
Supporting Information

"Visualizing" the partially reversible conversion of gold nanoclusters *via Au₂₃(S-c-C₆H₁₁)₁₇ intermediate*

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I. Experimental section

1.1. Materials

All chemicals were purchased commercially and were used without prior purification. Tetrachloroauric (III) acid ($\text{HAuCl}_4 \cdot 3\text{H}_2\text{O}$, > 99.9% metals basis), tetraoctylammonium bromide (TOAB, ≥ 98%), cyclohexane thiol ($\text{S}-c\text{-C}_6\text{H}_{11}$; 97%), sodium borohydride (NaBH_4 , 99.99%, trace metal basis), trans-2-[3-(4-tert-butylphenyl)-2-methyl-2 propenylidene] malononitrile (DCTB, > 99.0%) were purchased from Sigma-Aldrich. MeOH (HPLC grade, 99.9%, Spectrochem), toluene (HPLC grade, 99.9%, Spectrochem), Dimethylene chloride (DCM, HPLC grade, 99.9%, Spectrochem), Tetrahydrofuran (THF, HPLC grade, 99.9%, Spectrochem), 4-Flurothiophenol (*p*-FC₆H₄SH; 98%), 4-Chlorothiophenol (*p*-ClC₆H₄SH; 97%), 4-Bromothiophenol (*p*-BrC₆H₄SH; 95%), Thiophenol (C₆H₅SH; 97%), PTFE syringe filters (0.2 μm, Corning, Germany) and BioBeads S-X1 (BioRad) were used as purchased. The water used in all experiments was ultrapure (resistivity: 18.2 MΩ cm), produced by a Milli-Q NANO pure water system. All glassware was thoroughly cleaned with aqua regia (HCl: HNO₃ = 3:1, v: v) and rinsed with abundant pure water.

1.2. Methods

Synthesis of $[\text{Au}_{23}(\text{S}-c\text{-C}_6\text{H}_{11})_{16}]^-$ nanocluster. The synthesis of $[\text{Au}_{23}(\text{S}-c\text{-C}_6\text{H}_{11})_{16}]^-$ was conducted as per the previously reported method.¹⁹ Firstly, $\text{HAuCl}_4 \cdot 3\text{H}_2\text{O}$ (0.6 mmol, 236.3 mg) and tetraoctylammonium bromide (TOAB, 0.696 mmol, 380 mg) were dissolved in 30 mL of methanol in a 50 mL tri-necked round-bottom flask. The color of the solution changed from yellow to dark reddish-orange upon vigorous stirring for 15 min. To this solution, an excess of 1-cyclohexanethiol was added (3.2 mmol, 390 μL) was added to the mixture at room temperature. The solution turned yellowish from reddish brown almost immediately, indicating the formation of Au(I) from Au(III) complexes. After 15 min, NaBH_4 (6 mmol, 227 mg dissolved freshly in 12 mL of cold nano pure water) was rapidly added to the solution under vigorous stirring. The formation of a black-colored solution indicated the formation of Au NCs, which was further precipitated out of the solution using methanol. The reaction mixture was further allowed to stir overnight which finally yielded the pure $[\text{Au}_{23}(\text{S}-c\text{-C}_6\text{H}_{11})_{16}]^-$ NC.

Transformation of $[\text{Au}_{23}(\text{S}-c\text{-C}_6\text{H}_{11})_{16}]^-$ nanocluster using FBT: 5 mg of $[\text{Au}_{23}(\text{S}-c\text{-C}_6\text{H}_{11})_{16}]^-$ was dissolved in 1 mL of dichloromethane and was centrifuged at (25 °C, 10000 rpm) for 10 minutes to remove any insoluble particles. 0.25 mL *p*-Flurothiophenol was added to the supernatant and heated at 40 °C. The reaction was continued for two hours. Small aliquots were taken from the reaction mixture for the UV-Vis spectroscopy and MALDI-MS measurements. The obtained product was purified using size-exclusion chromatography (SEC), filtered using a 0.2 µm PTFE syringe filter, and dried using rotavapor. Single crystals of the as-prepared NCs were obtained by layering the DCM solution of NCs with hexane in less than a week at room temperature.

Transformation of $[\text{Au}_{23}(\text{S}-c\text{-C}_6\text{H}_{11})_{16}]^-$ nanocluster using CBT/ BBT: 5 mg of $[\text{Au}_{23}(\text{S}-c\text{-C}_6\text{H}_{11})_{16}]^-$ was dissolved in 1 mL of dichloromethane and was centrifuged at (25 °C, 10000 rpm) for 10 minutes to remove any insoluble particles. 290 mg *p*-chlorothiophenol/ 378 mg *p*-bromothiophenol was added to the supernatant and heated at 40 °C. The reaction was continued for three hours. Small aliquots were taken from the reaction mixture for the UV-Vis spectroscopy and MALDI-MS measurements. The obtained product was purified using size-exclusion chromatography (SEC), filtered using a 0.2 µm PTFE syringe filter, and dried using rotavapor. Single crystals of the as-prepared NCs were obtained by layering the DCM solution of NCs with hexane in less than a week at room temperature.

II. Characterization

The absorption spectra were obtained on a UV-3800 SHIMADZU UV-Vis NIR spectrometer using dichloromethane as a solvent. The mass spectra of the NC samples were acquired by using a Brucker Microflex MALDI-TOF mass spectrometer. DCTB was used as a matrix and the sample was prepared as 1 mg in 100 µL of dichloromethane for the same. From the stock solution, various amounts of the stock solution were taken and mixed with 1 µL of the analyte solution. The molecules were ionized with the Nd: YAG laser ($\lambda=266$ nm). The analyte concentration was optimized to get a well-resolved spectrum. ESI mass spectra were recorded using a Waters Q-TOF mass spectrometer equipped with a Z-Spray source. The source temperature was kept at 70 °C. The sample was directly infused into the chamber at 5 µL/min. The spray voltage was kept at 3.0 kV and the cone voltage at 30 V. To prepare the ESI sample, clusters were dissolved in dichloromethane (1 mg/ mL) and diluted (2: 1 v) using methanol solvent. Femtosecond transient absorption measurements were conducted by dissolving the NCs in DCM and exciting with a 400

nm pump with a fluence of 490 nJ/cm² and a white light probe (500-730 nm) was used to measure the absorption in the excited state.

2.1 Single Crystal X-Ray Diffraction

The single crystal of Au₂₅(FC₆H₄S)₁₈ with dimensions of 0.125 mm x 0.225 mm x 0.288 mm was mounted on Bruker D8 Quest - Microfocus SCXRD (single crystal X-ray diffractometer) with Bruker PHOTON III C14, CPAD detector using MoK α radiation ($\lambda = 0.71073 \text{ \AA}$). The total exposure time for data collection was 1.87 hours. The frames were integrated with the Bruker SAINTS software package using a narrow-frame algorithm. The crystal structure was solved and refined by SHELXTL software package.^{S1, S2, S3, S4} The extra framework cation, tetraoctyl ammonium was not detected properly based on the single crystal data. However, we could identify it through ESI-MS data.

III. Computational Details

The density functional theory (DFT) calculations were performed using Generalized gradient approximation as implemented in Perdew–Burke–Ernzerhof (PBE) functional with the help of Vienna Ab Initio Software Package (VASP).^{S5, S6} To describe the electron-ion interactions projector augmented wave (PAW) method was employed along with an energy cutoff of 470 eV.^{S7, S8} We have fit the nanocluster inside a $30 \times 30 \times 30 \text{ \AA}^3$ cubic cell to avoid periodic interactions. A gamma-centered ($1 \times 1 \times 1$) k-point was used to sample the Brillouin zone. DFT-D3 dispersion correction as implemented by Grimme was used.^{S9} The equations used for the calculation of cohesive energy of the core (E_c) and shell-to-core binding energy (E_{sc}) are as follows:

$$E_c = \frac{E_{\text{cluster}} - n \times E_{\text{Au}} - E_{\text{shell}}}{n + n_{\text{shell-int}}}$$

$$E_{sc} = \frac{E_{\text{cluster}} - E_{\text{core}} - E_{\text{shell}}}{n_{\text{shell-int}}}$$

Where E_{cluster} is the total energy of the cluster, E_{Au} is the energy of a single gold atom, E_{core} is the energies of gold core and E_{shell} is the single point energies of the surrounding shell at the optimized geometries of the clusters, n is the number of core gold atoms and $n_{\text{shell-int}}$ is the number of shell units interacting with the core.

