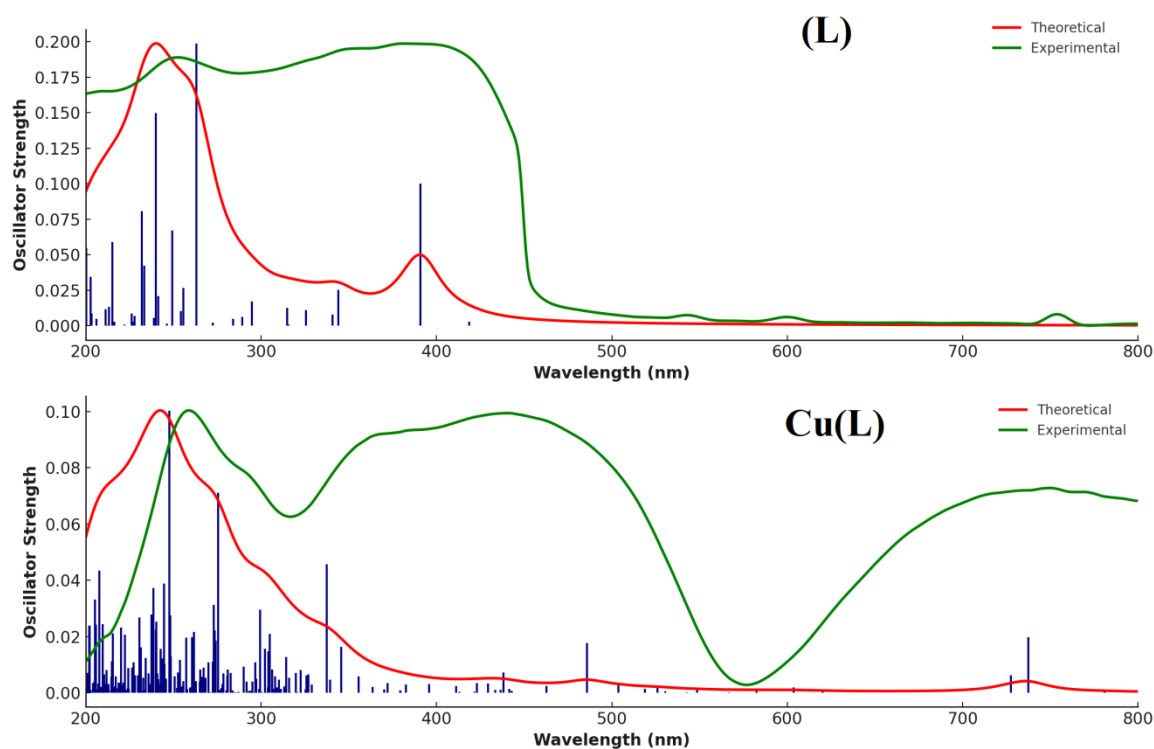


Figure S8. Experimental (green) and calculated (red) UV-Vis spectra of the studied compounds: L and Cu(L) . Oscillator strengths are represented as vertical navy-blue lines.

Table S3. The most important electronic transitions appearing in the studied compounds and the fluorescence emission maximum of L and Cu(L) .

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Absorption spectroscopy					
X-ray determined coordinates					
Experimental			Calculated		
L (nm)	Cu(L) (nm)	L (nm)	Cu(L) (nm)	f	Orbital transition
253	258	239.9		0.150	$\pi \rightarrow \pi^*$
			247.8	0.100	$n/\pi \rightarrow \pi^*$
		263.1		0.199	$n/\pi \rightarrow \pi^*$
			275.4	0.071	$n/\pi/\sigma/d(\text{Cu}) \rightarrow \pi^*/d(\text{Cu})$
		370	337.5	0.046	$n/\sigma/d(\text{Cu}) \rightarrow \pi^*/d(\text{Cu})$
382		390.7		0.100	$n/\pi \rightarrow \pi^*$
	442		485.7	0.018	$n/\sigma/d(\text{Cu}) \rightarrow \pi^*/d(\text{Cu})$
	749		737.4	0.020	$n/d(\text{Cu}) \rightarrow \pi^*/d(\text{Cu})$
DMSO solution					
Experimental			Calculated		
	Cu(L) (nm)		Cu(L) (nm)	f	Orbital transition
216			218.1	0.097	$n/d(\text{Cu}) \rightarrow \sigma^*$
			243.9	0.081	$\pi/d(\text{Cu}) \rightarrow \pi^*$
278			322.6	0.029	$n/d(\text{Cu}) \rightarrow \pi^*$
340			350.1	0.035	$\pi \rightarrow d(\text{Cu})/\pi^*$
540			468.4	0.024	$n/d(\text{Cu}) \rightarrow \pi^*$
701			605.4	0.012	$n/d(\text{Cu}) \rightarrow \pi^*$
Fluorescence					
Compound	ExW	EmW	Intensity		
L	470	481	36		
Cu(L)	-	-	-		

Used abbreviations: f - oscillator strength, d(Cu) — d orbital of a copper cation, σ — σ orbital of the organic ligand, n — non-bonding orbital, π — π orbital of the organic ligand, * — antibonding orbital.

