Mechanochemical Generation of Aryl Barium Nucleophiles from Unactivated Barium Metal

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1. Chemicals and instrumentation

Materials were obtained from commercial suppliers and purified using standard procedures unless otherwise noted. Solvents were purchased from commercial suppliers and further dried over molecular sieve (MS 4Å). Barium pieces (> 99.99%, product no. 474711) were purchased from Aldrich and stored in a glovebox to prevent oxidation. All reactions were performed using grinding vessels in the Retsch MM 400 (Figure S1). Both jars and balls were made of stainless steel (SUS400B and SUS420J2, respectively, Figure S2). NMR spectra were recorded on JEOL JNM-EC X400P and JNM-ECS400 spectrometers (¹H: 392 or 396 or 399 or 401 MHz, ¹³C: 99 or 100 MHz). Tetramethylsilane (¹H) and CDCl₃ (¹³C) were employed as external standards, respectively. Multiplicity was recorded as follows: s = singlet, brs = broad singlet, d = doublet, t = triplet, q = quartet, and m = multiplet. 1,1,2,2-Tetrachloroethane, dibromomethane, and anisole were used as an internal standard to determine the NMR yields. Recycling preparative gel permeation chromatography (GPC) was conducted with a JAI LaboACE LC-5060 using CHCl₃ as the eluent with JAIGEL-1H. Preparative thin-layer chromatography (PTLC) was performed using Wakogel B5-F silica-coated plates (0.75 mm) prepared in our laboratory. High-resolution mass spectra were recorded at the Global Facility Center, Hokkaido University.



Figure S1. Retsch MM400 used in this study.



Figure S2. Stainless-steel jar (5 mL) and two stainless-steel balls (10 mm) used in this study.

2. List of substrates used in this study

Aryl halides, except **1k**, were obtained from commercial suppliers and were used as received. **1k** was synthesized according to the reported procedures.^[1]



Figure S3. The list of aryl halides used in this study.

Electrophiles, except **3k**, were obtained from commercial suppliers. **3a**, **3b**, and **3d** were distilled before use. **3c**, **3e-3j**, and **5a-5e** were used as received. **3k** was synthesized according to the reported procedures.^[2]



Figure S4. The list of electrophiles used in this study.

3. General procedure for the generation of aryl barium nucleophile and subsequent reactions with electrophiles

Caution: You should store barium metal in a glovebox, because barium metal is rapidly oxidated in air. If you use barium metal stored in air for a long time, reproducibility will become poor.

Procedure for nucleophilic addition to various electrophiles (A)



4-Iodobiphenyl (1a, 1.0 mmol, 4.0 equiv) was placed in a milling jar (stainless steel, 5 mL) loaded with two grinding balls (stainless-steel, diameter: 10 mm). The milling jar was then transferred to a glovebox. In a glovebox, Ba pieces (99.99 %, Aldrich) (137.3 mg, 1.0 mmol, 4.0 equiv) were placed in a milling jar. After the milling jar was removed from the glovebox, the jar was opened under air, and then an electrophile (**3a-d**, 0.25 mmol, 1.0 equiv) and THF (160 μ L, 2.0 mmol, 8.0 equiv) were quickly added to the jar. After the jar was closed without purging with inert gas, the jar was placed in a ball mill (Retsch MM 400, 30 min, 30 Hz). After grinding for 30 min, the jar was opened in air, quenched with H₂O, and extracted with CH₂Cl₂ three times. The organic layer was washed with brine and dried over MgSO₄, and then filtrated. After the removal of the solvents under reduced pressure, yield of the corresponding products was determined by ¹H NMR analysis with tetrachloroethane or anisole as the internal standard.

Procedure for nucleophilic addition to hydrosilanes (B)



A solid organic halide (1, 1.0 mmol, 4.0 equiv) was placed in a milling jar (stainless steel, 5 mL) loaded with two grinding balls (stainless steel, diameter: 10 mm). The milling jar was then transferred to a glovebox. In a glovebox, Ba pieces (99.99 %, Aldrich) (137.3 mg, 1.0 mmol, 4.0 equiv) were placed in a milling jar. After the milling jar was removed from the glovebox, the jar was opened under air, then a hydrosilane (**3**, 0.25 mmol, 1.0 equiv) and THF (160 μ L, 2.0 mmol, 8.0 equiv) were added to the jar. After the jar was closed without purging with inert gas, the jar was placed in a ball mill

(Retsch MM 400, 30 min, 30 Hz). After grinding for 30 min, the jar was opened in air, quenched with H₂O, and extracted with CH₂Cl₂ three times. The organic layer was washed with brine, dried over MgSO₄, and then filtrated. After the removal of the solvents under reduced pressure, the NMR yield of the corresponding products was determined by ¹H NMR analysis with tetrachloroethane or anisole as the internal standard. The crude material was purified by recycling gel permeation chromatography (GPC) using CHCl₃ as the eluent or flash column chromatography (SiO₂, only hexane) or preparative thin-layer chromatography (PTLC) to give the corresponding product.

When a liquid organic iodide 1 was used, 1 was added to the jar at the same time as a hydrosilane and THF.

Procedure for nucleophilic addition to diaryl ketones (C)



A solid organic halide (1, 0.50 or 0.375 mmol, 2.0 or 1.5 equiv) was placed in a milling jar (stainless steel, 5 mL) loaded with two grinding balls (stainless steel, diameter: 10 mm). The milling jar was then transferred to a glovebox. In a glovebox, Ba pieces (99.99 %, Aldrich) (0.50 or 0.375 mmol, 2.0 or 1.2 equiv) were placed in a milling jar. After the milling jar was removed from the glovebox, the jar was opened under air, then a diaryl ketone (5, 0.25 mmol, 1.0 equiv) and THF (160 μ L, 2.0 mmol, 8.0 equiv) were added to the jar. After the jar was closed without purging with inert gas, the jar was placed in a ball mill (Retsch MM 400, 30 min, 30 Hz). After grinding for 30 min, the jar was opened in air, quenched with H₂O, and extracted with CH₂Cl₂ three times. The organic layer was washed with brine, dried over MgSO₄, and then filtrated. After the removal of the solvents under reduced pressure, the NMR yield of the corresponding products was determined by ¹H NMR analysis with dibromomethane as the internal standard. The crude material was purified by flash column chromatography (SiO₂, typically hexane/CH₂Cl₂ = 50/50, then hexane/Et₂O =95/5 to 75/25).

When a liquid organic iodide 1 was used, 1 was added to the jar at the same time as the hydrosilane and THF.

4. Protonation and deuteration experiments

Aryl barium species are highly unstable. Once formed under mechanochemical conditions, they can decompose rapidly. Therefore, the mass balance did not match in the following protonation experiments. The crude reaction mixture was carefully checked, but no byproducts were detected.