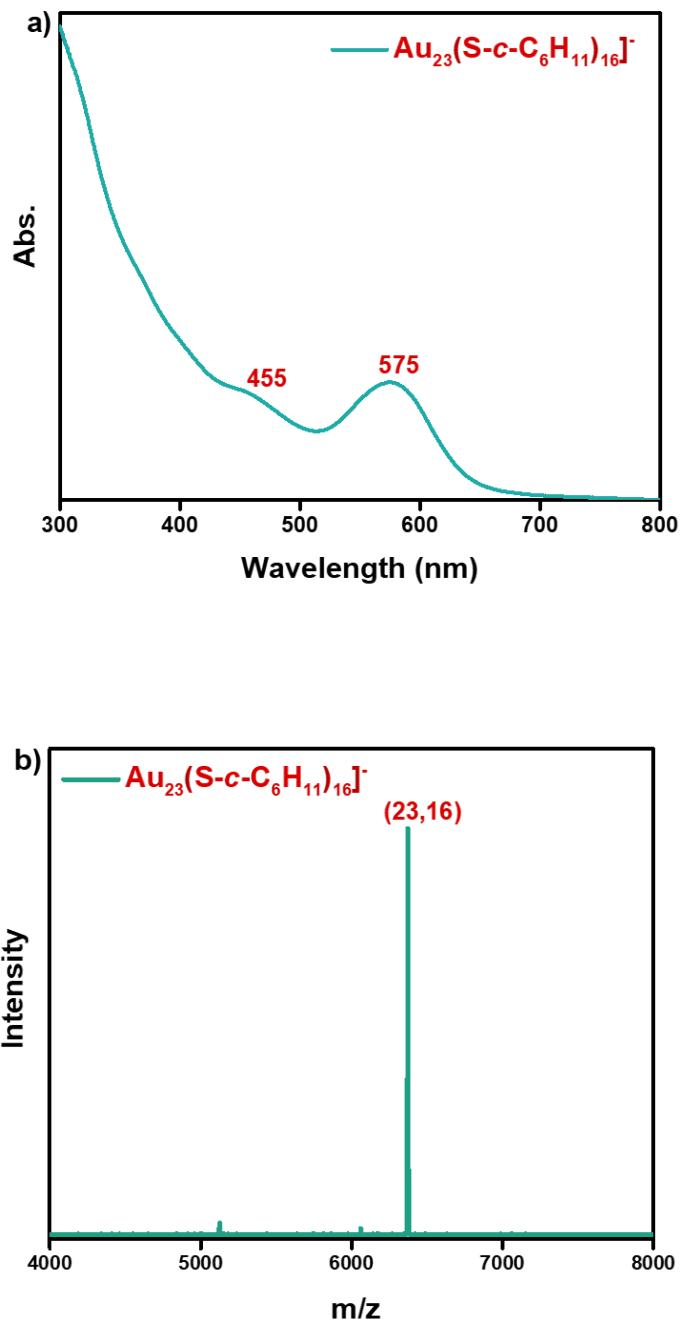
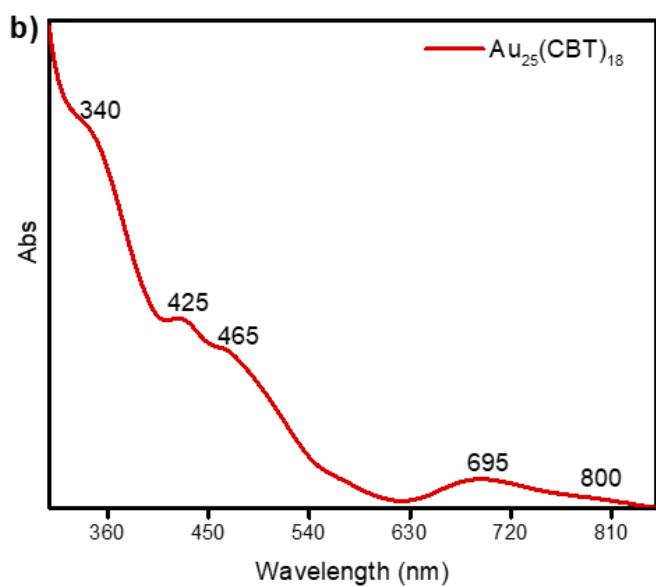
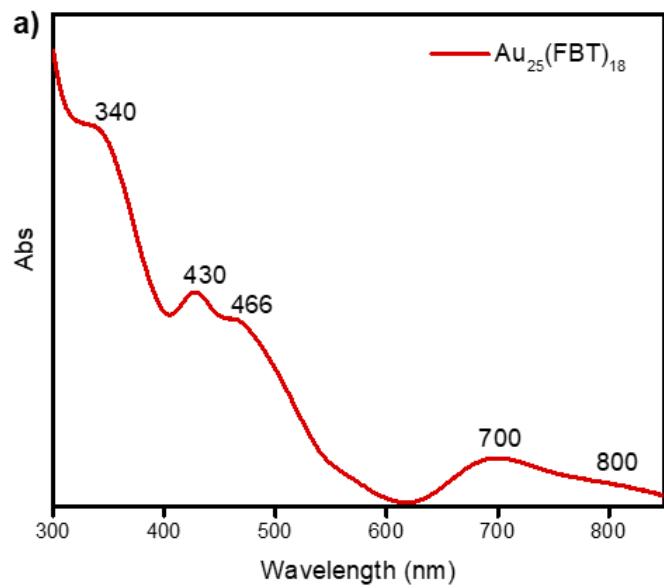


Figure S1. a) UV-vis absorption spectrum and b) MALDI-MS spectrum of the synthesized $[\text{Au}_{23}(\text{S}-\text{c}-\text{C}_6\text{H}_{11})_{16}]^+$ NC.



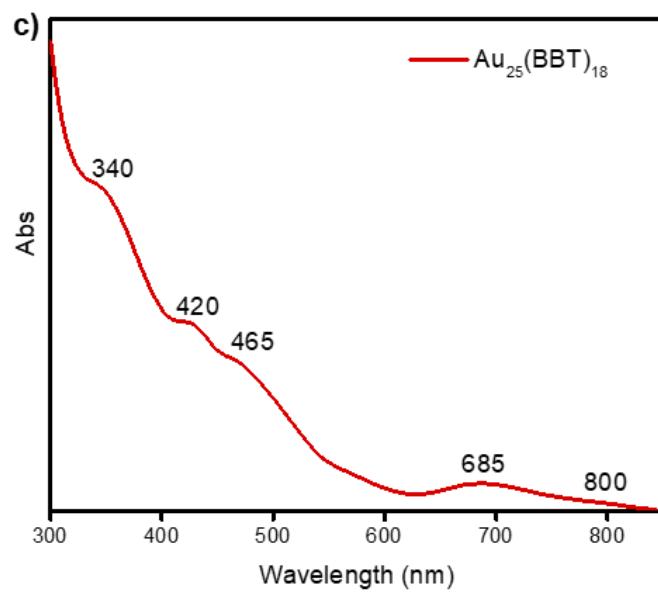
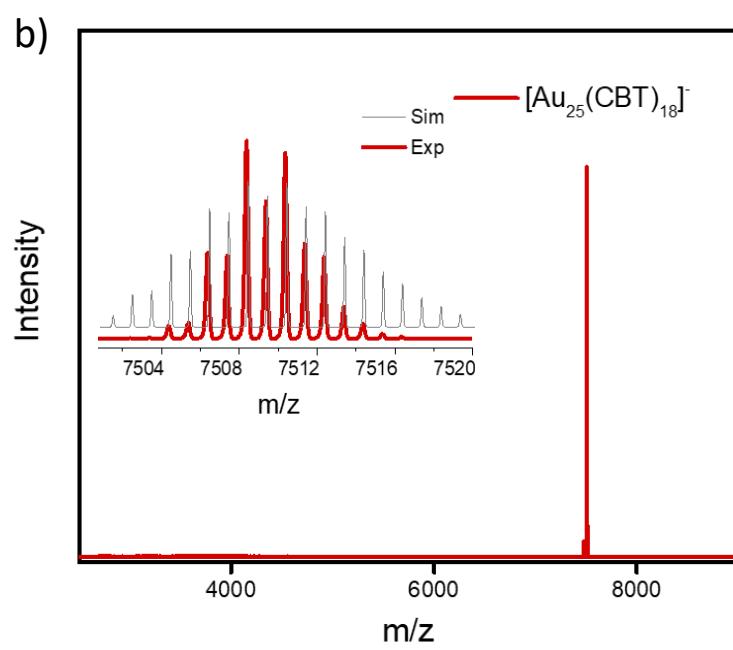
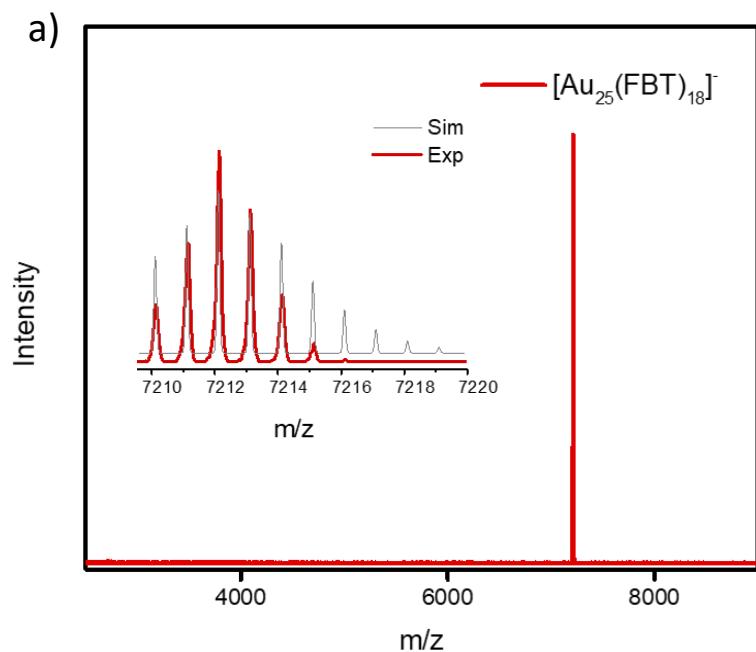


Figure S2. UV-vis absorption spectra of a) $[\text{Au}_{25}(\text{FBT})_{18}]^-$, b) $[\text{Au}_{25}(\text{CBT})_{18}]^-$ and c) $[\text{Au}_{25}(\text{BBT})_{18}]^-$ NCs.



