

Supporting Information

Determination of divalent metal ions regulated proton concentration and polarity at the interface of anionic phospholipid membrane

Pratima Mandal, Snigdha Roy, Manisha Karmakar, Sushil Ranjan Bhatta, Chandi Charan Ghosh, Arunabha Thakur and Partha Pratim Parui*

Department of Chemistry, Jadavpur University, Kolkata 700032, India.

Contents

| | | |
|-----------------|--|--------|
| Table S1 | LUV interfacial charge values in presence of bivalent metal ions. | p. S2 |
| Fig. S1 | Synthetic procedure of compound RGG. | p. S3 |
| Fig. S2 | ¹ H NMR spectrum of compound 2 . | p. S4 |
| Fig. S3 | ¹ H NMR spectrum of compound RGG. | p. S5 |
| Fig. S4 | ¹³ C NMR spectrum of compound RGG. | p. S6 |
| Fig. S5 | HR-MS spectrum of compound RGG. | p. S7 |
| Fig. S6 | UV-Vis absorption spectrum of RGG. | p. S8 |
| Fig. S7 | Incubation time-dependent fluorescence changes of RGG. | p. S9 |
| Fig. S8 | Solvent composition dependent ¹ H NMR spectra (down-filed region) of RGG. | p. S10 |
| Fig. S9 | Fluorescence microscopic observation | p. S11 |
| Fig. S10 | LUV-concentration dependence fluorescence studies of RGG. | p. S12 |
| Fig. S11 | pH-dependence fluorescence studies of RGG in presence of LUV. | p. S13 |
| Fig. S12 | Fluorescence studies of RGG in ethanol/buffer medium. | p. S14 |
| Fig. S13 | UV-Vis absorption spectra of GPP in the presence of LUV | p. S15 |
| Fig. S14 | UV-Vis absorption spectra of GPP in the presence of LUV and metal ions. | p. S16 |
| Fig. S15 | Fluorescence studies of RGG in the presence of metal ions (M ²⁺). | p. S17 |
| Fig. S16 | DLS studies for LUVs in presence of metal ions. | p. S18 |
| Fig. S17 | Fluorescence studies of RGG in the presence of Na ⁺ /K ⁺ ions and LUV. | p. S19 |

Table S1 Different bivalent metal ion concentration dependent zeta potential (ZP) for various lipid LUV system. ^a

| M ²⁺ | [M ²⁺] (mM) | ZP (mV) | | |
|------------------|----------------------------|---------|-------|-------|
| | | DMPG | DOPG | POPG |
| | Absence | -61.0 | -58.7 | -54.4 |
| Ca ²⁺ | 0.1 | -60.5 | -57.0 | -53.6 |
| | 0.2 | -58.0 | -55.8 | -52.0 |
| | 0.3 | -56.5 | -53.5 | -50.2 |
| Mg ²⁺ | 0.1 | -59.0 | -57.1 | -53.9 |
| | 0.2 | -58.6 | -55.8 | -52.7 |
| | 0.3 | -56.9 | -53.9 | -51.3 |
| Zn ²⁺ | 0.1 | -59.6 | -57.3 | -53.7 |
| | 0.3 | -58.3 | -56.2 | -52.8 |
| | 0.5 | -56.3 | -54.5 | -51.6 |