Ph	1 0 mmol	Ba(0) piece additive 5 ml jar 10 mm ball × 2 (stainless-steel) ball milling (30 Hz time, in air	Ph-BaX	CH ₃ CO ₂ H (1 5 ml jar 10 mm ball = (stainless-st ball milling 10 min	10 equiv) × 2 eel) (30 Hz)	Ph- 2 GC yield ^b
entry	halide	Ba (equiv)	additive	time	yield of 2 (%)	conversion of 1a, j (%)
1	1j (X = B	r) 1.0	THF (2.0 equiv)	60 min	29	63
2	1j	1.0	MTBE (2.0 equiv)	60 min	11	45
3	1j	1.0	2-MeTHF (2.0 equiv)	60 min	16	39
4	1j	1.0	1,4-dioxane (2.0 equiv)	60 min	13	29
5	1j	1.0	THP (2.0 equiv)	60 min	16	41
6	1a (X =) 1.0	THF (2.0 equiv)	60 min	52	>99
7	1a	1.0	THF (2.0 equiv)	30 min	52	>99
8	1a	1.0	THP (2.0 equiv)	30 min	22	37
9	1a	1.0	none	30 min	33	>99
10 ^c	1a	1.0	THF (2.0 equiv)	30 min	44	97
11 ^c	1a	1.5	THF (2.0 equiv)	30 min	54	>99
12	1a	1.0	THF (4.0 equiv)	30 min	52	98
13	1a	2.0	THF (2.0 equiv)	30 min	50	>99
14	1a	2.0	THF (4.0 equiv)	30 min	29	>99
15	1k (X = C	CI) 1.0	THF (2.0 equiv)	30 min	11	28

Table S1. Optimization study on protonation

^{*a*}Reactions were performed using Retsch MM400, a stainless-steel milling jar (5 mL), and a stainless-steel balls (10 mm × 2). Conditions: **1a** or **1j** (0.50 mmol), Barium (>99.99 %, Aldrich) (1.0–2.0 equiv), additive (2.0–4.0 equiv). ^{*b*}Yields were determined by GC analysis with dibenzyl as an internal standard. ^{*c*}Barium (>99.99%, Aldrich) that was stored under air for 1 month was used.

5. Reaction using barium metal with different purity

1. Dendritic pieces, purified by distillation, >99.99%, Aldrich-474711



2. Dendritic pieces, purified by distillation, >99.9%, Aldrich-441880



6. Barbier-type ketone arylation with 1c and magnesium metal

The reaction of sterically hindered **1c** in the presence of magnesium metal afforded the desired product **6g** in poor yield (31%), and significant amounts of other byproducts were formed (pinacol coupling byproduct **7g**: 24%; reduction byproduct **8g**: 22%).



In contrast, the reaction of **1c** with barium metal proceeded smoothly to give the desired tertiary alcohol **6g** in a better yield (66%) without the generation of these byproducts.



7. Details of scaled-up reaction



1d (841.9 mg, 3.0 mmol, 1.2 equiv) was placed in a milling jar (stainless steel, 10 mL) loaded with two grinding balls (stainless steel, diameter: 10 mm). The milling jar was then transferred to a glovebox. In a glovebox, Ba pieces (99.99 %, Aldrich) (435.4 mg, 3.0 mmol, 1.2 equiv) were placed in a milling jar. After the milling jar was removed from the glovebox, the jar was opened under air, then a di(4-methoxyphenyl)ketone (**5a**, 605.6 mg 2.5 mmol, 1.0 equiv) and THF (1.6 mL, 20 mmol, 8.0 equiv) were added to the jar. After the jar was closed without purging with inert gas, the jar was placed in a ball mill (Retsch MM 400, 30 min, 30 Hz). After grinding for 30 min, the jar was opened in air, quenched with H₂O, and extracted with CH₂Cl₂ three times. The organic layer was washed with brine, dried over MgSO₄, and then filtrated. After the removal of the solvents under reduced pressure, the NMR yield of the corresponding products was determined by ¹H NMR analysis with dibromomethane (314.6 mg) as the internal standard. The crude material was purified by flash column chromatography (SiO₂, hexane/CH₂Cl₂ = 50/50, then hexane/Et₂O = 95/5 to 85/15) to give the corresponding product **6a** in 77% yield (757.8 mg, 1.91 mmol) as a white solid.

8. Details of solution-based reactions

Ph-H Ba(0) piece (1.0 equiv) THF (0.5 M) rt, 2h Ph-Bal HCl aq. (1.0 M) then extraction Ph-Bal trace 5% conv. of 1a

Commercially available Ba pieces (99.99 %, Aldrich) 69.1 mg, 0.50 mmol, 1.0 equiv) were placed in an oven-dried reaction tube. After being sealed with a screw cap containing a TeflonTM-coated rubber septum, the vial was connected to a nitrogen line through a needle. 4-Iodobiphenyl (**1a**, 141.9 mg, 0.50 mmol, 1.0 equiv) and THF (1.0 mL) were added to the vial, and then the reaction mixture was stirred at rt for 2 h. After 2 h, the reaction mixture was quenched with 1M HCl, and extracted with ethyl acetate three times. The organic layer was washed with brine, dried over MgSO₄, and then filtrated. Dibenzyl (51.4 mg) was added to the solution as the internal standard, followed by GC analysis. The yields of protonation compound **2** and the conversion of **1a** were determined by GC.

Procedure for solution-based substitution reaction with Rieke barium

Procedure for solution-based reaction with unactivated barium



According to Rieke's method,^[3] lithium (2.2 mmol, 15.0 mg) and biphenyl (2.2 mmol, 344.7 mg) in dry THF (5.0 mL) were stirred under N₂ for 2 h. To a well-suspended solution of BaI₂ (1.1 mmol) in dry THF (5.0 mL), the prepared lithium biphenylide was transferred via a disposable syringe at room temperature. The reaction mixture was stirred for 1 h at room temperature. Theoretically, 1.1 mmol Rieke barium should be generated in situ.

The above suspension was cooled to -78 °C. Then, 2-iodotoluene (**1b**, 221.6 mg, 1.0 mmol) was added via a disposable syringe at -78 °C under N₂, and the mixture was allowed to warm to rt and stirred for 1 h. Then, hydrosilane (**3e**, 610 μ L, 4.0 mmol) was added at rt. The resulting mixture was stirred for 15 h. Finally, the mixture was quenched with NH₄Cl aq and extracted with Et₂O three times. The organic layer was washed with brine, dried over MgSO₄, and then filtrated. After the removal of the solvents under reduced pressure, the resulting crude mixture was analyzed by ¹H NMR with tetrachloroethane (27.5 mg) as an internal standard to obtain the NMR yield of **4b** in 56% yield.

9. Details of mechanistic study



Procedure for the reaction with chiral hydrosilanes (R)-3i

Two grinding balls (stainless, diameter: 10 mm) were placed in a milling jar (stainless, 5 mL). The milling jar was then transferred to a glovebox. In the glovebox, Ba pieces (99.99 %, Aldrich) (59.2 mg, 0.43 mmol, 4.3 equiv) were placed in a milling jar. After the milling jar was removed from glovebox, the jar was opened under air, then 1-iodonaphthalene (1e, 99.4 mg, 0.39 mmol, 3.9 equiv), a chiral hydrosilane [(R)-3i, 28.5 mg, 0.10 mmol, 1.0 equiv] and THF (65 µL, 0.8 mmol, 8.0 equiv) were added to the jar. After the jar was closed without purging with inert gas, the jar was placed in a ball mill (Retsch MM 400, 30 min, 30 Hz). After grinding for 30 min, the jar was opened in air, quenched with H₂O, and extracted with CH₂Cl₂ three times. The organic layer was washed with brine and dried over MgSO₄, and then filtrated. After the removal of the solvents under reduced pressure, the NMR yield of the corresponding products was determined by ¹H NMR analysis with tetrachloroethane (32.6 mg) as the internal standard. The crude material was purified by flash column chromatography (SiO₂, only hexane) and then purified by recycling gel permeation chromatography (GPC) using CHCl₃ as the eluent to give the corresponding product (*S*)-40 in 49% yield (20.1 mg, 0.05 mmol, 78% ee) as a white solid.