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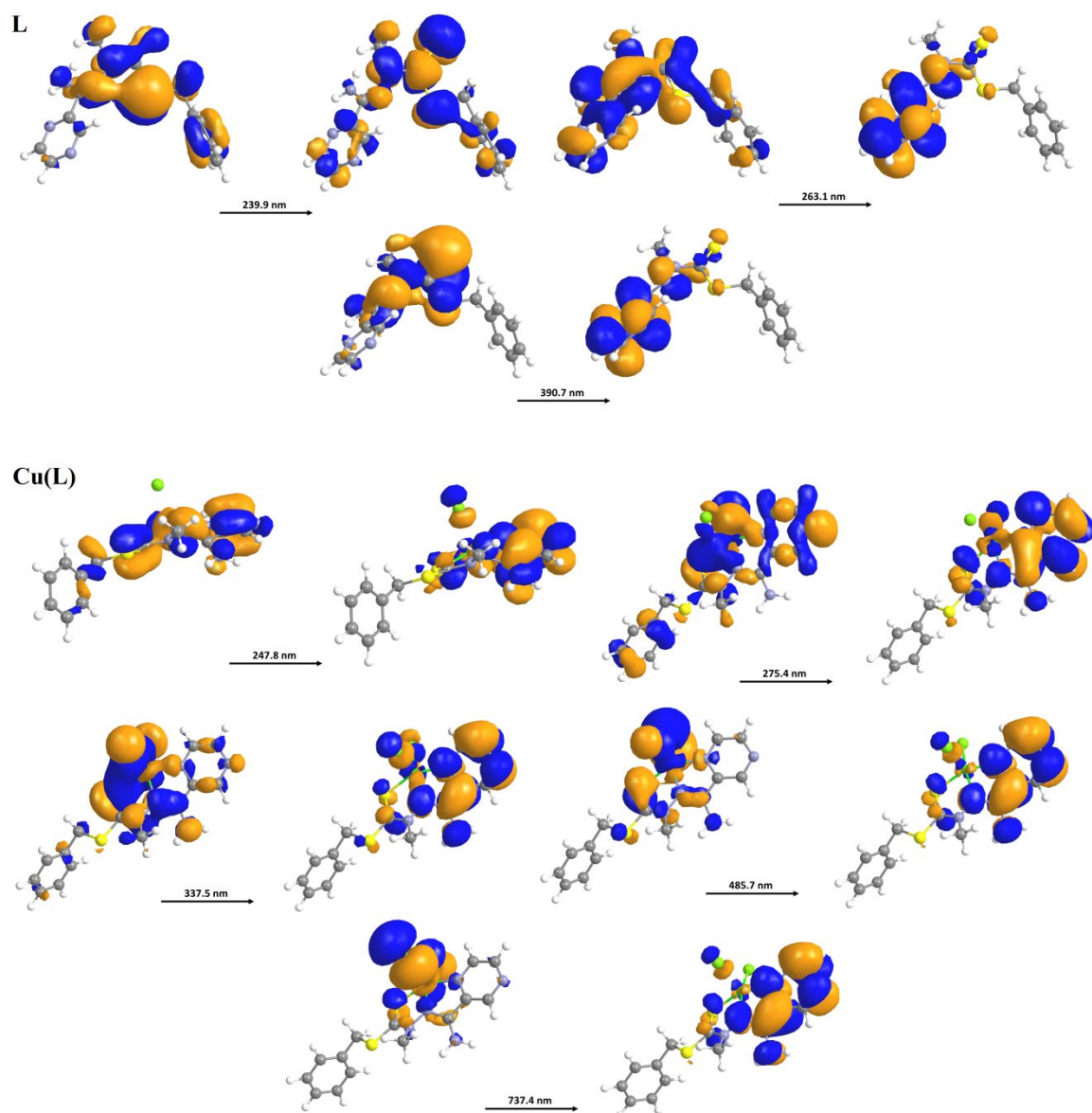


Figure S9. The main molecular orbital transitions in the studied compounds, **L** and **Cu(L)**, which are associated with the solid state absorption spectra of these compounds.

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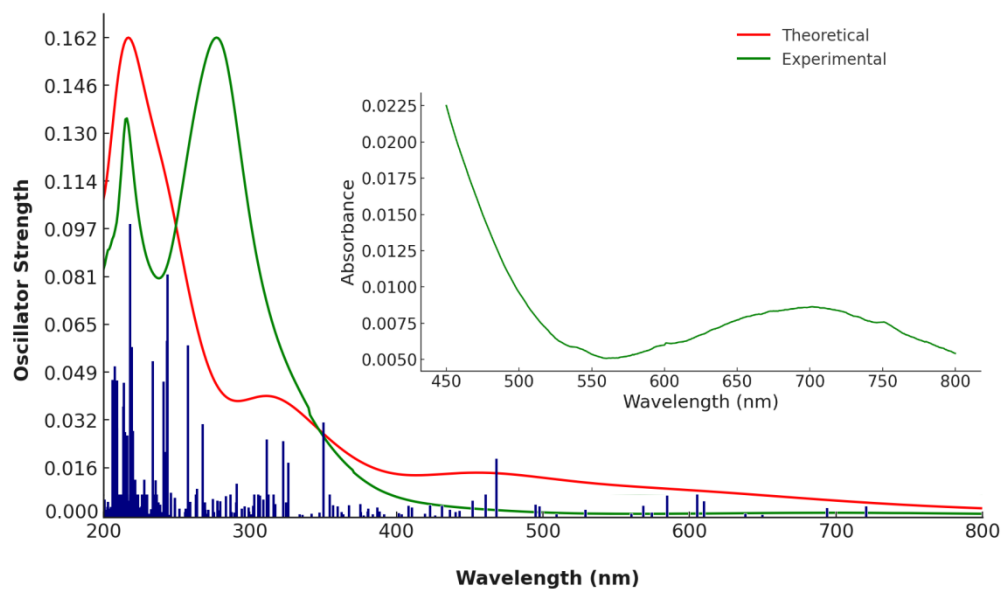


Figure S10. Experimental (green) and calculated (red) UV-Vis spectra of **Cu(L)** in DMSO. Oscillator strengths are represented as vertical navy-blue lines.

