## Supporting information

## Title: Granular aqueous suspensions with controlled inter-particular friction and adhesion

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Figure S1 :



Figure S 1: Carboxylate content (-CH<sub>3</sub>COO<sup>-</sup>, MAA) of 62 wt% determined by indirect titration.



Figure S 2: Schematic of the thermostatically controlled drum used for temperature-varying experiments. The drum is connected to a thermosted bath and a peristaltic pump to controll the temperature and water flow around the drum cell.



Figure S 3: Avalanche angle ( ${}^{\theta}c$ ) as a function of rotation speed ( $\omega$ ) for poly(methyl methacrylate) (PMMA) particles (P60) in pure water and hydrolyzed poly(sodium methacrylate) (PMAA-Na) particles (HP60) in pure water and in presence of NaCl (I = 0.5 M). For  $\omega \leq 0.2 \, \text{o.s}^{-1}$ , the avlanche angle ( ${}^{\theta}c$ ) is independent of  $\omega$ , i.e. measurements are performed in a quasi static regime.



Figure S 4: Swelling behavior of hydrolyzed particles (HP60) in the presence of NaCl,  $CaCl_2$ , and  $LaCl_3$ . (Left) Particle diameter (d) as a function of ionic strength (I). (Right) Microscopy images of HP60 particle suspensions in water, and in the presence of NaCl,  $CaCl_2$ , and  $LaCl_3$  at I = 0.5 M.



Figure S 5: Rotating drum images of HP60 particles in presence of mono- (Na<sup>+</sup>) and multivalent cations (Ca<sup>2+</sup>, La<sup>3+</sup>) with increasing ionic strength from 0 to 0.5 mol.L<sup>-1</sup>. The avalanche angles displayed in Figure 4 in the paper were determined from a serie of 20 to 30 images for each data point.