¹H NMR (401 MHz, CDCl₃, δ): 0.68 (s, 3H), 1.07–1.43 (m, 5H), 1.47–1.62 (m, 1H), 1.62–1.80 (m, 4H), 1.94 (d, *J* = 10.4 Hz, 1H), 7.29–7.37 (m, 2H), 7.38–7.64 (m, 10H), 7.79 (d, *J* = 6.1 Hz, 1H), 7.84 (d, *J* = 8.0 Hz, 1H), 7.89 (d, *J* = 8.4 Hz, 1H), 7.95 (d, *J* = 8.0 Hz, 1H). ¹³C NMR (101 MHz, CDCl₃, δ): -4.4 (CH₃), 25.2 (CH), 27.0 (CH₂), 28.1 (CH₂), 28.35 (CH₂), 28.40 (CH₂), 125.2 (CH), 125.5 (CH), 125.7 (CH), 126.6 (CH), 127.2 (CH), 127.5 (CH), 128.9 (CH), 129.0 (CH), 129.1 (CH), 130.3 (CH), 133.6 (C), 134.5 (C), 135.2 (CH) 135.4 (CH), 136.4 (C), 137.4 (C), 141.1 (C), 141.6 (C). EI (m/z): [M]⁺ calcd for C₂₉H₃₀Si: 406.2111, found: 406.2105. [α]_D²⁶–7.9 (c 1.61 in CHCl₃, 78% ee). Daicel CHIRALCEL[®] OD-3, hexane 100%, 0.5 mL/min, 40 °C, *S* isomer: t_S = 36.08 min; for the racemic compound: *R* isomer: t_R = 30.19 min, *S* isomer: t_S = 37.87 min.

10. Computational details

As it is difficult to determine the structure of ArBaI in a solid state, the AFIR method^[4] is used to suggest the possible geometry of both the monomeric and dimeric species via a comprehensive and unbiased sampling calculation. This initial sampling calculation was conducted at the semi-empirical GFN-xTB level of theory^[5] as implemented in ORCA 4.0, ^[6] and a collision energy of 200 kJ·mol⁻¹ was applied to the entire molecule for any possible bond rearrangements.^[4] The AFIR sampling calculation rendered in total 3729 possible isomers and conformers for dimeric ArBaI(THF)₂, of which the ones having the lowest energies were further reoptimized at the DFT level of theory with the B3LYP hybrid functional^[7] as implemented in Gaussian 16.^[8] Note that the energy profile shown in this study was generated with the geometries and energies derived from the following DFT calculations. In order to describe the dispersion properly, an explicit dispersion correction term called GD3,^[9] was also employed in the DFT calculations. The polarized triple- ζ 6-311G(d,p) basis set^[10] was used for C, H, O, and Si atoms during both the geometry optimization and the single-point calculation processes. The Stuttgart/Dresden pseudo-potential basis set SDD,^[11] as well as a dpolarization function,^[12] are used for Ba ($\zeta = 0.438$) and I ($\zeta = 0.289$) atoms. All the minima and transition states were fully optimized without any constraints. Frequency calculations were carried out to characterize all the optimized structures as minima or transition states. Transition states were identified by having one imaginary frequency. An intrinsic reaction coordinate (IRC) calculation^[13] was performed for each transition state to ensure that it connected the correct reactants and products. The free energies were computed at 298.15 K and 1 atm. All the geometries shown in this article are visualized by the CYLview software.^[14]

NBO Analysis on Intermediate II



Bonding orbitals involving Si atom

(Occupancy) Bond orbital/ Coefficients/ Hybrids

9. (1.62375) BD (1) C 3 -Si 85

(81.67%) 0.9037* C 3 s(21.09%)p 3.74(78.90%)d 0.00(0.01%)
(18.33%) 0.4281*Si 85 s(18.24%)p 2.83(51.58%)d 1.66(30.18%)

88. (1.81345) BD (1)Si 85 - H 86

(25.64%) 0.5064*Si 85 s(16.22%)p 3.90(63.34%)d 1.26(20.44%) (74.36%) 0.8623* H 86 s(99.91%)p 0.00(0.09%)

89. (1.74000) BD (1)Si 85 - C 87

(17.89%) 0.4229*Si 85 s(18.03%)p 2.93(52.81%)d 1.62(29.16%)
(82.11%) 0.9062*C 87 s(26.74%)p 2.74(73.26%)d 0.00(0.01%)

90. (1.94672) BD (1)Si 85 - C 98

(26.62%) 0.5159*Si 85 s(23.66%)p 2.91(68.96%)d 0.31(7.37%) (73.38%) 0.8566* C 98 s(32.03%)p 2.12(67.91%)d 0.00(0.05%)

91. (1.93477) BD (1)Si 85 - C 102

(26.07%) 0.5106*Si 85 s(24.05%)p 2.74(65.84%)d 0.42(10.11%)
(73.93%) 0.8598* C 102 s(32.14%)p 2.11(67.81%)d 0.00(0.05%)



Figure S5. Visualised bonding orbitals involving the Si atom

Cartesian Coordinates of Optimized Structures

(E: Electronic Energy; G: Gibbs Free Energy at 298.15 Kelvin and under 1 atm)

 $(\mathbf{E} = -773.13566312 \text{ a.u.}; \mathbf{G} = -772.84639608 \text{ a.u.})$

01

Μ

С	-0.036517556013	-6.783214924819	4.427970938485
С	-0.893523270465	-5.705223841684	4.183202298798
С	-0.532921403465	-4.654590471228	3.306375160038
С	0.737354496103	-4.763339620861	2.708047068188
С	1.606373114511	-5.835568474701	2.945458051636
С	1.213988938233	-6.852313264665	3.810453641672
Н	-0.339465405460	-7.578790654058	5.104977007953
Н	1.079027879957	-3.986203222369	2.020439394793
Н	2.577184358572	-5.876251076766	2.459606781025
Н	1.871685538278	-7.692736937031	4.006936132036
Ba	-1.911405475589	-2.474663235208	2.466957356781
0	-0.624303017405	-1.286210744773	4.560818495655
С	0.283305138384	-0.177205627753	4.338839892670
С	0.024950699582	-2.113220585371	5.561409198964
С	1.668455536826	-0.825438569594	4.300288415703
Н	0.177665568556	0.531741250052	5.169545537595
Н	0.001577042112	0.304806843003	3.402519620451
С	1.522597688822	-2.072751394825	5.214771649866
Н	-0.182242885887	-1.683468476888	6.548696157416
Н	-0.402644817640	-3.112543170802	5.490057250075
Н	1.896639269576	-1.120942293038	3.274922640002
Н	2.445815316928	-0.138531217750	4.638397583615
Н	2.130135126694	-1.998537221297	6.118371651410
Н	1.808008903242	-2.980514631665	4.682855708679
0	-2.623907969352	-4.193749103827	0.463069241534
С	-2.509488029664	-5.625524460742	0.677433424141
С	-2.143550730519	-3.967944152956	-0.886915091443
С	-1.244807764868	-6.066037320055	-0.085830025254
Н	-3.414762016201	-6.103742391581	0.286168189221
Η	-2.435061436945	-5.791918596830	1.751139471527
С	-0.899573679502	-4.847709317213	-0.981560938686

