

Supplementary Information

Enhanced H₂ Recovery by Coupling the Water-Gas Shift Reaction with in-situ Carbon Capture and Mineralization using Earth Abundant Ca- and Mg-Silicates and Hydroxides

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Table S1. Information of the Pt/Al₂O₃ catalyst

<i>Catalyst</i>	<i>Pt/Al₂O₃</i>
Surface area	163.131m ² / g
Pore volume	0.303cc/g
Average pore diameter	6.282nm

Table S2.1. Selected saturated pressure from steam table

<i>t</i> [°C]	<i>T</i> [K]	<i>P_s</i> [bar]
200	473.15	15.5467
220	493.15	23.1929
250	523.15	39.7594
280	553.15	64.1646
300	573.15	85.8771

Table S2.2. Selected specific volume (m³/kg) for superheated steam from the steam table

<i>t</i> [°C]	<i>5 bar</i>	<i>10bar</i>	<i>20bar</i>	<i>50bar</i>
200	0.425	0.206	/	/
220	0.445	0.217	0.102	/
250	0.474	0.233	0.110	/
280	0.503	0.248	0.120	0.042
300	0.523	0.258	0.125	0.045

Table S3. Residual formate concentrations in the aqueous phase determined by NMR

<i>Sorbent Type</i>	<i>Formate Concentration (ppm)</i>
Blank Experiment	1702.15
Mg(OH) ₂	102.46
Mg ₂ SiO ₄	147.99
Ca(OH) ₂	196.74
CaSiO ₃	131.97

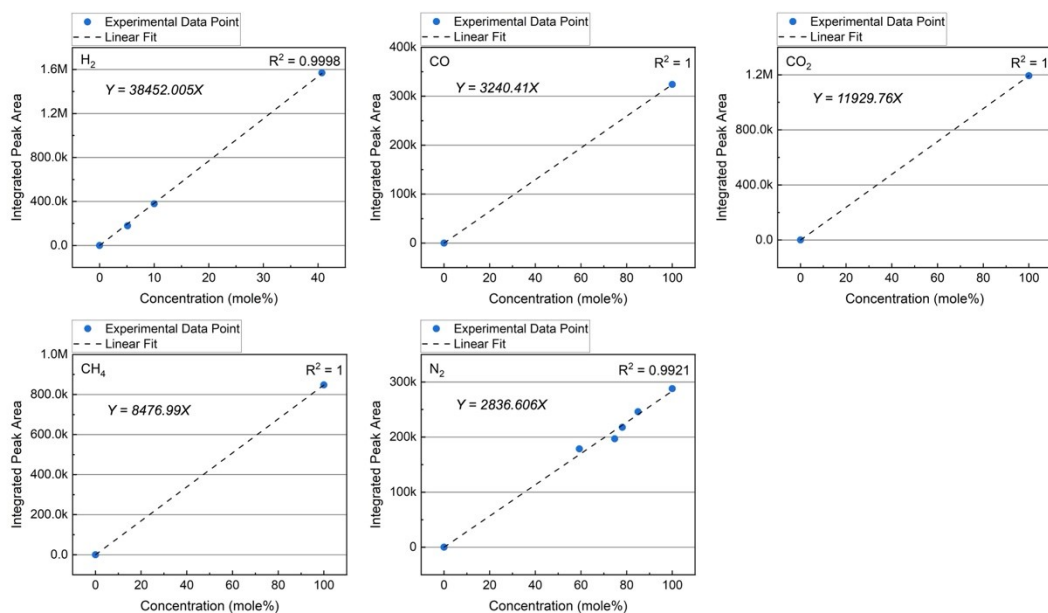


Figure S1. Gas chromatography (GC) calibration curves for different gas components.