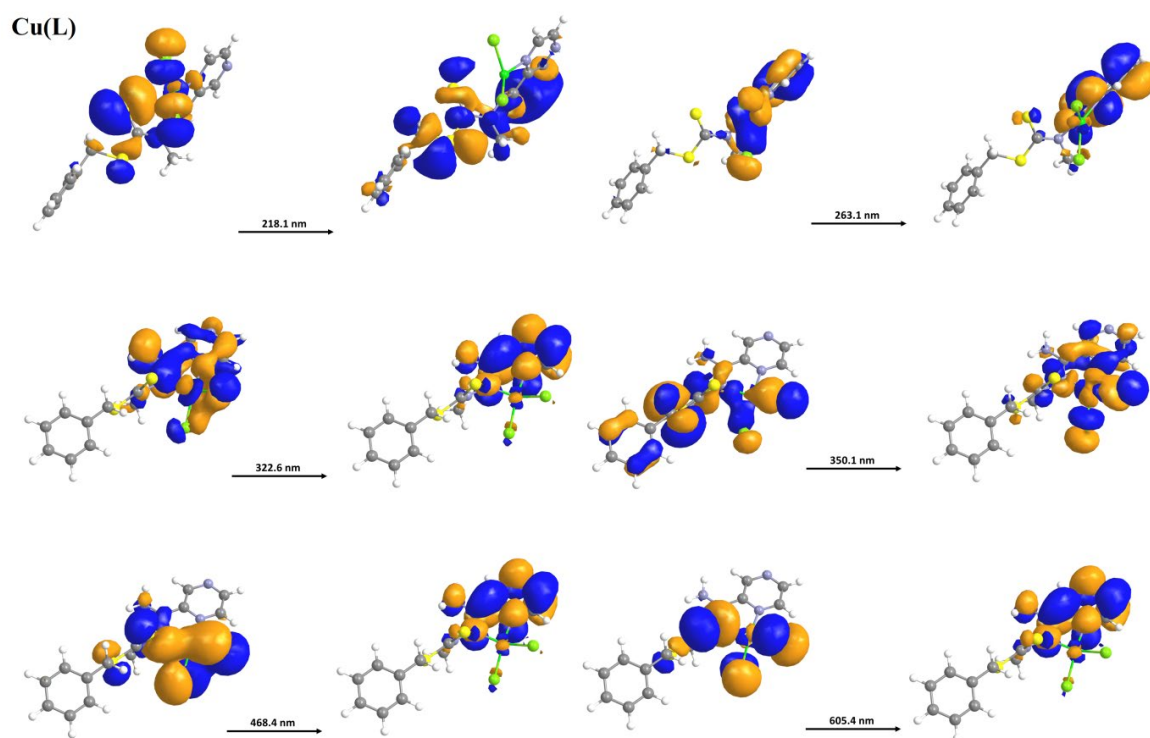


Figure S11. The main molecular orbital transitions for **Cu(L)** in DMSO, which are associated with the solution absorption spectra of this compound.

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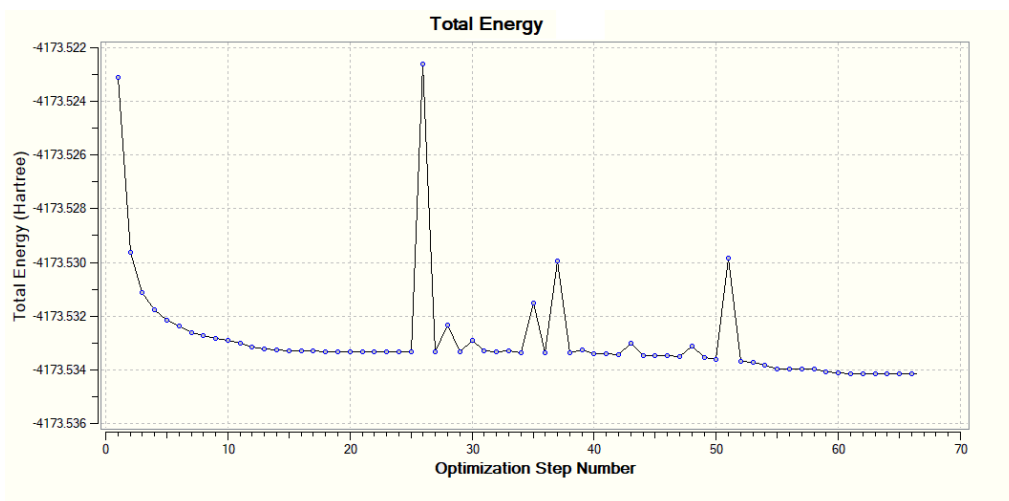


Figure S12. Changes in total energy during optimization of **Cu(L)** complex in water, showing overall decrease in energy, suggesting the system's energetic stability in aqueous environment.

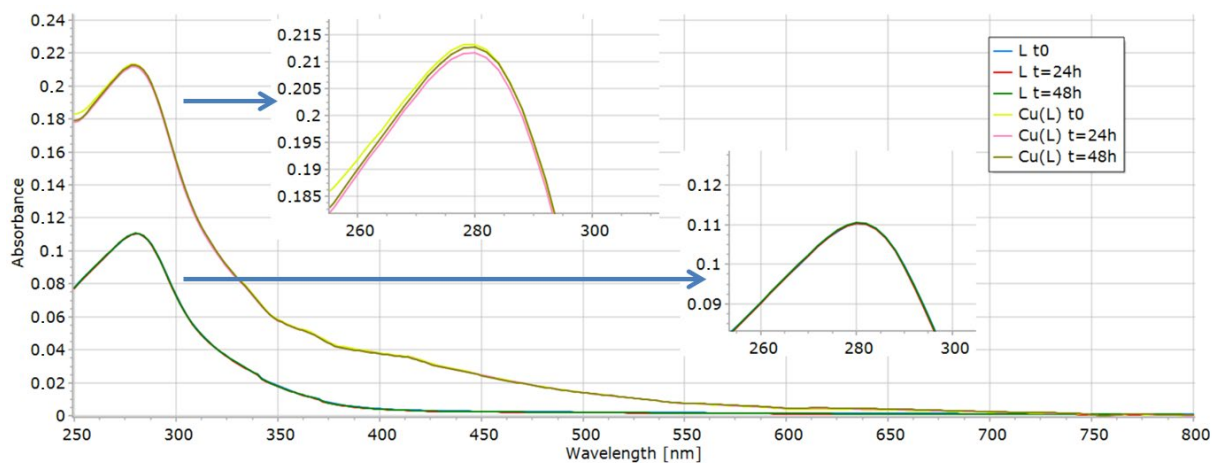


Figure S13. Time-dependent UV-Vis-monitored stability of **L** and **Cu(L)** in DMSO/H₂O (1:1) mixture (C = 5 μ M; T = 24°C).