Η	-2.925357480696	-4.273270605387	-1.593953059752
Н	-1.939759834721	-2.902813746001	-0.997686719397
Н	-0.434101220310	-6.282066839714	0.609594329770
Н	-1.443673096280	-6.962794335616	-0.675619965803
Н	-0.669437023935	-5.125145195490	-2.011314761801
Н	-0.046103476034	-4.300583523859	-0.576475146059
Ι	0.041614611162	-0.823421410377	0.387277455791
С	-2.245803399108	-5.658546059295	4.873526612416
Н	-3.065430234129	-5.640507435292	4.143152384565
Н	-2.415526728127	-6.517499394930	5.528467489149
Н	-2.345489235476	-4.754844658738	5.488053842075

D

(E = -1546.33776865 a.u.; G = -1545.72917366 a.u.)

С	-6.821684757979	-0.922524167192	-0.477972356623
С	-5.477501328414	-1.284431286895	-0.332991848294
С	-4.429481067533	-0.463878073595	-0.817459009189
С	-4.841393300276	0.711704176674	-1.472850298194
С	-6.181013317308	1.085489931153	-1.638261288483
С	-7.178970866302	0.261582679489	-1.127759958181
Η	-7.604758112356	-1.566556079050	-0.082709012924
Η	-4.092927170800	1.402872347303	-1.880470989703
Η	-6.441047162631	2.007246455506	-2.151924000349
Η	-8.224933539996	0.531827559746	-1.232935650320
Ba	-1.735849652464	-0.202806306015	-0.340152420564
0	-1.710010225089	-2.380544213676	1.285473165325
С	-1.497576400031	-3.721891207963	0.772272882705
С	-1.498285142524	-2.356192004980	2.724114144769
С	-0.690665583483	-4.431150017878	1.852994554855
Η	-0.977048351160	-3.636034978671	-0.183200024002
Η	-2.474127278573	-4.191324853641	0.613325134963
С	-1.261549045559	-3.810271576668	3.137008086673
Η	-0.627144648725	-1.730862065749	2.922207781972
Η	-2.381118890409	-1.911331788341	3.189366395138
Н	-0.803854457717	-5.515809823733	1.811431316200

Η	0.367461622260	-4.181238496008	1.752419535182
Н	-0.580493050174	-3.882232136616	3.986155564751
Н	-2.204225463678	-4.295219268356	3.406256593713
0	-2.858911810760	1.210854914271	1.742378630198
С	-4.040908275687	0.689473040597	2.393652872448
С	-3.007020013835	2.647931807218	1.740309477902
С	-5.193989966994	1.628813145206	1.995293020814
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Η	-2.055399242796	-3.513915704720	-2.782787584312
Η	-0.371773108092	-3.463522334143	1.187457324358
Η	-4.253041057355	-2.983004177580	-1.840692334397
Η	-2.540244637872	-2.975328561578	2.109929982954
Η	-4.536931741117	-2.718685784835	0.617085313766
С	0.580260933975	-5.997383066902	-1.776703261666
Η	0.489482832331	-6.457889381830	-0.789348700628
Η	-0.276475789842	-6.320460342712	-2.375681631148
Η	1.489548219623	-6.385804253739	-2.245521448499
С	0.692216777912	-3.420601588062	-3.414344628052
Η	1.606895875051	-3.707074566645	-3.936668738312
Η	-0.141119663531	-3.845420798260	-3.982643301841
Н	0.589991893906	-2.333539412461	-3.444007381087

11. Characterization of products

[1,1'-Biphenyl]-4-yldimethyl(phenyl)silane (4a).

The reaction was performed according to the general procedure. The reaction was conducted with aryl iodide **1a** (281.4 mg, 1.0 mmol) and hydrosilane **3e** (33.9 mg, 0.25 mmol). The resulting crude mixture was analyzed by ¹H NMR with anisole (25.2 mg) as an internal standard to obtain the NMR yield of **4a** in 79% yield. The product **4a** was purified by GPC and obtained in 84% yield (60.1 mg, 0.21 mmol) as a colorless liquid. ¹H and ¹³C NMR of the product **4a** were in agreement with the literature.^[15] ¹H NMR (399 MHz, CDCl₃, δ): 0.59 (s, 6H), 7.31–7.40 (m, 4H), 7.40–7.47 (m, 2H), 7.54–7.62 (m, 8H). ¹³C NMR (100 MHz, CDCl₃, δ): –2.2 (CH₃), 126.7 (CH), 127.3 (CH), 127.5 (CH), 128.0 (CH), 128.9 (CH), 129.3 (CH), 134.3 (CH), 134.8 (CH), 137.1 (C), 138.3 (C), 141.2 (C), 142.0 (C). EI (m/z): [M]⁺ calcd for C₂₀H₂₀Si: 288.1329, found: 288.1327.

Dimethyl(phenyl)(o-tolyl)silane (4b).



The reaction was performed according to the general procedure. The reaction was conducted with aryl iodide **1b** (216.4 mg, 1.0 mmol) and hydrosilane **3e** (34.3 mg, 0.25 mmol). The resulting crude mixture was analyzed by ¹H NMR with tetrachloroethane (47.8 mg) as an internal standard to obtain the NMR yield of **4b** in 79% yield. The product **4b** was purified by PTLC and obtained in 91% yield (51.6 mg, 0.23 mmol) as a colorless liquid. ¹H and ¹³C NMR of the product **4b** were in agreement with the literature.^[16]

¹H NMR (399 MHz, CDCl₃, δ): 0.58 (s, 6H), 2.27 (s, 3H), 7.12–7.23 (m, 2H), 7.27–7.39 (m, 4H), 7.45–7.51 (m, 3H). ¹³C NMR (100 MHz, CDCl₃, δ): –1.3 (*C*H₃), 23.3 (*C*H₃), 125.1 (*C*H), 128.0 (*C*H), 129.0 (*C*H), 129.7 (*C*H), 130.0 (*C*H), 134.1 (*C*H), 135.5 (*C*H), 136.3 (*C*), 139.0 (*C*), 144.2 (*C*). EI (m/z): [M]⁺ calcd for C₁₅H₁₈Si: 226.1172, found: 226.1172.
(2-Isopropylphenyl)dimethyl(phenyl)silane (4c).



