

Supporting Information

Efficient and Convenient Synthesis of Methyl (S)-5-Chloro-2-Hydroxy-1-Oxo-2,3-Dihydro-1H-Indene-2-Carboxylate: A Key Intermediate for (S)-Indoxacarb

Using Aqueous TBHP as Oxidant

Yun Zhang,^{†,a,b} Yao Du,^{†,a,b} Yan-Biao Chen,^{a,b} Jia-Huan Nie,^{a,b} Yue Xiong,^{a,b} Bao-Dong Cui,^{a,b} Xue-Qing Mou,^{a,b} Ming-Qiang Zhou,^{*,a,c} Yong-Zheng Chen^{*,a,b}

^a Key Laboratory of Biocatalysis & Chiral Drug Synthesis of Guizhou Province, Generic Drug Research Center of Guizhou Province, Green Pharmaceuticals Engineering Research Center of Guizhou Province, School of Pharmacy, Zunyi Medical University, Zunyi 563006, P. R. China.

^b Key Laboratory of Basic Pharmacology of Ministry of Education and Joint International Research Laboratory of Ethnomedicine of Ministry of Education, Zunyi Medical University, Zunyi 563006, P. R. China.

^c National Engineering Research Center of Chiral Drugs, Chengdu Institute of Organic Chemistry, Chinese Academy of Sciences, Chengdu 610041, P. R. China;

[†] These authors contributed equally to this work.

Table of Contents

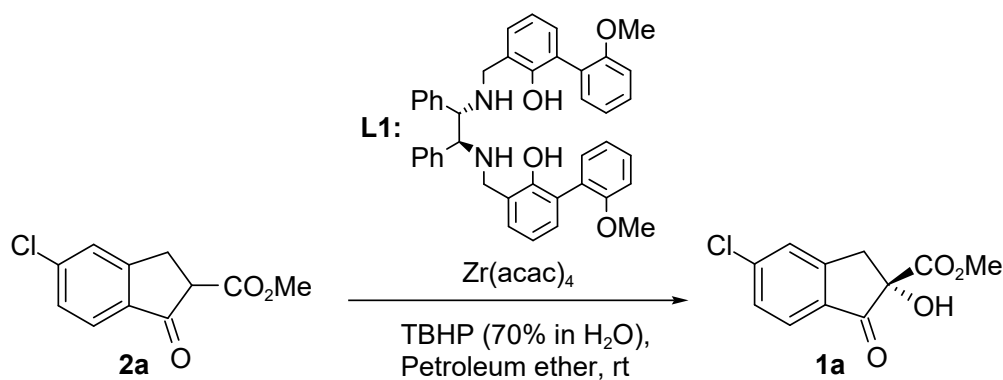
1. General information	S3
2. Reaction optimization tables	S3
3. Preparation of chiral ligan L1.....	S5
4. General procedure for the synthesis of β-keto esters.....	S6
5. General procedure for the enantioselective hydroxylation of β-keto ester	S12
6. Gram-scale experiment	S18
7. The recyclability of the catalytic system	S18
8. Mechanistic studies	S19
9. References	S22
10. Copies of ^1H and ^{13}C NMR spectras.....	S23
11. Copies of HPLC chromatogram	S46

1. General Information

TLC analysis was performed on pre-coated, glass-backed silica gel plates and visualized with UV light. Column chromatography was performed on silica gel (200-300 mesh). ^1H and ^{13}C NMR spectra were recorded on Agilent-400 MHz. Chemical shifts are reported in ppm versus tetramethylsilane with either tetramethylsilane or the residual solvent resonance used as an internal standard. Abbreviations were used in the description of NMR data as follows: chemical shift (δ , ppm), multiplicity (s = singlet, d = doublet, t = triplet, dd = doublet of doublets, m = multiplet), coupling constant (J , Hz). Mass spectra were determined on an Agilent 6520 QTOF., using electron spray ionization (ESI). Melting points are performed on an SGWX-4 digital visual melting point apparatus without correction. All other commercial chemicals were used without further purification.

2. Reaction optimization tables

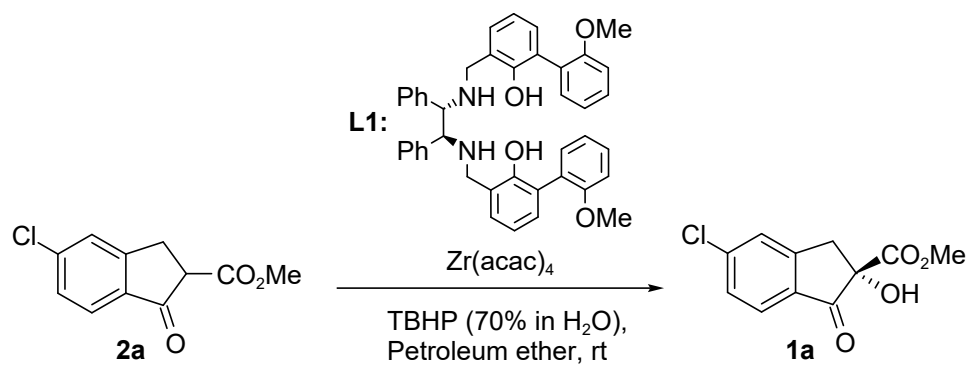
2.1 Optimization of the amount of $\text{Zr}(\text{acac})_4$ and **L1** (Table S1)



Entry ^a	$\text{Zr}(\text{acac})_4$ (equiv.)	L1 (equiv.)	Yield ^b	ee ^c
1	0.100	0.110	88%	99%
2	0.050	0.055	88%	99%
3	0.040	0.044	89%	99%
4	0.030	0.033	82%	99%
5	0.020	0.022	32%	99%

^a Reaction conditions: $\text{Zr}(\text{acac})_4$ (x equiv.), **L1** (y equiv.), Petroleum ether (4.3 mL), 30 min. Then, **2a** (0.25 mmol, 1.0 equiv.), TBHP (70% in H_2O) (0.375 mmol, 1.5 equiv.), 24 h. ^b Isolated yield and purification by filtration. ^c Determined by chiral HPLC.

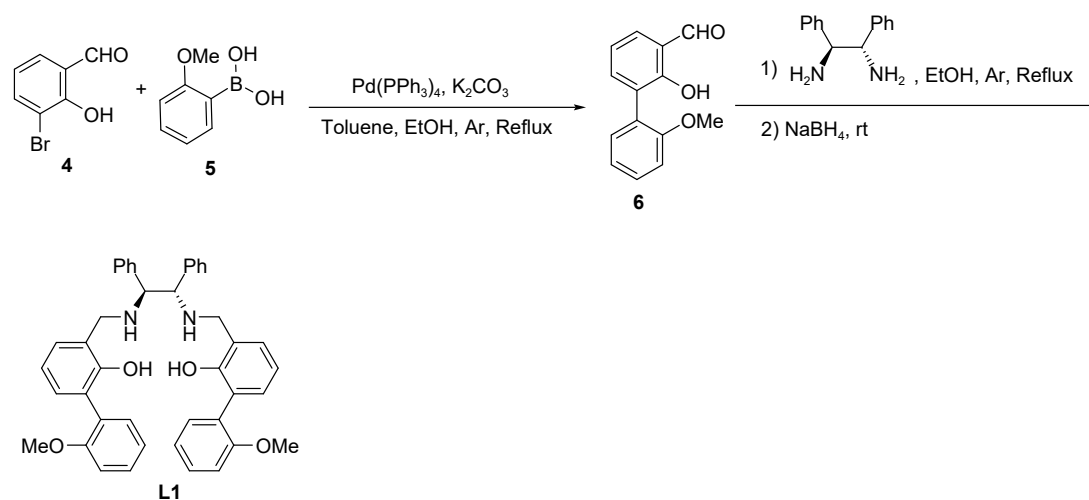
2.2 Optimization of the amount of TBHP and the volume of petroleum ether (Table S2)



Entry ^a	TBHP (70% in H ₂ O) (equiv.)	Volume of petroleum ether	Yield ^b	ee ^c
1	1.5	4.3 mL	89%	99%
2	2.0	4.3 mL	82%	99%
3	1.2	4.3 mL	84%	99%
4	1.0	4.3 mL	78%	99%
5	1.5	3.5 mL	92%	99%
6	1.5	3.0 mL	86%	99%

^a Reaction conditions: Zr(acac)₄ (0.01 mmol, 0.04 equiv.), L1 (0.011 mmol, 0.044 equiv.), Petroleum ether (x mL), 30 min. Then, **2a** (0.25 mmol, 1.0 equiv.), TBHP (70% in H₂O) y equiv.), 24 h. ^b Isolated yield and purification by filtration. ^c Determined by chiral HPLC.

3. Preparation of chiral ligan L1

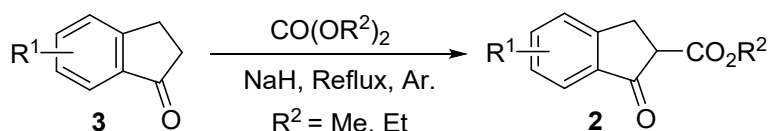


Under argon, 60 mL of toluene and 15 mL of ethanol were added to a mixture of 3-bromo-2-hydroxybenzaldehyde (2.010 g, 10 mmol), (2-methoxyphenyl) boronic acid (1.823 g, 12 mmol) and $\text{Pd}(\text{PPh}_3)_4$ (0.254 g, 0.22 mmol), giving a yellow solution, to which solid anhydrous K_2CO_3 (3.040 g, 22 mmol) was added. The reaction mixture was stirred for 4 hours under reflux and then allowed to cool to room temperature. Afterwards, the mixture was transferred to a separating funnel to which water and ethyl acetate were added. The water phase was extracted with ethyl acetate, and the organic phase was combined and dried with Na_2SO_4 . Purification using column chromatography on silica gel (petroleum ether/dichloromethane = 3/1) to give the desired product **6** as a white solid (2.043 g, 89%). ^1H NMR (CDCl_3 , 400 MHz) δ 11.31 (s, 1H), 9.94 (s, 1H), 7.56 (t, $J = 8.3$ Hz, 2H), 7.38 (t, $J = 7.9$ Hz, 1H), 7.29 (d, $J = 7.4$ Hz, 1H), 7.12 – 6.98 (m, 3H), 3.80 (s, 3H). Analytical data are in agreement with those reported in the literature^[1].

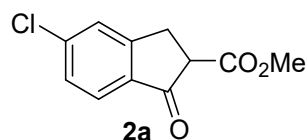
To a solution of $(1S,2S)$ -1,2-diphenylethane-1,2-diamine (0.849 g, 4.0 mmol) in ethanol (120 mL) was added 2-hydroxy-2'-methoxy-[1,1'-biphenyl]-3-carbaldehyde **6** (1.826 g, 8.0 mmol). The reaction mixture refluxed for 14 h under argon. After cooling to 0 °C, to the above reaction mixture was added sodium borohydride (0.454 g, 12 mmol). The resulting mixture was stirred at room temperature for 5 h, and a second portion of sodium borohydride (0.227 g, 6 mmol) was added. After stirring at room temperature for 1 h, the reaction mixture was poured into water 50 mL and the ethanol

was removed in vacuo. The suspension was extracted with dichloromethane and the combined organic layers were washed with brine. The organic layer was dried over anhydrous Na_2SO_4 and the solvent was concentrated under vacuum. The residue was purified by column chromatography on silica gel (petroleum ether/ethyl acetate = 5/1) to **L1** as a white solid (2.394 g, 94%). $^1\text{H NMR}$ (400 MHz, CDCl_3) δ 7.37 – 7.33 (m, 2H), 7.31 (dd, $J = 7.4, 1.5$ Hz, 2H), 7.24 – 7.18 (m, 6H), 7.16 (dd, $J = 7.4, 1.7$ Hz, 2H), 7.04 (t, $J = 7.5$ Hz, 2H), 7.00 (d, $J = 8.3$ Hz, 2H), 6.98 – 6.93 (m, 4H), 6.86 – 6.82 (m, 2H), 6.79 (t, $J = 7.4$ Hz, 2H), 4.01 (s, 2H), 3.86 (d, $J = 13.5$ Hz, 2H), 3.77 (s, 6H), 3.65 (d, $J = 13.5$ Hz, 2H). Analytical data are in agreement with those reported in the literature [2].

4. General procedure for the synthesis of β -keto esters [2,3]

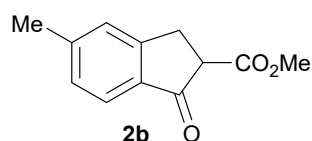


To a suspension of NaH (7.5 mmol, 60% dispersion in mineral oil) in dimethyl carbonate or diethyl carbonate (4.8 mL) was added a solution of indanone **3** (3.0 mmol) in dimethyl carbonate or diethyl carbonate (4.8 mL) dropwise at room temperature. The resulting mixture was heated to reflux until full conversion of **3** as indicated by TLC. The resulting solid was dissolved in H_2O (12 mL) and HCl (2 M, 3 mL), extracted with DCM and dried over Na_2SO_4 . Purification using column chromatography on silica gel to give the desired product **2**.

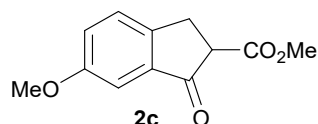


Following general procedure and **2a** was obtained as a yellow solid (579 mg, 86%) after purification by flash chromatography. $^1\text{H NMR}$ (400 MHz, CDCl_3) δ (duplicate signals are observed for some hydrogens; asterisks indicate those corresponding to the enol rotamer) 10.34* (s, 1H), 7.71 (d, $J = 8.2$ Hz, 1H), 7.56* (d, $J = 8.2$ Hz, 1H), 7.51

(s, 1H), 7.46* (s, 1H), 7.39 (d, $J = 8.2$ Hz, 1H), 3.86* (s, 3H), 3.80 (s, 3H), 3.76 (dd, $J = 8.3, 3.9$ Hz, 1H), 3.56 (dd, $J = 17.6, 4.0$ Hz, 1H), 3.51* (s, 2H), 3.36 (dd, $J = 17.5, 8.4$ Hz, 1H). ^{13}C NMR (100 MHz, CDCl_3) δ (duplicate signals are observed for some carbons; asterisks indicate those corresponding to the enol rotamer) 198.1, 169.2, 155.1, 144.8*, 142.2, 135.5*, 133.7, 128.8, 127.5*, 126.9, 125.9, 125.3*, 121.8*, 53.3, 53.11*, 53.09, 51.5*, 32.5*, 30.0. Analytical data are in agreement with those reported in the literature [2].

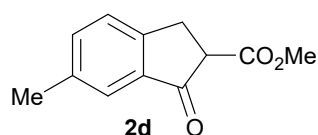


Following general procedure and **2b** was obtained as a brown liquid (490 mg, 80%) after purification by flash chromatography. ^1H NMR (400 MHz, CDCl_3) δ (duplicate signals are observed for some hydrogens; asterisks indicate those corresponding to the enol rotamer) 10.46* (s, 1H), 7.67 (d, $J = 7.9$ Hz, 1H), 7.54* (d, $J = 7.8$ Hz, 1H), 7.31 (s, 1H), 7.21 (d, $J = 7.9$ Hz, 1H), 3.85* (s, 3H), 3.80 (s, 3H), 3.74 (dd, $J = 8.2, 4.0$ Hz, 1H), 3.52 (dd, $J = 17.3, 3.7$ Hz, 1H), 3.48* (s, 2H), 3.33 (dd, $J = 17.2, 8.2$ Hz, 1H), 2.46 (s, 3H), 2.44* (s, 3H). ^{13}C NMR (100 MHz, CDCl_3) δ (duplicate signals are observed for some carbons; asterisks indicate those corresponding to the enol rotamer) 199.0, 169.8, 154.3, 147.0, 133.0, 129.3*, 129.2, 127.0*, 126.9, 124.64, 124.60*, 53.5, 53.4*, 52.9, 52.8*, 32.4*, 30.2, 22.3, 22.2*. Analytical data are in agreement with those reported in the literature [2].

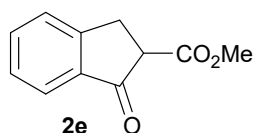


Following general procedure and **2c** was obtained as a yellow solid (362 mg, 55%) after purification by flash chromatography. ^1H NMR (400 MHz, CDCl_3) δ (duplicate signals are observed for some hydrogens; asterisks indicate those corresponding to the enol rotamer) 10.39* (s, 1H), 7.77* (s, 1H), 7.58 – 7.47* (m, 1H), 7.39 (d, $J = 7.9$ Hz,

1H), 7.25 – 7.18 (m, 2H), 7.00* (d, $J = 7.8$ Hz, 1H), 3.98* (d, $J = 26.9$ Hz, 2H), 3.85* (s, 3H), 3.84 (s, 3H), 3.80 (s, 3H), 3.75* (s, 3H), 3.65 (d, $J = 16.8$ Hz, 1H), 3.47 (d, $J = 17.4$ Hz, 1H), 3.31 (dd, $J = 16.8, 7.6$ Hz, 1H), 3.18* (d, $J = 16.9$ Hz, 1H). ^{13}C NMR (100 MHz, CDCl_3) δ (duplicate signals are observed for some carbons; asterisks indicate those corresponding to the enol rotamer) 200.9*, 199.6, 172.1*, 169.7, 159.9*, 159.8, 146.7, 145.3*, 136.5, 134.8*, 127.3, 125.8*, 125.5*, 125.2, 106.3*, 105.7, 56.1*, 55.8, 54.0, 53.6*, 52.9, 51.4*, 38.8*, 29.7. Analytical data are in agreement with those reported in the literature [2].

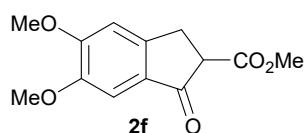


Following general procedure and **2d** was obtained as a gray solid (412 mg, 67%) after purification by flash chromatography. ^1H NMR (400 MHz, CDCl_3) δ (duplicate signals are observed for some hydrogens; asterisks indicate those corresponding to the enol rotamer) 10.38* (s, 1H), 7.57 (s, 1H), 7.51 – 7.42 (m, 2H), 7.39* (d, $J = 8.0$ Hz, 1H), 7.35* (d, $J = 7.7$ Hz, 1H), 7.24 (d, $J = 7.6$ Hz, 1H), 3.85 (s, 3H), 3.79* (s, 3H), 3.74 (dd, $J = 8.1, 3.8$ Hz, 1H), 3.51 (dd, $J = 17.2, 3.4$ Hz, 1H), 3.46* (s, 2H), 3.33 (dd, $J = 17.1, 8.1$ Hz, 1H), 2.43 (s, 3H), 2.40* (s, 3H). ^{13}C NMR (100 MHz, CDCl_3) δ (duplicate signals are observed for some carbons; asterisks indicate those corresponding to the enol rotamer) 199.7, 169.8, 151.1, 140.5*, 138.0, 136.9, 136.8*, 135.5, 130.6*, 126.3, 124.7, 124.5*, 121.2*, 53.6, 52.9, 51.3*, 32.2*, 30.0, 21.5*, 21.2. Analytical data are in agreement with those reported in the literature [2].

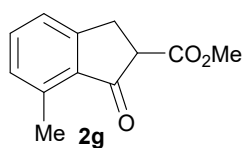


Following general procedure and **2e** was obtained as an orange liquid (437 mg, 77%) after purification by flash chromatography. ^1H NMR (400 MHz, CDCl_3) δ (duplicate signals are observed for some hydrogens; asterisks indicate those corresponding to the

enol rotamer) 10.39* (s, 1H), 7.78 (d, $J = 7.7$ Hz, 1H), 7.64 (t, $J = 7.5$ Hz, 1H), 7.51 (d, $J = 7.6$ Hz, 1H), 7.49 – 7.44* (m, 2H), 7.40 (t, $J = 7.3$ Hz, 1H), 3.86* (s, 3H), 3.80 (s, 3H), 3.75 (dd, $J = 8.0, 3.9$ Hz, 1H), 3.58 (dd, $J = 17.2, 3.6$ Hz, 1H), 3.52* (s, 2H), 3.39 (dd, $J = 17.3, 8.3$ Hz, 1H). ^{13}C NMR (100 MHz, CDCl_3) δ (duplicate signals are observed for some carbons; asterisks indicate those corresponding to the enol rotamer) 199.6, 169.6, 153.7, 135.7, 135.53*, 135.45, 135.2*, 128.3*, 127.5, 126.8, 126.5*, 124.9, 124.7*, 53.4, 53.1*, 53.0, 52.8*, 30.3, 30.0*. Analytical data are in agreement with those reported in the literature [2].

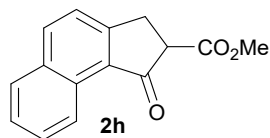


Following general procedure and **2f** was obtained as a white solid (709 mg, 94%) after purification by flash chromatography. ^1H NMR (400 MHz, CDCl_3) δ 7.18 (s, 1H), 6.92 (s, 1H), 3.99 (s, 3H), 3.91 (s, 3H), 3.80 (s, 3H), 3.74 (dd, $J = 7.8, 3.5$ Hz, 1H), 3.47 (dd, $J = 17.0, 3.3$ Hz, 1H), 3.29 (dd, $J = 17.1, 7.9$ Hz, 1H). ^{13}C NMR (100 MHz, CDCl_3) δ 198.1, 170.0, 156.2, 149.9, 149.4, 128.0, 107.4, 104.9, 56.5, 56.3, 53.5, 52.9, 30.1. Analytical data are in agreement with those reported in the literature [2].

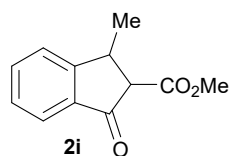


Following general procedure and **2g** was obtained as a white solid (527 mg, 86%) after purification by flash chromatography. ^1H NMR (400 MHz, CDCl_3) δ (duplicate signals are observed for some hydrogens; asterisks indicate those corresponding to the enol rotamer) 10.76* (s, 1H), 7.45 (t, $J = 7.5$ Hz, 1H), 7.28 (d, $J = 7.6$ Hz, 1H), 7.11 (d, $J = 7.4$ Hz, 1H), 3.83* (s, 2H), 3.78 (s, 3H), 3.69 (dd, $J = 8.4, 4.3$ Hz, 1H), 3.49 (dd, $J = 17.2, 4.2$ Hz, 1H), 3.44* (s, 1H), 3.30 (dd, $J = 17.2, 8.4$ Hz, 1H), 2.63* (s, 3H), 2.60 (s, 3H). ^{13}C NMR (100 MHz, CDCl_3) δ (duplicate signals are observed for some carbons; asterisks indicate those corresponding to the enol rotamer) 200.3, 169.9, 154.3,

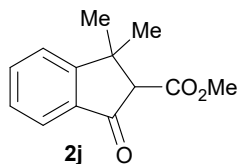
143.9*, 139.9, 134.8, 132.7, 129.7, 129.3*, 128.9*, 123.9, 122.3*, 53.6, 52.77*, 52.76, 51.2*, 32.2*, 29.9, 18.5*, 18.4.



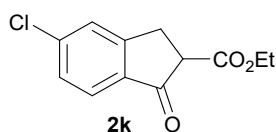
Following general procedure and **2h** was obtained as a yellow solid (539 mg, 75%) after purification by flash chromatography. ^1H NMR (400 MHz, CDCl_3) δ 9.05 (d, $J = 8.3$ Hz, 1H), 8.05 (d, $J = 8.4$ Hz, 1H), 7.87 (d, $J = 8.1$ Hz, 1H), 7.66 (t, $J = 7.6$ Hz, 1H), 7.56 (d, $J = 7.5$ Hz, 1H), 7.51 (d, $J = 8.5$ Hz, 1H), 3.87 – 3.82 (m, 1H), 3.81 (s, 3H), 3.63 (dd, $J = 17.5, 2.4$ Hz, 1H), 3.42 (dd, $J = 17.6, 7.8$ Hz, 1H). ^{13}C NMR (100 MHz, CDCl_3) δ 199.8, 169.9, 157.4, 136.8, 132.9, 129.6, 129.4, 129.3, 128.4, 127.0, 124.0, 123.7, 53.7, 52.9, 30.7.



Following general procedure and **2i** was obtained as a brown liquid (346 mg, 56%) after purification by flash chromatography. ^1H NMR (400 MHz, CDCl_3) δ (duplicate signals are observed for some hydrogens; asterisks indicate those corresponding to the enol rotamer) 10.55* (s, 1H), 7.76 (d, $J = 7.7$ Hz, 1H), 7.70 – 7.64 (m, 1H), 7.64 – 7.60* (m, 1H), 7.53 (d, $J = 7.7$ Hz, 1H), 7.48 – 7.45* (m, 1H), 7.42 (t, $J = 7.4$ Hz, 1H), 3.88* (s, 3H), 3.87 – 3.82 (m, 1H), 3.81 (s, 3H), 3.73* (s, 1H), 3.67* (dd, $J = 14.5, 7.3$ Hz, 1H), 3.32 (d, $J = 4.5$ Hz, 1H), 1.50 (d, $J = 7.1$ Hz, 3H), 1.42* (d, $J = 7.3$ Hz, 3H). ^{13}C NMR (100 MHz, CDCl_3) δ (duplicate signals are observed for some carbons; asterisks indicate those corresponding to the enol rotamer) 199.0, 169.6, 158.1, 135.7, 135.6*, 134.8, 129.8*, 128.1, 127.1*, 125.2, 124.5, 123.6*, 120.9*, 62.1, 58.1*, 52.8, 51.3*, 39.3*, 37.7, 20.1, 17.1*.

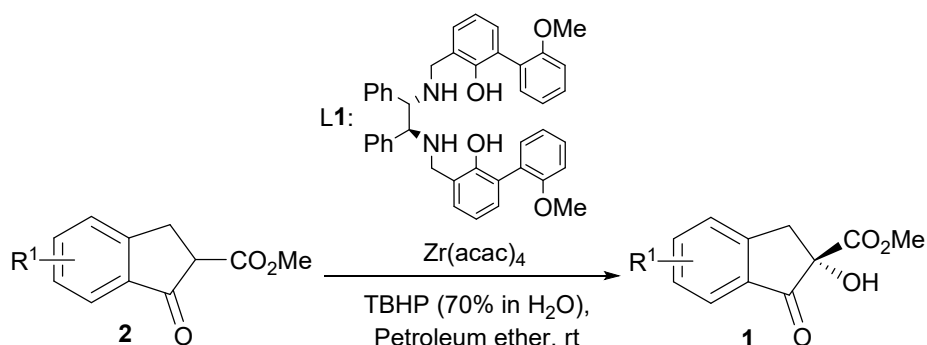


Following general procedure and **2j** was obtained as a brown liquid (519 mg, 79%) after purification by flash chromatography. ^1H NMR (400 MHz, CDCl_3) δ (duplicate signals are observed for some hydrogens; asterisks indicate those corresponding to the enol rotamer) 10.70* (s, 1H), 7.76 (d, $J = 7.7$ Hz, 1H), 7.68 – 7.63 (m, 1H), 7.61* (d, $J = 7.6$ Hz, 1H), 7.50 (d, $J = 7.8$ Hz, 1H), 7.47 – 7.42* (m, 2H), 7.40 (d, $J = 7.4$ Hz, 1H), 7.35* (td, $J = 7.4, 1.0$ Hz, 1H), 3.89* (s, 3H), 3.75 (s, 3H), 3.50 (s, 1H), 1.57 (s, 3H), 1.43* (s, 6H), 1.36 (s, 3H). ^{13}C NMR (100 MHz, CDCl_3) δ (duplicate signals are observed for some carbons; asterisks indicate those corresponding to the enol rotamer) 199.9, 169.7, 162.3, 155.2*, 135.8, 134.3, 134.2*, 130.2*, 128.1, 127.0*, 124.3, 123.5, 121.9*, 121.1*, 111.7*, 65.4, 52.2, 51.3*, 45.2*, 42.8, 30.6, 26.3*, 24.7.



Following general procedure and **2k** was obtained as a brown liquid (473 mg, 66%) after purification by flash chromatography. ^1H NMR (400 MHz, CDCl_3) δ (duplicate signals are observed for some hydrogens; asterisks indicate those corresponding to the enol rotamer) 10.43* (s, 1H), 7.71 (d, $J = 8.2$ Hz, 1H), 7.56* (d, $J = 8.0$ Hz, 1H), 7.51 (s, 1H), 7.46* (s, 1H), 7.38 (d, $J = 7.9$ Hz, 1H), 7.36* (s, 1H), 4.33* (q, $J = 6.7$ Hz, 2H), 4.25 (q, $J = 6.9$ Hz, 2H), 3.74 (dd, $J = 8.0, 3.2$ Hz, 1H), 3.55 (dd, $J = 17.3, 3.9$ Hz, 1H), 3.51* (s, 1H), 3.36 (dd, $J = 17.4, 8.2$ Hz, 1H), 1.37* (t, $J = 7.2$ Hz, 1H), 1.32 (t, $J = 7.0$ Hz, 3H). ^{13}C NMR (100 MHz, CDCl_3) δ (duplicate signals are observed for some carbons; asterisks indicate those corresponding to the enol rotamer) 198.2, 168.8, 155.2, 144.8*, 142.2, 135.6*, 133.8, 128.8, 127.5*, 126.9, 125.9, 125.3*, 121.7*, 62.1, 60.4*, 53.5, 32.6*, 30.1, 29.8*, 14.6*, 14.3.

5. General procedure for the enantioselective hydroxylation of β -keto ester



General procedure A:

To a solution of **L1** (0.014 g, 0.022 mmol) in petroleum ether (7.00 mL) was added $Zr(acac)_4$ (0.010 g, 0.020 mmol). The mixture was stirred at room temperature for 30 min. Then, β -keto ester **2** (0.50 mmol) and TBHP (70% in H_2O , 0.096 mL, 0.75 mmol) were added sequentially. The resulting solution was stirred at room temperature for 24 hours. The precipitates were formed during the course of the reaction, which was filtered and washed thoroughly with cold petroleum ether to give the desired product **1**.

General procedure B:

To a solution of **L1** (0.014 g, 0.022 mmol) in petroleum ether (7.00 mL) was added $Zr(acac)_4$ (0.010 g, 0.020 mmol). The mixture was stirred at room temperature for 30 min. Then, β -keto ester **2** (0.50 mmol) and TBHP (70% in H_2O , 0.096 mL, 0.75 mmol) were added sequentially, and the resulting solution was stirred at room temperature. After 24 hours, additional TBHP (70% in H_2O , 0.048 mL, 0.375 mmol) was added once every 8 hours for a total of 3 times. The precipitates were formed during the course of the reaction, which was filtered and washed thoroughly with cold petroleum ether to give the desired product **1**.

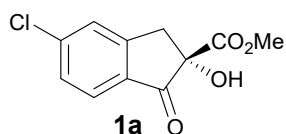
General procedure C:

To a solution of **L1** (0.014 g, 0.022 mmol) in petroleum ether (7.00 mL) was added $Zr(acac)_4$ (0.010 g, 0.020 mmol). The mixture was stirred at room temperature for 30

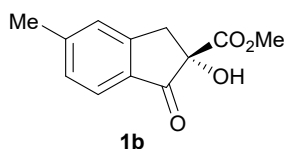
min. Then, β -keto ester **2** (0.50 mmol) and TBHP (70% in H₂O, 0.096 mL, 0.75 mmol) were added sequentially, and the resulting solution was stirred at room temperature. After 24 hours, additional TBHP (70% in H₂O, 0.048 mL, 0.375 mmol) was added once every 8 hours for a total of 3 times. The mixture was concentrated under vacuum. The residue was purified by column chromatography to give the desired product **1**.