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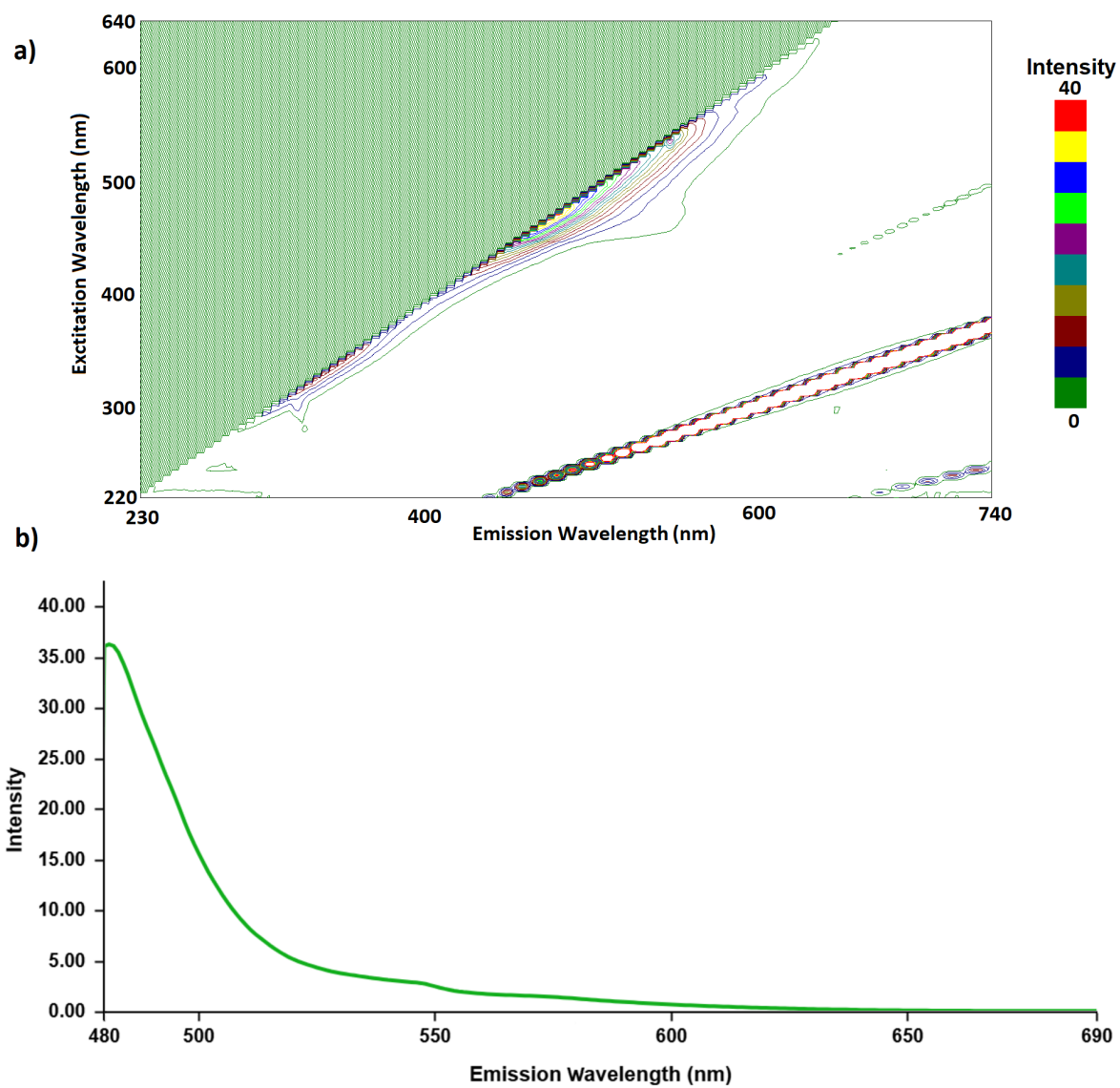


Figure S14. 3D fluorescence spectrum of ligand L (a) and 2D fluorescence spectrum of ligand L (b) obtained as a cross-section of the 3D spectrum at an excitation wavelength of 470 nm.

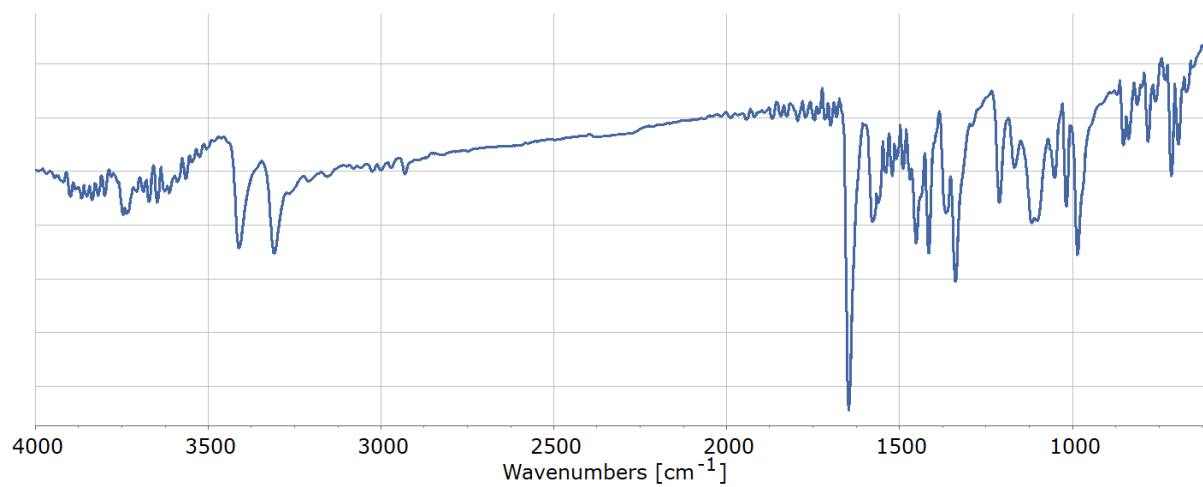


Figure S15. Experimental FTIR spectrum of the L ligand.

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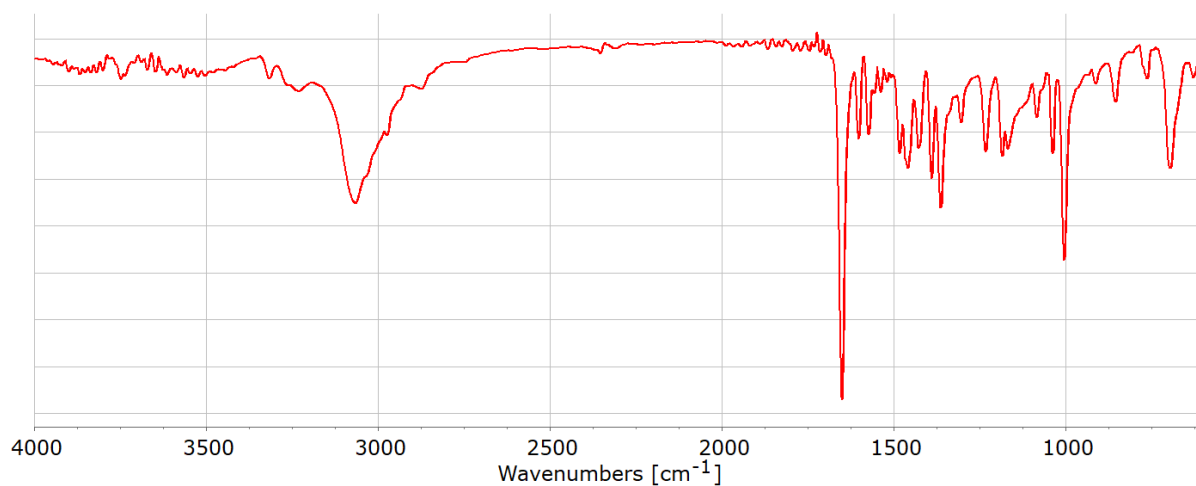


