

**Sm₆WO₁₂ Tungstate Supported Nickel-Based Catalysts with Enhanced Resistance to
Coking and Oxidation in Auto-thermal Reforming of Acetic Acid**

Xuemei Xie^{a,b}, Yingchun Xu^b, Mao Gan^b, Ying Su^b, Jinbo Liu^b, Lihong Huang^{a,b,*}

^a State Key Laboratory of Geohazard Prevention and Geoenvironment Protection (Chengdu University of Technology), Chengdu 610059, China

^b College of Materials and Chemistry & Chemical Engineering, Chengdu University of Technology, Chengdu 610059, China

* Corresponding author

E-mail address: huanglihong06@cdut.cn

Table S1. D_{Ni} , Ea and TOF- H_2 of NSW catalysts.

Catalysts	D_{Ni}^a (%)	TOF- H_2 ($10^{-2} s^{-1}$)	Ea (kJ/mol)
NS	1.6	1.56	57.6
NSW20	18.8	2.38	43.5
NSW40	1.8	1.61	68.0

^a The dispersion of Ni (D_{Ni}) was calculated by the results of H_2 -TPD assuming $H_{ad}/Ni^0=1$.

Table S2. Comparison of catalytic performance from reforming.

Catalyst	T (°C)	Time (h)	H_2 yield (%)	TOF- H_2 ($10^{-2} s^{-1}$)	reforming process	Ref.
Ni/ATP ¹	700	-	79	-	SR	[1]
Co/CaSBA-15 ²	600	5	72	-	SR	[2]
Co-Rh/CeO ₂ ³	700	30	76	-	SR	[3]
Ni@SiO ₂ -T ⁴	750	10	58	-	SR	[4]
Co-Ba-Al ⁵	650	10	69	0.4	ATR	[5]
Ni/SmMn ₂ O ₅ ⁶	700	10	76	0.6	ATR	[6]
Zn-Ni-Al-Fe-O ⁷	650	10	69	-	ATR	[7]
Ni/Sm ₆ WO ₁₂ ^{current work}	650	10	70	2.4	ATR	current work

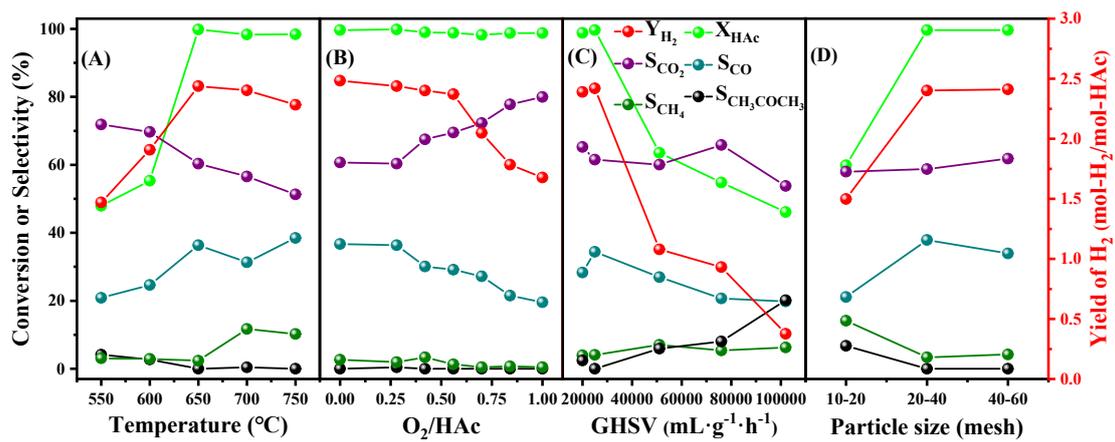


Fig. S1. Effect of (A) temperature, (B) O₂/HAc, (C) GHSV and (D) particle size on the reactivity of NS20W in ATR of HAC.

Notes and references

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