General procedure D:

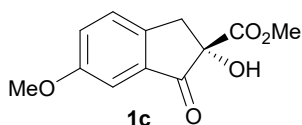
To a solution of **L1** (0.014 g, 0.022 mmol) in petroleum ether (7.00 mL) was added Zr(acac)₄ (0.010 g, 0.020 mmol). The mixture was stirred at room temperature for 30 min. Then, β -keto ester **2** (0.50 mmol) and TBHP (70% in H₂O, 0.096 mL, 0.75 mmol) were added sequentially. The resulting solution was stirred at room temperature for 24 hours. The mixture was concentrated under vacuum. The residue was purified by column chromatography to give the desired product **1**.



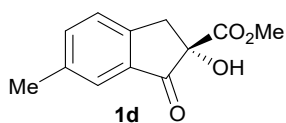
Following general procedure A and **1a** was obtained as a solid (111 mg, yield 92%, ee 99%) after purification by filtration. ¹H NMR (CDCl₃, 400 MHz): δ = 7.74 (d, J = 8.2 Hz, 1H), 7.50 (s, 1H), 7.42 (d, J = 8.2 Hz, 1H), 4.00 (s, 1H), 3.75 (s, 3H), 3.70 (d, J = 17.5 Hz, 1H), 3.24 (d, J = 17.5 Hz, 1H). ¹³C NMR(CDCl₃, 100 MHz): δ = 199.6, 171.7, 153.7, 143.0, 132.0, 129.2, 126.9, 126.5, 80.5, 53.8, 39.0. HRMS (ESI): m/z [M+Na]⁺ calcd for C₁₁H₉ClNaO₄: 263.0082; found: 263.0078. HPLC analysis CHIRALPAK OD-H, Hexane: *i*-PrOH=9:1, 25 °C, 1.0 mL/min flow rate, detection at 254 nm, t₁=13.99 min (major), t₂=17.75 min (minor). Analytical data are in agreement with those reported in the literature [2,3].



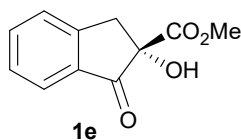
Following general procedure A and **1b** was obtained as a solid (109 mg, yield 99%, ee 97%) after purification by filtration. ^1H NMR (400 MHz, CDCl_3) δ 7.70 (d, $J = 7.9$ Hz, 1H), 7.30 (s, 1H), 7.24 (brs, 1H), 3.94 (s, 1H), 3.74 (s, 3H), 3.68 (d, $J = 17.2$ Hz, 1H), 3.21 (d, $J = 17.2$ Hz, 1H), 2.47 (s, 3H). ^{13}C NMR (100 MHz, DMSO) δ 200.7, 172.0, 152.6, 147.3, 131.2, 129.4, 127.2, 124.4, 80.0, 52.4, 40.7, 21.8. HRMS (ESI): calcd for $\text{C}_{12}\text{H}_{13}\text{O}_4$ $[\text{M}+\text{H}]^+$: 221.0808, found: 221.0805. HPLC analysis CHIRALPAK OD-H, Hexane: *i*-PrOH=47:3, 25 °C, 0.8 mL/min flow rate, detection at 254 nm, $t_1=20.20$ min (major), $t_2=24.92$ min (minor). Analytical data are in agreement with those reported in the literature [2].



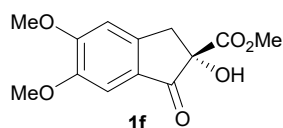
Following general procedure A and **1c** was obtained as a solid (106 mg, yield 90%, ee 94%) after purification by filtration. ^1H NMR (400 MHz, CDCl_3) δ 7.39 (d, $J = 8.5$ Hz, 1H), 7.28 (s, 1H), 7.22 (s, 1H), 4.11 (s, 1H), 3.85 (s, 3H), 3.74 (s, 3H), 3.66 (d, $J = 17.1$ Hz, 1H), 3.18 (d, $J = 16.8$ Hz, 1H). ^{13}C NMR (100 MHz, CDCl_3) δ 201.0, 172.1, 159.9, 145.3, 134.8, 127.3, 125.8, 106.3, 81.2, 55.8, 53.6, 38.8. HRMS (ESI): calcd for $\text{C}_{12}\text{H}_{13}\text{O}_5$ $[\text{M}+\text{H}]^+$: 237.0757, found: 237.0767. HPLC analysis CHIRALPAK OD-H, Hexane: *i*-PrOH=9:1, 25 °C, 1.0 mL/min flow rate, detection at 254 nm, $t_1=13.95$ min (major), $t_2=15.94$ min (minor). Analytical data are in agreement with those reported in the literature [2].



Following general procedure B and **1d** was obtained as a solid (90 mg, yield 81%, ee 95%) after purification by filtration. ^1H NMR (400 MHz, CDCl_3) δ 7.60 (s, 1H), 7.50 (d, $J = 7.8$ Hz, 1H), 7.39 (d, $J = 7.8$ Hz, 1H), 4.03 (s, 1H), 3.74 (s, 3H), 3.69 (d, $J = 17.2$ Hz, 1H), 3.21 (d, $J = 17.2$ Hz, 1H), 2.42 (s, 3H). ^{13}C NMR (100 MHz, CDCl_3) δ 201.0, 172.2, 149.8, 138.4, 137.6, 133.8, 126.3, 125.3, 80.8, 53.6, 39.1, 21.2. HRMS (ESI): calcd for $\text{C}_{12}\text{H}_{12}\text{NaO}_4$ $[\text{M}+\text{Na}]^+$: 243.0628, found: 243.0636. HPLC analysis CHIRALPAK OD-H, Hexane: *i*-PrOH=47:3, 25 °C, 0.8 mL/min flow rate, detection at 254 nm, t_1 =18.07 min (major), t_2 =21.32 min (minor). Analytical data are in agreement with those reported in the literature [2].

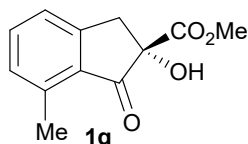


Following general procedure B and **1e** was obtained as a solid (87 mg, yield 85%, ee 90%) after purification by filtration. ^1H NMR (400 MHz, CDCl_3) δ 7.81 (d, $J = 7.7$ Hz, 1H), 7.68 (t, $J = 7.5$ Hz, 1H), 7.50 (d, $J = 7.6$ Hz, 1H), 7.44 (t, $J = 7.5$ Hz, 1H), 4.07 (s, 1H), 3.82 – 3.66 (m, 4H), 3.27 (d, $J = 17.4$ Hz, 1H). ^{13}C NMR (100 MHz, CDCl_3) δ 201.5, 171.9, 152.2, 136.3, 133.5, 128.3, 127.1, 124.5, 79.7, 52.5, 40.9. HRMS (ESI): calcd for $\text{C}_{11}\text{H}_{11}\text{O}_4$ $[\text{M}+\text{H}]^+$: 207.0652, found: 207.0655. HPLC analysis CHIRALPAK AD-H, Hexane: *i*-PrOH=22:3, 25 °C, 1.0 mL/min flow rate, detection at 254 nm, t_1 =18.07 min (major), t_2 =19.53 min (minor). Analytical data are in agreement with those reported in the literature [2].

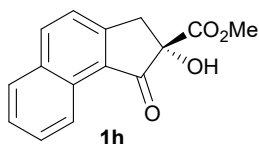


Following general procedure C and **1f** was obtained as a solid (60 mg, yield 45%, ee 97%) after purification by column chromatography. ^1H NMR (400 MHz, CDCl_3) δ 7.20 (s, 1H), 6.91 (s, 1H), 4.00 (s, 3H), 3.96 (d, $J = 11.0$ Hz, 1H), 3.92 (s, 3H), 3.75 (s, 3H), 3.65 (d, $J = 17.0$ Hz, 1H), 3.17 (d, $J = 17.0$ Hz, 1H). ^{13}C NMR (100 MHz, CDCl_3) δ

199.3, 172.3, 156.8, 150.0, 148.3, 126.2, 107.4, 105.3, 80.8, 56.5, 56.2, 53.5, 39.1. HRMS (ESI): calcd for C₁₃H₁₅O₆ [M+H]⁺: 267.0863, found: 267.0860. HPLC analysis CHIRALPAK AD-H, Hexane: *i*-PrOH=3:1, 25 °C, 1.0 mL/min flow rate, detection at 254 nm, t₁=19.22 min (major), t₂=23.06 min (minor). Analytical data are in agreement with those reported in the literature [2].

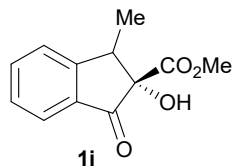


Following general procedure C and **1g** was obtained as a solid (69 mg, yield 62%, ee 53%) after purification by column chromatography. ¹H NMR (400 MHz, CDCl₃) δ 7.51 (dd, *J* = 13.4, 5.9 Hz, 1H), 7.30 (d, *J* = 7.6 Hz, 1H), 7.18 (d, *J* = 7.4 Hz, 1H), 4.03 (s, 1H), 3.74 (s, 3H), 3.70 (d, *J* = 17.3 Hz, 1H), 3.20 (d, *J* = 17.3 Hz, 1H), 2.63 (s, 3H). ¹³C NMR (100 MHz, CDCl₃) δ 201.6, 172.2, 152.9, 140.6, 135.6, 131.2, 130.0, 123.8, 80.5, 53.6, 38.8, 18.5. HRMS (ESI): calcd for C₁₂H₁₃O₄ [M+H]⁺: 221.0808, found: 221.0804. HPLC analysis CHIRALPAK OD-H, Hexane: *i*-PrOH=9:1, 25 °C, 1.0 mL/min flow rate, detection at 254 nm, t₁=9.22 min (major), t₂=11.40 min (minor).

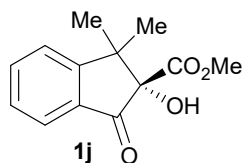


Following general procedure C and **1h** was obtained as a solid (39 mg, yield 30%, ee 30%) after purification by column chromatography. ¹H NMR (400 MHz, CDCl₃) δ 9.00 (d, *J* = 8.3 Hz, 1H), 8.14 (d, *J* = 8.4 Hz, 1H), 7.93 (d, *J* = 8.1 Hz, 1H), 7.71 (t, *J* = 7.6 Hz, 1H), 7.60 (t, *J* = 7.6 Hz, 1H), 7.55 (d, *J* = 8.3 Hz, 1H), 4.12 (s, 1H), 3.84 (d, *J* = 14.7 Hz, 1H), 3.74 (s, 3H), 3.37 (d, *J* = 17.6 Hz, 1H). ¹³C NMR (100 MHz, CDCl₃) δ 201.1, 172.2, 156.1, 137.6, 133.1, 129.8, 129.7, 128.6, 127.9, 127.3, 124.2, 123.7, 80.7, 53.6, 39.6. HRMS (ESI): calcd for C₁₅H₁₃O₄ [M+H]⁺: 257.0808, found: 257.0808.

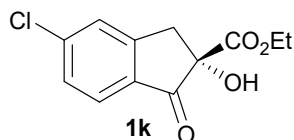
HPLC analysis CHIRALPAK OD-H, Hexane: *i*-PrOH=9:1, 25 °C, 1.0 mL/min flow rate, detection at 254 nm, t₁=16.85 min (major), t₂=19.27 min (minor).



Following general procedure C and **1i** was obtained as a solid (47 mg, yield 42%, ee 63%) after purification by column chromatography. ¹H NMR (400 MHz, CDCl₃) δ 7.80 (d, *J* = 7.7 Hz, 1H), 7.70 (td, *J* = 7.7, 1.1 Hz, 1H), 7.53 (dd, *J* = 7.8, 0.7 Hz, 1H), 7.44 (t, *J* = 7.5 Hz, 1H), 3.92 (s, 1H), 3.80 (q, *J* = 7.3 Hz, 1H), 3.73 (s, 3H), 1.34 (d, *J* = 7.3 Hz, 3H). ¹³C NMR (100 MHz, CDCl₃) δ 201.1, 172.5, 158.6, 136.4, 132.8, 128.3, 125.6, 125.2, 82.1, 53.7, 42.6, 16.2. HRMS (ESI): calcd for C₁₂H₁₃O₄ [M+H]⁺: 221.0808, found: 221.0807. HPLC analysis CHIRALPAK OD-H, Hexane: *i*-PrOH=22:3, 25 °C, 1.0 mL/min flow rate, detection at 254 nm, t₁=7.64 min (major), t₂=8.74 min (minor).



Following general procedure B and **1j** was obtained as a solid (63 mg, yield 53%, ee 2%) after purification by filtration. ¹H NMR (400 MHz, CDCl₃) δ 7.76 (d, *J* = 7.4 Hz, 1H), 7.67 (t, *J* = 7.2 Hz, 1H), 7.47 (d, *J* = 7.5 Hz, 1H), 7.41 (t, *J* = 7.4 Hz, 1H), 3.95 (s, 1H), 3.61 (s, 3H), 1.39 (s, 3H), 1.23 (s, 3H). ¹³C NMR (100 MHz, CDCl₃) δ 200.7, 172.0, 160.9, 136.1, 133.1, 128.0, 124.6, 123.3, 87.7, 53.3, 46.8, 29.9, 21.4. HRMS (ESI): calcd for C₁₃H₁₅O₄ [M+H]⁺: 235.0965, found: 235.0961. HPLC analysis CHIRALPAK OD-H, Hexane: *i*-PrOH=23:2, 25 °C, 0.8 mL/min flow rate, detection at 254 nm, t₁=11.52 min (major), t₂=12.60 min (minor).



Following general procedure D and **1k** was obtained as a solid (44 mg, yield 35%, ee 83%) after purification by column chromatography. ^1H NMR (400 MHz, CDCl_3) δ 7.74 (d, $J = 8.0$ Hz, 1H), 7.50 (s, 1H), 7.42 (d, $J = 8.2$ Hz, 1H), 4.22 (q, $J = 6.2$, 2H), 4.03 (s, 1H), 3.70 (d, $J = 17.8$ Hz, 1H), 3.23 (d, $J = 17.7$ Hz, 1H), 1.20 (t, $J = 7.1$ Hz, 3H). ^{13}C NMR (100 MHz, CDCl_3) δ 199.7, 171.2, 153.8, 142.9, 132.1, 129.1, 126.8, 126.5, 80.4, 63.1, 39.1, 14.1. HRMS (ESI): calcd for $\text{C}_{12}\text{H}_{11}\text{ClNaO}_4$ $[\text{M}+\text{Na}]^+$: 277.0238, found: 277.0254. HPLC analysis CHIRALPAK OD-H, Hexane: *i*-PrOH=9:1, 25 °C, 1.0 mL/min flow rate, detection at 254 nm, $t_1=10.29$ min (major), $t_2=12.49$ min (minor). Analytical data are in agreement with those reported in the literature [3].

6. Gram-scale experiment

To a solution of **L1** (0.841 g, 1.32 mmol) in petroleum ether (420.00 mL) was added $\text{Zr}(\text{acac})_4$ (0.585 g, 1.20 mmol). The mixture was stirred at room temperature for 30 min. Then, methyl 5-chloro-1-oxo-2,3-dihydro-1*H*-indene-2-carboxylate **2a** (6.739 g, 30.0 mmol) and TBHP (70% in H_2O , 5.79 mL, 45 mmol) were added sequentially. The resulting solution was stirred at room temperature for 36 hours. Fine white precipitates were formed during the course of the reaction. The white solid product **1a** was isolated by filtration and washed thoroughly with cold petroleum ether: the amount of the obtained product **1a** was 7.053 g (yield 98%, ee 91%).

7. The recyclability of the catalytic system

To a solution of **L1** (0.140 g, 0.22 mmol) in petroleum ether (70.0 mL) was added $\text{Zr}(\text{acac})_4$ (0.098 g, 0.20 mmol). The mixture was stirred at room temperature for 30 min. Then, methyl 5-chloro-1-oxo-2,3-dihydro-1*H*-indene-2-carboxylate **2a** (1.123 g, 5.0 mmol) and TBHP (70% in H_2O , 7.5 mmol) were added sequentially. The resulting solution was stirred at room temperature for 24 hours. Fine white precipitates were

formed during the course of the reaction. The white solid was isolated by filtration. Firstly, the filtrate solution containing the catalyst was decanted into another flask and fresh portions of ester **2a** and TBHP (70% in H₂O) were added to initiate the reaction, and the reaction was reperformed over 3 cycles. Then, the solid was washed thoroughly with cold petroleum ether to give the product **1a**.

First cycle: the amount of the obtained product **1a** was 1.095 g (yield 91%, ee 99%)

Second cycle: the amount of the obtained product **1a** was 0.902 g (yield 75%, ee 63%)

Third cycle: the amount of the obtained product **1a** was 0.517 g (yield 43%, ee 16%)

8. Mechanistic studies

8.1 HRMS Study

To a solution of **L1** (7.0 mg, 0.011 mmol) in petroleum ether (3.5 mL) was added Zr(acac)₄ (5.0 mg, 0.010 mmol). The mixture was stirred at room temperature for 30 min. Then, a small amount of the mixture was taken in dichloromethane for high resolution mass spectrometry. The HRMS of the in situ generated Zr(IV)-**L1** complex exhibits a ion peak at m/z 824.2404, whose mass correspond to [Zr^{IV}(**L1**)(acac)]⁺ (calculated m/z = 824.2397) (Figure S1)

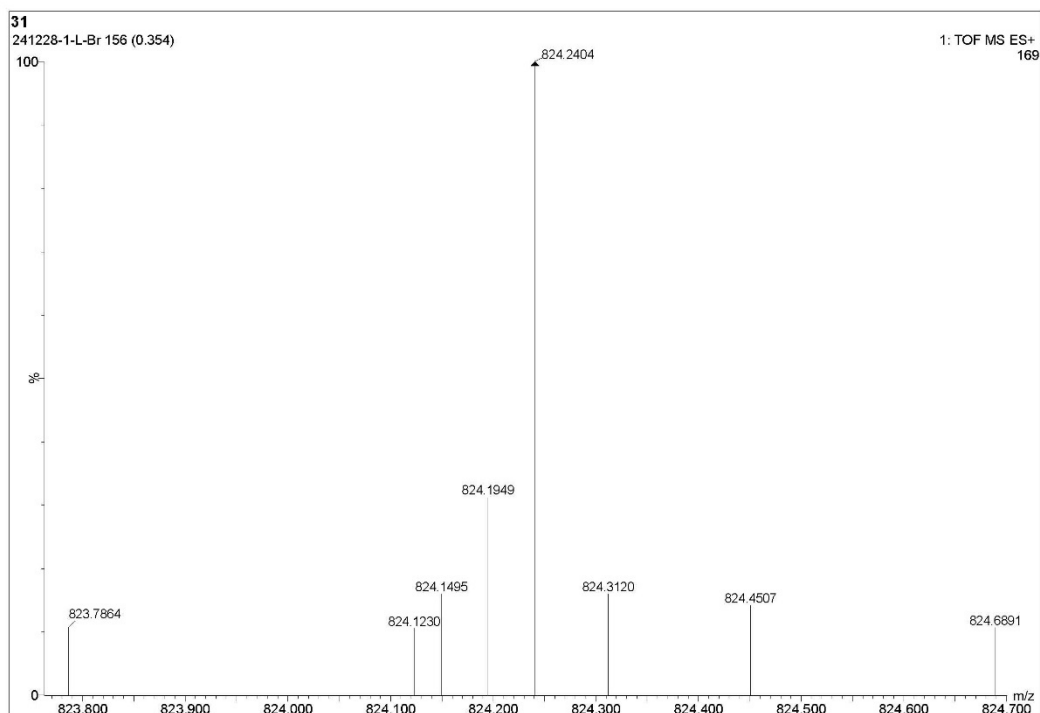


Figure S1 HRMS spectrum of Zr(IV)-**L1** generated in situ in the reaction of Zr(acac)₄ and **L1** in petroleum ether at room temperature.

8.2 NMR Study

A NMR tube was charged with **L1** (2.0 mg, 0.0031 mmol), Zr(acac)₄ (1.4 mg, 0.028 mmol) and cyclohexane-d₁₂ (1.0 mL) and the resulting solution was stirred at room temperature. After 30 min, the mixture was determined by ¹H NMR analysis. The full ¹H NMR spectra were given in Figure S2, compared with the ¹H NMR spectra of Zr(acac)₄, acetylacetonate and **L1**, significant chemical shift changes were observed at δ 5.48-5.06 ppm and 2.00-1.35 ppm. The spectra from δ 5.48 ppm to 5.06 ppm were enlarged in Figure S3. From Figure S3, the α-H (-CH-) of acac⁻ in Zr(acac)₄ (δ 5.33 ppm) disappeared and the α-H (-CH-) of acac⁻ in Zr(IV)-**L1** complex (δ 5.32 ppm) appeared. Furthermore, Figure S4 showed that the CH₃ of acac⁻ in Zr(acac)₄ (δ 1.80 ppm) disappeared and the CH₃ of acac⁻ in Zr(IV)-**L1** complex (δ 1.45 ppm) appeared.

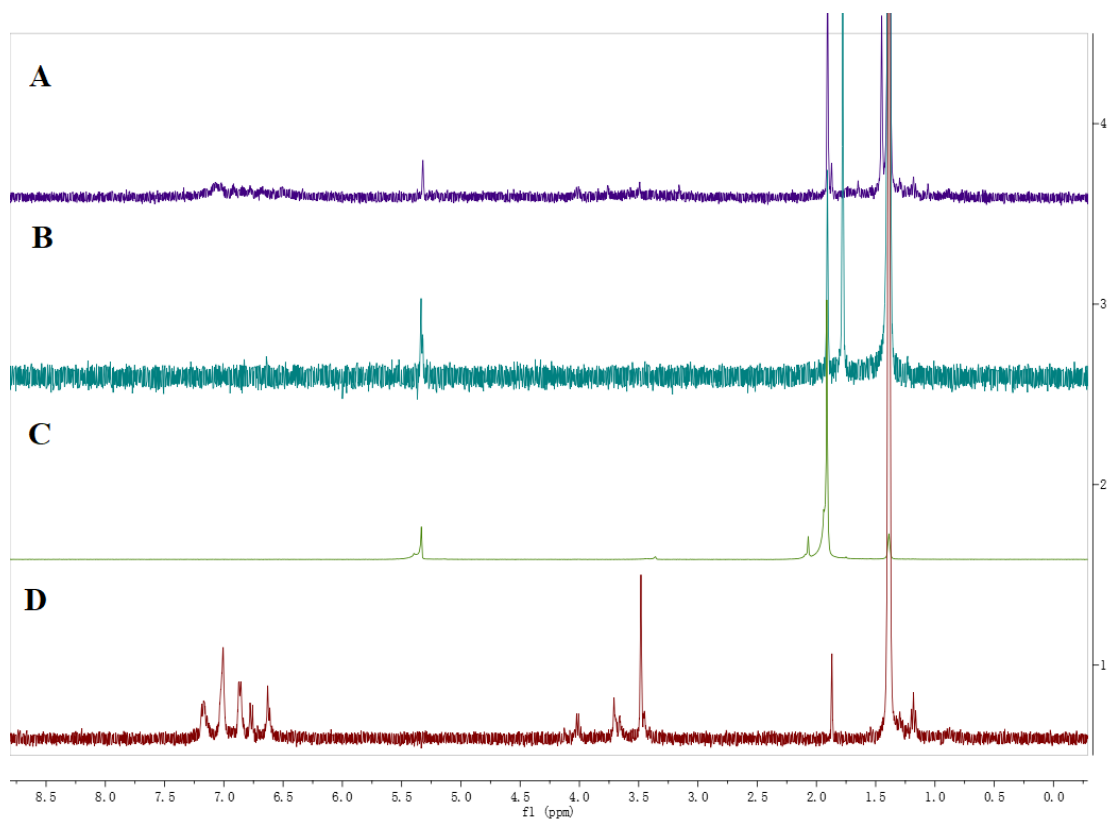


Figure S2 The formation of Zr(IV)-L1 complex evidenced by ^1H NMR analysis. (A) the resulting solution of the reaction between L1 and $\text{Zr}(\text{acac})_4$; (B) $\text{Zr}(\text{acac})_4$; (C) acetylacetonone (Hacac); (D) L1.

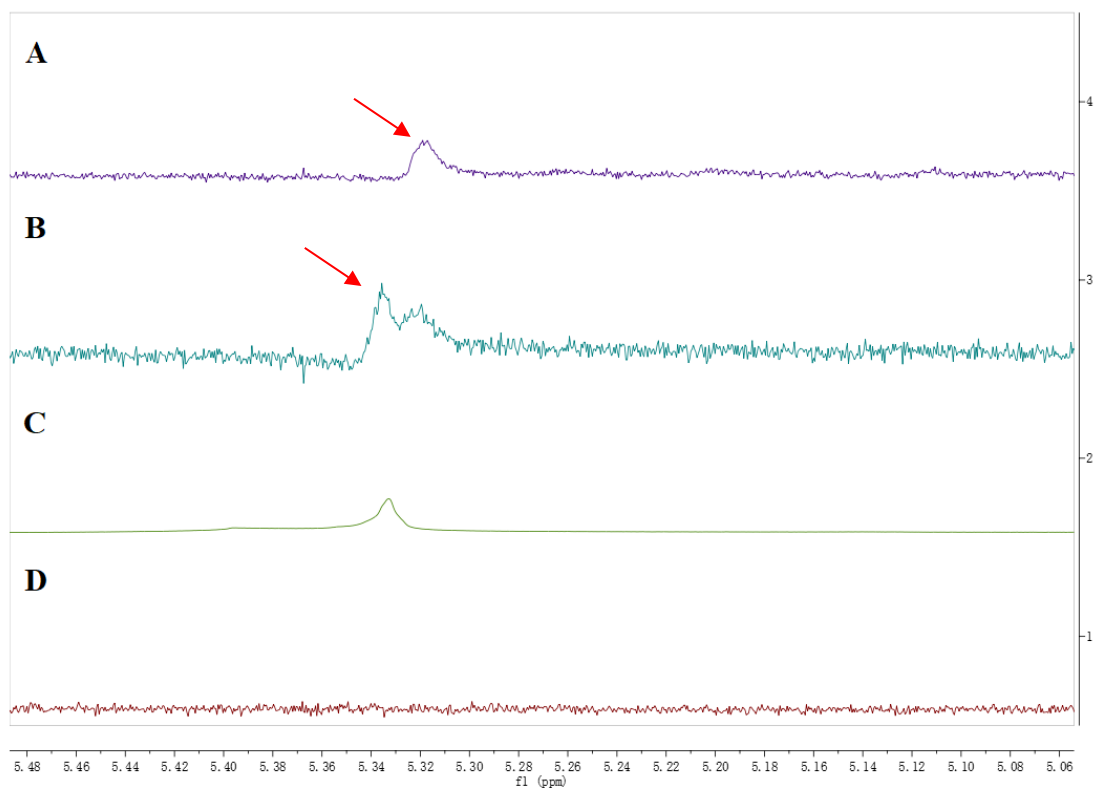


Figure S3 The enlarged spectra of Figure S2 scaled from δ 5.48 to 5.06 ppm. (A) the resulting solution of the reaction between L1 and Zr(acac)₄; (B) Zr(acac)₄; (C) acetylacetone (Hacac); (D) L1.

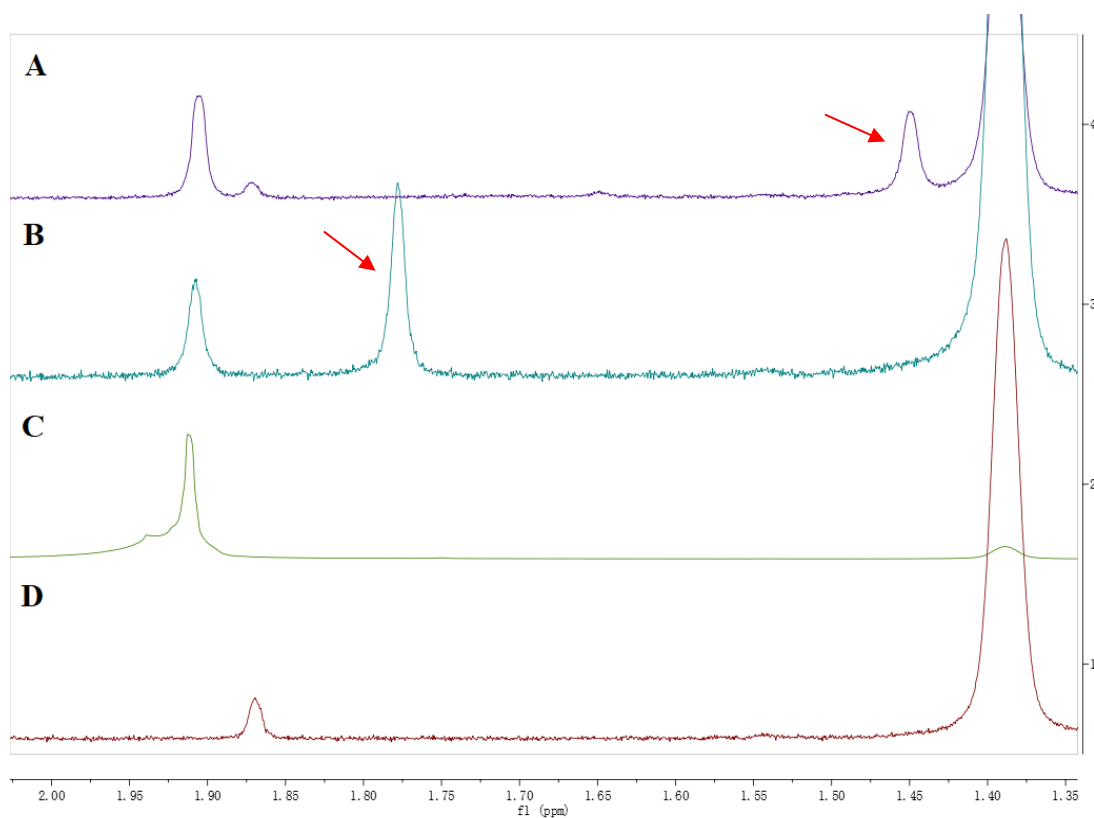


Figure S4 The enlarged spectra of Figure S2 scaled from δ 2.00 to 1.35 ppm. (A) the resulting solution of the reaction between L1 and Zr(acac)₄; (B) Zr(acac)₄; (C) acetylacetone (Hacac); (D) L1.