^a Lipid concentration was 1 mM

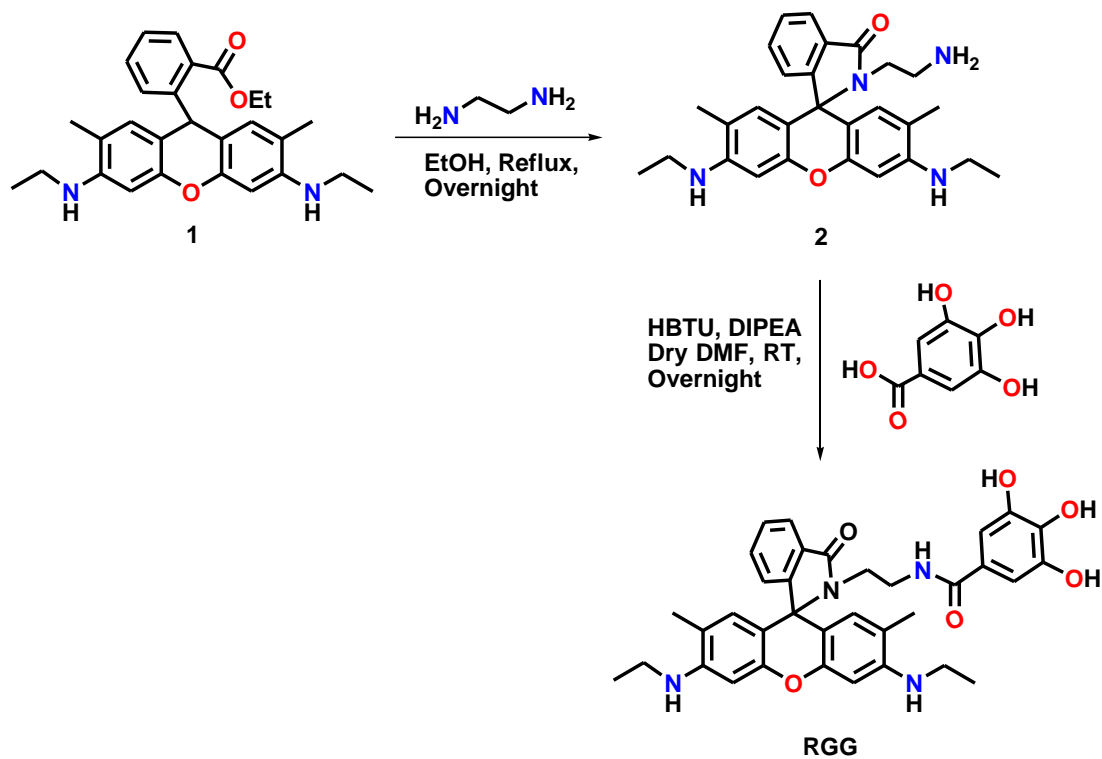


Fig. S1. Synthetic procedure of compound RGG.