The reaction was performed according to the general procedure. The reaction was conducted with aryl iodide **1c** (247.2 mg, 1.0 mmol) and hydrosilane **3e** (33.9 mg, 0.25 mmol). The resulting crude mixture was analyzed by ¹H NMR with anisole (25.4 mg) as an internal standard to obtain the NMR yield of **4c** in 80% yield. The product **4c** was purified by GPC and obtained in 82% yield (51.9 mg, 0.20 mmol) as a colorless liquid.

¹H NMR (399 MHz, CDCl₃, δ): 0.58 (s, 6H), 1.03 (d, *J* = 6.8 Hz, 6H), 2.97 (sept, *J* = 6.7Hz, 1H), 7.15–7.22 (m, 1H), 7.27–7.43 (m, 5H), 7.43–7.56 (m, 3H). ¹³C NMR (100 MHz, CDCl₃, δ): –0.7 (CH₃), 24.4 (CH₃), 34.1 (CH), 125.3 (CH), 125.5 (CH), 127.8 (CH), 129.0 (CH), 130.1 (CH), 134.1 (CH), 135.2 (C), 135.4 (C), 139.6 (C), 155.6 (C). EI (m/z): [M]⁺ calcd for C₁₇H₂₂Si: 254.1485, found: 254.1489

[1,1'-Biphenyl]-2-yldimethyl(phenyl)silane (4d).



4d

with the literature.^[17]

The reaction was performed according to the general procedure. The reaction was conducted with aryl iodide **1d** (281.8 mg, 1.0 mmol) and hydrosilane **3e** (34.1 mg, 0.25 mmol). The resulting crude mixture was analyzed by ¹H NMR with anisole (25.2 mg) as an internal standard to obtain the NMR yield of **4d** in 76% yield. The product **4d** was purified by GPC and obtained in 82% yield (59.4 mg, 0.21 mmol) as a colorless liquid by GPC separation. ¹H and ¹³C NMR of the product **4d** were in agreement

¹H NMR (399 MHz, CDCl₃, δ): 0.17 (s, 6H), 7.06–7.11 (m, 2H), 7.19–7.25 (m, 3H), 7.25–7.44 (m, 8H), 7.62 (dd, *J* = 7.5, 1.2 Hz, 1H). ¹³C NMR (100 MHz, CDCl₃, δ): –1.0 (*C*H₃), 126.4 (*C*H), 127.1 (*C*H), 127.7 (*C*H), 127.8 (*C*H), 128.8 (*C*H), 129.0 (*C*H), 129.5 (*C*H), 129.8 (*C*H), 134.1 (*C*H), 135.9 (*C*H), 136.5 (*C*), 140.1 (*C*), 144.2 (*C*), 149.7 (*C*). EI (m/z): [M]⁺ calcd for C₂₀H₂₀Si: 288.1329, found: 288.1332.

Dimethyl(naphthalen-1-yl)(phenyl)silane (4e).



4e

The reaction was performed according to the general procedure. The reaction was conducted with aryl iodide **1e** (255.6 mg, 1.0 mmol) and hydrosilane **3e** (33.6 mg, 0.25 mmol). The resulting crude mixture was analyzed by ¹H NMR with anisole (25.0 mg) as an internal standard to obtain the NMR yield of **4e** in 76% yield. The product **4e** was purified by GPC and obtained in 82% yield (53.1 mg, 0.20 mmol) as a colorless liquid. ¹H and ¹³C NMR of the product **4e** were in agreement with the literature.^[15] ¹H NMR (399 MHz, CDCl₃, δ): 0.70 (s, 6H), 7.30–7.58 (m, 8H), 7.72 (dd, *J* = 7.0, 1.0 Hz, 1H), 7.86 (dd, *J* = 8.0, 1.2 Hz, 1H), 7.91 (t, *J* = 8.5 Hz, 2H). ¹³C NMR (100 MHz, CDCl₃, δ): -0.81 (*CH*₃), 125.2 (*CH*), 125.5 (*CH*), 125.8 (*CH*), 128.0 (*CH*), 128.7 (*CH*), 129.1 (*CH*), 129.2 (*CH*), 130.4 (*CH*), 133.5 (*C*), 134.3 (*CH*), 134.8 (*CH*), 135.8 (*C*), 137.1 (*C*), 139.0 (*C*). EI (m/z): [M]⁺ calcd for C₁₈H₁₈Si: 262.1172, found: 262.1172.

(2-Methoxyphenyl)dimethyl(phenyl)silane (4f).



The reaction was performed according to the general procedure. The reaction was conducted with aryl iodide **1f** (235.7 mg, 1.0 mmol) and hydrosilane **3e** (34.3 mg, 0.25 mmol). The resulting crude mixture was analyzed by ¹H NMR with tetrachloroethane (45.4 mg) as an internal standard to obtain the NMR yield of **4f** in 94% yield. The product **4f** was purified by GPC and obtained in 61% yield (37.3 mg, 0.15 mmol) as a colorless liquid. ¹H and ¹³C NMR of product **4f** were in agreement with the literature.^[17]

¹H NMR (400 MHz, CDCl₃, δ): 0.55 (s, 6H), 3.74 (s, 3H), 6.83 (d, *J* = 8.4 Hz, 1H), 6.91 (dt, *J* = 9.9, 7.2 Hz, 1H), 7.24–7.28 (m, 1H) 7.30–7.39 (m, 4H), 7.50–7.60 (m, 2H). ¹³C NMR (100 MHz, CDCl₃, δ): –2.1 (*C*H₃), 55.1 (*C*H₃), 109.8 (*C*H), 120.6 (*C*H), 126.2 (*C*), 127.7 (*C*H), 128.8 (*C*H), 131.2 (*C*H), 134.3 (*C*H), 136.1 (*C*H), 139.1 (*C*), 164.6 (*C*). EI (m/z): [M]⁺ calcd for C₁₅H₁₈OSi: 242.1121, found: 242.1123.

4-[Dimethyl(phenyl)silyl]-N,N-diphenylaniline (4g).



The reaction was performed according to the general procedure. The reaction was conducted with aryl iodide **1g** (370.4 mg, 1.0 mmol) and hydrosilane **3e** (33.9 mg, 0.25 mmol). The resulting crude mixture was analyzed by ¹H NMR with tetrachloroethane (47.3 mg) as an internal standard to obtain the NMR yield of **4g** in 48% yield. The product **4g** was purified by GPC and obtained in 56% yield (53.1 mg, 0.14 mmol) as a colorless liquid. ¹H and ¹³C NMR of the product **4g** were in agreement with the literature.^[18]

¹H NMR (399 MHz, CDCl₃, δ): 0.53 (s, 6H), 6.99–7.05 (m, 4H), 7.08–7.13 (m, 4H), 7.22–7.29 (m, 4H), 7.32–7.39 (m, 5H), 7.51–7.58 (m, 2H). ¹³C NMR (100 MHz, CDCl₃, δ): –2.1 (CH₃), 122.4 (CH), 123.2 (CH), 124.9 (CH), 127.9 (CH), 129.1 (CH), 129.4 (CH), 130.8 (C), 134.3 (CH), 135.2 (CH), 138.7 (C), 147.6 (C), 148.7 (C). EI (m/z): [M]⁺ calcd for C₂₆H₂₅NSi: 379.1751, found: 379.1750.