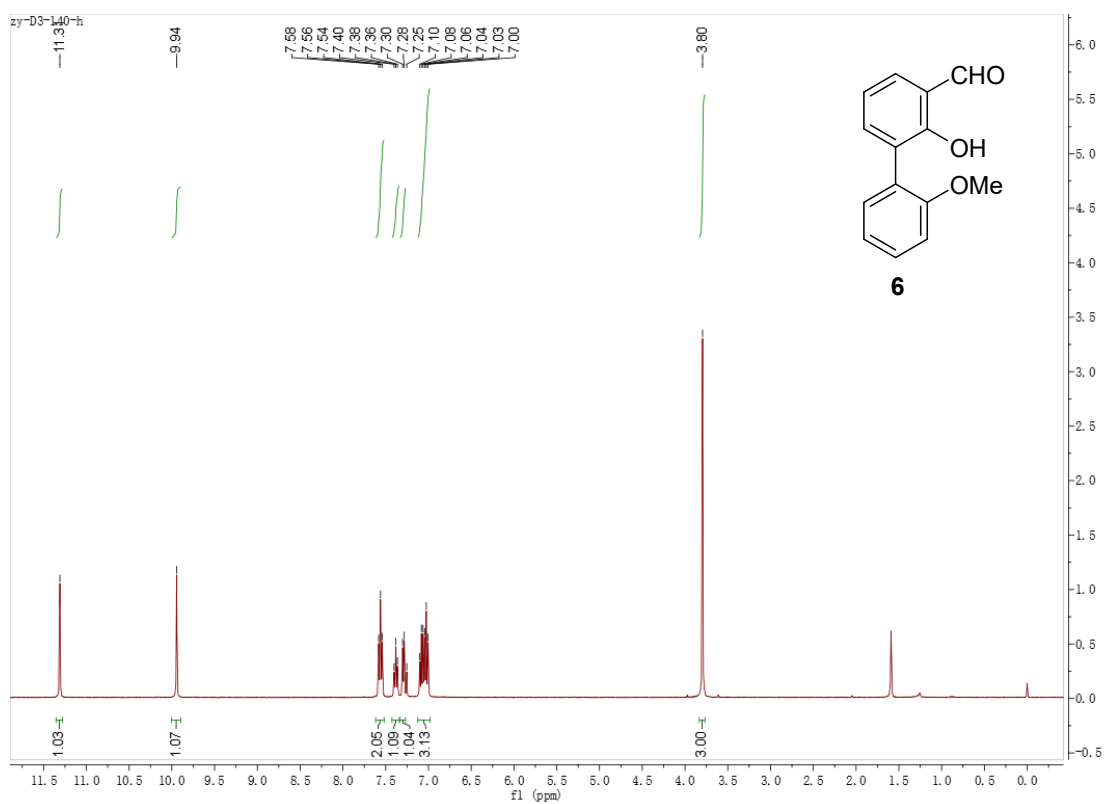
9. References

- [1] H.-L. Mu, W.-P. Ye, D.-P. Song, Y.-S. Li, *Organometallics*, 2010, **29**, 6282.
- [2] J. Chen, H.-Y. Gu, X.-Y. Zhu, W.-W. Nam and B. Wang, *Adv. Synth. Catal.*, 2020, **362**, 2976.
- [3] F. Yang, J. Zhao, X. Tang, G. Zhou, W. Song, Q. Meng, *Org. Lett.*, 2017, **19**, 448.

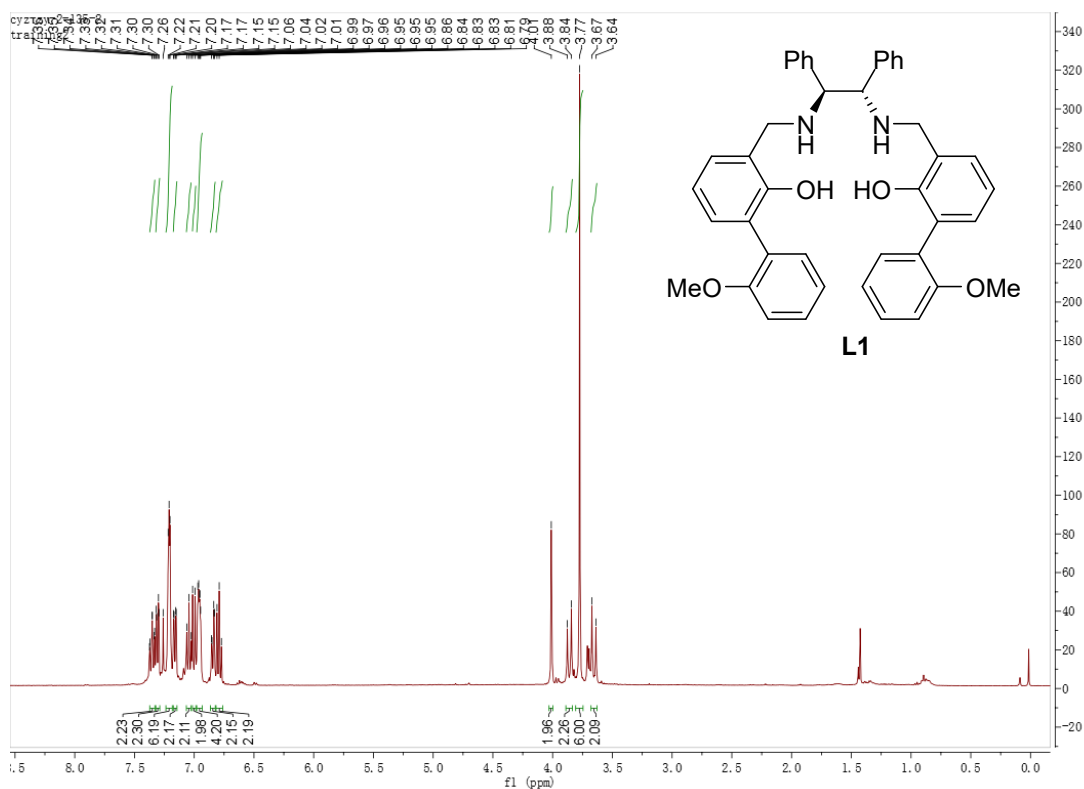
10. Copies of ^1H and ^{13}C NMR spectra