Figure S16. Experimental FTIR spectrum of the **Cu(L)** compound.

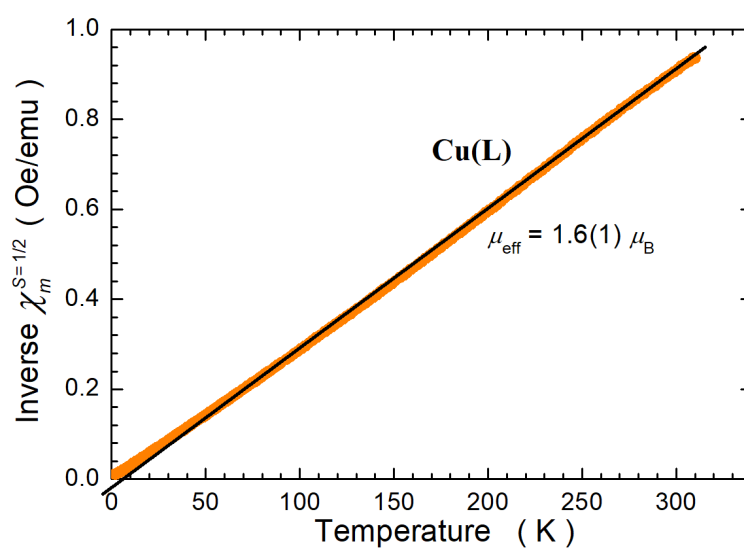


Figure S17. Temperature dependence of the inverse of the molar *spin* susceptibility of Cu ions, $\chi_m^{S=1/2}$ in the **Cu(L)** coordination compound (orange symbols). The straight black line approximates $1/T$ paramagnetic dependency. Its slope yields the magnitude of the inverse of the Curie constant, from which the value of the effective spin moment μ_{eff} is calculated and given in the units of Bohr magnetons μ_B .

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Pa	Pi	Cell-line	Description	Tissue/Organ	Type	IAP
0.633	0.004	A-375	Malignant melanoma	Skin	Melanoma	0.931
0.591	0.025	HeLa	Cervical adenocarcinoma	Cervix	Adenocarcinoma	0.885
0.558	0.004	NCI-H1299	Non-small cell lung carcinoma	Lung	Carcinoma	0.916
0.430	0.050	YAPC	Pancreatic carcinoma	Pancreas	Carcinoma	0.817
0.378	0.132	HCC1937	Breast Carcinoma	Breast	Carcinoma	0.806
0.363	0.104	SNU-5	Gastric Carcinoma	Stomach	Carcinoma	0.828
0.360	0.076	HGC-27	Gastric carcinoma	Stomach	Carcinoma	0.830
0.339	0.066	DU-145	Prostate carcinoma	Prostate	Carcinoma	0.896
0.339	0.122	HT1197	Carcinoma	Urinary bladder	Carcinoma	0.815
0.330	0.236	SK-LU-1	Adenocarcinoma	Lung	Carcinoma	0.810
0.328	0.050	HT	Lymphoma	Haematopoietic and lymphoid tissue	Leukemia	0.857
0.318	0.191	NCI-H358	Bronchioalveolar Carcinoma	Lung; Bronchiole	Carcinoma	0.805
0.314	0.136	KYSE-520	Esophageal squamous cell carcinoma	Esophagus	Carcinoma	0.858
0.314	0.272	GIST882	Gastrointestinal stromal tumor	Intestine	Carcinoma	0.824
0.305	0.240	NCI-H441	Papillary adenocarcinoma	Lung	Adenocarcinoma	0.813

Figure S18. Cancer cell lines the compound L is predicted to be active against. Pa – probably active; Pi – probably inactive; IAP - Invariant Accuracy of Prediction. Only results with a Pa value higher than 0.300 are included.