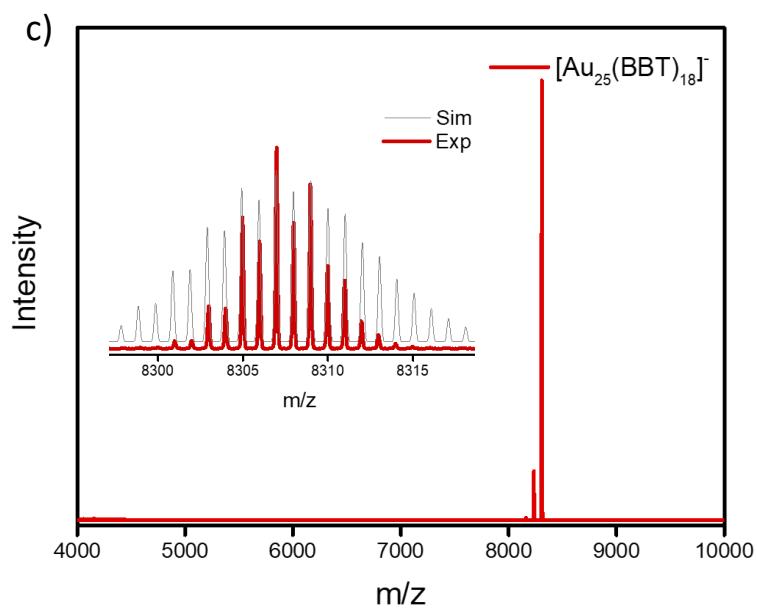


Figure S3. Negative-ion mode ESI mass spectra of a) $[\text{Au}_{25}(\text{FBT})_{18}]^-$, b) $[\text{Au}_{25}(\text{CBT})_{18}]^-$, and c) $[\text{Au}_{25}(\text{BBT})_{18}]^-$ NCs. The inset shows the respective experimental and simulated isotopic patterns.

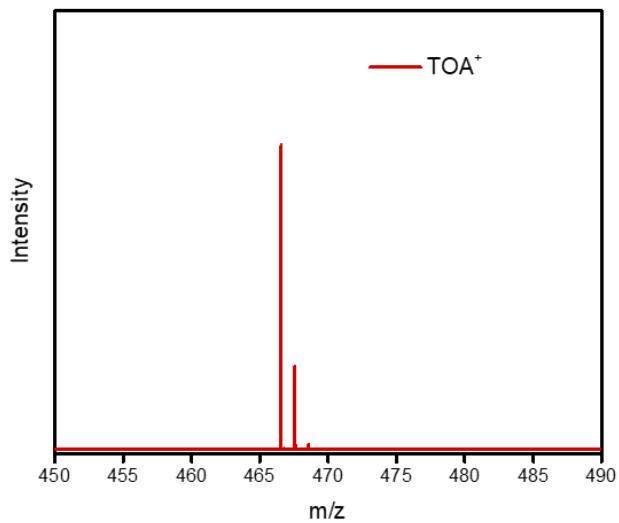


Figure S4. Positive-ion mode ESI mass spectra of $[\text{Au}_{25}(\text{XBT})_{18}]^-$ NCs showing the presence of TOA⁺ counter cation.

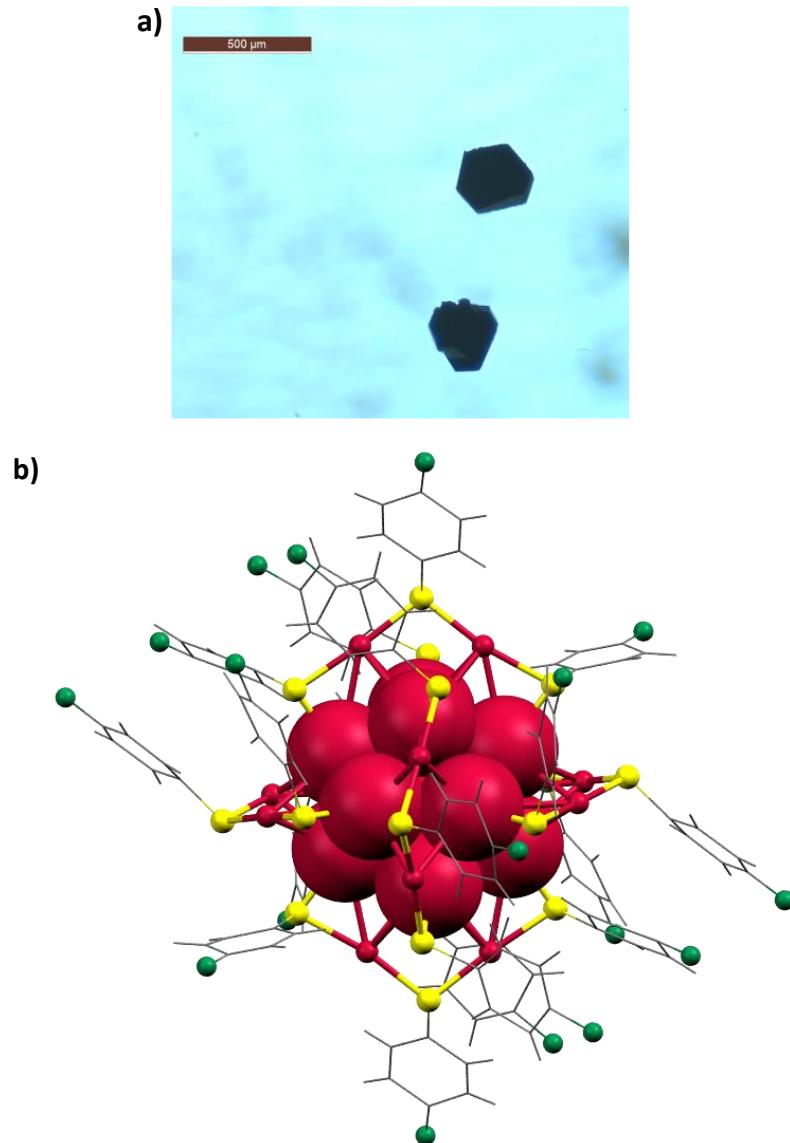


Figure S5. a) Optical microscope image of the single crystals of $\text{Au}_{25}(\text{FBT})_{18}$ NC. b) Total structure of the $[\text{Au}_{25}(\text{FBT})_{18}]^-$ NC. Color legend: red sphere, Au; yellow sphere, S; green sphere, F; grey stick, C and H.

Table S1. Crystal structure refinement parameters of $[\text{Au}_{25}(\text{FBT})_{18}]^- \text{NC}$.

Identification code	0061_260324_SG53A
Empirical formula	C108 H72 Au25 F18 S18
CCDC number	2334107
Formula weight	7212.89 g/mol
Temperature/K	100
Wavelength/ \AA	0.71073
Crystal system	Monoclinic
Space group	$C2/c$ (No. 15)
a/ \AA	29.6682(14)
b/ \AA	17.7115(9)
c/ \AA	34.7181(18)
$\alpha/^\circ$	90
$\beta/^\circ$	114.038(2)
$\gamma/^\circ$	90
Volume/ \AA^3	16661.1(15)
Z	4
$\rho_{\text{calc}} \text{g/cm}^3$	2.876
μ/mm^{-1}	22.190
F(000)	12580.0
Crystal size/mm ³	0.125 \times 0.225 \times 0.288
2 θ range for data collection/ $^\circ$	2.08 to 25.03
Index ranges	-35 \leq h \leq 32, -21 \leq k \leq 21, -41 \leq l \leq 41
Reflections collected	185497
Independent reflections	14690 [R(int) = 0.0498]
Data/restraints/parameters	14690/80/791
Goodness-of-fit on F^2	1.167
Final R indexes [$I >= 2\sigma(I)$]	R ₁ = 0.0610, wR ₂ = 0.1451
Largest diff. peak/hole / e \AA^{-3}	3.328 and -2.689

Table S2. Selected bond distances in $[\text{Au}_{25}(\text{FBT})_{18}]^- \text{NC}$.

Atoms	Bond length (Å)
Au(01) – Au(6A)	2.7534(8)
Au(01) – Au(6A)	2.7534(8)
Au(01) – Au(03)	2.7596(7)
Au(01) – Au(03)	2.7596(7)
Au(01) – Au(04)	2.7790(8)
Au(01) – Au(04)	2.7790(8)
Au(01) – Au(2A)	2.7923(7)
Au(01) – Au(2A)	2.7923(7)
Au(01) – Au(07)	2.7934(8)
Au(01) – Au(07)	2.7934(8)
Au(01) – Au(05)	2.8106(7)
Au(01) – Au(05)	2.8107(7)
Au(2A) – Au(03)	2.8113(11)
Au(2A) – Au(07)	2.9068(12)
Au(2A) – Au(6A)	2.9334(11)
Au(2A) – Au(04)	2.9498(11)
Au(2A) – Au(05)	2.9873(11)
Au(2A) – Au(10)	3.0754(12)
Au(2A) – Au(09)	3.1011(11)
Au(2A) – Au(11)	3.1864(11)
Au(6A) – Au(05)	2.8110(10)
Au(6A) – Au(07)	2.9563(11)
Au(6A) – Au(04)	2.9639(11)
Au(6A) – Au(03)	2.9825(11)
Au(6A) – Au(3C)	3.0368(11)
Au(6A) – Au(10)	3.0762(11)
Au(6A) – Au(13)	3.1576(11)
Au(3C) – Au(05)	3.1574(11)
Au(3C) – Au(07)	3.1819(11)
Au(2B) – Au(10)	2.20(5)
Au(2B) – Au(6B)	2.50(5)
Au(2B) – Au(03)	2.55(4)
Au(2B) – Au(09)	3.11(4)
Au(2B) – Au(04)	3.29(4)
Au(6B) – Au(13)	2.46(4)
Au(6B) – Au(04)	2.51(4)
Au(6B) – Au(3D)	2.75(6)
Au(6B) – Au(05)	3.14(4)
Au(6B) – Au(10)	3.21(4)
Au(3D) – Au(05)	2.23(4)