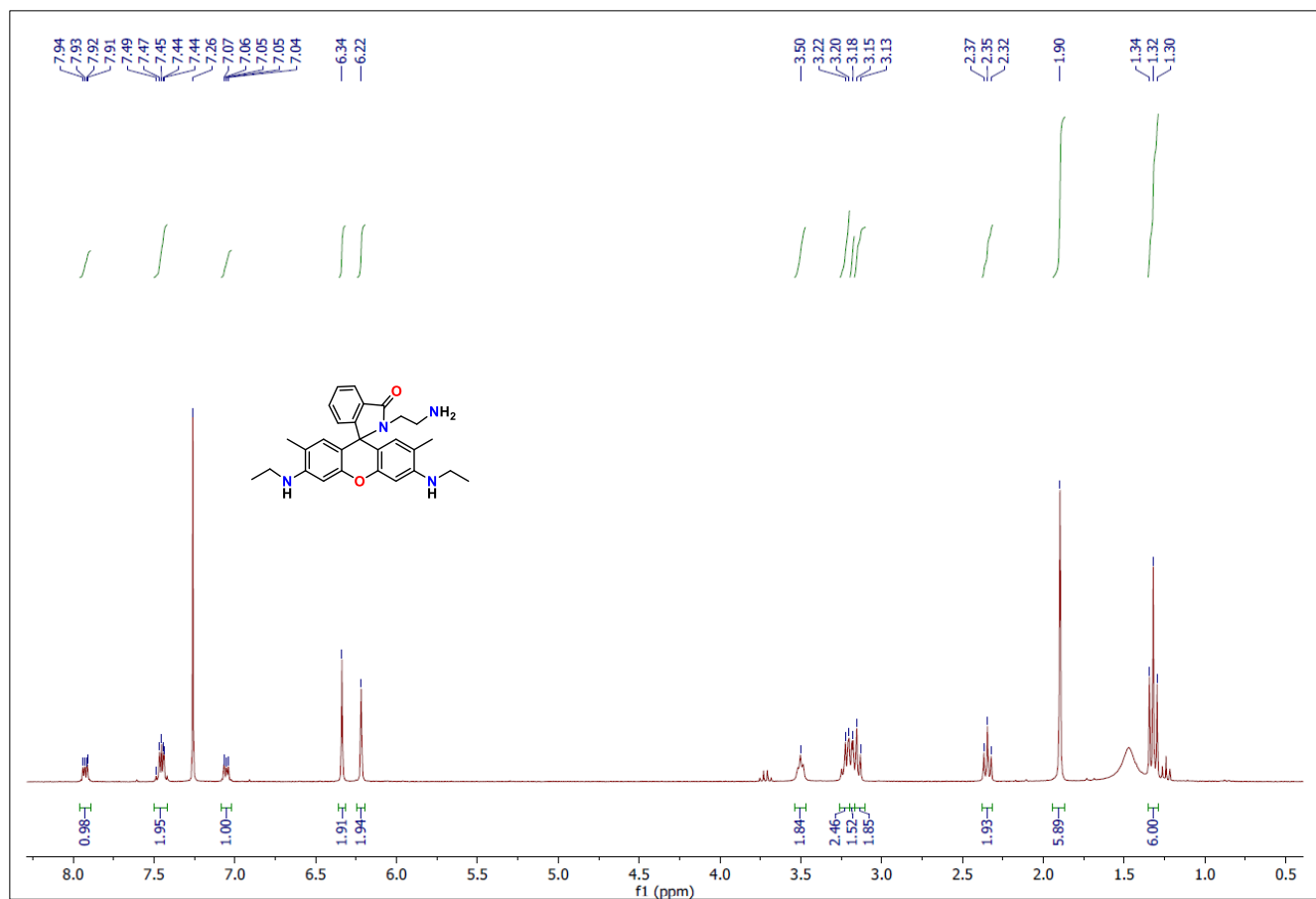


Fig. S2. ¹H NMR spectra of compound **2** in CDCl₃ as a solvent.