10.1 chiral ligand L1

^1H NMR of compound 6

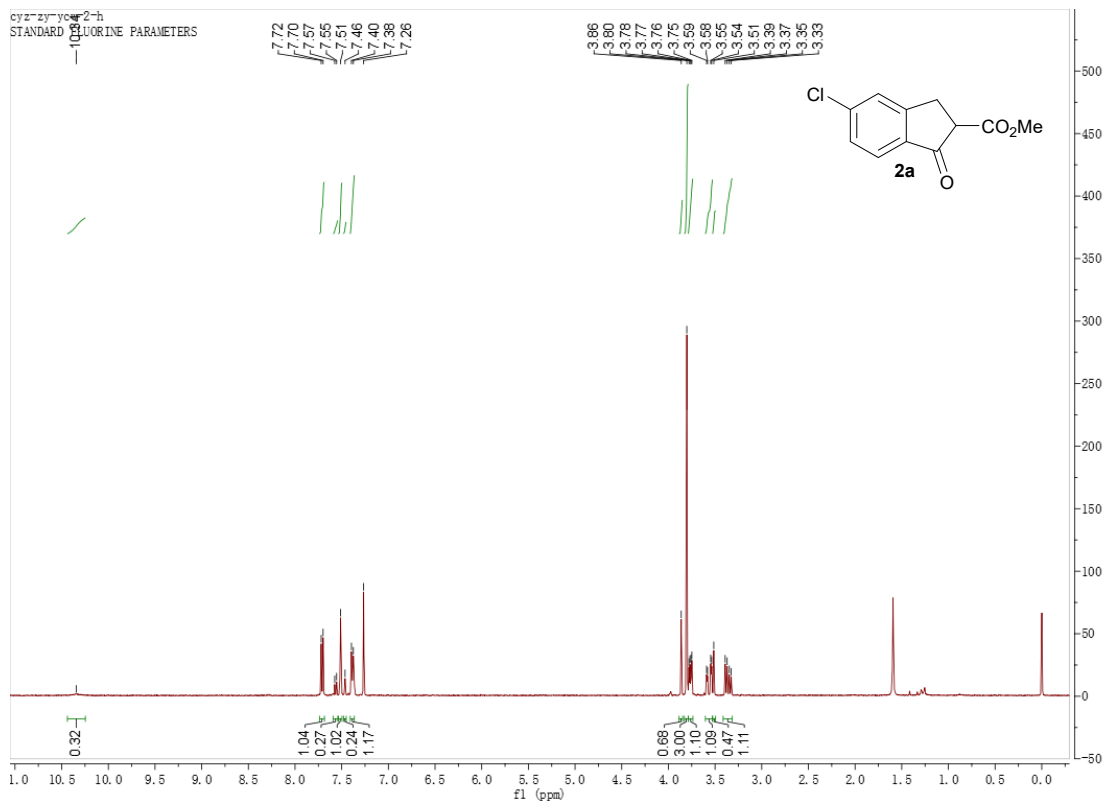


^1H NMR of compound L1

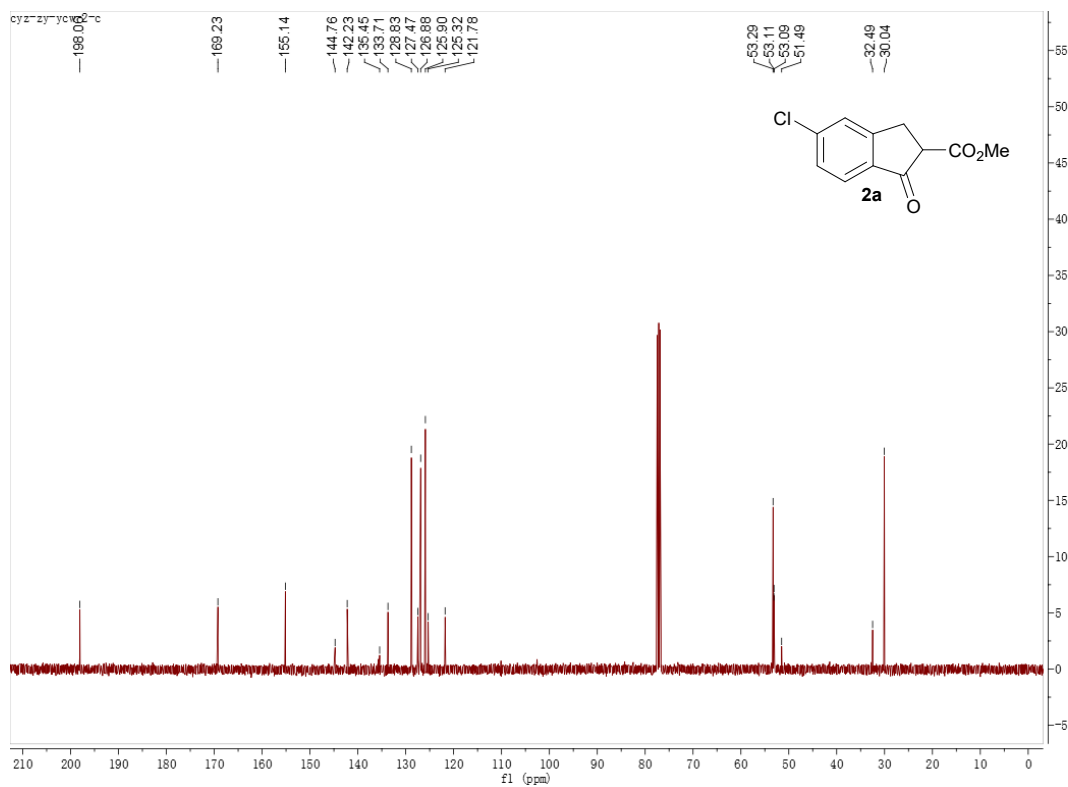


10.2 β -keto esters

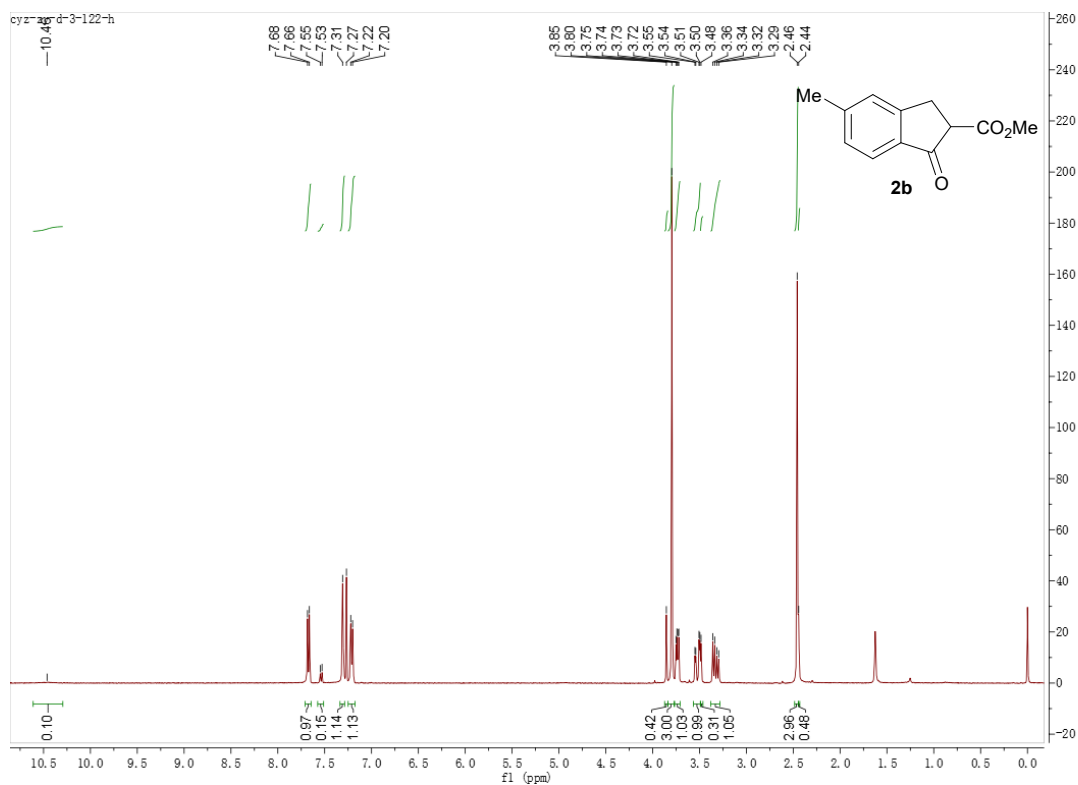
^1H NMR of compound 2a



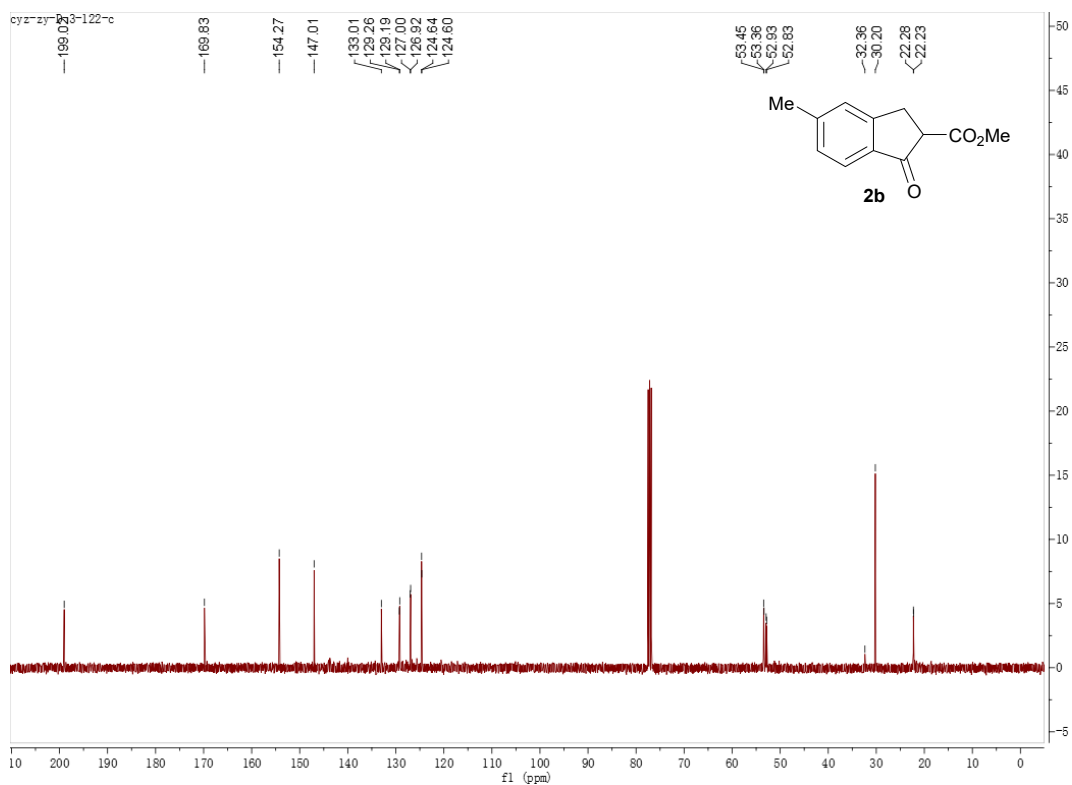
^{13}C NMR of compound 2a



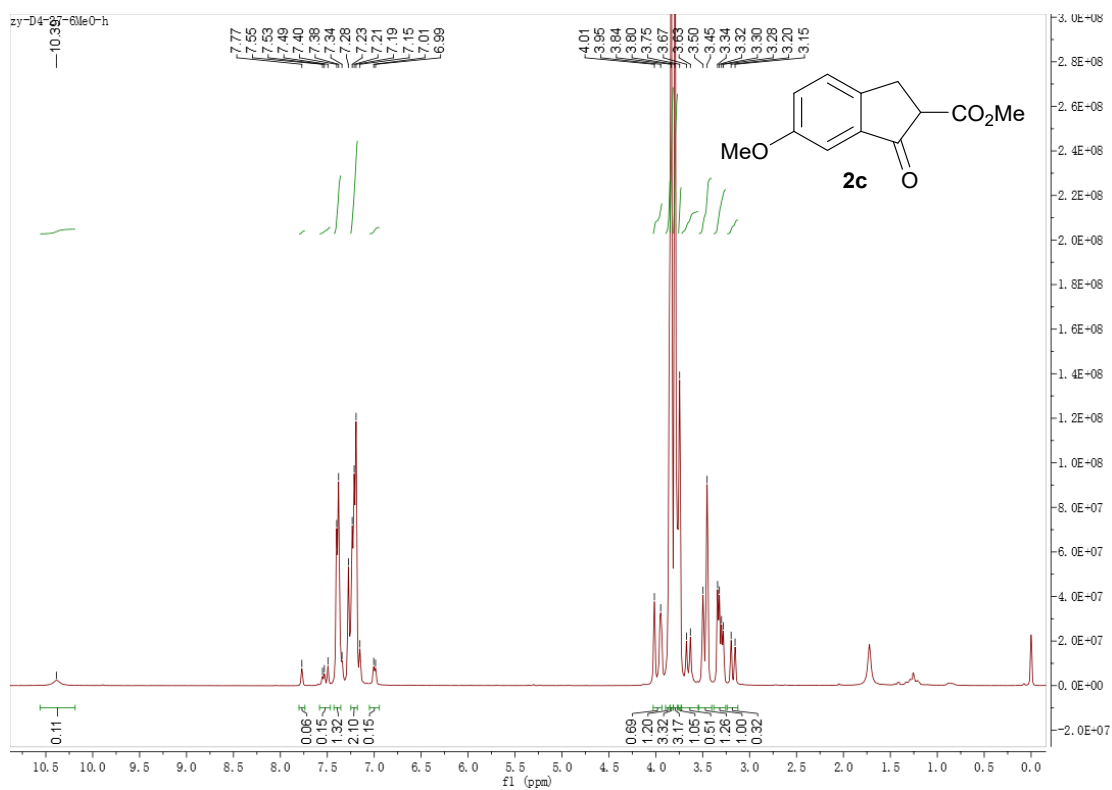
¹H NMR of compound 2b



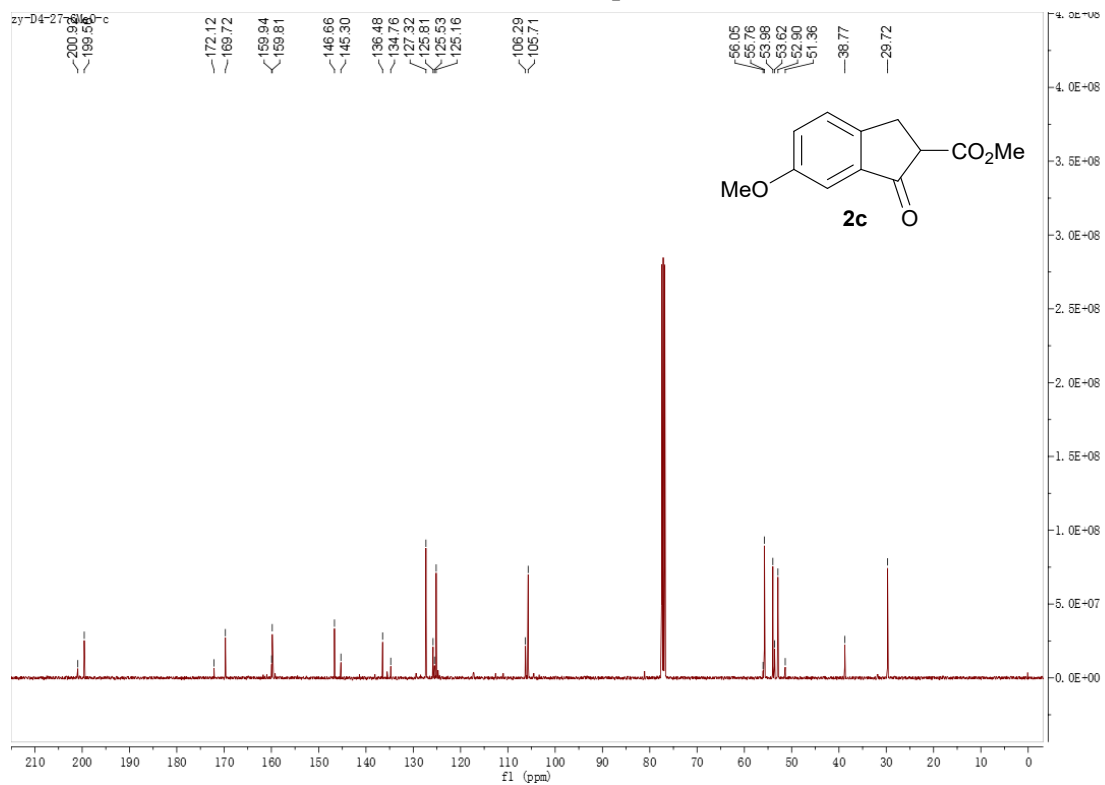
¹³C NMR of compound 2b



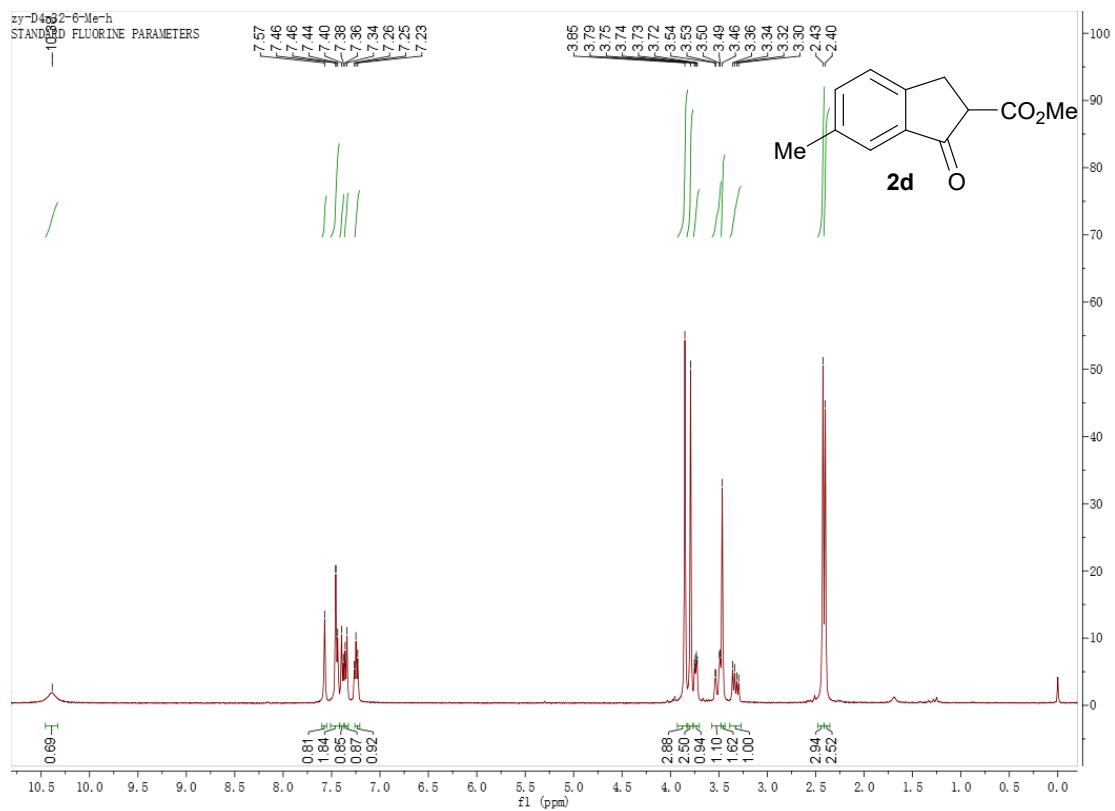
¹H NMR of compound 2c



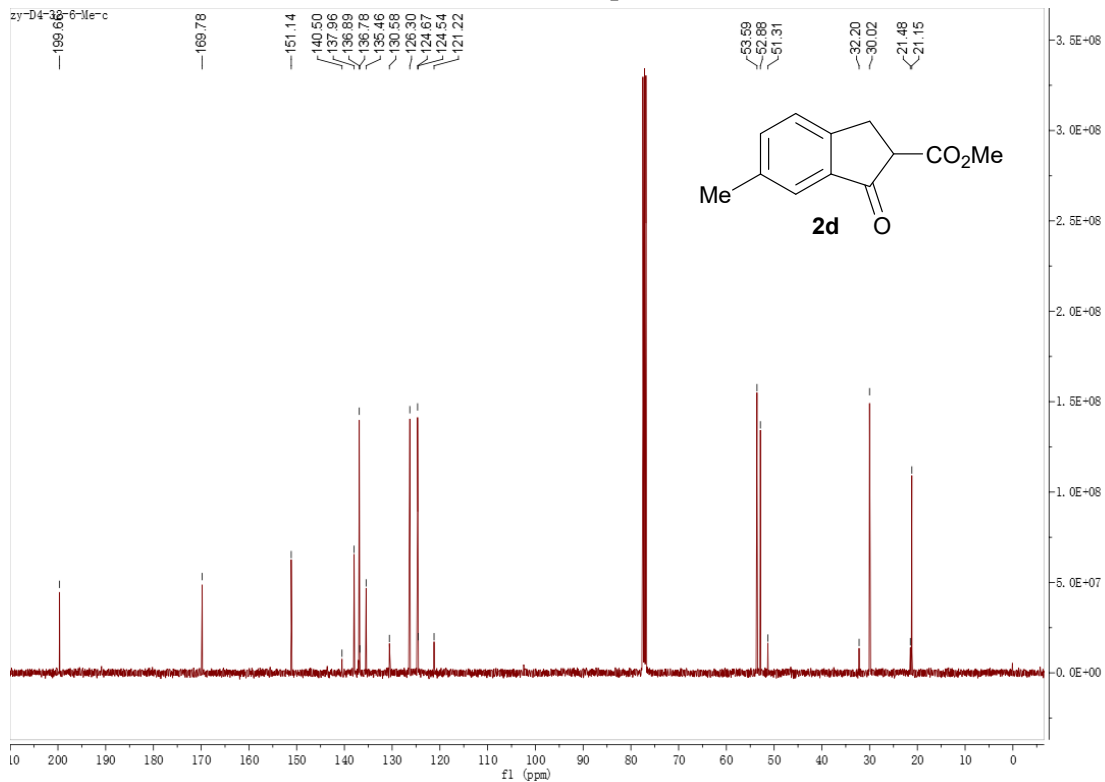
¹³C NMR of compound 2c



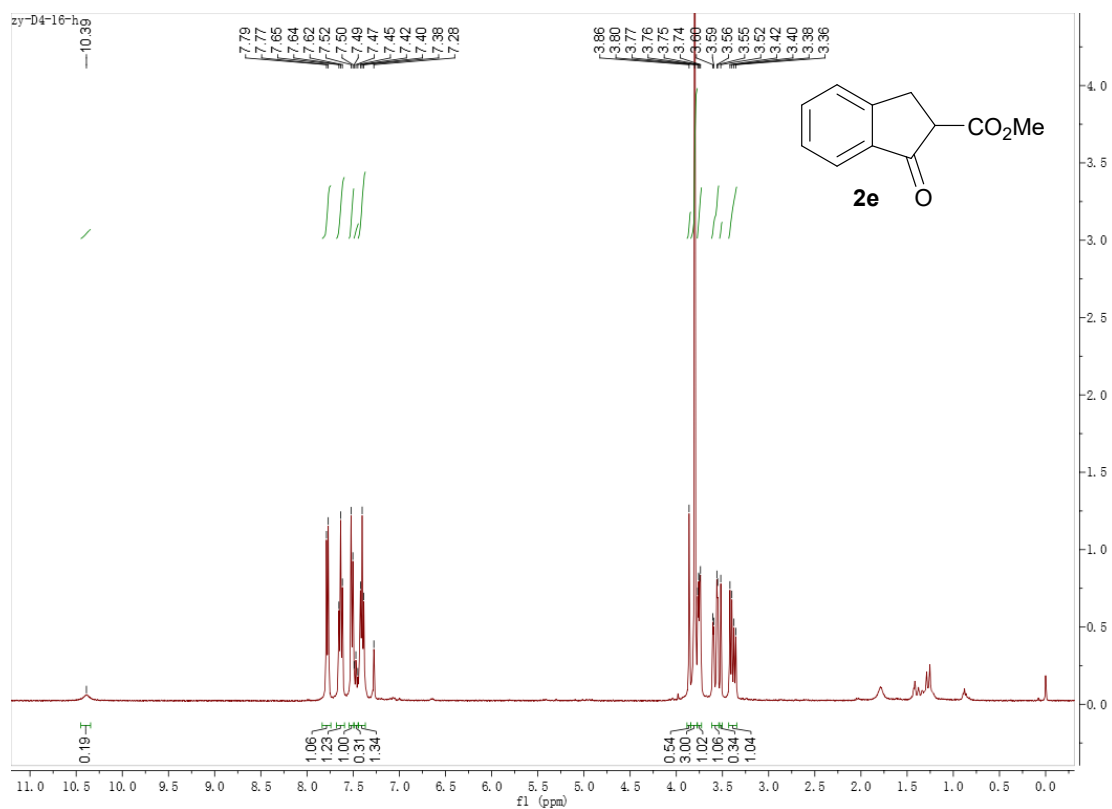
¹H NMR of compound 2d



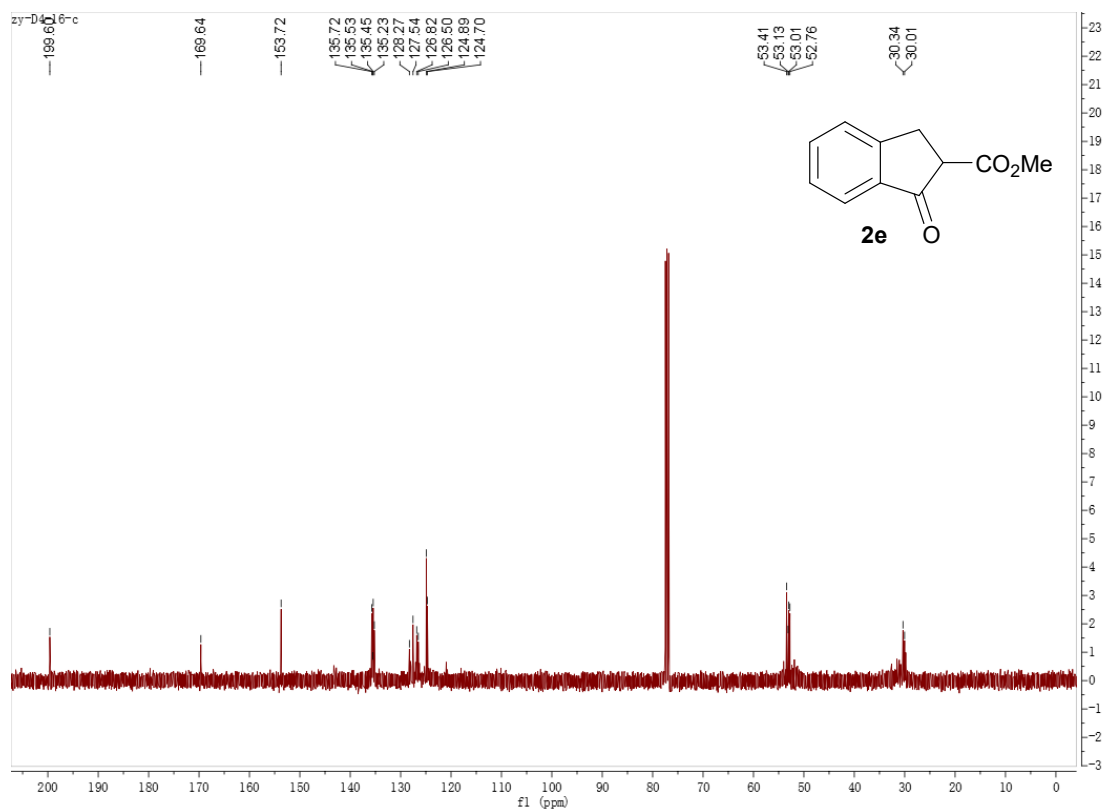
¹³C NMR of compound 2d



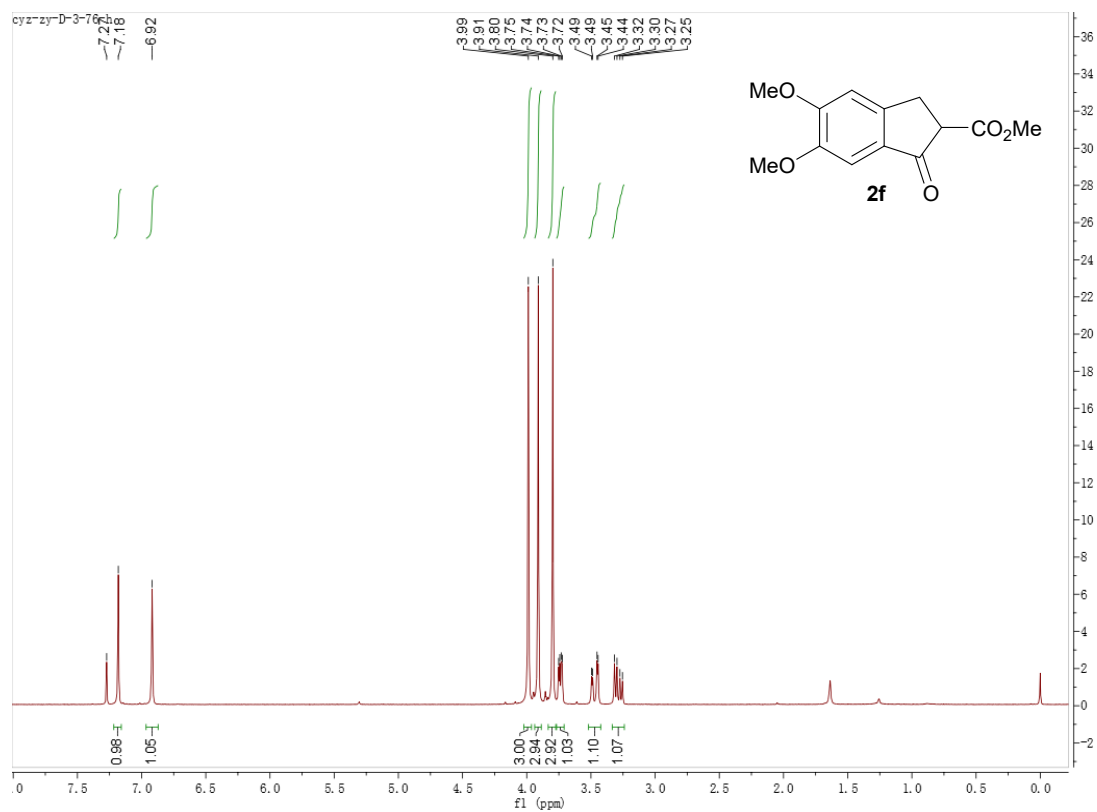
¹H NMR of compound 2e



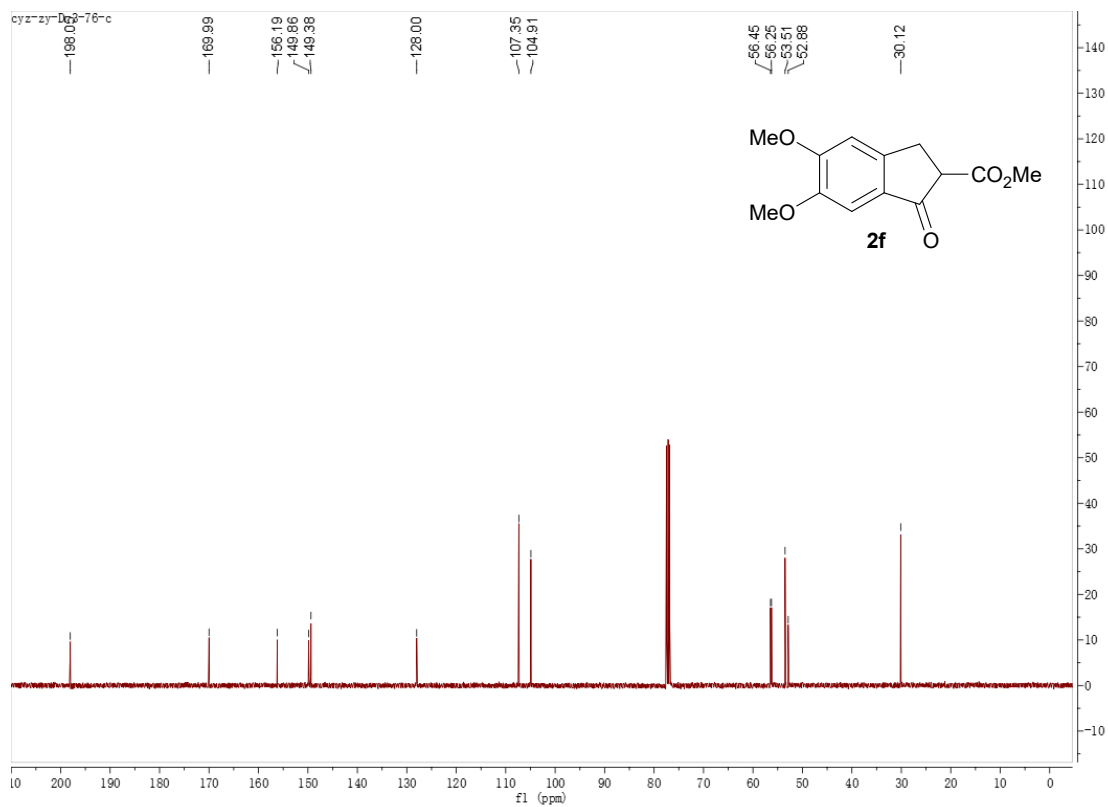
¹³C NMR of compound 2e



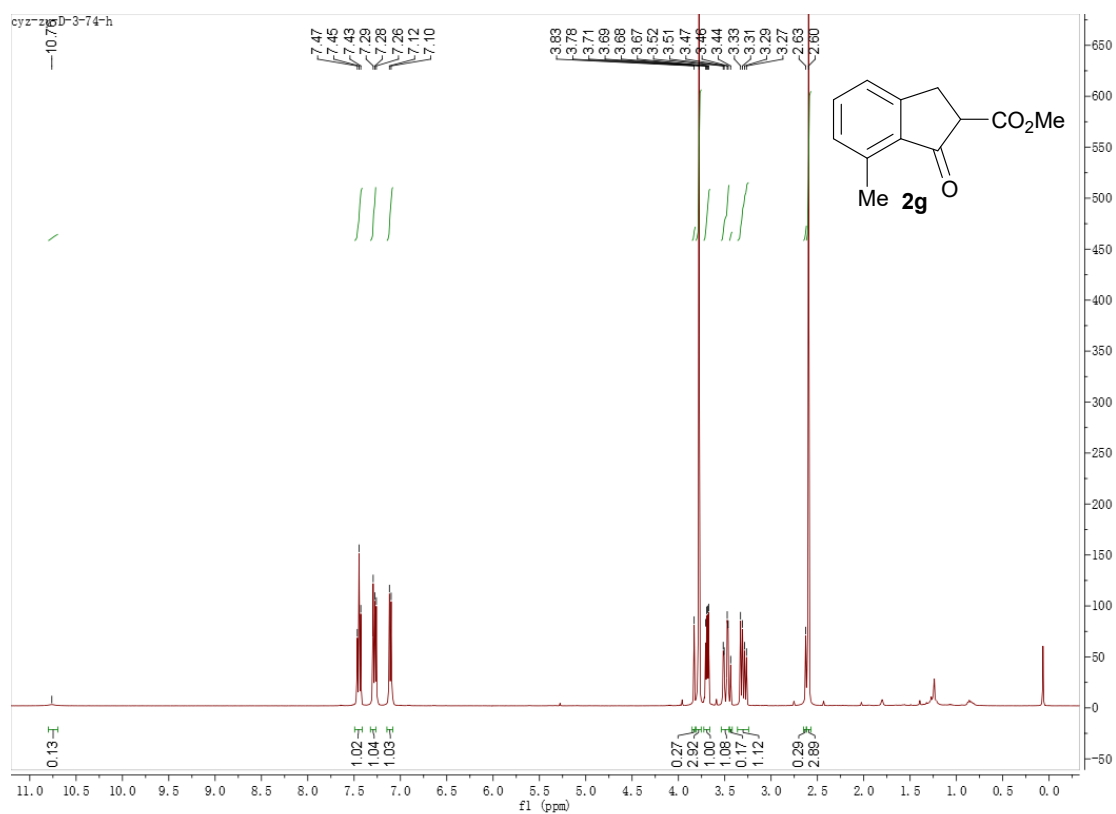
¹H NMR of compound 2f



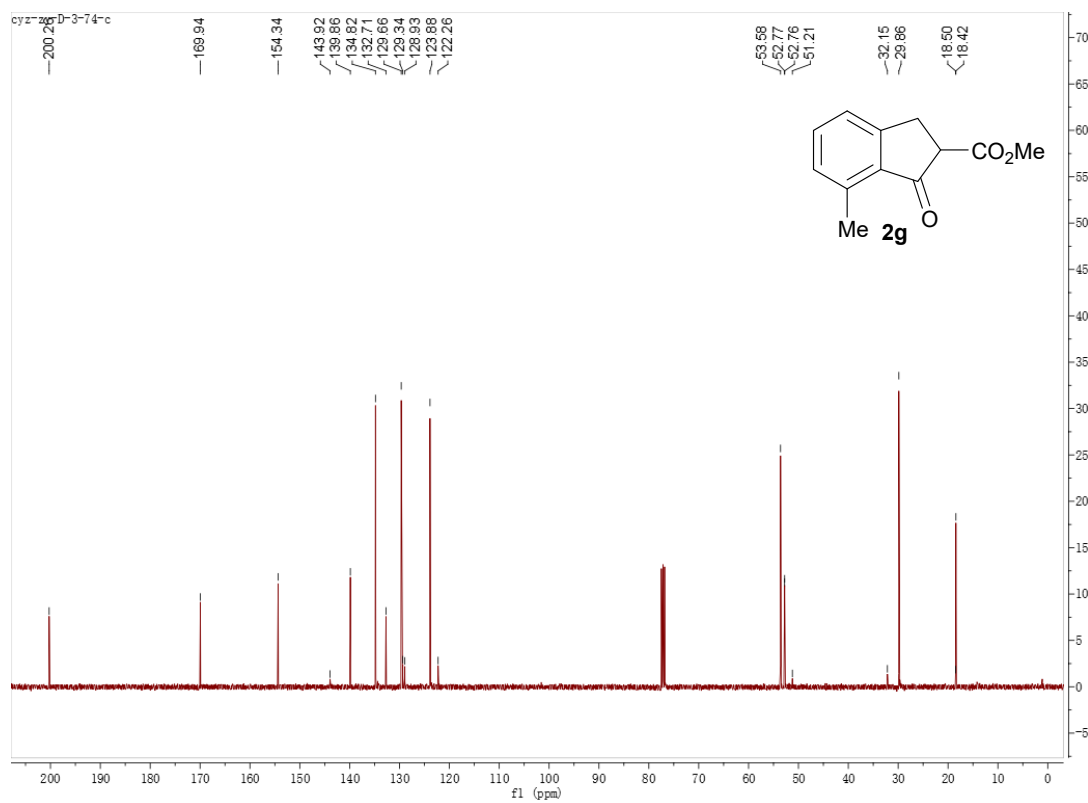
¹³C NMR of compound 2f



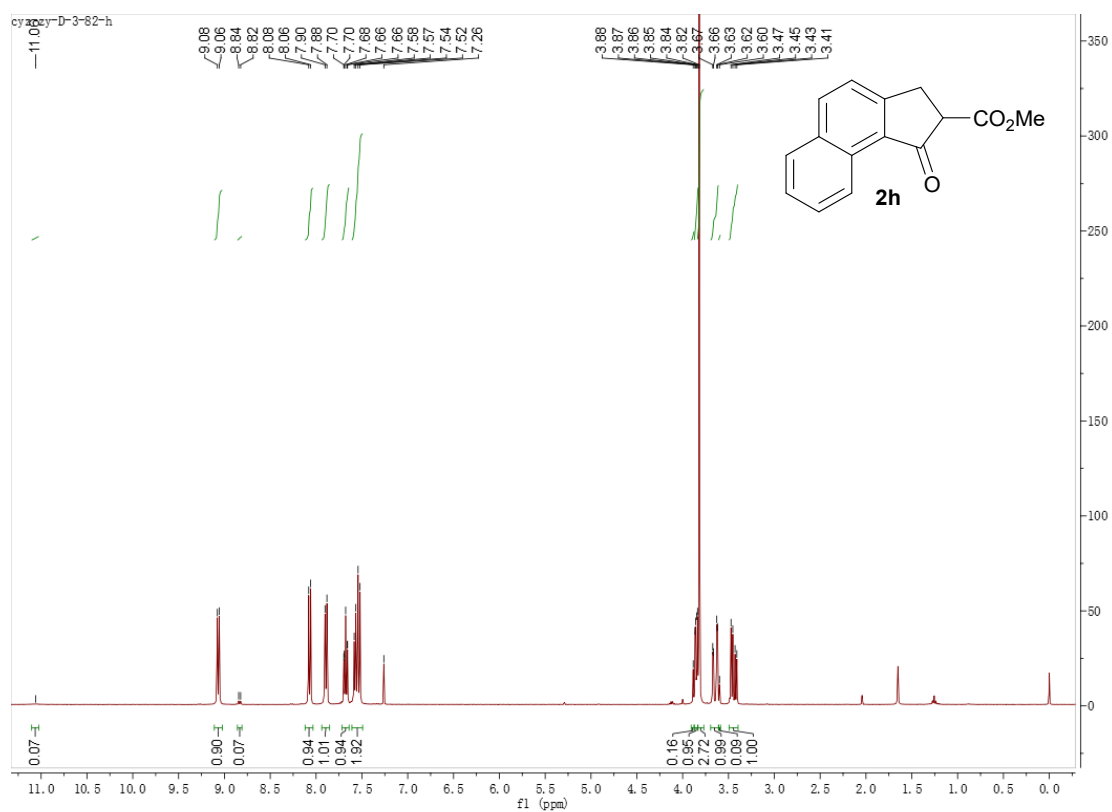
¹H NMR of compound 2g



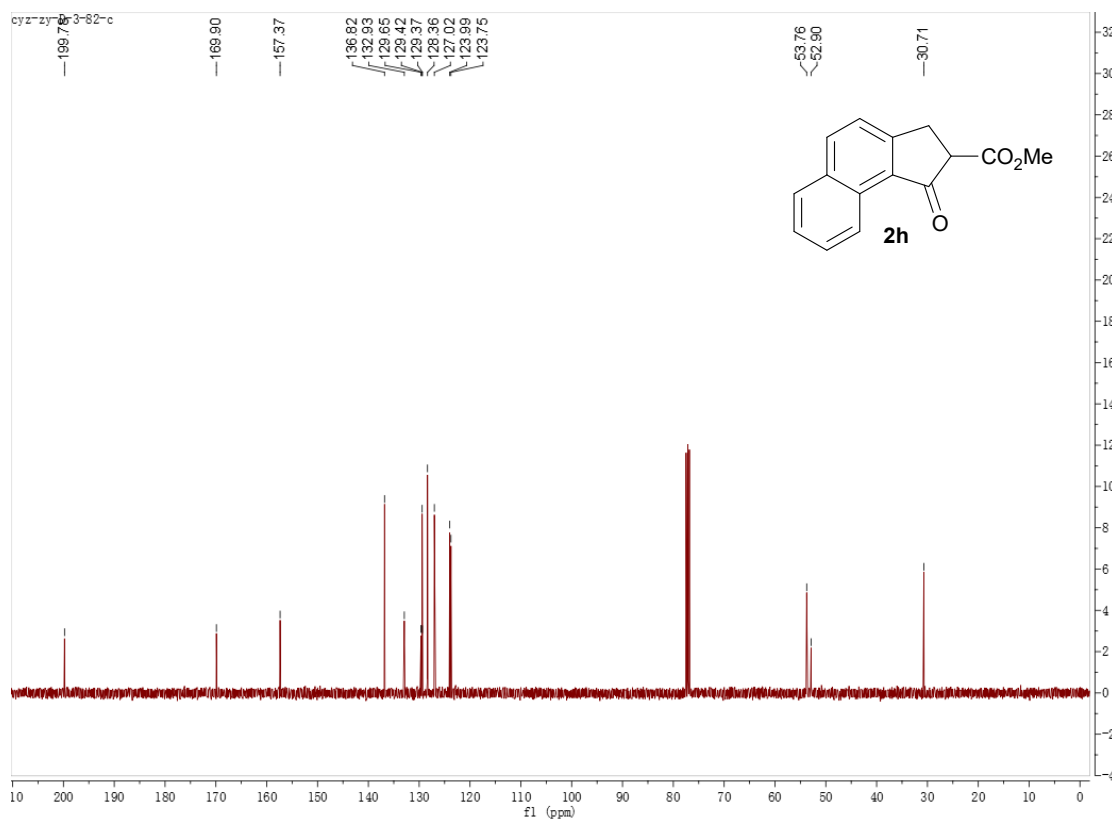
¹³C NMR of compound 2g



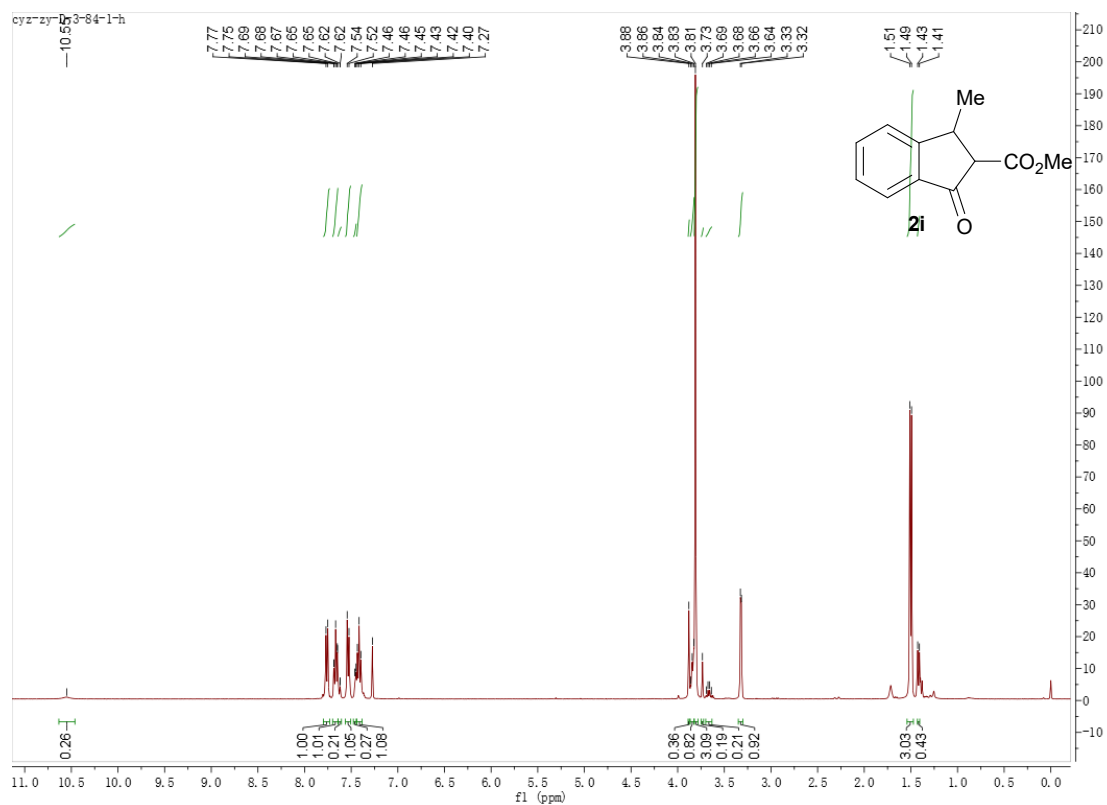
¹H NMR of compound 2h



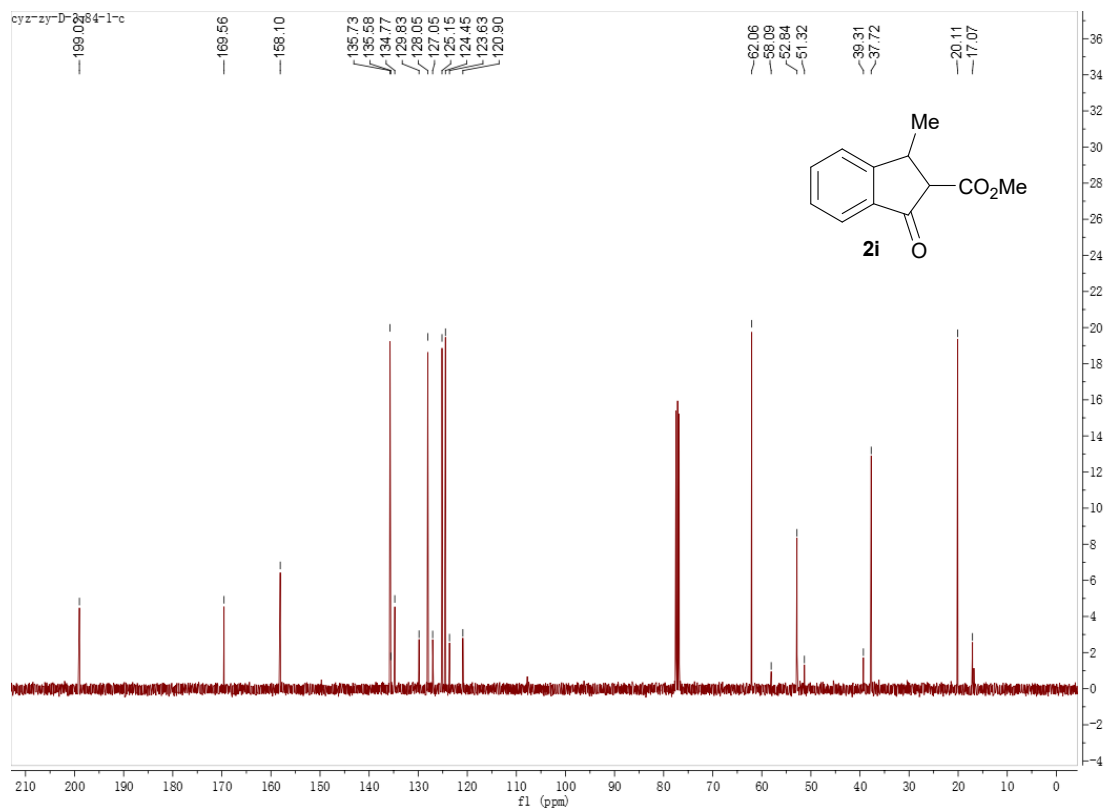
¹³C NMR of compound 2h



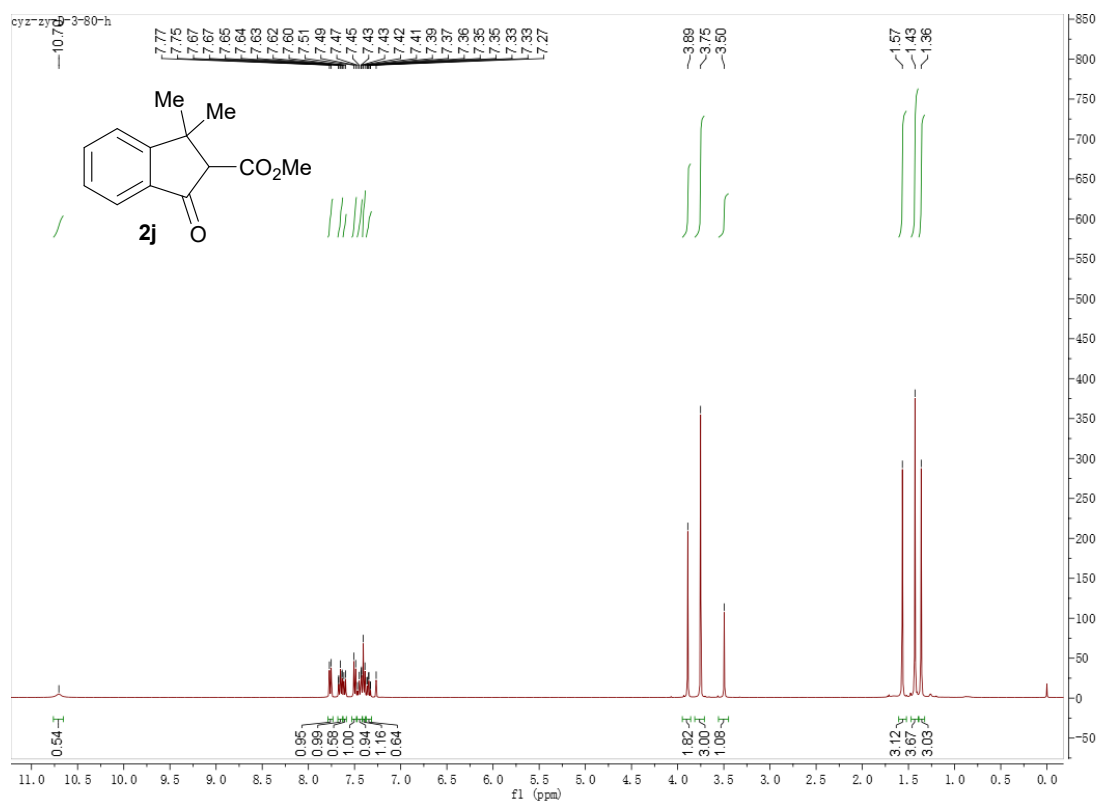
¹H NMR of compound 2i



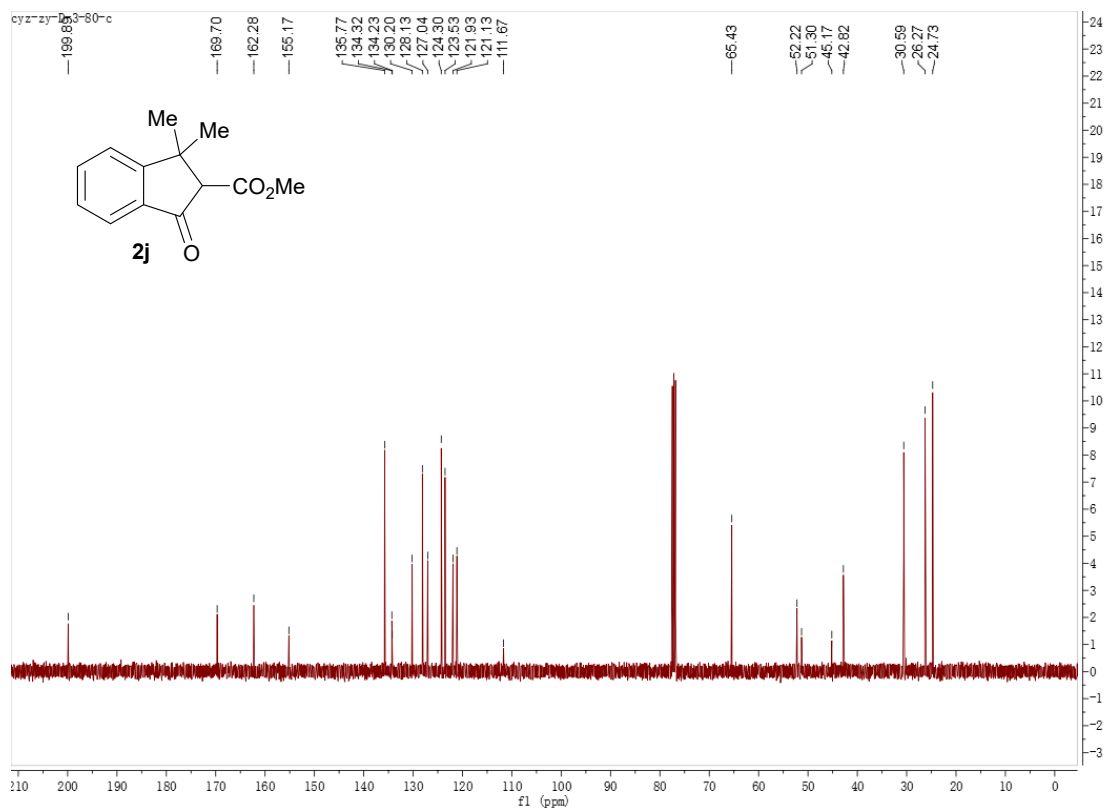