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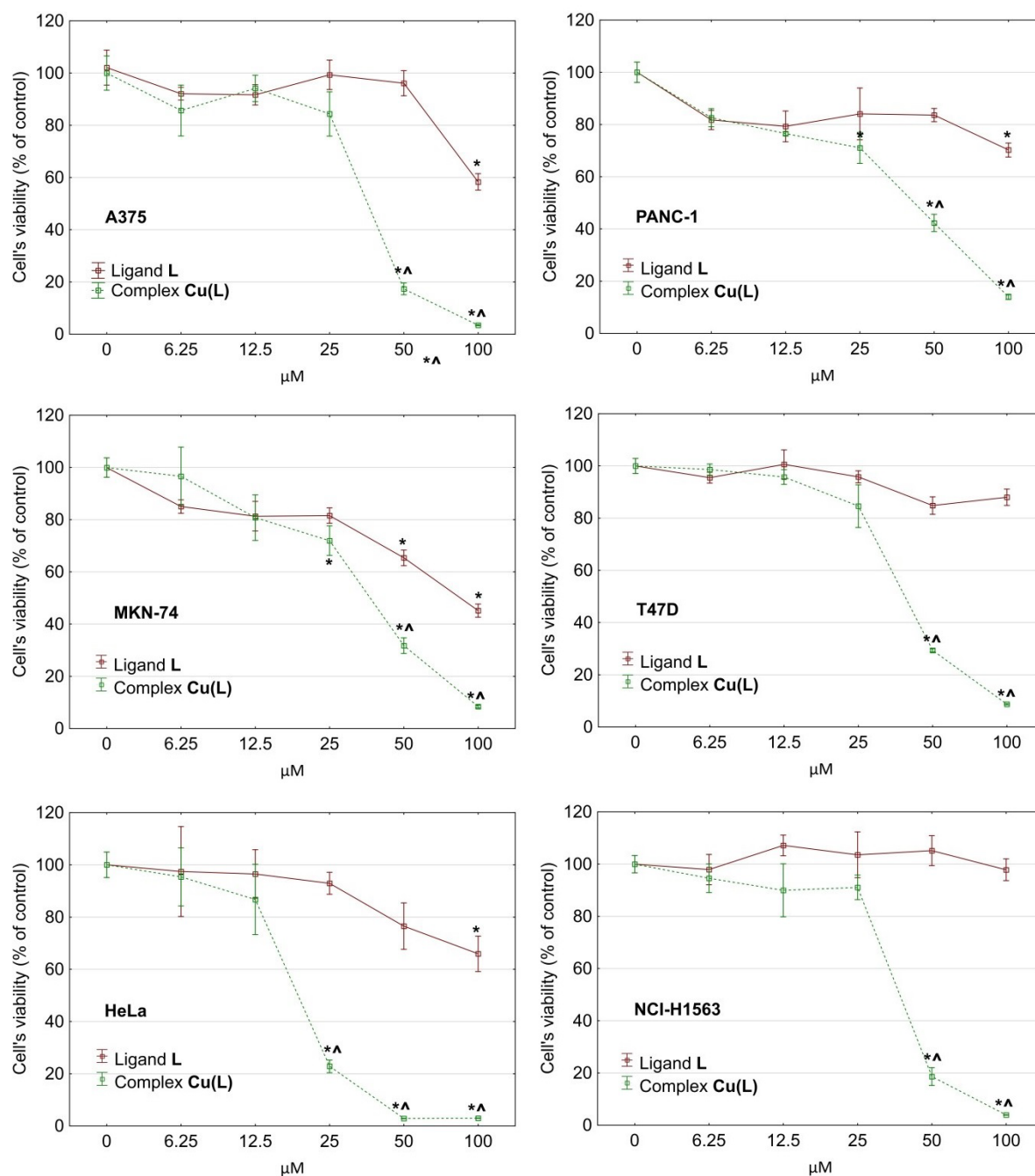
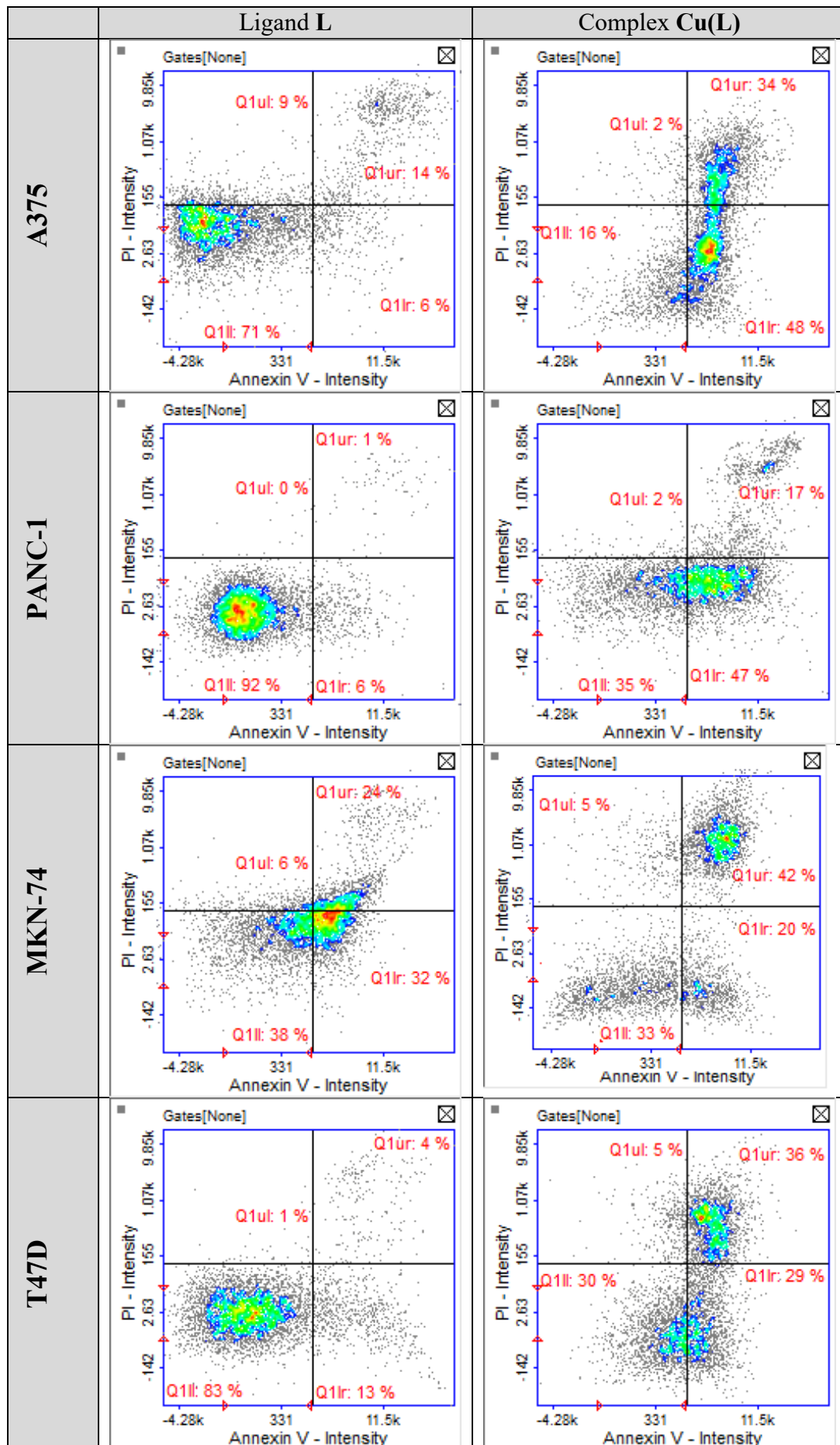


Figure S19. Cytotoxicity of ligand **L** and its complex, **Cu(L)**, based on MTT test results after 48 h incubation. Data are presented as % of control culture viability (mean \pm SD). * $p < 0.05$ vs. control, ^ $p < 0.05$ vs. ligand **L**.