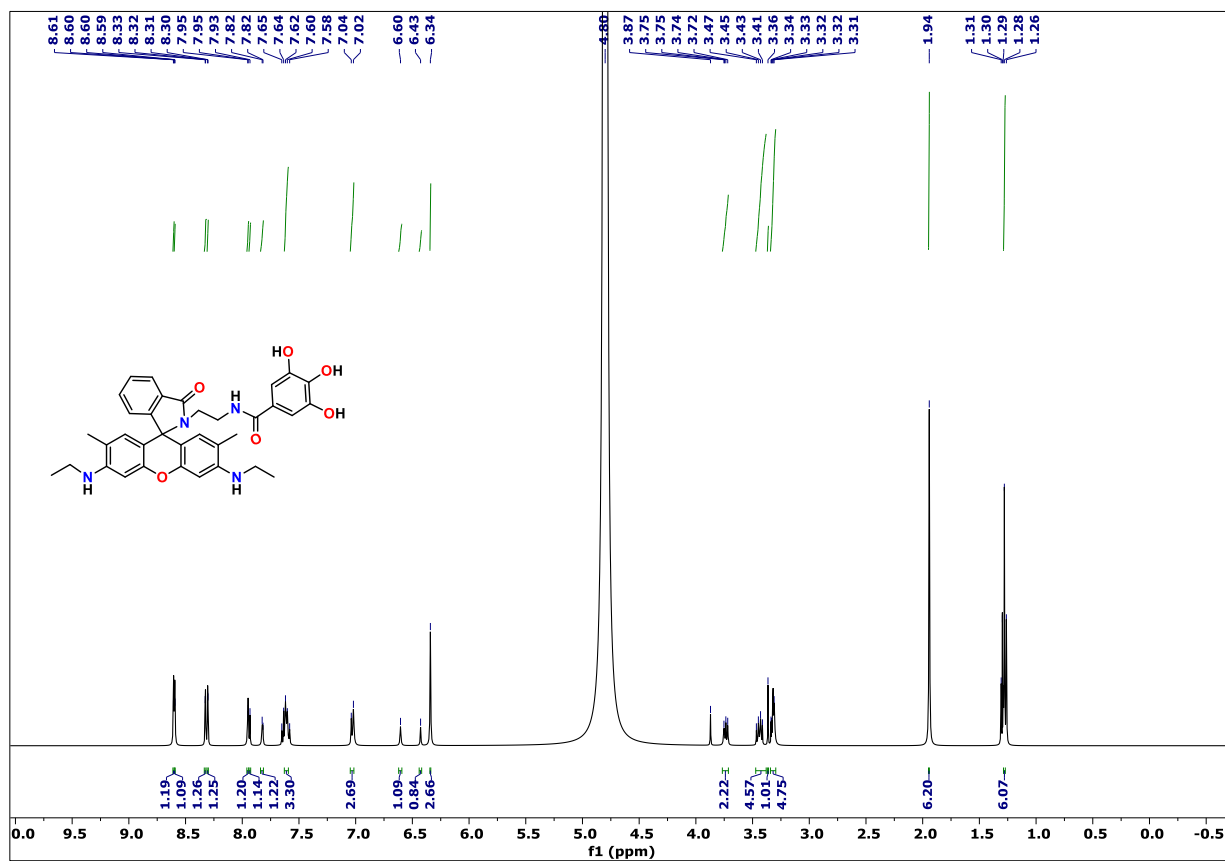


Fig. S3. ^1H NMR spectra of compound RGG in D_2O as a solvent.

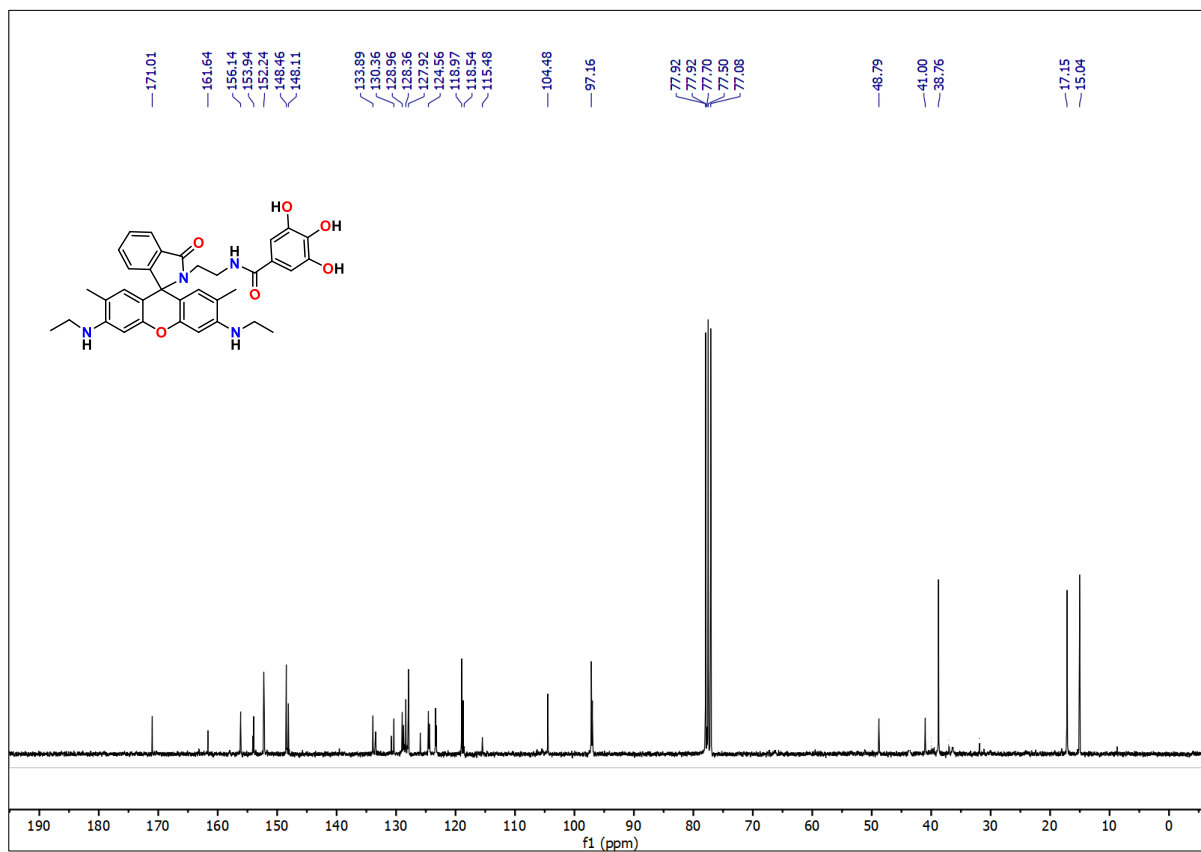


Fig. S4. ¹³C NMR spectra of compound RGG in CDCl₃ as a solvent.

