

SUPPORTING INFORMATION

for

Straightforward computational determination of energy-transfer
kinetics through the application of the Marcus theory

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1 Additional details about the application of the Marcus theory.

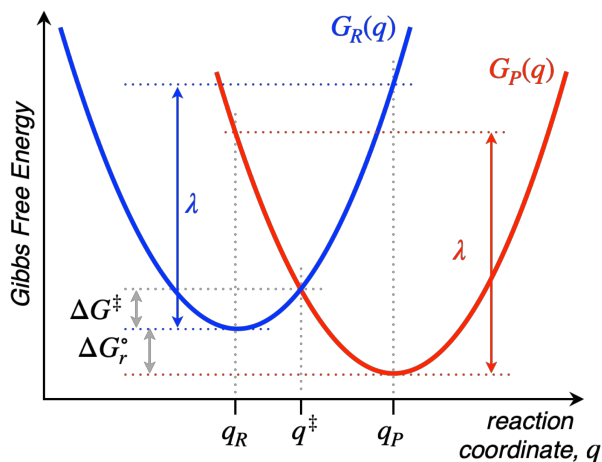
Classical vs. Semi-classical Marcus theory.

Free-energy basins of **reactants** and **products** states described as parabolic functions, where k_R and k_P determine their amplitudes:

$$G_R(q) = G_R(q_R) + \frac{1}{2} k_R (q - q_R)^2$$

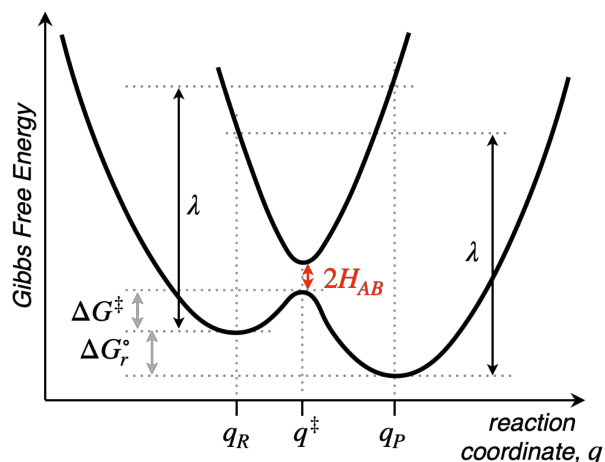
$$G_P(q) = G_P(q_P) + \frac{1}{2} k_P (q - q_P)^2$$

Classical Marcus Theory



$$k_{EnT} = \frac{k_B T}{h} \exp\left(-\frac{(\lambda + \Delta G_r^\circ)^2}{4\lambda k_B T}\right)$$

Semi-classical Marcus Theory

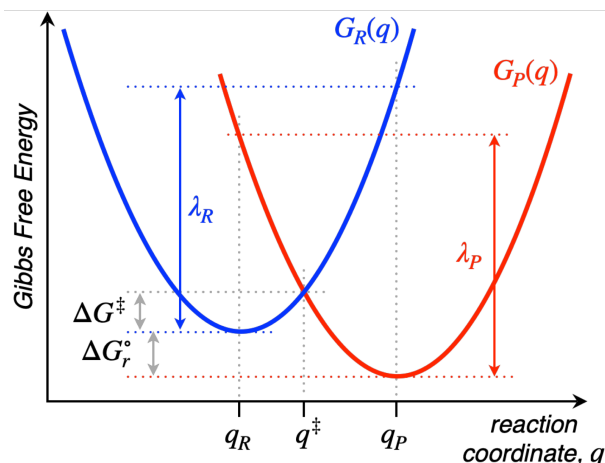


$$k_{EnT} = \frac{2\pi |H_{AB}|^2}{\hbar \sqrt{4\pi\lambda k_B T}} \exp\left(-\frac{(\lambda + \Delta G_r^\circ)^2}{4\lambda k_B T}\right)$$

Figure S 1: Comparison of the classical (left) and semi-classical (right) versions of the Marcus Theory and their impact on EnT rate constants. H_{AB} stands for the electronic coupling between initial and final states.

Symmetric vs. Asymmetric Marcus approach.

Symmetric approach:

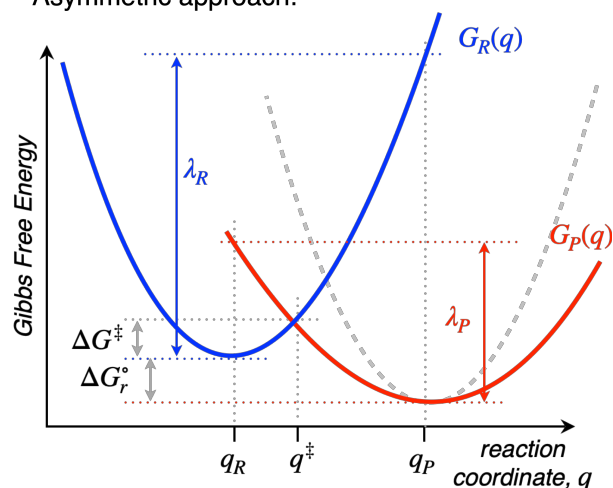


$$k_R = k_P$$

$$\lambda_R = \lambda_P = \lambda$$

$$\Delta G^\ddagger = \frac{(\lambda + \Delta G_r^\circ)^2}{4\lambda}$$

Asymmetric approach:



$$k_R \neq k_P$$

$$\lambda_R \neq \lambda_P$$

$$\Delta G^\ddagger = \lambda_R \left(\frac{-\lambda_P + \sqrt{\lambda_R \lambda_P + (\lambda_R - \lambda_P) \Delta G_r^\circ}}{\lambda_R - \lambda_P} \right)^2$$

Figure S 2: Graphical representation of the impact of the relative amplitude of parabolas describing reactants and products states on the height of the free-energy barrier. In the asymmetric scenario, the gray dashed parabola represents what would be assumed if the symmetric approach were applied to a system where the free-energy curves of reactants and products have distinct amplitudes.

2 Discussion about the existence of two minima on the triplet surface of **4**.

As mentioned in the main text and shown in Figure 5, we characterized two minima for substrate **4** on the triplet-state energy surface. The first one, labeled as ${}^3\mathbf{4}$, corresponds to the twisted configuration analogous to that obtained for **1-3** (see Figure S3 for a representation of the spin density on the optimized geometry of ${}^3\mathbf{4}$), which was obtained when relaxing the *Z* isomer on the triplet-state surface. The second (${}^3\mathbf{4}'$), obtained when steaming from the *E* conformer, is 6.8 kcal mol⁻¹ less stable and corresponds to a structure whereby the planarity of the system is preserved (Figure S3 and Figure 5, main text). The existence of this second minimum for **4** was ascribed to a higher delocalization of the spin density in the flat configuration compared to the other substrates, granted by the highly-conjugated naphthalene substituent, thus stabilizing a local minimum with $\Phi = 180^\circ$ by reducing the repulsion between electrons of the same spin. Note that a flat triplet minimum with a *Z*-like configuration could not be located, most likely due to steric repulsion between the oxygen atom of the ester group and the naphthalene substituent, which destabilizes the flat configuration, preventing an efficient conjugation of the π system. As a matter of fact, deviation from planarity of the naphthalene group with respect to the double bond was found to be larger in the *Z* isomer (54.1°) compared to the *E* one (20.6°).

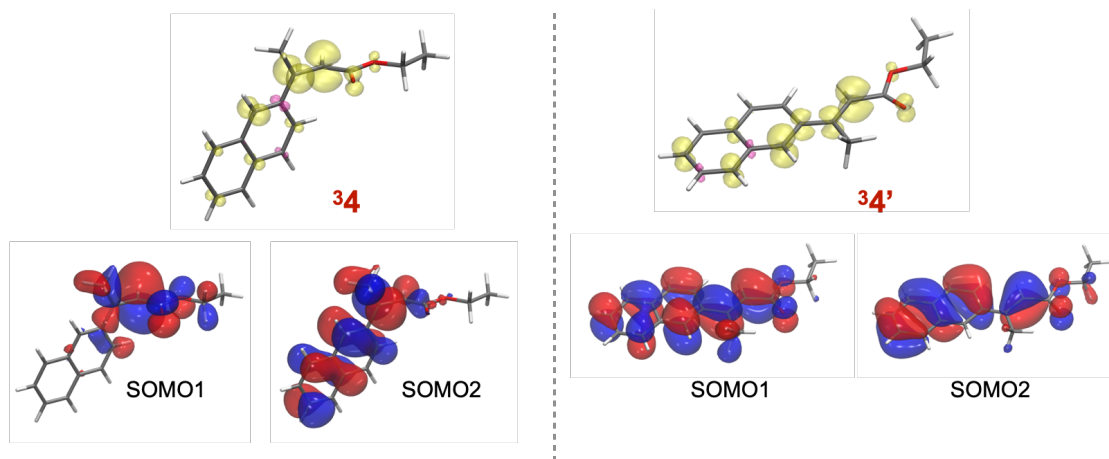


Figure S 3: Spin density distribution and singly occupied molecular orbitals (SOMOs) represented on the optimized structures of ${}^3\mathbf{4}$, $\Phi \approx 90^\circ$ (left) and ${}^3\mathbf{4}'$, $\Phi \approx 180^\circ$ (right)

3 Summary of the calculated parameters used in the Marcus theory.

Table S 1: Calculated reorganization energies (kcal mol⁻¹) for all the studied species on the reactants (λ_R) and products (λ_P) surfaces of the ³PC + substrate → PC + ³substrate reaction

pair	λ_R	λ_P
³ TX*/TX	4.8	4.1
1 _E / ³ 1	71.9	32.7
1 _Z / ³ 1	71.0	26.1
2 _E / ³ 2	61.3	21.7
2 _Z / ³ 2	59.8	33.0
3 _E / ³ 3	63.0	23.3
3 _Z / ³ 3	62.9	33.8
4 _E / ³ 4'	10.5	9.2
4 _Z / ³ 4	59.2	22.3
³ Ir-A/Ir-A	13.1	8.3
³ Ir-B/Ir-B	7.3	5.1
³ Ir-C/Ir-C	6.6	3.3
³ Ir-D/Ir-D	7.9	5.9
³ Ir-E/Ir-E	11.1	11.8
³ Ru/Ru	8.2	3.5

Table S 2: Kinetic and thermodynamic parameters (kcal mol⁻¹) for the sensitization of **3_E** and **3_Z** by Ir(III)- and Ru(II)-based photocatalysts (PCs).