Au(3D) – Au(07)	2.60(4)
Au(3D) – Au(13)	3.06(4)
Au(03) – Au(04)	2.8924(11)
Au(03) – Au(05)	2.9065(11)
Au(03) – Au(07)	3.0674(11)
Au(03) – Au(09)	3.1089(11)
Au(03) – Au(08)	3.1211(11)
Au(03) – Au(10)	3.1220(11)
Au(04) – Au(07)	2.8334(11)
Au(04) – Au(05)	2.9886(11)
Au(04) – Au(11)	3.0933(11)
Au(04) – Au(13)	3.1515(11)
Au(04) – Au(08)	3.1746(11)
Au(05) – Au(07)	2.8910(11)
Au(05) – Au(13)	3.0976(11)
Au(05) – Au(09)	3.1894(11)
Au(07) – Au(08)	3.0335(11)
Au(07) – Au(11)	3.1132(11)
Au(6A) – S(01)	2.410(5)
Au(3C) – S(06)	2.298(6)
Au(3C) – S(03)	2.316(6)
Au(2B) – S(08)	2.51(4)
Au(6B) – S(01)	2.24(4)
Au(3D) – S(06)	2.53(4)
Au(3D) – S(03)	2.66(4)
Au(3D) – S(02)	2.412(5)
Au(04) – S(09)	2.385(5)
Au(05) – S(04)	2.396(5)
Au(07) – S(03)	2.415(5)
Au(08) – S(05)	2.319(5)
Au(08) – S(02)	2.332(5)
Au(09) – S(07)	2.304(6)
Au(09) – S(04)	2.312(6)
Au(10) – S(07)	2.297(5)
Au(10) – S(01)	2.318(5)
Au(11) – S(08)	2.289(6)
Au(11) – S(05)	2.311(5)
Au(13) – S(06)	2.294(6)
Au(13) – S(09)	2.317(6)

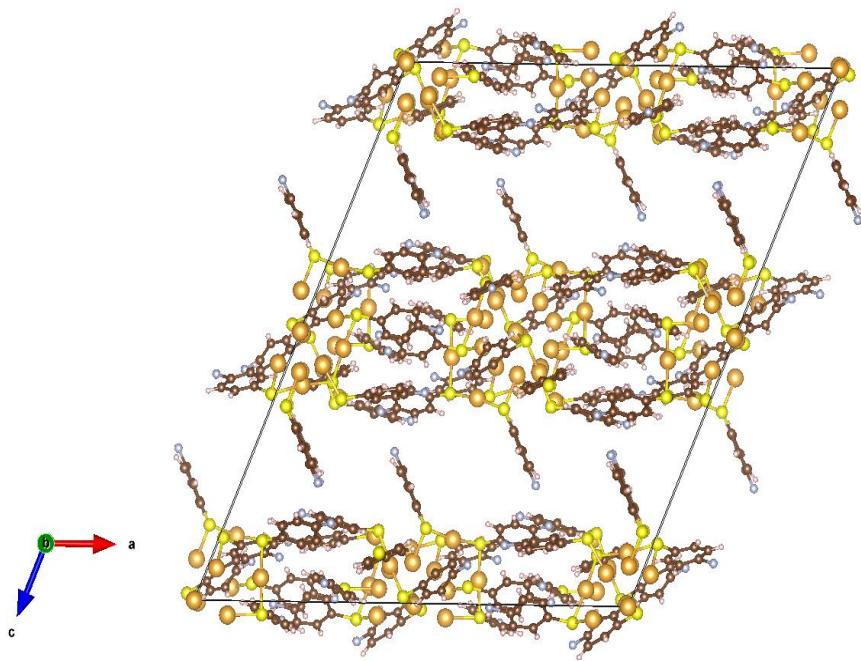
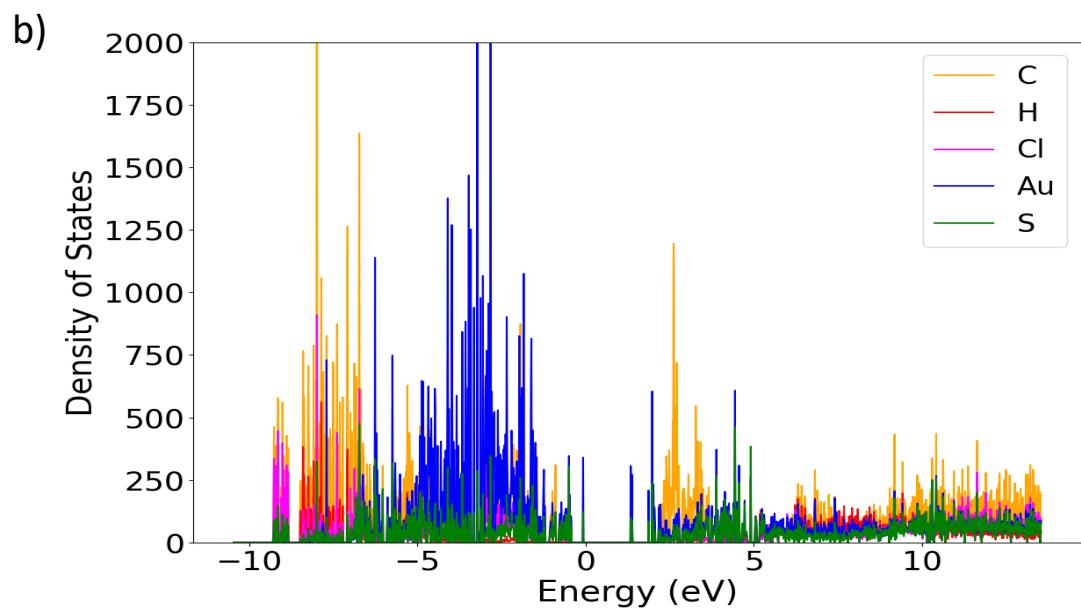
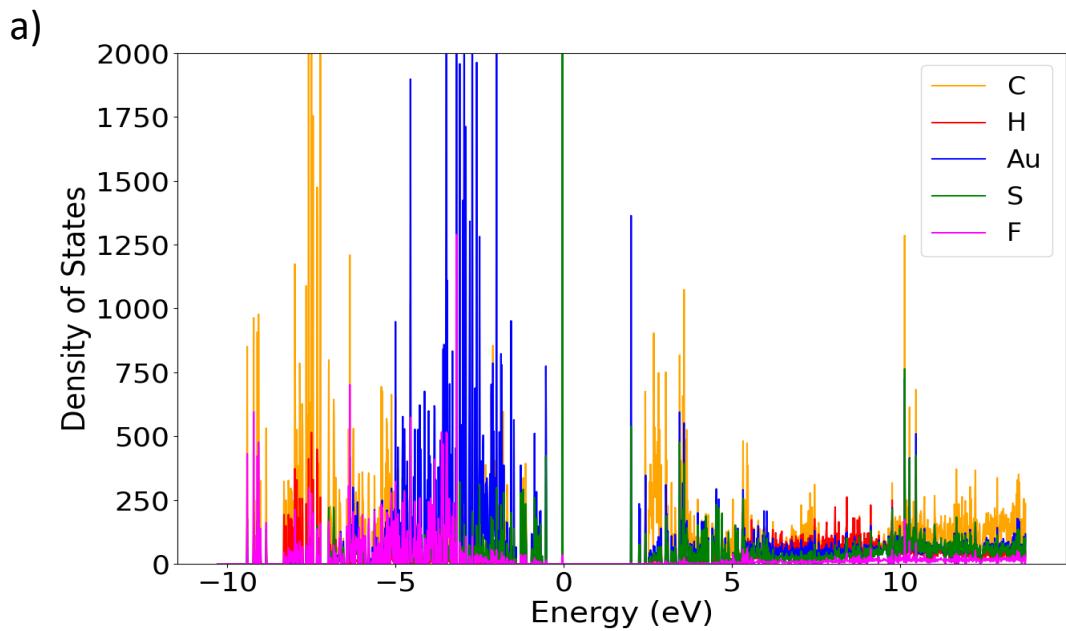


Figure S6. Relaxed structure of the $[\text{Au}_{25}(\text{XBT})_{18}]^-$ NCs, X = F, Cl, Br. Color legend: Golden yellow, Au; yellow, S; Blue, F, Cl, Br; Brown, C.

Table S3. Electronic properties of the $[Au_{25}(XBT)_{18}]^-$ NCs, X = F, Cl, Br.

Lattice vectors	a= 29.899 b= 17.874 c= 35.269	a= 29.899 b= 17.874 c= 35.269	a= 29.899 b= 17.874 c= 35.269
Chemical Formula	$Au_{100}S_{72}C_{432}H_{288}F_{72}$	$Au_{100}S_{72}C_{432}H_{288}Cl_{72}$	$Au_{100}S_{72}C_{432}H_{288}Br_{72}$
Number of Atoms	964	964	964
Total Energy (eV)	-54996566.84262	-55706315.966422088	-59941358.851653107
Fermi Energy (eV)	-5.72342	-5.54215765	-5.48438
Highest occupied state (eV)	-5.72342	-5.55704335	-5.492457
Lowest unoccupied state (eV)	-4.36278	-4.20388263	-4.16609
Estimated HOMO-LUMO gap (eV)	1.37640	1.35316072	1.32630
Cell Volume (\AA^3)	17252.96791	17252.96791	17252.96791