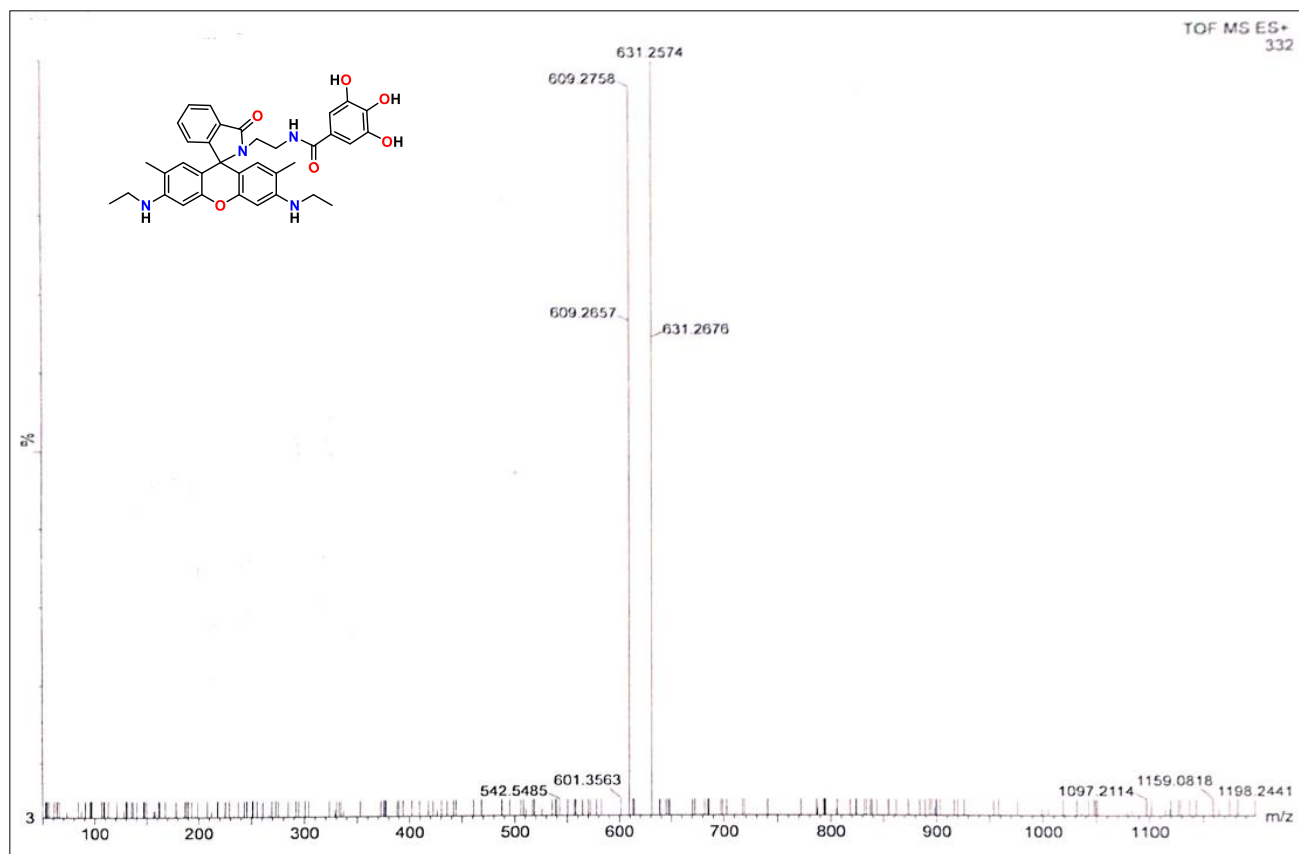


Fig. S5. HR-MS spectrum of compound RGG.

