

Supporting Information

Rapid synthesis of uniform silicalite-1 zeolite via a micro-liquid film reactor-assisted approach

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Figures and Tables

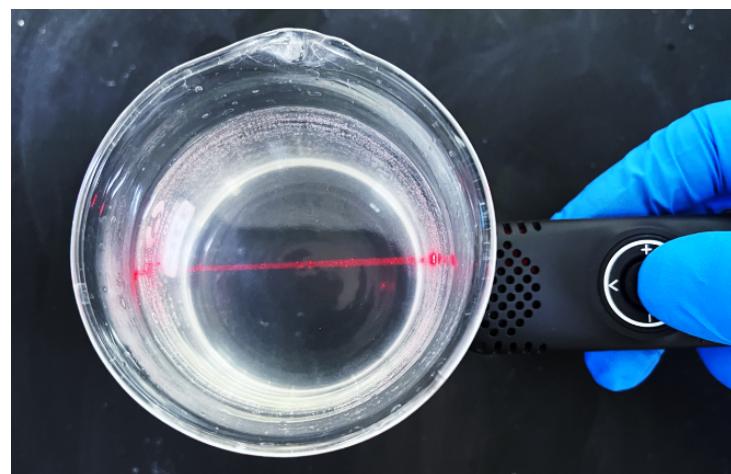


Fig. S1 Photo for Tyndall effect of the zeolite precursor pretreated in MLF reactor for 10 min.

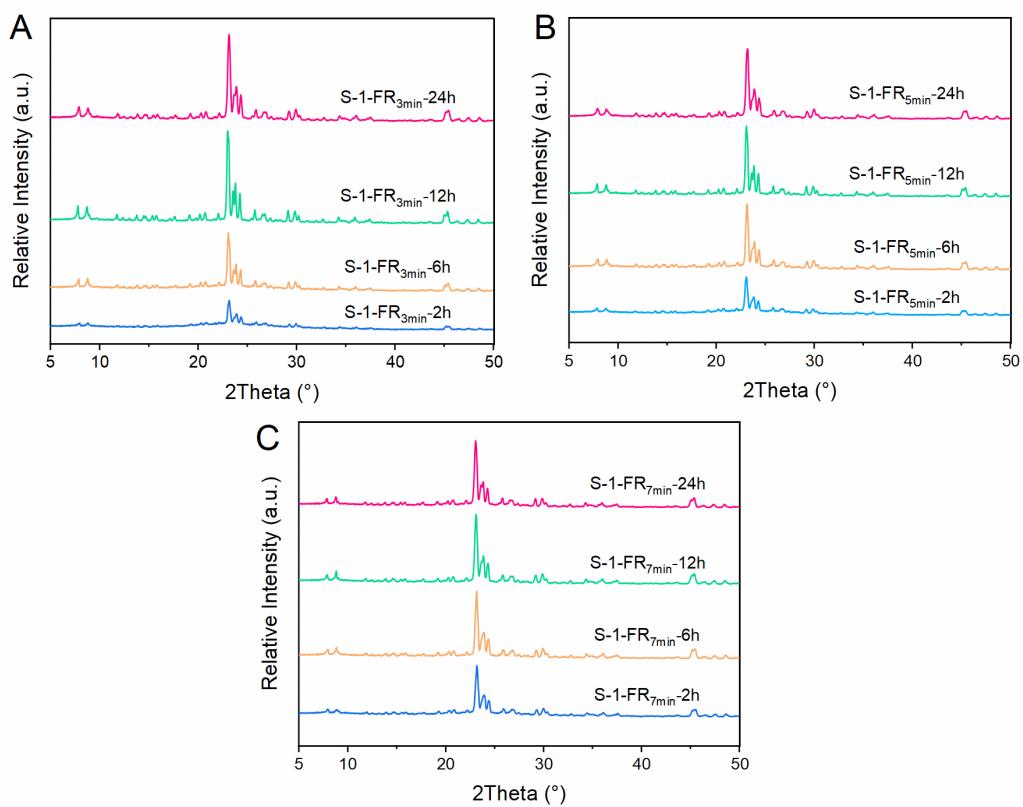


Fig. S2 XRD patterns of samples derived from MLF reactor: (A) S-1-FR_{3min}- y , (B) S-1-FR_{5min}- y , (C) S-1-FR_{7min}- y ($y=2, 6, 12, 24$ h).

