

Supporting Information

Exploring Efficient and Air-Stable d² Re(V) Alkylidyne Catalysts: Toward Room Temperature Alkyne Metathesis

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1. X-ray Crystallography Studies

Intensity data of **12b** 2CHCl_3 , **14** $0.5\text{C}_7\text{H}_8$ (C_7H_8 = toluene) and **15** were collected on a Rigaku OD Xcalibur, Gemini Ultra, Sapphire3 diffractometer with monochromatized Cu-K α at 100 or 173 K. Diffraction data were processed using the CrysAlisPro software (version 1.171.41.93a, Rigaku Oxford Diffraction, 2021). Empirical absorption corrections were performed using spherical harmonics, implemented in SCALE3 ABSPACK scaling algorithm in the CrysAlisPro software suite.

Using Olex2,¹ the structures of **12b** 2CHCl_3 , **14** $0.5\text{C}_7\text{H}_8$ (C_7H_8 = toluene) and **15** were solved with the ShelXT² structure solution program using Intrinsic Phasing and was refined with the ShelXL³ refinement package using Least Squares minimisation. All of non-hydrogen atoms were refined anisotropically with a riding model for the hydrogen atoms except noted separately.

Further crystallographic details and structural parameters are summarized in Tables S1-S2. The crystallographic data have been deposited with the Cambridge Crystallographic Data Centre as supplementary publications, CCDC reference numbers 2360919 (**12b** 2CHCl_3), 2289258 (**14** $0.5\text{C}_7\text{H}_8$) and 2360920 (**15**). Copies of these data can be obtained free of charge from the Cambridge Crystallographic Data Centre via www.ccdc.cam.ac.uk/data_request/cif.

Table S1. Crystallographic details for **12b** 2CHCl_3 and **14** $0.5\text{C}_7\text{H}_8$.

| Compound | 12b 2CHCl_3 | 14 $0.5\text{C}_7\text{H}_8$ |
|---|---|---|
| CCDC No. | 2360919 | 2289258 |
| Empirical formula | $\text{C}_{105.5}\text{H}_{55.5}\text{Cl}_{6.5}\text{F}_{48}\text{O}_4\text{P}_4\text{Re}_2\text{Zn}$ | $\text{C}_{47.5}\text{H}_{41}\text{O}_3\text{P}_2\text{Re}$ |
| Formula weight | 3091.07 | 907.94 |
| Temperature/K | 173.00(10) | 100.00(10) |
| Crystal system | triclinic | triclinic |
| Space group | P-1 | P-1 |
| a/Å | 17.1380(3) | 14.9819(6) |
| b/Å | 18.3645(4) | 14.9880(4) |
| c/Å | 21.0670(5) | 19.2809(6) |
| $\alpha/^\circ$ | 107.093(2) | 75.285(3) |
| $\beta/^\circ$ | 92.870(2) | 68.908(3) |
| $\gamma/^\circ$ | 113.868(2) | 83.041(3) |
| Volume/Å ³ | 5686.2(2) | 3904.8(2) |
| Z | 2 | 4 |
| $\rho_{\text{calc}}/\text{g}/\text{cm}^3$ | 1.805 | 1.544 |
| μ/mm^{-1} | 7.436 | 7.188 |
| F(000) | 3006.0 | 1820.0 |
| Crystal size/mm ³ | 0.22 × 0.18 × 0.06 | 0.1 × 0.08 × 0.08 |
| Radiation | Cu K α ($\lambda = 1.54184$) | CuK α ($\lambda = 1.54184$) |
| 2 θ range for data collection/° | 4.472 to 145.422 | 6.604 to 142.106 |
| Index ranges | -21 ≤ h ≤ 15, -22 ≤ k ≤ 21, -24 ≤ l ≤ 26 | -17 ≤ h ≤ 18, -18 ≤ k ≤ 13, -21 ≤ l ≤ 23 |
| Reflections collected | 34705 | 22474 |
| Independent reflections | 21495 [$R_{\text{int}} = 0.0368$, $R_{\text{sigma}} = 0.0568$] | 14375 [$R_{\text{int}} = 0.0320$, $R_{\text{sigma}} = 0.0495$] |
| Data/restraints/parameters | 21495/953/1916 | 14375/642/1148 |
| Goodness-of-fit on F ² | 1.035 | 1.048 |
| Final R indexes [$I \geq 2\sigma(I)$] | $R_1 = 0.0537$, $wR_2 = 0.1450$ | $R_1 = 0.0355$, $wR_2 = 0.0855$ |
| Final R indexes [all data] | $R_1 = 0.0632$, $wR_2 = 0.1564$ | $R_1 = 0.0451$, $wR_2 = 0.0931$ |
| Largest diff. peak/hole / e Å ⁻³ | 2.96/-1.62 | 1.68/-1.07 |

Table S2. Crystallographic details for **15**.

| | |
|---|---|
| Compound | 15 |
| CCDC No. | 2360920 |
| Empirical formula | C ₄₆ H ₄₁ O ₂ P ₂ ReS |
| Formula weight | 905.99 |
| Temperature/K | 173.00(10) |
| Crystal system | triclinic |
| Space group | P-1 |
| a/Å | 11.0053(5) |
| b/Å | 11.4390(4) |
| c/Å | 17.8263(6) |
| α/° | 85.487(3) |
| β/° | 78.475(3) |
| γ/° | 62.411(4) |
| Volume/Å ³ | 1948.62(15) |
| Z | 2 |
| ρ _{calc} /cm ³ | 1.544 |
| μ/mm ⁻¹ | 7.667 |
| F(000) | 908.0 |
| Crystal size/mm ³ | 0.1 × 0.1 × 0.08 |
| Radiation | Cu Kα (λ = 1.54184) |
| 2θ range for data collection/° | 5.06 to 149.238 |
| Index ranges | -13 ≤ h ≤ 7, -14 ≤ k ≤ 13, -22 ≤ l ≤ 22 |
| Reflections collected | 11819 |
| Independent reflections | 7585 [R _{int} = 0.0159, R _{sigma} = 0.0208] |
| Data/restraints/parameters | 7585/0/471 |
| Goodness-of-fit on F ² | 1.050 |
| Final R indexes [I >= 2σ (I)] | R ₁ = 0.0159, wR ₂ = 0.0393 |
| Final R indexes [all data] | R ₁ = 0.0166, wR ₂ = 0.0397 |
| Largest diff. peak/hole / e Å ⁻³ | 0.51/-0.53 |

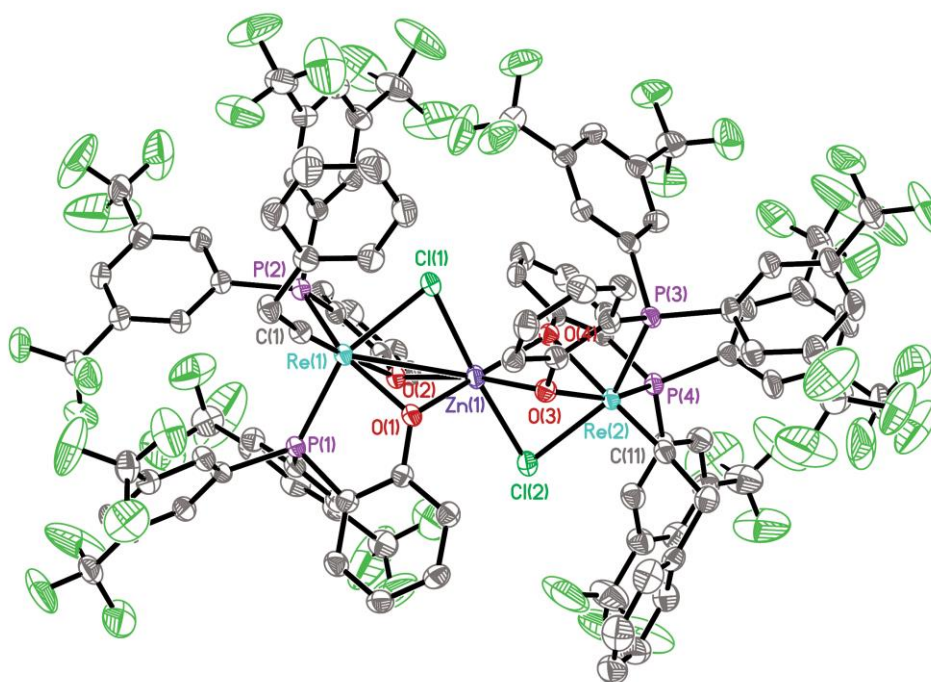


Figure S1. ORTEP diagram of the ZnCl_2 -bridged complex **12b** with thermal ellipsoids at the 40% probability level. All hydrogen atoms as well as solvent atoms are omitted for clarity.

| Selected bond lengths (Å) in 12b : | | | |
|---|------------|-------------|------------|
| Re(1)-Zn(1) | 3.0505(7) | Re(2)-Zn(1) | 3.0527(7) |
| Re(1)-C(1) | 1.763(5) | Re(2)-C(11) | 1.765(6) |
| Re(1)-Cl(1) | 2.4790(13) | Re(2)-Cl(2) | 2.5040(14) |
| Re(1)-P(1) | 2.3479(13) | Re(2)-P(3) | 2.3327(14) |
| Re(1)-P(2) | 2.3269(14) | Re(2)-P(4) | 2.3303(14) |
| Re(1)-O(1) | 2.175(4) | Re(2)-O(3) | 2.100(4) |
| Re(1)-O(2) | 2.109(4) | Re(2)-O(4) | 2.167(4) |
| Zn(1)-Cl(1) | 2.4348(16) | Zn(1)-Cl(2) | 2.4386(16) |
| Zn(1)-O(1) | 2.096(4) | Zn(1)-O(2) | 2.118(4) |
| Zn(1)-O(3) | 2.115(4) | Zn(1)-O(4) | 2.103(4) |

| Selected bond angles (deg.) in 12b : | | | |
|---|------------|-------------------|------------|
| C(1)-Re(1)-Zn(1) | 136.15(18) | C(11)-Re(2)-Zn(1) | 138.84(19) |

| | | | |
|-------------------|------------|-------------------|------------|
| C(1)-Re(1)-Cl(1) | 99.96(18) | C(11)-Re(2)-Cl(2) | 103.00(19) |
| C(1)-Re(1)-P(1) | 98.50(18) | C(11)-Re(2)-P(3) | 95.4(2) |
| C(1)-Re(1)-P(2) | 97.64(18) | C(11)-Re(2)-P(4) | 96.3(2) |
| C(1)-Re(1)-O(1) | 177.3(2) | C(11)-Re(2)-O(3) | 106.1(2) |
| C(1)-Re(1)-O(2) | 107.1(2) | C(11)-Re(2)-O(4) | 177.3(2) |
| P(1)-Re(1)-Cl(1) | 155.77(5) | P(3)-Re(2)-Cl(2) | 153.17(5) |
| P(2)-Re(1)-Cl(1) | 88.59(5) | P(4)-Re(2)-Cl(2) | 96.61(5) |
| P(2)-Re(1)-P(1) | 104.36(5) | P(4)-Re(2)-P(3) | 100.67(5) |
| O(1)-Re(1)-Cl(1) | 77.90(10) | O(3)-Re(2)-Cl(2) | 76.79(11) |
| O(1)-Re(1)-P(1) | 83.99(10) | O(3)-Re(2)-P(3) | 79.43(11) |
| O(1)-Re(1)-P(2) | 80.79(11) | O(3)-Re(2)-P(4) | 157.47(11) |
| O(2)-Re(1)-Cl(1) | 79.65(11) | O(4)-Re(2)-Cl(2) | 77.82(11) |
| O(2)-Re(1)-P(1) | 80.02(11) | O(4)-Re(2)-P(3) | 84.76(11) |
| O(2)-Re(1)-P(2) | 154.06(11) | O(4)-Re(2)-P(4) | 81.01(11) |
| O(2)-Re(1)-O(1) | 74.20(15) | O(4)-Re(2)-O(3) | 76.56(15) |
| Re(2)-Zn(1)-Re(1) | 167.10(3) | Cl(2)-Zn(1)-Cl(1) | 175.44(5) |
| Cl(1)-Zn(1)-Re(1) | 52.28(3) | Cl(1)-Zn(1)-Re(2) | 122.98(4) |
| Cl(2)-Zn(1)-Re(1) | 132.24(4) | Cl(2)-Zn(1)-Re(2) | 52.83(3) |
| O(1)-Zn(1)-Re(1) | 45.47(11) | O(3)-Zn(1)-Re(1) | 123.72(11) |
| O(1)-Zn(1)-Re(2) | 147.15(11) | O(3)-Zn(1)-Re(2) | 43.39(11) |
| O(1)-Zn(1)-Cl(1) | 80.38(12) | O(3)-Zn(1)-Cl(1) | 99.94(12) |
| O(1)-Zn(1)-Cl(2) | 102.66(12) | O(3)-Zn(1)-Cl(2) | 78.00(12) |
| O(2)-Zn(1)-Re(1) | 43.69(10) | O(4)-Zn(1)-Re(1) | 140.98(11) |
| O(2)-Zn(1)-Re(2) | 126.94(11) | O(4)-Zn(1)-Re(2) | 45.21(10) |
| O(2)-Zn(1)-Cl(1) | 80.51(11) | O(4)-Zn(1)-Cl(1) | 95.12(12) |
| O(2)-Zn(1)-Cl(2) | 103.47(12) | O(4)-Zn(1)-Cl(2) | 80.48(12) |
| O(2)-Zn(1)-O(1) | 75.65(15) | O(4)-Zn(1)-O(1) | 117.10(15) |
| O(3)-Zn(1)-O(1) | 165.26(15) | O(4)-Zn(1)-O(2) | 165.85(15) |
| O(3)-Zn(1)-O(2) | 89.82(15) | O(4)-Zn(1)-O(3) | 77.61(15) |
| Zn(1)-Cl(1)-Re(1) | 76.74(4) | Zn(1)-Cl(2)-Re(2) | 76.27(4) |
| Zn(1)-O(1)-Re(1) | 91.15(15) | Zn(1)-O(2)-Re(1) | 92.38(15) |
| Zn(1)-O(3)-Re(2) | 92.82(15) | Zn(1)-O(4)-Re(2) | 91.26(14) |

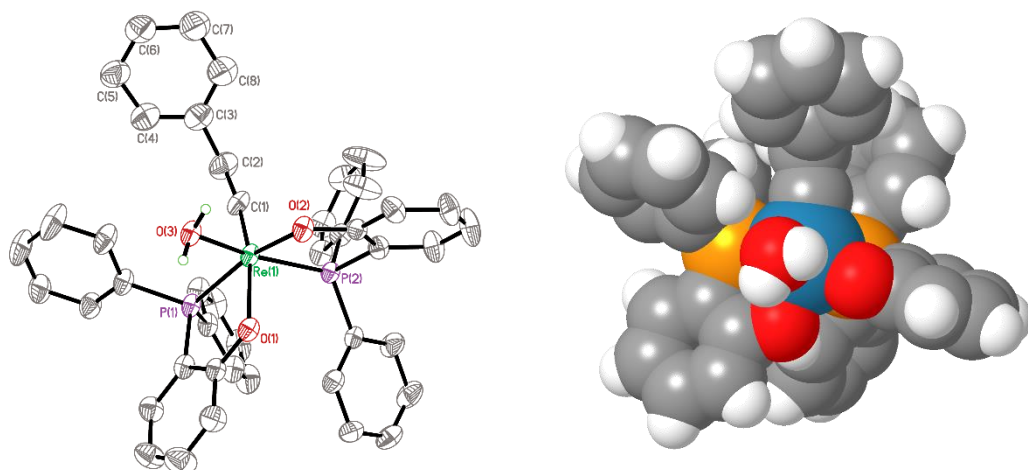


Figure S2. Left: ORTEP diagram of the aqua complex **14** with thermal ellipsoids at the 40% probability level. All hydrogen atoms except those of the water ligand as well as solvent atoms are omitted for clarity. Right: Space-filling model of the aqua complex **14**.

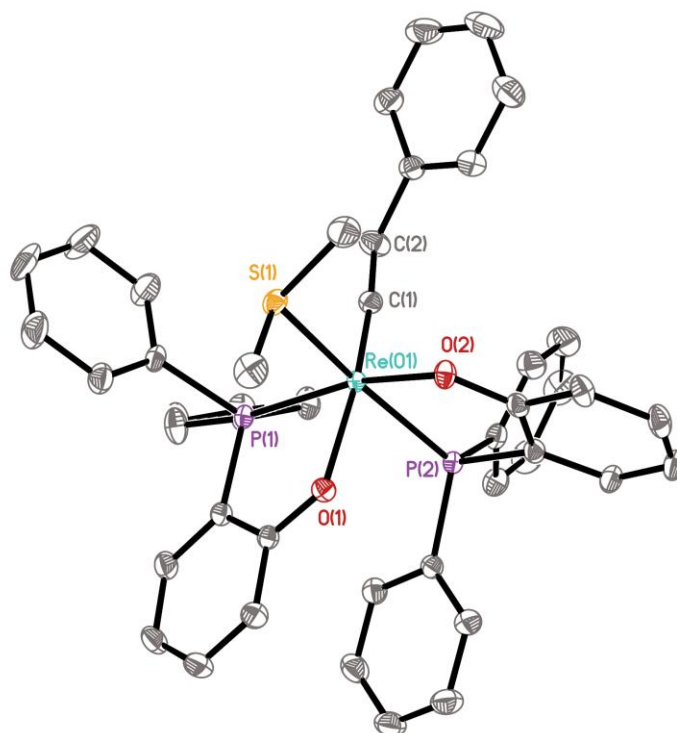


Figure S3. ORTEP diagram of the SMe_2 complex **15** with thermal ellipsoids at the 40% probability level. All hydrogen atoms as well as solvent atoms are omitted for clarity. Selected bond lengths (\AA) and angles (deg.): Re(01)-C(1) 1.7690(18), Re(01)-S(1) 2.4524(4), Re(01)-P(1) 2.3370(4), Re(01)-P(2) 2.3811(4), Re(01)-O(1) 2.1581(12), Re(01)-O(2) 2.0991(12), C(1)-C(2) 1.480(3), C(1)-Re(01)-S(1) 94.15(6), C(1)-Re(01)-P(1) 90.14(6), C(1)-Re(01)-P(2) 95.29(6), C(1)-Re(01)-O(1) 168.90(7), C(1)-Re(01)-O(2) 108.09(7), C(2)-C(1)-Re(01) 172.07(15), P(1)-Re(01)-S(1) 90.445(15), P(1)-Re(01)-P(2) 102.386(15), P(2)-Re(01)-S(1) 164.018(15), O(1)-Re(01)-S(1) 89.50(4), O(2)-Re(01)-S(1) 83.80(4).

2. *In situ* NMR Studies

In situ $^{31}\text{P}\{^1\text{H}\}$ NMR Spectra for the Reactions of Anhydrous ZnCl_2 with Re(V) Alkylidynes **10** and **11**.

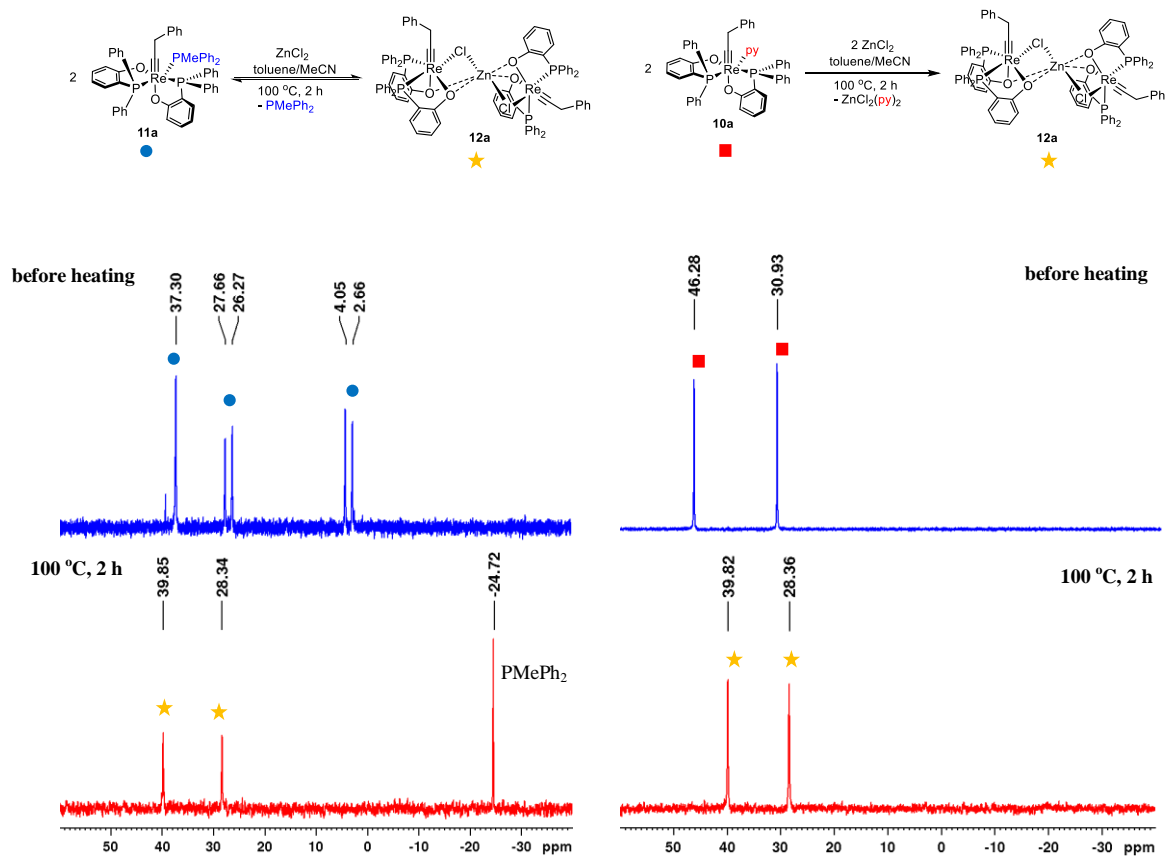


Figure S4. The stacked *in situ* $^{31}\text{P}\{^1\text{H}\}$ NMR spectra for the reactions of anhydrous ZnCl_2 with $\text{Re}(\equiv\text{CCH}_2\text{Ph})(\text{P}^{\text{Ph}}\text{O})_2(\text{PMePh}_2)$ (**11a**) (left) and $\text{Re}(\equiv\text{CCH}_2\text{Ph})(\text{P}^{\text{Ph}}\text{O})_2(\text{py})$ (**10a**) (right) in the mixed solvent of toluene and acetonitrile (v/v = 3:1). Blue circles: signals of **11a**; red squares: signals of **10a**; yellow stars: signals of **12a**. **Note:** Both two reactions produce the same product **12a**.

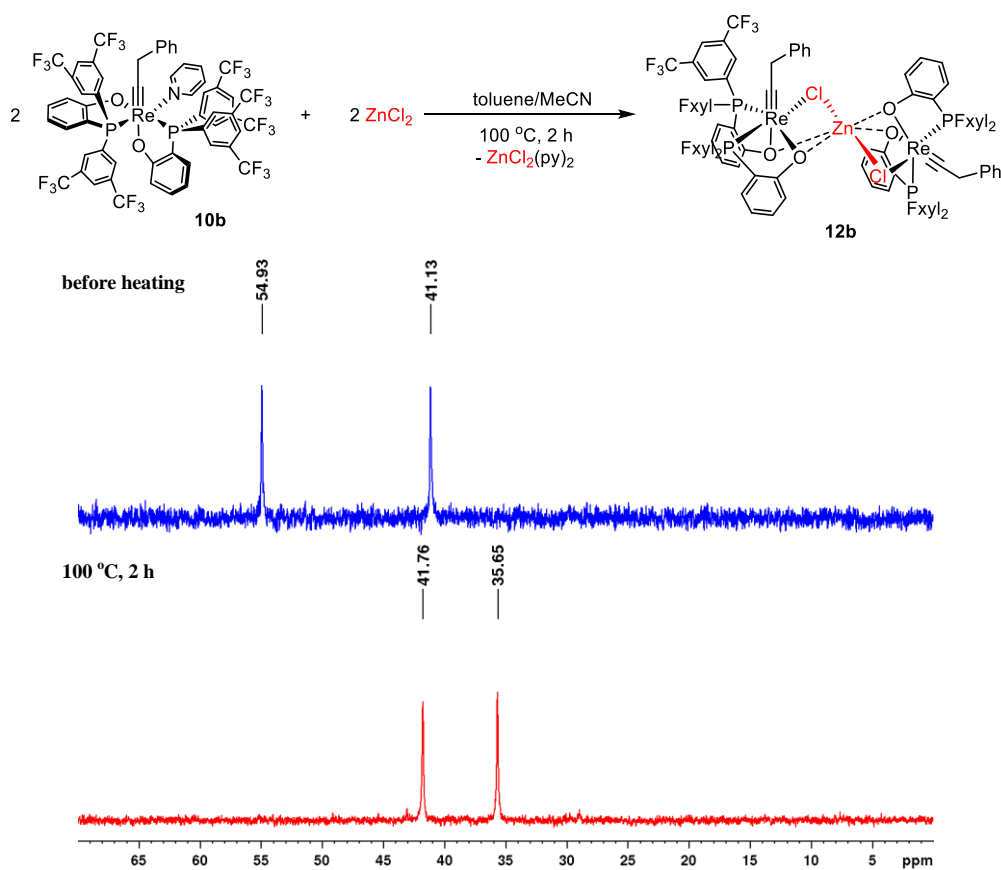


Figure S5. The stacked *in situ* $^{31}\text{P}\{^1\text{H}\}$ NMR spectra of the reactions of $\text{Re}(\equiv\text{CCH}_2\text{Ph})(\text{F}_{\text{xyl}}\text{PO})_2(\text{py})$ (**10b**) with anhydrous ZnCl_2 in the mixed solvent of toluene and acetonitrile (v/v = 3:1).

Spectroscopic Evidence for the Formation of Re(V) Alkylidyne-RuCl₂(*p*-cymene) Intermediate **13**.

Generation of **13.** To a J-Young NMR tube were added the Re(V) alkylidyne Re(≡CCH₂Ph)(^{Ph}PO)₂(PMePh₂) (**11a**) (10.4 mg, 0.01 mmol), (*p*-cymene)RuCl₂ (6.5 mg, 0.0212 mmol) and 0.5 mL of dry C₆D₆ under nitrogen. The mixture was monitored by NMR spectroscopy under the conditions specified in *Figures S7-S8*.