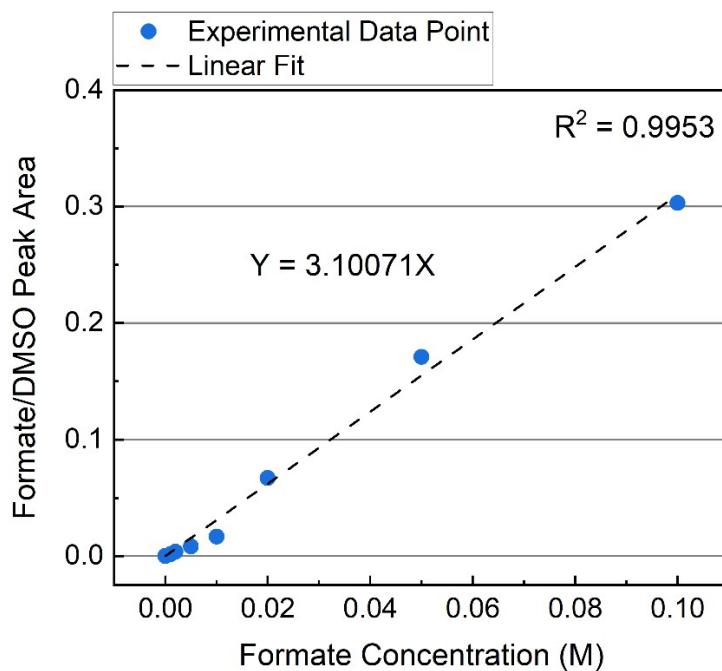


Figure S2. Formate concentration calibration line determined using liquid-NMR with DMSO internal standards.

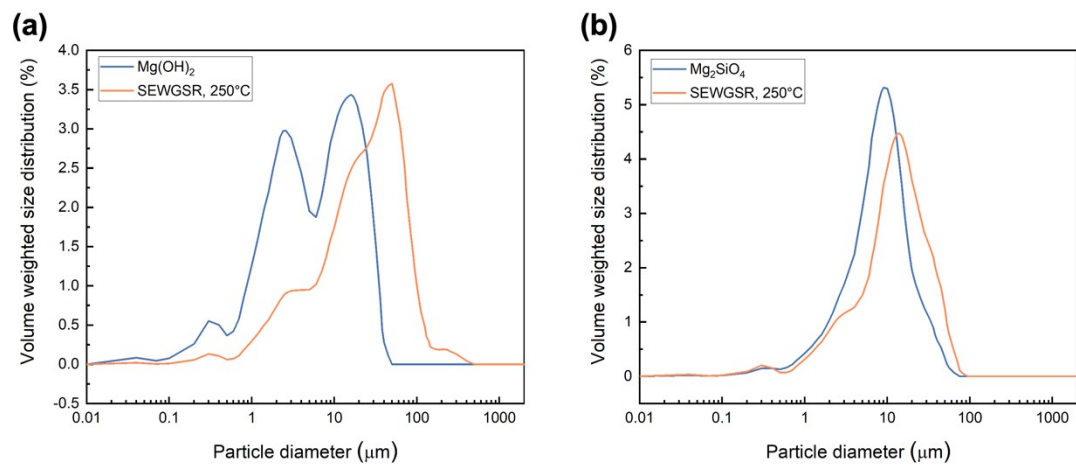


Figure S3. Particle size distribution of the unreacted and carbonated sorbents for (a) $Mg(OH)_2$; (b) Mg_2SiO_4 ;