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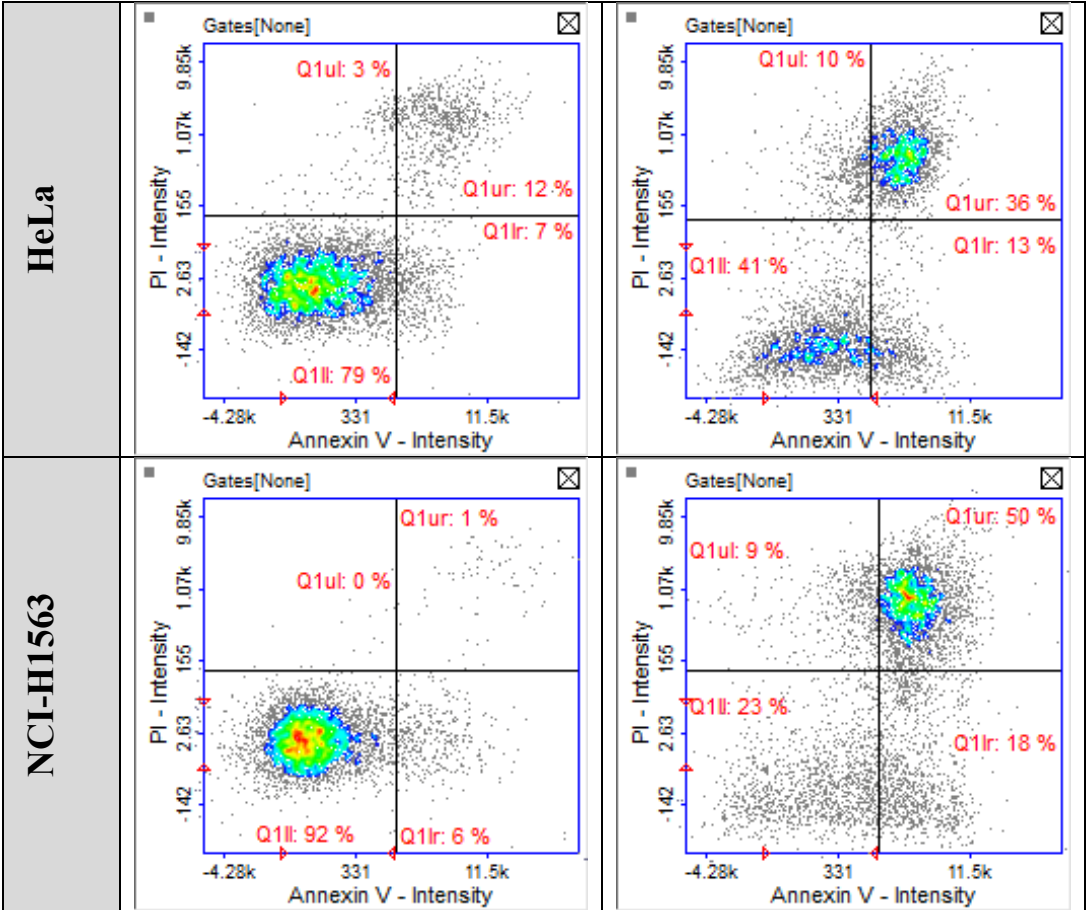


Figure S20. Apoptosis/necrosis detection by image cytometry. The results show one representative experiment. Q1II—live, Q1Ir—early apoptotic, Q1ur—late apoptotic, and Q1uI—necrotic cells.

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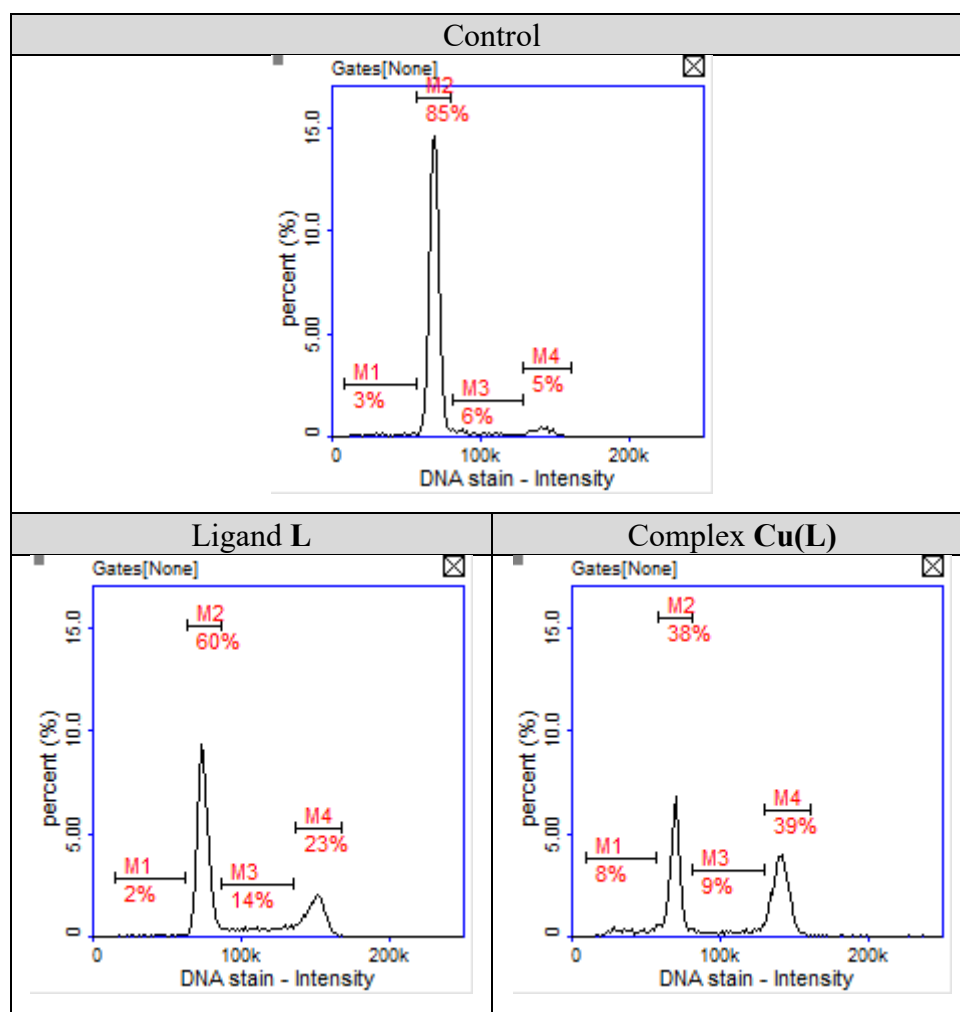


Figure S21. Representative histograms of cell cycle analysis in HeLa cells treated with ligand **L** (100 μM) and its complex, **Cu(L)** (in concentration corresponding to IC_{50} value – 17.50 μM) or DMSO as a vehicle in control culture for 48 h. M1—subG1, M2—G1, M3—S, M4—G2/M phase.