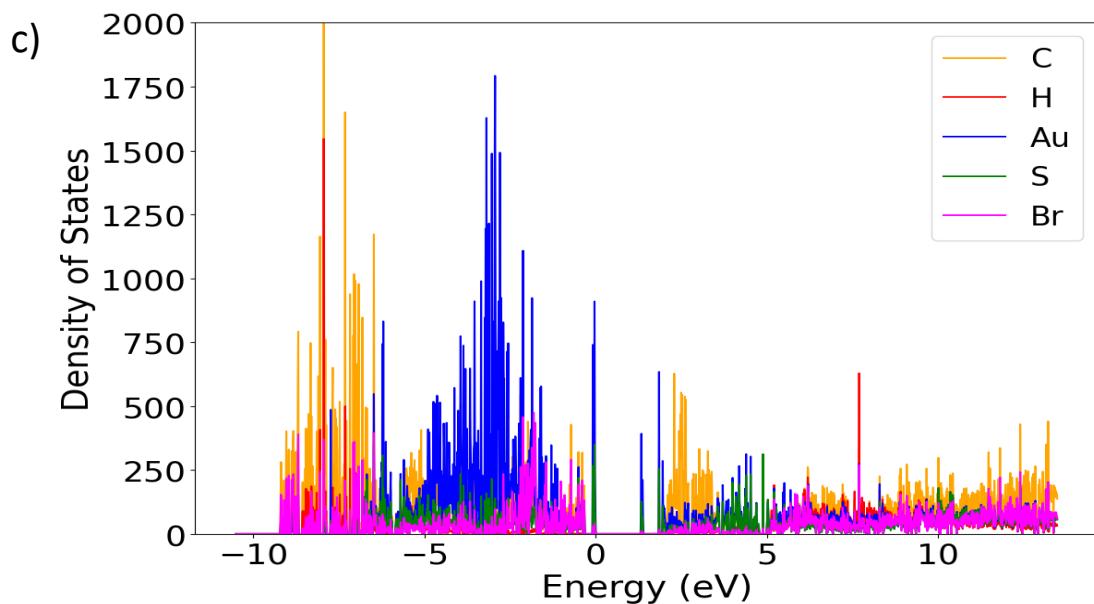


Figure S7. Density of states of the a) [Au₂₅(FBT)₁₈]⁻, b) [Au₂₅(CBT)₁₈]⁻, c) [Au₂₅(BBT)₁₈]⁻ NCs.

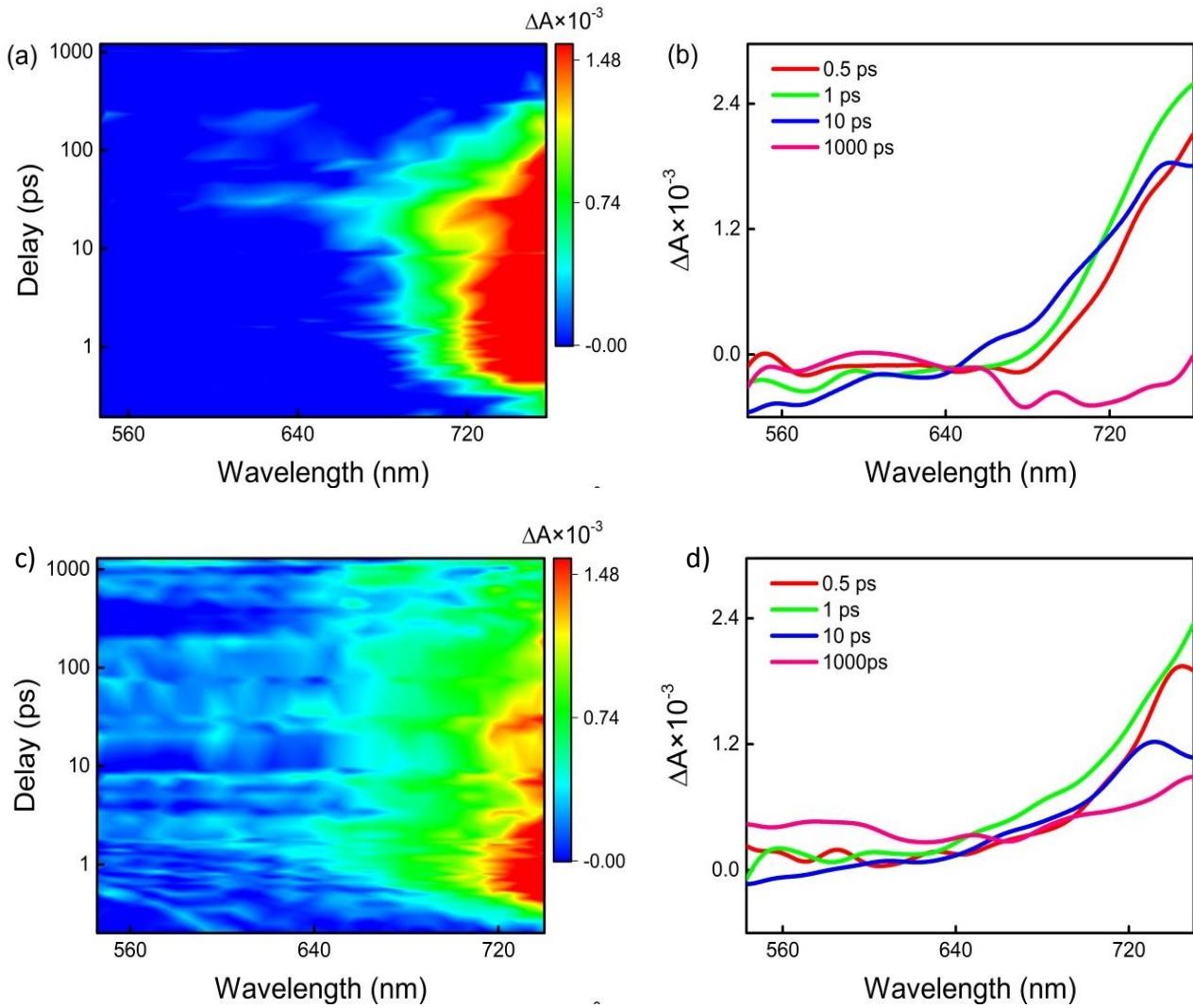


Figure S8. Contour plot illustrating the ultrafast transient absorption spectrum of (a) $[\text{Au}_{23}(\text{S}-c\text{-C}_6\text{H}_{11})_{16}]^-$ NC and (c) $\text{Au}_{23}(\text{S}-c\text{-C}_6\text{H}_{11})_{17}$ NCs dissolved in dichloromethane (DCM), with (b & d) corresponding cross-sections of the contour plot at selected probe delay times. The excitation wavelength is 400 nm, and the excitation fluence is 490 nJ/cm².

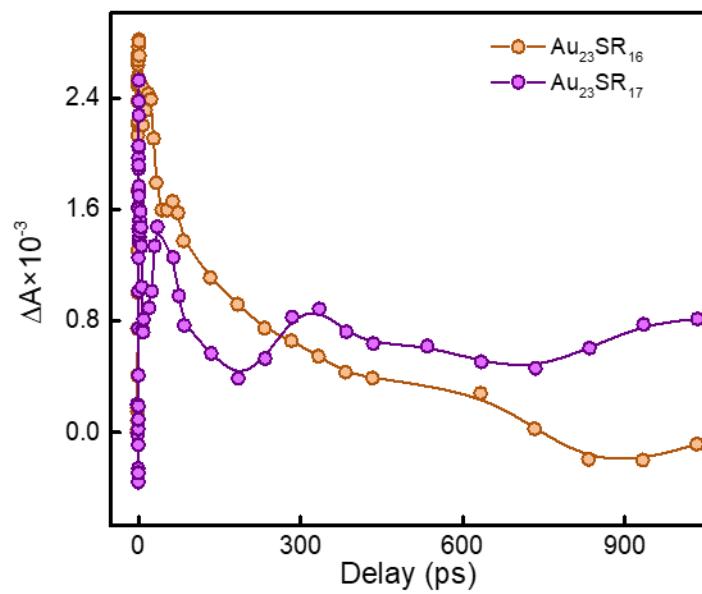
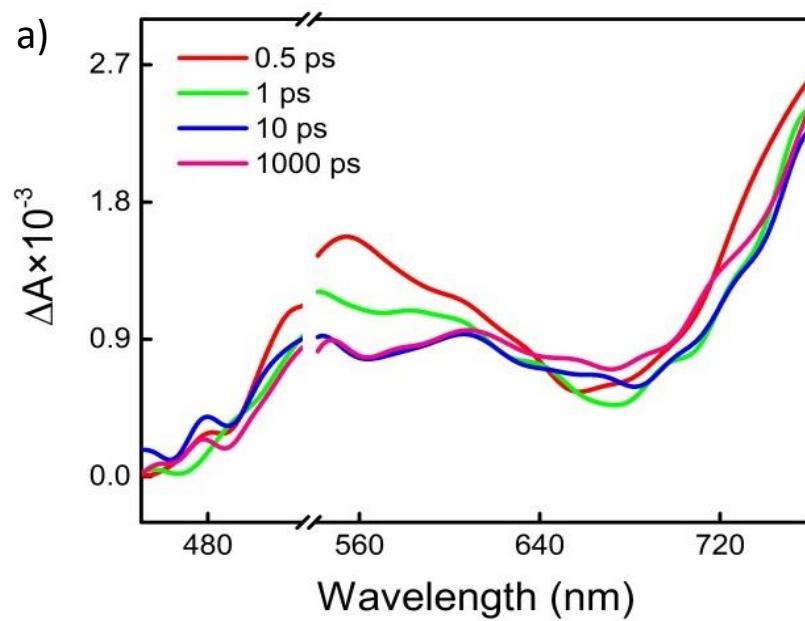


Figure S9. TA decay kinetics of the $[\text{Au}_{23}(\text{S}-c\text{-C}_6\text{H}_{11})_{16}]^-$ and the $\text{Au}_{23}(\text{S}-c\text{-C}_6\text{H}_{11})_{17}$ species at 750 nm.