¹³C NMR of compound 2i



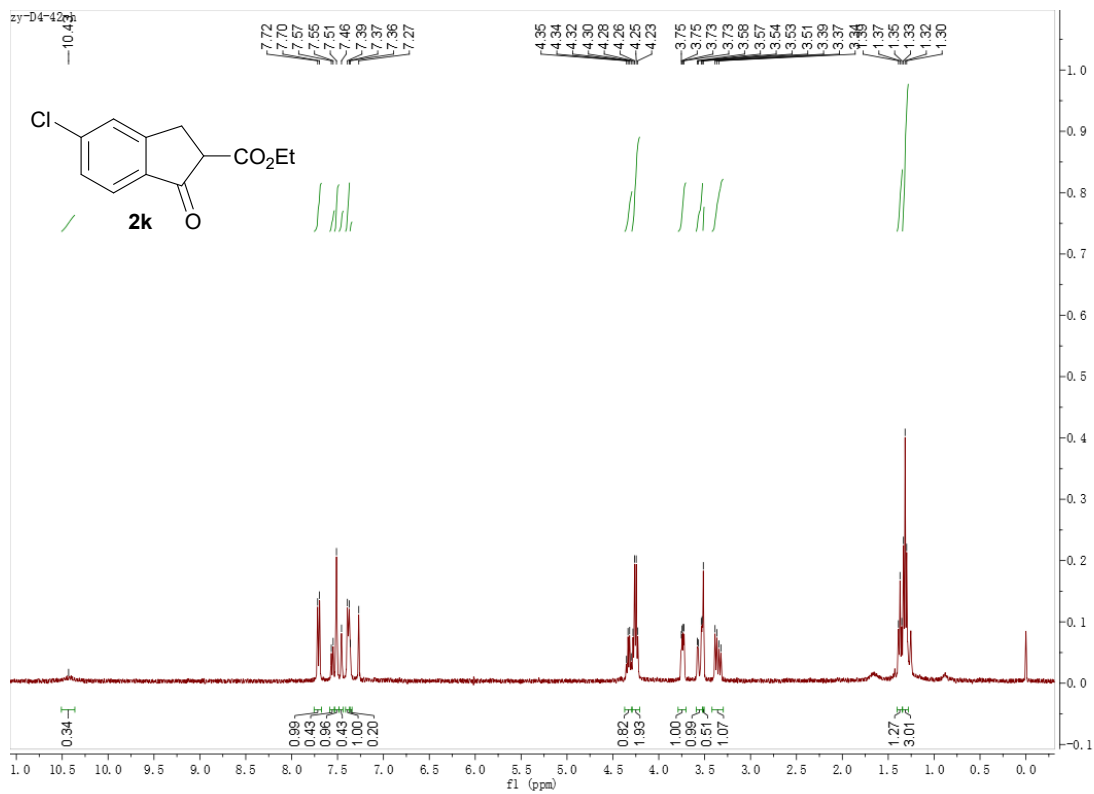
¹H NMR of compound 2j



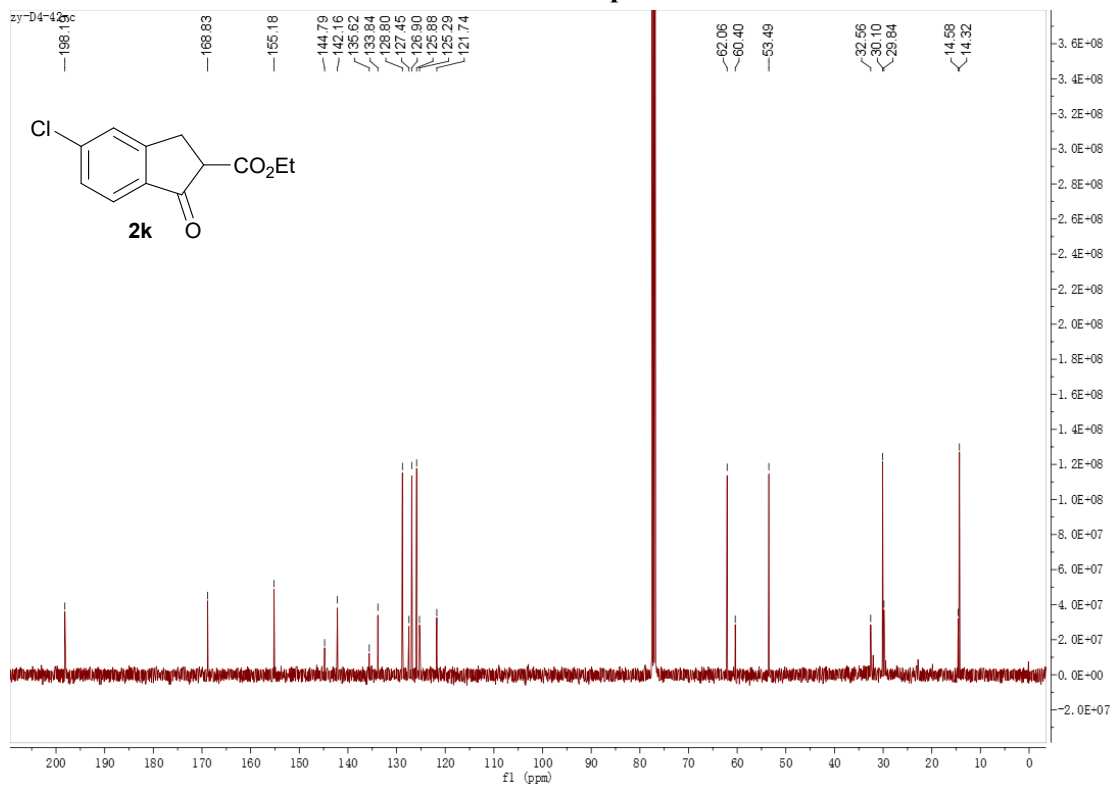
¹³C NMR of compound 2j



¹H NMR of compound 2k

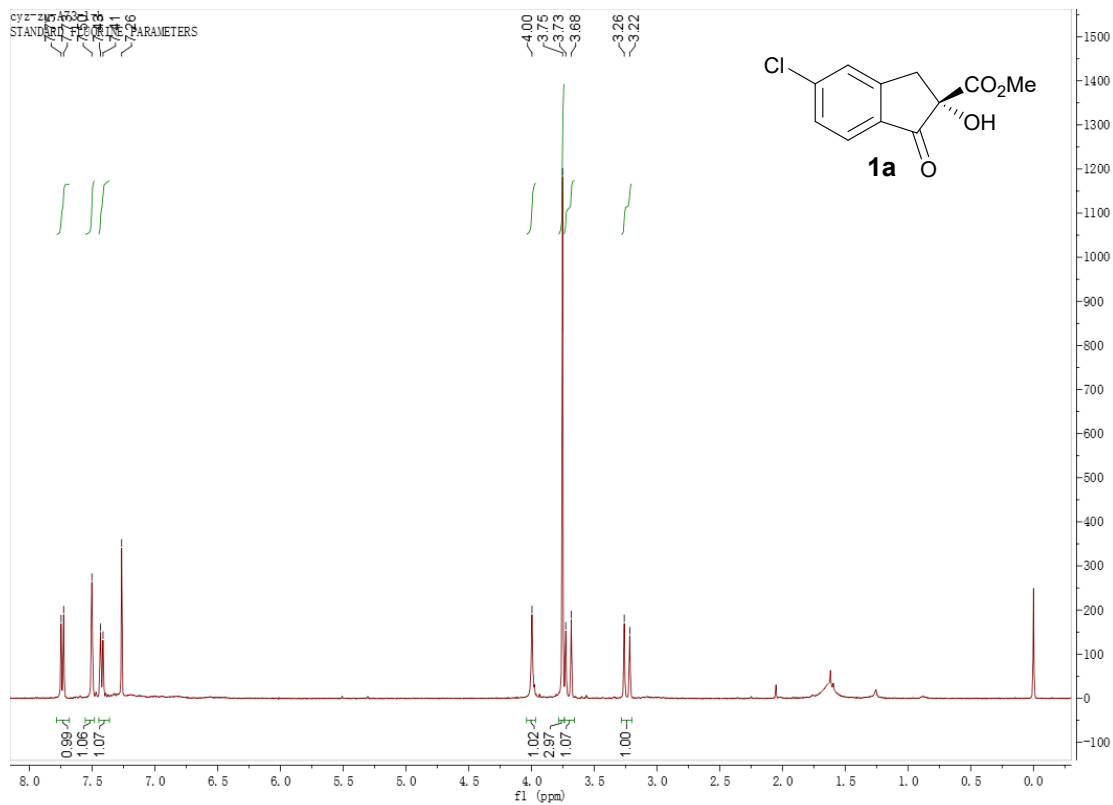


¹³C NMR of compound 2k

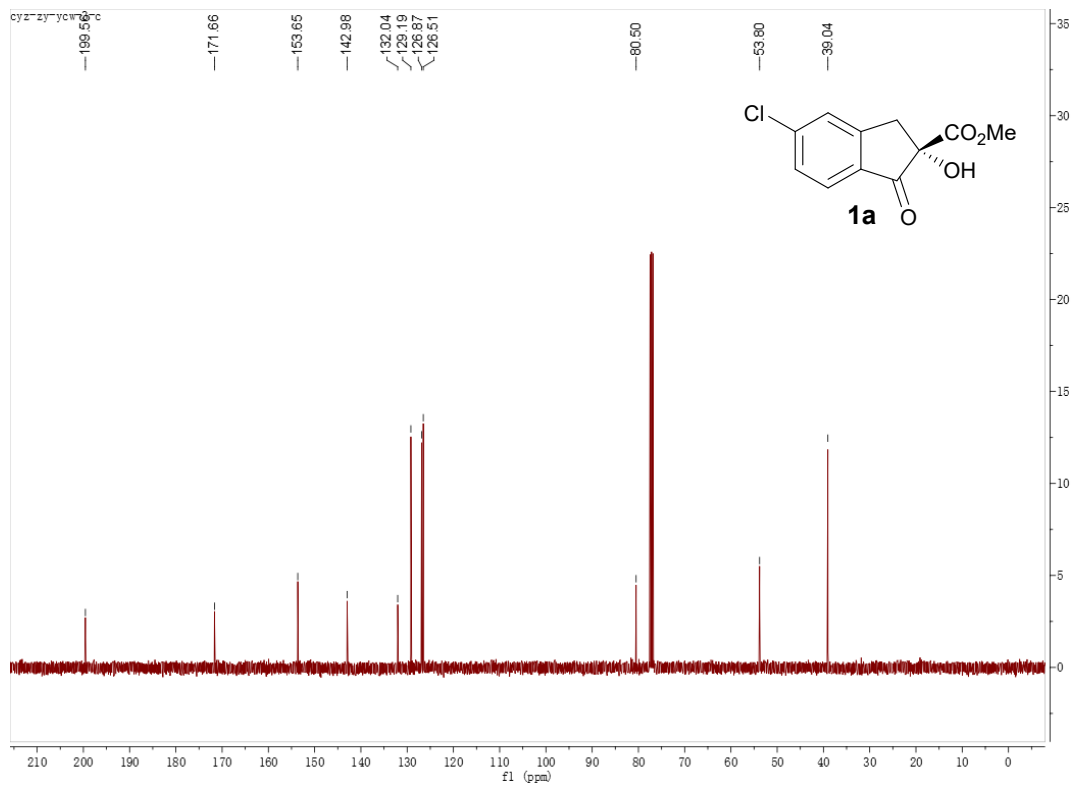


10.3 The products of enantioselective hydroxylation

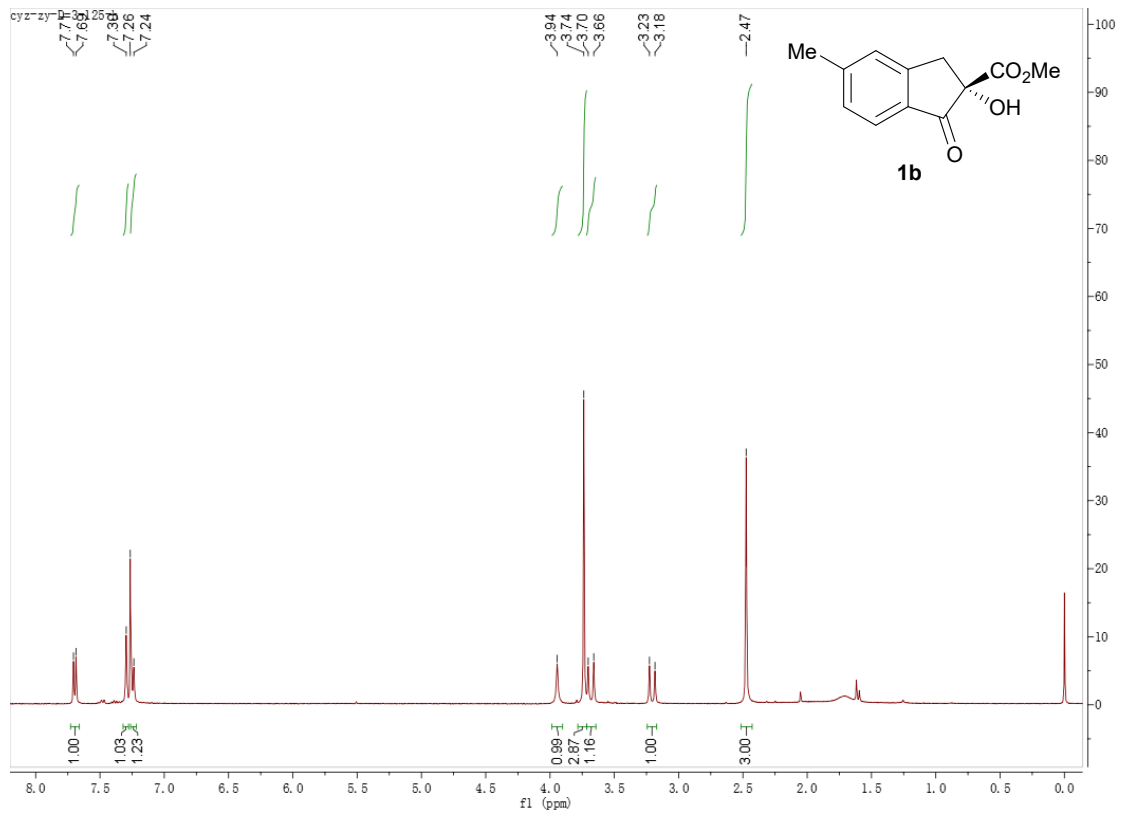
¹H NMR of compound 1a



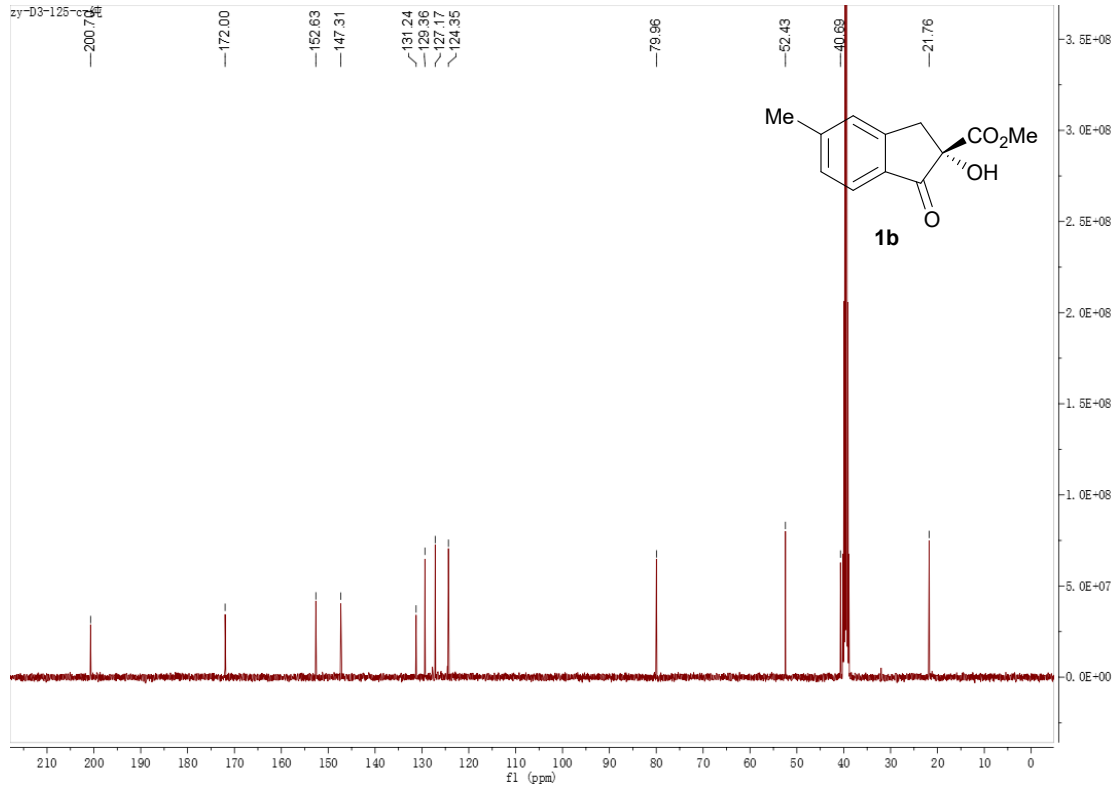
¹³C NMR of compound 1a



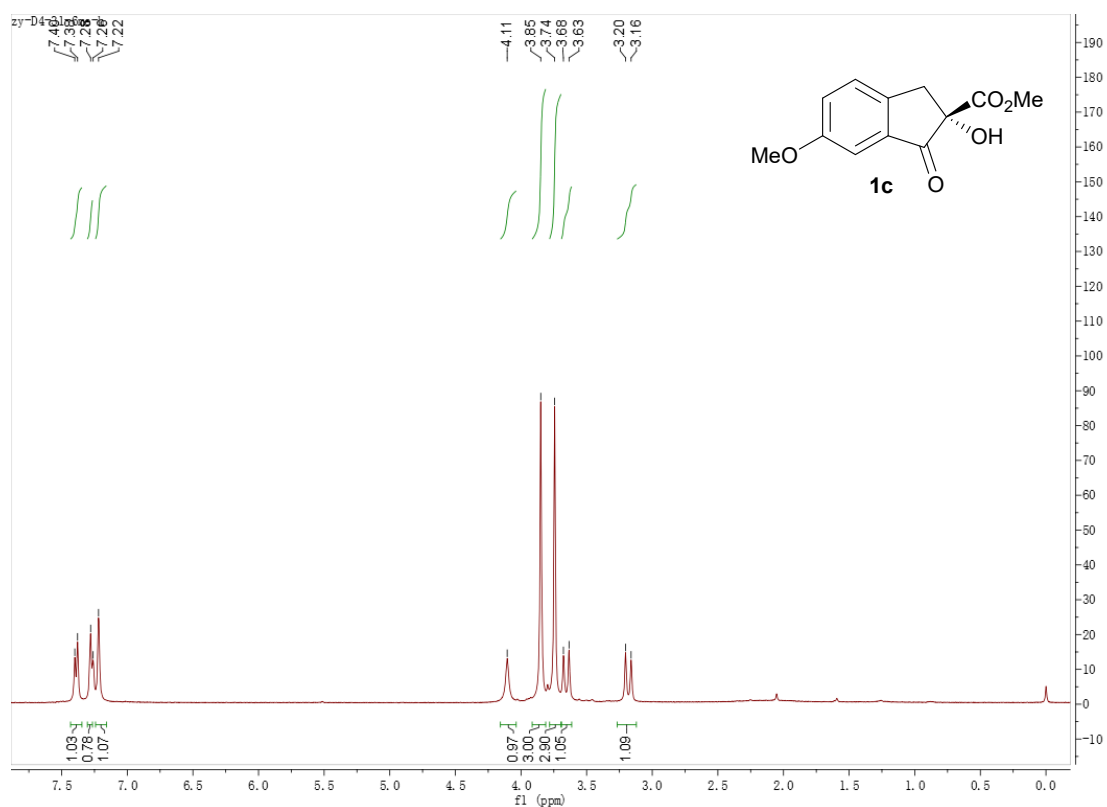
¹H NMR of compound 1b



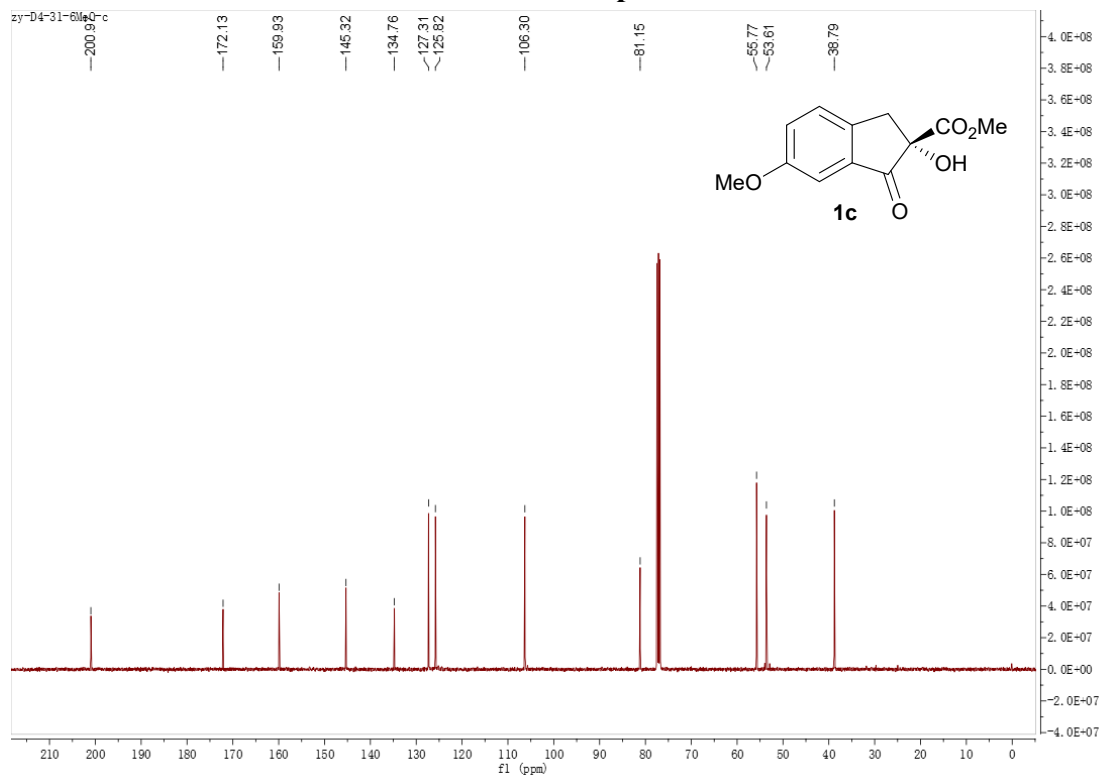
¹³C NMR of compound 1b



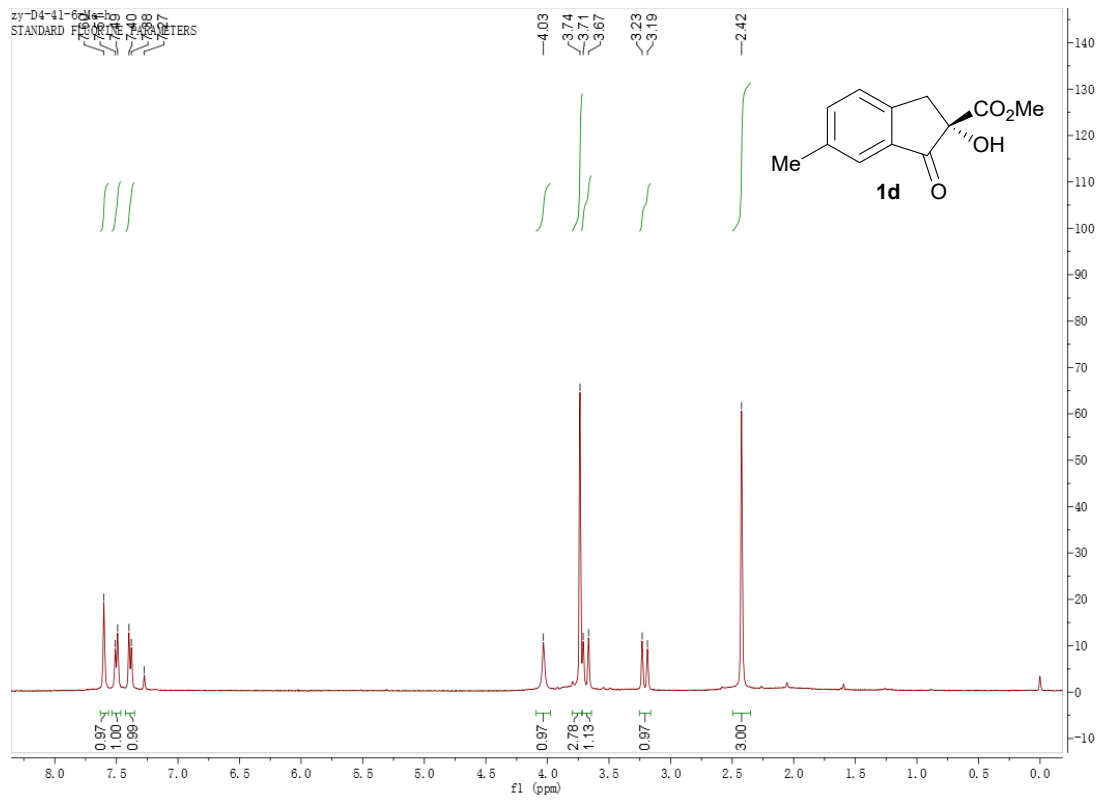
¹H NMR of compound 1c



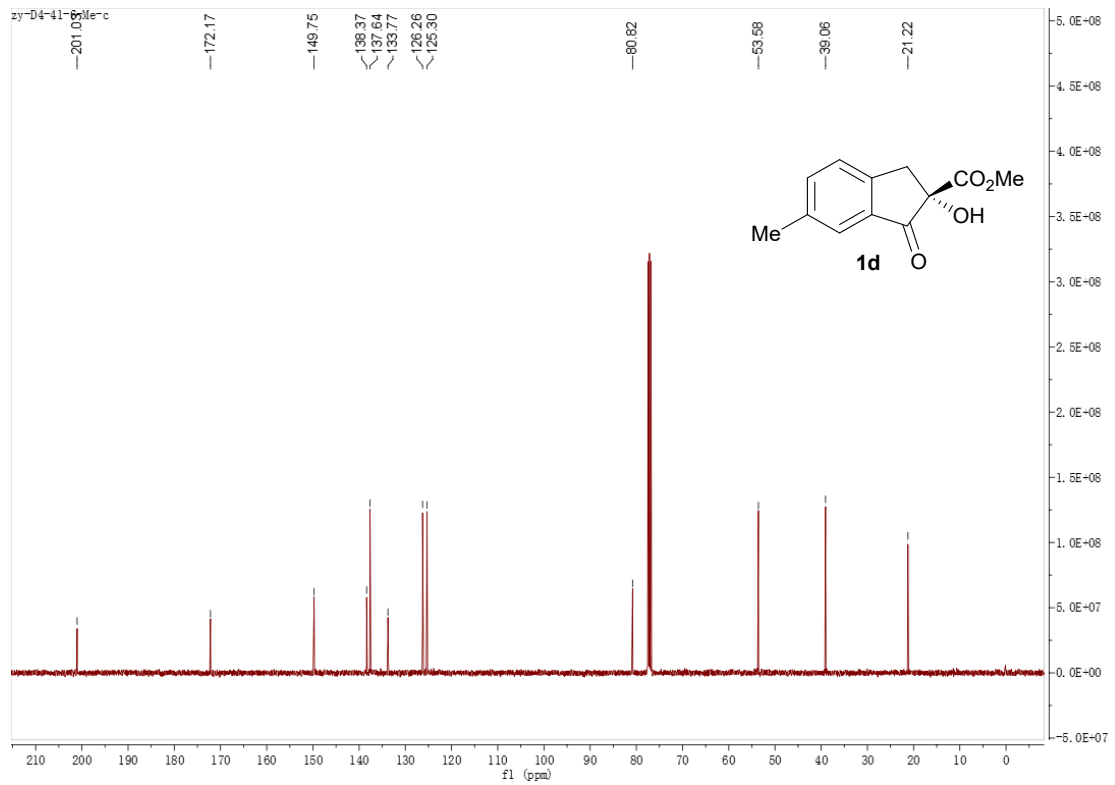
¹³C NMR of compound 1c



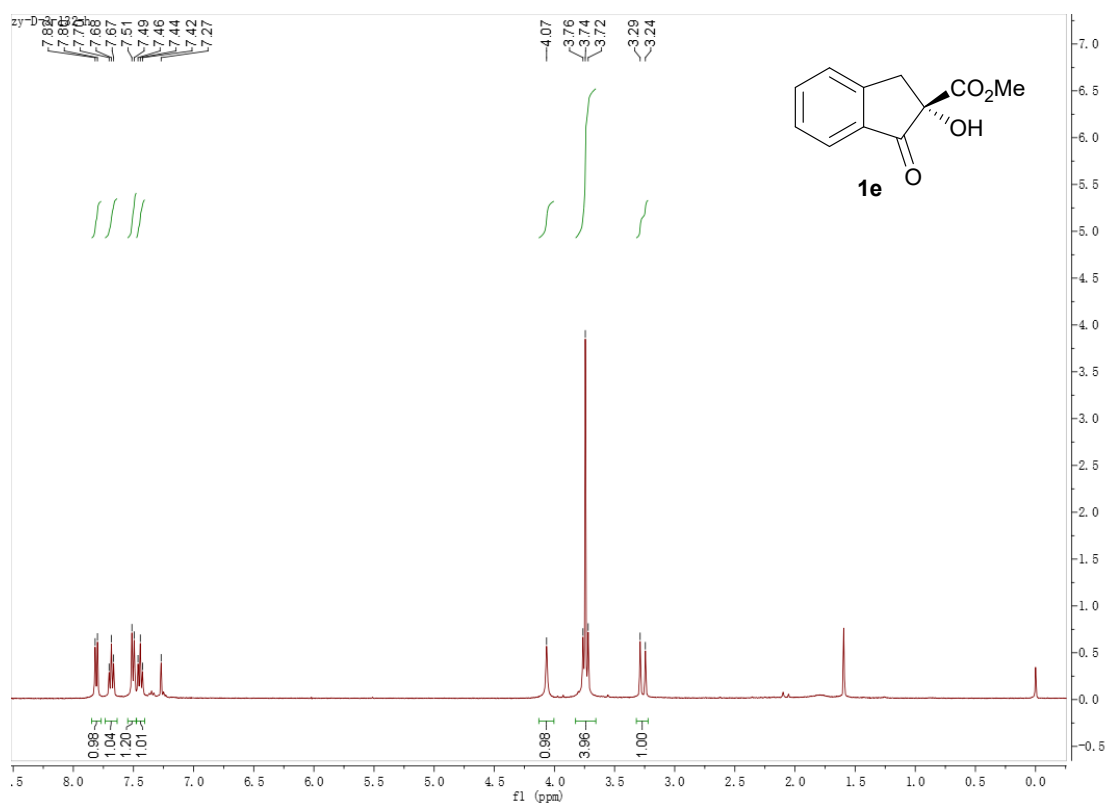
¹H NMR of compound 1d



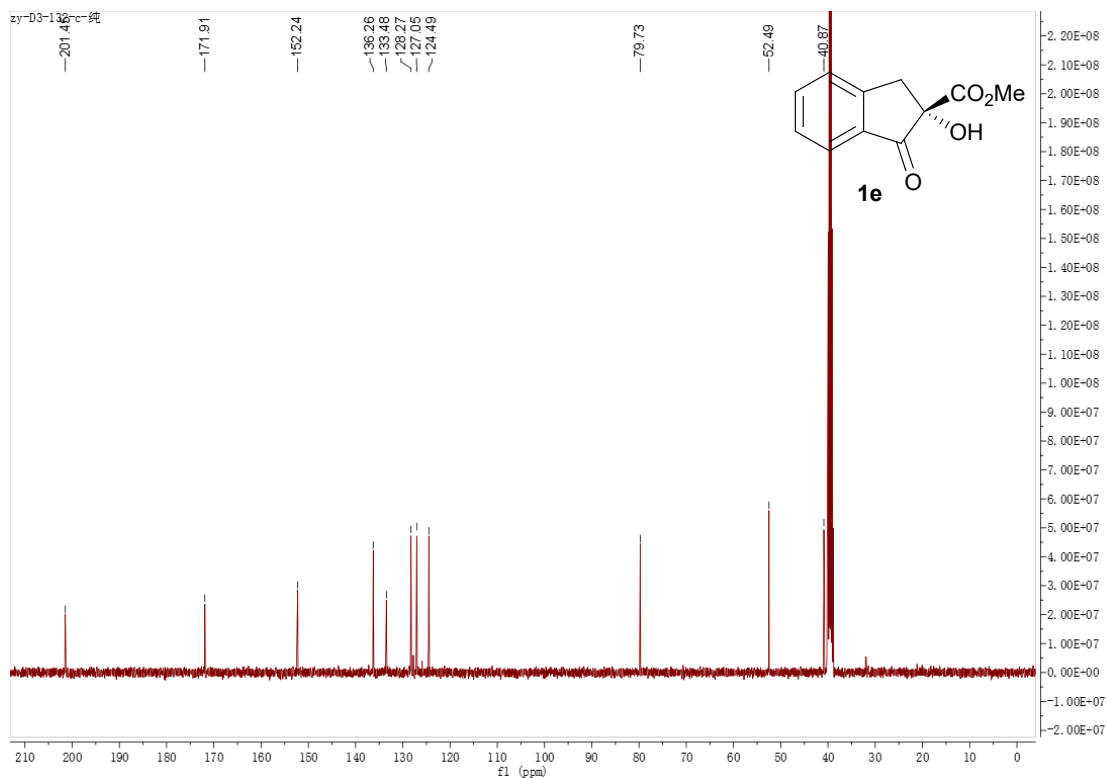
¹³C NMR of compound 1d



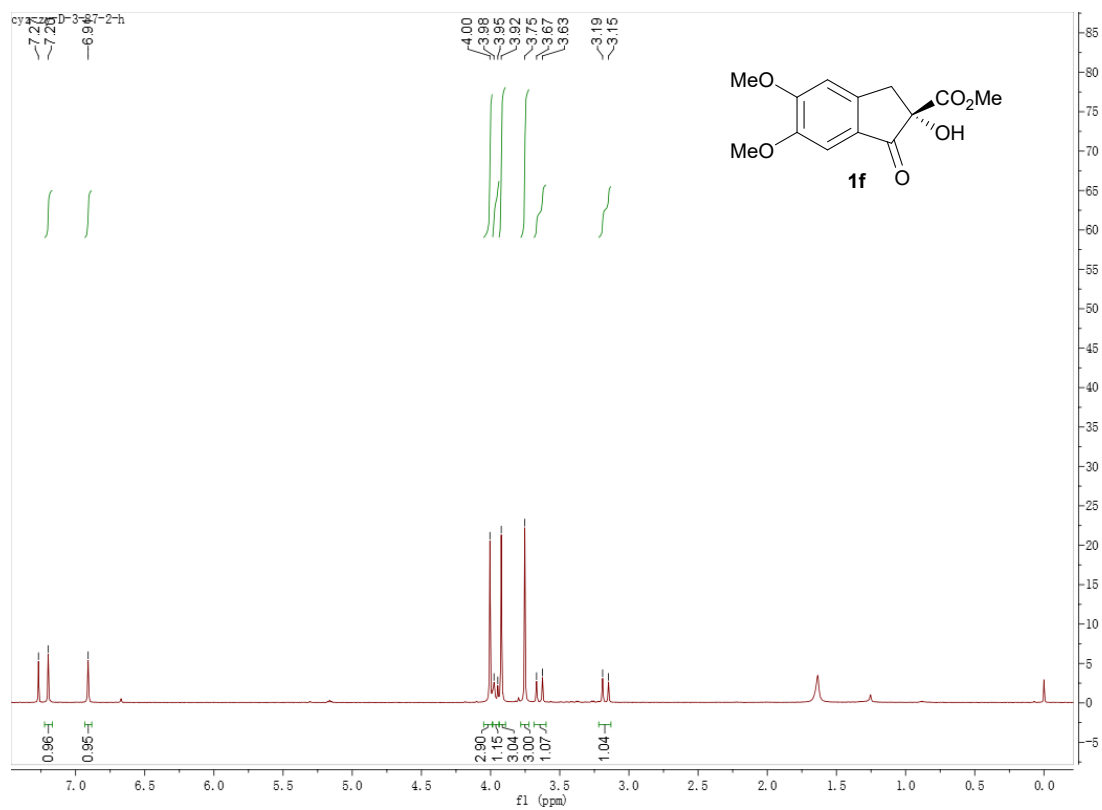
¹H NMR of compound 1e



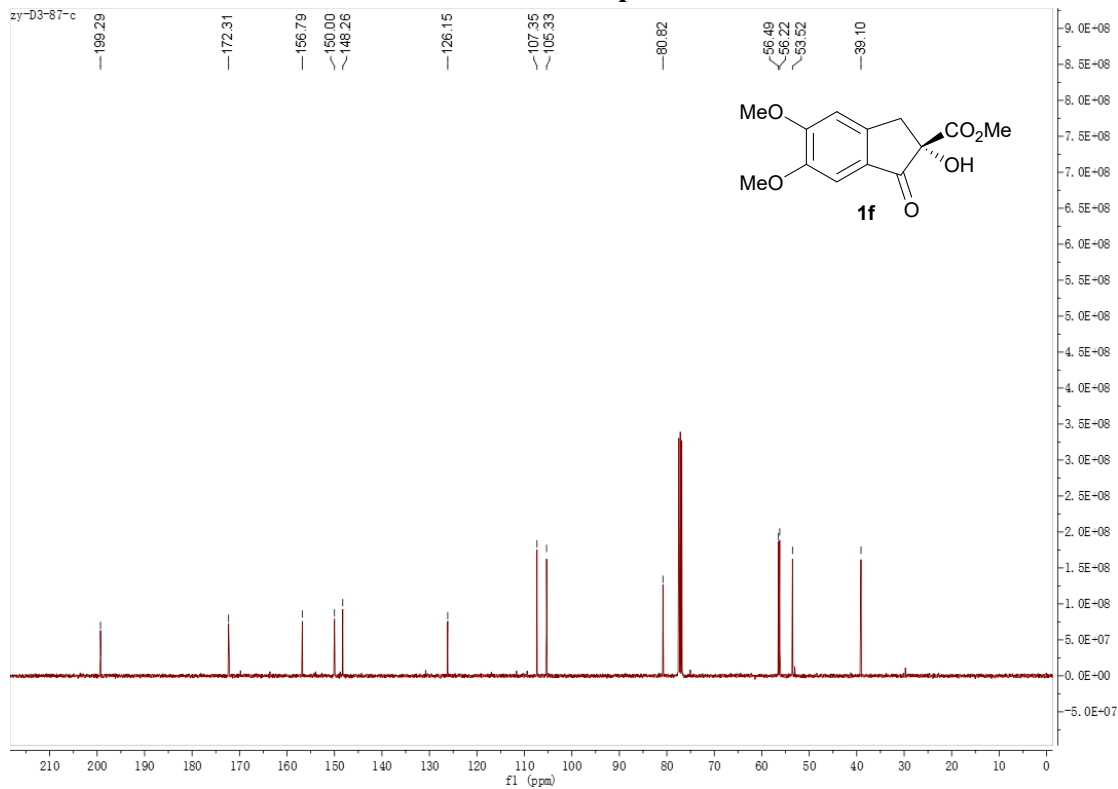
¹³C NMR of compound 1e



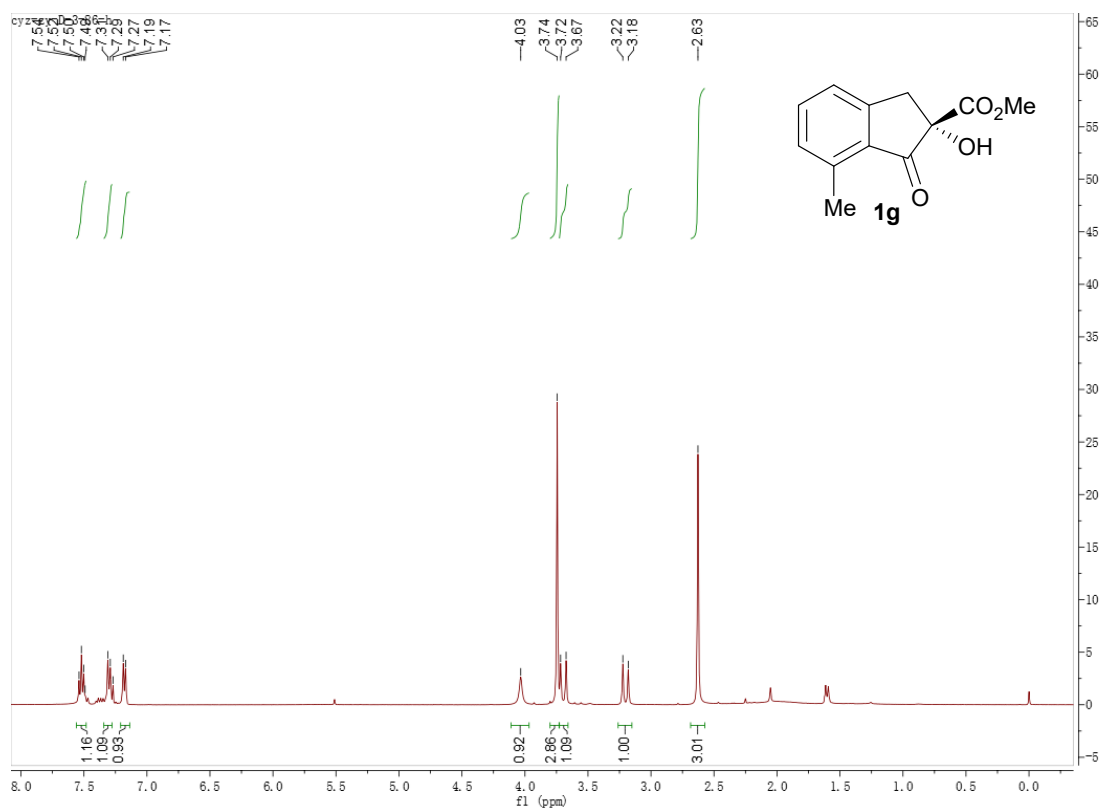
¹H NMR of compound 1f



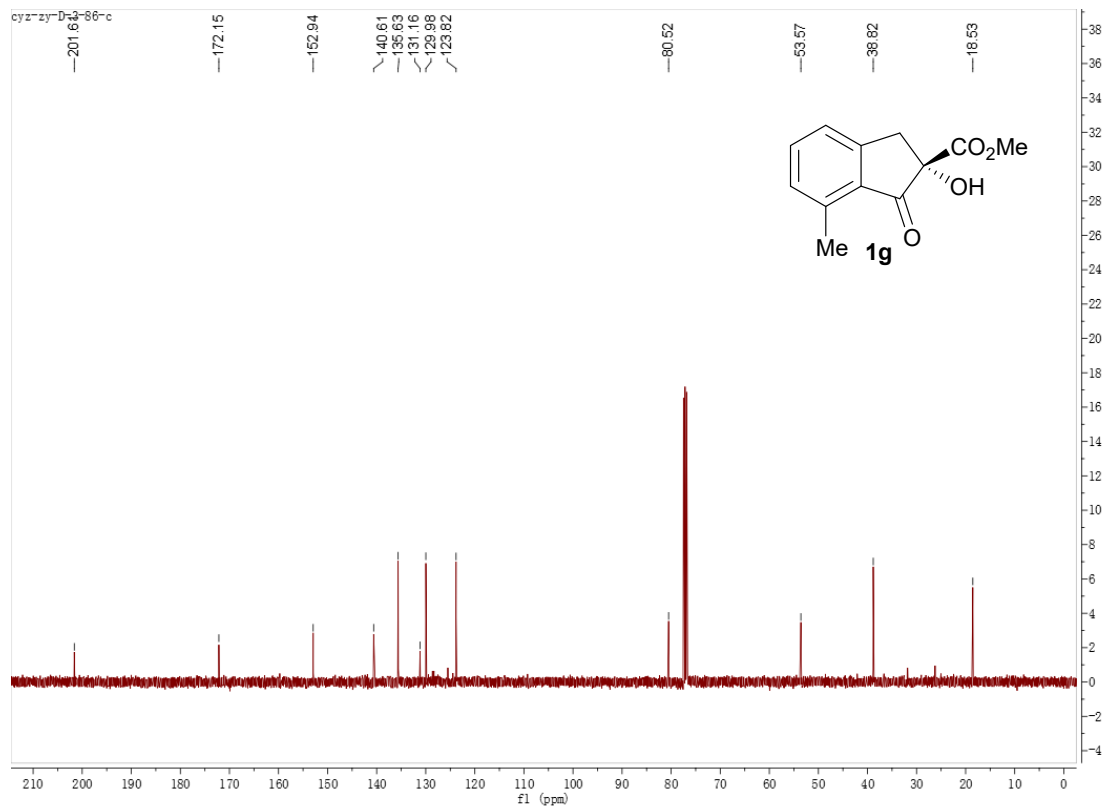
¹³C NMR of compound 1f



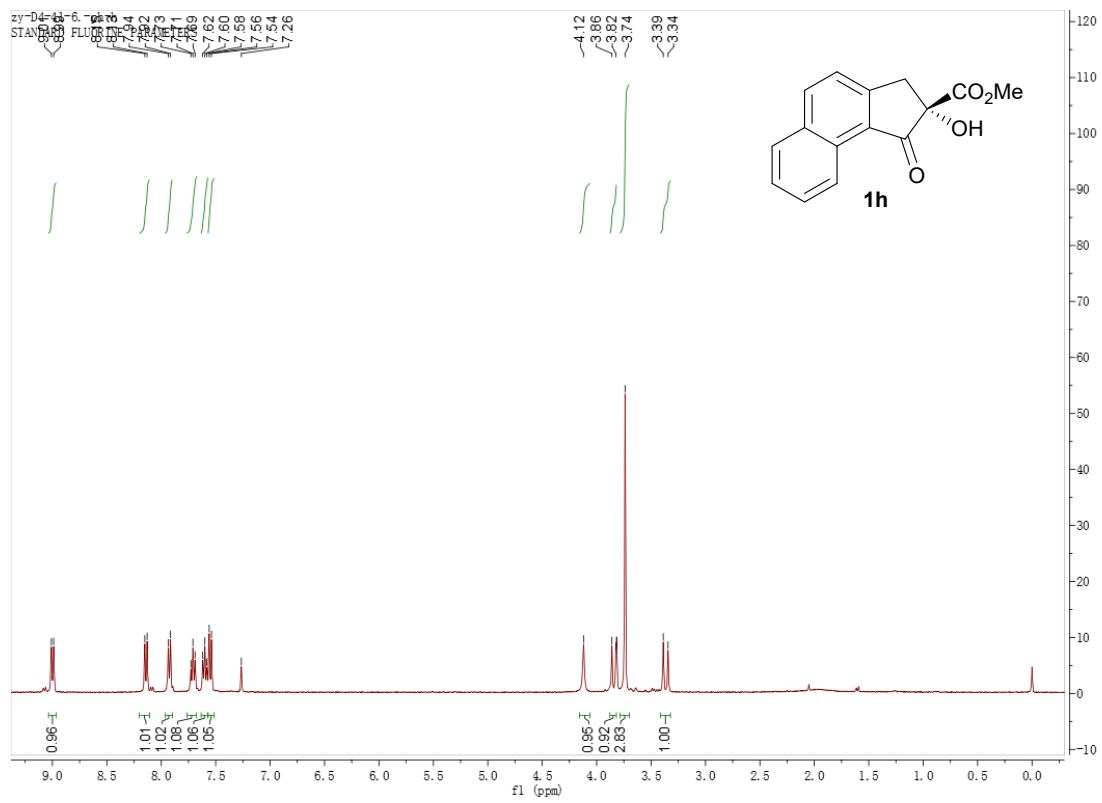
¹H NMR of compound 1g



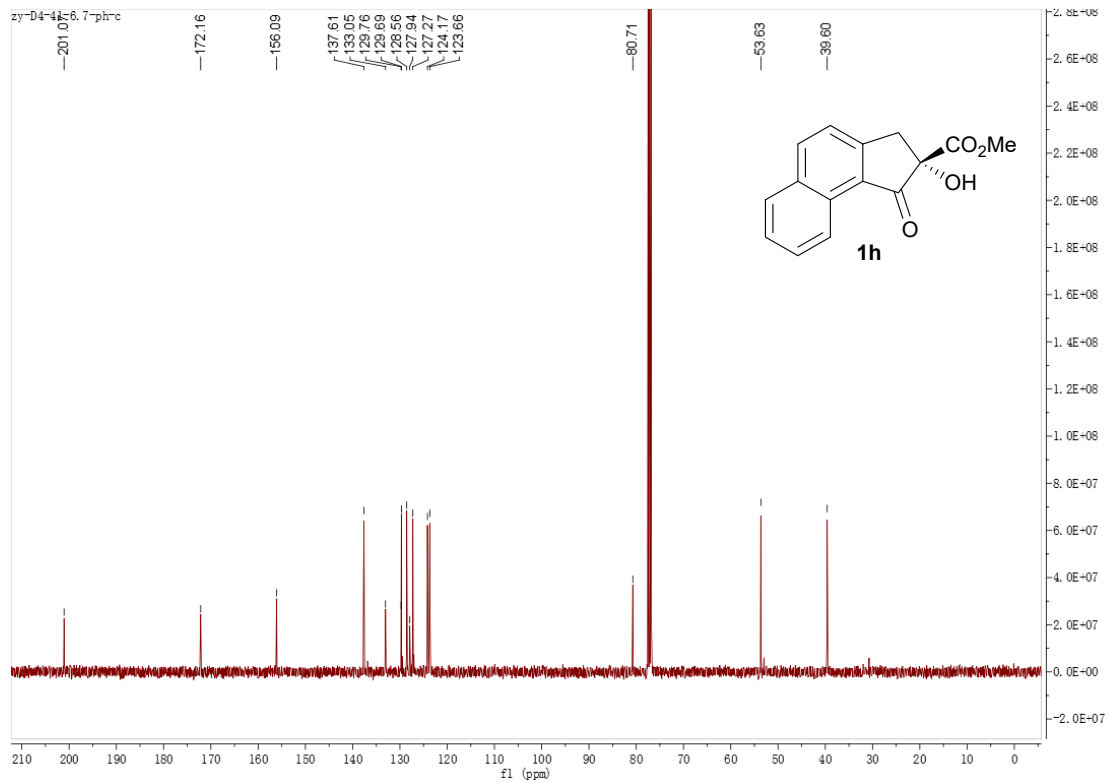
¹³C NMR of compound 1g



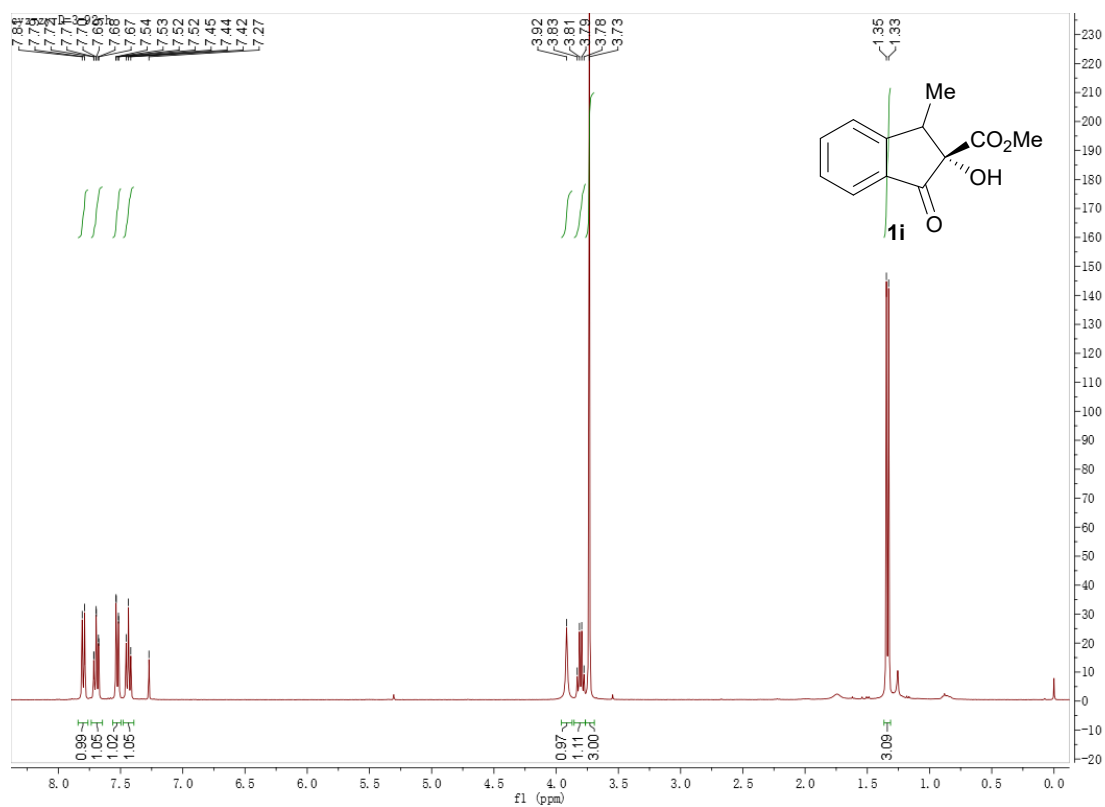
¹H NMR of compound 1h



