Supporting information

Triggering the phase transition of Molybdenum Di-Selenide (MoSe₂) 1T@2H by introducing Copper (Cu⁺): Experimental Insights and DFT Analysis for hydrogen evolution reaction

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1.Material and reagents

Selenium powder (Se) and Cupric chloride $CuCl_2.2H_2O$ were purchased from Sigma Aldrich and sodium molybdate dihydrate (Na₂MoO₄. 2H₂O), and hydrazine hydrate solution (N₂H₄. H₂O) were purchased from Qualigens Fine Chemicals Ltd, sulphuric acid (H₂SO₄; assay 97%), The above chemicals were of analytical grade and used without any further purification.

1.1 Material characterizations

The crystallinity of powdered Pristine MoSe₂ and 3% & 5% Cu-MoSe₂ samples was examined utilizing a PANalytical EMPYREAN X-ray system with a monochromatic CuK α (λ = 1.5406 Å) radiation source. X-ray photoelectron spectroscopy (XPS) was conducted utilizing the PHI - VERSAPROBE III from Japan to ascertain the elemental makeup and their chemical states. High-resolution scanning electron microscopy (HRSEM) (Thermo-Fisher Scientific, USA) was used to document the morphology of the synthesized Pristine MoSe₂, 3% Cu-MoSe2@CC, and 5% Cu-MoSe2@CC samples. The Energy Dispersive X-ray Analysis (EDX) (Oxford instruments) was used to do elemental analysis accompanied by colour mapping.

2. X-ray diffraction patterns (XRD)



Figure S1- X-ray diffraction patterns of Pristine MoSe₂@CC, 3% Cu-MoSe₂@CC, and 5% Cu-MoSe₂@CC catalyst coated on Carbon cloth

3. X-ray photoelectron spectroscopy (XPS)



Figure S2 -high resolution XPS spectra of Carbon C1s - samples Pristine MoSe₂ and 5%Cu-MoSe₂



Figure S3- Core-level XPS spectra of Cu-2p

4. Electrochemical activity



Figure S4-Linear polarization curves indicating the increased current density of adding Cu on MoSe₂



Figure S5 -CV curves between 0.02 and 0.12 of the Pristine MoSe₂ & 5%Cu-MoSe₂ samples with at different scan rates



Figure S6 - SEM images with different magnification of after stability of 80hrs

5. DFT



Figure S7 The total density of states as a function of energy for Pristine MoSe₂(2H) and Cu-MoSe₂(1T)