PC	substrate	$\Delta G^{\ddagger}_{exp.}$	ΔG°_r	λ_R	λ_P	$\Delta G^{\ddagger}_{symm.}$	$\Delta\Delta G^{\ddagger}_{symm.}$	$\Delta G^{\ddagger}_{asymm.}$	$\Delta\Delta G^{\ddagger}_{asymm.}$
Ir-A	3_E	8.2	-3.1	76.1	31.6	12.0	-3.8	9.9	-1.7
	3_Z	9.7	-3.2	76.0	42.0	13.2	-3.5	12.0	-2.3
Ir-B	3_E	6.5	-8.2	70.3	28.4	8.6	-2.1	6.0	+0.5
	3_Z	8.1	-8.4	70.2	38.1	9.8	-1.7	8.3	-0.3
Ir-C	3_E	5.4	-15.4	69.6	26.6	5.6	-0.2	2.3	+3.0
	3_Z	6.7	-15.6	69.5	37.0	6.7	0.0	4.7	+2.0
Ir-D	3_E	5.1	-12.8	70.9	29.2	6.9	-1.8	4.1	+1.1
	3_Z	6.4	-13.0	70.9	39.6	8.1	-1.7	6.4	0.0
Ir-E	3_E	4.6	-15.3	74.1	35.0	7.1	-2.5	4.6	+0.1
	3_Z	5.0	-15.5	74.0	45.5	8.2	-3.2	6.7	-1.7
Ru	3_E	-	-0.6	71.2	26.8	11.9	-	9.9	-
	3_Z	-	-0.8	71.2	37.3	13.2	-	12.1	-
						MAE =	2.0	MAE =	1.3

4 Analysis of the theoretical rate constants derived from the application of the asymmetric Marcus approach.

Owing to the good agreement between calculated barriers and those derived from experimental rate constants (see main text), the theoretical rate constants are also in line with the experimental ones, as shown in Table S3. Note however, that rate constants are strongly sensitive to computational error due to their exponential dependence on free-energy barriers. In fact, a difference of only 1 kcal mol⁻¹ in the height of the barrier, which falls within the range of computational uncertainty, entails variations of one order of magnitude on the derived rate constant. On these grounds, we consider that the agreement between experimental and theoretical rate constants is more than acceptable. Additionally, the theoretical barriers can also reproduce the experimental selectivity trends.[1] Specifically, they lead to a k_Z/k_E ratio >1 for substrate **1**, indicating minor *E*-isomer sensitization and thus, major *E*-isomer formation; and k_Z/k_E ratios <1 for substrates **2** and **3**, for which a preference for the *Z*-isomer formation was also experimentally reported.[1] In fact, as shown in Figure S4, the construction of microkinetic models based on theoretical rate constants can correctly reproduce the experimental selectivity trend. Unfortunately, a quantitative theoretical determination of reaction outcomes in terms of amounts of *E* and *Z* isomer would require extremely accurate barriers with errors significantly smaller than 1 kcal mol⁻¹, which is out of reach for DFT, and most likely, even for more sophisticated and demanding computational methods. As a matter of fact, decreasing the $\Delta\Delta G^\ddagger$ between *E*- and *Z*-isomer sensitization by only 1.5 kcal mol⁻¹ for substrates **2** and **3** leads to *E*:*Z* product ratios that are in good quantitative agreement with experimental ones (see Figure S4)

Obviously, the above analysis cannot apply to substrate **4**, whose sensitization is diffusion controlled. In these cases, a *E*:*Z* ratio close to 50:50 (as experimentally found by Kerzig, Gilmour and co-workers)[1] may be always expected, as diffusion barriers cannot discern between *E* and *Z* isomers of the same molecule and hence, the sensitization process cannot be selective at all.

Table S 3: Comparison of experimental and theoretical rate constants and constants and barrier ratios for the sensitization of *Z* and *E* isomers of substrates **1-3**

Substrate	k_{EnT} exp. / $10^9 \text{ M}^{-1} \text{ s}^{-1}$	k_{EnT} calc. $10^9 \text{ M}^{-1} \text{ s}^{-1}$	k_Z/k_E exp.	k_Z/k_E calc.	$\Delta G_Z^\ddagger/\Delta G_E^\ddagger$ exp.	$\Delta G_Z^\ddagger/\Delta G_E^\ddagger$ calc.
1_Z	0.583	0.142	30.7	39.3	0.73	0.74
1_E	0.019	0.004				
2_Z	2.60	0.54	0.38	0.03	1.14	1.62
2_E	6.76	18.9				
3_Z	3.25	0.14	0.46	0.02	1.05	1.51
3_E	7.02	8.38				

Equations used in microkinetic models to produce the results shown in Figure S4:



k_1 was fitted to 10^{-1} s^{-1} , so that the reaction is completed within 1h, as in the experimental reference.[1] k_{-1} was estimated from the experimental lifetime of TX^* (77 μs), assuming first-order kinetics. k_2 and k_3 (EnT events) were derived from theoretical free-energy barriers. Finally, k_4 and k_5 were both set to 10^{11} s^{-1} for the sake of being faster than EnT processes and equal for subs^* to yield either *E* or *Z* products. Initial concentrations:[1]

$$[\text{subs}_Z] = 6 \times 10^{-2} \text{ M}; [\text{TX}] = 6 \times 10^{-4} \text{ M}; [\text{subs}_E] = [\text{subs}^*] = [\text{TX}^*] = 0.$$

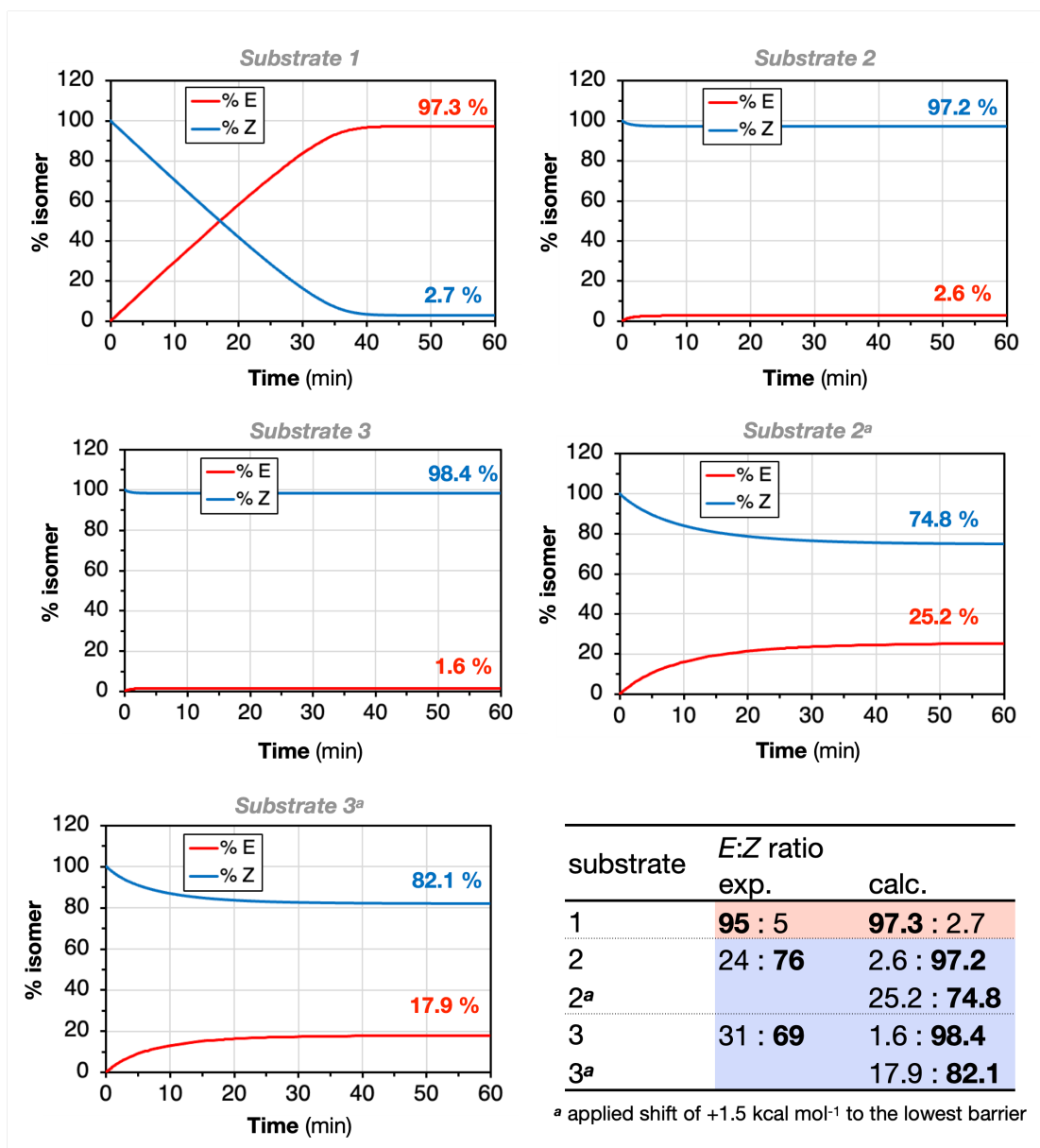


Figure S 4: Evolution of *E*:*Z* product ratios as function of reaction time as derived from the microkinetic model presented above. Asterisk marks denote that in those models, the free-energy barrier for the fastest EnT event has been shifted by +1.5 kcal mol⁻¹ and the table compares *E*:*Z* ratios at the end of the simulation with experimental ones.[1] Within the table, red and blue coloring represents experimental selectivity towards *E* or *Z* isomers, respectively, which is qualitatively, if not quantitatively reproduced by our microkinetic models in all cases.

5 Analysis of the factors governing the EnT free-energy barriers from excited photocatalysts to $\mathbf{3}_E$ and $\mathbf{3}_Z$.

As shown in Figure S5, neither the computationally-determined barriers nor the experimental ones are solely determined by the triplet energy of the analyzed PCs, which are implicit in the reaction free energies (Figure S5a,d), although they roughly correlate. Conversely, our results also evidence that their intrinsic reorganization energies have an important weight on the height of the barriers (Figure S5b,c,e,f), correlating even better with the barriers than the reaction energies themselves.

Consequently, this causes that PCs that entail distinct reaction energies display similar barriers, due to the compensating effect of the reorganization energies. For instance, even though the EnT from **Ir-A** to $\mathbf{3}_E$ is thermodynamically less favorable than that of **Ir-B** to $\mathbf{3}_Z$ (ΔG_r° of -3.1 vs -8.4 kcal mol⁻¹, respectively, entries 1 and 4 in Table S2), they both exhibit very similar barriers, which is explained by the fact that the λ_P of the latter is significantly higher.

Similarly, although **Ir-A** and **Ru** exhibit rather different triplet energies (see Table S5) they were predicted to show similar barriers, owing to the much smaller reorganization energies determined for **Ru** compared to **Ir-A** (Table S1).

It is also interesting to note that Gilmour and Kerzig observed that thioxanthone-based photocatalysts sensitize substrates faster than Ir-based photocatalysts with similar triplet energies.[1] This can be easily rationalized using the Marcus theory and the above-reported parameters as follows:

According to the Marcus theory, barriers are a function of reaction free energies (ΔG_r°) and reorganization energies (λ parameters). The more exergonic the reaction, the lower the barrier; and the smaller the reorganization energy, the lower the barrier too (see eq. 3 and 4 in the main text). When comparing the **Ir-D** and **TX** catalysts, although they exhibit similar triplet energies, which translate into almost equal reaction free energies (see Table 1 in the main text and Table S2), they sensitize alkene substrates at a different rate. Hence, the discrepancies in the rate constants (and thus the barriers) arise from their distinct reorganization energies. As shown in Table S1, the reorganization energies calculated for **TX** (4.1-4.8 kcal mol⁻¹) are smaller than those measured for **Ir-D** (5.9-7.9 kcal mol⁻¹). Consequently, the EnT barriers are lower for **TX**, explaining the experimental outcomes by

Gilmour and Kerzig and co-workers.[1]

To further support these conclusions, we have carried out additional calculations on the sensitization of 3_E and 3_Z by 2,7-Dimethoxy-9*H*-thioxanthen-9-one (**MeOTX**), which was also found to sensitize substrates faster than Ir-based PCs with similar triplet energies. Specifically, the reorganization energy of **MeOTX** (4.2 kcal mol⁻¹, both on reactants and products surfaces) was computed to be very similar to that of **TX** (4.4 kcal mol⁻¹, on average) and hence, also smaller than that of Ir-based PCs, which are typically larger than 6 kcal mol⁻¹ when averaged over reactants and products surfaces. Also in qualitative agreement with experimental findings,[1] free-energy barriers obtained for **MeOTX** were found to be *ca.* 5 kcal mol⁻¹ higher than those involving **TX**. This can be explained by the lower triplet energy of **MeOTX**,[1] which leads to less favorable EnT thermodynamics. Specifically, the sensitization free energies by **MeOTX** (-2.5 and -2.7 kcal mol⁻¹ for 3_E and 3_Z , respectively) are significantly less exergonic than those involving **TX** (-11.9 and -12.1 kcal mol⁻¹). Thus, although both PCs display similar reorganization energies, their distinct reaction kinetics are due to different thermodynamic driving forces.