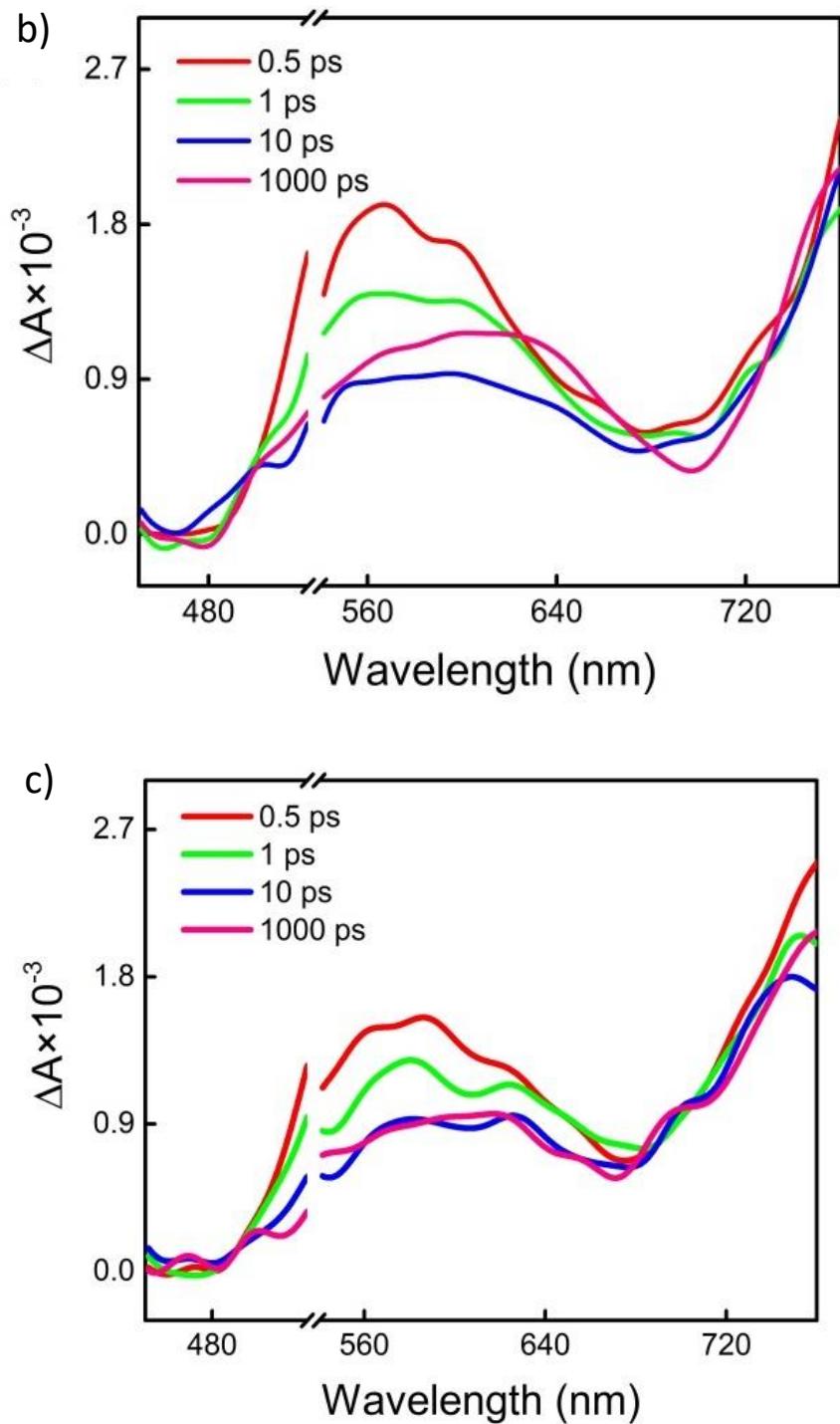


Figure S10. The time-associated femtosecond transient absorption spectra of the a) $[\text{Au}_{25}(\text{FBT})_{18}]^-$, b) $[\text{Au}_{25}(\text{CBT})_{18}]^-$, c) $[\text{Au}_{25}(\text{BBT})_{18}]^-$ NCs at different delay times.

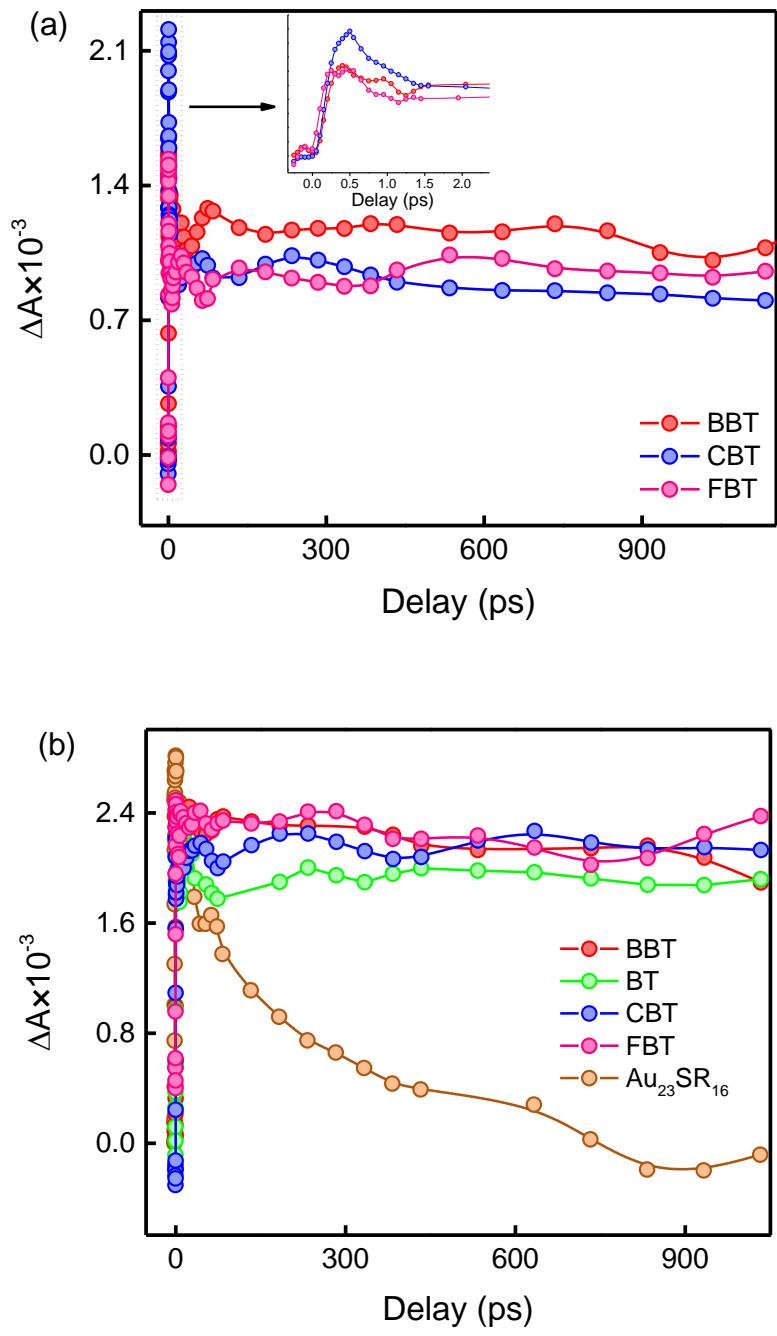


Figure S11. TA decay kinetics of the $[Au_{25}(XBT)_{18}]$ NCs at (a) 560 nm. TA decay kinetics of the $[Au_{25}(XBT)_{18}]$ NCs and the $[Au_{25}(S-c-C_6H_{11})_{16}]$ at (b) 750 nm. Inset of (a) shows the early time decay feature of PIA.