Dimethyl(phenyl)(4-(trifluoromethyl)phenyl)silane (4h).





The reaction was performed according to the general procedure. The reaction was conducted with aryl iodide **1h** (271.8 mg, 1.0 mmol) and hydrosilane **3e** (34.4 mg, 0.25 mmol). The resulting crude mixture was analyzed by ¹H NMR with anisole (47.6 mg) as an internal standard to obtain the NMR yield of **4h** in 57% yield. The product **4h** was purified by GPC and obtained in 48% yield (33.7 mg, 0.208 mmol) as a colorless liquid. ¹H and ¹³C NMR of the product **4h** were in agreement with the literature.^[15]

¹H NMR (399 MHz, CDCl₃, δ): 0.58 (s, 6H), 7.32–7.44 (m, 3H), 7.46–7.55 (m, 2H), 7.55–7.66 (m, 4H). ¹³C NMR (100 MHz, CDCl₃, δ): –2.5 (*C*H₃), 124.5 (q, *J*_{C-F} = 3.8 Hz, *C*H), 126.6 (q, *J*_{C-F} = 223.0 Hz, *C*), 128.1 (*C*H), 129.6 (*C*H), 131.2 (q, *J*_{C-F} = 32.3 Hz, *C*), 134.3 (*C*H), 134.6 (*C*H), 137.2 (*C*), 143.5 (*C*). EI (m/z): [M]⁺ calcd for C₁₅H₁₅F₃Si: 280.0890, found: 280.0891.

(4-Fluorophenyl)dimethyl(phenyl)silane (4i).



The reaction was performed according to the general procedure. The reaction was conducted with aryl iodide **1i** (223.7 mg, 1.0 mmol) and hydrosilane **3e** (33.5 mg, 0.25 mmol). The resulting crude mixture was analyzed by ¹H NMR with tetrachloroethane (28.5 mg) as an internal standard to obtain the NMR yield of **4i** in 58% yield. The product **4i** was purified by flash column chromatography (SiO₂, only hexane) and obtained in 36% yield (20.2 mg, 0.09 mmol) as a colorless liquid. ¹H and ¹³C NMR of the product **4i** were in agreement with the literature.^[16]

¹H NMR (399 MHz, CDCl₃, δ): 0.54 (s, 6H), 6.99–7.09 (m, 2H), 7.31–7.39 (m, 3H), 7.45–7.54 (m, 4H). ¹³C NMR (100 MHz, CDCl₃, δ): –2.2 (*C*H₃), 115.1 (d, *J*_{C-F} = 19.2 Hz, *C*H₂), 128.0 (*C*H), 129.4 (*C*H), 133.8 (d, *J*_{C-F} = 2.8 Hz, *C*), 134.2 (*C*H), 136.2 (d, *J*_{C-F} = 7.7 Hz, *C*H), 138.1 (*C*), 163.9 (d, *J*_{C-F} = 248.1 Hz, *C*). EI (m/z): [M]⁺ calcd for C₁₄H₁₅FSi: 230.0922, found: 230.0924.

Methyldiphenyl(o-tolyl)silane (4j).





¹H NMR (399 MHz, CDCl₃, δ): 0.87 (s, 3H), 2.22 (s, 3H), 7.12 (t, *J* = 7.6 Hz, 1H), 7.18 (d, *J* = 8.0 Hz, 1H), 7.26–7.42 (m, 8H), 7.46–7.53 (m, 4H). ¹³C NMR (100 MHz, CDCl₃, δ): –2.2 (CH₃), 23.7 (CH₃), 125.1 (CH), 128.0 (CH), 129.4 (CH), 130.0 (CH), 130.1 (CH), 134.5 (C), 135.3 (CH), 136.7 (C), 137.1 (CH), 144.7 (C). EI (m/z): [M]⁺ calcd for C₂₀H₂₀Si: 288.1329, found: 288.1337.

Benzyldimethyl(o-tolyl)silane (4k).





The reaction was performed according to the general procedure. The reaction was conducted with aryl iodide **1b** (219.1 mg, 1.0 mmol) and hydrosilane **3g** (38.1 mg, 0.25 mmol). The resulting crude mixture was analyzed by ¹H NMR with tetrachloroethane (47.3 mg) as an internal standard to obtain the NMR yield of **4k** in 68% yield. The product **4k** was purified by PTLC (hexane/ethyl acetate, 98:2) and GPC and obtained in 69% yield (60.1 mg, 0.18 mmol) as a colorless liquid. ¹H NMR of the product **4k** was in agreement with the literature.^[19]

¹H NMR (399 MHz, CDCl₃, δ): 0.28 (s, 6H), 2.38 (s, 2H), 2.44 (s, 3H), 6.94 (d, *J* = 8.0 Hz, 2H), 7.03–7.09 (m, 1H), 7.12–7.33 (m, 5H), 7.41 (d, *J* = 7.6Hz, 1H). ¹³C NMR (100 MHz, CDCl₃, δ): –2.0 (CH₃), 23.4 (CH₃), 26.3 (CH₂), 124.2 (CH), 125.1 (CH), 128.2 (CH), 128.4 (CH), 129.5 (CH), 130.0 (CH), 134.9 (CH), 136.8 (C), 140.0 (C), 143.8 (C). EI (m/z): [M]⁺ calcd for C₁₆H₂₀Si: 240.1329, found: 240.1326.

(1,1'-Biphenyl)-4-yltriethylsilane (4l).





¹H NMR (399 MHz, CDCl₃, δ): 0.82 (q, *J* = 7.8 Hz, 6H), 0.99 (t, *J* = 8.0 Hz, 9H), 7.33 (t, *J* = 6.6 Hz, 1H), 7.43 (t, *J* = 7.8 Hz, 2H), 7.54–7.63 (m, 6H). ¹³C NMR (100 MHz, CDCl₃, δ): 3.5 (*C*H₂), 7.6 (*C*H₃), 126.5 (*C*H), 127.3 (*C*H), 127.4 (*C*H), 128.9 (*C*H), 134.8 (*C*H), 136.4 (*C*), 141.3 (*C*), 141.5 (*C*). EI (m/z): [M]⁺ calcd for C₁₈H₂₄Si: 268.1642, found: 268.1641.

(1,1'-Biphenyl)-4-yl(*tert*-butyl)dimethylsilane (4m).

4m

The reaction was performed according to the general procedure. The reaction was conducted with aryl iodide **1a** (279.0 mg, 1.0 mmol) and hydrosilane **3i** (28.3 mg, 0.24 mmol). The resulting crude mixture was analyzed by ¹H NMR with tetrachloroethane (20.2 mg) as an internal standard to obtain the NMR yield of **4m** in 59% yield. The product **4m** was purified by GPC and obtained in 46% yield (30.0 mg, 0.11 mmol) as a colorless liquid. ¹H and ¹³C NMR of the product **4m** were in agreement with the literature.^[21]

¹H NMR (399 MHz, CDCl₃, δ): 0.30 (s, 6H), 0.91 (s, 9H), 7.34 (t, *J* = 7.2 Hz, 1H), 7.44 (t, *J* = 7.6 Hz, 2H), 7.56–7.63 (m, 6H). ¹³C NMR (100 MHz, CDCl₃, δ): –6.0 (CH₃), 17.1 (*C*), 26.7 (CH₃), 126.3 (CH), 127.3 (CH), 127.5 (CH), 128.9 (CH), 135.1 (CH), 136.7 (*C*), 141.2 (*C*), 141.6 (*C*). EI (m/z): [M]⁺ calcd for C₁₈H₂₄Si: 268.1642, found: 268.1642.