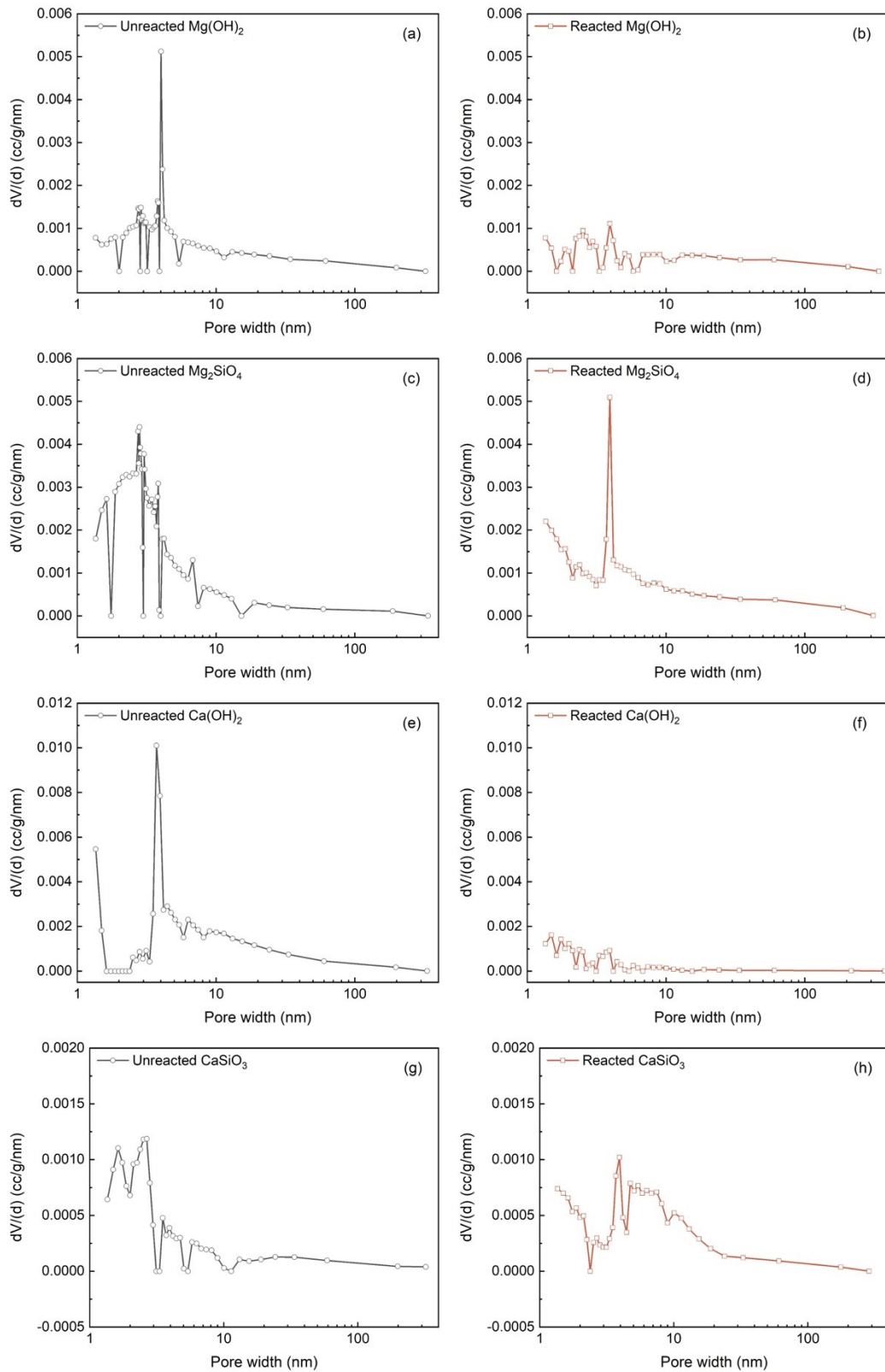


Figure S4. The pore size distribution of (a) unreacted $\text{Mg}(\text{OH})_2$; (b) reacted $\text{Mg}(\text{OH})_2$; (c) unreacted Mg_2SiO_4 ; (d) reacted Mg_2SiO_4 ; (e) unreacted $\text{Ca}(\text{OH})_2$; (f) reacted $\text{Ca}(\text{OH})_2$; (g) unreacted CaSiO_3 ; (h) reacted CaSiO_3 determined by the BJH model.