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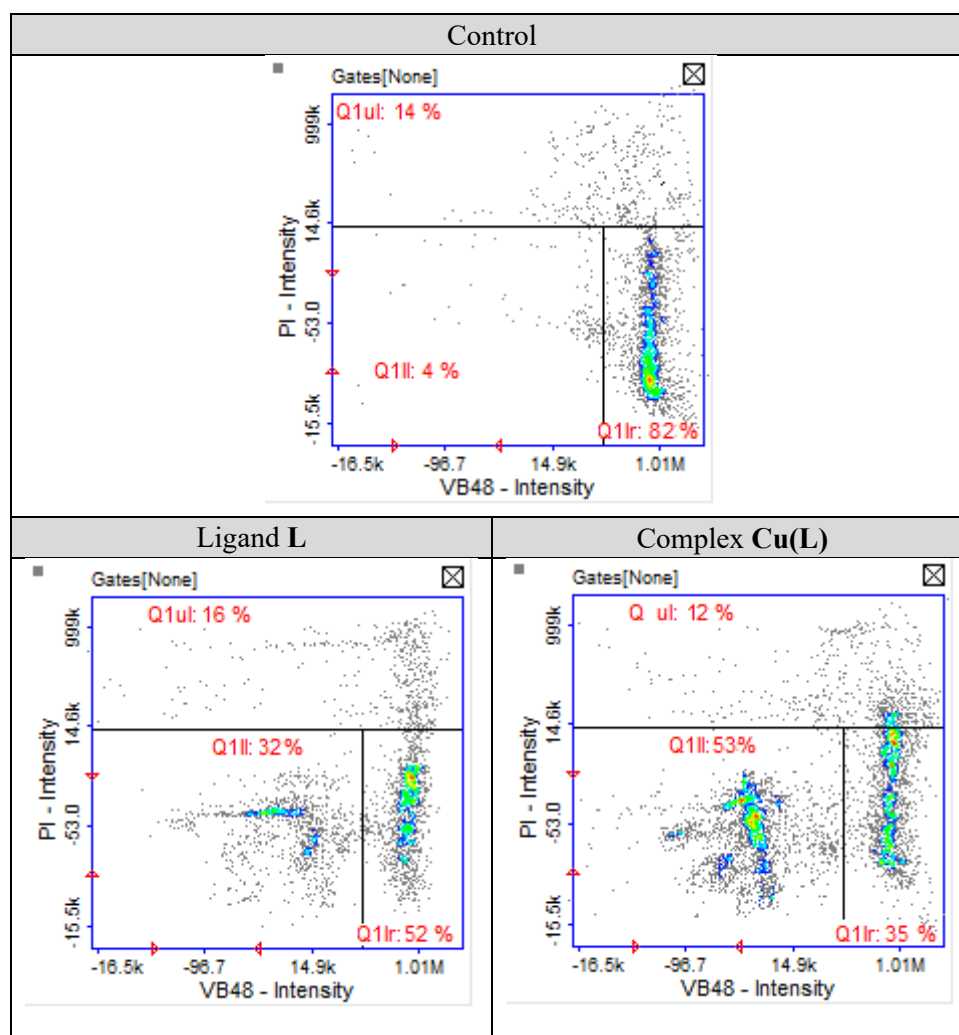


Figure S22. Representative histograms of thiol level distribution in HeLa cells treated with ligand **L** (100 μ M) and its complex, **Cu(L)** (in concentrations corresponding to IC_{50} value – 17.50 μ M) or DMSO as a vehicle in control culture for 48 h. Cells were stained with VitaBright-48TM and propidium iodide (PI). Q1Ir – healthy cells, Q1II – subpopulation with decreased thiol levels, Q1ul – dead cells (PI-positive).

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Table S4. Antimicrobial activity of tested compounds presented as minimal inhibitory concentration (MIC), minimal bactericidal concentration (MBC), and minimal fungicidal concentration (MFC) [mg/L].

Microorganism	Compound				Reference
	Ligand L		Complex Cu(L)		Vancomycin
Gram-positive bacteria	MIC	MBC	MIC	MBC	(Van)
<i>S. aureus</i> ATCC 25923	>500	>500	250	500	0.98
<i>S. aureus</i> ATCC BAA-1707*	>500	>500	250	250	0.98
<i>S. epidermidis</i> ATCC 12228	>500	>500	250	500	0.98
<i>M. luteus</i> ATCC 10240	500	>500	250	250	0.12
<i>E. faecalis</i> ATCC 29212	>500	>500	250	>500	1.95
<i>B. cereus</i> ATCC 10876	>500	>500	250	250	0.98
Gram-negative bacteria	MIC	MBC	MIC	MBC	(Cip)
<i>S. typhimurium</i> ATCC 14028	>500	>500	>500	>500	0.061
<i>E. coli</i> ATCC 25922	>500	>500	>500	>500	0.015
<i>P. mirabilis</i> ATCC 12453	>500	>500	>500	>500	0.03
<i>K. pneumoniae</i> ATCC 13883	>500	>500	>500	>500	0.12
<i>P. aeruginosa</i> ATCC 9027	>500	>500	>500	>500	0.49
Yeasts	MIC	MFC	MIC	MFC	(Nys)
<i>C. albicans</i> ATCC 102231	>500	>500	250	>500	0.24
<i>C. parapsilosis</i> ATCC 22019	>500	>500	62.5	>500	0.24
<i>C. glabrata</i> ATCC 90030	>500	>500	>500	>500	0.48

* Methicillin-resistant *Staphylococcus aureus* (MRSA)