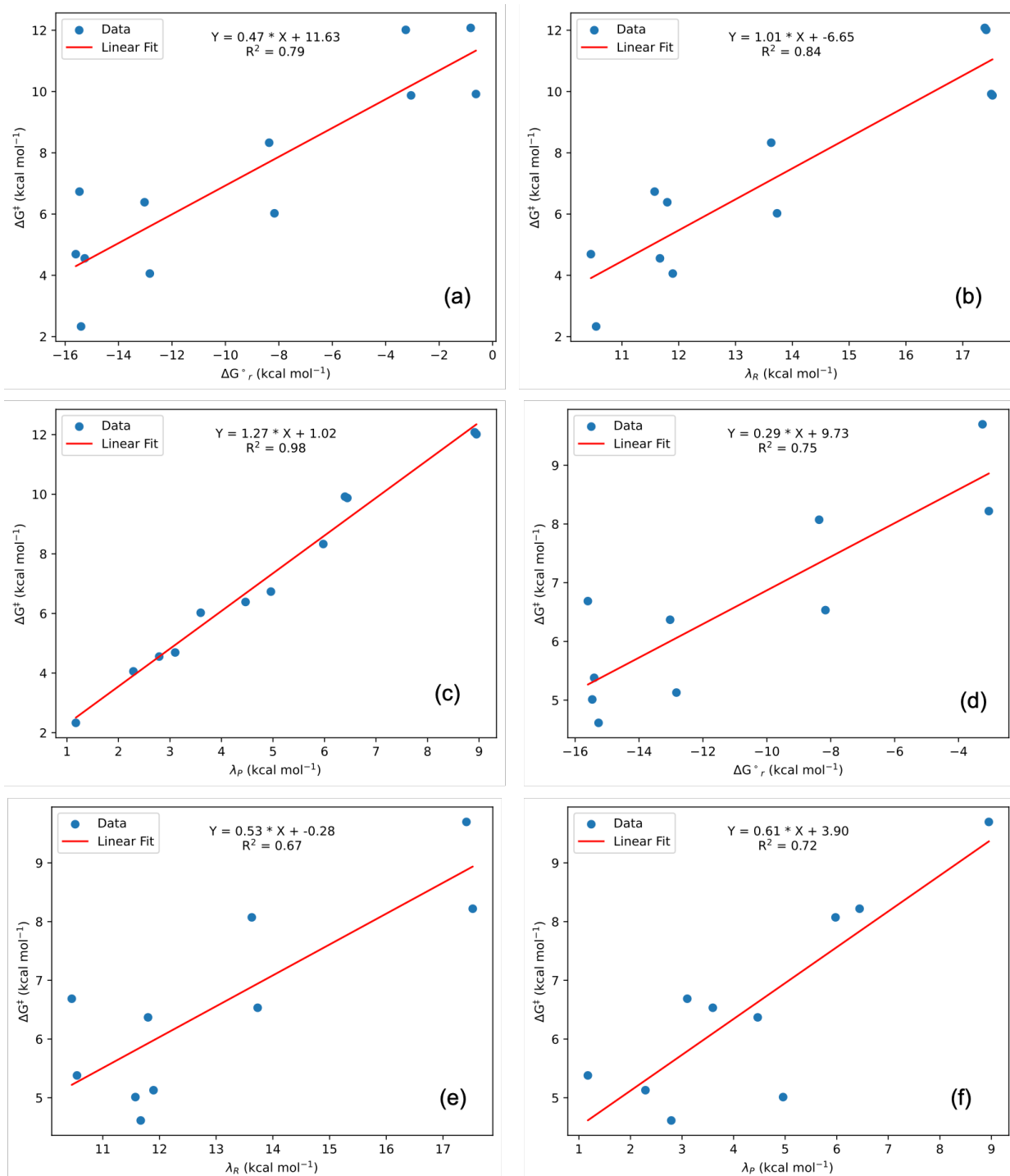


Figure S 5: Linear fitting of the ΔG_r° for EnT processes from Ir- and Ru-based photocatalysts to $\mathbf{3}_E$ and $\mathbf{3}_Z$ and contributions to λ_R and λ_P of the PCs with the free-energy barriers obtained either computationally through the asymmetric Marcus equation (a-c) or from experimental rate constants[1] (d-f).

6 Methodology benchmark and calculated triplet energies for the studied photocatalysts.

Table S 4: Comparison of the performance of several density functionals in predicting the triplet energy of **TX**. The experimental value of 63.4 kcal mol⁻¹ was determined from the phosphorescence emission wavelength in 2-Methyltetrahydrofuran ($\epsilon = 6.97$).[1]

E[ρ]	$\Delta E_{triplet-singlet}$	
	IEF-PCM (MeCN, $\epsilon = 35.69$)	IEF-PCM (THF, $\epsilon = 7.43$)
B3LYP-D3BJ	60.3	61.0
ω B97X-D	67.0	67.9
PBE0-D3BJ	61.2	61.9
TPSSH-D3BJ	58.6	59.2
PBE-D3BJ	55.5	56.1

Table S 5: Comparison of the DFT (B3LYP-D3BJ)-derived triplet energies (kcal mol⁻¹) for Ir- and Ru-based photocatalysts with the experimental ones.[1]

PC	$\Delta E_{triplet-singlet}$	
	exp.	calc.
Ir-A	55.6	51.3
Ir-B	59.3	57.0
Ir-C	62.5	61.9
Ir-D	63.4	62.6
Ir-E	67.1	64.2
Ru	50.5	49.5

7 Cartesian coordinates (Å) of the optimized structures at the B3LYP-D3BJ/cc-pVDZ,LANL2DZ(f) level

1_E

Atom X Y Z

1 C -1.0181 0.3787 0.0131
2 C -0.2812 1.5059 0.0290
3 H 0.8102 1.5017 -0.0371
4 C -0.3870 -0.9805 -0.1032
5 H 0.7103 -0.9328 -0.1658
6 H -0.6680 -1.6079 0.7607
7 H -0.7725 -1.5039 -0.9956
8 C -0.9509 2.8134 0.1356
9 O -2.1638 2.9683 0.2101
10 O -0.0769 3.8339 0.1409
11 C -0.6436 5.1643 0.2419
12 H -1.3281 5.3196 -0.6065
13 H -1.2370 5.2239 1.1674
14 B -2.5889 0.4205 0.1171
15 O -3.4001 0.3219 -0.9841
16 O -3.2569 0.3195 1.3103
17 C -4.7825 0.4248 -0.5085
18 C -4.6527 0.0013 1.0039
19 C -5.2065 1.8818 -0.7006
20 C -5.6494 -0.4968 -1.3557
21 C -4.8097 -1.5031 1.2307
22 C -5.5439 0.7804 1.9616
23 H -4.5656 2.5604 -0.1233
24 H -5.1036 2.1410 -1.7652
25 H -6.2565 2.0331 -0.4102
26 H -5.2668 -1.5255 -1.3612
27 H -6.6832 -0.5078 -0.9772

28 H -5.6711 -0.1324 -2.3936
29 H -4.5107 -1.7400 2.2625
30 H -5.8514 -1.8260 1.0892
31 H -4.1678 -2.0792 0.5481
32 H -5.3800 0.4270 2.9907
33 H -5.3278 1.8558 1.9284
34 H -6.6052 0.6261 1.7128
35 C 0.5069 6.1467 0.2356
36 H 1.0894 6.0634 -0.6950
37 H 0.1179 7.1737 0.3105
38 H 1.1808 5.9679 1.0878

1_z

Atom X Y Z

1 C -0.9859 0.2637 0.1096
2 C -0.3940 1.4766 0.0128
3 H 0.6803 1.5389 -0.1725
4 C -2.4587 0.0573 0.3605
5 H -3.0602 0.4821 -0.4591
6 H -2.6933 -1.0112 0.4546
7 H -2.7858 0.5818 1.2712
8 C -1.0901 2.7723 0.1403
9 O -2.2840 2.9424 0.3297
10 O -0.2122 3.7937 0.0167
11 C -0.7660 5.1260 0.1218
12 H -1.5382 5.2504 -0.6536
13 H -1.2587 5.2281 1.1017
14 B -0.0649 -0.9918 -0.0502
15 O 1.2911 -0.9213 -0.2467
16 O -0.5460 -2.2739 -0.0040
17 C 1.7506 -2.2720 -0.5787
18 C 0.6122 -3.1719 0.0297
19 C 1.8277 -2.3428 -2.1045
20 C 3.1280 -2.4796 0.0345

21 C 0.8409 -3.5272 1.4994
22 C 0.2812 -4.4197 -0.7763
23 H 0.8378 -2.2016 -2.5628
24 H 2.4904 -1.5427 -2.4652
25 H 2.2348 -3.3077 -2.4398
26 H 3.1281 -2.2727 1.1123
27 H 3.4677 -3.5143 -0.1257
28 H 3.8525 -1.8052 -0.4458
29 H -0.0807 -3.9677 1.9072
30 H 1.6539 -4.2591 1.6118
31 H 1.0875 -2.6351 2.0936
32 H -0.5263 -4.9781 -0.2797
33 H -0.0479 -4.1709 -1.7932
34 H 1.1606 -5.0785 -0.8422
35 C 0.3752 6.1055 -0.0474
36 H 0.8555 5.9828 -1.0306
37 H -0.0049 7.1360 0.0267
38 H 1.1369 5.9598 0.7342

³1

Atom X Y Z

1 C -1.2831 0.5983 0.6766
2 C -0.6177 1.6122 -0.1333
3 H -0.1142 1.3585 -1.0727
4 C -0.4892 -0.0917 1.7477
5 H -0.1349 0.6453 2.4940
6 H -1.0877 -0.8544 2.2647
7 H 0.4188 -0.5664 1.3325
8 C -0.6305 3.0063 0.2748
9 O -1.1824 3.4315 1.2859
10 O 0.0387 3.7984 -0.5976
11 C 0.0688 5.2073 -0.2762
12 H -0.9664 5.5768 -0.2015
13 H 0.5398 5.3367 0.7114

14 B -2.7899 0.3510 0.4980
15 O -3.5552 0.9927 -0.4507
16 O -3.5097 -0.5231 1.2817
17 C -4.9537 0.6994 -0.1394
18 C -4.8415 -0.6351 0.6856
19 C -5.4712 1.8781 0.6878
20 C -5.7289 0.5830 -1.4441
21 C -4.8231 -1.8873 -0.1929
22 C -5.8620 -0.7836 1.8053
23 H -4.9228 1.9714 1.6367
24 H -5.3241 2.8057 0.1154
25 H -6.5432 1.7734 0.9099
26 H -5.2722 -0.1482 -2.1234
27 H -6.7690 0.2801 -1.2482
28 H -5.7484 1.5589 -1.9518
29 H -4.5314 -2.7505 0.4232
30 H -5.8140 -2.0890 -0.6252
31 H -4.0964 -1.7892 -1.0128
32 H -5.7045 -1.7399 2.3262
33 H -5.7766 0.0263 2.5411
34 H -6.8843 -0.7823 1.3970
35 C 0.8447 5.9051 -1.3724
36 H 0.3593 5.7601 -2.3501
37 H 0.8921 6.9854 -1.1655
38 H 1.8738 5.5180 -1.4324