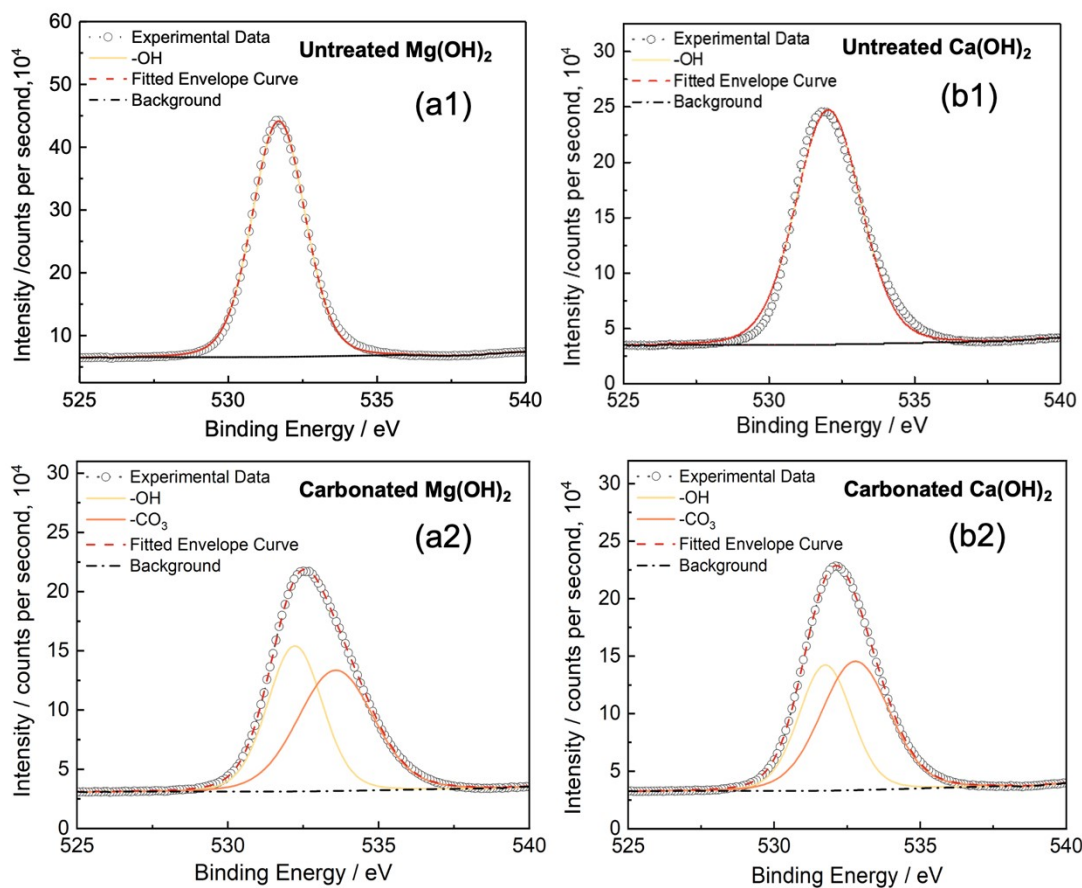


Figure S5. O1s XPS spectra of unreacted (a-1) Mg(OH)₂, (b-1) Ca(OH)₂ and carbonated (a-2) Mg(OH)₂; (b-2) Ca(OH)₂ sorbents respectively.