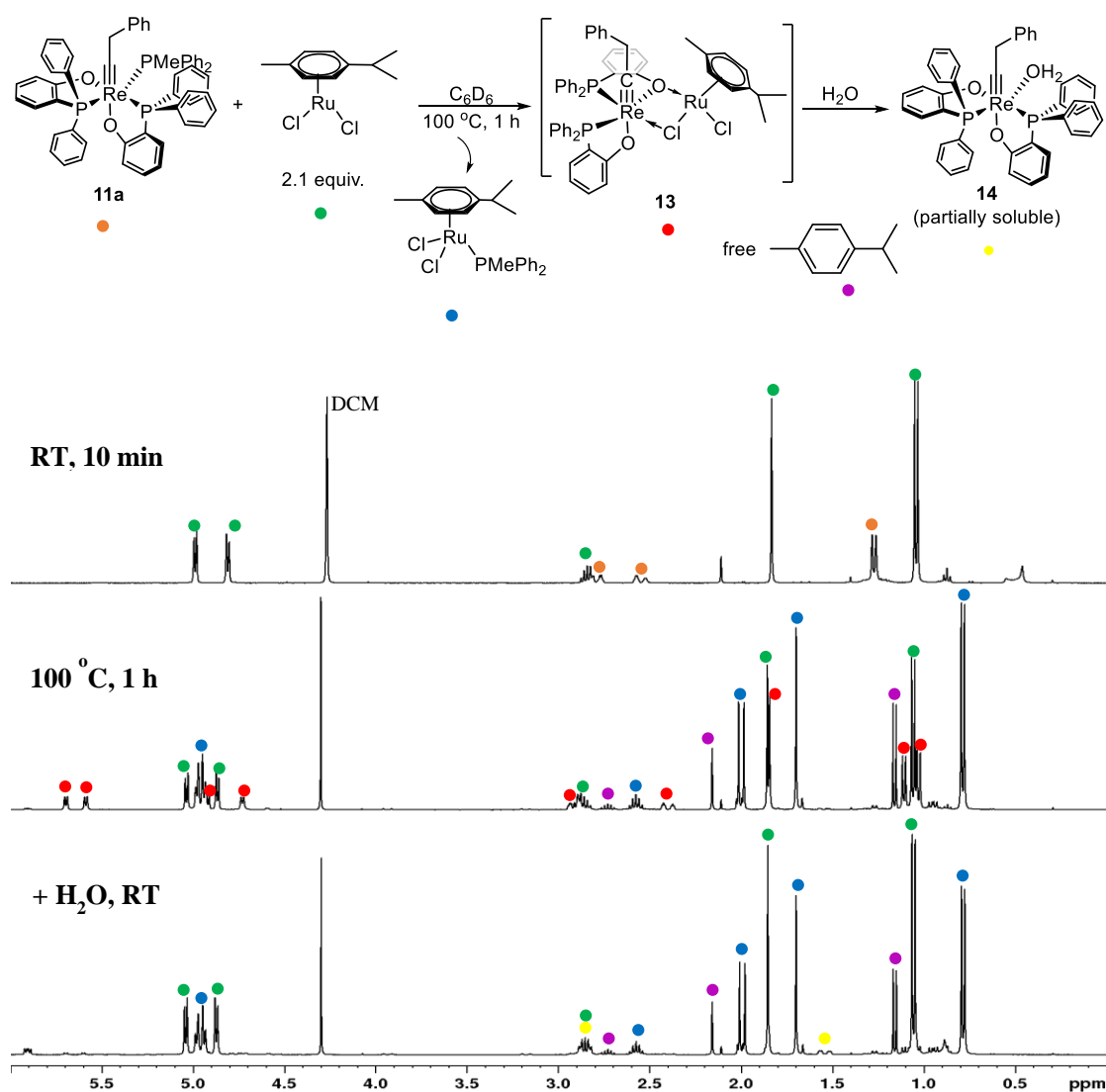


Figure S6. The stacked ¹H NMR spectra (upfield region) for the reaction of Re(≡CCH₂Ph)(^{Ph}PO)₂(PMePh₂) (**11a**) with (*p*-cymene)RuCl₂ (2.1 equiv.) in C₆D₆ at 20 °C. Top: before heating; Middle: after heating at 100 °C for 1 h; Bottom: after hydrolysis. Convention for peak assignments: orange dots, signals of the PMePh₂-coordinated complex **11a**; red dots, signals of the intermediate **13**; blue dots, signals

of (*p*-cymene)RuCl₂(PMePh₂); green dots, signals of (*p*-cymene)RuCl₂; yellow dots, signals of the aqua complex **14**.

Notes:

Before heating, the ¹H NMR spectrum only showed peaks of Re(≡CCH₂Ph)(^{*P*}PO)₂(PMePh₂) (**11a**) (orange) and (*p*-cymene)RuCl₂ (green) (in 1:2.1 molar ratio). Characteristic signals of **11a**: 2.79 ppm (dq, *J* = 19.7, 3.5 Hz, Re≡CCH₂Ph), 2.55 (dq, *J* = 19.8, 3.6 Hz, Re≡CCH₂Ph), 1.27 ppm (d, *J* = 8.4 Hz, PMePh₂). Characteristic signals of (*p*-cymene)RuCl₂: 4.98 ppm (d, *J* = 5.9 Hz, cymene ring), 4.81 ppm (d, *J* = 5.9 Hz, cymene ring), 2.84 ppm (sept, *J* = 6.9 Hz, CH(CH₃)₂), 1.83 ppm (s, CH₃), 1.04 ppm (d, *J* = 6.9 Hz, CH(CH₃)₂).

After heating at 100 °C for 1 h, Re(V) alkylidyne **11a** has been largely consumed to give the intermediate **13** (red), (*p*-cymene)RuCl₂(PMePh₂) (blue) and free cymene (purple). Characteristic ¹H signals of **13**: 5.70 ppm (d, *J* = 5.7 Hz, cymene ring), 5.59 ppm (d, *J* = 5.64 Hz, cymene ring), 4.92 ppm (d, *J* = 5.6 Hz, cymene ring), 4.73 ppm (d, *J* = 5.7 Hz, cymene ring), 2.88 ppm (sept, *J* = 7.1 Hz, CH(CH₃)₂), 2.40 ppm (dt, *J* = 19.8 Hz, 3.3 Hz, Re≡CCH₂Ph), 2.92 ppm (d = 19.2, 3.3 Hz, Re≡CCH₂Ph), 1.84 ppm (s, *Me*), 1.11 ppm (d, *J* = 6.8 Hz, CH(CH₃)₂), 1.03 ppm (d, *J* = 7.0 Hz, CH(CH₃)₂). The splitting pattern of the two methylene protons changed from “dq” to “dt”, indicating the absence of the PMePh₂ ligand. The four protons of the *p*-cymene ring as well as two methyl protons of the *i*Pr group of *p*-cymene became magnetically inequivalent, revealing the coordination of RuCl₂(*p*-cymene) moiety to the Re center.

Upon addition of water, the signals of the intermediate **13** (red) disappeared, with the appearance of the signals of the aqua complex **14** (yellow). In the meanwhile, the integrals of the peaks of (*p*-cymene)RuCl₂ increased, consistent with the reaction of the intermediate **13** with water to give **14** and (*p*-cymene)RuCl₂. It is noted that, at that moment, some yellow crystals of the poorly soluble aqua complex **14** were deposited on the wall of the NMR tube.

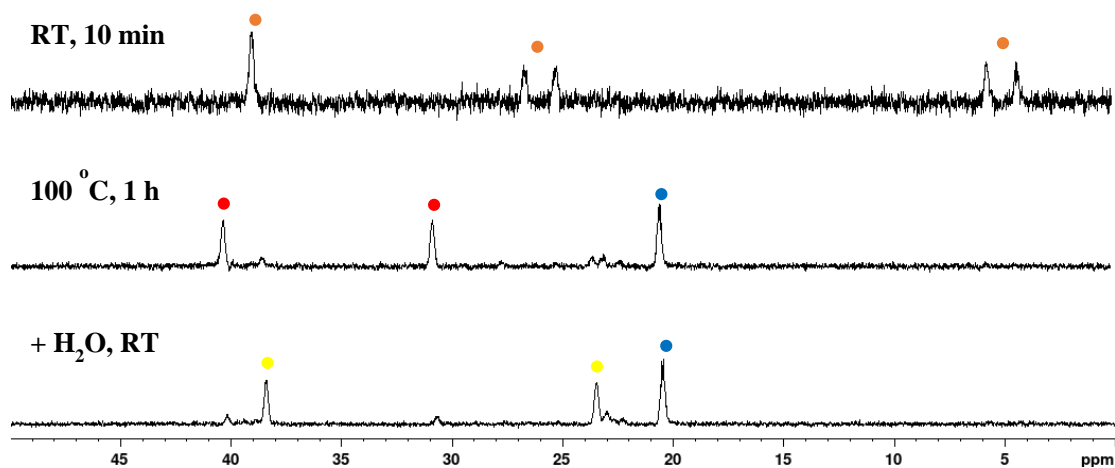
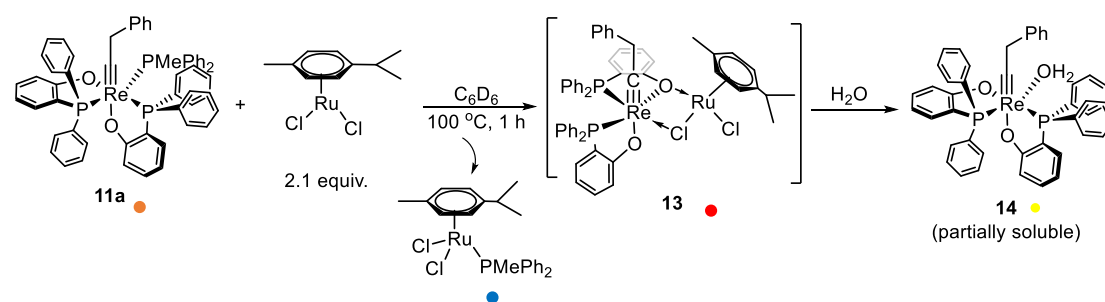


Figure S7. The stacked $^{31}\text{P}\{^1\text{H}\}$ NMR spectra for the reaction of $\text{Re}(\equiv\text{CCH}_2\text{Ph})(^{\text{Ph}}\text{PO})_2(\text{PMePh}_2)$ (**11a**) with $(p\text{-cymene})\text{RuCl}_2$ (2.1 equiv.) in C_6D_6 at $20\text{ }^\circ\text{C}$. Top: before heating; Middle: after heating at $100\text{ }^\circ\text{C}$ for 1 h; Bottom: after hydrolysis. Convention for peak assignments: orange dots, signals of the PMePh_2 -coordinated complex **11a**; red dots, signals of the intermediate **13**; blue dots, signals of $(p\text{-cymene})\text{RuCl}_2(\text{PMePh}_2)$; yellow dots, signals of the aqua complex **14**.

Notes:

Before heating, the *in situ* NMR spectrum only showed signals of the $\text{Re}(\text{V})$ alkylidyne **11a** (orange) at 38.9 ppm, 26.0 ppm, and 5.0 ppm. After heating at $100\text{ }^\circ\text{C}$ for 1 h, the signals of **11a** disappeared and the spectrum showed two singlet signals of the intermediate **13** (red) at 40.1 ppm and 30.5 ppm and one singlet signal of $(p\text{-cymene})\text{RuCl}_2(\text{PMePh}_2)$ (blue) at 20.4 ppm. Upon addition of water, $(p\text{-cymene})\text{RuCl}_2(\text{PMePh}_2)$ remains unchanged as indicated by the signal at 20.4 ppm, the intermediate **13** was largely consumed to give the aqua complex **14** which shows two singlets at 38.4 ppm and 23.4 ppm.

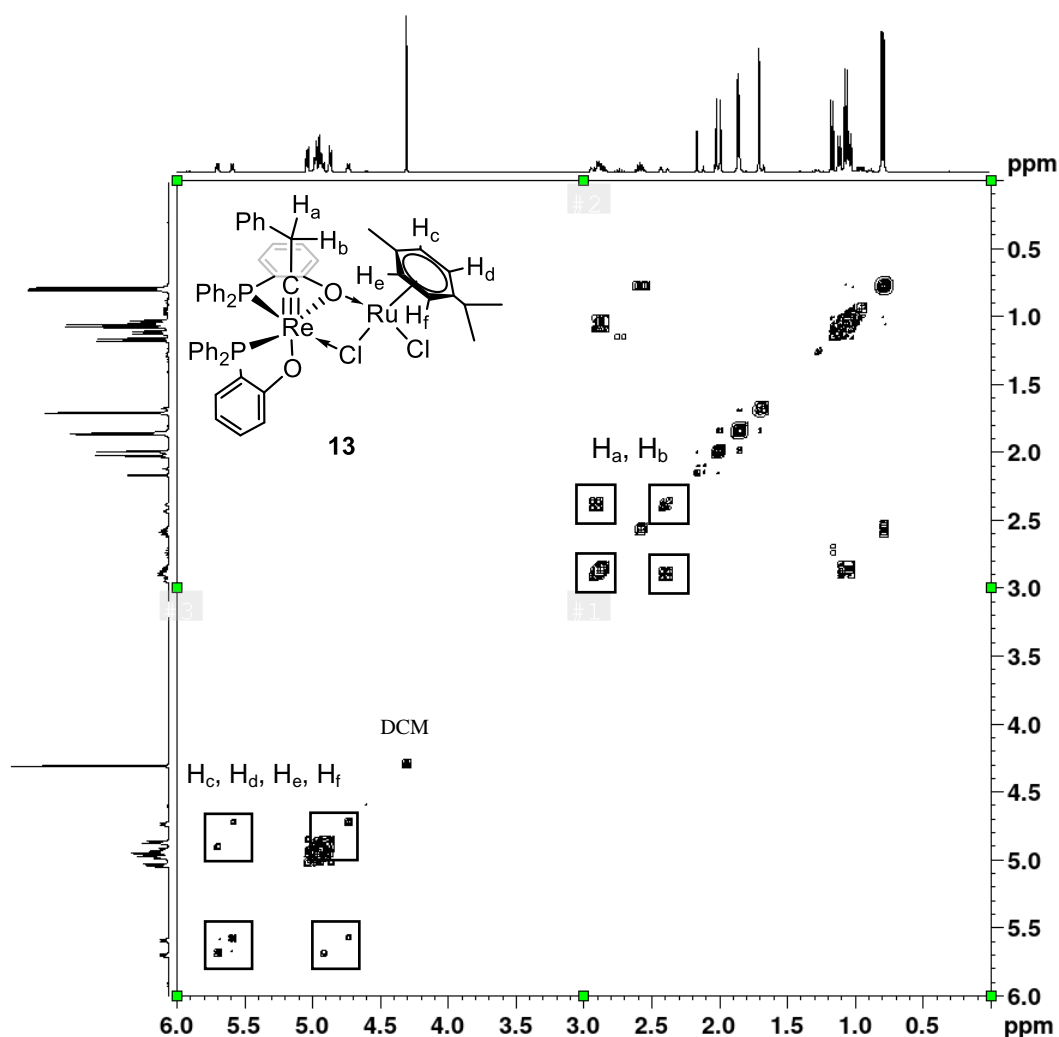


Figure S8. The *in situ* ^1H - ^1H COSY NMR spectrum for the reaction of $\text{Re}(\equiv\text{CCH}_2\text{Ph})(^{\text{Ph}}\text{PO})_2(\text{PMePh}_2)$ (**11a**) with (*p*-cymene) RuCl_2 (2.1 equiv.) in C_6D_6 at 20°C , recorded after heating the mixture at 100°C for 1 h.

Notes:

The ^1H - ^1H COSY NMR spectrum confirms that signals of the methylene protons (H_a , H_b) of $\text{Re}\equiv\text{CCH}_2\text{Ph}$ of **13** appeared at 2.40 ppm and 2.92 ppm (the signal at 2.92 ppm is partially overlapped with the signal of $\text{CH}(\text{CH}_3)_2$ of *p*-cymene); the signals of *i*Pr of the intermediate **13** appeared at 1.03 and 1.11 ppm for $\text{CH}(\text{CH}_3)_2$, and 2.86 ppm for $\text{CH}(\text{CH}_3)_2$, the signal of MeC_6H_4 -*i*Pr of **13** appeared at 1.84 ppm, the signals of the four protons (H_{c-f}) of the cymene ring appeared as doublets at 4.73, 4.92, 5.59 and 5.70 ppm.

3. Experimental Details

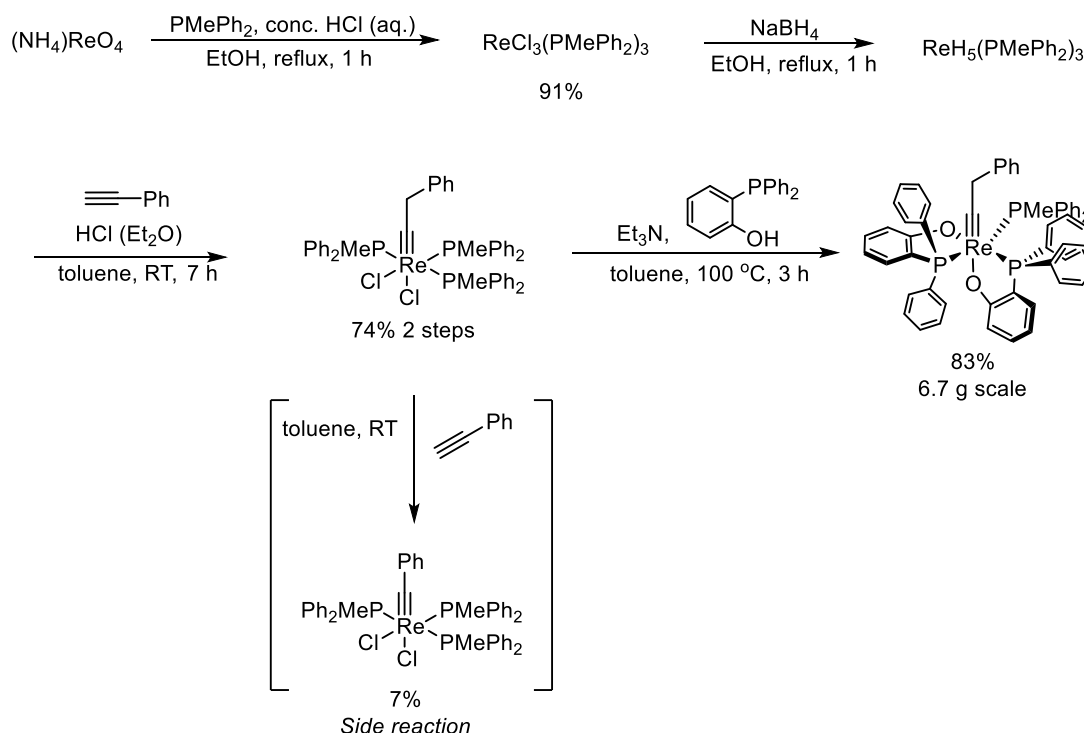
General Considerations

All manipulations were carried out under a nitrogen atmosphere using standard Schlenk techniques unless stated otherwise. Solvents were freshly distilled under nitrogen from sodium benzophenone (hexane, diethyl ether, tetrahydrofuran, toluene), or CaH₂ (dichloromethane, acetonitrile). Methanol and ethanol were bubbled with N₂ for about 20 min before use. Deuterated solvents were dried over CaH₂ (CD₂Cl₂, CDCl₃, CD₃CN) or sodium benzophenone (C₆D₆), distilled under nitrogen or vacuum transferred, degassed by three freeze-pump-thaw cycles and stored in a sealed tube with 4 Å or 3 Å (CD₃CN) molecular sieves. (*p*-cymene)RuCl₂,⁴ (*p*-cymene)OsCl₂,⁵ Ph₂P(*o*-C₆H₄-OH),⁶ **10a**,⁷ **10b**,⁸ and alkyne substrates **17-19**, **22-24**, **29-32**, **34**,^{9,10} **20**,¹¹ **21**,¹² **26**,¹³ **27**,¹⁴ **28**,¹⁵ **33**,¹⁶ **35**,¹⁷ **41**,¹⁸ **43**,⁷ **45**¹⁹ were prepared following the procedures described in the literatures. All other reagents were purchased from commercial suppliers and used without further purification. Powdered 5 Å molecular sieves (5 Å MS) was purchased from Sigma-Aldrich and activated prior to use either by heating at 150 °C under vacuum for about 24 h or heating with a heat gun (at 450 °C) under vacuum for about 5 min.

¹H, ¹³C{¹H}, and ³¹P{¹H} NMR spectra were collected on a Bruker-400 spectrometer. ¹H and ¹³C{¹H} NMR shifts are reported in ppm and are relative to the solvent signal²⁰ (¹H NMR, CDCl₃ at 7.26 ppm, CD₂Cl₂ at 5.32 ppm, CD₃CN at 1.94 ppm; ¹³C{¹H} NMR, CDCl₃ at 77.16 ppm, CD₂Cl₂ at 53.84 ppm). ³¹P{¹H} chemical shifts are relative to 85% H₃PO₄. HRMS were recorded by using a chemical ionization (CI) or electrospray ionization (ESI) mass spectrometer. FT-IR spectra were recorded on a Bruker ALPHA spectrometer with an ATR attachment, and selected peaks are reported in cm⁻¹. Elemental analysis was performed by MEDAC Ltd (Egham, UK).

Revised Synthesis of $\text{Re}(\equiv\text{CCH}_2\text{Ph})(^{\text{Ph}}\text{PO})_2(\text{PMePh}_2)$ (**11a**)

The syntheses of the Re(V) alkylidyne complexes $\text{Re}(\equiv\text{CCH}_2\text{Ph})\text{Cl}_2(\text{PMePh}_2)_3$ ²¹ and $\text{Re}(\equiv\text{CCH}_2\text{Ph})(^{\text{Ph}}\text{PO})_2(\text{PMePh}_2)$ (**11a**)⁸ have been reported previously. Herein, we reported an optimized synthetic protocol for the large scale preparation of these compounds.



mer-ReCl₃(PMePh₂)₃.²² To a 250 mL 3-neck flask equipped with a gas inlet, a condenser and a stir bar were added $(\text{NH}_4)\text{ReO}_4$ (6.50 g, 24.2 mmol), concentrated hydrochloric acid (aq., 18.4 mL) and 100 mL of degassed ethanol. The mixture was stirred for 30 min until the solid was fully dissolved, to which was added PMePh_2 (25.00 g, 125.0 mmol). Under vigorous stirring, the mixture was refluxed for 1 hour until a yellow suspension was obtained. *Note: Normally, the powdered product will precipitate suddenly, accompanied with the formation of oily phosphine oxide. The vigorous stirring is crucial to avoid the formation of dark green chunks, which prevent the reaction to go completion.* The mixture was cooled to room temperature

and the air-stable yellow powder was collected by filtration, washed with ethanol (30 ml \times 3) and dried under vacuum. Yield: 20.07 g, 91%.

$^1\text{H NMR}$ (400 MHz, CDCl_3) δ 13.90 (d, $J = 7.7$ Hz, 8H), 9.83 (d, $J = 7.6$ Hz, 4H), 9.42 (t, $J = 7.5$ Hz, 2H), 8.86 (t, $J = 7.5$ Hz, 4H), 8.58 (t, $J = 7.6$ Hz, 8H), 7.89 (t, $J = 7.5$ Hz, 4H), -0.69 (s, 6H), -2.12 (s, 3H).²³

$\text{Re}(\equiv\text{CCH}_2\text{Ph})\text{Cl}_2(\text{PMePh}_2)_3$. To a 250 mL 3-neck flask equipped with a gas inlet, a condenser and a stir bar were added $\text{ReCl}_3(\text{PMePh}_2)_3$ (9.5 g, 10.6 mmol), NaBH_4 (5.25 g, 140 mmol) and 130 mL of degassed ethanol. The mixture was vigorously stirred at room temperature for 20 min and then heated to reflux for about 1 hour until gas evolution ceased. After cooling to room temperature, the solvent was completely evaporated under vacuum to give a pale red solid. The red $\text{ReH}_5(\text{PMePh}_2)_3$ was extracted with toluene (45 mL \times 3) and filtered off to remove the white inorganic salts. The combined filtrates were collected in a 250 mL Schlenk flask charged with a stir bar, to which was added phenylacetylene (2.94 g, 28.7 mmol, ca. 3.15 mL). At room temperature, 28.7 mL of 1 M ethereal solution of HCl was added dropwise to the stirred solution over 1 h. During the course of addition, the solution color changed gradually from red to brown, followed by gentle gas evolution. After the addition, the reaction mixture was further stirred at room temperature for 6 - 7 hours until a large amount of yellow precipitant was formed. The volume of the solution was reduced to one-half under vacuum and 120 mL of hexane was added to the solution to precipitate more solid. The solid was separated by cannula filtration, sequentially washed with hexane (60 mL \times 2), diethyl ether (60 mL \times 3) as well as methanol (16 mL \times 3) and dried under vacuum to afford the desired product $\text{Re}(\equiv\text{CCH}_2\text{Ph})\text{Cl}_2(\text{PMePh}_2)_3$ as a yellow solid. Yield: 7.42 g, 74% for the two-step synthesis from $\text{ReCl}_3(\text{PMePh}_2)_3$.

Note:

A small amount of the side product, $\text{Re}(\equiv\text{CPh})\text{Cl}_2(\text{PMePh}_2)_3$, was generated from alkyne metathesis of $\text{Re}(\equiv\text{CCH}_2\text{Ph})\text{Cl}_2(\text{PMePh}_2)_3$ with excess phenylacetylene. Prolonged reaction time may increase the yield of the benzyldiyne side product.

$\text{Re}(\equiv\text{CPh})\text{Cl}_2(\text{PMePh}_2)_3$ could be isolated and purified by the following procedure: The toluene/hexane filtrate and Et_2O extracts were combined and evaporated to dryness to give an oily residue, which was washed with methanol (15 mL \times 3). The resulting orange solid was extracted with toluene (10 mL \times 3). The extracts were filtered by cannula into a long Schlenk tube and layered with hexane. After 2 weeks, red crystals (benzylidyne) together with orange crystals (benzyl alkylidyne) were deposited on the wall of the Schlenk tube. The red crystals were picked up by hands using a spatula. Repeating this procedure for two times gave pure $\text{Re}(\equiv\text{CPh})\text{Cl}_2(\text{PMePh}_2)_3$ as red crystals. Yield: 705 mg (7% for 2 steps).

Characterization data of $\text{Re}(\equiv\text{CCH}_2\text{Ph})\text{Cl}_2(\text{PMePh}_2)_3$:²¹

$^1\text{H NMR}$ (400 MHz, CDCl_3) δ 7.54 – 7.45 (m, 4H), 7.40 – 7.31 (m, 4H), 7.31 – 7.10 (m, 15H), 7.10 – 6.94 (m, 12H), 2.68 (q, $J = 3.6$ Hz, 2H), 2.14 (t, $J = 4.0$ Hz, 6H), 1.80 (d, $J = 8.8$ Hz, 3H).