2_E

Atom X Y Z

1 C -0.9746 0.2685 -0.0741
2 C -0.4207 1.5030 0.0271
3 H 0.6621 1.6008 0.1047
4 C -2.4594 0.0243 -0.1241
5 H -3.0089 0.9545 -0.2889
6 H -2.7018 -0.7076 -0.9099

7 H -2.8035 -0.4092 0.8306
8 C -1.1302 2.7940 0.0585
9 O -2.3335 3.0040 0.0179
10 O -0.2255 3.7996 0.1446
11 C -0.7641 5.1406 0.1852
12 H -1.3605 5.3115 -0.7249
13 H -1.4432 5.2247 1.0484
14 C 0.4065 6.0948 0.2862
15 H 1.0743 5.9907 -0.5831
16 H 0.0393 7.1322 0.3190
17 H 0.9905 5.9047 1.2001
18 C -0.0933 -0.9280 -0.1196
19 C -0.5577 -2.1754 0.3411
20 C 1.2187 -0.8587 -0.6303
21 C 0.2672 -3.3018 0.3211
22 H -1.5684 -2.2675 0.7385
23 C 2.0390 -1.9858 -0.6594
24 H 1.5923 0.0821 -1.0361
25 C 1.5694 -3.2131 -0.1784
26 H -0.1122 -4.2542 0.6966
27 H 3.0481 -1.9080 -1.0692
28 H 2.2118 -4.0955 -0.2017

2_Z

Atom X Y Z

1 C -1.1610 0.4306 -0.0602
2 C -0.5047 1.6113 -0.0649
3 H 0.5625 1.6204 -0.2948
4 C -0.4323 -0.8409 -0.4052
5 H 0.6194 -0.6553 -0.6630
6 H -0.4790 -1.5468 0.4411
7 H -0.9213 -1.3453 -1.2553
8 C -1.0969 2.9451 0.1587
9 O -2.2707 3.2258 0.3256

10 O -0.1176 3.8824 0.1301
11 C -0.5430 5.2513 0.3170
12 H -1.2649 5.5100 -0.4737
13 H -1.0656 5.3321 1.2833
14 C 0.6947 6.1210 0.2644
15 H 1.2032 6.0221 -0.7073
16 H 0.4147 7.1766 0.4032
17 H 1.4040 5.8430 1.0595
18 C -2.5951 0.2606 0.3032
19 C -3.0583 0.6153 1.5805
20 C -3.4912 -0.3291 -0.6026
21 C -4.3872 0.3912 1.9410
22 H -2.3659 1.0653 2.2931
23 C -4.8277 -0.5347 -0.2487
24 H -3.1471 -0.6116 -1.5996
25 C -5.2788 -0.1796 1.0255
26 H -4.7303 0.6663 2.9406
27 H -5.5161 -0.9791 -0.9706
28 H -6.3203 -0.3494 1.3063

³2

Atom X Y Z

1 C -1.4305 0.4930 -0.0738
2 C -0.6984 1.6479 0.4576
3 H -0.0136 1.5509 1.3075
4 C -0.7818 -0.2707 -1.1972
5 H 0.2664 0.0311 -1.3303
6 H -0.8050 -1.3591 -1.0206
7 H -1.3049 -0.0841 -2.1537
8 C -0.8043 2.9526 -0.1787
9 O -1.4943 3.1960 -1.1622
10 O -0.0419 3.8801 0.4482
11 C -0.0828 5.2130 -0.1106
12 H 0.2488 5.1675 -1.1603

13 H -1.1259 5.5674 -0.1066
14 C 0.8192 6.0864 0.7343
15 H 1.8546 5.7122 0.7192
16 H 0.8162 7.1146 0.3409
17 H 0.4725 6.1137 1.7790
18 C -2.7241 0.1574 0.4262
19 C -3.3226 0.9110 1.4790
20 C -3.4746 -0.9318 -0.1055
21 C -4.5826 0.5906 1.9677
22 H -2.7740 1.7528 1.9068
23 C -4.7356 -1.2421 0.3912
24 H -3.0588 -1.5296 -0.9168
25 C -5.3020 -0.4883 1.4301
26 H -5.0149 1.1849 2.7757
27 H -5.2893 -2.0814 -0.0355
28 H -6.2923 -0.7376 1.8155

3_E

Atom X Y Z

1 C -1.1078 0.7335 -0.0523
2 C -0.3849 1.8793 -0.1441
3 B -2.6371 0.5617 0.0860
4 O -3.1922 -0.6945 0.2110
5 O -3.5889 1.5605 0.1001
6 C -4.6412 -0.5548 0.0971
7 C -4.8638 0.9535 0.4791
8 C -4.9995 -0.8683 -1.3572
9 C -5.3015 -1.5579 1.0335
10 C -5.0260 1.1791 1.9835
11 C -5.9788 1.6528 -0.2861
12 H -4.5363 -0.1478 -2.0472
13 H -4.6255 -1.8720 -1.6077
14 H -6.0879 -0.8550 -1.5147
15 H -4.9328 -1.4561 2.0622

16 H -6.3940 -1.4226 1.0360
17 H -5.0862 -2.5812 0.6911
18 H -4.9644 2.2575 2.1911
19 H -5.9993 0.8126 2.3414
20 H -4.2308 0.6743 2.5516
21 H -6.0534 2.7003 0.0425
22 H -5.7950 1.6450 -1.3681
23 H -6.9459 1.1651 -0.0896
24 H -0.5462 -0.2074 -0.0634
25 C -1.0219 3.2456 -0.0736
26 H -0.7373 3.7581 0.8613
27 H -0.6705 3.8847 -0.8999
28 H -2.1138 3.1792 -0.1019
29 C 1.0959 1.8447 -0.3028
30 C 1.8950 2.9243 0.1218
31 C 1.7455 0.7421 -0.8946
32 C 3.2854 2.8899 -0.0091
33 H 1.4309 3.7989 0.5777
34 C 3.1326 0.7103 -1.0342
35 H 1.1529 -0.0914 -1.2740
36 C 3.9124 1.7829 -0.5868
37 H 3.8807 3.7359 0.3410
38 H 3.6078 -0.1533 -1.5044
39 H 4.9984 1.7587 -0.6967

3_Z

Atom X Y Z

1 C -1.1235 0.6643 0.7557
2 C -0.3910 1.7004 0.2843
3 B -2.6593 0.4809 0.7046
4 O -3.5131 1.0907 -0.1857
5 O -3.3107 -0.3869 1.5556
6 C -4.8765 0.8081 0.2512
7 C -4.6920 -0.5068 1.0919

8 C -5.3257 2.0057 1.0917
9 C -5.7607 0.6650 -0.9799
10 C -4.7613 -1.7789 0.2450
11 C -5.6041 -0.6259 2.3051
12 H -4.7055 2.1174 1.9931
13 H -5.2199 2.9207 0.4911
14 H -6.3771 1.9087 1.3995
15 H -5.3626 -0.0801 -1.6806
16 H -6.7802 0.3671 -0.6904
17 H -5.8234 1.6304 -1.5041
18 H -4.4179 -2.6304 0.8510
19 H -5.7891 -1.9837 -0.0888
20 H -4.1132 -1.7041 -0.6405
21 H -5.4010 -1.5706 2.8316
22 H -5.4496 0.2001 3.0111
23 H -6.6601 -0.6298 1.9938
24 H -0.5555 -0.1620 1.2022
25 C 1.1169 1.6884 0.3132
26 H 1.5307 1.7937 -0.7041
27 H 1.5017 2.5447 0.8930
28 H 1.5051 0.7571 0.7489
29 C -1.0128 2.9417 -0.2571
30 C -0.6332 3.4562 -1.5090
31 C -1.9718 3.6451 0.4912
32 C -1.2192 4.6210 -2.0111
33 H 0.1112 2.9285 -2.1085
34 C -2.5487 4.8161 -0.0024
35 H -2.2575 3.2697 1.4747
36 C -2.1788 5.3063 -1.2593
37 H -0.9234 4.9951 -2.9936
38 H -3.2900 5.3484 0.5975
39 H -2.6315 6.2207 -1.6483

Atom X Y Z

1 C -0.9863 0.3824 1.4503
2 C -0.1331 1.5713 1.3265
3 B -2.4102 0.3215 0.8796
4 O -3.2382 -0.7741 0.9955
5 O -2.9803 1.3620 0.1789
6 C -4.3803 -0.5426 0.1107
7 C -4.3873 1.0241 -0.0271
8 C -4.0661 -1.2613 -1.2030
9 C -5.6244 -1.1351 0.7578
10 C -5.1726 1.7245 1.0836
11 C -4.8159 1.5453 -1.3919
12 H -3.1732 -0.8375 -1.6860
13 H -3.8706 -2.3227 -0.9909
14 H -4.9089 -1.1978 -1.9068
15 H -5.7770 -0.7477 1.7733
16 H -6.5174 -0.9065 0.1559
17 H -5.5255 -2.2293 0.8181
18 H -4.9566 2.8023 1.0465
19 H -6.2563 1.5835 0.9595
20 H -4.8816 1.3502 2.0761
21 H -4.7851 2.6451 -1.3947
22 H -4.1557 1.1819 -2.1898
23 H -5.8470 1.2327 -1.6177
24 H -0.5856 -0.4858 1.9947
25 C -0.1966 2.5877 2.4374
26 H 0.8011 2.7832 2.8698
27 H -0.5827 3.5579 2.0740
28 H -0.8554 2.2484 3.2487
29 C 0.6875 1.7771 0.1768
30 C 1.5051 2.9369 0.0310
31 C 0.7141 0.8259 -0.8870
32 C 2.2907 3.1237 -1.1012
33 H 1.5160 3.6926 0.8169

34 C 1.5040 1.0230 -2.0123
35 H 0.0982 -0.0723 -0.8061
36 C 2.3005 2.1730 -2.1332
37 H 2.9062 4.0225 -1.1851
38 H 1.5020 0.2760 -2.8097
39 H 2.9191 2.3262 -3.0195

4_E

Atom X Y Z

1 C -0.9448 0.2529 0.2369
2 C -0.4205 1.4948 0.0741
3 H 0.6265 1.6075 -0.2047
4 C -2.3937 0.0096 0.5674
5 H -2.9117 -0.4326 -0.3004
6 H -2.4839 -0.7129 1.3930
7 H -2.9009 0.9416 0.8274
8 C -1.1208 2.7836 0.2159
9 O -2.2894 2.9919 0.5071
10 O -0.2500 3.7925 -0.0322
11 C -0.7818 5.1331 0.0654
12 H -1.6204 5.2350 -0.6416
13 H -1.1827 5.2831 1.0803
14 C 0.3472 6.0911 -0.2484
15 H 0.7354 5.9212 -1.2647
16 H -0.0162 7.1283 -0.1850
17 H 1.1756 5.9696 0.4668
18 C -0.0792 -0.9420 0.0654
19 C 1.3462 -0.8539 0.1627
20 C -0.6387 -2.1871 -0.1924
21 C 2.1448 -1.9585 -0.0098
22 H 1.8116 0.1008 0.4067
23 C 0.1593 -3.3453 -0.3866
24 H -1.7198 -2.3018 -0.2711
25 C 1.5847 -3.2356 -0.2960