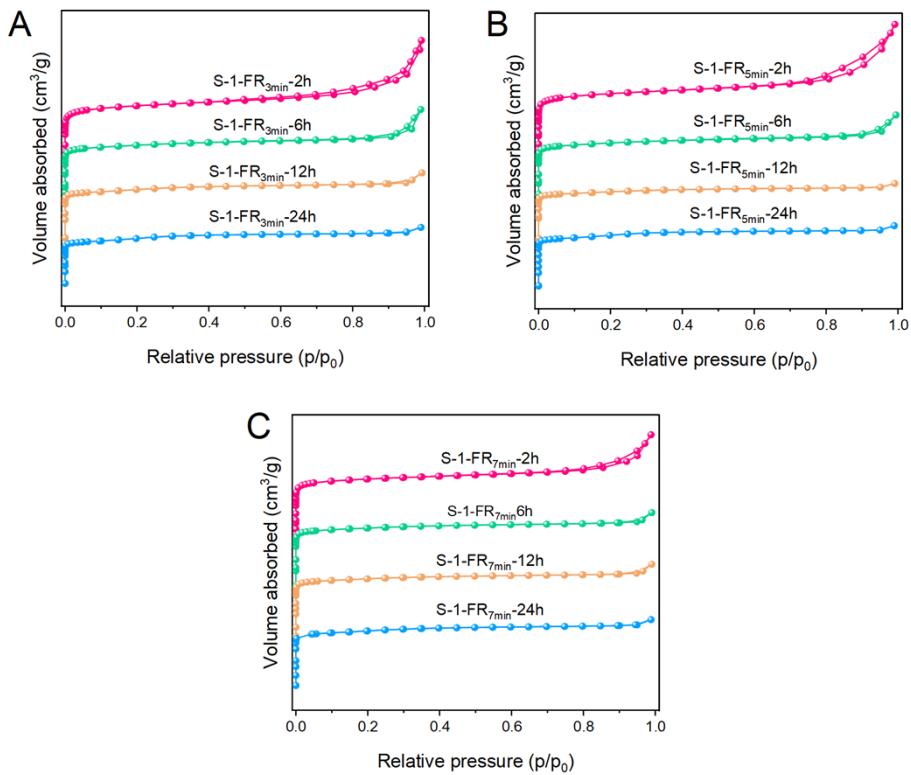


Fig. S3 Nitrogen adsorption-desorption isotherms of S-1 zeolite samples derived from MLF reactor: (A) S-1-FR_{3min}-y, (B) S-1-FR_{5min}-y, (C) S-1-FR_{7min}-y (y=2, 6, 12, 24 h).

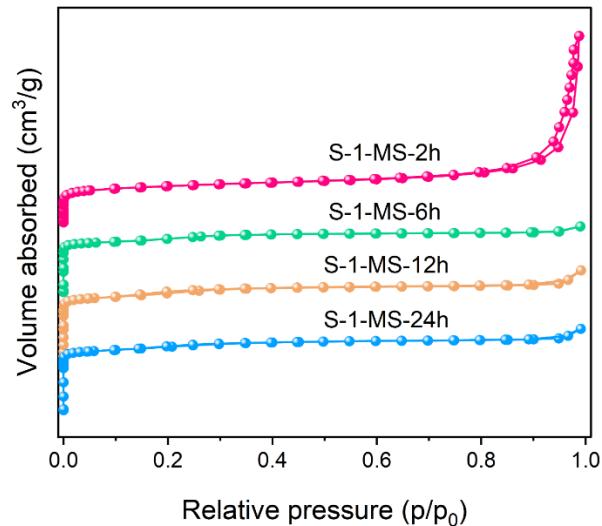


Fig. S4 Nitrogen adsorption-desorption isotherms of S-1-MS-y samples (y=2, 6, 12, 24 h).