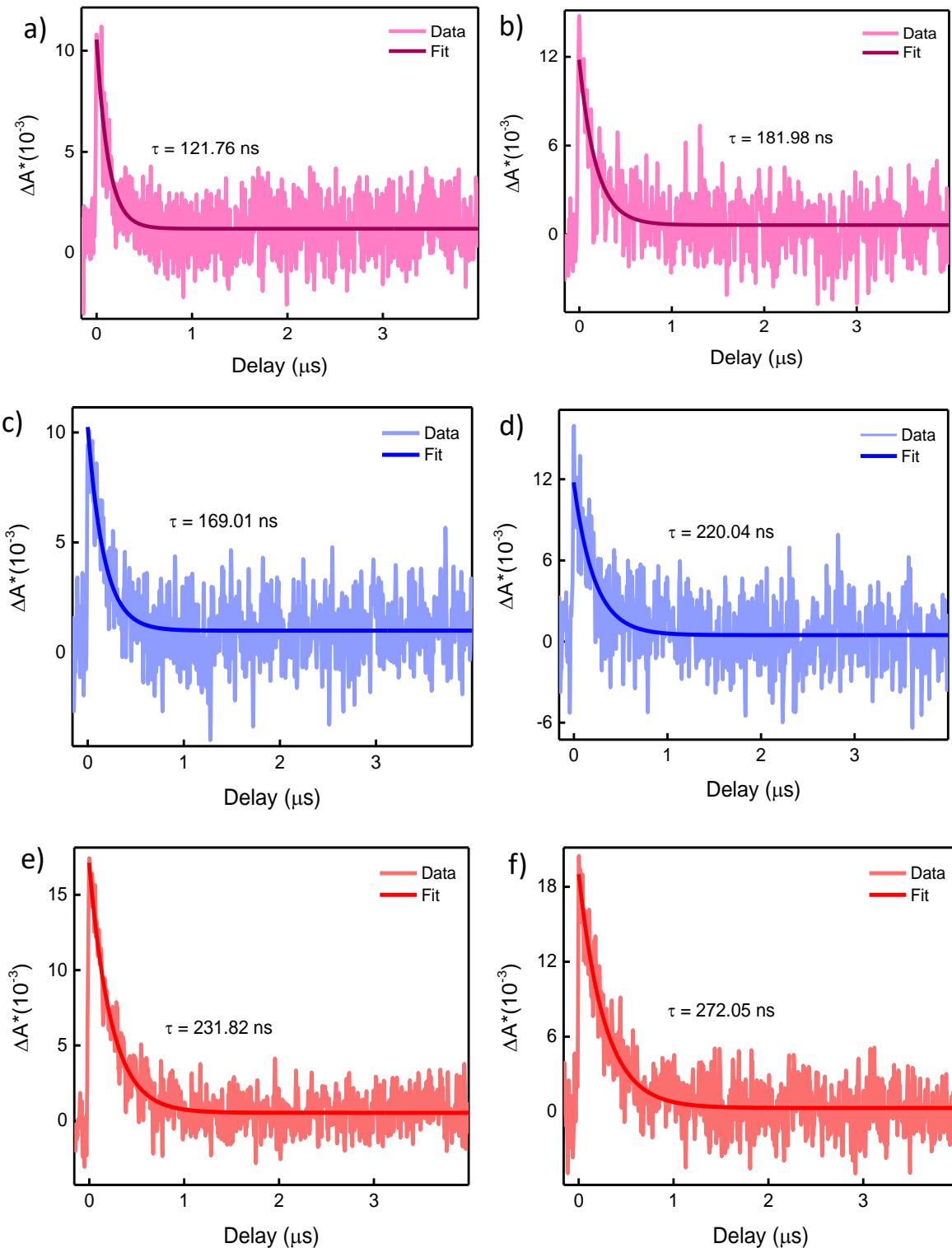


Figure S12. Nanosecond transient absorption decay kinetics of a), b) $[\text{Au}_{25}(\text{FBT})_{18}]^-$ NC; c), d) $[\text{Au}_{25}(\text{CBT})_{18}]^-$ NC; e), f) $[\text{Au}_{25}(\text{BBT})_{18}]^-$ NCs at 560 nm and 750 nm wavelength, respectively.

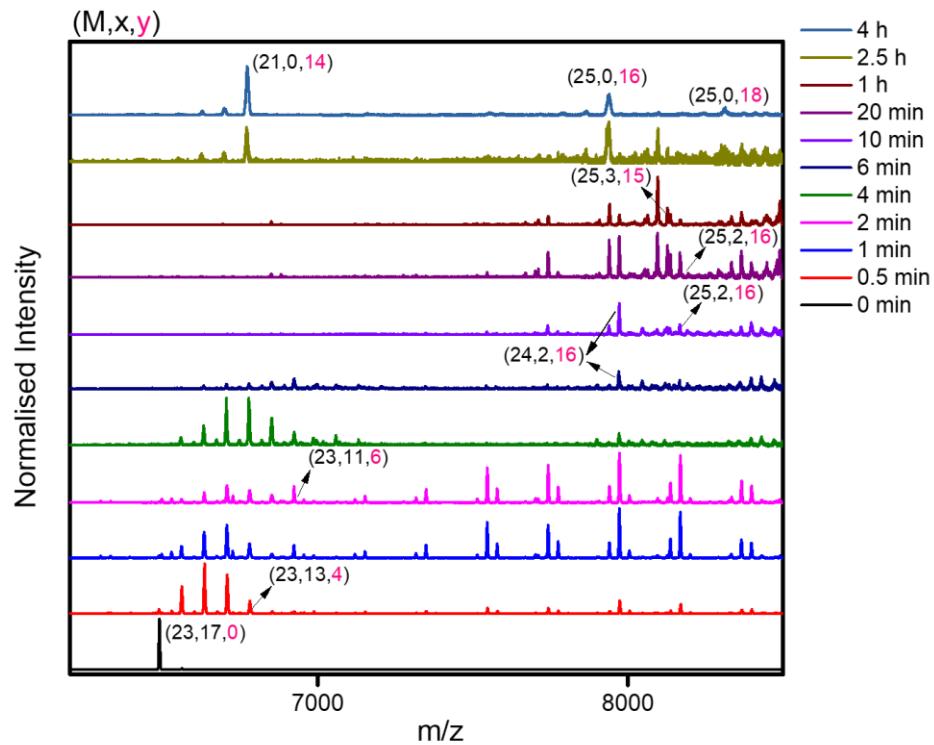


Figure S13 a. Time-dependent MALDI-MS spectra of the aliquots taken during the transformation reaction of $\text{Au}_{23}(\text{S}-c\text{-C}_6\text{H}_{11})_{17}$ intermediate species with BBT. ($M = \text{number of metal atoms}$, $x = \text{number of cyclohexanethiolate ligands}$, $y = \text{number of bromobzenethiolate ligands in each species}$).

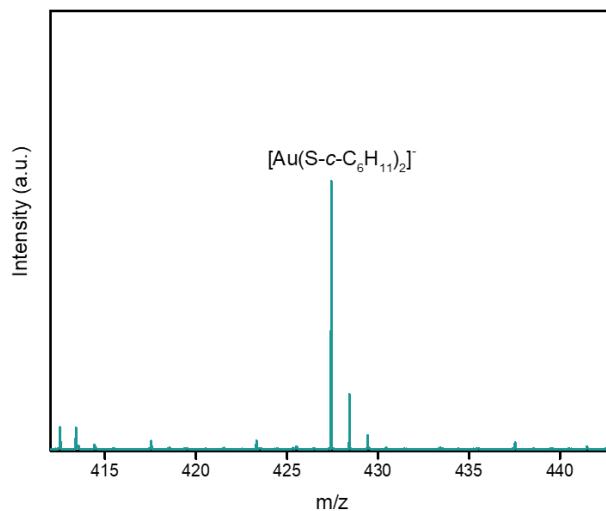


Figure S13 b. Negative mode ESI-MS spectrum of the $[\text{Au}_{23}(\text{S}-c\text{-C}_6\text{H}_{11})_{16}]^-$ nanoclusters at low range showing the presence of $\text{Au}(\text{S}-c\text{-C}_6\text{H}_{11})_2$ complex.

Table S4. Optimized structure coordinates of a) $[\text{Au}_{23}(\text{SC}_6\text{H}_{11})_{16}]^-$ NC, b) $\text{Au}_{23}(\text{SC}_6\text{H}_{11})_{17}$, c) $\text{Au}_{23}(\text{SC}_6\text{H}_{11})_{11}(\text{BBT})_6$, d) $\text{Au}_{24}(\text{SC}_6\text{H}_{11})_2(\text{BBT})_{16}$, e) $\text{Au}_{25}(\text{SC}_6\text{H}_{11})_2(\text{BBT})_{16}$ and f) $[\text{Au}_{25}(\text{BBT})_{18}]^-$ NCs.

a) $\text{Au}_{23}(\text{S}-c\text{-C}_6\text{H}_{11})_{16}$

Au	15.318386151	14.859476536	15.025961020
Au	14.447284570	12.224959686	14.863527453
Au	15.967930274	17.601810981	14.917166678
Au	16.504536438	13.074043281	13.249575970
Au	13.998098032	16.733273335	16.597286586
Au	17.968017113	15.684234418	16.134212154
Au	12.362642886	14.044246017	13.551292102
Au	12.786353950	14.090446579	16.318446428
Au	17.700752324	15.729175244	13.434698520
Au	16.879185930	12.811254568	16.001980282
Au	13.580341428	16.666370187	13.830798521
Au	15.429488818	15.094197671	18.458073653
Au	15.000455134	14.862316658	11.444304342
Au	19.825749147	13.358722911	17.695943094
Au	10.443568741	16.179727663	12.263315010
Au	19.482551714	14.038997815	14.634655080
Au	10.875717161	15.618687022	15.122520106
Au	19.239272012	11.814118125	12.720858675
Au	11.302598878	17.872422035	17.057834856
Au	19.164315297	18.296354055	14.195912140
Au	11.107849880	11.734659456	15.606409317
Au	21.166843754	16.105834297	15.894804183
Au	8.830324843	13.714775179	14.271935772
S	17.523907108	16.082252351	18.422192469
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S	18.351040772	11.598814523	17.377324218

S	12.338081544	17.498396491	11.988997675
S	17.262203974	19.548524719	14.625625915
S	12.852901209	10.497731784	14.716629860
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S	16.630024893	16.520081110	11.442307308
S	17.367965599	11.801881595	11.382049175
S	13.333682025	18.898926720	17.468372485
S	20.914988455	12.132449710	14.280030496
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S	21.220537279	15.187463995	18.028672735
S	8.660571148	14.674771048	12.178237961
S	21.141187193	17.160468058	13.812510888
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H	16.385475124	17.800197114	20.564882765
H	18.133733807	17.592452254	20.716001335

b) Au₂₃(S-c-C₆H₁₁)₁₇

Au	14.595505462	13.611298533	15.165907741
Au	13.258756896	11.092075003	15.081562088
Au	15.364255709	16.384732751	14.915254865
Au	15.660531938	11.777315459	13.403761261
Au	13.358677702	15.651443266	16.573570024
Au	17.283436706	14.472883102	16.248306584
Au	11.512295332	12.899694651	13.753593917
Au	11.958230549	13.124052091	16.536409238
Au	16.942635555	14.375278175	13.504540594
Au	16.540248273	11.774966985	16.075611283
Au	13.004607662	15.369012763	13.806576807
Au	14.712863561	14.134926966	18.582726674
Au	14.218759282	13.492378360	11.590156700