26 C -0.4171 -4.6160 -0.6669
27 C 2.3811 -4.3965 -0.4850
28 C 0.3815 -5.7261 -0.8477
29 H -1.5043 -4.6969 -0.7361
30 C 1.7930 -5.6156 -0.7559
31 H -0.0705 -6.6964 -1.0624
32 H 3.2296 -1.8681 0.0814
33 H 3.4675 -4.3077 -0.4132
34 H 2.4135 -6.5022 -0.9012

4z

Atom X Y Z

1 C -1.2388 0.5652 -0.1798
2 C -0.4930 1.6906 -0.1156
3 H 0.5704 1.6328 -0.3549
4 C -0.6096 -0.7316 -0.6150
5 H 0.4531 -0.6087 -0.8650
6 H -0.7101 -1.4894 0.1799
7 H -1.1348 -1.1362 -1.4963
8 C -0.9833 3.0476 0.1923
9 O -2.1339 3.3997 0.3867
10 O 0.0598 3.9128 0.2103
11 C -0.2660 5.2953 0.4802
12 H -0.9783 5.6478 -0.2825
13 H -0.7689 5.3571 1.4583
14 C 1.0290 6.0784 0.4552
15 H 1.5164 5.9999 -0.5291
16 H 0.8257 7.1413 0.6575
17 H 1.7274 5.7071 1.2213
18 C -2.6786 0.4829 0.1900
19 C -3.1081 0.8519 1.4988
20 C -3.6068 -0.0222 -0.7037
21 C -4.4235 0.7153 1.8747
22 H -2.3725 1.2362 2.2056

23 C -4.9789 -0.1502 -0.3553
24 H -3.2997 -0.3144 -1.7103
25 C -5.3995 0.2207 0.9638
26 C -5.9470 -0.6444 -1.2734
27 C -6.7697 0.0803 1.3153
28 C -7.2704 -0.7672 -0.9039
29 H -5.6230 -0.9258 -2.2782
30 C -7.6857 -0.4018 0.4027
31 H -8.0047 -1.1472 -1.6169
32 H -4.7378 0.9916 2.8839
33 H -7.0857 0.3628 2.3224
34 H -8.7361 -0.5038 0.6830

³4'

Atom X Y Z

1 C -0.9519 0.2117 0.0096
2 C -0.3982 1.5464 0.0028
3 H 0.6741 1.6713 -0.1277
4 C -2.4395 0.0029 0.1686
5 H -2.8596 -0.5354 -0.6979
6 H -2.6623 -0.6088 1.0592
7 H -2.9580 0.9596 0.2677
8 C -1.0990 2.8030 0.1537
9 O -2.3037 3.0003 0.3211
10 O -0.2077 3.8379 0.0862
11 C -0.7641 5.1597 0.2220
12 H -1.5172 5.3166 -0.5675
13 H -1.2842 5.2341 1.1911
14 C 0.3798 6.1462 0.1169
15 H 0.8867 6.0569 -0.8567
16 H -0.0023 7.1740 0.2159
17 H 1.1212 5.9736 0.9127
18 C -0.1086 -0.8974 -0.1332
19 C 1.3325 -0.8055 -0.2910

20 C -0.6569 -2.2568 -0.1301
21 C 2.1155 -1.9086 -0.4275
22 H 1.8118 0.1713 -0.3016
23 C 0.1457 -3.3953 -0.2704
24 H -1.7302 -2.3972 -0.0150
25 C 1.5704 -3.2505 -0.4248
26 C -0.4045 -4.7141 -0.2656
27 C 2.3684 -4.3853 -0.5645
28 C 0.4132 -5.8273 -0.4064
29 H -1.4840 -4.8318 -0.1487
30 C 1.8019 -5.6722 -0.5567
31 H -0.0243 -6.8274 -0.4005
32 H 3.1959 -1.7953 -0.5441
33 H 3.4481 -4.2667 -0.6812
34 H 2.4424 -6.5487 -0.6672

³4

Atom X Y Z

1 C -1.5684 0.8157 -0.3571
2 C -0.7032 1.7853 0.3250
3 H -0.0545 1.4879 1.1565
4 C -0.9943 0.1284 -1.5672
5 H 0.0946 0.2658 -1.6237
6 H -1.2040 -0.9537 -1.5626
7 H -1.4289 0.5434 -2.4958
8 C -0.6187 3.1621 -0.1394
9 O -1.2476 3.6188 -1.0872
10 O 0.2461 3.8906 0.6059
11 C 0.3975 5.2767 0.2228
12 H 0.7473 5.3195 -0.8211
13 H -0.5889 5.7657 0.2635
14 C 1.3856 5.9061 1.1807
15 H 2.3617 5.3993 1.1271
16 H 1.5306 6.9663 0.9221

17 H 1.0184 5.8489 2.2172
18 C -2.9020 0.5921 0.0805
19 C -3.4205 1.2870 1.2361
20 C -3.7710 -0.2966 -0.5870
21 C -4.7008 1.0905 1.6750
22 H -2.7658 1.9812 1.7669
23 C -5.0991 -0.5130 -0.1493
24 H -3.4279 -0.8373 -1.4698
25 C -5.5869 0.1884 1.0052
26 C -5.9771 -1.4111 -0.8227
27 C -6.9161 -0.0323 1.4386
28 C -7.2690 -1.6056 -0.3764
29 H -5.6082 -1.9454 -1.7015
30 C -7.7453 -0.9119 0.7636
31 H -7.9301 -2.2973 -0.9026
32 H -5.0680 1.6276 2.5529
33 H -7.2792 0.5051 2.3180
34 H -8.7687 -1.0735 1.1077

TX

Atom X Y Z

1 C -9.1456 -0.3853 0.0081
2 C -7.7590 -0.4653 0.0198
3 C -6.9823 0.7096 0.0147
4 C -7.6078 1.9733 -0.0023
5 C -9.0183 2.0233 -0.0139
6 C -9.7835 0.8671 -0.0089
7 C -6.8786 3.2651 -0.0089
8 C -5.3952 3.2740 0.0032
9 C -4.5954 2.1128 0.0206
10 C -3.1911 2.2199 0.0308
11 H -2.5813 1.3143 0.0441
12 C -2.5864 3.4703 0.0239
13 C -3.3703 4.6369 0.0067

14 C -4.7527 4.5307 -0.0034
15 H -9.7364 -1.3034 0.0123
16 H -7.2646 -1.4387 0.0330
17 H -9.4816 3.0098 -0.0271
18 H -10.8730 0.9285 -0.0180
19 H -1.4969 3.5397 0.0319
20 H -2.8940 5.6187 0.0014
21 H -5.3890 5.4156 -0.0168
22 S -5.2359 0.4706 0.0310
23 O -7.5043 4.3295 -0.0246

³TX

Atom X Y Z

1 C -9.1421 -0.4126 0.0085
2 C -7.7501 -0.4704 0.0200
3 C -6.9987 0.7167 0.0145
4 C -7.6118 2.0132 -0.0029
5 C -9.0257 2.0266 -0.0142
6 C -9.7680 0.8485 -0.0086
7 C -6.8795 3.2666 -0.0092
8 C -5.4282 3.2970 0.0026
9 C -4.5936 2.1306 0.0203
10 C -3.1909 2.2096 0.0307
11 H -2.6015 1.2906 0.0440
12 C -2.5643 3.4539 0.0242
13 C -3.3616 4.6142 0.0071
14 C -4.7520 4.5388 -0.0034
15 H -9.7313 -1.3294 0.0127
16 H -7.2340 -1.4324 0.0334
17 H -9.5138 2.9999 -0.0274
18 H -10.8587 0.9068 -0.0177
19 H -1.4766 3.5227 0.0322
20 H -2.8821 5.5957 0.0019
21 H -5.3645 5.4390 -0.0165

22 S -5.2609 0.5132 0.0304
23 O -7.5333 4.3783 -0.0251

Ir-A

Atom X Y Z

1 N -1.8879 -1.0683 0.2227
2 H 0.7437 -3.7588 3.6816
3 H -0.6975 3.7488 3.5577
4 H 4.0151 -1.0898 2.8483
5 H -4.0120 1.0518 3.0473
6 H 3.0908 -3.0194 4.0970
7 H -3.0145 3.0339 4.1487
8 H 0.6115 2.4978 1.8778
9 H -0.6683 -2.5842 2.0307
10 H 4.6501 0.6128 1.7193
11 H -4.7132 -0.7022 2.0398
12 H -5.3469 -2.7265 0.7365
13 H 3.4487 3.4467 -1.3109
14 H -3.6929 -3.6753 -0.9226
15 H 1.2308 2.3023 -1.3819
16 H -1.4857 -2.5371 -1.1822
17 C 0.3556 -2.2426 2.1936
18 C -0.4002 2.1660 2.1188
19 C 2.1884 -0.7395 1.7214
20 C -2.2573 0.6468 1.8285
21 C 1.1531 -2.9094 3.1290
22 C -1.1393 2.8760 3.0702
23 C 2.9883 -1.4105 2.6623
24 C -2.9987 1.3616 2.7848
25 C 2.4716 -2.4965 3.3660
26 C -2.4407 2.4772 3.4058
27 C 2.6498 0.4076 0.9378
28 C -2.7646 -0.5346 1.1284
29 C 3.9132 1.0084 1.0220

30 C -4.0193 -1.1299 1.3178
31 C -4.3715 -2.2605 0.5893
32 C 3.2539 2.5914 -0.6649
33 C -3.4609 -2.7906 -0.3306
34 C 2.0184 1.9608 -0.7128
35 C -2.2321 -2.1638 -0.4835
36 Ir -0.0885 -0.0856 0.0439
37 C 4.2194 2.1031 0.2216
38 H 5.2016 2.5734 0.2871
39 C -0.9379 1.0406 1.4708
40 C 1.4649 -2.2732 -1.6360
41 C 0.2417 -0.7279 -2.8689
42 C -0.6183 0.4229 -2.8660
43 C -1.0909 0.9535 -4.0902
44 C -0.6893 0.3316 -5.3215
45 C 0.1385 -0.7545 -5.3248
46 C 0.6291 -1.3161 -4.0967
47 C 1.4869 -2.4387 -4.0392
48 C 1.9053 -2.9117 -2.8087
49 H 2.5682 -3.7729 -2.7289
50 H 1.8096 -2.9186 -4.9645
51 H -1.0604 0.7502 -6.2582
52 H 0.4417 -1.2190 -6.2641
53 C -1.7501 2.0300 -1.6250
54 C -1.9405 2.0820 -4.0260
55 H -2.3265 2.5194 -4.9480
56 C -2.2705 2.6148 -2.7930
57 H -2.9245 3.4822 -2.7081
58 N -0.9427 0.9695 -1.6576
59 N 0.6524 -1.2163 -1.6619
60 H 1.7675 -2.6211 -0.6479
61 H -1.9826 2.4255 -0.6358
62 C 0.8503 -1.1480 1.4636
63 N 1.7184 0.9006 0.0641