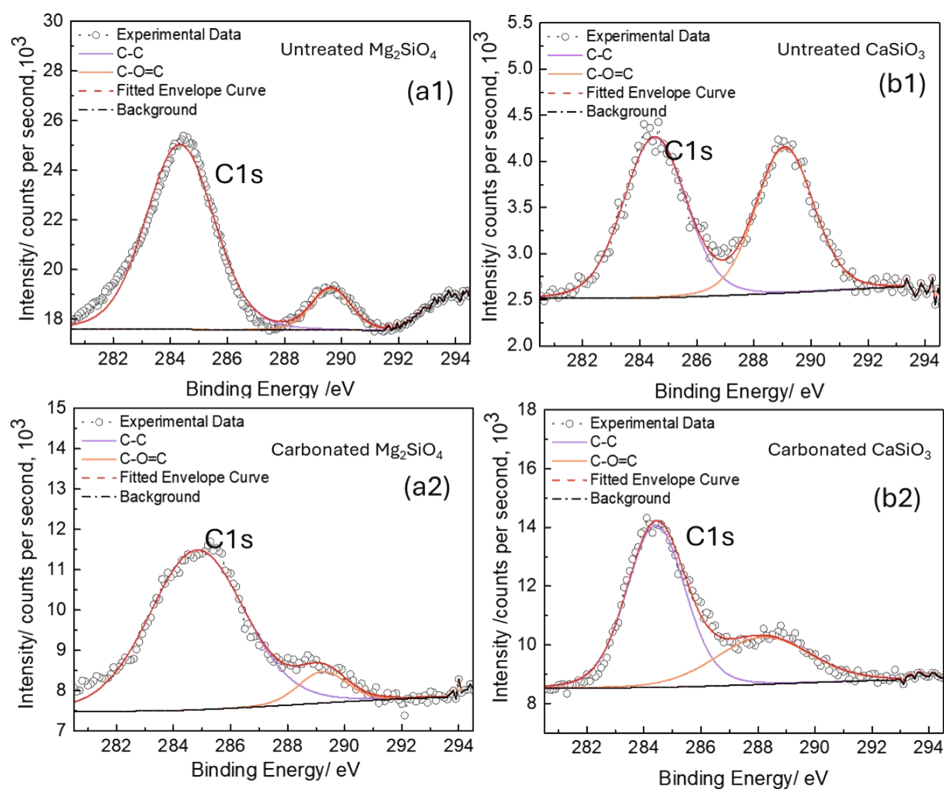


Figure S6. C1s XPS spectra of unreacted (a-1) Mg_2SiO_4 , (b-1) $CaSiO_3$ and carbonated (a-2) Mg_2SiO_4 ; (b-2) $CaSiO_3$ sorbents respectively.