1-[(1,1'-biphenyl)-4-yl]-1,1,2,2,2-pentamethyldisilane (4n).



4n

The reaction was performed according to the general procedure. The reaction was conducted with aryl iodide **1a** (279.5 mg, 1.0 mmol) and hydrosilane **3j** (32.7 mg, 0.25 mmol). The product **4n** was purified by flash column chromatography (SiO₂, only hexane) and GPC and obtained in 17% yield (11.7 mg, 0.04 mmol) as a yellow oil.

¹H NMR (401 MHz, CDCl₃, δ): 0.08 (s, 9H), 0.36 (s, 6H), 7.31–7.38 (m, 1H), 7.40–7.48 (m, 2H), 7.49–7.64 (m, 6H). ¹³C NMR (100 MHz, CDCl₃, δ): –3.8 (*C*H₃), –2.1 (*C*H₃), 126.6 (*C*H), 127.2 (*C*H), 127.4 (*C*H), 128.9 (*C*H), 134.4 (*C*H), 138.6 (*C*H), 141.1 (*C*), 141.3 (*C*). EI (m/z): [M]⁺ calcd for C₁₇H₂₄Si₂: 284.1411, found: 284.1411.

(S)-(1,1'-biphenyl)-4-yl(cyclohexyl)(methyl)(naphthalen-1-yl)silane [(S)-40].



The reaction was performed according to the general procedure. The reaction was conducted with aryl iodide **1e** (99.4 mg, 0.39 mmol) and hydrosilane (*R*)-**3k** (28.5 mg, 0.10 mmol). The product (*S*)-**4o** was purified by flash column chromatography (SiO₂, only hexane) and GPC and obtained in 49% yield (20.1 mg, 0.05 mmol, 78% ee) as a white solid. ¹H and ¹³C NMR of the product (*S*)-**4o** were in agreement with the literature.^[2]

¹H NMR (401 MHz, CDCl₃, δ): 0.68 (s, 3H), 1.07–1.43 (m, 5H), 1.47–1.62 (m, 1H), 1.62–1.80 (m, 4H), 1.94 (d, *J* = 10.4 Hz, 1H), 7.29–7.37 (m, 2H), 7.38–7.64 (m, 10H), 7.79 (d, *J* = 6.1 Hz, 1H), 7.84 (d, *J* = 8.0 Hz, 1H), 7.89 (d, *J* = 8.4 Hz, 1H), 7.95 (d, *J* = 8.0 Hz, 1H). ¹³C NMR (101 MHz, CDCl₃, δ): –4.4 (*C*H₃), 25.2 (*C*H), 27.0 (*C*H₂), 28.1 (*C*H₂), 28.35 (*C*H₂), 28.40 (*C*H₂), 125.2 (*C*H), 125.5 (*C*H), 125.7 (*C*H), 126.6 (*C*H), 127.2 (*C*H), 127.5 (*C*H), 128.9 (*C*H), 129.0 (*C*H), 129.1 (*C*H), 130.3 (*C*H), 133.6 (*C*), 134.5 (*C*), 135.2 (*C*H) 135.4 (*C*H), 136.4 (*C*), 137.4 (*C*), 141.1 (*C*), 141.6 (*C*). EI (m/z): [M]⁺ calcd for C₂₉H₃₀Si: 406.2111, found: 406.2105. [α]D²⁶ –7.9 (c 1.61 in CHCl₃, 78% ee). Daicel CHIRALCEL[®] OD-3, hexane 100%, 0.5 mL/min, 40 °C, *S* isomer: t_S = 36.08 min; for the racemic compound: *R* isomer: t_R = 30.19 min, *S* isomer: t_S = 37.87 min. The absolute configuration of **40** was determined by the retention time of the reported HPLC trace.^[2]

[1,1'-Biphenyl]-4-ylbis(4-methoxyphenyl)methanol (6a)



The reaction was performed according to the general procedure. The reaction was conducted with aryl iodide **1a** (142.1 mg, 0.50 mmol) and diaryl ketone **5a** (61.1 mg, 0.25 mmol). The product **6a** was purified by flash column chromatography (SiO₂, hexane/CH₂Cl₂ = 50/50 then hexane/Et₂O = 95/5 to 80/20) and obtained in 83% yield (82.9 mg, 0.21 mmol) as a white solid.

¹H NMR (396 MHz, CDCl₃, δ): 2.74 (s, 1H, OH), 3.81 (s, 6H), 6.83–6.88 (m, 4H), 7.18–7.25 (m, 4H), 7.33–7.38 (m, 3H), 7.40–7.46 (m, 2H), 7.51–7.56 (m, 2H), 7.56–7.62 (m, 2H). ¹³C NMR (100 MHz, CDCl₃, δ): 55.4 (*C*H₃), 81.4 (*C*), 113.3 (*C*H), 126.7 (*C*H), 127.2 (*C*H), 127.4 (*C*H), 128.3 (*C*H), 128.9 (*C*H), 129.3 (*C*H), 139.5 (*C*), 140.0 (*C*), 140.8 (*C*), 146.5 (*C*), 158.8 (*C*). EI (m/z): [M]⁺ calcd for C₂₇H₂₄O₃: 396.1720, found: 396.1706.

[1,1'-Biphenyl]-4-yldiphenylmethanol (6b)



The reaction was performed according to the general procedure. The reaction was conducted with aryl iodide **1a** (141.2 mg, 0.50 mmol) and diaryl ketone **5b** (46.0 mg, 0.25 mmol). The product **6b** was purified by flash column chromatography (SiO₂, hexane/CH₂Cl₂ = 50/50 then hexane/Et₂O = 90/10) and obtained in 94% yield (80.1 mg, 0.24 mmol) as a colorless oil.

¹H NMR (396 MHz, CDCl₃, δ): 2.83 (s, 1H, OH), 7.28–7.38 (m, 13H), 7.43 (t, *J* = 7.5 Hz, 2H), 7.54 (d, *J* = 7.9 Hz, 2H), 7.59 (d, *J* = 8.3 Hz, 2H). ¹³C NMR (100 MHz, CDCl₃, δ): 82.0 (*C*), 126.8 (*C*H), 127.2 (*C*H), 127.5 (*C*H), 128.05 (*C*H), 128.13 (*C*H), 128.5 (*C*H), 128.9 (*C*H), 140.2 (*C*), 140.7 (*C*), 146.0 (*C*), 146.9 (*C*). EI (m/z): [M]⁺ calcd for C₂₅H₂₀O: 336.1509, found: 336.1500.