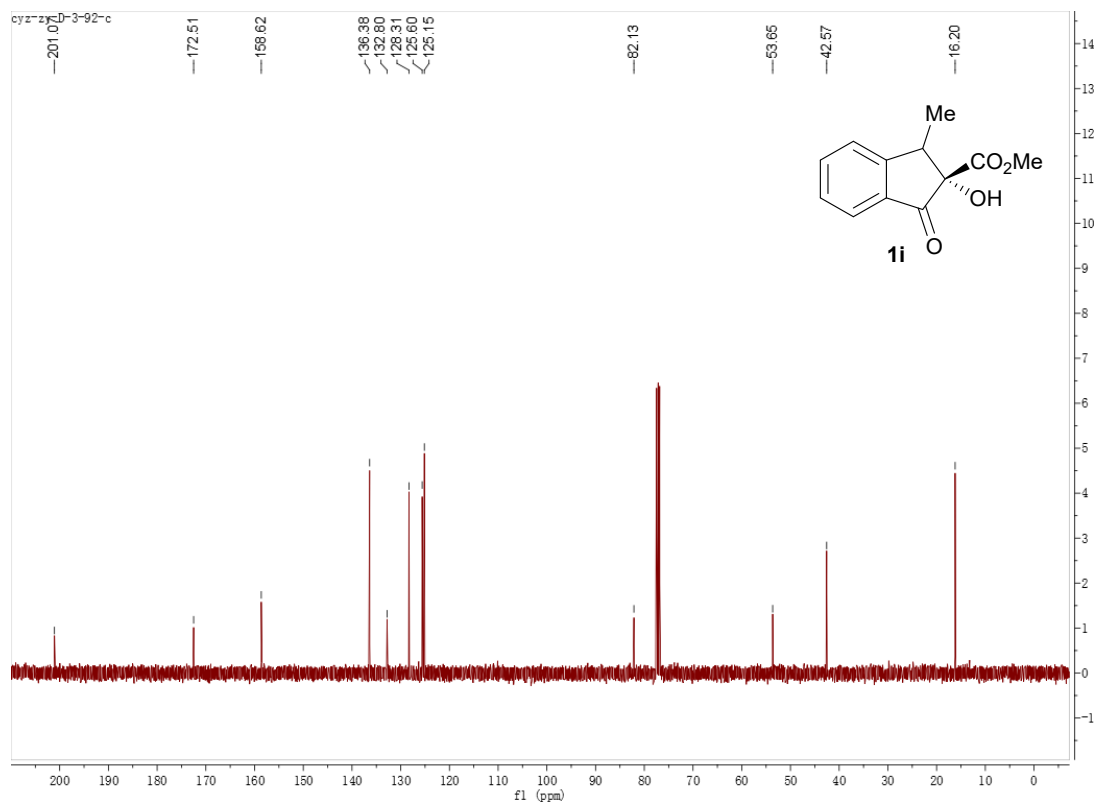
¹³C NMR of compound 1h



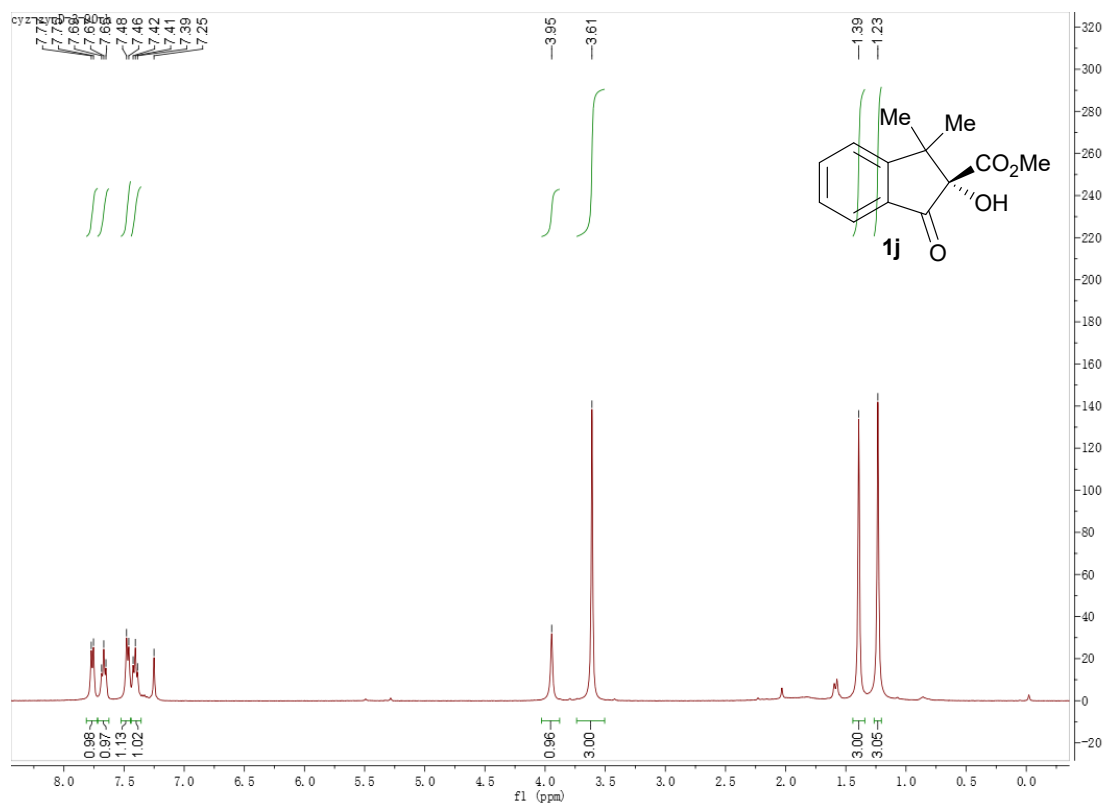
¹H NMR of compound 1i



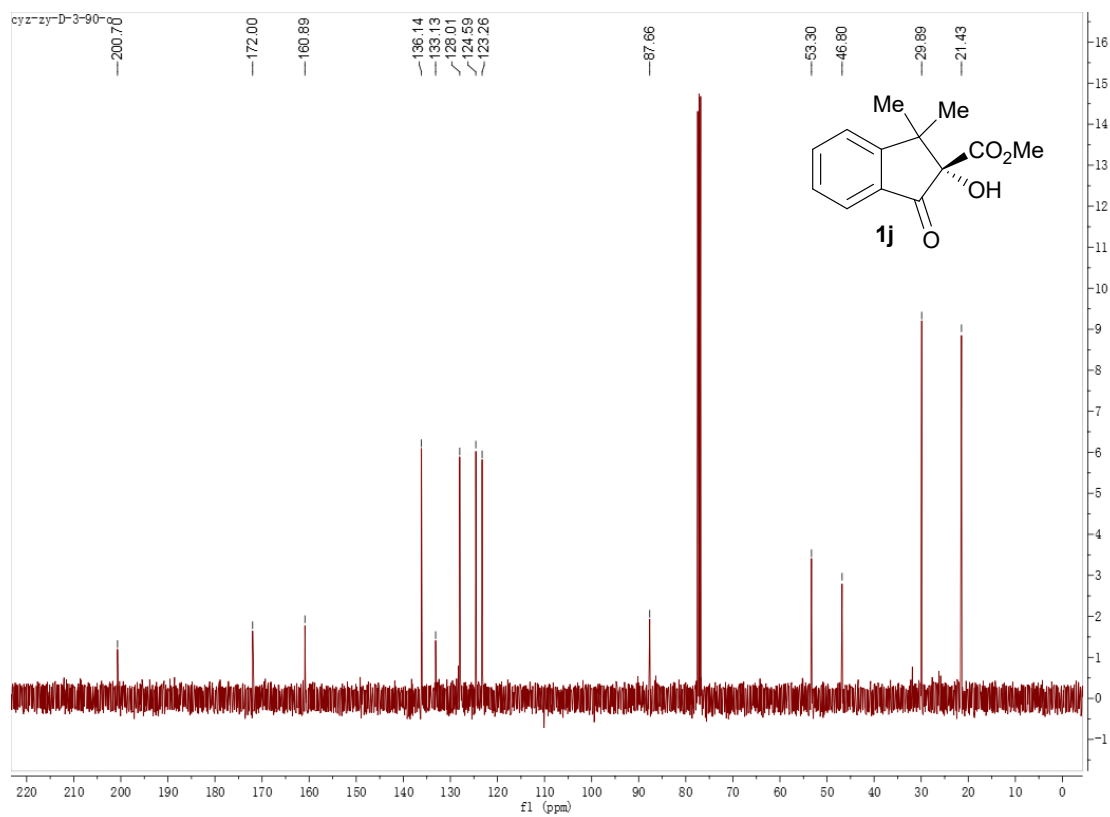
¹³C NMR of compound 1i



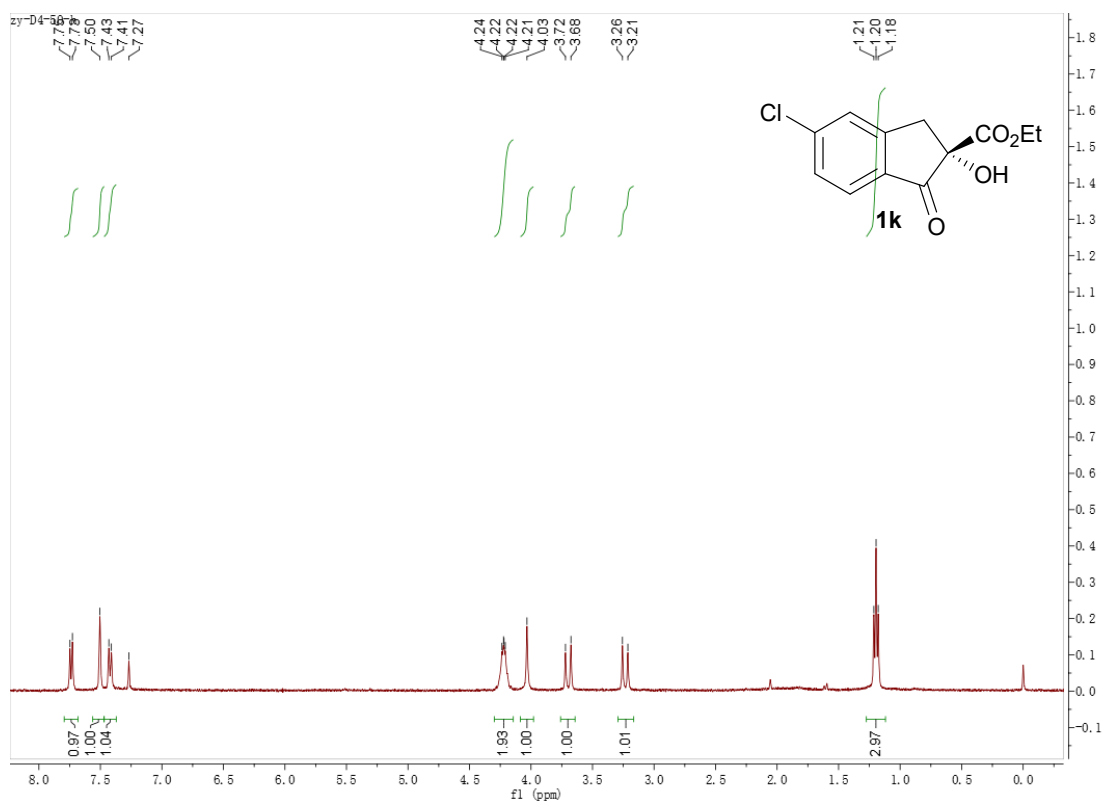
¹H NMR of compound 1j



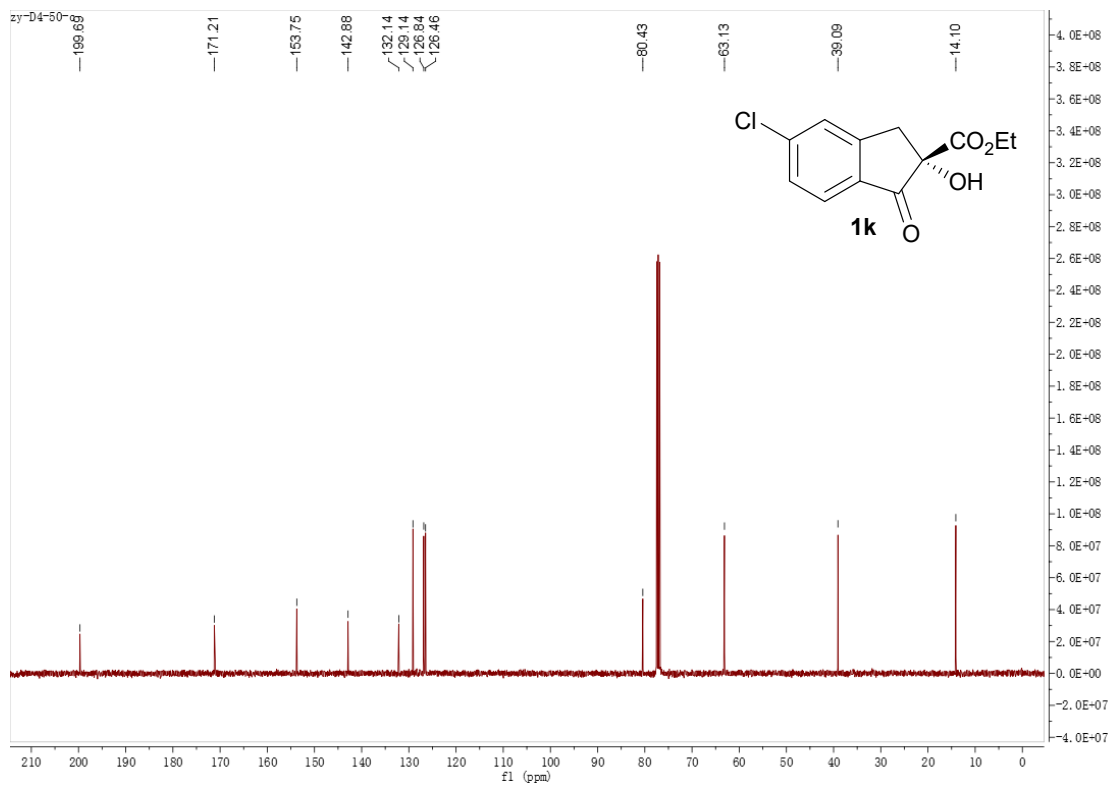
¹³C NMR of compound 1j



¹H NMR of compound 1k



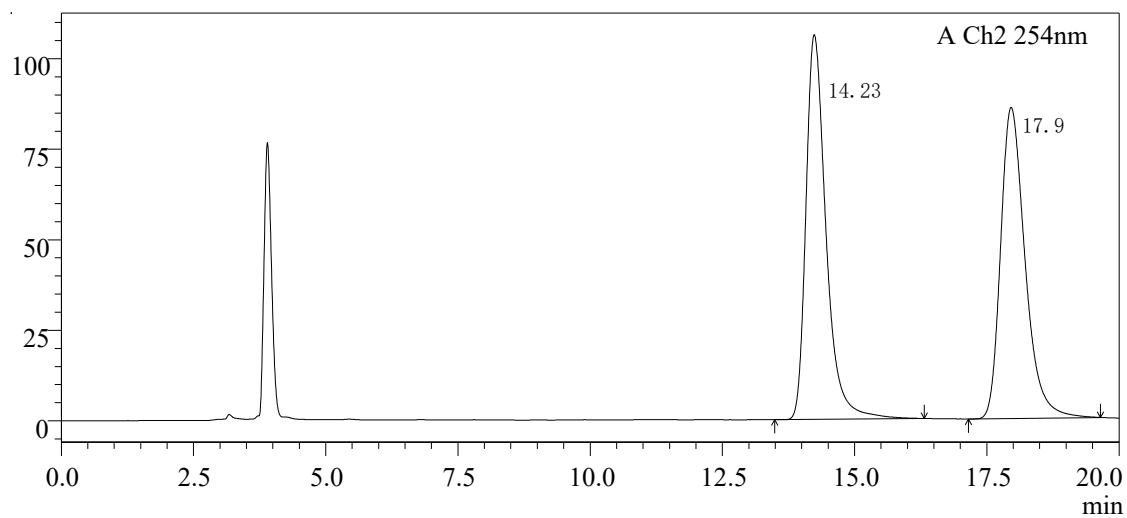
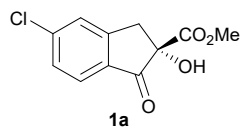
¹³C NMR of compound 1k



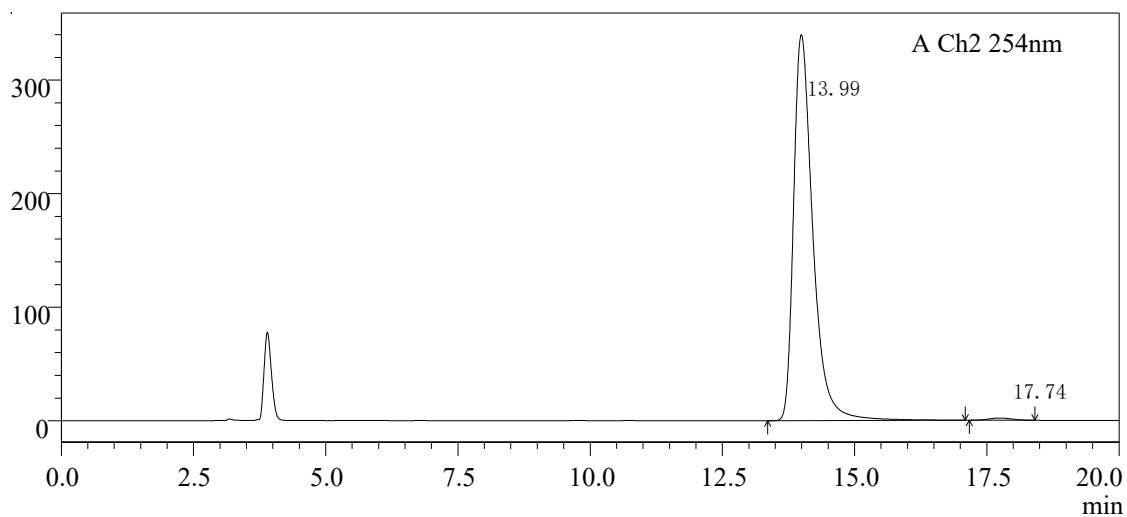
11. Copies of HPLC chromatogram

11.1 HPLC chromatogram of the Products

HPLC chromatogram of 1a

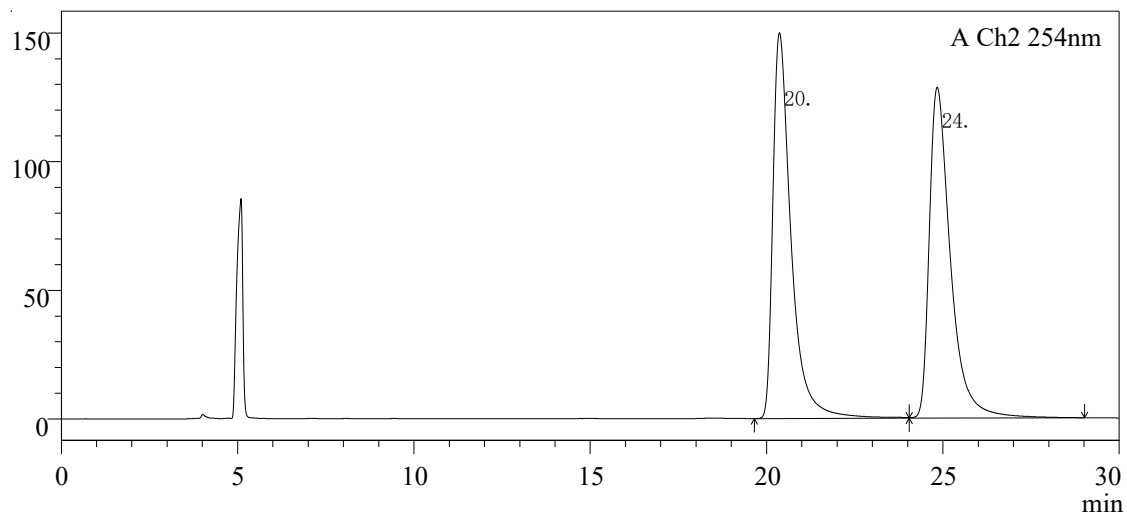
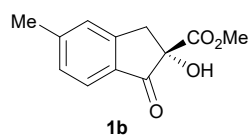


ID#	Ret. Time	Area	Height	Area %	Resolution
1	14.238	2800481	106246	50.367	--
2	17.959	2759621	85968	49.633	5.010

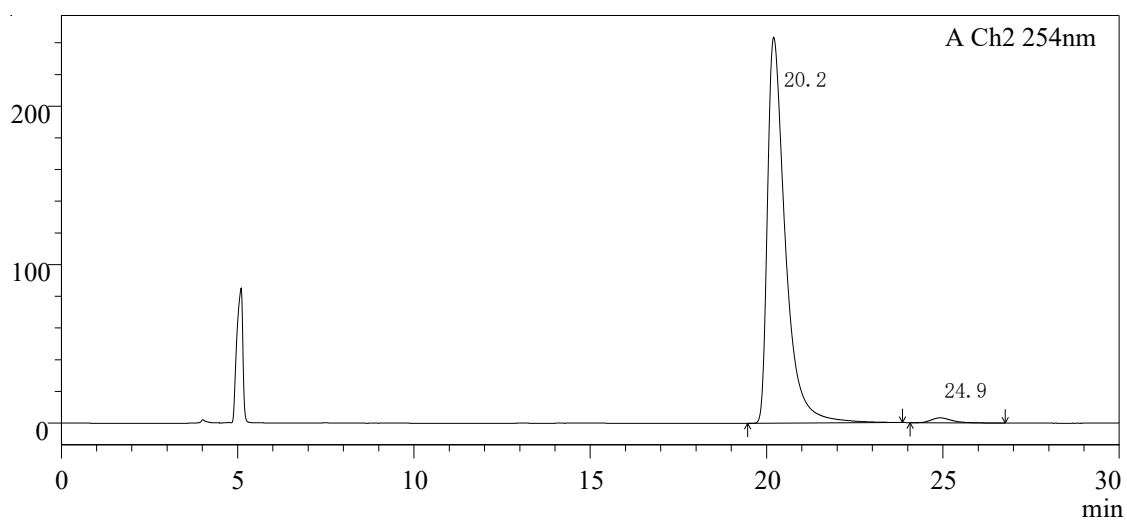


ID#	Ret. Time	Area	Height	Area %	Resolution
1	13.993	8635758	339860	99.321	--
2	17.747	59069	1980	0.679	5.222

HPLC chromatogram of 1b

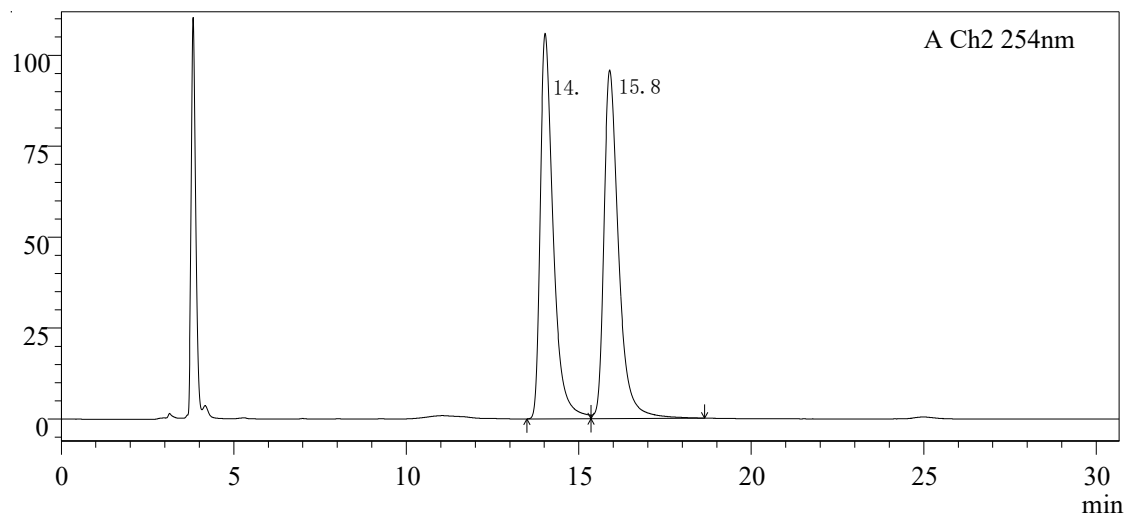
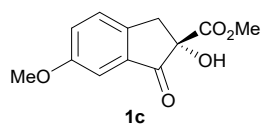


ID#	Ret. Time	Area	Height	Area %	Resolution
1	20.364	5405857	149908	50.064	--
2	24.838	5392134	128585	49.936	4.669

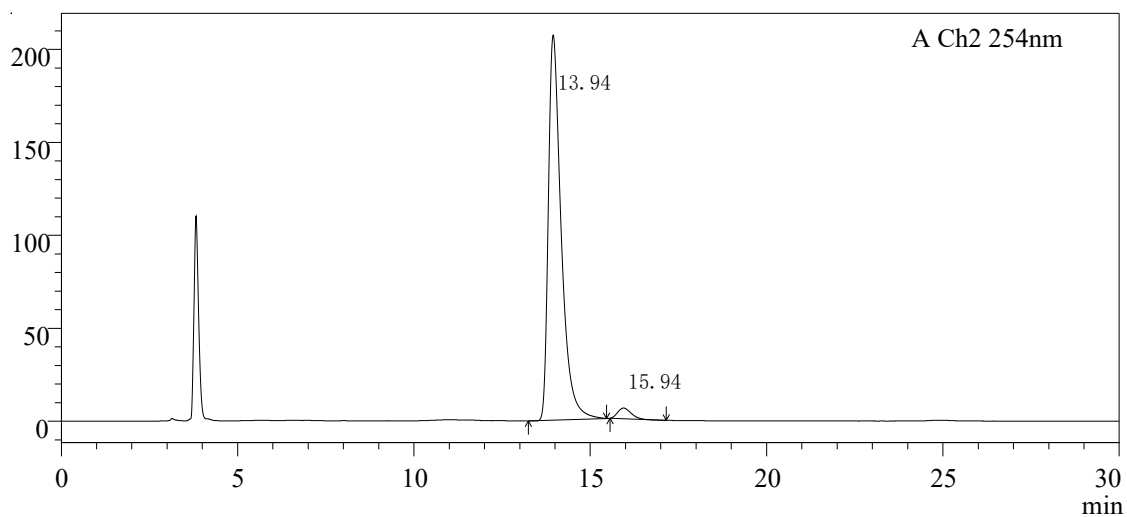


ID#	Ret. Time	Area	Height	Area %	Resolution
1	20.201	8633736	243759	98.554	--
2	24.920	126653	3108	1.446	4.852