³Ir-A

Atom X Y Z

1 N -1.8819 -1.0789 0.2736
2 H 0.8629 -3.9040 3.5626
3 H -0.8208 3.8888 3.4348
4 H 4.1027 -1.1845 2.7387
5 H -4.1052 1.1435 2.9390
6 H 3.2119 -3.1597 3.9283
7 H -3.1431 3.1692 3.9784
8 H 0.5472 2.5840 1.8352
9 H -0.6048 -2.6754 1.9919
10 H 4.7307 0.5305 1.5831
11 H -4.8049 -0.6209 1.9027
12 H -5.3805 -2.6778 0.6197
13 H 3.3992 3.4538 -1.3049
14 H -3.6479 -3.6848 -0.9138
15 H 1.1538 2.3363 -1.2761
16 H -1.4049 -2.5736 -1.0737
17 C 0.4201 -2.3372 2.1478
18 C -0.4661 2.2566 2.0700
19 C 2.2511 -0.7952 1.6745
20 C -2.3229 0.6998 1.7823
21 C 1.2445 -3.0310 3.0298
22 C -1.2346 2.9931 2.9677
23 C 3.0721 -1.4956 2.5641
24 C -3.0875 1.4433 2.6871
25 C 2.5689 -2.6111 3.2383
26 C -2.5437 2.5874 3.2765
27 C 2.6858 0.3785 0.9128
28 C -2.8035 -0.5070 1.1047
29 C 3.9594 0.9552 0.9430
30 C -4.0731 -1.0779 1.2388
31 C -4.3928 -2.2265 0.5195

32 C 3.2265 2.5880 -0.6668
33 C -3.4375 -2.7893 -0.3305
34 C 1.9772 1.9785 -0.6608
35 C -2.1903 -2.1835 -0.4290
36 Ir -0.0855 -0.0857 0.1473
37 C 4.2323 2.0675 0.1512
38 H 5.2229 2.5232 0.1704
39 C -0.9828 1.0915 1.4639
40 C 1.4931 -2.2739 -1.5441
41 C 0.2343 -0.7115 -2.7817
42 C -0.6039 0.4110 -2.7783
43 C -1.0951 0.9680 -4.0037
44 C -0.6820 0.3368 -5.2334
45 C 0.1436 -0.7486 -5.2371
46 C 0.6420 -1.3235 -4.0114
47 C 1.4932 -2.4261 -3.9578
48 C 1.9287 -2.9037 -2.7000
49 H 2.5975 -3.7603 -2.6210
50 H 1.8175 -2.9078 -4.8819
51 H -1.0494 0.7543 -6.1734
52 H 0.4460 -1.2090 -6.1802
53 C -1.7747 2.0316 -1.5295
54 C -1.9402 2.0748 -3.9418
55 H -2.3268 2.5155 -4.8625
56 C -2.2879 2.6105 -2.6802
57 H -2.9493 3.4722 -2.5945
58 N -0.9411 0.9672 -1.5477
59 N 0.6563 -1.2118 -1.5533
60 H 1.8061 -2.6251 -0.5607
61 H -2.0197 2.4278 -0.5439
62 C 0.8951 -1.2014 1.4573
63 N 1.7141 0.9099 0.1125

Ir-B

Atom X Y Z

1 N 0.8061 -1.0838 -1.6597
2 N 1.0369 -1.0842 1.5244
3 N -1.8363 -1.0857 0.1327
4 H 0.6275 -0.8545 -5.0270
5 H -0.6425 0.8107 -4.9966
6 H 2.1149 -2.8385 -4.8519
7 H 2.7650 -3.6807 -2.5589
8 H -2.7323 3.6871 -2.5766
9 H 1.8687 -2.4723 -0.5557
10 H -1.8930 2.5957 -0.5237
11 H 0.8394 -3.6863 3.6642
12 H -0.8674 3.6909 3.6499
13 H 4.0363 -0.8459 3.0638
14 H -4.0094 0.8170 3.0449
15 H 3.1448 -2.8359 4.2574
16 H -3.1329 2.8057 4.2278
17 H 0.4957 2.5941 1.9036
18 H -0.4461 -2.4794 1.8846
19 H 4.6405 0.8262 1.9562
20 H -4.6675 -0.8513 1.9637
21 H -5.2567 -2.8386 0.5918
22 H 3.5899 3.7016 -1.0646
23 H -3.5922 -3.6853 -1.1119
24 H 1.3996 2.5990 -1.3771
25 H -1.4088 -2.4778 -1.3356
26 C 0.4452 -0.6051 -2.8856
27 C -0.4321 0.5711 -2.8449
28 C 0.9152 -1.2366 -4.0486
29 C -0.9120 1.1959 -4.0109
30 C 1.7484 -2.3458 -3.9498
31 C 2.1114 -2.8174 -2.6834
32 C -2.0847 2.8094 -2.6561
33 C 1.6176 -2.1524 -1.5674

34 C -1.6103 2.1902 -1.4972
35 C 0.5532 -2.1560 2.1772
36 C -0.4903 2.1907 2.1426
37 C 2.2760 -0.6010 1.8297
38 C -2.2472 0.5730 1.7931
39 C 1.2722 -2.8202 3.1638
40 C -1.2594 2.8128 3.1289
41 C 3.0474 -1.2314 2.8197
42 C -3.0197 1.2006 2.7879
43 C 2.5473 -2.3440 3.4879
44 C -2.5312 2.3203 3.4568
45 C 2.6775 0.5779 1.0526
46 C -2.7197 -0.6046 1.0548
47 C 3.9233 1.2087 1.2269
48 C -3.9624 -1.2354 1.2279
49 C -4.2918 -2.3464 0.4590
50 C 3.3364 2.8216 -0.4666
51 C -3.3744 -2.8206 -0.4853
52 C 2.0997 2.1963 -0.6425
53 C -2.1607 -2.1560 -0.6146
54 Ir 0.0018 0.0262 -0.0007
55 C 1.7233 1.0615 0.1061
56 C -0.7705 1.0579 -1.5454
57 C 4.2562 2.3304 0.4716
58 C -1.7369 2.3146 -3.9217
59 H 5.2232 2.8183 0.6094
60 H -2.1077 2.7975 -4.8280
61 C -0.9514 1.0583 1.4393

³Ir-B

Atom X Y Z

1 N 0.7818 -1.0846 -1.6258
2 N 1.0091 -1.0610 1.5372
3 N -1.8318 -1.1015 0.1267

4 H 0.6189 -0.8459 -5.0470
5 H -0.5463 0.9564 -5.0150
6 H 2.0900 -2.8298 -4.8665
7 H 2.7534 -3.6660 -2.5458
8 H -2.7285 3.7381 -2.5360
9 H 1.8507 -2.4815 -0.5474
10 H -1.9306 2.5809 -0.4783
11 H 0.7689 -3.5968 3.7471
12 H -0.9286 3.6846 3.6662
13 H 4.0216 -0.8432 3.0497
14 H -4.0259 0.7655 3.0485
15 H 3.0960 -2.7815 4.3001
16 H -3.1776 2.7593 4.2398
17 H 0.4475 2.6226 1.9043
18 H -0.5024 -2.4146 1.9433
19 H 4.6532 0.7807 1.8916
20 H -4.6419 -0.9350 1.9922
21 H -5.1888 -2.9416 0.6349
22 H 3.6409 3.6154 -1.1804
23 H -3.5251 -3.7537 -1.0835
24 H 1.4164 2.5619 -1.4413
25 H -1.3683 -2.4857 -1.3341
26 C 0.4308 -0.5636 -2.8943
27 C -0.3743 0.5951 -2.8580
28 C 0.9115 -1.2268 -4.0679
29 C -0.8396 1.2925 -4.0192
30 C 1.7270 -2.3262 -3.9690
31 C 2.0979 -2.8047 -2.6714
32 C -2.0766 2.8656 -2.6080
33 C 1.5972 -2.1459 -1.5553
34 C -1.6282 2.2113 -1.4598
35 C 0.5042 -2.1020 2.2212
36 C -0.5311 2.2024 2.1453
37 C 2.2568 -0.5937 1.8224

38 C -2.2641 0.5581 1.7887
39 C 1.2159 -2.7537 3.2209
40 C -1.3071 2.8043 3.1396
41 C 3.0232 -1.2135 2.8224
42 C -3.0435 1.1658 2.7905
43 C 2.5035 -2.2969 3.5224
44 C -2.5697 2.2887 3.4646
45 C 2.6742 0.5627 1.0198
46 C -2.7134 -0.6342 1.0571
47 C 3.9368 1.1660 1.1639
48 C -3.9352 -1.2999 1.2487
49 C -4.2398 -2.4233 0.4879
50 C 3.3728 2.7558 -0.5606
51 C -3.3216 -2.8798 -0.4651
52 C 2.1193 2.1587 -0.7103
53 C -2.1277 -2.1856 -0.6115
54 Ir -0.0150 0.0596 -0.0240
55 C 1.7290 1.0531 0.0714
56 C -0.7727 1.0950 -1.5202
57 C 4.2883 2.2614 0.3792
58 C -1.6644 2.3903 -3.8875
59 H 5.2683 2.7275 0.4966
60 H -2.0112 2.9079 -4.7860
61 C -0.9816 1.0729 1.4367

Ir-C

Atom X Y Z

1 N 0.8007 -1.0918 -1.6595
2 N 1.0393 -1.0919 1.5207
3 N -1.8351 -1.0938 0.1377
4 H 0.6281 -0.8532 -5.0267
5 H -0.6292 0.8315 -4.9926
6 H 2.1039 -2.8449 -4.8537
7 H 2.7464 -3.6968 -2.5622

8 H 1.8537 -2.4892 -0.5571
9 H -1.8966 2.6131 -0.5183
10 H 0.8530 -3.7022 3.6509
11 H 4.0388 -0.8477 3.0595
12 H -4.0118 0.8357 3.0350
13 H 3.1552 -2.8444 4.2456
14 H -3.1185 2.8489 4.2059
15 H 0.5001 2.6130 1.9020
16 H -0.4381 -2.4947 1.8759
17 H 4.6331 0.8438 1.9597
18 H -4.6668 -0.8531 1.9675
19 H -5.2523 -2.8479 0.6067
20 H -3.5872 -3.7022 -1.0924
21 H 1.3932 2.6175 -1.3806
22 H -1.4052 -2.4937 -1.3228
23 C 0.4442 -0.6077 -2.8845
24 C -0.4260 0.5723 -2.8421
25 C 0.9122 -1.2388 -4.0486
26 C -0.8973 1.2081 -4.0040
27 C 1.7390 -2.3525 -3.9508
28 C 2.0978 -2.8297 -2.6851
29 C -2.0453 2.7961 -2.6449
30 C 1.6062 -2.1653 -1.5682
31 C -1.5978 2.1903 -1.4783
32 C 0.5597 -2.1682 2.1696
33 C -0.4814 2.1912 2.1217
34 C 2.2766 -0.6046 1.8271
35 C -2.2485 0.5735 1.7877
36 C 1.2818 -2.8326 3.1533
37 C -1.2696 2.7989 3.0900
38 C 3.0511 -1.2355 2.8143
39 C -3.0209 1.2113 2.7743
40 C 2.5553 -2.3521 3.4783
41 C -2.5385 2.3366 3.4381

42 C 2.6725 0.5778 1.0548
43 C -2.7190 -0.6081 1.0570
44 C 3.9113 1.2184 1.2318
45 C -3.9609 -1.2396 1.2338
46 C -4.2882 -2.3551 0.4710
47 C 3.3062 2.8067 -0.4410
48 C -3.3704 -2.8336 -0.4708
49 C 2.0750 2.1962 -0.6410
50 C -2.1577 -2.1687 -0.6039
51 Ir 0.0009 0.0189 -0.0003
52 C 1.7163 1.0583 0.1066
53 C -0.7675 1.0549 -1.5403
54 C 4.2428 2.3455 0.4837
55 C -1.7147 2.3325 -3.9180
56 H 5.1963 2.8599 0.6034
57 H -2.0921 2.8431 -4.8040
58 C -0.9499 1.0554 1.4340
59 F -0.7866 3.8966 3.7310
60 F 3.6174 3.9063 -1.1779
61 F -2.8427 3.8932 -2.5468