$^{31}\text{P}\{^1\text{H}\}$ NMR (162 MHz, CDCl_3) δ -7.19 (t, $J = 11.4$ Hz, 1P), -9.59 (d, $J = 11.5$ Hz, 2P).

Characterization data of $\text{Re}(\equiv\text{CPh})\text{Cl}_2(\text{PMePh}_2)_3$:²¹

$^1\text{H NMR}$ (400 MHz, CDCl_3) δ 7.45 – 6.85 (m, 35H), 2.20 (t, $J = 4.0$ Hz, 6H), 1.81 (d, $J = 8.6$ Hz, 3H).

$^{31}\text{P}\{^1\text{H}\}$ NMR (162 MHz, CDCl_3) δ -3.41 (t, $J = 10.9$ Hz, 1 P), -11.18 (d, $J = 11.6$ Hz, 2P).

$\text{Re}(\equiv\text{CCH}_2\text{Ph})(^{\text{Ph}}\text{PO})_2(\text{PMePh}_2)$ (11a). To a 250 mL Schlenk flask equipped with a stir bar were added $\text{Re}(\equiv\text{CCH}_2\text{Ph})\text{Cl}_2(\text{PMePh}_2)_3$ (7.42 g, 7.7 mmol), (2-hydroxyphenyl)diphenylphosphine (4.50 g, 16.2 mmol), 80 mL of toluene and Et_3N (2.34 g, ca. 3.2 mL, 23.2 mmol). The mixture was stirred at 100 °C for 3 hours. After cooling to room temperature, the precipitate (triethylamine hydrochloride) was filtered off and discarded. The filtrate was evaporated under vacuum to give an orange

residue which was washed with hexane (50 mL × 2) and methanol (20 mL × 3) and dried under vacuum to afford the desired product as a yellow solid. Yield: 6.66 g, 83%.

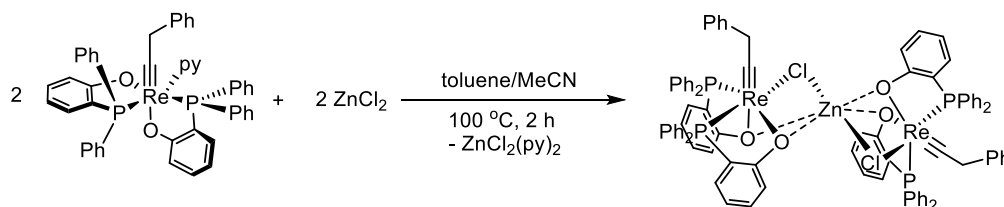
$^1\text{H NMR}$ (400.1 MHz, CDCl_3) δ 7.36-7.23 (m, 5H), 7.21-7.10 (m, 15H), 7.05-6.90 (m, 14H), 6.84-6.76 (m, 5H), 6.59-6.42 (m, 4H), 2.80 (dq, $J = 19.8, 3.2$ Hz, 1H), 2.56 (dq, $J = 19.6, 2.8$ Hz, 1H), 1.31 (d, $J = 8.5$ Hz, 3H).⁸

$^1\text{H NMR}$ (400 MHz, C_6D_6) δ 7.81 – 7.71 (m, 2H), 7.65 – 7.50 (m, 4H), 7.47 – 7.33 (m, 3H), 7.28 – 7.18 (m, 2H), 7.15 – 6.68 (m, 30H), 6.60 – 6.45 (m, 2H), 2.79 (dq, $J = 19.7, 3.5$ Hz, 1H), 2.55 (dq, $J = 19.9, 3.6$ Hz, 1H), 1.27 (dd, $J = 8.5, 1.0$ Hz, 3H).

$^{31}\text{P}\{^1\text{H}\}$ NMR (162.0 MHz, CDCl_3): δ 37.28 (br s, 1P), 27.00 (br d, $J = 225.3$ Hz, 1P), 3.39 (br d, $J = 224.0$ Hz, 1P).⁸

$^{31}\text{P}\{^1\text{H}\}$ NMR (162 MHz, C_6D_6) δ 38.92 (br s, 1P), 25.88 (dd, $J = 222.9, 5.6$ Hz, 1P), 4.99 (d, $J = 223.9$, 1P).

Synthesis of the ZnCl_2 -bridged Complex $[\text{Re}(\equiv\text{CCH}_2\text{Ph})(^{\text{Ph}}\text{PO})_2]_2(\mu\text{-ZnCl}_2)$ (**12a**)



A mixture of $\text{Re}(\equiv\text{CCH}_2\text{Ph})(^{\text{Ph}}\text{PO})_2(\text{py})$ (**10a**, 185 mg, 0.20 mmol) and anhydrous ZnCl_2 (136 mg, 1.00 mmol) in 15 mL of toluene and 5 mL of MeCN was heated at 100 °C for 2 hours. The remaining solid was filtered off, washed with 5 mL of toluene and discarded. The combined filtrate was concentrated to ca. 2 mL. To this solution was added 10 mL of hexane to precipitate a pale yellow solid, which was collected by filtration. The solid was extracted with dichloromethane (DCM) (3 mL × 2) with the addition of 1 drop of acetonitrile, and the combined extracts were concentrated to ca. 1 mL and precipitated by the addition of 10 mL of hexane. This procedure was repeated 3 times to ensure complete removal of $\text{ZnCl}_2(\text{py})_2$. The final product

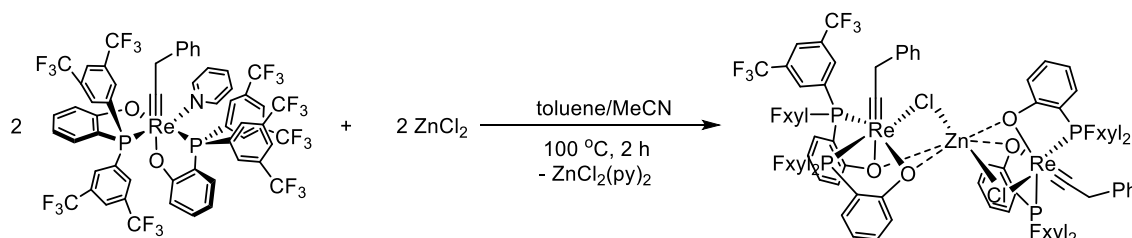
obtained was a pale yellow solid, which was dried under vacuum. Yield: 108 mg, 59%.

$^1\text{H NMR}$ (400 MHz, CD_3CN) δ 7.53 – 6.65 (m, 66 H), 2.83 (dt, $J = 19.2, 3.1$ Hz, 2H), 2.48 – 2.35 (m, 2H).

$^{31}\text{P}\{^1\text{H}\}$ NMR (162 MHz, CD_3CN) δ 37.99, 27.51.

Elem. Anal. Calcd for $\text{C}_{38}\text{H}_{70}\text{Cl}_2\text{O}_4\text{P}_4\text{Re}_2\text{Zn}$: C, 57.94; H, 3.87. Found: C, 56.82; H, 4.17.

Synthesis of the ZnCl_2 -bridged Complex $[\text{Re}(\equiv\text{CCH}_2\text{Ph})(^{\text{FxyI}}\text{PO})_2]_2(\mu\text{-ZnCl}_2)$ (**12b**)



To a NMR tube were added $\text{Re}(\equiv\text{CCH}_2\text{Ph})(^{\text{FxyI}}\text{PO})_2(\text{py})$ (11.8 mg, 8 μmol), anhydrous ZnCl_2 (5.5 mg, 40 μmol), 0.3 mL of toluene and 0.1 mL of acetonitrile. The mixture was heated at 100 $^\circ\text{C}$ for 2 hours. The reaction was monitored by $^{31}\text{P}\{^1\text{H}\}$ NMR spectroscopy (**Figure S6**). The volatiles were evaporated under vacuum, and the remaining solid was dissolved in 0.5 mL of CD_3CN . $^1\text{H NMR}$ spectrum indicated that it is a mixture of complex **12b** and $\text{ZnCl}_2(\text{py})_2$ (**Figure S12**). The solvent was evaporated to dryness and the residue was taken into 0.5 mL of CDCl_3 with the addition of 1 drop of acetonitrile. After standing at room temperature for 1 day, pale yellow crystals deposited on the wall of the NMR tube, which were collected for single-crystal X-ray diffraction analysis.

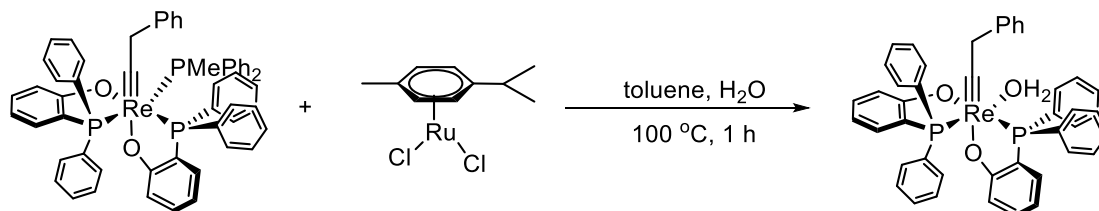
$^1\text{H NMR}$ (400 MHz, CD_3CN) δ 8.22 – 7.99 (m, 14H), 7.87 – 7.78 (m, 6H), 7.61 (s, 2H), 7.59 (s, 2H), 7.54 (s, 2H), 7.51 (s, 2H), 7.45 – 7.37 (m, 2H), 7.32 – 7.22 (m, 4H), 7.14 – 7.04 (m, 8H), 7.03 – 6.96 (m, 2H), 6.78 (t, $J = 7.4$ Hz, 2H), 6.68 (t, $J = 7.5$ Hz, 2H), 6.37 (dd, $J = 8.2, 6.1$ Hz, 2H), 3.41 (dt, $J = 19.9, 3.3$ Hz, 2H), 3.22 – 3.10 (m, 2H).

$^{19}\text{F NMR}$ (376 MHz, CD_3CN) δ -63.17, -63.47, -63.75, -63.76.

$^{31}\text{P}\{^1\text{H}\}$ NMR (162 MHz, CD_3CN) δ 41.82, 35.94

$^{31}\text{P}\{^1\text{H}\}$ NMR (162 MHz, CDCl_3) δ 39.84, 35.78

Synthesis of the Aqua Complex $\text{Re}(\equiv\text{CCH}_2\text{Ph})(^{\text{Ph}}\text{PO})_2(\text{H}_2\text{O})$ (14)



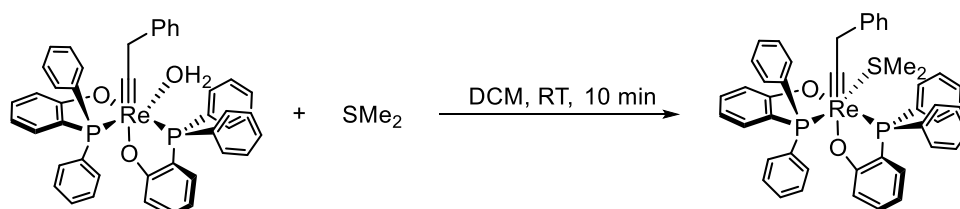
To a Schlenk flask equipped with a stir bar were added $\text{Re}(\equiv\text{CCH}_2\text{Ph})(^{\text{Ph}}\text{PO})_2(\text{PMePh}_2)$ (1.000 g, 0.96 mmol), (*p*-cymene) RuCl_2 (391 mg, 1.28 mmol) and 28 mL of dry toluene. The resulting mixture was stirred at 100 °C with vigorous stirring for 1 hour. The hot solution was filtered to another Schlenk flask to remove a trace amount of black impurities originated from (*p*-cymene) RuCl_2 . To the filtrate was added 140 μL of distilled water (7.78 mmol) and the mixture was stirred at 100 °C for another 1 h. After the reaction, the mixture was cooled to 0 °C to precipitate a yellow crystalline solid, which was collected by filtration, washed with toluene (15 mL \times 4) and dried under vacuum. Yield: 744 mg, 90%.

^1H NMR (400 MHz, CD_2Cl_2) δ 8.25 (dd, $J = 9.7, 8.0$ Hz, 2H), 7.84 (s, 2H, H_2O), 7.55 – 6.66 (m, 26H), 6.34 (dd, $J = 5.7, 3.3$ Hz, 2H), 6.14 (dd, $J = 11.5, 7.4$ Hz, 2H), 5.28 – 5.21 (m, 1H), 2.65 (dt, $J = 19.1, 3.3$ Hz, 1H), 1.37 – 1.27 (m, 1H).

$^{31}\text{P}\{^1\text{H}\}$ NMR (162 MHz, CD_2Cl_2) δ 38.18 (s), 23.40 (s).

Elem. Anal. Calcd for $\text{C}_{44}\text{H}_{37}\text{O}_3\text{P}_2\text{Re}$: C, 61.31; H, 4.33. Found: C, 61.08; H, 4.32.

Synthesis of the SMe_2 Adduct $\text{Re}(\equiv\text{CCH}_2\text{Ph})(^{\text{Ph}}\text{PO})_2(\text{SMe}_2)$ (15)



To a Schlenk flask equipped with a stir bar were added $\text{Re}(\equiv\text{CCH}_2\text{Ph})(^{\text{Ph}}\text{PO})_2(\text{OH}_2)$ (200 mg, 0.232 mmol), SMe_2 (50 μL , 0.681 mmol) and 5 mL of DCM. The resulting

mixture was stirred at room temperature for 10 mins until all solid was dissolved. The solution was concentrated to ca. 1 mL, to which was added 15 mL of hexane. The solution volume was reduced to ca. 5 mL and a pale brown solid was precipitated out. The solid was collected by filtration, washed with hexane:Et₂O = 3:1 (4 mL × 3) and dried under vacuum. Pale brown solid. Yield: 185 mg. 88%. Single crystals of **15** suitable for X-ray diffraction analysis were obtained by layering its DCM solution with hexane.

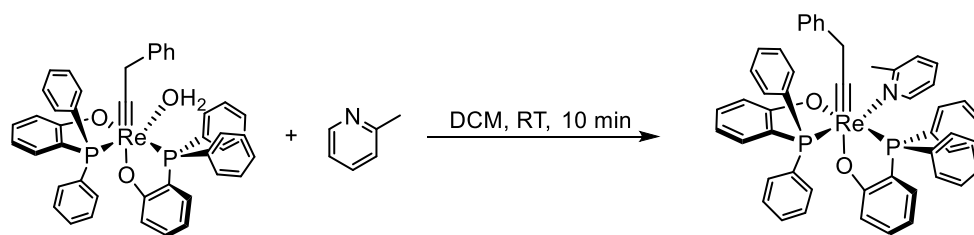
¹H NMR (400 MHz, CDCl₃) δ 7.56 – 7.40 (m, 2H), 7.34 – 7.02 (m, 22H), 7.00 – 6.80 (m, 6H), 6.55 (t, *J* = 7.2 Hz, 1H), 6.46 (t, *J* = 7.3 Hz, 1H), 6.37 (dd, *J* = 8.3, 5.5 Hz, 1H), 2.72 (dt, *J* = 19.2, 3.4 Hz, 1H), 2.30 – 2.20 (m, 1H), 2.12 (brs, 6H, *SMe*₂).

³¹P{¹H} NMR (162 MHz, CDCl₃) δ 38.81 (d, *J* = 5.1 Hz), 27.62 (d, *J* = 4.7 Hz).

¹³C{¹H} NMR (101 MHz, CDCl₃) δ 272.20 (t, *J* = 12.5 Hz), 179.64 (d, *J* = 20.9 Hz), 174.53 (d, *J* = 22.6 Hz), 142.05 (d, *J* = 56.4 Hz), 140.03 (d, *J* = 48.6 Hz), 136.53 (s), 135.97 (s), 134.91 (s), 133.84 (s), 133.64 – 131.75 (m), 130.98 (s), 130.52 (s), 129.34 (d, *J* = 2.2 Hz), 129.25 (d, *J* = 2.3 Hz), 129.19 (s), 128.84 (s), 128.59 (s), 128.18 – 127.15 (m), 126.41 (s), 121.32 (d, *J* = 6.8 Hz), 120.65 (d, *J* = 8.1 Hz), 119.61 (d, *J* = 2.2 Hz), 119.08 (d, *J* = 2.2 Hz), 118.95 (s), 118.43 (s), 116.11 (d, *J* = 6.8 Hz), 114.45 (d, *J* = 6.2 Hz), 54.72 (s), 19.87 (s).

Elem. Anal. Calcd for C₄₆H₄₁O₂P₂ReS: C, 60.98; H, 4.56. Found: C, 59.44; H, 4.85.

Synthesis of the 2-Picoline Adduct Re(≡CCH₂Ph)(^{Ph}PO)₂(^{2-Me}py) (**16**)



To a Schlenk flask equipped with a stir bar were added Re(≡CCH₂Ph)(^{Ph}PO)₂(OH₂) (200 mg, 0.232 mmol), 2-picoline (26 mg, 0.283 mmol) and 5 mL of DCM. The resulting mixture was stirred at room temperature for 10 mins until all solid was dissolved. The solution was concentrated to ca. 0.5 mL and 15 mL

of hexane was added to precipitate a bright yellow solid. The bright yellow solid was collected by filtration, washed with hexane:Et₂O= 5:1 (5mL × 3) and dried under vacuum. Yield: 176 mg. 81%.

¹H NMR (400 MHz, CDCl₃) δ 8.29 (d, *J* = 5.9 Hz, 1H), 7.41 – 7.27 (m, 5H), 7.25 – 7.15 (m, 5H), 7.16 – 6.97 (m, 15H), 6.94 (td, *J* = 7.9, 1.8 Hz, 2H), 6.91 – 6.83 (m, 3H), 6.70 (dd, *J* = 10.5, 7.6 Hz, 2H), 6.61 – 6.44 (m, 4H), 2.83 (dt, *J* = 19.1, 3.6 Hz, 1H), 2.69 (s, 3H), 2.27 – 2.14 (m, 1H).

³¹P{¹H} NMR (162 MHz, CDCl₃) δ 44.28, 24.61.

¹³C{¹H} NMR (101 MHz, CDCl₃) δ 269.67 (t, *J* = 15.6 Hz), 179.10 (d, *J* = 20.1 Hz), 174.31 (d, *J* = 22.5 Hz), 162.61 (s), 154.57 (s), 143.32 (d, *J* = 60.3 Hz), 139.66 (d, *J* = 43.8 Hz), 136.63 (s), 135.82 (s), 135.22 (s), 134.34 – 131.59 (m), 129.50 – 127.12 (m), 126.14 (s), 125.34 (s), 122.45 (d, *J* = 6.9 Hz), 121.45 (s), 120.73 (d, *J* = 8.2 Hz), 120.45 (s), 119.93 (s), 119.42 (s), 115.92 (d, *J* = 6.8 Hz), 114.80 (d, *J* = 6.1 Hz), 54.57 (s), 28.36 (s).

Rate of Alkyne Metathesis Using Re(V) Alkylidyne Catalysts

General Procedure for the Model Reaction (Homometathesis of 1-Methoxy-4-(1-propyn-1-yl)benzene) (Figure 5 & Table 1).

To a Schlenk tube equipped with a stir bar were added 750 mg of activated 5 Å molecular sieves (MS) and catalyst (0.5 - 5 mol%). The tube was capped with a rubber septum, evacuated and refilled with N₂ for three times. To the Schlenk tube were added 1-methoxy-4-(1-propyn-1-yl)benzene (73.1 mg, 75 μL, 0.50 mmol), the internal standard CH₂Ph₂ (126 mg, 126 μL, 0.75 mmol) and distilled toluene (5 mL). The mixture was stirred at specified temperatures. Samples (ca. 0.1 mL) were carefully taken at specified time by needle syringes, dilute with 0.5 mL of wet dichloromethane, filtered with a PTFE syringe filter, dried under vacuum and re-dissolved in CDCl₃ for ¹H NMR measurements.

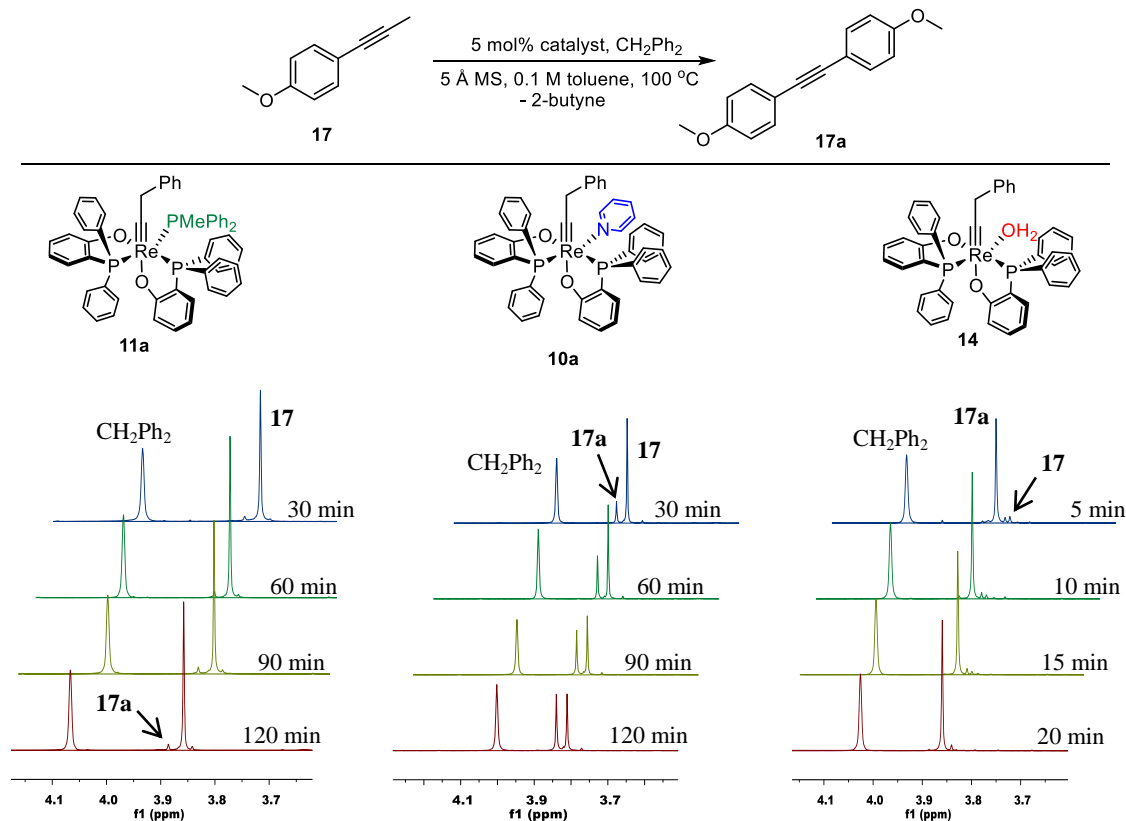


Figure S9. Stacked ¹H NMR spectra of homometathesis of 1-methoxy-4-(1-propyn-1-yl)benzene catalyzed by **11a** (left), **10a** (middle) and **14** (right) at 100 °C.

General Procedures for Metathesis Reactions (Tables 2 & 3).

General Procedure A.

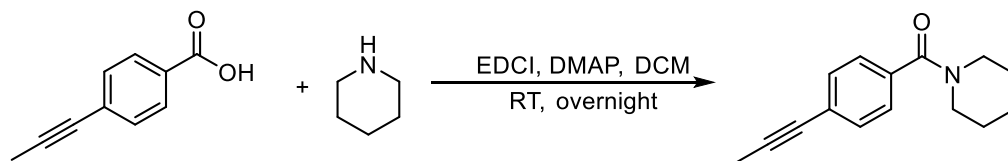
A Schlenk tube equipped with a stir bar and 5 Å molecular sieves (MS) was heated with a heat gun (at 450 °C) under vacuum for ca. 5 min to activate the MS. After cooling to room temperature, the Schlenk tube was evacuated and refilled with N₂ three times. To the Schlenk tube were added substrate(s), Re(≡CCH₂Ph)(^{Ph}PO)₂(H₂O) (**14**), and distilled toluene under N₂ flow. The mixture was stirred at specified temperatures for specified times. After the reaction, the mixture was diluted with dichloromethane in air, filtered through Celite and washed thoroughly with dichloromethane. Evaporation of the volatiles on a rotary evaporator gave a residue, which was purified by column chromatography to yield the desired product.

General Procedure B. (Using nnpurified toluene)

A Schlenk tube equipped with a stir bar and 5 Å molecular sieves (MS) was heated with a heat gun (at 450 °C) under vacuum for ca. 5 min to activate the MS. After cooling to room temperature, the Schlenk tube was evacuated and refilled with N₂ three times. To the Schlenk tube were added substrate(s), Re(≡CCH₂Ph)(^{Ph}PO)₂(H₂O) (**14**, 5 mol%), and unpurified toluene (commercial AR grade toluene stored in air, used without drying or degassing processes) under N₂ flow. The Schlenk tube was sealed and the mixture was stirred at room temperatures for specified times. After the reaction, the mixture was diluted with dichloromethane in air, filtered through Celite and washed thoroughly with dichloromethane. Evaporation of the volatiles on a rotary evaporator gave a residue, which was purified by column chromatography to yield the desired product.

Synthesis of New Substrates

Piperidin-1-yl(4-(prop-1-yn-1-yl)phenyl)methanone (25)



To a stirred solution of 4-(prop-1-yn-1-yl)benzoic acid (160 mg, 1.0 mmol) and piperidine (86 mg, 1.0 mmol) in anhydrous DCM (5 mL) were added *N*-(3-dimethylaminopropyl)-*N'*-ethylcarbodiimide hydrochloride (EDCI, 288 mg, 1.5 mmol) and 4-(dimethylamino)pyridine (12 mg, 0.10 mmol) at 0 °C. The resulting mixture was brought to room temperature and stirred overnight. After the reaction, the mixture was concentrated on a rotary evaporator to give an oily residue, which was purified by flash column chromatography (hexane:EA = 3:1) to give the desired product as white crystals. Yield: 227 mg, 99%.

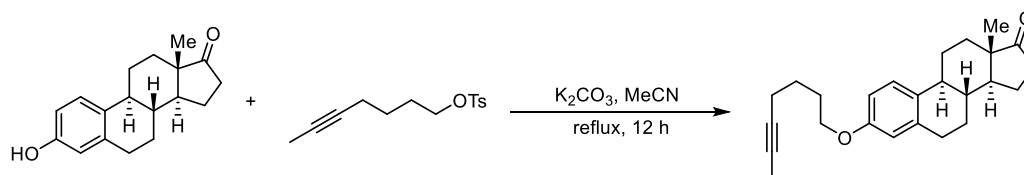
$^1\text{H NMR}$ (400 MHz, CDCl_3) δ 7.39 (d, $J = 8.1$ Hz, 2H), 7.30 (d, $J = 8.1$ Hz, 2H), 3.68 (br, 2H), 3.32 (br, 2H), 2.05 (s, 3H), 1.66 (br, 4H), 1.51 (br, 2H).