[1,1'-Biphenyl]-4-yl(phenyl)(p-tolyl)methanol (6c)



The reaction was performed according to the general procedure. The reaction was conducted with aryl iodide **1a** (141.3 mg, 0.50 mmol) and diaryl ketone **5c** (50.0 mg, 0.25 mmol). The product **6c** was purified by flash column chromatography (SiO₂, hexane/CH₂Cl₂ = 50/50 then hexane/Et₂O = 90/10 to 80/20) and obtained in 86% yield (76.3 mg, 0.21 mmol) as a colorless oil.

¹H NMR (396 MHz, CDCl₃, δ): 2.35 (s, 3H), 2.79 (s, 1H, OH), 7.11–7.16 (m, 2H), 7.17–7.22 (m, 2H), 7.27–7.38 (m, 8H), 7.43 (t, *J* = 7.5 Hz, 2H), 7.51–7.56 (m, 2H), 7.56–7.61 (m, 2H). ¹³C NMR (100 MHz, CDCl₃, δ): 21.2 (*C*H₃), 81.9 (*C*), 126.7 (*C*H), 127.2 (*C*H), 127.38 (*C*H), 127.44 (*C*H), 128.0 (*C*H), 128.1 (*C*H), 128.5 (*C*H), 128.8 (*C*H), 128.9 (*C*H), 137.2 (*C*), 140.1 (*C*), 140.8 (*C*), 144.1 (*C*), 146.2 (*C*) 147.1 (*C*). EI (m/z): [M]⁺ calcd for C₂₆H₂₂O: 350.1665, found: 350.1648.

Di([1,1'-biphenyl]-4-yl)(phenyl)methanol (6d)



The reaction was performed according to the general procedure. The reaction was conducted with aryl iodide **1a** (140.2 mg, 0.50 mmol) and diaryl ketone **5d** (66.0 mg, 0.25 mmol). The product **6d** was purified by flash column chromatography (SiO₂, hexane/CH₂Cl₂ = 50/50 then hexane/Et₂O = 90/10 to 80/20) and obtained in 86% yield (90.1 mg, 0.21 mmol) as a white solid.

¹H NMR (400 MHz, CDCl₃, δ): 2.86 (s, 1H, OH), 7.28–7.49 (m, 15H), 7.52–7.66 (m, 8H). ¹³C NMR (100 MHz, CDCl₃, δ): 81.9 (*C*), 126.8 (*C*H), 127.2 (*C*H), 127.50 (*C*H), 127.53 (*C*H), 128.1 (*C*H), 128.2 (*C*H), 128.5 (*C*H), 128.9 (*C*H), 140.2 (*C*), 140.7 (*C*), 145.9 (*C*), 146.9 (*C*). EI (m/z): [M]⁺ calcd for C₃₁H₂₄O: 412.1822, found: 412.1807.

[1,1'-Biphenyl]-4-yl(4-(dimethylamino)phenyl)(phenyl)methanol (6e)



6e

The reaction was performed according to the general procedure. The reaction was conducted with aryl iodide **1a** (141.1 mg, 0.50 mmol) and diaryl ketone **5e** (58.4 mg, 0.25 mmol). The product **6e** was purified by flash column chromatography (SiO₂, hexane/CH₂Cl₂ = 50/50 then hexane/Et₂O = 80/20 to 75/25) and obtained in 69% yield (67.9 mg, 0.17 mmol) as a orange oil.

¹H NMR (396 MHz, CDCl₃, δ): 2.74 (s, 1H, OH), 2.95 (s, 6H), 6.68 (d, J = 9.1 Hz, 2H), 7.10–7.15 (m, 2H), 7.26–7.46 (m, 10H), 7.51–7.56 (m, 2H), 7.57–7.62 (m, 2H). ¹³C NMR (100 MHz, CDCl₃, δ): 40.6 (CH₃), 81.8 (C), 111.9 (CH), 126.6 (CH), 127.1 (CH), 127.2 (CH), 127.4 (CH), 128.0 (CH), 128.4 (CH), 128.9 (CH), 129.0 (CH), 134.9 (C), 139.8 (C), 140.9 (C), 146.6 (C), 147.5 (C) 149.8 (C). EI (m/z): [M]⁺ calcd for C₂₇H₂₅NO: 379.1931, found: 379.1915.

Bis(4-methoxyphenyl)(o-tolyl)methanol (6f)



The reaction was performed according to the general procedure. The reaction was conducted with aryl iodide **1b** (82.5 mg, 0.375 mmol) and diaryl ketone **5a** (61.0 mg, 0.25 mmol). The product **6f** was purified by flash column chromatography (SiO₂, hexane/CH₂Cl₂ = 50/50 then hexane/Et₂O = 90/10 to 85/15) and obtained in 72% yield (60.9 mg, 0.18 mmol) as a colorless oil.

¹H NMR (396 MHz, CDCl₃, δ): 2.15 (s, 3H), 2.90 (s, 1H, OH), 3.80 (s, 6H), 6.76 (d, J = 7.5 Hz, 1H), 6.83 (d, J = 9.1 Hz, 4H), 6.98–7.07 (m, 1H), 7.08–7.23 (m, 6H). ¹³C NMR (100 MHz, CDCl₃, δ): 22.3 (CH₃), 55.4 (CH₃), 82.7 (C), 113.3 (CH), 125.0 (CH), 127.8 (CH), 129.0 (CH), 129.5 (CH), 132.6 (CH), 138.0 (C), 139.3 (C), 145.0 (C), 158.6 (C). EI (m/z): [M]⁺ calcd for C₂₂H₂₂O₃: 334.1564, found: 334.1557.

(2-Isopropylphenyl)bis(4-methoxyphenyl)methanol (6g)



The reaction was performed according to the general procedure. The reaction was conducted with aryl iodide **1c** (91.7 mg, 0.375 mmol) and diaryl ketone **5a** (60.7 mg, 0.25 mmol). The product **6g** was purified by flash column chromatography (SiO₂, hexane/CH₂Cl₂ = 50/50 then hexane/Et₂O = 90/10 to 80/20) and obtained in 66% yield (59.8 mg, 0.16 mmol) as a colorless oil.

¹H NMR (396 MHz, CDCl₃, δ): 0.96 (d, *J* = 6.7 Hz, 6H), 2.89 (s, 1H, OH), 3.28 (sept, *J* = 6.7 Hz, 1H), 3.81 (s, 6H), 6.71 (d, *J* = 8.3 Hz, 1H), 6.84 (d, *J* = 8.7 Hz, 4H), 6.99 (t, *J* = 7.3 Hz, 1H), 7.14 (d, *J* = 9.1 Hz, 4H), 7.29 (t, *J* = 7.5 Hz, 1H), 7.38 (d, *J* = 7.9 Hz, 1H). ¹³C NMR (100 MHz, CDCl₃, δ): 24.3 (CH₃), 30.1 (CH), 55.4 (CH₃), 82.6 (C), 113.2 (CH), 124.6 (CH), 128.1 (CH), 128.2 (CH), 129.1 (CH), 129.2 (CH), 140.0 (C), 144.0 (C), 149.4 (C), 158.6 (C). EI (m/z): [M]⁺ calcd for C₂₄H₂₆O₃: 362.1877, found: 362.1871.

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