³Ir-C

Atom X Y Z

1 N 0.8401 -1.0716 -1.6119
2 N 0.9851 -1.1089 1.4625
3 N -1.8514 -1.0474 0.1576
4 H 0.7194 -0.8797 -4.9820
5 H -0.6091 0.7483 -4.9914
6 H 2.2764 -2.8022 -4.7542
7 H 2.9148 -3.5951 -2.4419
8 H 1.9299 -2.4058 -0.4678
9 H -2.0324 2.5319 -0.5651
10 H 0.8383 -3.8572 3.4195
11 H 4.0435 -0.9817 2.9566

12 H -3.9840 0.9261 3.1050
13 H 3.1836 -3.0556 4.0114
14 H -3.0446 2.9410 4.2092
15 H 0.5414 2.6518 1.8565
16 H -0.4780 -2.5471 1.7538
17 H 4.5823 0.8874 2.0524
18 H -4.6431 -0.8330 2.0875
19 H -5.2812 -2.7847 0.6943
20 H -3.6793 -3.5837 -1.1087
21 H 1.4000 2.6186 -1.3828
22 H -1.4910 -2.4041 -1.3612
23 C 0.4903 -0.6140 -2.8460
24 C -0.4210 0.5349 -2.8353
25 C 1.0024 -1.2392 -3.9941
26 C -0.9087 1.1271 -4.0134
27 C 1.8744 -2.3142 -3.8649
28 C 2.2315 -2.7589 -2.5871
29 C -2.1602 2.6756 -2.6978
30 C 1.6907 -2.1054 -1.4873
31 C -1.7010 2.1106 -1.5143
32 C 0.5251 -2.2393 2.0512
33 C -0.4404 2.2506 2.1062
34 C 2.2541 -0.6230 1.7939
35 C -2.2570 0.6204 1.8070
36 C 1.2608 -2.9603 2.9673
37 C -1.2061 2.8745 3.0758
38 C 3.0439 -1.3445 2.7162
39 C -2.9969 1.2854 2.8124
40 C 2.5668 -2.5007 3.3031
41 C -2.4813 2.4111 3.4399
42 C 2.6403 0.5851 1.1028
43 C -2.7251 -0.5567 1.1145
44 C 3.8758 1.2468 1.3024
45 C -3.9660 -1.2002 1.3166

46 C -4.3232 -2.2869 0.5380
47 C 3.3012 2.8227 -0.4092
48 C -3.4336 -2.7392 -0.4655
49 C 2.0728 2.2058 -0.6298
50 C -2.2189 -2.0868 -0.6128
51 Ir -0.0246 0.0655 0.0448
52 C 1.7059 1.0893 0.1319
53 C -0.8068 1.0273 -1.5552
54 C 4.2088 2.3684 0.5514
55 C -1.7826 2.2090 -3.9559
56 H 5.1541 2.8933 0.6950
57 H -2.1696 2.6864 -4.8560
58 C -0.9439 1.1165 1.4356
59 F -0.7186 3.9741 3.7072
60 F 3.6301 3.9188 -1.1473
61 F -3.0119 3.7287 -2.6282

Ir-D

Atom X Y Z

1 N 0.8049 -1.0866 -1.6553
2 N 1.0329 -1.0865 1.5226
3 N -1.8338 -1.0889 0.1317
4 H 0.6692 -0.8783 -5.0217
5 H 2.1561 -2.8570 -4.8145
6 H 2.7782 -3.6867 -2.5087
7 H 1.8524 -2.4659 -0.5274
8 H -1.9062 2.5759 -0.4816
9 H 0.7898 -3.6905 3.6540
10 H 4.0139 -0.8729 3.0919
11 H 3.0947 -2.8555 4.2722
12 H -3.0807 2.8462 4.2205
13 H 0.5350 2.5773 1.8899
14 H -0.4641 -2.4707 1.8618
15 H -4.6818 -0.8806 1.9316

16 H -5.2441 -2.8615 0.5430
17 H -3.5576 -3.6916 -1.1479
18 H 1.3655 2.5799 -1.4076
19 H -1.3802 -2.4693 -1.3386
20 C 0.4568 -0.6119 -2.8886
21 C -0.4233 0.5625 -2.8418
22 C 0.9445 -1.2528 -4.0411
23 C -0.9264 1.2353 -3.9663
24 C 1.7782 -2.3604 -3.9193
25 C 2.1257 -2.8251 -2.6482
26 C -2.0751 2.7920 -2.6026
27 C 1.6157 -2.1531 -1.5443
28 C -1.6132 2.1718 -1.4503
29 C 0.5334 -2.1553 2.1671
30 C -0.4503 2.1730 2.1203
31 C 2.2735 -0.6091 1.8398
32 C -2.2493 0.5621 1.7864
33 C 1.2351 -2.8269 3.1605
34 C -1.2183 2.7938 3.0953
35 C 3.0283 -1.2493 2.8384
36 C -2.9725 1.2356 2.7832
37 C 2.5080 -2.3593 3.4973
38 C -2.4891 2.3503 3.4529
39 C 2.6708 0.5673 1.0560
40 C -2.7285 -0.6137 1.0489
41 C 3.8939 1.2439 1.1853
42 C -3.9699 -1.2554 1.2035
43 C -4.2803 -2.3643 0.4220
44 C 3.2849 2.8008 -0.4902
45 C -3.3528 -2.8291 -0.5140
46 C 2.0582 2.1770 -0.6691
47 C -2.1424 -2.1562 -0.6252
48 Ir 0.0004 0.0130 -0.0004
49 C 1.7183 1.0494 0.0994

50 C -0.7748 1.0462 -1.5390
51 C 4.2301 2.3595 0.4324
52 C -1.7496 2.3489 -3.8821
53 H 5.1892 2.8578 0.5620
54 H -2.1196 2.8443 -4.7780
55 C -0.9456 1.0465 1.4394
56 F -0.7215 3.8806 3.7345
57 F 3.5880 3.8886 -1.2394
58 F -2.8784 3.8778 -2.4915
59 F -4.2138 0.8102 3.1432
60 F -0.6174 0.8102 -5.2214
61 F 4.8262 0.8207 2.0815

³Ir-D

Atom X Y Z

1 N 0.8171 -1.0663 -1.6649
2 N 1.0136 -1.0913 1.5253
3 N -1.7761 -1.0958 0.1203
4 H 0.6535 -0.8646 -5.0292
5 H 2.1910 -2.8043 -4.8281
6 H 2.8528 -3.6123 -2.5266
7 H 1.9071 -2.4153 -0.5397
8 H -1.9182 2.5752 -0.4728
9 H 0.7356 -3.6976 3.6459
10 H 3.9828 -0.8999 3.1158
11 H 3.0390 -2.8767 4.2833
12 H -3.0876 2.8921 4.1655
13 H 0.5636 2.5814 1.8781
14 H -0.4966 -2.4637 1.8390
15 H -4.6509 -0.9430 2.0112
16 H -5.2082 -2.9105 0.6180
17 H -3.5355 -3.6750 -1.1589
18 H 1.4167 2.5681 -1.4209
19 H -1.3784 -2.4581 -1.3758

20 C 0.4506 -0.5999 -2.8945
21 C -0.4470 0.5612 -2.8424
22 C 0.9446 -1.2299 -4.0499
23 C -0.9765 1.2190 -3.9637
24 C 1.8073 -2.3155 -3.9311
25 C 2.1772 -2.7682 -2.6621
26 C -2.1256 2.7738 -2.5947
27 C 1.6573 -2.1090 -1.5553
28 C -1.6414 2.1670 -1.4442
29 C 0.4996 -2.1589 2.1593
30 C -0.4267 2.1875 2.1051
31 C 2.2539 -0.6196 1.8505
32 C -2.2708 0.5555 1.7581
33 C 1.1889 -2.8356 3.1572
34 C -1.1975 2.8190 3.0710
35 C 2.9963 -1.2676 2.8532
36 C -3.0003 1.2786 2.7620
37 C 2.4618 -2.3751 3.5047
38 C -2.4973 2.3756 3.4089
39 C 2.6688 0.5551 1.0707
40 C -2.7099 -0.5986 1.0856
41 C 3.9024 1.2122 1.2045
42 C -3.9558 -1.2967 1.2548
43 C -4.2625 -2.3843 0.4825
44 C 3.3344 2.7654 -0.4909
45 C -3.3244 -2.8245 -0.5116
46 C 2.0980 2.1605 -0.6743
47 C -2.1189 -2.1455 -0.6373
48 Ir -0.0077 0.0330 -0.0023
49 C 1.7359 1.0494 0.1040
50 C -0.7930 1.0510 -1.5417
51 C 4.2622 2.3156 0.4445
52 C -1.8119 2.3229 -3.8741
53 H 5.2283 2.7991 0.5787

54 H -2.1990 2.8080 -4.7685
55 C -0.9228 1.0728 1.4186
56 F -0.7157 3.9058 3.7239
57 F 3.6620 3.8401 -1.2462
58 F -2.9376 3.8503 -2.4802
59 F -4.2555 0.8714 3.1013
60 F -0.6810 0.7884 -5.2180
61 F 4.8198 0.7820 2.1108

Ir-E

Atom X Y Z

1 N 0.7767 -1.0277 -1.6510
2 N 1.0113 -1.0823 1.5047
3 N -1.0035 1.0317 1.4424
4 N -1.8293 -1.1047 0.1110
5 H 0.6676 -0.7921 -5.0228
6 H -0.7002 0.9288 -4.9917
7 H 2.1638 -2.7712 -4.8128
8 H 2.7576 -3.6111 -2.5085
9 H -2.7628 3.6578 -2.6436
10 H 1.8206 -2.4198 -0.5237
11 H -1.9711 2.6134 -0.5375
12 H 0.7908 -3.6158 3.7136
13 H -0.7704 3.6430 3.5582
14 H 3.9736 -0.7652 3.1069
15 H -3.9433 0.7591 3.0908
16 H 3.0669 -2.7245 4.3387
17 H -3.0354 2.7639 4.2486
18 H 0.4626 2.4676 1.7417
19 H -0.4646 -2.4917 1.8697
20 H 4.6187 0.8817 1.8808
21 H -4.5931 -0.9121 2.0482
22 H -5.1999 -2.9052 0.6935
23 H 3.6959 3.5731 -1.1360

24 H -3.5884 -3.7226 -1.0673
25 H 1.4665 2.5551 -1.5071
26 H -1.4325 -2.4955 -1.3737
27 C 0.4430 -0.5570 -2.8896
28 C -0.4476 0.6095 -2.8466
29 C 0.9356 -1.1750 -4.0394
30 C -0.9331 1.2428 -3.9760
31 C 1.7734 -2.2815 -3.9204
32 C 2.1053 -2.7502 -2.6482
33 C -2.1058 2.7899 -2.6475
34 C 1.5904 -2.0952 -1.5361
35 C -1.6517 2.1902 -1.4885
36 C 0.5196 -2.1387 2.1746
37 C -0.5127 2.1183 2.0717
38 C 2.2371 -0.5867 1.8285
39 C -2.2264 0.5418 1.8022
40 C 1.2264 -2.7602 3.1989
41 C -1.2106 2.7678 3.0823
42 C 2.9946 -1.1658 2.8487
43 C -2.9698 1.1561 2.8117
44 C 2.4850 -2.2633 3.5402
45 C -2.4607 2.2787 3.4597
46 C 2.6415 0.5702 1.0099
47 C -2.6878 -0.6451 1.0592
48 C 3.8684 1.1946 1.1576
49 C -3.9103 -1.2814 1.2861
50 C -4.2494 -2.3985 0.5251
51 C 3.3524 2.7213 -0.5521
52 C -3.3614 -2.8548 -0.4492
53 C 2.1153 2.1345 -0.7410
54 C -2.1592 -2.1777 -0.6275
55 Ir -0.0148 0.0167 -0.0522
56 C 1.7054 1.0370 0.0374
57 C -0.8016 1.0704 -1.5432