$^{13}\text{C}\{^1\text{H}\}$ NMR (101 MHz, CDCl_3) δ 169.84, 135.34, 131.48, 126.84, 125.30, 87.36, 79.21, 48.84 (br), 43.00 (br), 26.20 (br), 25.90 (br), 24.57, 4.38.

IR (ATR, cm^{-1}): 2997, 2936, 2855, 2252, 1625, 1431, 1275, 1106, 1125, 1101, 845, 761, 729

HRMS (ESI) Calcd. for $[\text{C}_{15}\text{H}_{17}\text{NNaO}]^+$ $[\text{M}+\text{Na}]^+$: 250.1202; Found: 250.1210.

Estrone hept-5-ynyl ether (39)



A mixture of estrone (270 mg, 1.0 mmol), hept-5-yn-1-yl 4-methylbenzenesulfonate (320 mg, 1.2 mmol) and K_2CO_3 (346 mg, 2.5 mmol) in 4 mL of MeCN was refluxed for 12 hours. After cooling to room temperature, the solvent was removed on a rotary evaporator. The residue was partitioned between dichloromethane (DCM) and water and the aqueous phase was extracted with DCM for 2 times. The combined DCM

extracts was washed with sat. NaCl (aq.), dried over Na₂SO₄ and evaporated to dryness. The resulting residue was purified by flash column chromatography (hexane:EA = 10:1) to give the desired product as a white solid. Yield: 238 mg, 65%.

¹H NMR (400 MHz, CDCl₃) δ 7.19 (d, *J* = 8.6 Hz, 1H), 6.71 (dd, *J* = 8.5, 2.1 Hz, 1H), 6.69 (d, *J* = 2.2 Hz, 1H), 3.95 (t, *J* = 6.3 Hz, 2H), 3.01 – 2.76 (m, 2H), 2.50 (dd, *J* = 18.8, 8.5 Hz, 1H), 2.43 – 2.32 (m, 1H), 2.28 – 1.82 (m, 9H), 1.78 (t, *J* = 2.2 Hz, 3H), 1.70 – 1.39 (m, 8H), 0.91 (s, 3H).

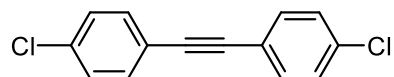
¹³C{¹H} NMR (101 MHz, CDCl₃) δ 221.06, 157.18, 137.79, 132.00, 126.40, 114.64, 112.20, 78.90, 75.97, 67.42, 50.51, 48.12, 44.09, 38.49, 35.98, 31.69, 29.76, 28.56, 26.67, 26.03, 25.65, 21.70, 18.56, 13.96, 3.58.

IR: (ATR, cm⁻¹): 2919, 2862, 1736, 1608, 1498, 1310, 1280, 1157, 817, 734.

HRMS (CI) Calcd. for [C₂₅H₃₂O₂]⁺ [M]⁺: 364.2402; Found: 364.2413.

Products of Alkyne Metathesis Reactions

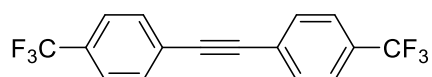
4,4'-Dichlorophenylacetylene (18a)



Synthesized following general procedure A. 4-Chlorophenyl propyne (45.2 mg, 0.30 mmol), Re(≡CCH₂Ph)(^{Ph}PO)₂(H₂O) (5.2 mg, 0.006 mmol, 2 mol%), 5 Å molecular sieves (450 mg) and toluene (3 mL). RT, 24 h. Purified by column chromatography (hexane). White solid. Yield: 31.0mg, 84%.

¹H NMR (400 MHz, CDCl₃) δ 7.45 (d, *J* = 8.5 Hz, 4H), 7.33 (d, *J* = 8.5 Hz, 4H). The NMR spectroscopic data are consistent with those reported in the literature.²⁴

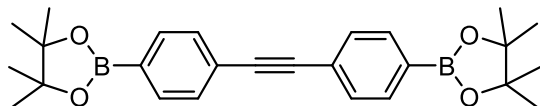
4,4'-Ditrifluoromethylphenylacetylene (19a)



Synthesized following general procedure A. 4-Trifluoromethylphenyl propyne (55.3 mg, 0.30 mmol), Re(≡CCH₂Ph)(^{Ph}PO)₂(H₂O) (5.2 mg, 0.006 mmol, 2 mol%), 5 Å molecular sieves (450 mg) and toluene (3 mL). 60 °C, 4 h. Purified by column chromatography (hexane). White solid. Yield: 38.1mg, 81%.

$^1\text{H NMR}$ (400 MHz, CDCl_3) δ 7.69 – 7.58 (m, 8H). The NMR spectroscopic data are consistent with those reported in the literature.²⁵

Diphenylacetylene-4,4'-diboronic acid bis(pinacol) ester (20a)



Synthesized following general procedure A. 4,4,5,5-tetramethyl-2-(4-(prop-1-yn-1-yl)phenyl)-1,3,2-dioxaborolane (72.6 mg, 0.30 mmol), $\text{Re}(\equiv\text{CCH}_2\text{Ph})(^{\text{Ph}}\text{PO})_2(\text{H}_2\text{O})$ (5.2 mg, 0.006 mmol, 2 mol%), 5 Å molecular sieves (450 mg) and toluene (3 mL). 60 °C, 4 h. Purified by column chromatography (Hex:EA=20:1 to DCM:MeOH=10:1). White solid. Yield: 57.0mg, 88%.

$^1\text{H NMR}$ (400 MHz, CDCl_3) δ 7.79 (d, $J = 7.7$ Hz, 4H), 7.54 (d, $J = 7.9$ Hz, 4H), 1.35 (s, 24H). The NMR spectroscopic data are consistent with those reported in the literature.²⁶

1,18-Dibromooctadec-9-yne (21a)



Synthesized following general procedure A. 11-Bromoundec-2-yne (69.4 mg, 0.30 mmol), $\text{Re}(\equiv\text{CCH}_2\text{Ph})(^{\text{Ph}}\text{PO})_2(\text{H}_2\text{O})$ (5.2 mg, 0.006 mmol, 2 mol%), 5 Å molecular sieves (450 mg) and toluene (3 mL). 60 °C, 3 h. Purified by column chromatography (Hex:EA=10:1 to DCM:MeOH=10:1). Colorless oil. Yield: 30.0mg, 49%.

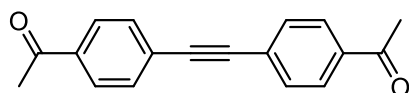
$^1\text{H NMR}$ (400 MHz, CDCl_3) δ 3.40 (t, $J = 6.9$ Hz, 4H), 2.18 – 2.09 (m, 4H), 1.85 (quintet, $J = 7.0$ Hz, 4H), 1.51 – 1.22 (m, 20H).

$^{13}\text{C}\{^1\text{H}\}$ NMR (101 MHz, CDCl_3) δ 80.33, 34.18, 32.93, 29.20, 29.08, 28.84, 28.80, 28.25, 18.85.

IR: (ATR, cm^{-1}): 2927, 2854, 1462, 1433, 1246, 723

HRMS (CI): Calcd. for $[\text{C}_{18}\text{H}_{32}\text{Br}_2]^+$ $[\text{M}]^+$: 408.0850; Found: 408.0857.

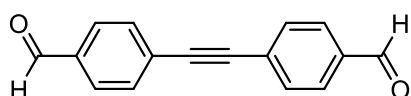
4,4'-Diacetylphenylacetylene (22a)



Synthesized following general procedure A. 4-Acetylphenyl propyne (47.5 mg, 0.30 mmol), $\text{Re}(\equiv\text{CCH}_2\text{Ph})(^{\text{Ph}}\text{PO})_2(\text{H}_2\text{O})$ (5.2 mg, 0.006 mmol, 2 mol%), 5 Å molecular sieves (450 mg) and toluene (3 mL). RT, 24 h. Purified by column chromatography (hexane:EA=10:1). White solid. Yield: 33.0mg, 84%.

$^1\text{H NMR}$ (400 MHz, CDCl_3) δ 7.96 (d, $J = 8.4$ Hz, 4H), 7.64 (d, $J = 8.4$ Hz, 4H), 2.63 (s, 6H). The NMR spectroscopic data are consistent with those reported in the literature.²⁷

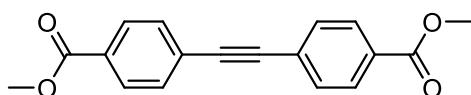
4,4'-(Ethyne-1,2-diyl)dibenzaldehyde (23a)



Synthesized following general procedure A. 4-(Prop-1-yn-1-yl)benzaldehyde (43.2 mg, 0.30 mmol), $\text{Re}(\equiv\text{CCH}_2\text{Ph})(^{\text{Ph}}\text{PO})_2(\text{H}_2\text{O})$ (5.2 mg, 0.006 mmol, 2 mol%), 5 Å molecular sieves (450 mg) and toluene (3 mL). 60 °C, 12 h. Purified by column chromatography (hexane:DCM = 10:1 to DCM). Pale yellow solid. Yield: 31.2mg, 88%.

$^1\text{H NMR}$ (400 MHz, CDCl_3) δ 10.04 (s, 2H), 7.90 (d, $J = 7.8$ Hz, 4H), 7.71 (d, $J = 7.8$ Hz, 4H). The NMR spectroscopic data are consistent with those reported in the literature.²⁵

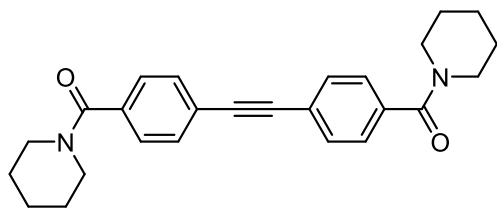
4,4'-Bis(methoxycarbonyl)diphenylacetylene (24a)



Synthesized following general procedure A. 4-Propynyl benzoic acid methyl ester (52.3 mg, 0.30 mmol), $\text{Re}(\equiv\text{CCH}_2\text{Ph})(^{\text{Ph}}\text{PO})_2(\text{H}_2\text{O})$ (5.2 mg, 0.006 mmol, 2 mol%), 5 Å molecular sieves (450 mg) and toluene (3 mL). 60 °C, 4 h. Purified by column chromatography (hexane:EA=40:1 to 10:1). White solid. Yield: 41.4mg, 94%.

$^1\text{H NMR}$ (400 MHz, CDCl_3) δ 8.04 (d, $J = 8.3$ Hz, 4H), 7.61 (d, $J = 8.4$ Hz, 4H), 3.94 (s, 6H). The NMR spectroscopic data are consistent with those reported in the literature.²⁷

(Ethyne-1,2-diylbis(4,1-phenylene))bis(piperidin-1-ylmethanone) (25a)



Synthesized following general procedure A. Piperidin-1-yl(4-(prop-1-yn-1-yl)phenyl)methanone (68.2 mg, 0.3 mmol), $\text{Re}(\equiv\text{CCH}_2\text{Ph})(^{\text{Ph}}\text{PO})_2(\text{H}_2\text{O})$ (5.2 mg, 0.006 mmol, 2 mol%), 5 Å molecular sieves (450 mg) and toluene (3 mL). 60 °C, 4 h. Purified by column chromatography (hexane:EA=40:1 to 10:1 to DCM). White solid. Yield: 50.8mg, 85%.

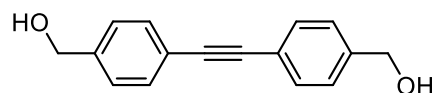
$^1\text{H NMR}$ (400 MHz, CDCl_3) δ 7.55 (d, $J = 8.3$ Hz, 4H), 7.38 (d, $J = 8.3$ Hz, 4H), 3.70 (br, 4H), 3.34 (br, 4H), 1.68 (br, 8H), 1.53 (br, 4H).

$^{13}\text{C NMR}$ (101 MHz, CDCl_3) δ 169.71, 136.49, 131.80, 127.12, 124.25, 90.05, 48.92 (br), 43.32 (br), 26.66 (br), 25.76 (br), 24.69.

IR (ATR, cm^{-1}): 2998, 2937, 2856, 1623, 1439, 1276, 1107, 1001, 919, 850, 730.

HRMS (ESI) Calcd. for $[\text{C}_{26}\text{H}_{28}\text{N}_2\text{NaO}_2]^+$ $[\text{M}+\text{Na}]^+$: 423.2043; Found: 423.2054

(Ethyne-1,2-diylbis(4,1-phenylene))dimethanol (26a)

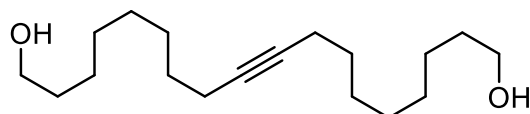


Synthesized following general procedure A. (4-(Prop-1-yn-1-yl)phenyl)methanol (43.9 mg, 0.3 mmol), $\text{Re}(\equiv\text{CCH}_2\text{Ph})(^{\text{Ph}}\text{PO})_2(\text{H}_2\text{O})$ (5.2 mg, 0.006 mmol, 2 mol%), 5 Å molecular sieves (450 mg) and toluene (3 mL). 60 °C, 4 h. Purified by column chromatography (DCM:MeOH= 50:1). White solid. Yield: 34.0 mg, 95%.

$^1\text{H NMR}$ (400 MHz, MeOD) δ 7.49 (d, $J = 8.1$ Hz, 4H), 7.37 (d, $J = 8.0$ Hz, 4H), 4.62 (s, 4H).

The spectroscopic data are in accordance with the literature.²⁸

9-Octadecyne-1,18-diol (27a)

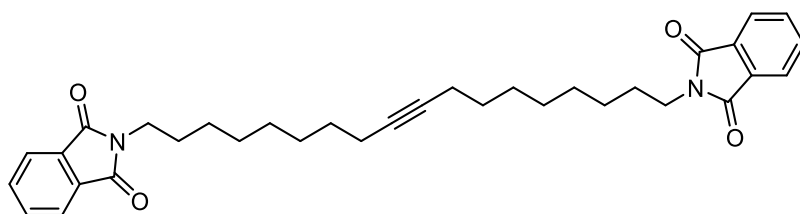


Synthesized following general procedure A. Undec-9-yn-1-ol (50.5 mg, 57 μ L, 0.30 mmol), $\text{Re}(\equiv\text{CCH}_2\text{Ph})(^{\text{Ph}}\text{PO})_2(\text{H}_2\text{O})$ (5.2 mg, 0.006 mmol, 2 mol%), 5 Å molecular sieves (450 mg) and toluene (3 mL). 60 °C, 4 h. Purified by column chromatography (DCM to DCM:MeOH=10:1). White solid. Yield: 38.6 mg, 91%.

$^1\text{H NMR}$ (400 MHz, CDCl_3) δ 3.62 (t, $J = 6.7$ Hz, 4H), 2.24 – 2.03 (m, 4H), 1.73 (br, 2H), 1.56 (dt, $J = 13.5, 6.7$ Hz, 4H), 1.50 – 1.42 (m, 4H), 1.42 – 1.24 (m, 16H).

The spectroscopic data are in accordance with the literature.⁸

2,2'-(Octadec-9-yne-1,18-diyl)bis(isoindoline-1,3-dione) (28a)



Synthesized following general procedure A. 2-(Undec-9-yn-1-yl)isoindoline-1,3-dione (89.2 mg, 0.30 mmol), $\text{Re}(\equiv\text{CCH}_2\text{Ph})(^{\text{Ph}}\text{PO})_2(\text{H}_2\text{O})$ (5.2 mg, 0.006 mmol, 2 mol%), 5 Å molecular sieves (450 mg) and toluene (3 mL). 60 °C, 3 h or RT, 16 h. Purified by column chromatography (hexane:EA = 4:1). White solid. Yield: 78.5 mg, 97%.

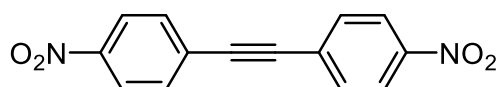
$^1\text{H NMR}$ (400 MHz, CDCl_3) δ 7.86 - 7.80 (m, 4H), 7.73 - 7.77 (m, 4H), 3.67 (t, $J = 7.2$ Hz, 4H), 2.11 (t, $J = 7.0$ Hz, 4H), 1.66 (quintet, $J = 6.9$ Hz, 4H), 1.61 - 1.20 (m, 24H).

$^{13}\text{C}\{^1\text{H}\}$ NMR (101 MHz, CDCl_3) δ 168.61, 133.96, 132.33, 123.29, 80.33, 38.20, 29.24, 29.15, 28.91, 28.73, 26.97, 18.87.

HRMS (ESI): Calcd. for $[\text{C}_{34}\text{H}_{40}\text{N}_2\text{NaO}_4]^+$ $[\text{M}+\text{Na}]^+$: 563.2880; Found: 563.2888.

The spectroscopic data are in accordance with the literature.²⁹

1,2-Bis(4-nitrophenyl)ethyne (29a)



Synthesized following general procedure A. 1-Nitro-4-propynylbenzene (48.3 mg, 0.30 mmol), $\text{Re}(\equiv\text{CCH}_2\text{Ph})(^{\text{Ph}}\text{PO})_2(\text{H}_2\text{O})$ (12.9 mg, 0.015 mmol, 5 mol%), 5 Å

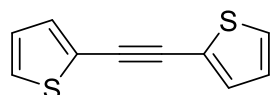
molecular sieves (450 mg) and toluene (3 mL). 60 °C, 12 h. Purified by column chromatography (hexane: DCM = 2:1). Yellow solid. Yield: 27.0 mg, 67 %.

$^1\text{H NMR}$ (400.1 MHz, CDCl_3) δ 8.26 (d, $J = 8.9$ Hz, 4H), 7.72 (d, $J = 8.9$ Hz, 4H).

$^{13}\text{C}\{^1\text{H}\}$ NMR (101 MHz, CDCl_3) δ 147.79, 132.78, 129.02, 123.94, 92.15.

The NMR spectroscopic data are consistent with those reported in the literature.³⁰

1,2-Di(thiophen-2-yl)ethyne (30a)

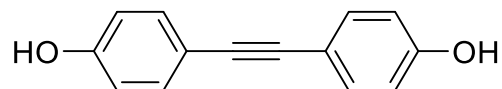


Synthesized following general procedure A. 2-(Prop-1-yn-1-yl)thiophene (36.7 mg, 0.30 mmol), $\text{Re}(\equiv\text{CCH}_2\text{Ph})(^{\text{Ph}}\text{PO})_2(\text{H}_2\text{O})$ (5.2 mg, 0.006 mmol, 2 mol%), 5 Å molecular sieves (450 mg) and toluene (3 mL). 80 °C, 4 h. Purified by column chromatography (hexane: DCM= 100:1). White solid. Yield: 24.8 mg, 87 %.

$^1\text{H NMR}$ (400.1 MHz, CDCl_3) δ 7.37 – 7.25 (m, 4H), 7.01 (dd, $J = 5.1, 3.7$ Hz, 2H).

The NMR spectroscopic data are consistent with those reported in the literature.²⁵

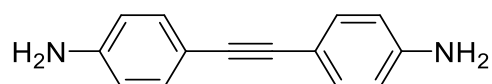
4,4'-Dihydroxyphenylacetylene (31a)



Synthesized following general procedure A. 4-Hydroxyphenyl propyne (40.0 mg, 0.30 mmol), $\text{Re}(\equiv\text{CCH}_2\text{Ph})(^{\text{Ph}}\text{PO})_2(\text{H}_2\text{O})$ (12.9 mg, 0.015 mmol, 5 mol%), 5 Å molecular sieves (450 mg) and toluene (3 mL). 100 °C, 8 h. Purified by column chromatography (hexane: EA= 2:1). White solid. Yield: 27.5 mg, 87 %.

$^1\text{H NMR}$ (400 MHz, CD_3CN) δ 7.38 (d, $J = 8.7$ Hz, 4H), 7.30 (br, 2H), 6.84 (d, $J = 8.7$ Hz, 4H). The NMR spectroscopic data are consistent with those reported in the literature.³¹

4,4'-Dianilinoacetylene (32a)

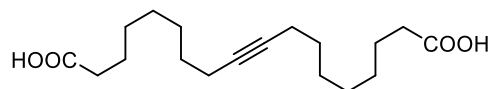


Synthesized following general procedure A. 4-Anilino propyne (39.4 mg, 0.30 mmol), $\text{Re}(\equiv\text{CCH}_2\text{Ph})(^{\text{Ph}}\text{PO})_2(\text{H}_2\text{O})$ (12.9 mg, 0.015 mmol, 5 mol%), 5 Å molecular sieves

(450 mg) and toluene (3 mL). 100 °C, 8 h. Purified by column chromatography (hexane:EA= 2:1). White solid. Yield: 25.4mg, 81%.

$^1\text{H NMR}$ (400 MHz, CDCl_3) δ 7.30 (d, $J = 8.5$ Hz, 4H), 6.62 (d, $J = 8.4$ Hz, 4H), 3.77 (s, 4H). The NMR spectroscopic data are consistent with those reported in the literature.³²

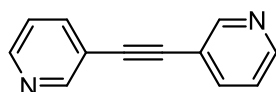
8- Hexadecyne-1,16- dicarboxylic acid (33a)



Synthesized following general procedure A. 9-Undecynoic acid (54.7 mg, 0.30 mmol), $\text{Re}(\equiv\text{CCH}_2\text{Ph})(^{\text{Ph}}\text{PO})_2(\text{H}_2\text{O})$ (12.9 mg, 0.015 mmol, 5 mol%), 5 Å molecular sieves (450 mg) and toluene (3 mL). 100 °C, 16 h. Purified by column chromatography (DCM:HCOOH = 100:1 to DCM:EA:HCOOH = 100:10:1). White solid. Yield: 32.3 mg, 69%.

$^1\text{H NMR}$ (400 MHz, CDCl_3) δ 2.35 (t, $J = 7.2$ Hz, 4H), 2.17 – 2.11 (m, 4H), 1.65 (quintet, $J = 7.1$ Hz, 4H), 1.51 – 1.27 (m, 16H). The NMR spectroscopic data are consistent with those reported in the literature.⁸

1,2-Di(pyridin-3-yl)ethyne (34a)



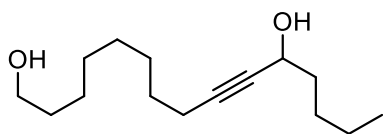
Synthesized following general procedure A. 3-(Prop-1-yn-1-yl)pyridine (35.1 mg, 0.30 mmol), $\text{Re}(\equiv\text{CCH}_2\text{Ph})(^{\text{Ph}}\text{PO})_2(\text{H}_2\text{O})$ (12.9 mg, 0.015 mmol, 5 mol%), 5 Å molecular sieves (450 mg) and toluene (3 mL). 100 °C, 3 days. Purified by column chromatography (hexane:EA:Et₃N = 100:10:1 to 100:50:1). Pale yellow solid. Yield: 4.0 mg, 15%. Note: after heating at 100 °C for 16 h, ca. 6% conversion was observed by $^1\text{H NMR}$ spectroscopy.

$^1\text{H NMR}$ (400 MHz, CDCl_3) δ 8.80 (br s, 2 H), 8.60 (br s, 2H), 7.85 (br d, $J = 7.9$ Hz, 2 H), 7.33 (dd, $J = 7.4, 5.1$ Hz, 2H).

$^{13}\text{C}\{^1\text{H}\}$ NMR (101 MHz, CDCl_3) δ 152.29, 149.09, 138.84, 123.39, 120.00, 89.35.

The NMR spectroscopic data are consistent with those reported in the literature.³³

Pentadec-9-yne-1,11-diol (36)



Synthesized following general procedure A. Undec-9-yn-1-ol (168.3 mg, 1.0 mmol), oct-2-yn-4-ol (31.6 mg, 0.25 mmol), $\text{Re}(\equiv\text{CCH}_2\text{Ph})(^{\text{Ph}}\text{PO})_2(\text{H}_2\text{O})$ (10.8 mg, 5 mol%), 5 Å molecular sieves (2.50 g) and distilled toluene (7.5 mL). 60 °C, 12 h. Purified by column chromatography (hexane:EA = 5:1 to 2:1). Yellow oil. Yield: 40.5 mg, 67%.

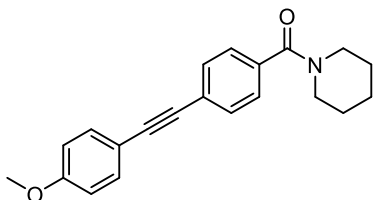
$^1\text{H NMR}$ (400 MHz, CDCl_3) δ 4.34 (tt, $J = 6.5, 1.8$ Hz, 1H), 3.63 (t, $J = 6.6$ Hz, 2H), 2.20 (td, $J = 7.0, 1.9$ Hz, 2H), 1.81 (br, 2H), 1.72 – 1.27 (m, 18H), 0.91 (t, $J = 7.1$ Hz, 3H).

$^{13}\text{C}\{^1\text{H}\}$ NMR (101 MHz, CDCl_3) δ 100.11, 85.54, 63.16, 62.87, 38.05, 32.82, 29.33, 29.07, 28.74, 28.68, 27.52, 25.73, 22.53, 18.76, 14.18.

IR (ATR, cm^{-1}): 3317, 2928, 2856, 1460, 1438, 1036, 1009.

HRMS (CI) Calcd. for $[\text{C}_{15}\text{H}_{29}\text{O}_2]^+$ $[\text{M}+\text{H}]^+$: 241.2162; Found: 241.2172.

(4-((4-Methoxyphenyl)ethynyl)phenyl)(piperidin-1-yl)methanone (37)



Synthesized following general procedure A. Piperidin-1-yl(4-(prop-1-yn-1-yl)phenyl)methanone (68.2 mg, 0.3 mmol), 1-methoxy-4-(prop-1-yn-1-yl)benzene (131.6 mg, 0.9 mmol), $\text{Re}(\equiv\text{CCH}_2\text{Ph})(^{\text{Ph}}\text{PO})_2(\text{H}_2\text{O})$ (12.9 mg, 0.015 mmol, 5 mol%), 5 Å molecular sieves (1.80 g) and toluene (5 mL). 60 °C, 4.5 h. Purified by column chromatography (hexane:EA = 10:1 to 3:1). Pale brown crystals. Yield: 66.0 mg, 69%.

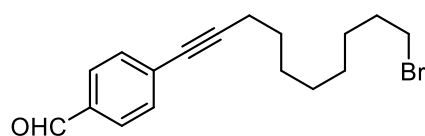
$^1\text{H NMR}$ (400 MHz, CDCl_3) δ 7.52 (d, $J = 8.1$ Hz, 2H), 7.46 (d, $J = 8.7$ Hz, 2H), 7.36 (d, $J = 8.1$ Hz, 2H), 6.87 (d, $J = 8.7$ Hz, 2H), 3.82 (s, 3H), 3.69 (br, 2H), 3.33 (br, 2H), 1.67 (br, 4H), 1.51 (br, 2H).