HPLC chromatogram of 1c

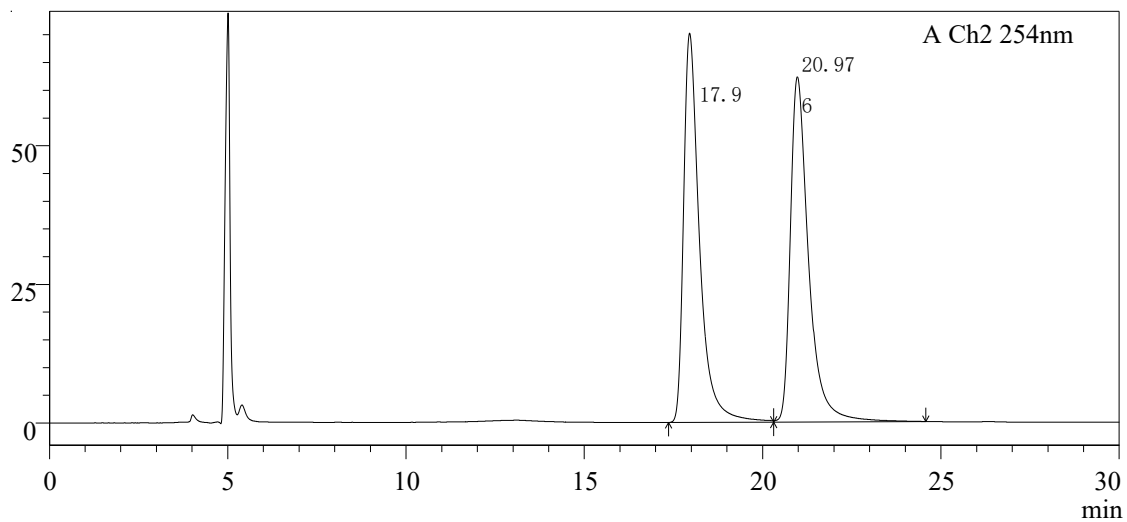
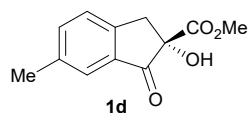


ID#	Ret. Time	Area	Height	Area %	Resolution
1	14.026	2771384	105971	49.488	--
2	15.898	2828716	95792	50.512	2.691

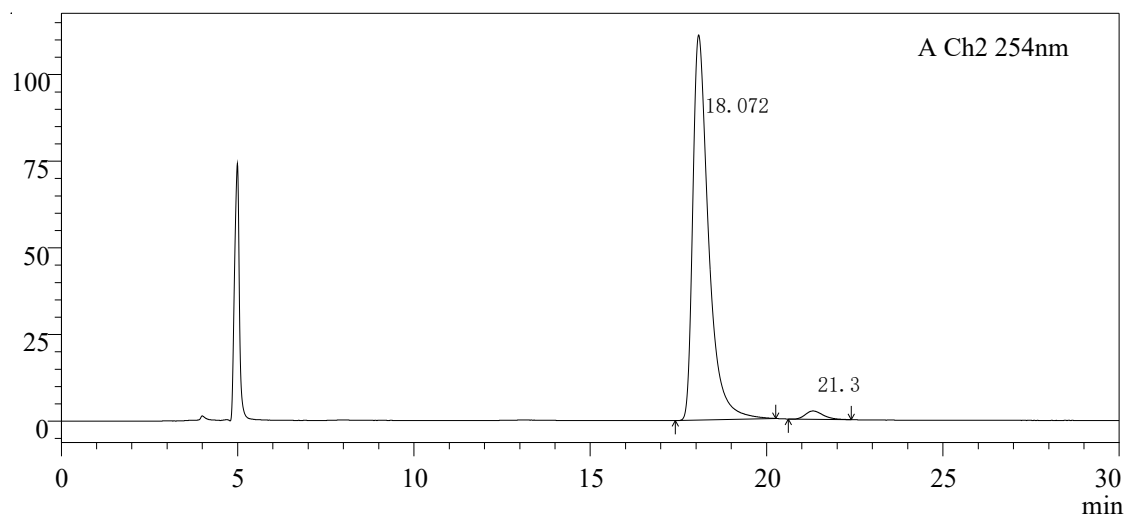


ID#	Ret. Time	Area	Height	Area %	Resolution
1	13.948	5264852	207216	97.216	--
2	15.944	150797	5746	2.784	2.945

HPLC chromatogram of 1d

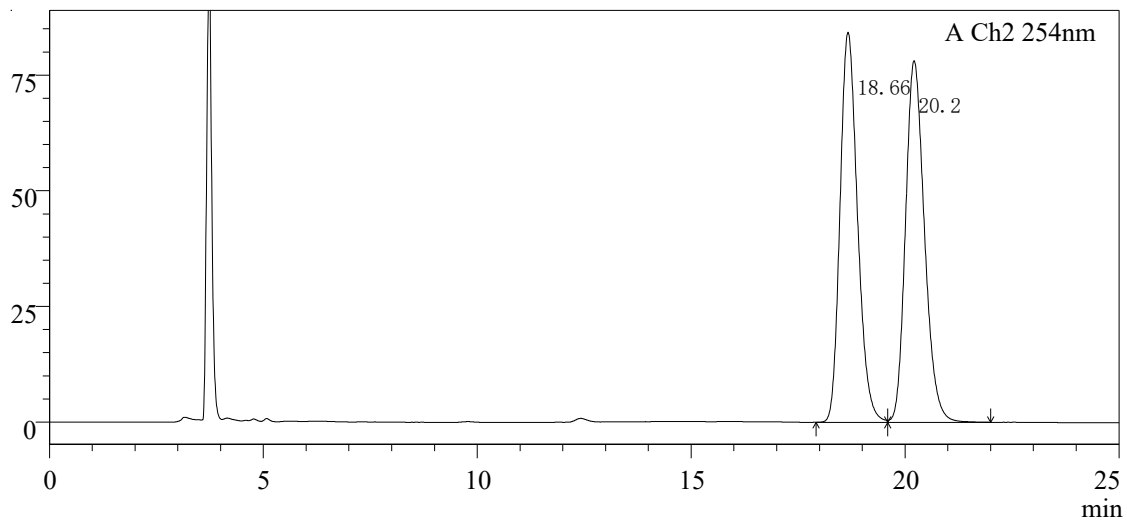
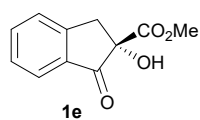


ID#	Ret. Time	Area	Height	Area %	Resolution
1	17.950	2194267	70188	49.776	--
2	20.976	2214056	62258	50.224	3.694

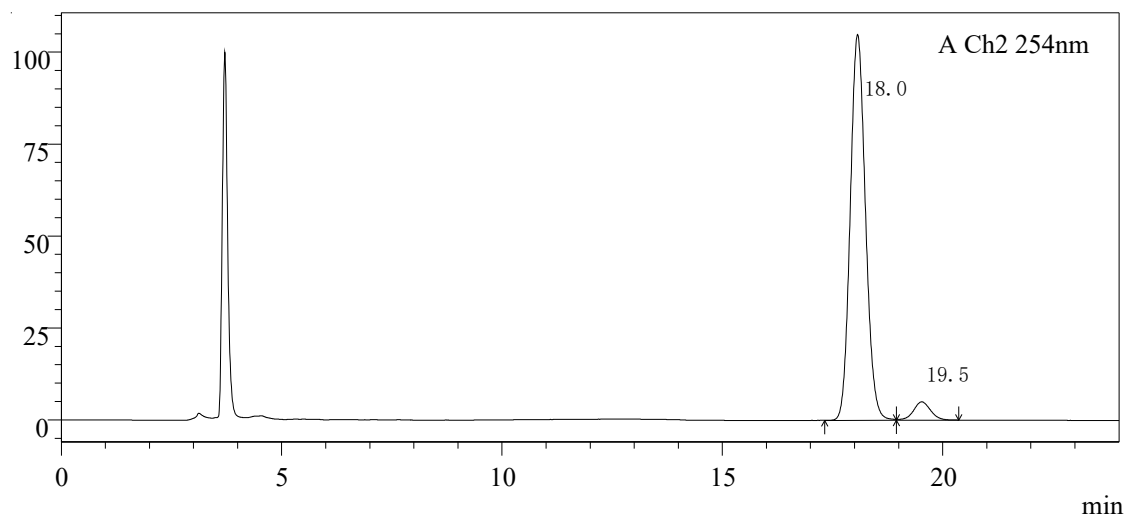


ID#	Ret. Time	Area	Height	Area %	Resolution
1	18.072	3470787	111158	97.652	--
2	21.321	83465	2399	2.348	3.872

HPLC chromatogram of 1e

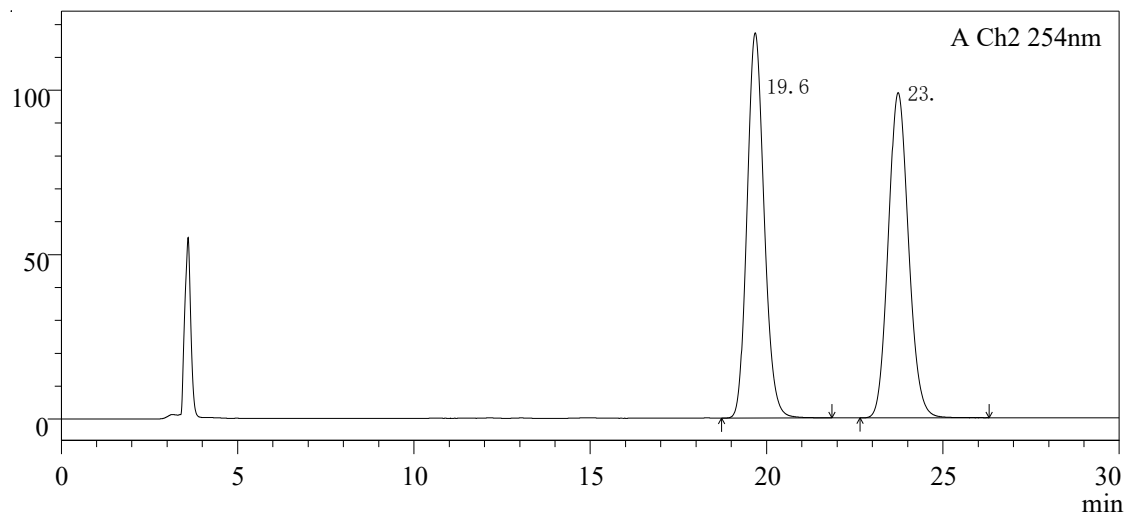
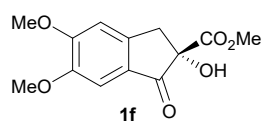


ID#	Ret. Time	Area	Height	Area %	Resolution
1	18.666	2395628	84315	49.865	--
2	20.212	2408642	78160	50.135	1.999

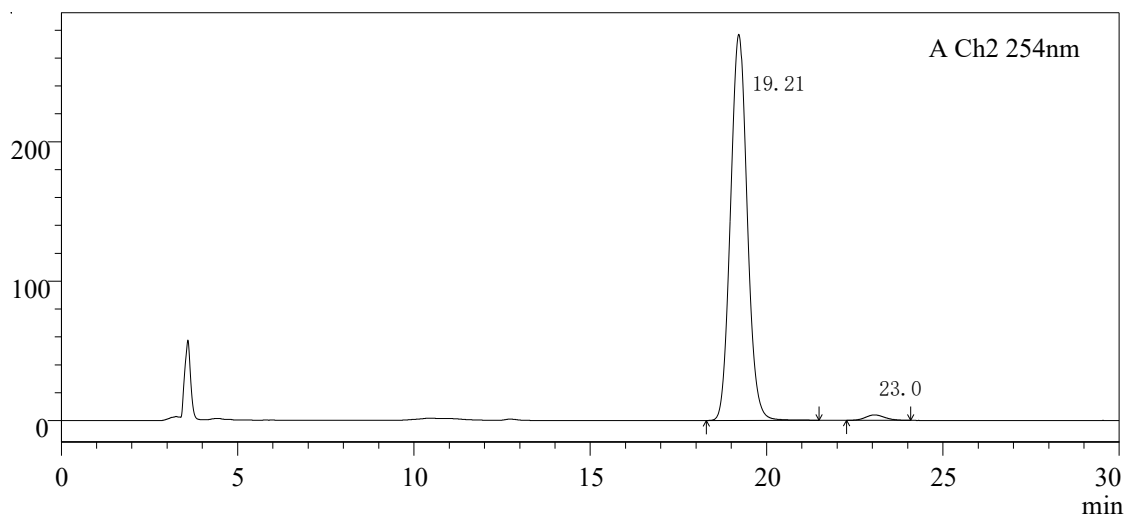


ID#	Ret. Time	Area	Height	Area %	Resolution
1	18.068	2522781	104962	94.879	--
2	19.526	136152	4992	5.121	2.179

HPLC chromatogram of 1f

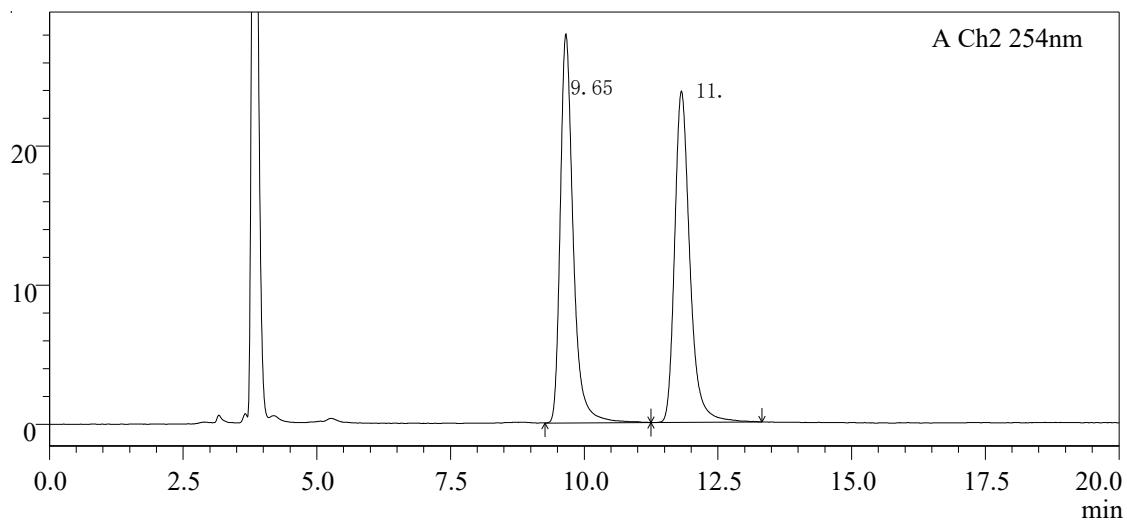
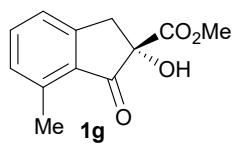


ID#	Ret. Time	Area	Height	Area %	Resolution
1	19.674	3943280	117186	50.033	--
2	23.727	3938122	98908	49.967	4.202