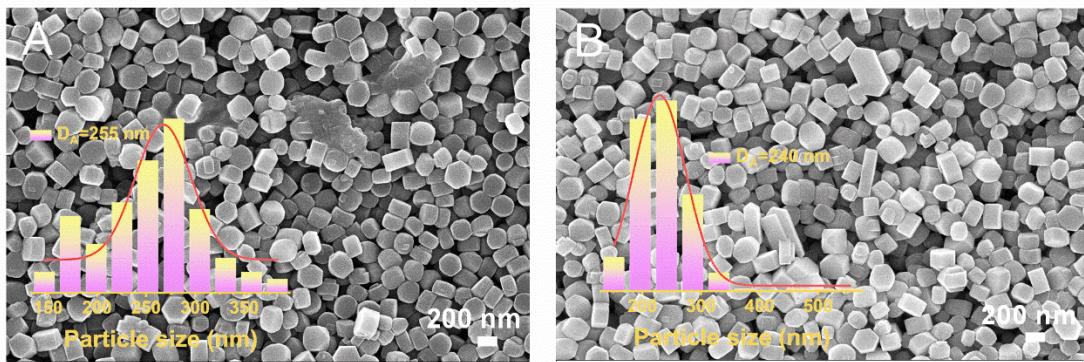


Fig. S5 SEM images of S-1 zeolite samples derived from mechanical stirring of 12 h (A) and 24 h (B) under hydrothermal conditions (170 °C, 2 h). The insets in A and B are crystal size distribution diagrams.

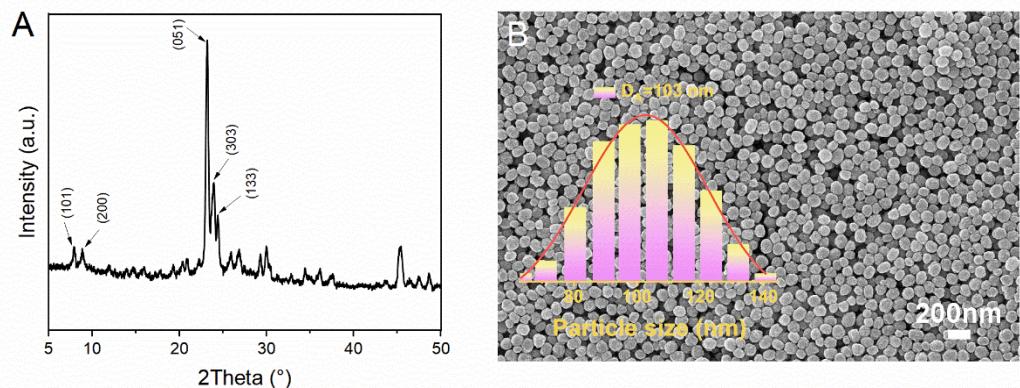


Fig. S6 XRD pattern (A) and SEM image (B) of S-1 zeolite sample derived from MLF reactor for a mixing time of 10 min after static hydrothermal treatment at 90 °C. The inset in B is crystal size distribution diagram.

Table S1. Properties of the various S-1 zeolite samples.

Samples	S_{BET} ^a (m ² /g)	S_{micro} ^a (m ² /g)	V_{total} ^b (cm ³ /g)	V_{micro} ^c (cm ³ /g)
S-1-FR _{3min} -2h	424	352	0.41	0.16
S-1-FR _{3min} -6h	364	337	0.28	0.16
S-1-FR _{3min} -12h	370	362	0.22	0.17
S-1-FR _{3min} -24h	373	359	0.19	0.16
S-1-FR _{5min} -2h	433	359	0.43	0.16
S-1-FR _{5min} -6h	401	357	0.27	0.16
S-1-FR _{5min} -12h	388	374	0.19	0.17
S-1-FR _{5min} -24h	400	394	0.21	0.18
S-1-FR _{7min} -2h	435	394	0.32	0.17
S-1-FR _{7min} -6h	436	414	0.23	0.18
S-1-FR _{7min} -12h	407	391	0.23	0.18
S-1-FR _{7min} -24h	424	415	0.21	0.18

^a Determined by BET method. ^b Determined by Volume adsorbed at p/p₀ = 0.995. ^c Determined by t-plot method.