$^{13}\text{C}\{^1\text{H}\}$ NMR (101 MHz, CDCl_3) δ 169.83, 159.89, 135.76, 133.23, 131.48, 127.04, 124.92, 115.06, 114.13, 90.82, 87.57, 55.41, 48.83 (br), 43.35 (br), 26.64 (br), 25.68 (br), 24.65.

IR (ATR, cm^{-1}): 2999, 2934, 2854, 2214, 1624, 1600, 1568, 1515, 1434, 1279, 1247, 1174, 1137, 1106, 1026, 1000, 832.

HRMS (ESI) Calcd. for $[\text{C}_{21}\text{H}_{21}\text{NNaO}_2]^+$ $[\text{M}+\text{Na}]^+$: 342.1465; Found: 342.1468.

4-(10-Bromodec-1-yn-1-yl)benzaldehyde (**38**)



Synthesized following general procedure A. 4-(Prop-1-yn-1-yl)benzaldehyde (43.3 mg, 0.3 mmol), 11-bromoundec-2-yne (208.1 mg, 0.9 mmol), $\text{Re}(\equiv\text{CCH}_2\text{Ph})(^{\text{Ph}}\text{PO})_2(\text{H}_2\text{O})$ (12.9 mg, 0.015 mmol, 5 mol%), 5 Å molecular sieves (1.80 g) and toluene (6 mL). 60 °C, 4 h. Purified by column chromatography (hexane to hexane:EA = 100:1). Colorless oil. Yield: 59.2 mg, 61%.

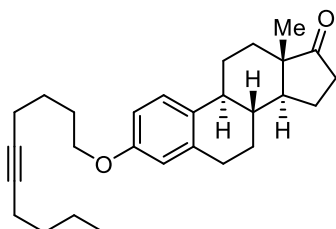
^1H NMR (400 MHz, CDCl_3) δ 9.98 (s, 1H), 7.79 (d, $J = 7.9$ Hz, 2H), 7.52 (d, $J = 8.0$ Hz, 2H), 3.40 (t, $J = 6.8$ Hz, 2H), 2.44 (t, $J = 7.0$ Hz, 2H), 1.95 – 1.77 (m, 2H), 1.68 – 1.56 (m, 2H), 1.54 – 1.28 (m, 8H).

$^{13}\text{C}\{^1\text{H}\}$ NMR (101 MHz, CDCl_3) δ 191.64, 135.07, 132.21, 130.69, 129.63, 95.32, 80.31, 34.12, 32.89, 29.05, 28.91, 28.75, 28.56, 28.23, 19.67.

IR (ATR, cm^{-1}): 2928, 2854, 2729, 2226, 1699, 1600, 1561, 1206, 1164, 828.

HRMS (CI) Calcd. for $[\text{C}_{17}\text{H}_{22}\text{BrO}]^+$ $[\text{M}+\text{H}]^+$: 321.0849; Found: 321.0858.

Estrone dec-5-ynyl ether (**40**)



Synthesized following general procedure A. The estrone derivative **39** (72.9 mg, 0.2 mmol), 5-decyne (69.1 mg, 0.5 mmol), $\text{Re}(\equiv\text{CCH}_2\text{Ph})(^{\text{Ph}}\text{PO})_2(\text{H}_2\text{O})$ (8.6 mg, 5 mol%),

5 Å molecular sieves (0.30 g) and distilled toluene (2 mL). 60 °C, 5 h. Purified by column chromatography (hexane:EA = 10:1). Pale yellow solid. Yield: 58.9 mg, 72%.

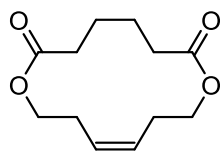
$^1\text{H NMR}$ (400 MHz, CDCl_3) δ 7.19 (d, $J = 8.6$ Hz, 1H), 6.71 (dd, $J = 8.6, 2.6$ Hz, 1H), 6.64 (d, $J = 2.5$ Hz, 1H), 3.95 (t, $J = 6.4$ Hz, 2H), 2.89 (dd, $J = 7.0, 3.0$ Hz, 2H), 2.50 (dd, $J = 18.8, 8.6$ Hz, 1H), 2.44 – 2.34 (m, 1H), 2.30 – 1.81 (m, 11H), 1.72 – 1.31 (m, 12H), 0.98 – 0.85 (m, 6H).

$^{13}\text{C}\{^1\text{H}\}$ NMR (101 MHz, CDCl_3) δ 220.88, 157.14, 137.71, 131.92, 126.33, 114.59, 112.15, 80.73, 79.64, 67.38, 50.46, 48.06, 44.04, 38.45, 35.92, 31.65, 31.29, 29.72, 28.52, 26.63, 25.99, 25.72, 22.01, 21.65, 18.56, 18.49, 13.91, 13.71.

IR (ATR, cm^{-1}): 2950, 2928, 2859, 1738, 1558, 1507, 1457.

HRMS (CI) Calcd. for $[\text{C}_{28}\text{H}_{39}\text{O}_2]^+$ $[\text{M}+\text{H}]^+$: 407.2945; Found: 407.2939.

1,8-Dioxacyclotetradec-11-yne-2,7-dione (42)



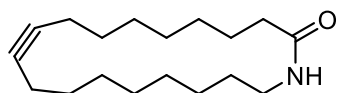
Synthesized following general procedure A. Di(pent-3-yn-1-yl) adipate (**41**) (83.5 mg, 0.30 mmol), $\text{Re}(\equiv\text{CCH}_2\text{Ph})(^{\text{Ph}}\text{PO})_2(\text{H}_2\text{O})$ (12.9 mg, 0.015 mmol), 5 Å molecular sieves (1.0 g) and toluene (60 mL). 80 °C, 8 h. Purified by column chromatography (hexane:EA = 5:1). White needle crystals. Yield: 62.0 mg, 92%.

$^1\text{H NMR}$ (400 MHz, CDCl_3) δ 4.13 (t, $J = 5.4$ Hz, 4H), 2.57 – 2.46 (m, 4H), 2.43 – 2.32 (m, 4H), 1.77 – 1.72 (m, 4H).

$^{13}\text{C}\{^1\text{H}\}$ NMR (101 MHz, CDCl_3) δ 173.28, 78.03, 62.67, 35.08, 25.12, 19.24.

HRMS (ESI) Calcd. for $[\text{C}_{12}\text{H}_{16}\text{NaO}_4]^+$ $[\text{M}+\text{Na}]^+$: 247.0941; Found: 247.0947.

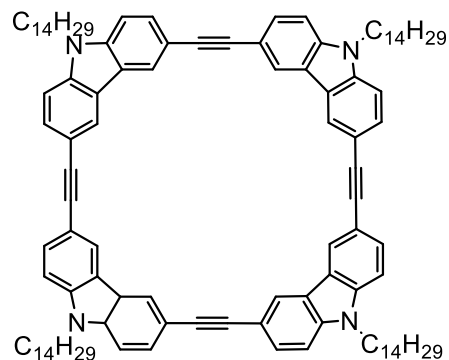
Azacyclononadec-10-yn-2-one (44)



Synthesized following general procedure A. *N*-(Undec-9-yn-1-yl)undec-9-ynamide (**43**) (66.3 mg, 0.20 mmol), $\text{Re}(\equiv\text{CCH}_2\text{Ph})(^{\text{Ph}}\text{PO})_2(\text{H}_2\text{O})$ (8.6 mg, 0.01 mmol), 5 Å molecular sieves (0.60 g) and toluene (40 mL). 100 °C, 12 h. Purified by column chromatography (hexane:EA = 2:1). White crystals. Yield: 38.0 mg, 68%.

$^1\text{H NMR}$ (400 MHz, CDCl_3) δ 5.56 (br, 1H), 3.30 (td, $J = 5.8, 5.7$ Hz, 2H), 2.25 – 2.06 (m, 6H), 1.71 – 1.57 (m, 2H), 1.55 – 1.46 (m, 2H), 1.46 – 1.36 (m, 8H), 1.38 – 1.27 (m, 10H). The NMR spectroscopic data are consistent with those reported in the literature.⁷

Macrocycle 46



Synthesized following general procedure A. 3,6-Di(prop-1-yn-1-yl)-9-tetradecylcarbazole (**45**) (83.5 mg, 0.30 mmol), $\text{Re}(\equiv\text{CCH}_2\text{Ph})(^i\text{PrPO})_2(\text{H}_2\text{O})$ (8.6 mg, 0.01 mmol), 5 Å molecular sieves (0.60 g) and toluene (10 mL). 60 °C, 14 h. Purified by column chromatography (hexane:chloroform = 1:1). White solid. Yield: 38.0 mg, 98%.

$^1\text{H NMR}$ (400 MHz, CDCl_3) δ 8.37 (d, $J = 1.0$ Hz, 2H), 7.71 (dd, $J = 8.4, 1.4$ Hz, 2H), 7.36 (d, $J = 8.5$ Hz, 2H), 4.27 (t, $J = 7.0$ Hz, 2H), 1.95 – 1.80 (m, 2H), 1.46 – 1.13 (m, 22H), 0.88 (t, $J = 6.8$ Hz, 3H).

$^{13}\text{C}\{^1\text{H}\}$ NMR (101 MHz, CDCl_3) δ 140.24, 129.38, 124.06, 122.77, 114.55, 109.00, 89.23, 43.45, 32.08, 29.85, 29.81, 29.78, 29.73, 29.68, 29.53, 29.15, 27.46, 22.85, 14.29.

The NMR spectroscopic data are consistent with those reported in the literature.¹⁵

4. NMR Spectra

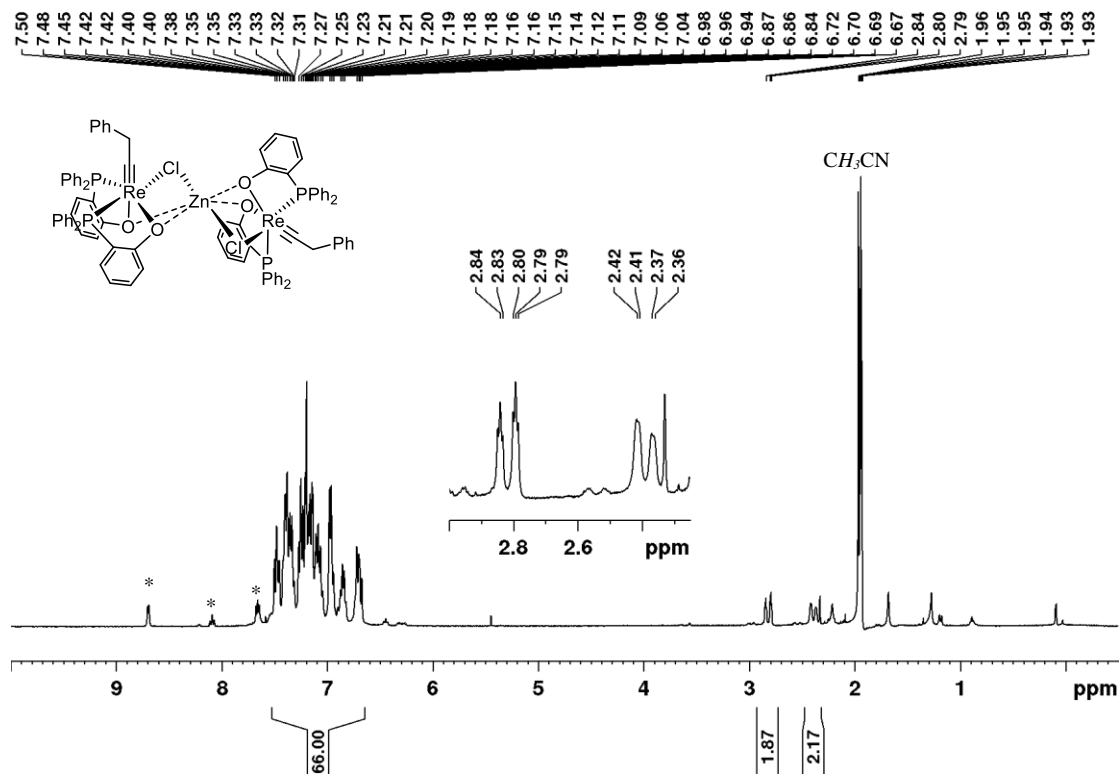


Figure S10. The ^1H NMR spectrum of $[\text{Re}(\equiv\text{CCH}_2\text{Ph})(\text{Ph}^h\text{PO})_2]_2(\mu\text{-ZnCl}_2)$ (12a) in CD_3CN at 400.1 MHz. * The signals of remaining $\text{ZnCl}_2(\text{py})_2$.

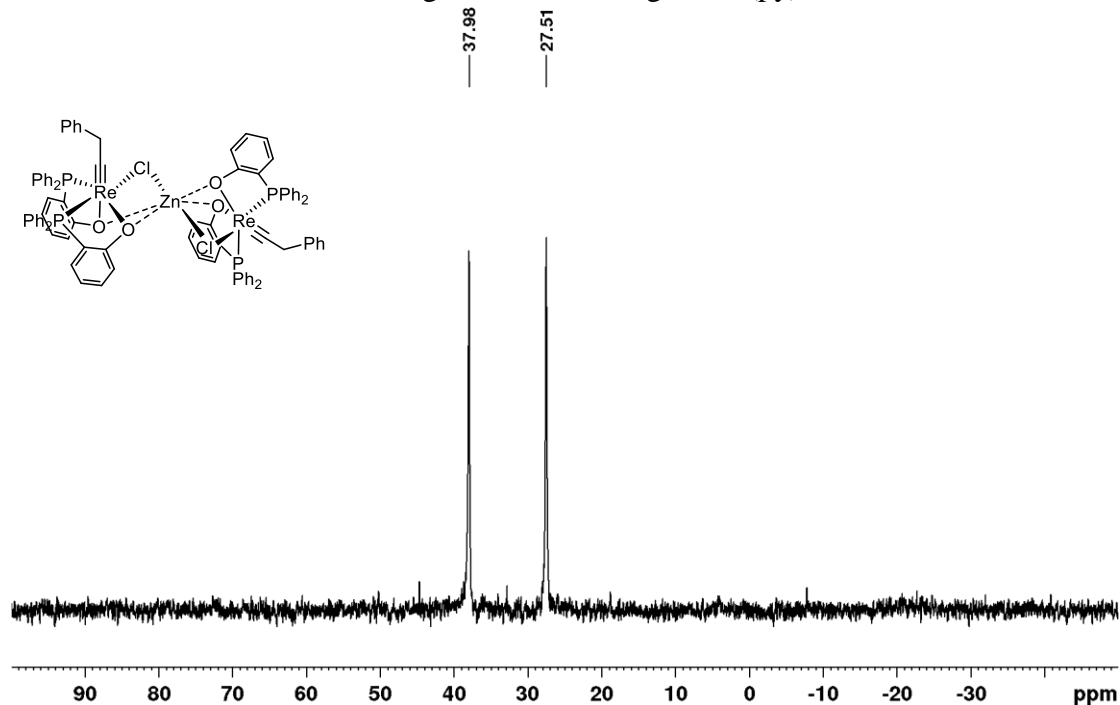


Figure S11. The $^{31}\text{P}\{^1\text{H}\}$ NMR spectrum of $[\text{Re}(\equiv\text{CCH}_2\text{Ph})(\text{Ph}^h\text{PO})_2]_2(\mu\text{-ZnCl}_2)$ (12a) in CD_3CN at 162.0 MHz.

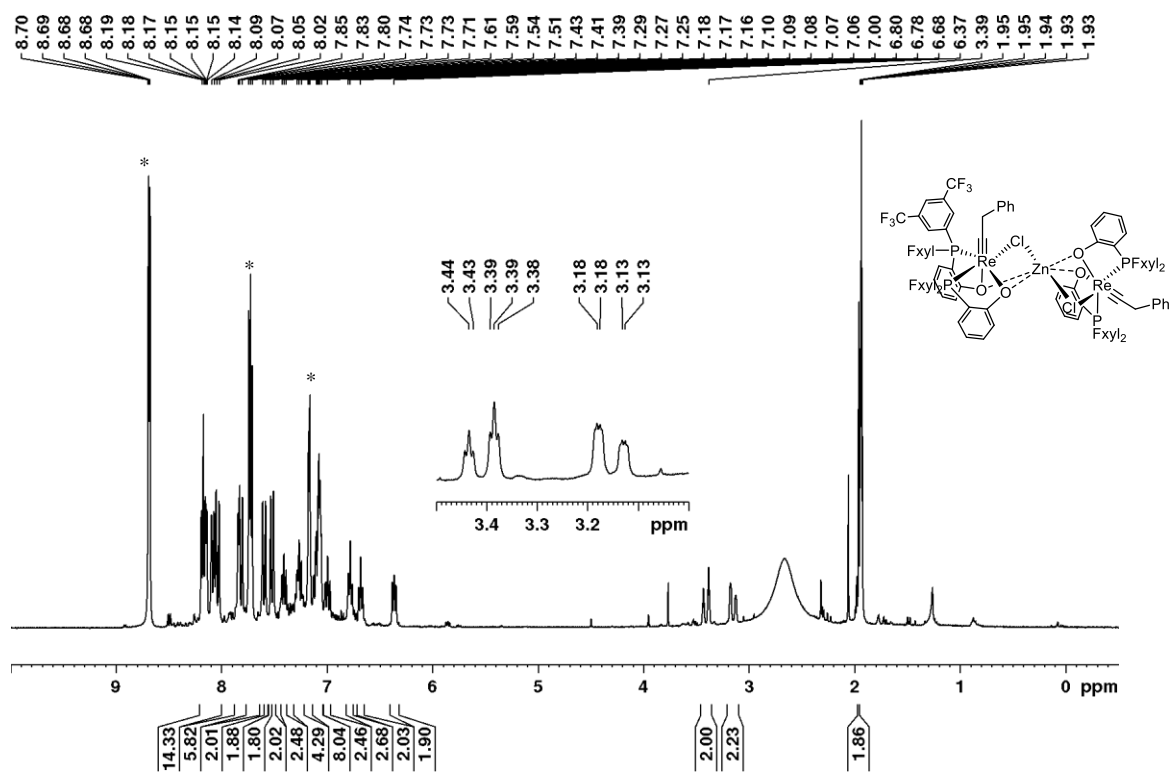


Figure S12. The ^1H NMR spectrum of $[\text{Re}(\equiv\text{CCH}_2\text{Ph})(\text{FxyP}(\text{O})_2)_2](\mu\text{-ZnCl}_2)$ (**12b**) in CD_3CN at 400.1 MHz. * The signals of remaining $\text{ZnCl}_2(\text{py})_2$.

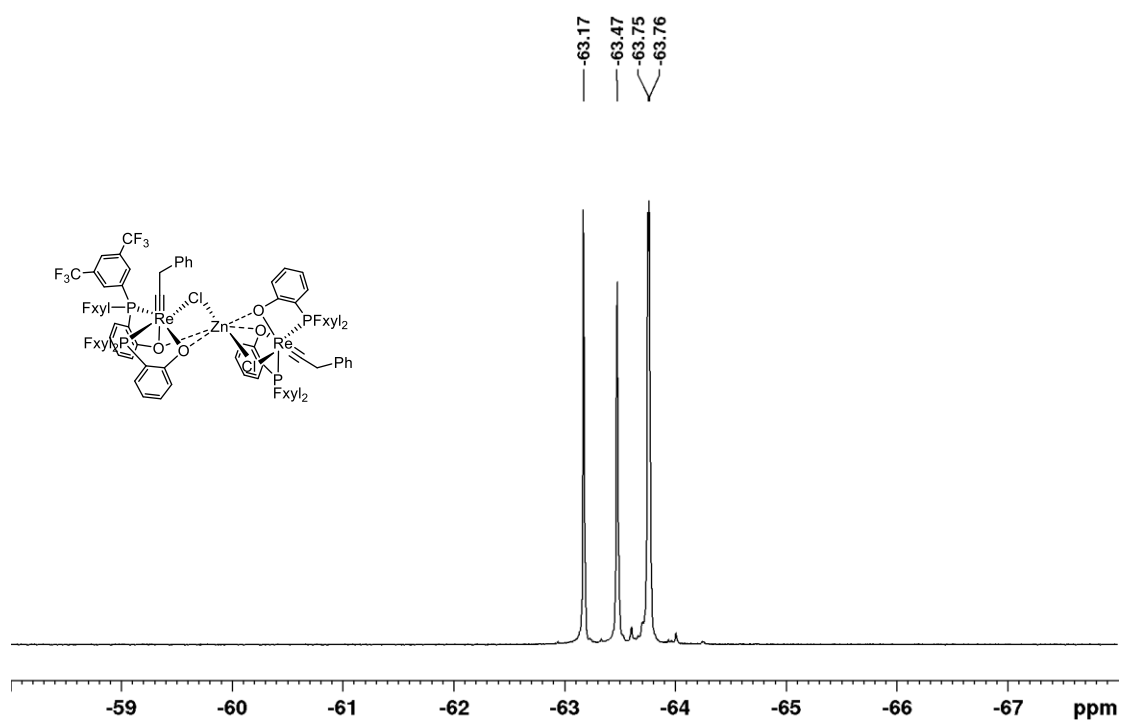


Figure S13. The $^{19}\text{F}\{^1\text{H}\}$ NMR spectrum of $[\text{Re}(\equiv\text{CCH}_2\text{Ph})(\text{FxyP}(\text{O})_2)_2](\mu\text{-ZnCl}_2)$ (**12b**) in CD_3CN at 376.5 MHz.

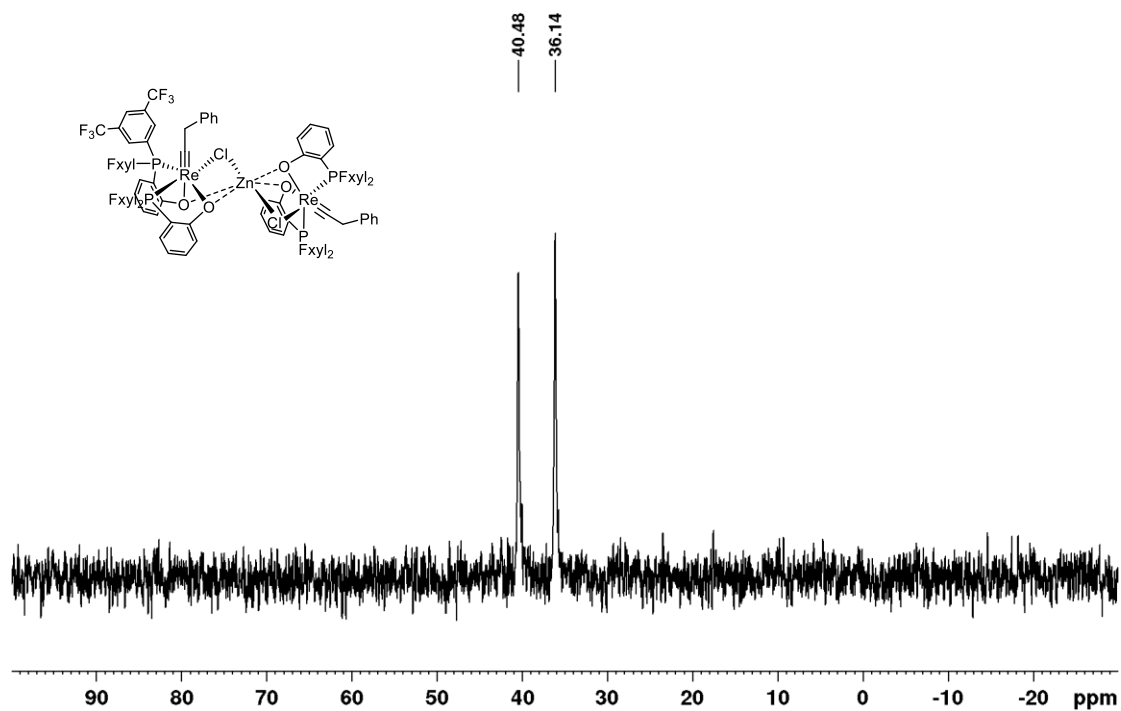


Figure S14. The $^{31}\text{P}\{^1\text{H}\}$ NMR spectrum of $[\text{Re}(\equiv\text{CCH}_2\text{Ph})(\text{Fxy}^{\text{P}}\text{O})_2]_2(\mu\text{-ZnCl}_2)$ (12b) in CDCl_3 at 162.0 MHz.

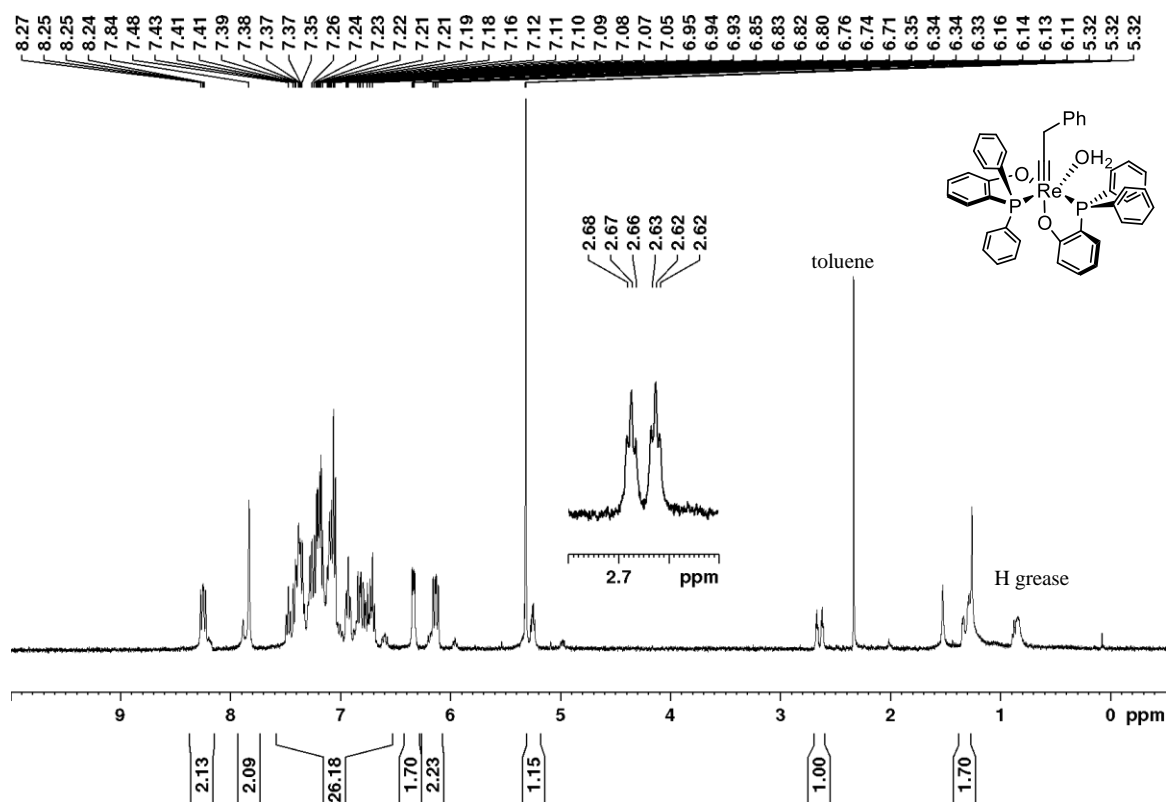


Figure S15. The ^1H NMR spectrum of $\text{Re}(\equiv\text{CCH}_2\text{Ph})(\text{Ph}^{\text{P}}\text{O})_2(\text{H}_2\text{O})$ (14) in CD_2Cl_2 at 400.1 MHz. The signal of coordinated H_2O appeared as a singlet at 7.84 ppm.