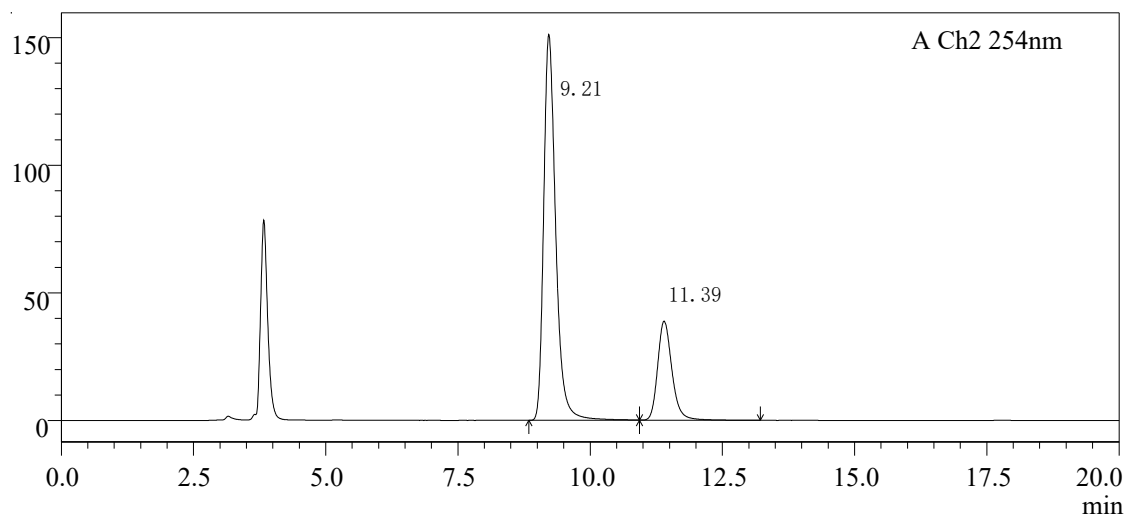


ID#	Ret. Time	Area	Height	Area %	Resolution
1	19.215	9030907	276886	98.389	--
2	23.061	147825	3880	1.611	4.128

HPLC chromatogram of 1g

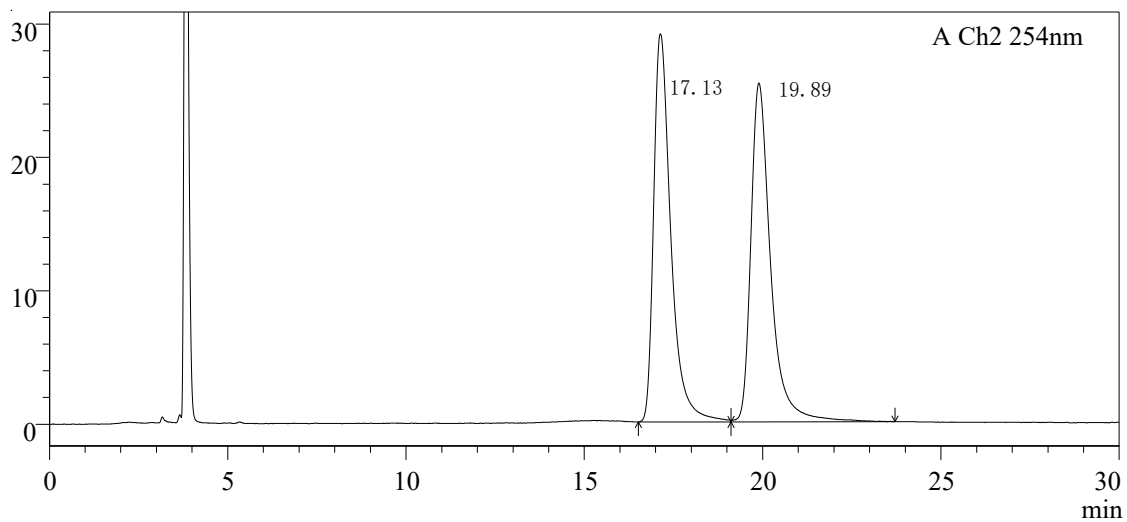
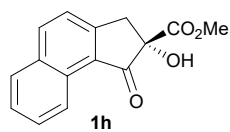


ID#	Ret. Time	Area	Height	Area %	Resolution
1	9.656	471840	27986	49.967	--
2	11.817	472456	23838	50.033	4.660

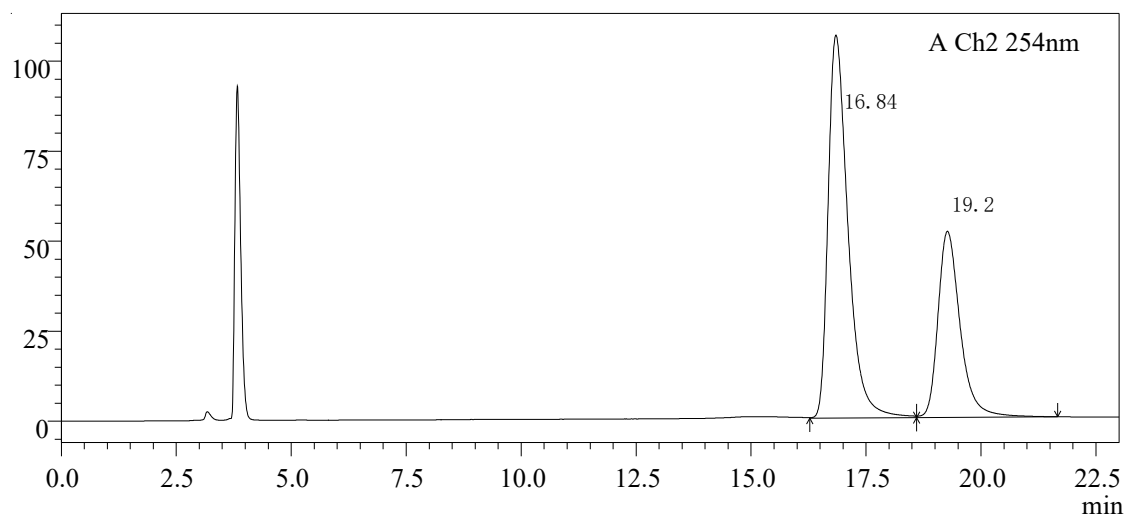


ID#	Ret. Time	Area	Height	Area %	Resolution
1	9.219	2383650	151142	76.506	--
2	11.398	731981	38859	23.494	4.933

HPLC chromatogram of 1h

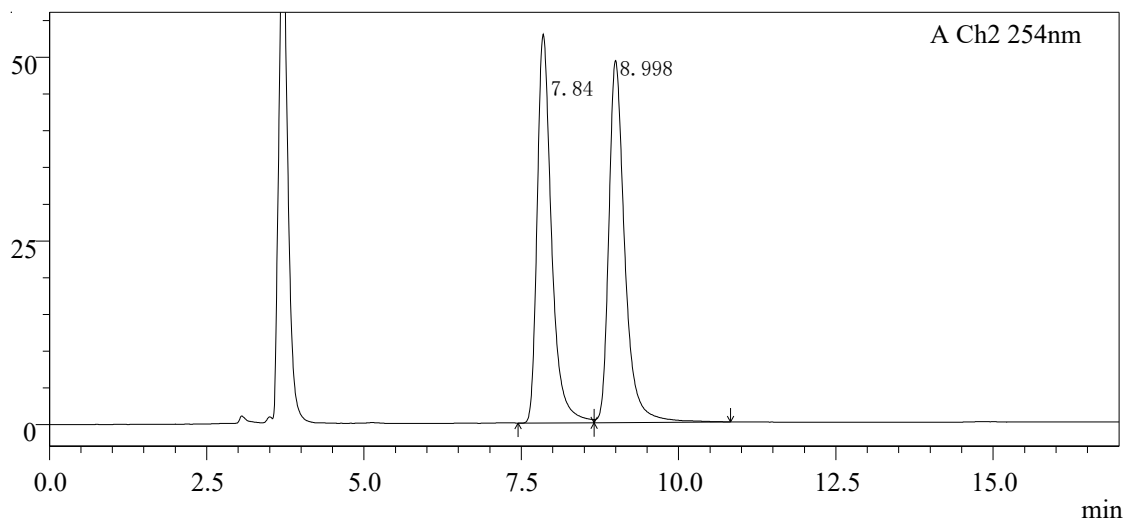
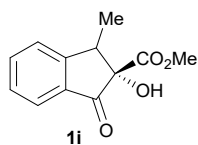


ID#	Ret. Time	Area	Height	Area %	Resolution
1	17.134	965728	29101	49.651	--
2	19.893	979312	25395	50.349	3.131

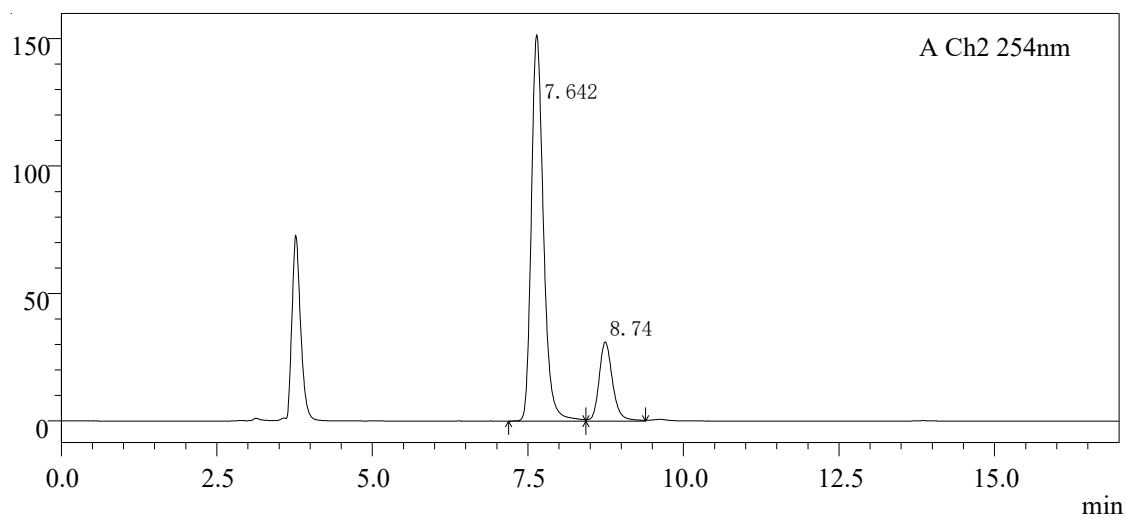


ID#	Ret. Time	Area	Height	Area %	Resolution
1	16.845	3197817	106386	64.839	--
2	19.270	1734149	51706	35.161	3.016

HPLC chromatogram of 1i

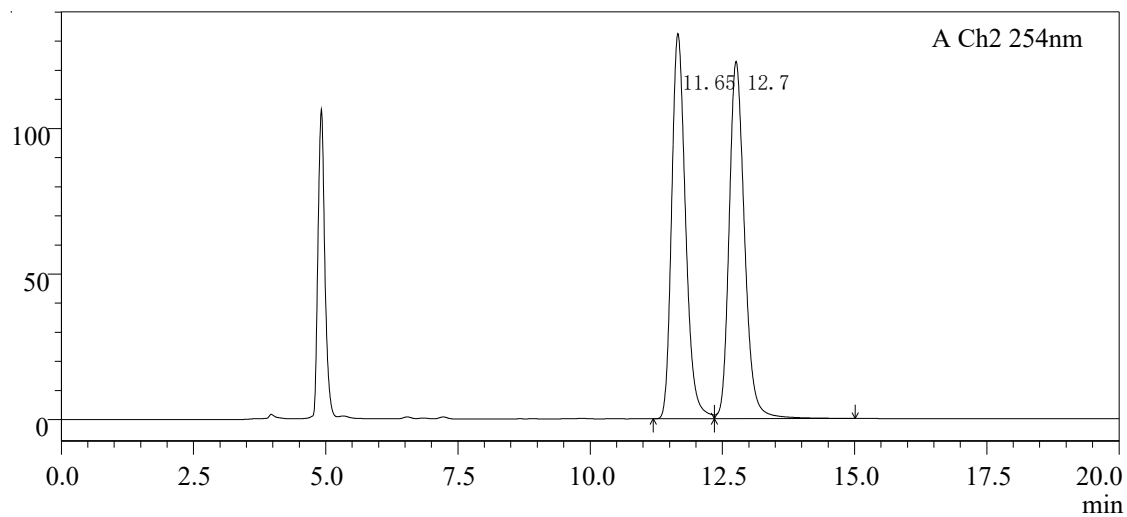
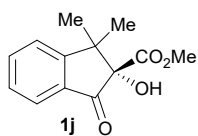


ID#	Ret. Time	Area	Height	Area %	Resolution
1	7.848	840912	52921	49.182	--
2	8.998	868874	49266	50.818	2.733

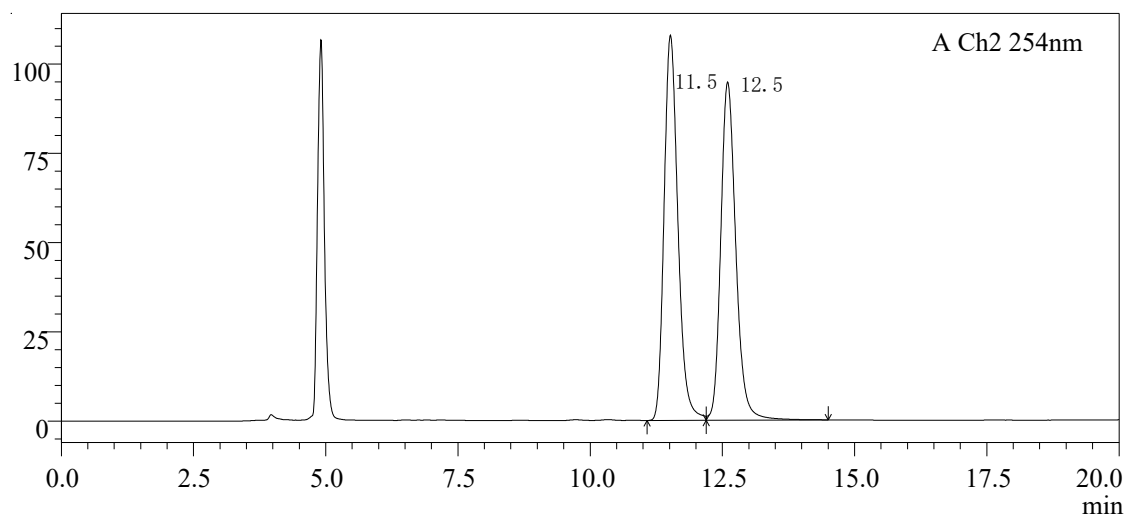


ID#	Ret. Time	Area	Height	Area %	Resolution
1	7.642	2052377	151496	81.487	--
2	8.743	466283	31046	18.513	2.988

HPLC chromatogram of 1j

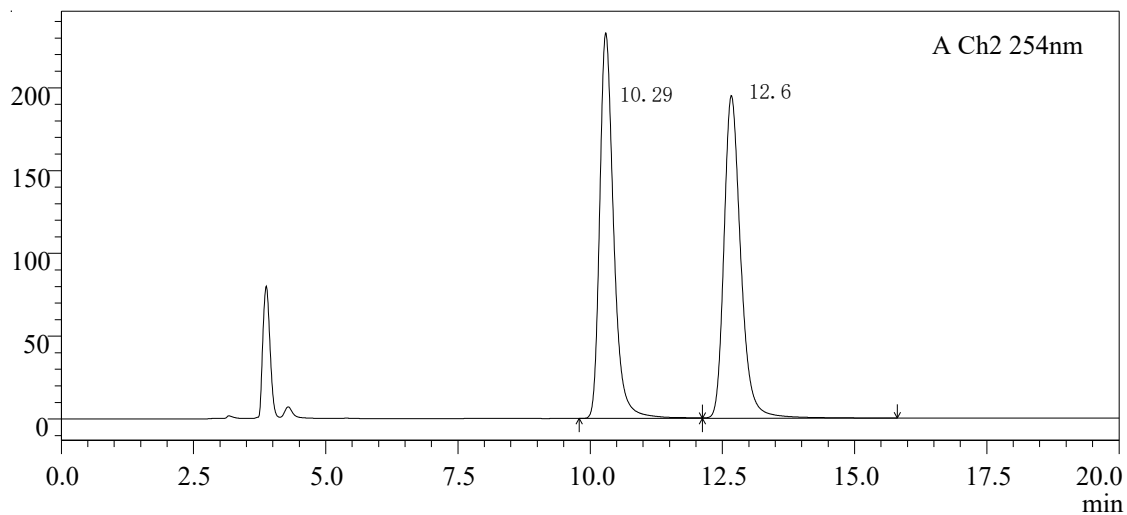
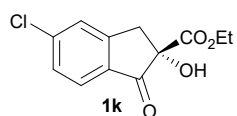


ID#	Ret. Time	Area	Height	Area %	Resolution
1	11.658	2435065	132387	49.519	--
2	12.759	2482350	122777	50.481	2.206

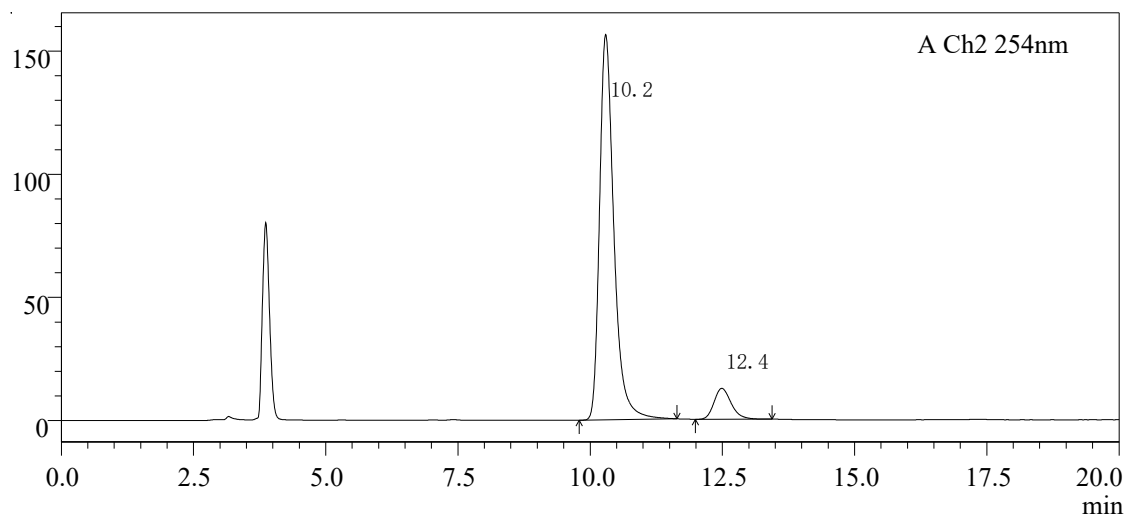


ID#	Ret. Time	Area	Height	Area %	Resolution
1	11.516	1954801	107965	50.970	--
2	12.599	1880401	94763	49.030	2.207

HPLC chromatogram of 1k



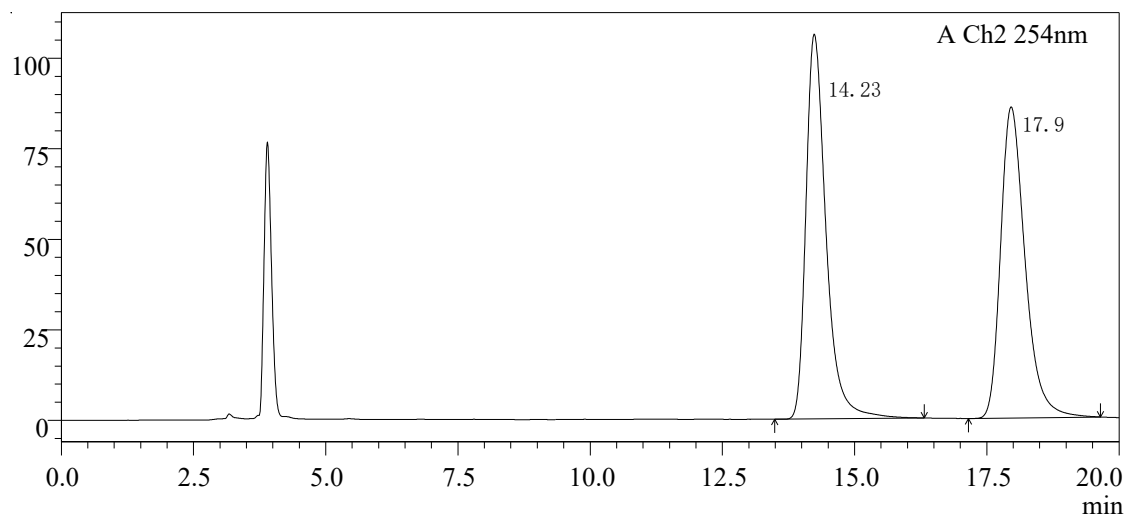
ID#	Ret. Time	Area	Height	Area %	Resolution
1	10.294	4292526	232864	49.857	--
2	12.670	4317075	194856	50.143	4.608



ID#	Ret. Time	Area	Height	Area %	Resolution
1	10.293	2965045	156536	91.258	--
2	12.490	284018	12559	8.742	4.106

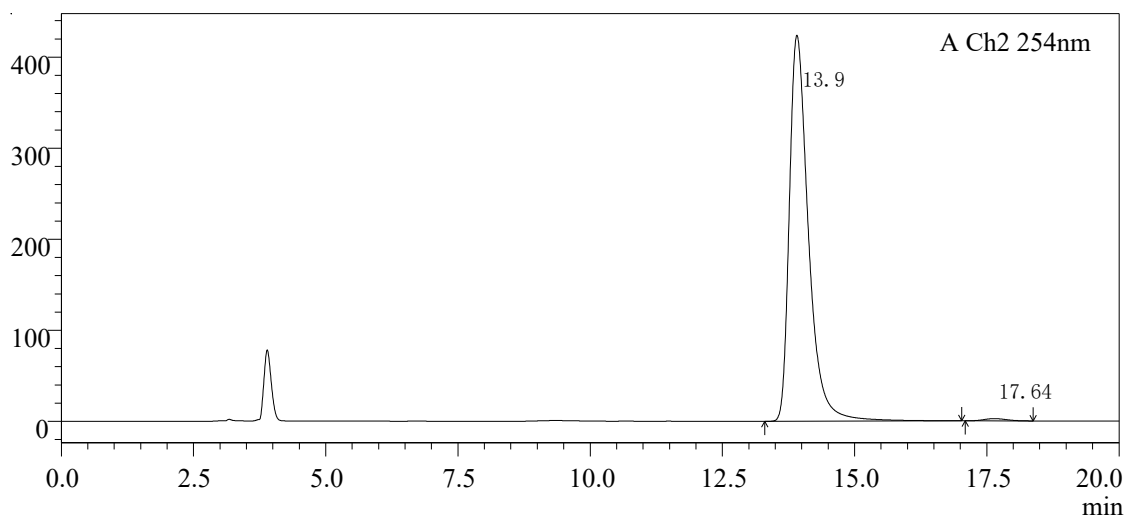
11.2 HPLC chromatogram of the recyclability of the catalytic system

racemate



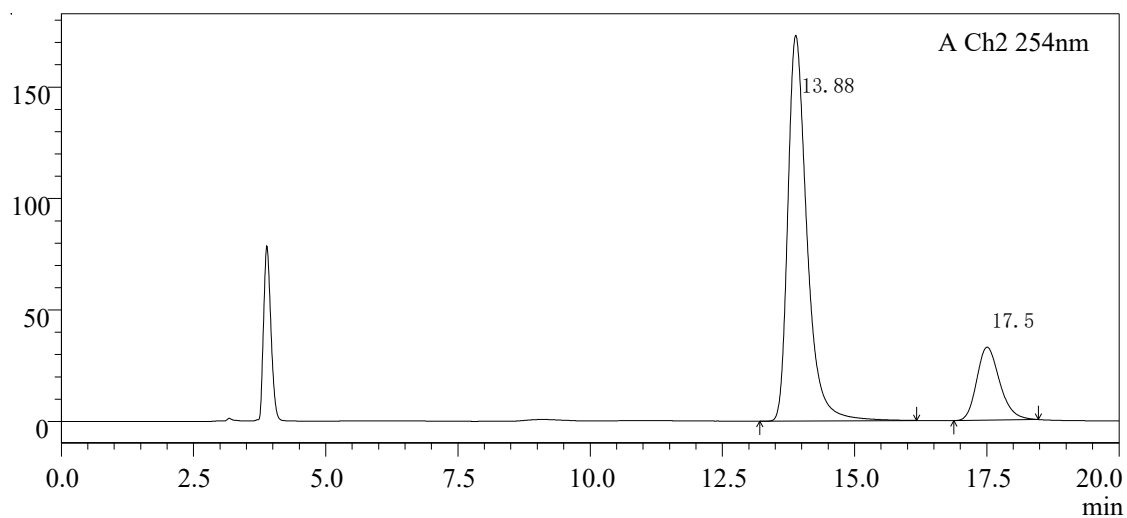
ID#	Ret. Time	Area	Height	Area %	Resolution
1	14.238	2800481	106246	50.367	--
2	17.959	2759621	85968	49.633	5.010

First cycle



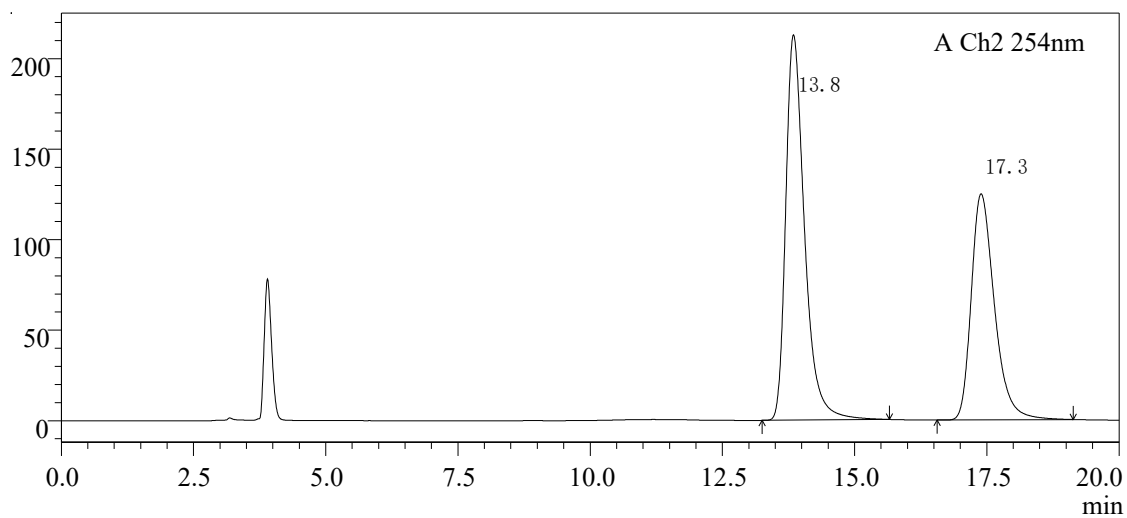
ID#	Ret. Time	Area	Height	Area %	Resolution
1	13.909	10672976	423955	99.300	--
2	17.646	75200	2527	0.700	5.240

Second cycle



ID#	Ret. Time	Area	Height	Area %	Resolution
1	13.887	4352306	173029	81.433	--
2	17.505	992315	32783	18.567	5.085

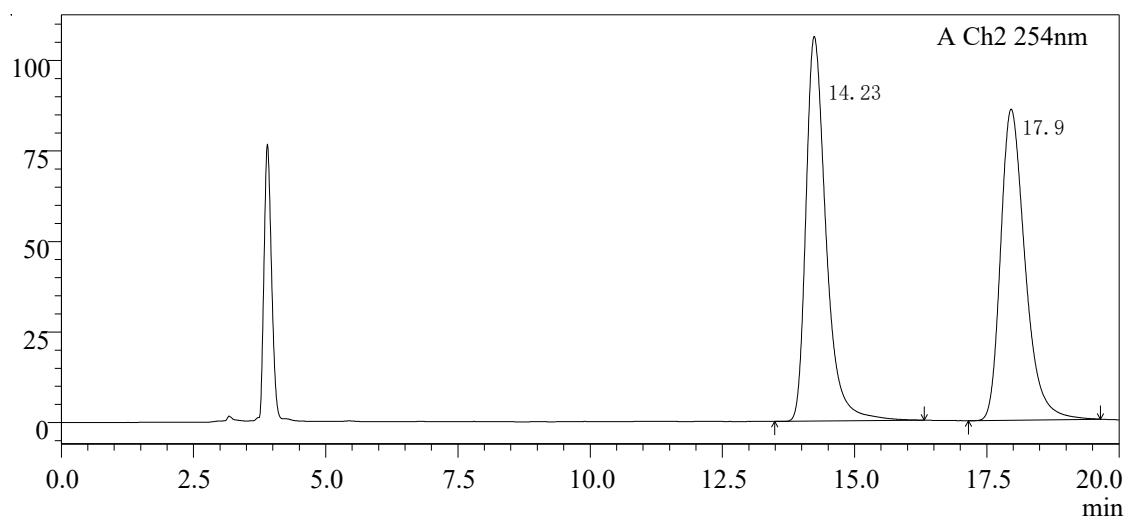
Third cycle



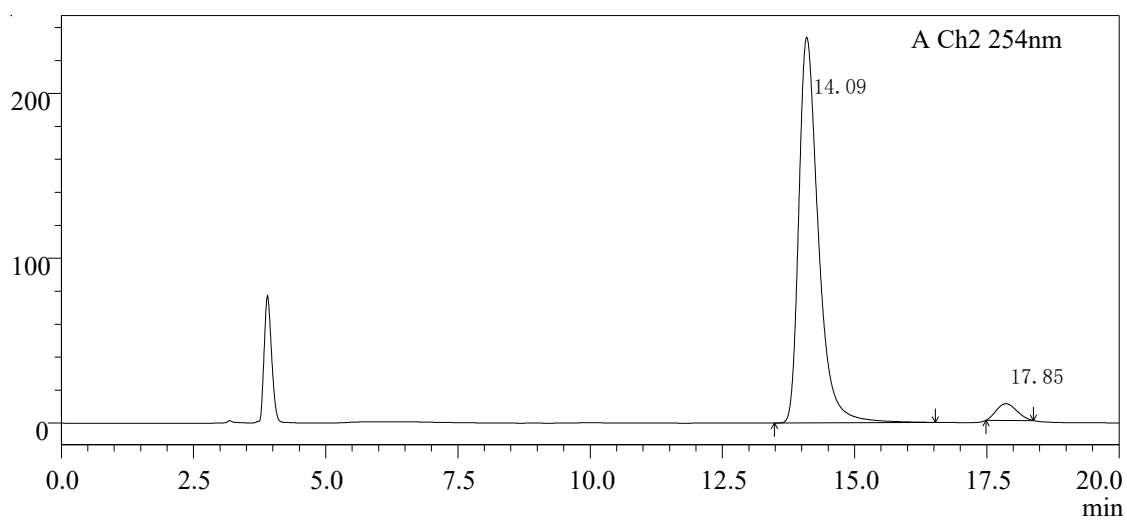
ID#	Ret. Time	Area	Height	Area %	Resolution
1	13.844	5279032	212910	58.058	--
2	17.392	3813669	124911	41.942	5.028

11.3 HPLC chromatogram of gram-scale experiment

racemate



ID#	Ret. Time	Area	Height	Area %	Resolution
1	14.238	2800481	106246	50.367	--
2	17.959	2759621	85968	49.633	5.010



ID#	Ret. Time	Area	Height	Area %	Resolution
1	14.096	6012567	234063	95.657	--
2	17.859	272968	10143	4.343	5.413