58 N 4.2083 2.2531 0.3872
59 N -1.7483 2.3174 -3.8673
60 C -2.2316 2.9724 -5.0989
61 H -1.3750 3.3837 -5.6477
62 H -2.7544 2.2328 -5.7177
63 H -2.9184 3.7782 -4.8243
64 C 5.5261 2.8881 0.5898
65 H 5.5880 3.2644 1.6183
66 H 6.3136 2.1449 0.4139
67 H 5.6323 3.7170 -0.1160

³Ir-E

Atom X Y Z

1 N 0.7904 -1.0280 -1.6604
2 N 1.0248 -1.0815 1.5045
3 N -0.9631 1.0380 1.4213
4 N -1.7876 -1.1143 0.0945
5 H 0.6578 -0.7954 -5.0316
6 H -0.7272 0.9122 -4.9921
7 H 2.1734 -2.7602 -4.8294
8 H 2.7915 -3.5918 -2.5286
9 H -2.7730 3.6454 -2.6341
10 H 1.8550 -2.4082 -0.5382
11 H -1.9551 2.6108 -0.5310
12 H 0.7932 -3.6071 3.7205
13 H -0.7707 3.5985 3.6015
14 H 3.9828 -0.7639 3.1140
15 H -4.0148 0.8054 3.0371
16 H 3.0691 -2.7183 4.3478
17 H -3.0923 2.7692 4.2279
18 H 0.4869 2.4499 1.7986
19 H -0.4553 -2.4853 1.8670
20 H 4.6347 0.8778 1.8837
21 H -4.6205 -0.9005 2.0576

22 H -5.2320 -2.8748 0.6980
23 H 3.7207 3.5716 -1.1336
24 H -3.6087 -3.6961 -1.0742
25 H 1.4885 2.5578 -1.5069
26 H -1.4372 -2.5111 -1.3739
27 C 0.4444 -0.5611 -2.8967
28 C -0.4517 0.6013 -2.8486
29 C 0.9350 -1.1751 -4.0495
30 C -0.9509 1.2294 -3.9754
31 C 1.7843 -2.2735 -3.9348
32 C 2.1297 -2.7375 -2.6645
33 C -2.1137 2.7794 -2.6411
34 C 1.6156 -2.0866 -1.5495
35 C -1.6463 2.1855 -1.4843
36 C 0.5286 -2.1348 2.1755
37 C -0.4998 2.0905 2.0870
38 C 2.2500 -0.5855 1.8301
39 C -2.2518 0.5195 1.7619
40 C 1.2314 -2.7541 3.2037
41 C -1.2193 2.7404 3.1041
42 C 3.0039 -1.1634 2.8537
43 C -3.0247 1.1849 2.7951
44 C 2.4901 -2.2582 3.5465
45 C -2.5154 2.2689 3.4506
46 C 2.6570 0.5710 1.0115
47 C -2.6757 -0.6040 1.0731
48 C 3.8855 1.1925 1.1601
49 C -3.9450 -1.2738 1.2914
50 C -4.2808 -2.3672 0.5389
51 C 3.3746 2.7205 -0.5502
52 C -3.3786 -2.8343 -0.4502
53 C 2.1361 2.1364 -0.7401
54 C -2.1492 -2.1652 -0.6236
55 Ir -0.0003 0.0193 -0.0550

56 C 1.7238 1.0404 0.0381
57 C -0.7954 1.0673 -1.5447
58 N 4.2282 2.2502 0.3899
59 N -1.7674 2.3022 -3.8623
60 C -2.2643 2.9518 -5.0916
61 H -1.4138 3.3624 -5.6502
62 H -2.7917 2.2088 -5.7024
63 H -2.9497 3.7574 -4.8130
64 C 5.5473 2.8822 0.5939
65 H 5.6093 3.2572 1.6229
66 H 6.3332 2.1375 0.4176
67 H 5.6557 3.7117 -0.1109

Ru

Atom X Y Z

1 Ru -0.0332 -0.0934 -0.0163
2 C 1.4591 -2.3037 -1.6003
3 C 0.2689 -0.7438 -2.8440
4 C -0.5632 0.4207 -2.8387
5 C -1.0299 0.9658 -4.0571
6 C -0.6515 0.3289 -5.2888
7 C 0.1469 -0.7798 -5.2941
8 C 0.6338 -1.3505 -4.0681
9 C 1.4627 -2.4940 -4.0053
10 C 1.8733 -2.9641 -2.7705
11 H 2.5133 -3.8417 -2.6841
12 H 1.7694 -2.9920 -4.9263
13 H -1.0177 0.7533 -6.2248
14 H 0.4296 -1.2552 -6.2344
15 C -1.6434 2.0477 -1.5806
16 C -1.8476 2.1167 -3.9834
17 H -2.2289 2.5698 -4.8996
18 C -2.1517 2.6515 -2.7442
19 H -2.7798 3.5369 -2.6494

20 N -0.8659 0.9628 -1.6191
21 N 0.6742 -1.2239 -1.6283
22 H 1.7635 -2.6582 -0.6162
23 H -1.8650 2.4523 -0.5937
24 N 0.9902 -1.1586 1.4645
25 C 2.2516 -0.6825 1.6983
26 C 0.5571 -2.1952 2.1863
27 C 2.6811 0.4339 0.9123
28 C 3.1181 -1.2484 2.6619
29 C 1.3525 -2.8147 3.1664
30 H -0.4525 -2.5461 1.9756
31 C 3.9756 0.9735 1.0931
32 N 1.7873 0.9369 0.0063
33 C 4.4289 -0.6830 2.8289
34 C 2.6324 -2.3474 3.4070
35 H 0.9448 -3.6582 3.7228
36 C 4.8401 0.3808 2.0766
37 C 4.3389 2.0754 0.2853
38 C 2.1595 1.9764 -0.7446
39 H 5.0924 -1.1252 3.5735
40 H 3.2681 -2.8134 4.1612
41 H 5.8372 0.8026 2.2103
42 C 3.4286 2.5711 -0.6311
43 H 5.3287 2.5217 0.3917
44 H 1.4240 2.3514 -1.4554
45 H 3.6748 3.4184 -1.2705
46 N -1.8477 -1.1154 0.1933
47 C -2.6661 -0.5711 1.1456
48 C -2.2809 -2.1822 -0.4830
49 C -2.1753 0.5761 1.8466
50 C -3.9425 -1.0974 1.4512
51 C -3.5374 -2.7663 -0.2442
52 H -1.6056 -2.5881 -1.2355
53 C -2.9612 1.1846 2.8523

54 N -0.9372 1.0383 1.4919
55 C -4.7254 -0.4606 2.4747
56 C -4.3705 -2.2303 0.7218
57 H -3.8349 -3.6373 -0.8274
58 C -4.2549 0.6318 3.1469
59 C -2.4166 2.3114 3.5101
60 C -0.4469 2.1019 2.1332
61 H -5.7091 -0.8724 2.7043
62 H -5.3488 -2.6678 0.9263
63 H -4.8564 1.1071 3.9231
64 C -1.1601 2.7637 3.1482
65 H -2.9891 2.8102 4.2934
66 H 0.5420 2.4403 1.8263
67 H -0.7085 3.6278 3.6347

³Ru

Atom X Y Z

1 Ru -0.0057 -0.1080 0.0394
2 C 1.6111 -2.2087 -1.6311
3 C 0.2764 -0.7257 -2.8285
4 C -0.6243 0.3839 -2.7885
5 C -1.1678 0.9017 -3.9859
6 C -0.7873 0.2992 -5.2337
7 C 0.0822 -0.7541 -5.2730
8 C 0.6422 -1.2991 -4.0673
9 C 1.5455 -2.3866 -4.0348
10 C 2.0280 -2.8355 -2.8181
11 H 2.7261 -3.6696 -2.7589
12 H 1.8543 -2.8600 -4.9679
13 H -1.2100 0.7041 -6.1540
14 H 0.3665 -1.2037 -6.2251
15 C -1.7757 1.9281 -1.4783
16 C -2.0612 1.9919 -3.8727
17 H -2.5046 2.4222 -4.7717

18 C -2.3625 2.4984 -2.6211
19 H -3.0477 3.3365 -2.4999
20 N -0.9252 0.9032 -1.5614
21 N 0.7532 -1.1871 -1.6348
22 H 1.9718 -2.5456 -0.6598
23 H -1.9859 2.3058 -0.4790
24 N 1.0117 -1.1507 1.5208
25 C 2.2840 -0.6937 1.7346
26 C 0.5468 -2.1618 2.2601
27 C 2.7262 0.4052 0.9366
28 C 3.1416 -1.2635 2.7013
29 C 1.3361 -2.7815 3.2423
30 H -0.4742 -2.4830 2.0577
31 C 4.0319 0.9201 1.0982
32 N 1.8363 0.9160 0.0339
33 C 4.4646 -0.7230 2.8524
34 C 2.6297 -2.3392 3.4635
35 H 0.9148 -3.6050 3.8175
36 C 4.8917 0.3224 2.0824
37 C 4.4044 2.0037 0.2691
38 C 2.2137 1.9334 -0.7404
39 H 5.1238 -1.1677 3.5988
40 H 3.2574 -2.8097 4.2215
41 H 5.8979 0.7254 2.2036
42 C 3.4960 2.5036 -0.6472
43 H 5.4028 2.4340 0.3582
44 H 1.4764 2.3093 -1.4494
45 H 3.7532 3.3363 -1.3007
46 N -1.8076 -1.1370 0.1956
47 C -2.6679 -0.5256 1.1004
48 C -2.2441 -2.1969 -0.4975
49 C -2.1853 0.5958 1.7763
50 C -3.9901 -1.0262 1.3312
51 C -3.5242 -2.7309 -0.3253

52 H -1.5404 -2.6397 -1.2032
53 C -2.9752 1.2550 2.7599
54 N -0.8878 1.0285 1.4581
55 C -4.7908 -0.3390 2.3055
56 C -4.4015 -2.1434 0.5993
57 H -3.8183 -3.6019 -0.9092
58 C -4.3054 0.7428 2.9881
59 C -2.4124 2.3371 3.4458
60 C -0.3511 2.0626 2.1889
61 H -5.8014 -0.7045 2.4969
62 H -5.4021 -2.5518 0.7518
63 H -4.9252 1.2444 3.7338
64 C -1.0685 2.7245 3.1540
65 H -2.9905 2.8648 4.2050
66 H 0.6690 2.3469 1.9351
67 H -0.6023 3.5525 3.6875

Bibliography

- [1] Till J. B. Zähringer, Max Wienhold, Ryan Gilmour, and Christoph Kerzig. Direct observation of triplet states in the isomerization of alkenylboronates by energy transfer catalysis. *J. Am. Chem. Soc.*, 145(39):21576–21586, 2023. PMID: 37729087.