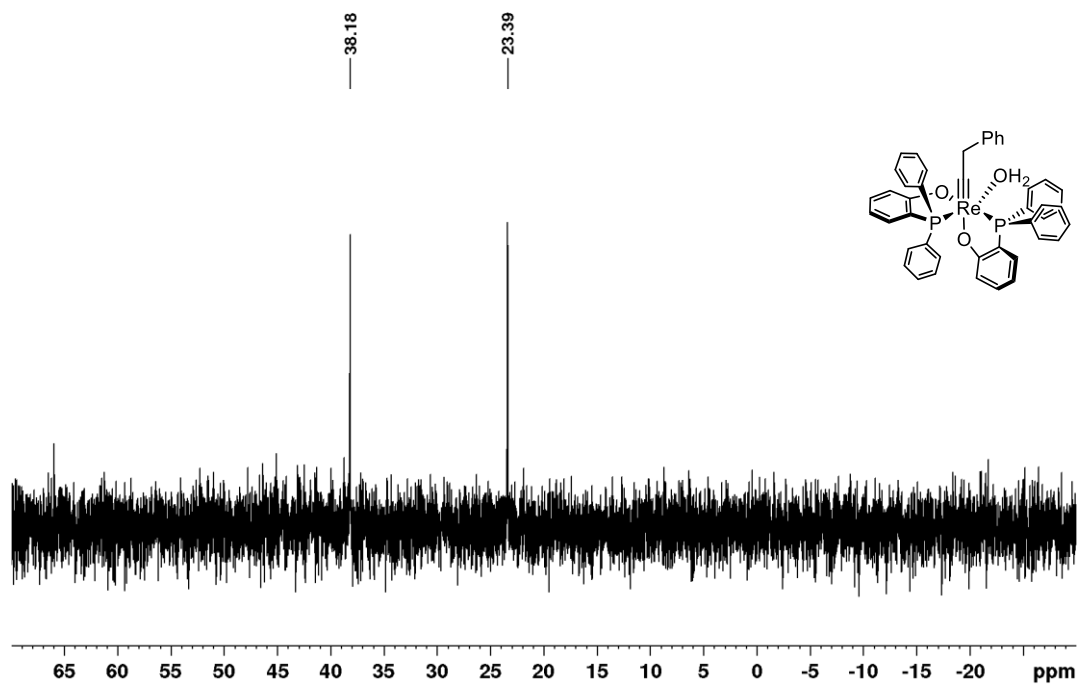


Figure S16. The $^{31}\text{P}\{^1\text{H}\}$ NMR spectrum of $\text{Re}(\equiv\text{CCH}_2\text{Ph})(^{\text{Ph}}\text{PO})_2(\text{H}_2\text{O})$ (**14**) in CD_2Cl_2 at 162.0 MHz.

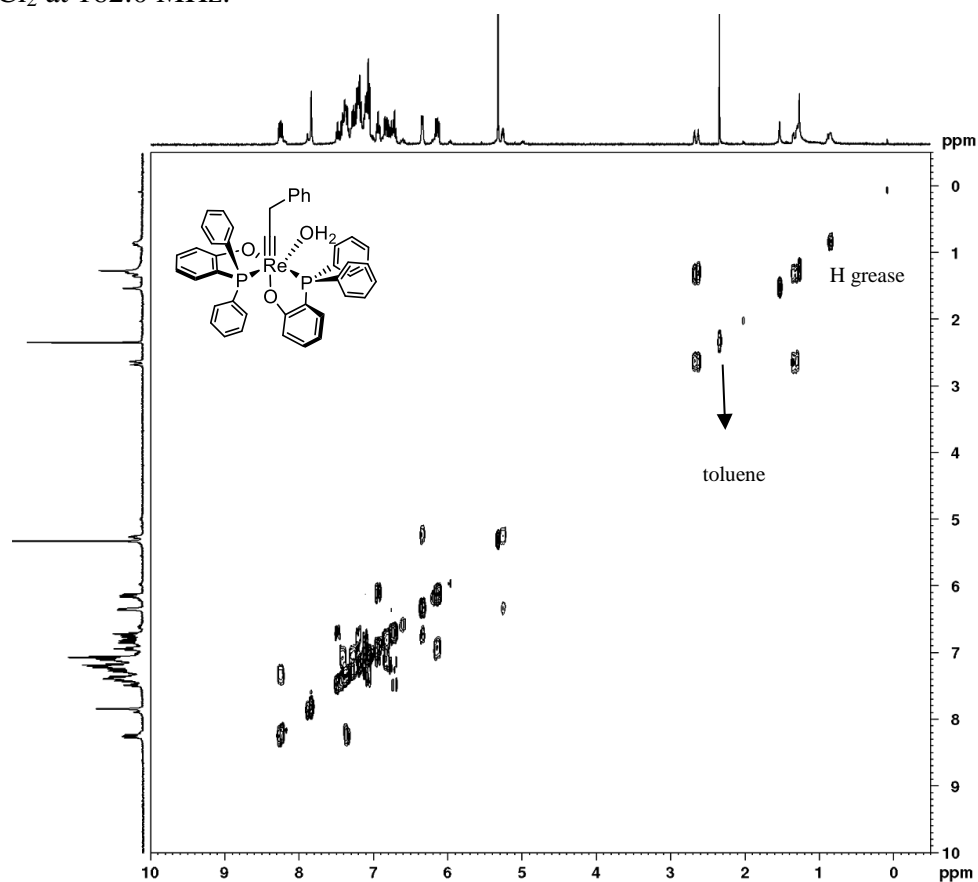


Figure S17. The COSY NMR spectrum of $\text{Re}(\equiv\text{CCH}_2\text{Ph})(^{\text{Ph}}\text{PO})_2(\text{H}_2\text{O})$ (**14**) in CD_2Cl_2 at 400.1 MHz. The COSY NMR spectrum confirms that the signals of the methylene protons of $\text{Re}\equiv\text{CCH}_2\text{Ph}$ appear at 2.65 and 1.37 ppm.

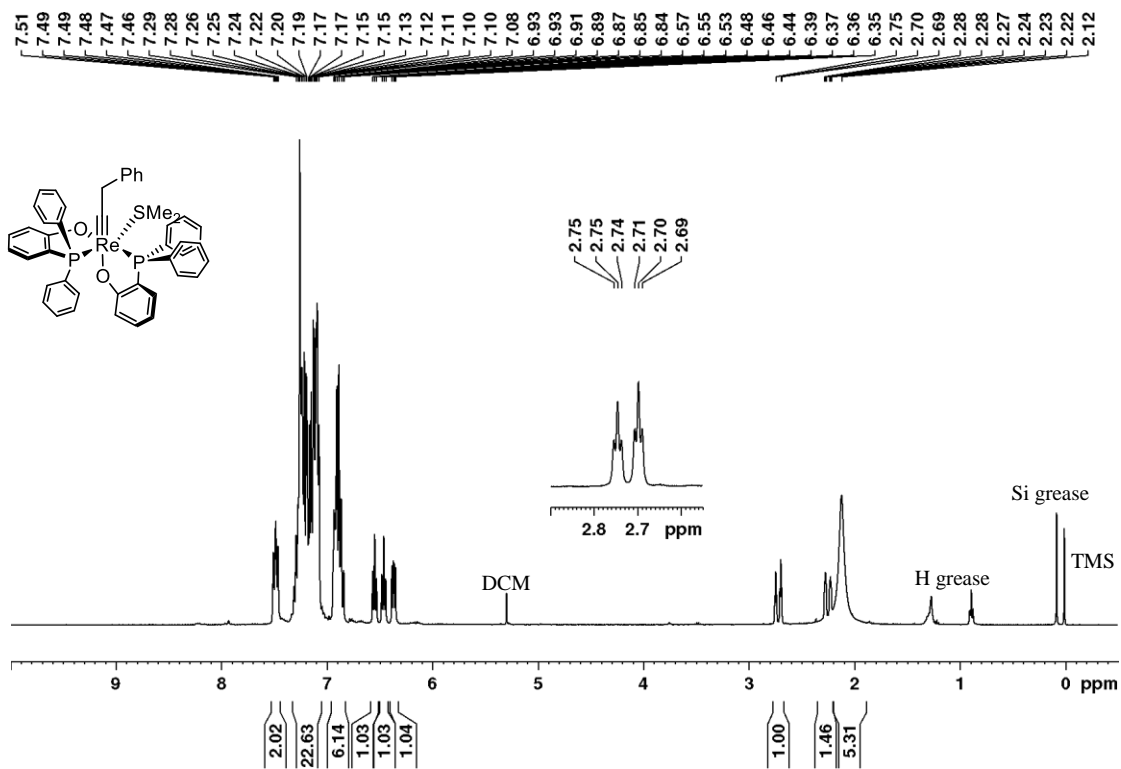


Figure S18. The ^1H NMR spectrum of $\text{Re}(\equiv\text{CCH}_2\text{Ph})(\text{PhPO})_2(\text{SMe}_2)$ (**15**) in CDCl_3 at 400.1 MHz.

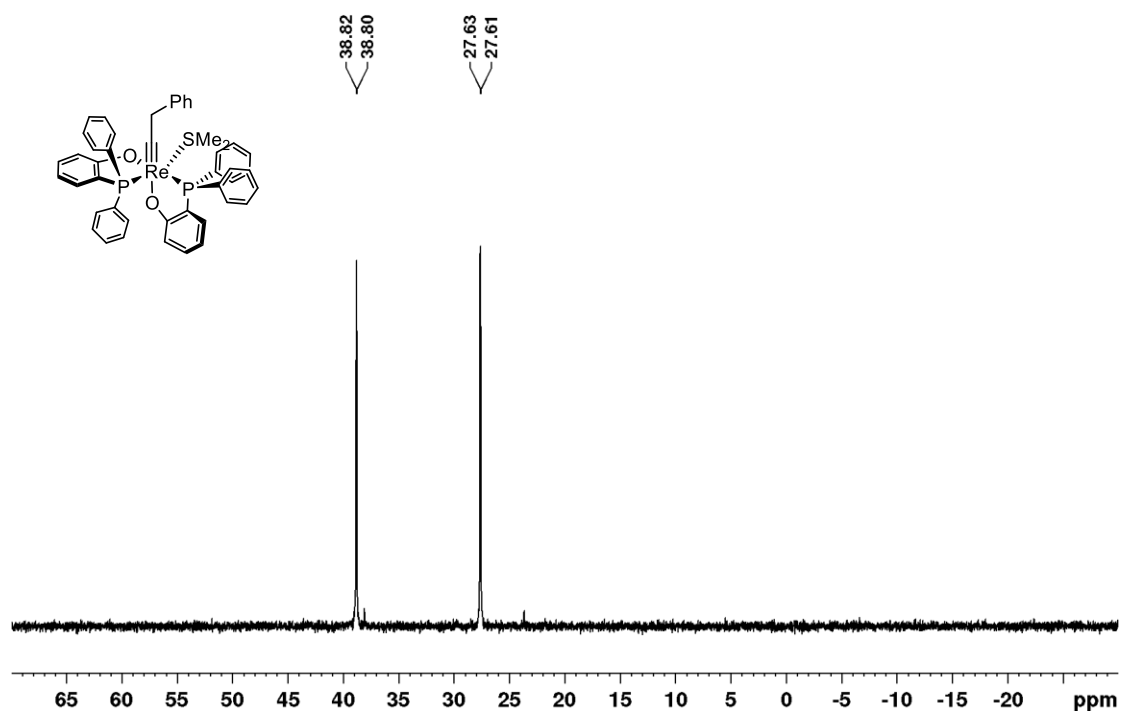


Figure S19. The $^{31}\text{P}\{^1\text{H}\}$ NMR spectrum of $\text{Re}(\equiv\text{CCH}_2\text{Ph})(\text{PhPO})_2(\text{SMe}_2)$ (**15**) in CDCl_3 at 162.0 MHz.

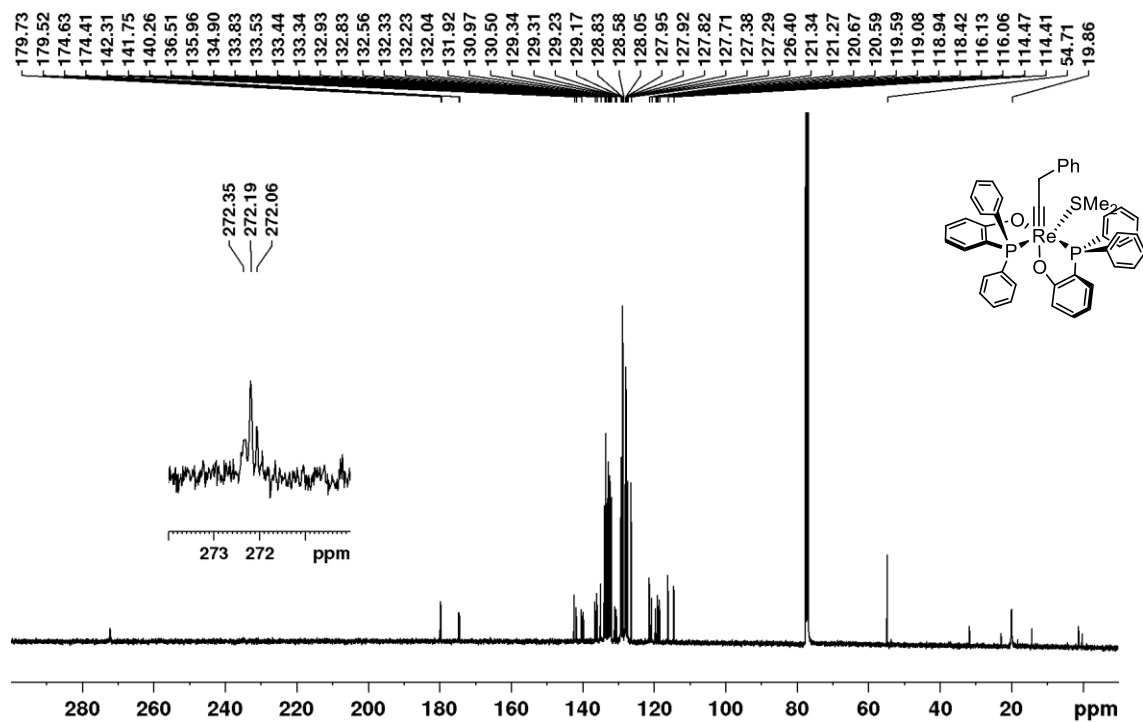


Figure S20. The $^{13}\text{C}\{^1\text{H}\}$ NMR spectrum of $\text{Re}(\equiv\text{CCH}_2\text{Ph})(\text{PhPO})_2(\text{SMe}_2)$ (**15**) in CDCl_3 at 100.6 MHz.

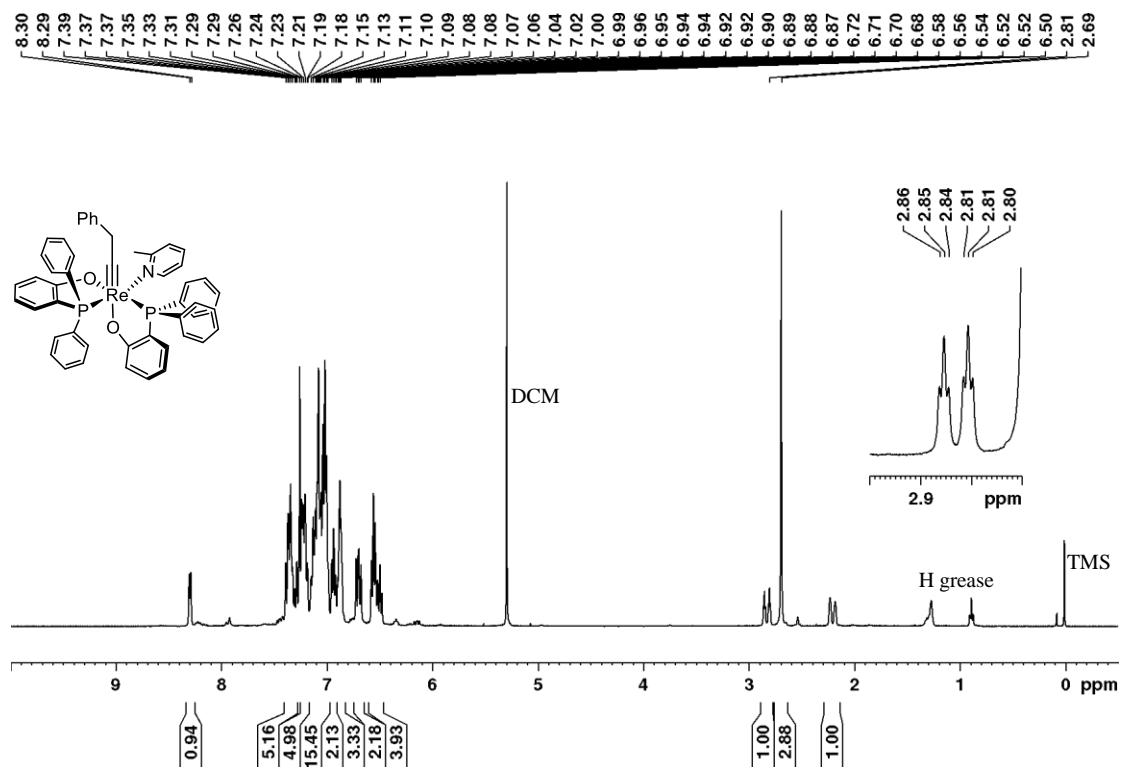


Figure S21. The ^1H NMR spectrum of $\text{Re}(\equiv\text{CCH}_2\text{Ph})(\text{PhPO})_2(2\text{-Me py})$ (**16**) in CDCl_3 at 400.1 MHz.

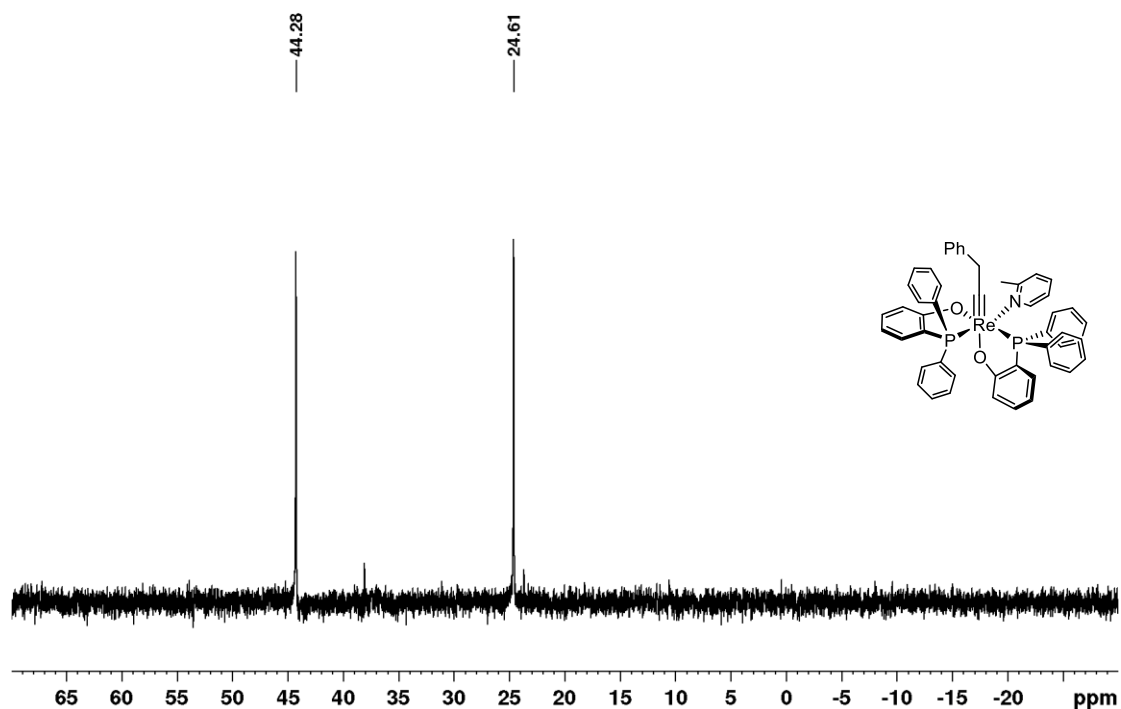


Figure S22. The $^{31}\text{P}\{^1\text{H}\}$ NMR spectrum of $\text{Re}(\equiv\text{CCH}_2\text{Ph})(\text{PhPO})_2(^{2\text{-Me}}\text{py})$ (16) in CDCl_3 at 162.0 MHz.

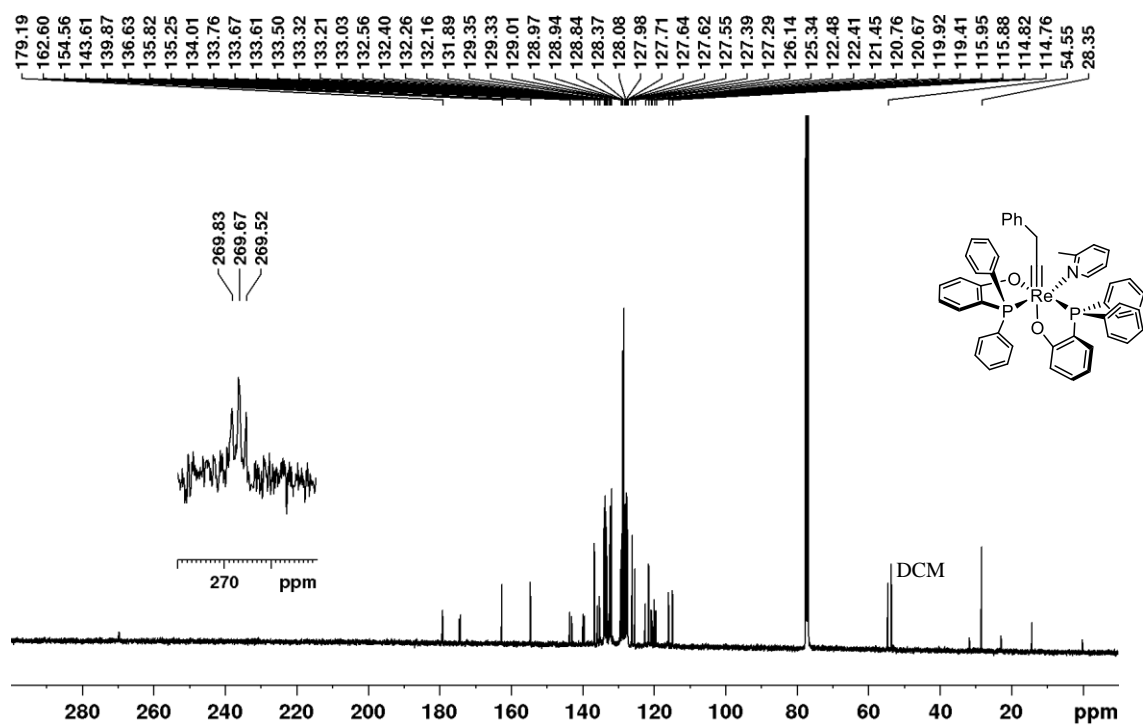


Figure S23. The $^{13}\text{C}\{^1\text{H}\}$ NMR spectrum of $\text{Re}(\equiv\text{CCH}_2\text{Ph})(\text{PhPO})_2(^{2\text{-Me}}\text{py})$ (16) in CDCl_3 at 100.6 MHz.

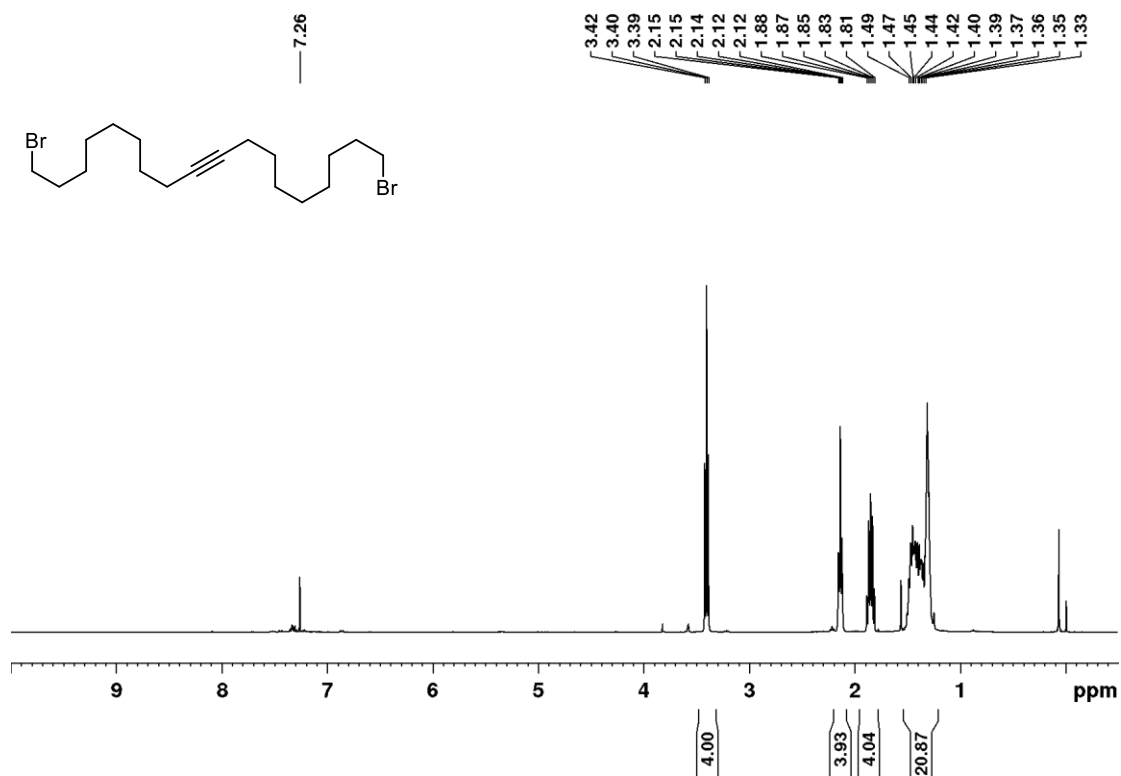


Figure S24. The ¹H NMR spectrum of 1,18-dibromooctadec-9-yne (**21a**) in CDCl₃ at 400.1 MHz.