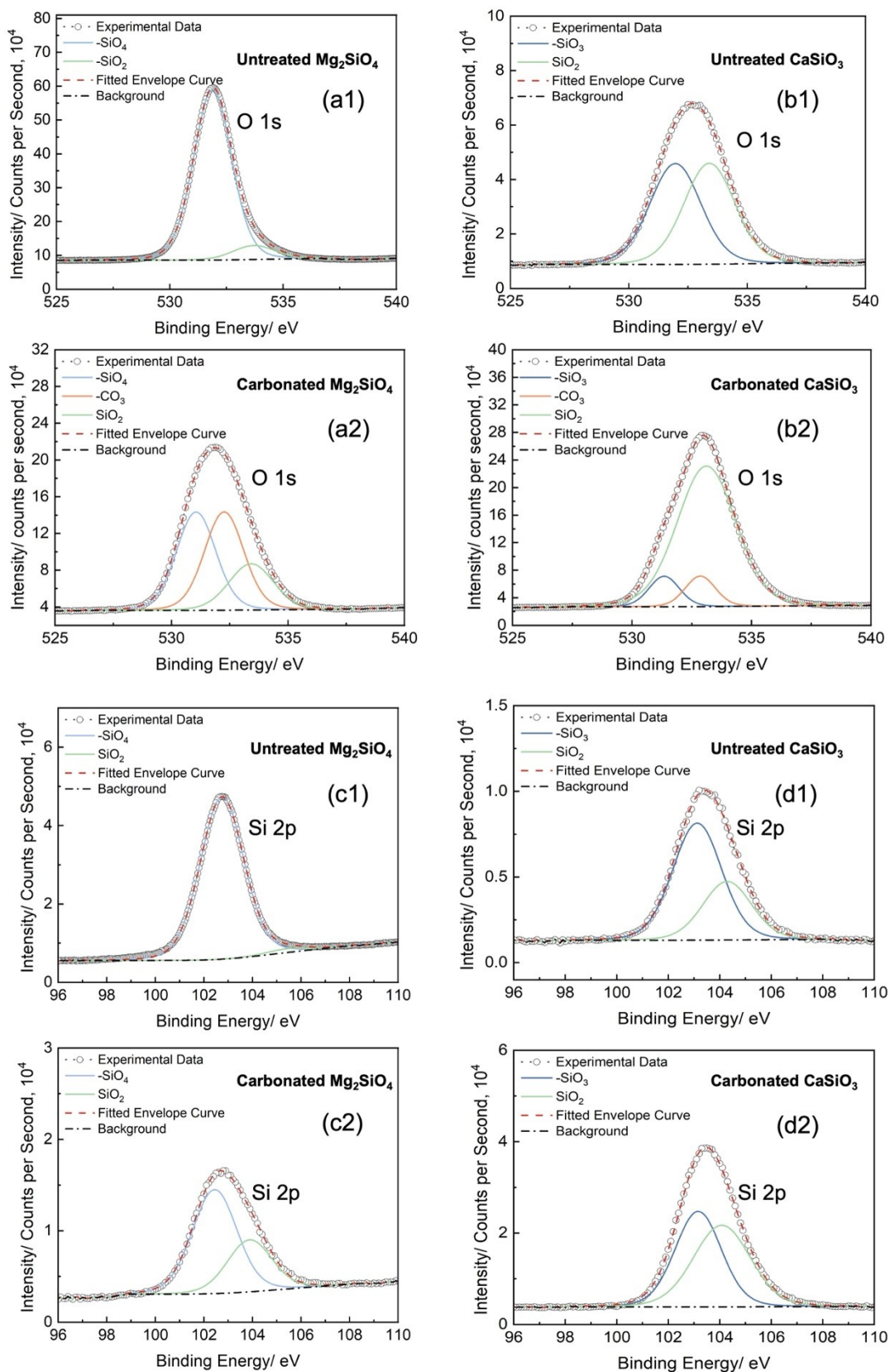


Figure S7. High resolution XPS spectra of ((a1-b2) O1s and (c1-d2) Si2p for unreacted and carbonated $CaSiO_3$, Mg_2SiO_4 sorbents, respectively

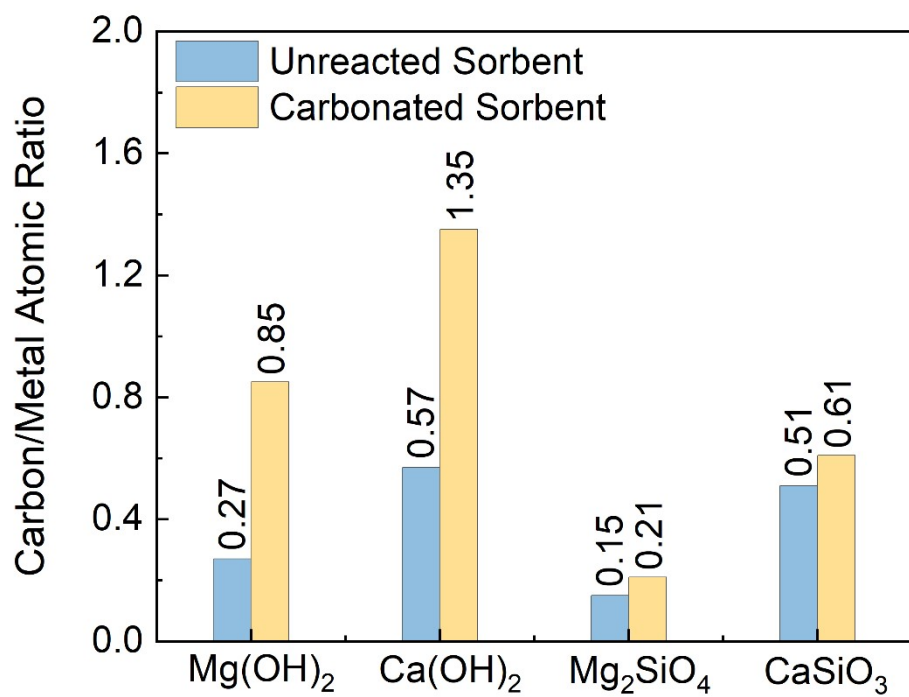


Figure S8. Carbon/Metal atomic ratios of the unreacted sorbents and corresponding carbonated products determined by XPS