Au	19.551415553	12.619016318	17.856939851
Au	9.779112082	15.019056502	12.388754099
Au	18.864794910	12.890065102	14.765768646
Au	10.013198780	14.483731775	15.302782462
Au	18.709028226	10.589783409	12.902596231
Au	10.584565185	16.819196659	17.205513909
Au	18.561785697	16.869768854	14.053463100
Au	9.606262696	10.879706143	15.831479786
Au	20.719826654	15.092947260	15.879921043
Au	7.708908375	12.927916819	14.412831468
S	16.825617838	15.102399243	18.494485499
S	12.565970802	11.875032448	11.839257959
S	18.346749121	10.709630049	17.477324053
S	11.693487859	16.304269704	12.054412415
S	16.742310706	18.242551651	14.433916860
S	11.309558239	9.499412130	15.123558306
S	12.859845496	12.738621742	18.692745771
S	15.868081809	15.124544559	11.507538366
S	16.607171045	9.871492207	12.303590871
S	12.653896220	17.816174459	17.381284544
S	20.584914214	11.395667635	14.022820312
S	8.601241473	15.704440649	16.828927419
S	20.854094808	14.517688345	18.122160440
S	7.822924579	13.746954883	12.264304722
S	20.574756141	15.788317832	13.667898402
S	7.754384234	12.077215910	16.574105448
S	14.948822840	9.951215093	16.384035452
C	22.557343204	13.886505721	18.552955097
C	23.097245383	12.747670719	17.677585674
C	23.658671585	13.211260426	16.332148807

C	24.684942273	14.335944842	16.501916050
C	24.093138557	15.507772057	17.293716096
C	23.554512032	15.051384019	18.653512307
C	21.666685136	12.369732510	12.888305782
C	21.003600922	12.798973136	11.580038598
C	21.978512366	13.618859817	10.723510275
C	23.303809275	12.884467131	10.490319296
C	23.937901896	12.437371863	11.812291004
C	22.960341601	11.588953223	12.637794940
C	21.789778633	17.181117633	13.504945233
C	21.512377810	17.958126864	12.215598461
C	22.540101005	19.083607544	12.030777814
C	23.976525683	18.549162087	12.064570078
C	24.241649315	17.750413147	13.345475570
C	23.214823772	16.624098826	13.525209936
C	17.057183589	14.834544764	10.125884279
C	18.183904136	15.866137768	10.259110380
C	19.160356857	15.782927264	9.080750187
C	18.436615794	15.928024029	7.739158283
C	17.325902401	14.880947032	7.609420891
C	16.335457653	14.957741819	8.780950702
C	16.157164325	18.710599836	12.741761844
C	17.233948310	19.492582435	11.985723925
C	16.722087120	19.909895190	10.599964624
C	15.419017620	20.709689798	10.700231585
C	14.352626280	19.929310280	11.476334977
C	14.856283594	19.508333151	12.864768590
C	11.257952973	18.035900418	12.511915191
C	11.040979490	18.247625201	14.005046408
C	10.735200440	19.714749176	14.320626554

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C	7.467385374	16.841943240	15.901378846
C	7.393659347	18.218608283	16.567242036
C	6.352836079	19.117115509	15.884950024
C	4.973874075	18.451311957	15.844381542
C	5.055313176	17.083086987	15.161467208
C	6.089703559	16.173228744	15.836905221
C	6.441316513	14.962117528	12.077829391
C	5.109172525	14.211368334	12.177772785
C	3.917942273	15.144457090	11.918567624
C	4.057595681	15.883299749	10.583400440
C	5.389929269	16.638095399	10.509705342
C	6.579945912	15.696351334	10.741255554
C	6.402119983	10.829149128	16.749617919
C	5.062993195	11.563574854	16.865413807
C	3.914817445	10.583488579	17.143585693
C	4.194787337	9.729918380	18.385520652
C	5.532672657	8.992747821	18.257323220
C	6.686604688	9.966746315	17.983772740
C	11.540620478	13.492343173	19.747619006
C	11.993955879	13.605752649	21.203773169
C	10.860953476	14.173208209	22.073797578
C	9.578523382	13.341762245	21.951331125
C	9.146035302	13.209675530	20.487285793
C	10.274044447	12.639606953	19.619674080
C	12.943552623	18.201547204	19.155932066
C	12.627683572	17.065513644	20.122210489
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C	12.152310199	18.759775757	21.948905591
C	12.465263018	19.894392084	20.967451810
C	12.173997016	19.476994745	19.518778728
C	17.403290152	10.257099379	18.990302464
C	16.832367880	11.443474494	19.758255887
C	15.994055107	10.971374483	20.950658971
C	16.804862476	10.051782198	21.870470013
C	17.400225166	8.872300190	21.092399085
C	18.239723493	9.348147061	19.895823756
C	16.338642973	9.867925977	10.480427497
C	16.686657010	11.182880720	9.790267952
C	16.341042072	11.132046047	8.296505136
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C	10.763178824	7.221814251	16.613041527
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C	11.627310864	9.198442914	17.934063021
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C	11.353923981	13.173514843	9.610112948
C	10.468466121	13.027233630	8.366140578
C	9.100400277	12.437144551	8.725205297
C	9.245991228	11.096932737	9.455443404
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C	16.750792541	16.947241430	18.545989450
C	18.053426235	17.505935800	17.967991027
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C	17.937911895	19.473275784	19.556032387
C	16.636530881	18.916061598	20.141035892
C	16.561357853	17.388411894	20.000290713
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C	14.775024398	7.821148296	14.534314252
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H	24.192478784	18.426224215	14.217356429
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H	17.738168509	16.875667750	10.294027339
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H	19.150285535	15.832447261	6.905899075
H	17.997800015	16.939016345	7.666855968
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H	5.027342947	13.737419528	13.168515677
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H	11.190348631	14.227929302	23.123206804
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H	13.962450700	7.409843939	15.155619921
H	14.738810685	6.114510079	13.200624673
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H	16.723799789	4.937150536	14.231842332
H	15.444178558	5.222908315	15.419404013
H	17.654421147	5.841281332	16.394630937
H	17.967940963	6.937631286	15.042013092
H	17.199526789	8.217427605	17.003641446
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c) **Au₂₃(S-*c*-C₆H₁₁)₁₁BBT₆**

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Au	13.209117328	15.594979795	16.629998655
Au	17.322191355	14.532992709	16.319382573
Au	11.467901214	12.742672730	13.786830281
Au	12.091524591	12.703822132	16.450995964
Au	16.928832144	14.233054830	13.582111947
Au	16.490762070	11.875459754	16.345434012
Au	12.865075198	15.292742184	13.918076673
Au	14.782787897	13.978124374	18.487729284

Au	14.182687507	13.251007830	11.725608582
Au	19.544236560	12.912274842	18.033118937
Au	9.659649214	14.829579564	12.482563335
Au	18.848683025	12.827268659	14.936770629
Au	10.003939079	14.151118083	15.479292537
Au	18.484641905	10.389129646	13.214555660
Au	10.448451396	16.194285264	17.767249788
Au	18.464583840	16.722750676	13.950721826
Au	9.625587998	10.496793296	15.548084144
Au	20.695423457	15.131101103	15.800469509
Au	7.689688084	12.603072366	14.499423393
S	16.731604549	15.265146200	18.506833966
S	12.545067758	11.633116659	11.968236226
S	18.294352588	11.003904320	17.867225673
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S	16.639141894	18.094333427	14.309709564
S	11.258953996	9.375994776	14.351152129
S	13.053069419	12.419790954	18.603032375
S	15.813964093	14.887745277	11.585948557
S	16.384181619	9.774562168	12.432241475
S	12.490157120	17.112372341	18.361534012
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S	8.499666674	15.154794212	17.089952543
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S	7.817337024	11.556873048	16.589515631
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Br	3.115536459	7.182096593	17.442447449

Br	2.619545900	17.301312588	11.147000504
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Br	15.730743091	11.250928852	6.160162685
Br	4.462106048	19.400507118	14.333751105
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C	18.322971089	9.669693196	20.291843916
C	16.330148366	17.062776345	18.435071099
C	17.562082660	17.832251271	17.952732155
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C	11.446720293	20.976586201	17.629858808
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C	16.266389058	10.217131923	10.715383498
C	17.343616585	10.717590671	9.970237561
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H	18.728406068	15.100309779	11.120716206
H	17.863942723	16.448788310	10.381241867
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H	15.228732370	14.161618771	8.784559213
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H	17.967307412	18.601674699	11.705706054
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H	13.972339148	18.685774346	13.314322876
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H	8.911995094	21.064982850	13.556276437
H	8.183023889	19.455880321	13.466620169
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H	14.410621049	7.207528137	15.794074837
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H	17.483524104	8.515907835	17.606002285
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Au	18.379853578	15.776972536	15.723019269
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Au	12.886265373	14.424619564	16.039813671
Au	17.903709739	15.285237317	13.058448927
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Au	13.654209842	16.390844168	12.961033595
Au	15.390238798	15.862329330	17.813505541
Au	15.131600434	14.105454466	11.235408139
Au	20.068970879	14.172542343	17.700702237
Au	10.503059101	16.052411594	11.707874891
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Au	11.127450086	15.896817327	14.511149092

Au	19.705513541	11.782398453	12.639709037
Au	11.643296058	18.306748243	16.467613130
Au	19.430497807	18.071726514	13.187066427
Au	10.942019540	12.282715160	15.896391272
Au	21.640475536	16.302091284	15.402089905
Au	8.782209718	14.137530898	14.185691821
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S	12.378964462	17.316011215	11.171275519
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S	13.275905916	15.041328388	18.313006612
S	16.821287824	15.668169761	10.966629103
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S	13.669231330	19.418394224	16.532386199
S	21.669584021	12.781615020	13.387589205
S	9.859456401	16.875093366	16.323544763
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Br	13.158512508	9.840655235	22.140083848
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Br	4.703878329	8.901098456	18.003798932
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C	18.036212164	15.075788446	9.703362828
C	19.348896130	15.838721188	9.914144167
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Au	14.823132463	14.599976948	17.877285506
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Au	13.776343932	17.114169994	16.739634091
Au	12.627302361	16.053470250	13.545801509
Au	17.179910714	14.095524537	16.464766112
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Au	16.600583680	16.842433007	16.597827543
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Au	14.857788704	10.345448507	13.172595204
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H	11.587181322	23.891136895	13.244485825

IV. References

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