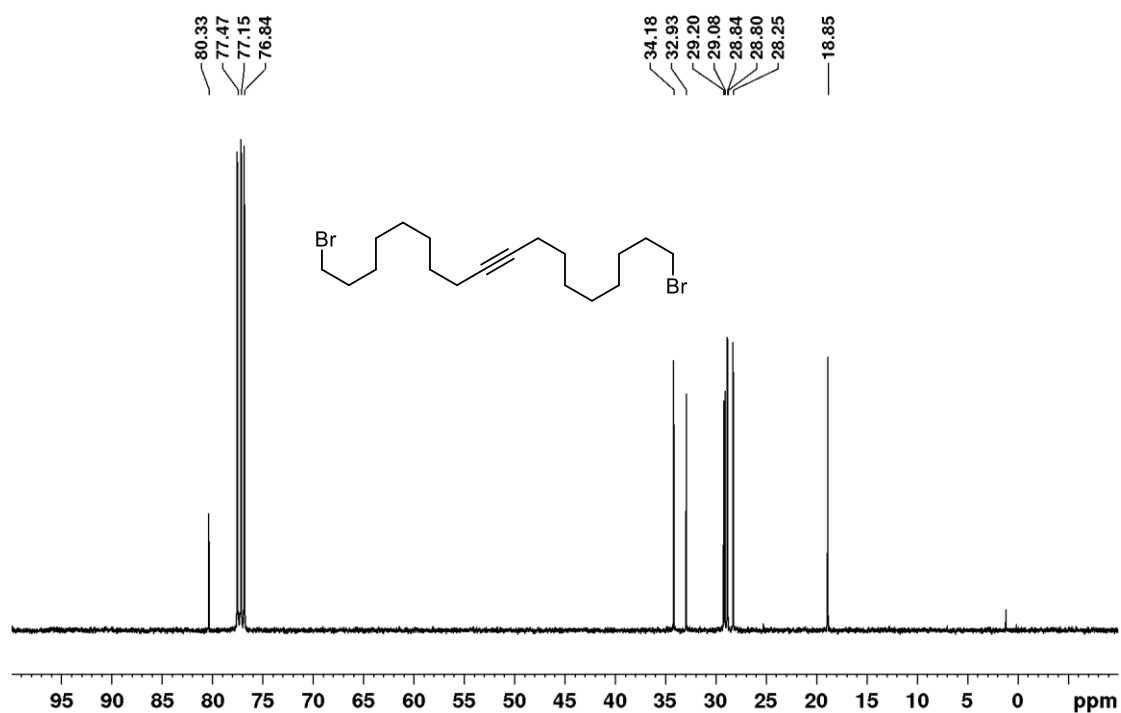


Figure S25. The ¹³C{¹H} NMR spectrum of 1,18-dibromooctadec-9-yne (**21a**) in CDCl₃ at 100.6 MHz.

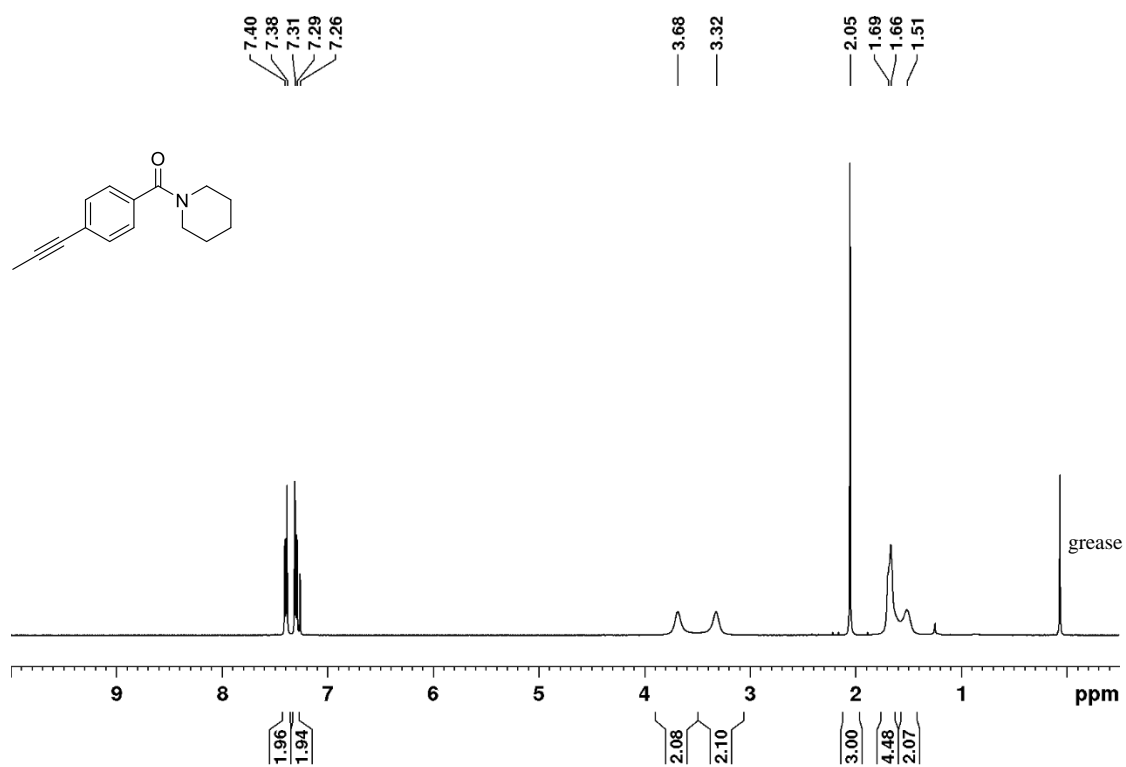


Figure S26. The ¹H NMR spectrum of piperidin-1-yl(4-(prop-1-yn-1-yl)phenyl)methanone (**25**) in CDCl₃ at 400.1 MHz.

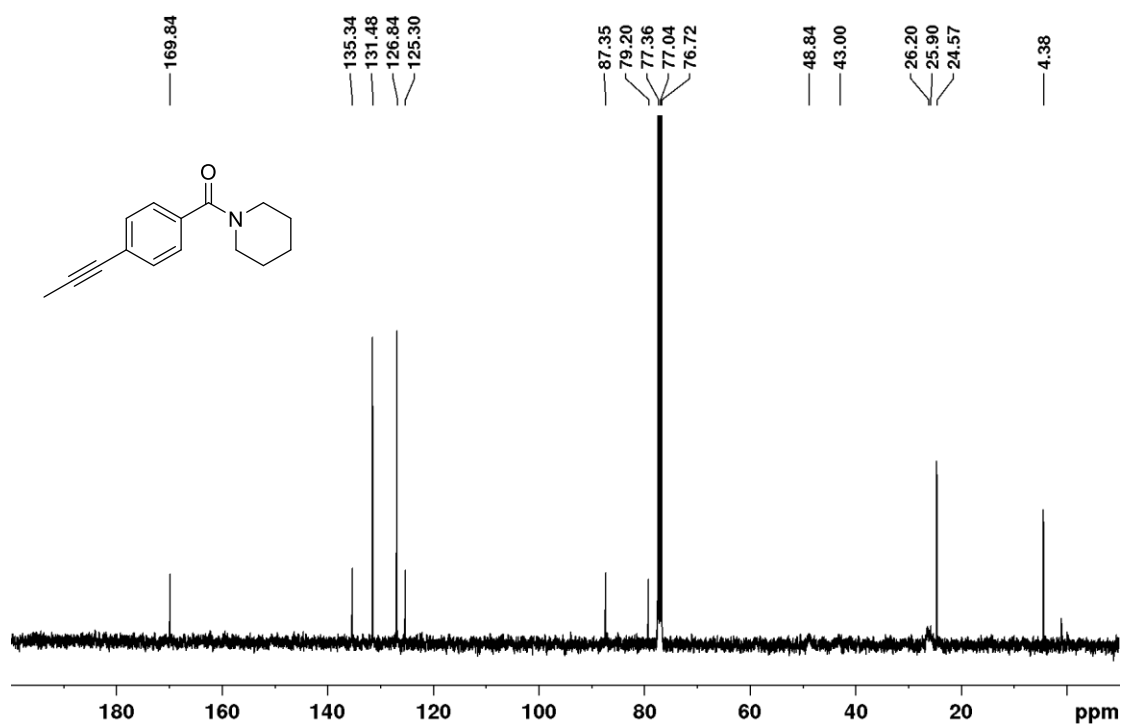


Figure S27. The ¹³C{¹H} NMR spectrum of piperidin-1-yl(4-(prop-1-yn-1-yl)phenyl)methanone (**25**) in CDCl₃ at 100.6 MHz.

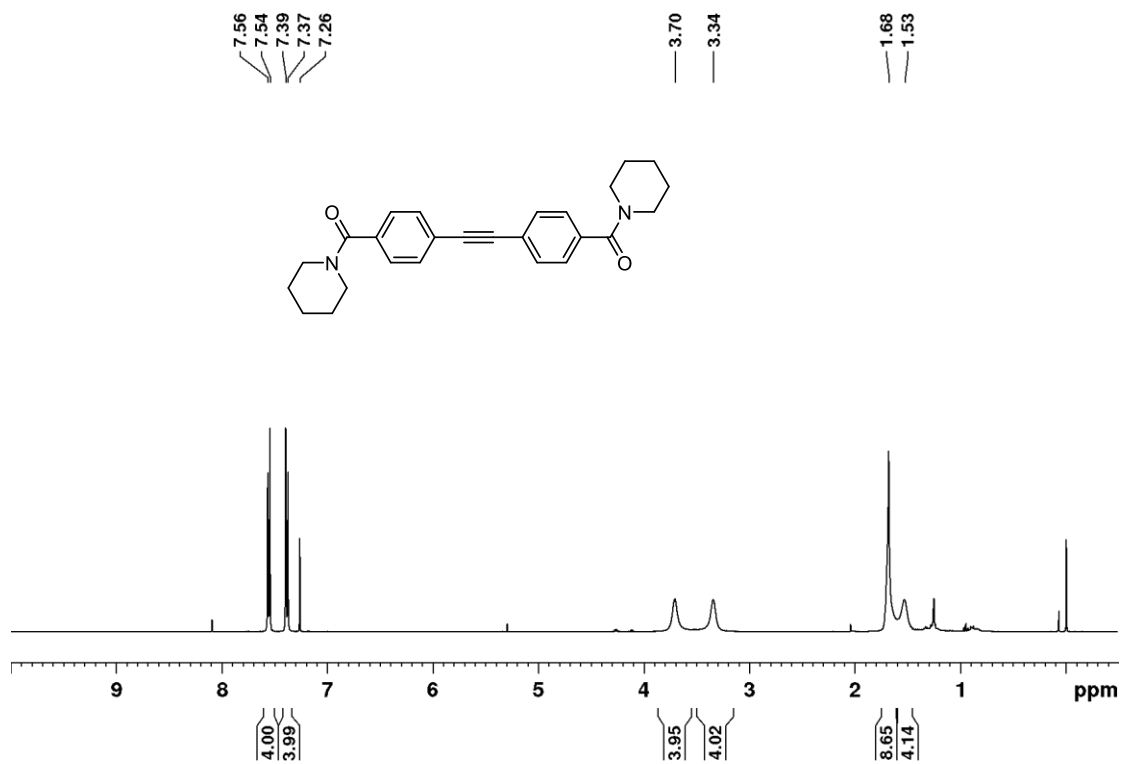


Figure S28. The ^1H NMR spectrum of (ethyne-1,2-diylbis(4,1-phenylene))bis(piperidin-1-ylmethanone) (**25a**) in CDCl₃ at 400.1 MHz.

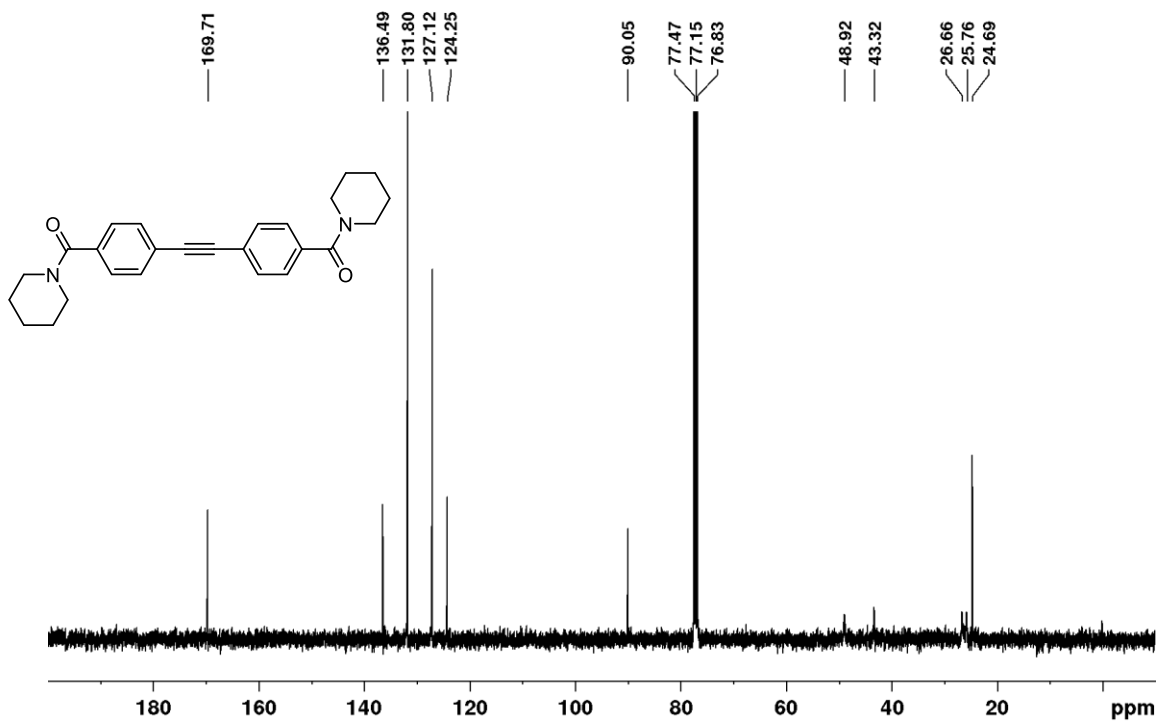


Figure S29. The $^{13}\text{C}\{^1\text{H}\}$ NMR spectrum of (ethyne-1,2-diylbis(4,1-phenylene))bis(piperidin-1-ylmethanone) (**25a**) in CDCl₃ at 100.6 MHz.

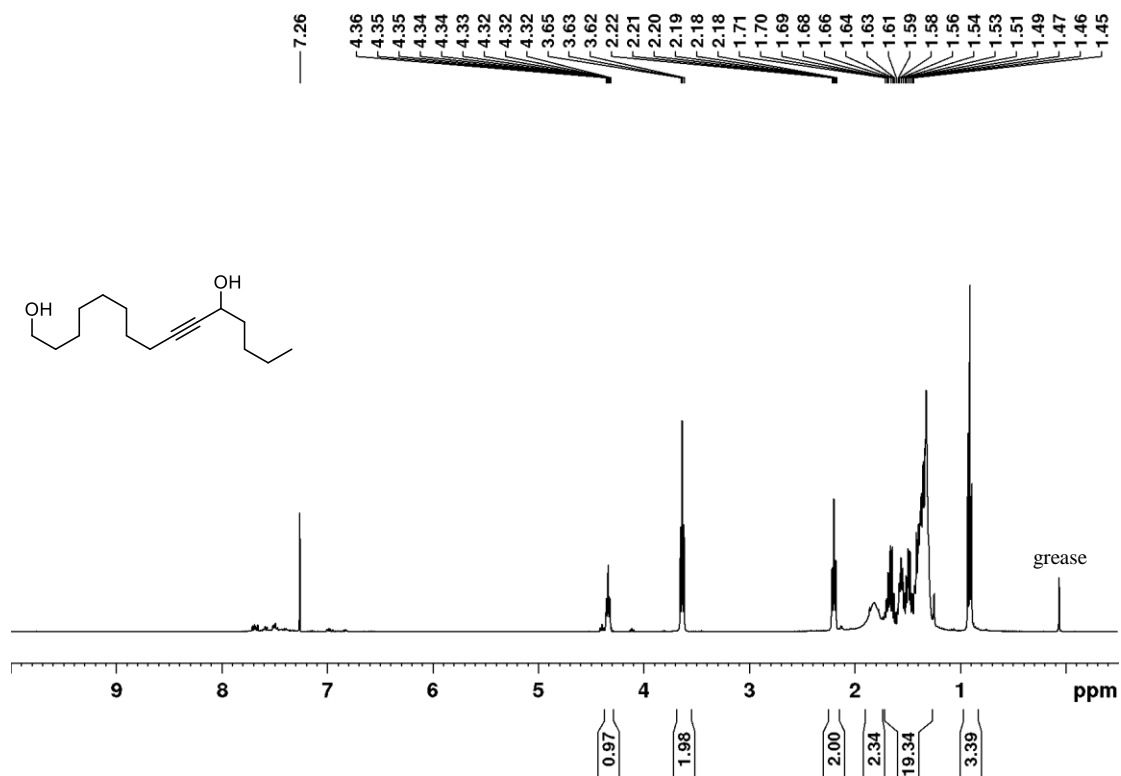


Figure S30. The ¹H NMR spectrum of pentadec-9-yne-1,11-diol (**36**) in CDCl₃ at 400.1 MHz.

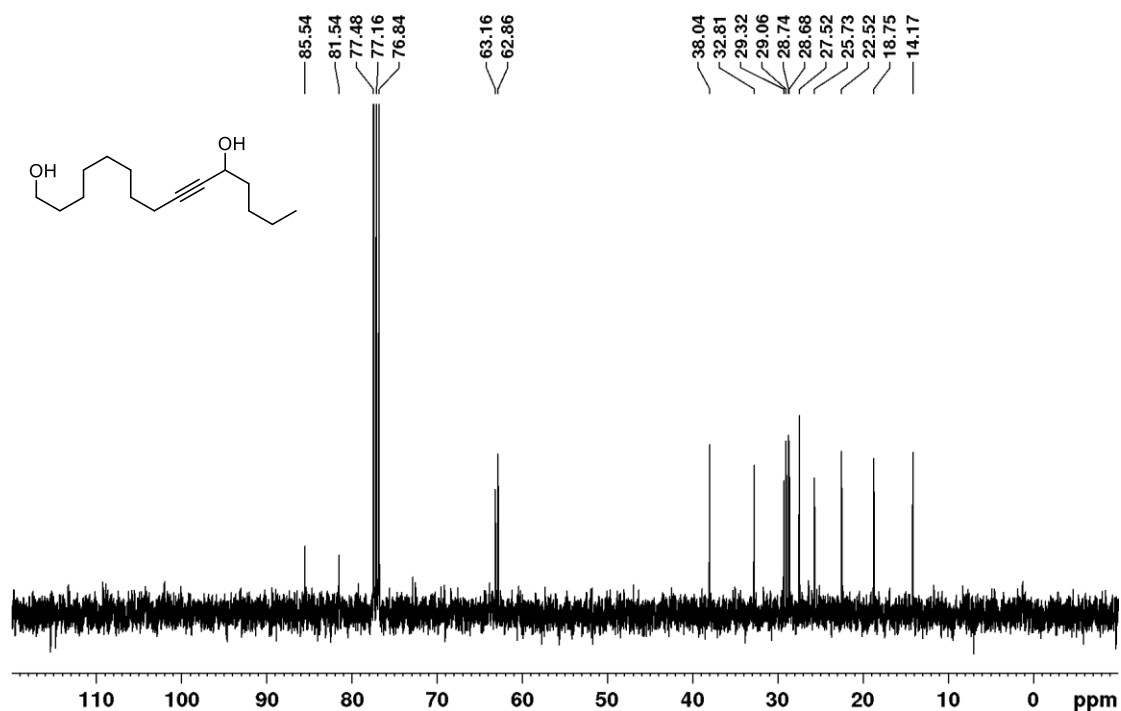


Figure S31. The ¹³C{¹H} NMR spectrum of pentadec-9-yne-1,11-diol (**36**) in CDCl₃ at 100.6 MHz.

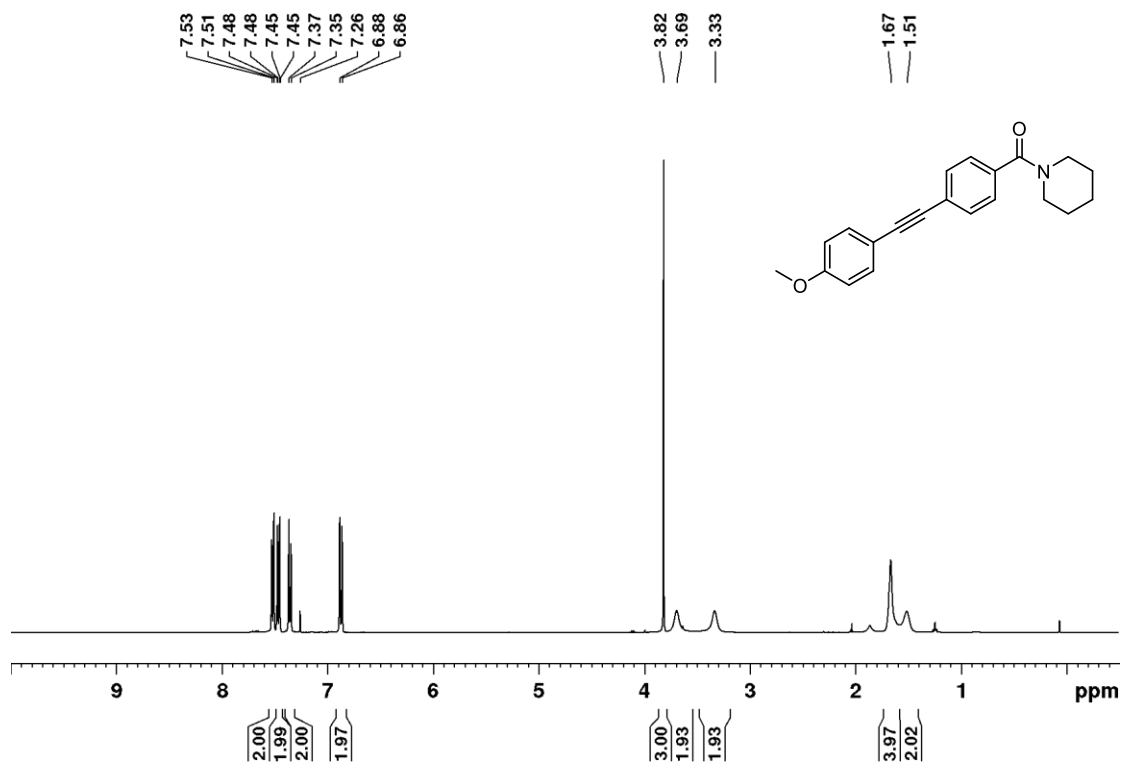


Figure S32. The ^1H NMR spectrum of (4-((4-methoxyphenyl)ethynyl)phenyl)(piperidin-1-yl)methanone (**37**) in CDCl_3 at 400.1 MHz.

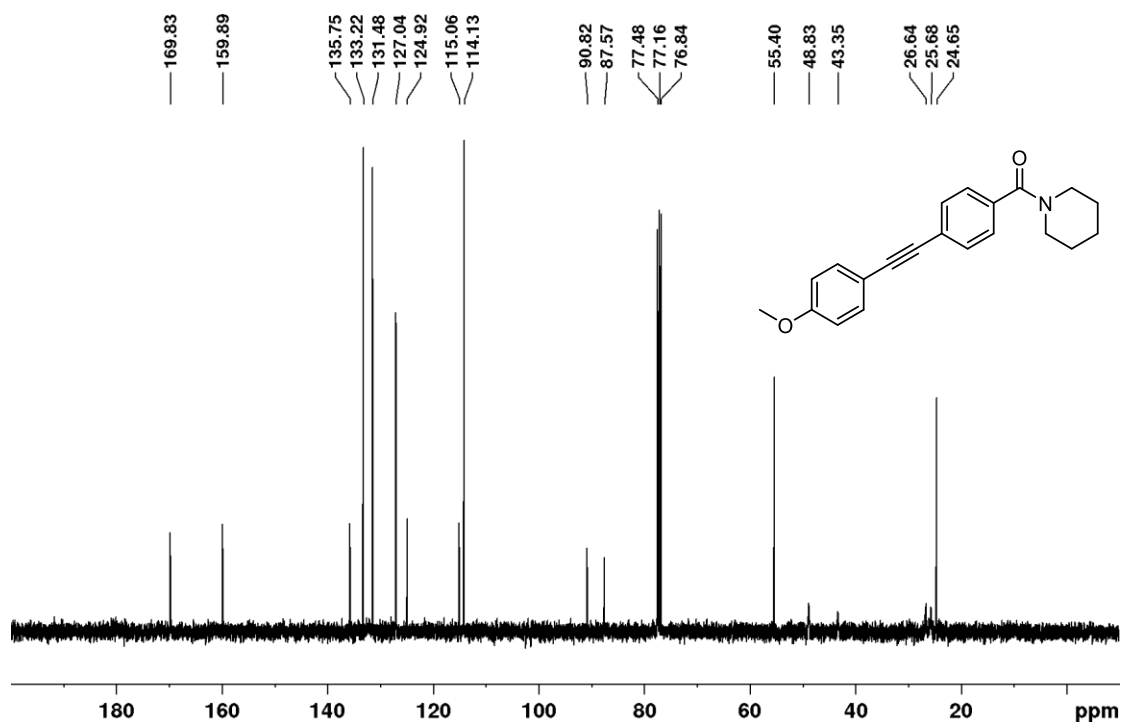


Figure S33. The $^{13}\text{C}\{^1\text{H}\}$ NMR spectrum of (4-((4-methoxyphenyl)ethynyl)phenyl)(piperidin-1-yl)methanone (**37**) in CDCl_3 at 100.6 MHz.

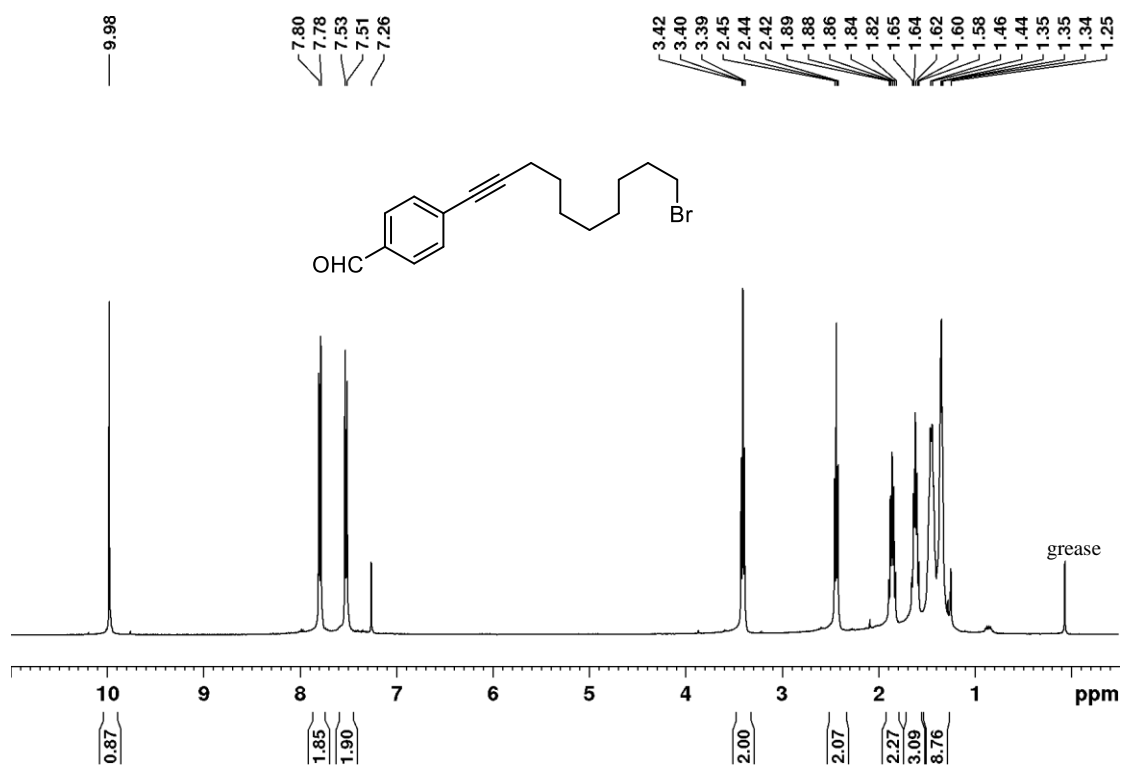


Figure S34. The ^1H NMR spectrum of 4-(10-bromodec-1-yn-1-yl)benzaldehyde (**38**) in CDCl_3 at 400.1 MHz.

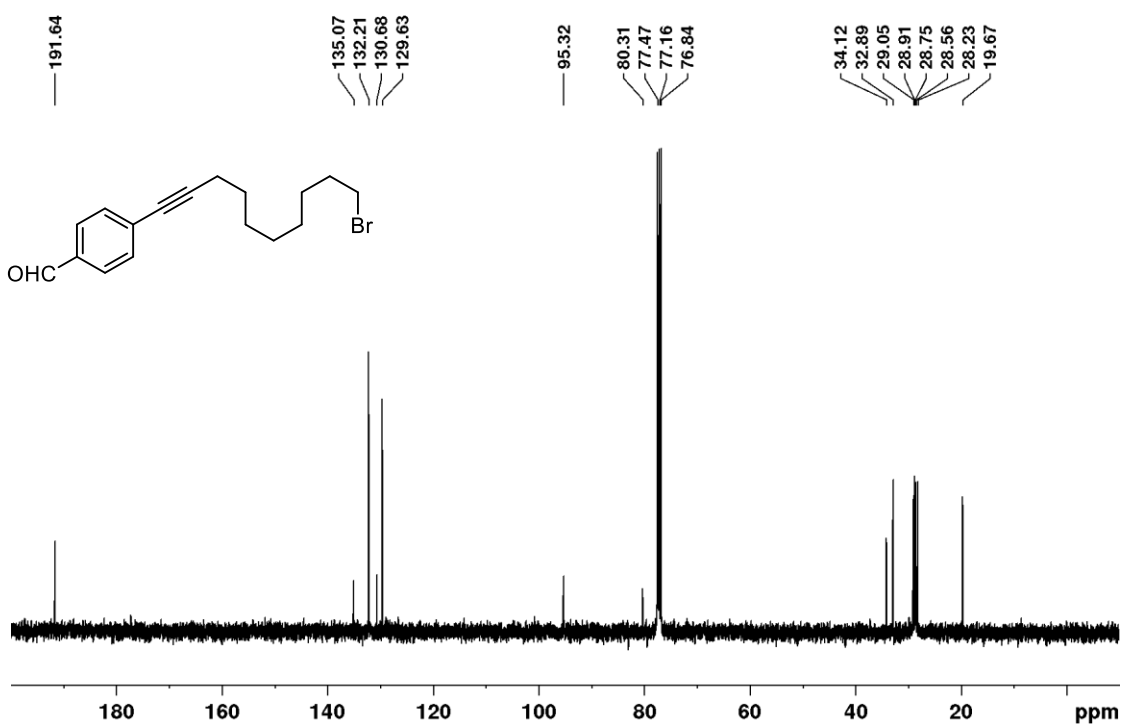


Figure S35. The $^{13}\text{C}\{^1\text{H}\}$ NMR spectrum of 4-(10-bromodec-1-yn-1-yl)benzaldehyde (**38**) in CDCl_3 at 100.6 MHz.

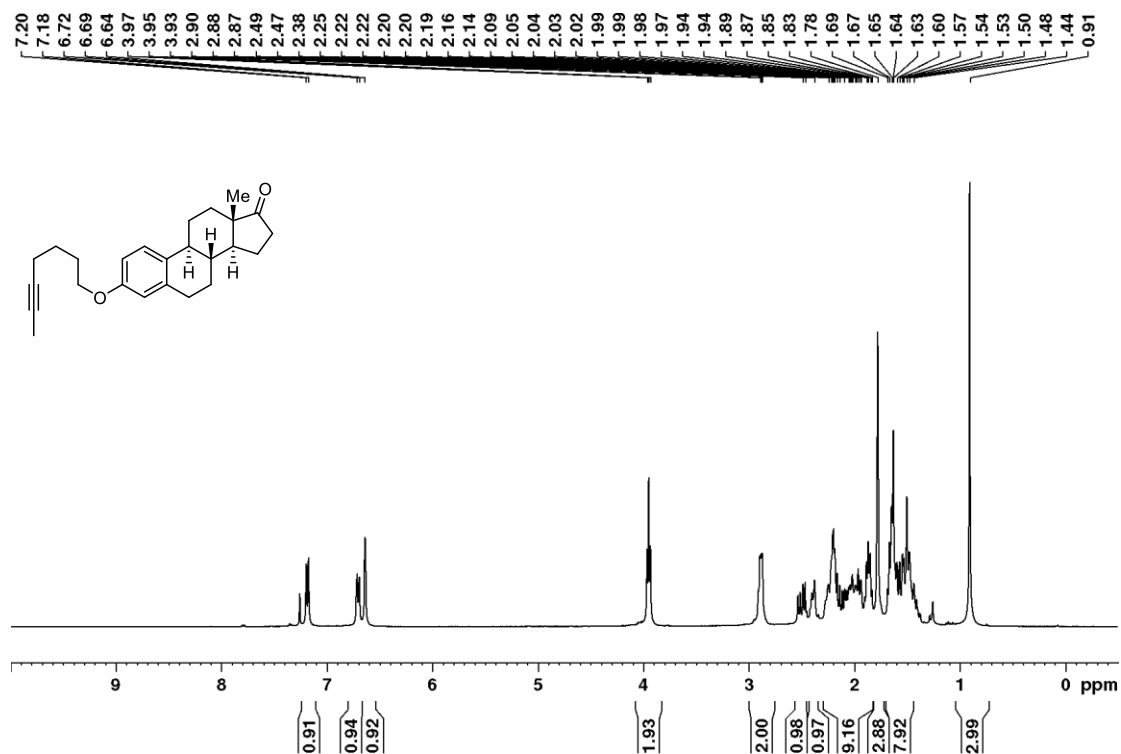


Figure S36. The ¹H NMR spectrum of estrone hept-5-ynyl ether (**39**) in CDCl₃ at 400.1 MHz.

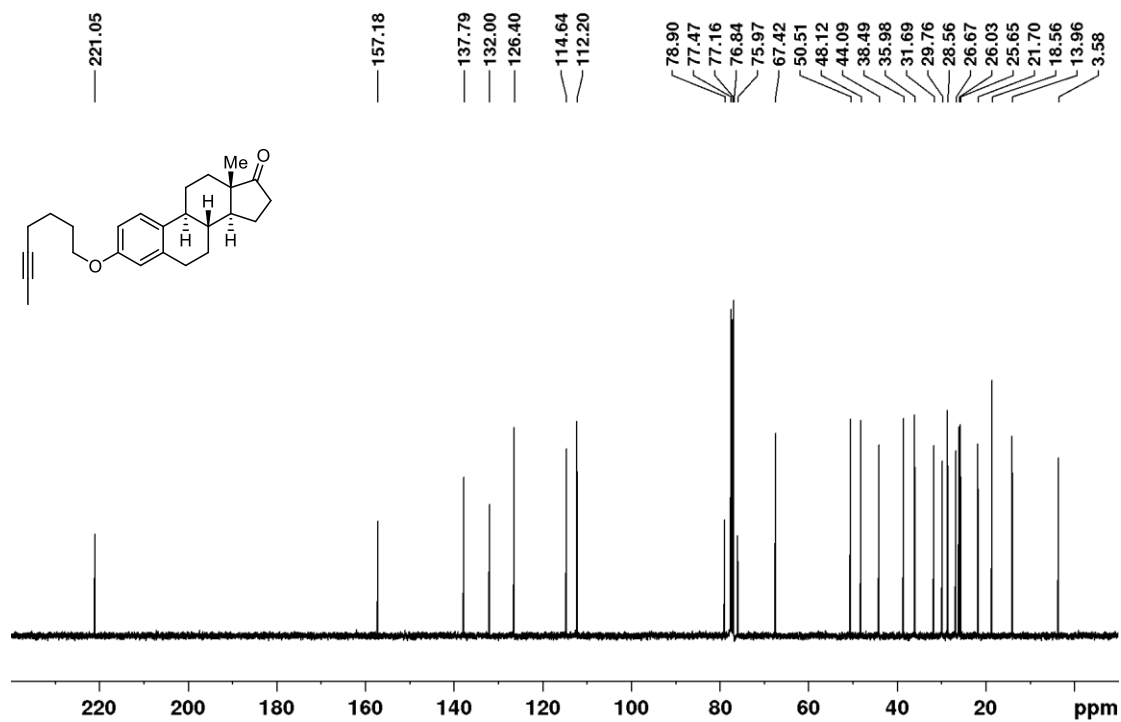


Figure S37. The ¹³C{¹H} NMR spectrum of estrone hept-5-ynyl ether (**39**) in CDCl₃ at 100.6 MHz.

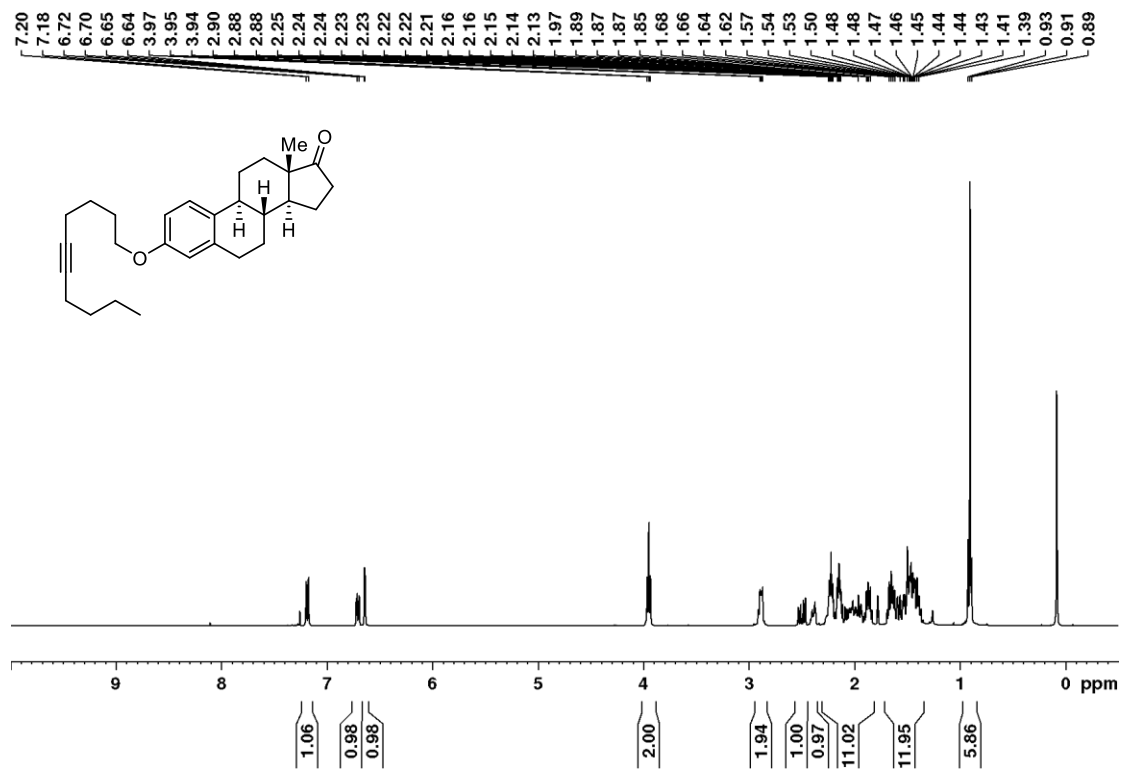


Figure S38. The ¹H NMR spectrum of estrone dec-5-ynyl ether (**40**) in CDCl₃ at 400.1 MHz.

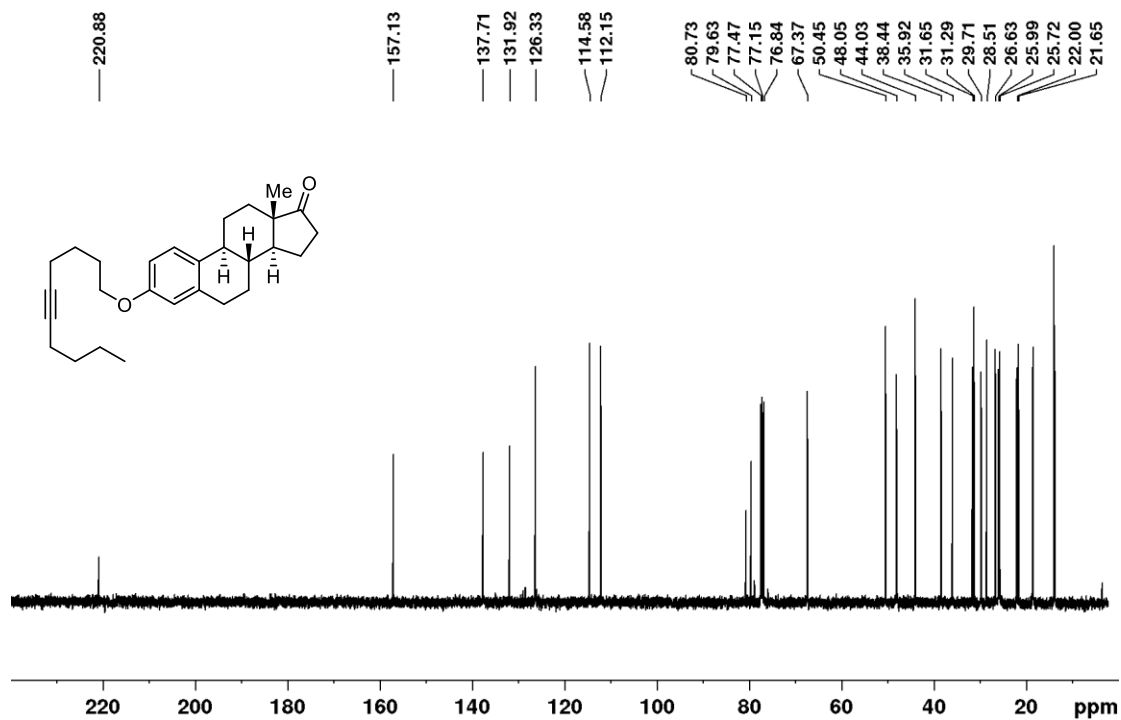


Figure S39. The ¹³C{¹H} NMR spectrum of estrone dec-5-ynyl ether (**40**) in CDCl₃ at 100.6 MHz.

5. Reference

- [1] Dolomanov, O. V.; Bourhis, L. J.; Gildea, R. J.; Howard, J. A. K.; Puschmann, H., OLEX2: a complete structure solution, refinement and analysis program. *J. Appl. Crystallogr.* **2009**, *42*, 339-341.
- [2] Sheldrick, G. M., SHELXT - Integrated space-group and crystal-structure determination. *Acta Cryst.* **2015**, *A71*, 3-8.
- [3] Sheldrick, G. M., Crystal structure refinement with SHELXL. *Acta Cryst.* **2015**, *C71*, 3-8.
- [4] Bennett, M. A.; Huang, T. N.; Matheson, T. W.; Smith, A. K., (η^6 -Hexamethylbenzene)ruthenium Complexes. *Inorganic Syntheses*, **1982**, *21*, 74-78.
- [5] Werner, H.; Zenkert, K., Arene(phosphine) metal complexes. XIII. Osmium(II) and osmium(0) complexes with p-cymene as the aromatic ligand. *J. Organomet. Chem.* **1988**, *345*, 151-166.
- [6] Kapadnis, P. B.; Hall, E.; Ramstedt, M.; Galloway, W. R.; Welch, M.; Spring, D. R., Towards quorum-quenching catalytic antibodies. *Chem. Commun.* **2009**, 538-540. The phosphine is also commercially available, for example, from Sigma-Aldrich Chemicals (CAS No.: 60254-10-6).
- [7] Cui, M.; Sung, H. H. Y.; Williams, I. D.; Jia, G., Alkyne Metathesis with d^2 Re(V) Alkylidyne Complexes Supported by Phosphino-Phenolates: Ligand Effect on Catalytic Activity and Applications in Ring-Closing Alkyne Metathesis. *J. Am. Chem. Soc.* **2022**, *144*, 6349-6360.
- [8] Cui, M.; Bai, W.; Sung, H. H. Y.; Williams, I. D.; Jia, G. Robust Alkyne Metathesis Catalyzed by Air Stable d^2 Re(V) Alkylidyne Complexes. *J. Am. Chem. Soc.* **2020**, *142*, 13339-13344.
- [9] Pschirer, N. G.; Bunz, U. H. F., Alkyne metathesis with simple catalyst systems: High yield dimerization of propynylated aromatics; scope and limitations. *Tetrahedron Lett.* **1999**, *40*, 2481-2484.
- [10] Yoneyama, H.; Numata, M.; Uemura, K.; Usami, Y.; Harusawa, S., Transformation of Carbonyl Compounds into Homologous Alkynes under Neutral Conditions: Fragmentation of Tetrazoles Derived from Cyanophosphates. *J. Org. Chem.* **2017**, *82*, 5538-5556.
- [11] Haydl, A. M.; Hilpert, L. J.; Breit, B., Regioconvergent and Enantioselective Rhodium-Catalyzed Hydroamination of Internal and Terminal Alkynes: A Highly Flexible Access to Chiral Pyrazoles. *Chem. Eur. J.* **2016**, *22*, 6547-6551.
- [12] Mathys, M.; Kraft, P., Synthesis by ring-closing alkyne metathesis with selective hydrogenation, and olfactory comparison of (7E)- and (7Z)-cyclohexadec-7-enone (Aurelione((R))). *Chem. Biodivers.* **2014**, *11*, 1597-1607.

- [13] Tomita, R.; Koike, T.; Akita, M., Photoredox-Catalyzed Stereoselective Conversion of Alkynes into Tetrasubstituted Trifluoromethylated Alkenes. *Angew. Chem. Int. Ed.* **2015**, *54*, 12923-12927.
- [14] Ahmed, T. S.; Montgomery, T. P.; Grubbs, R. H., Using stereoretention for the synthesis of E-macrocycles with ruthenium-based olefin metathesis catalysts. *Chem. Sci.* **2018**, *9*, 3580-3583.
- [15] Fürstner, A.; Rumbo, A., Ring-Closing Alkyne Metathesis. Stereoselective Synthesis of the Cytotoxic Marine Alkaloid Motuporamine C. *J. Org. Chem.* **2000**, *65*, 2608-2611.
- [16] Schulz, S.; Yildizhan, S.; Stritzke, K.; Estrada, C.; Gilbert, L. E., Macrolides from the scent glands of the tropical butterflies *Heliconius cydno* and *Heliconius pachinus*. *Org. Biomol. Chem.* **2007**, *5*, 3434-3441.
- [17] Barker, G.; Johnson, D. G.; Young, P. C.; Macgregor, S. A.; Lee, A. L., Chirality Transfer in Gold(I)-Catalysed Direct Allylic Etherifications of Unactivated Alcohols: Experimental and Computational Study. *Chem. Eur. J.* **2015**, *21*, 13748-13757.
- [18] Thiel, N. O.; Kaewmee, B.; Tran Ngoc, T.; Teichert, J. F., A Simple Nickel Catalyst Enabling an E-Selective Alkyne Semihydrogenation. *Chem. Eur. J.* **2020**, *26*, 1597-1603.
- [19] Zhang, W.; Moore, J. S., Arylene ethynylene macrocycles prepared by precipitation-driven alkyne metathesis. *J. Am. Chem. Soc.* **2004**, *126*, 12796.
- [20] Fulmer, G. R.; Miller, A. J.; Sherden, N. H.; Gottlieb, H. E.; Nudelman, A.; Stoltz, B. M.; Bercaw, J. E.; Goldberg, K. I., NMR chemical shifts of trace impurities: common laboratory solvents, organics, and gases in deuterated solvents relevant to the organometallic chemist. *Organometallics*, **2010**, *29*, 2176-2179.
- [21] Bai, W.; Wei, W.; Sung, H. H. Y.; Williams, I. D.; Lin, Z.; Jia, G., Syntheses of Re(V) Alkylidyne Complexes and Ligand Effect on the Reactivity of Re(V) Alkylidyne Complexes toward Alkynes. *Organometallics* **2018**, *37*, 559-569.
- [22] a) Chatt, J.; Leigh, G. J.; Mingos, D. M. P.; Paske, R. J. Complexes of Osmium, Ruthenium, Rhenium, and Iridium Halides with Some Tertiary Monophosphines and Monoarsines. *J. Chem. Soc. A*, **1968**, 2636 – 2641. b) Skupinski, W. A.; Huffman, J. C.; Bruno, J. W.; Caulton, K. G., Dinuclear elimination from rhenium hydrides and trimethylaluminum: rhenium/aluminum polyhydrides. *J. Am. Chem. Soc.* **1984**, *106*, 8128-8136.
- [23] Cotton, F. A.; Luck, R. L., Reduction of rhenium pentachloride in the presence of methylidiphenylphosphine to give $\text{mer-ReCl}_3(\text{PMePh}_2)_3$, $\text{ReCl}(\eta^2\text{-H}_2)(\text{PMePh}_2)_4$, $\text{ReH}_3(\text{PMePh}_2)_4$, or $\text{ReCl}(\text{CO})_3(\text{PMePh}_2)_2$, depending on conditions. *Inorg. Chem.* **1989**, *28*, 2181-2186.
- [24] Yang, H.; Liu, Z.; Zhang, W., Multidentate Triphenolsilane-Based Alkyne

Metathesis Catalysts. *Adv. Synth. Catal.* **2013**, *355*, 885-890.

[25] Zhang, W.; Kraft, S.; Moore, J. S., Highly active trialkoxymolybdenum(VI) alkylidyne catalysts synthesized by a reductive recycle strategy. *J. Am. Chem. Soc.* **2004**, *126*, 329-335.

[26] Takase, M.; Nakajima, A.; Takeuchi, T., Synthesis of an extended hexagonal molecule as a highly symmetrical ligand. *Tetrahedron Lett.* **2005**, *46*, 1739-1742.

[27] Mio, M. J.; Kopel, L. C.; Braun, J. B.; Gadzikwa, T. L.; Hull, K. L.; Brisbois, R. G.; Markworth, C. J.; Grieco, P. A., One-pot synthesis of symmetrical and unsymmetrical bisarylethynes by a modification of the sonogashira coupling reaction. *Org. Lett.* **2002**, *4*, 3199-3202.

[28] Adam, A.; Haberhauer, G., Twisting of Alkynes towards a Carbon Double Helix. *Chem. Eur. J.* **2017**, *23*, 12190-12197.

[29] Hillenbrand, J.; Leutzsch, M.; Yiannakas, E.; Gordon, C. P.; Wille, C.; Nothling, N.; Coperet, C.; Furstner, A., "Canopy Catalysts" for Alkyne Metathesis: Molybdenum Alkylidyne Complexes with a Tripodal Ligand Framework. *J. Am. Chem. Soc.* **2020**, *142*, 11279-11294.

[30] Korber, J. N.; Wille, C.; Leutzsch, M.; Furstner, A., From the Glovebox to the Benchtop: Air-Stable High Performance Molybdenum Alkylidyne Catalysts for Alkyne Metathesis. *J. Am. Chem. Soc.* **2023**, *145*, 26993-27009.

[31] Hillenbrand, J.; Leutzsch, M.; Furstner, A., Molybdenum Alkylidyne Complexes with Tripodal Silanolate Ligands: The Next Generation of Alkyne Metathesis Catalysts. *Angew. Chem., Int. Ed.* **2019**, *58*, 1-8.

[32] Schaubach, S.; Gebauer, K.; Ungeheuer, F.; Hoffmeister, L.; Ilg, M. K.; Wirtz, C.; Furstner, A., A Two-Component Alkyne Metathesis Catalyst System with an Improved Substrate Scope and Functional Group Tolerance: Development and Applications to Natural Product Synthesis. *Chem. Eur. J.* **2016**, *22*, 8494-8507.

[33] Bindl, M.; Stade, R.; Heilmann, E. K.; Picot, A.; Goddard, R.; Furstner, A., Molybdenum nitride complexes with Ph₃SiO ligands are exceedingly practical and tolerant precatalysts for alkyne metathesis and efficient nitrogen transfer agents. *J. Am. Chem. Soc.* **2009**, *131*, 9468-9470.