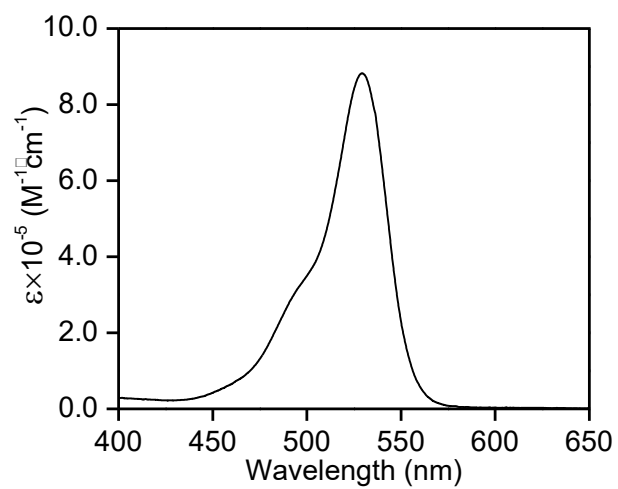


Fig. S6. UV-Vis absorption spectrum of RGG (2 μM) in 2 mM Britton–Robinson buffer, at pH 2.5. The spectrum was recorded 1 hours after addition of RGG.

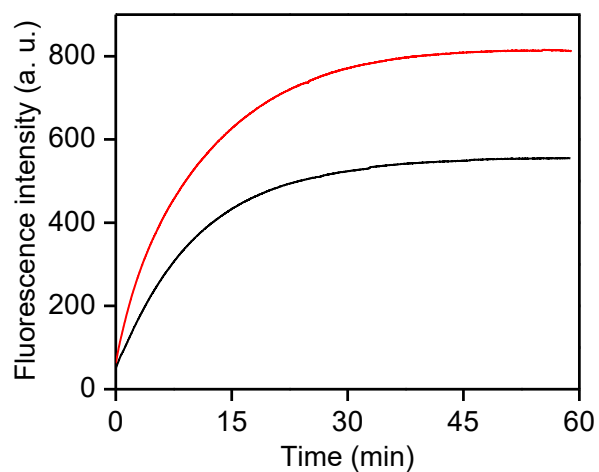


Fig. S7. Changes of fluorescence intensities of RGG at 555 nm in absence (black) and presence (red) of DMPG (1 mM) LUV were plotted against of exposure time of RGG in 2 mM Britton–Robinson buffer, pH 2.5 (black) or HEPES, pH 5.5 at 25 °C. Excitation and emission wavelengths were 530 and 560 nm.

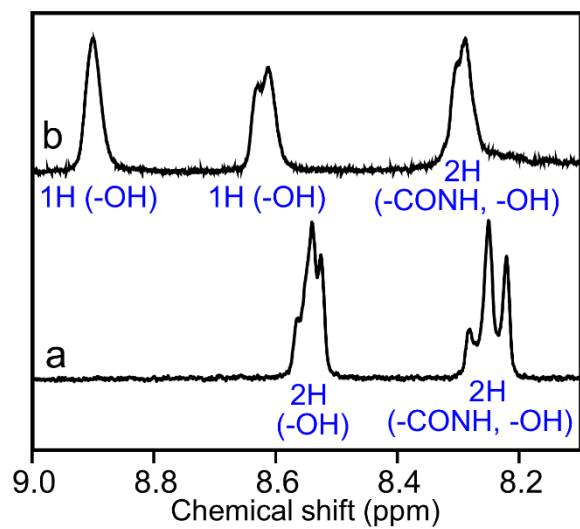


Fig. S8. ¹H NMR spectra of RGG (1.0 mM) in the downfield region in (a) D₂O and (b) 6:4 DMSO-*d*₆ and D₂O medium.

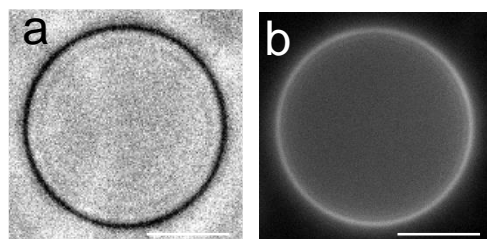


Fig. S9. (a) Phase contrast and (b) fluorescence microscopic observation of DOPG/DOPC (2:1) GUVs (total lipid, 0.5 mM) in 1 mM HEPES buffer, pH 7.0, containing 200 mM sucrose at 25°C. Images were captured after 2 hours of RGG (0.5 μ M) addition in the solution. White bars represent 5 μ m.

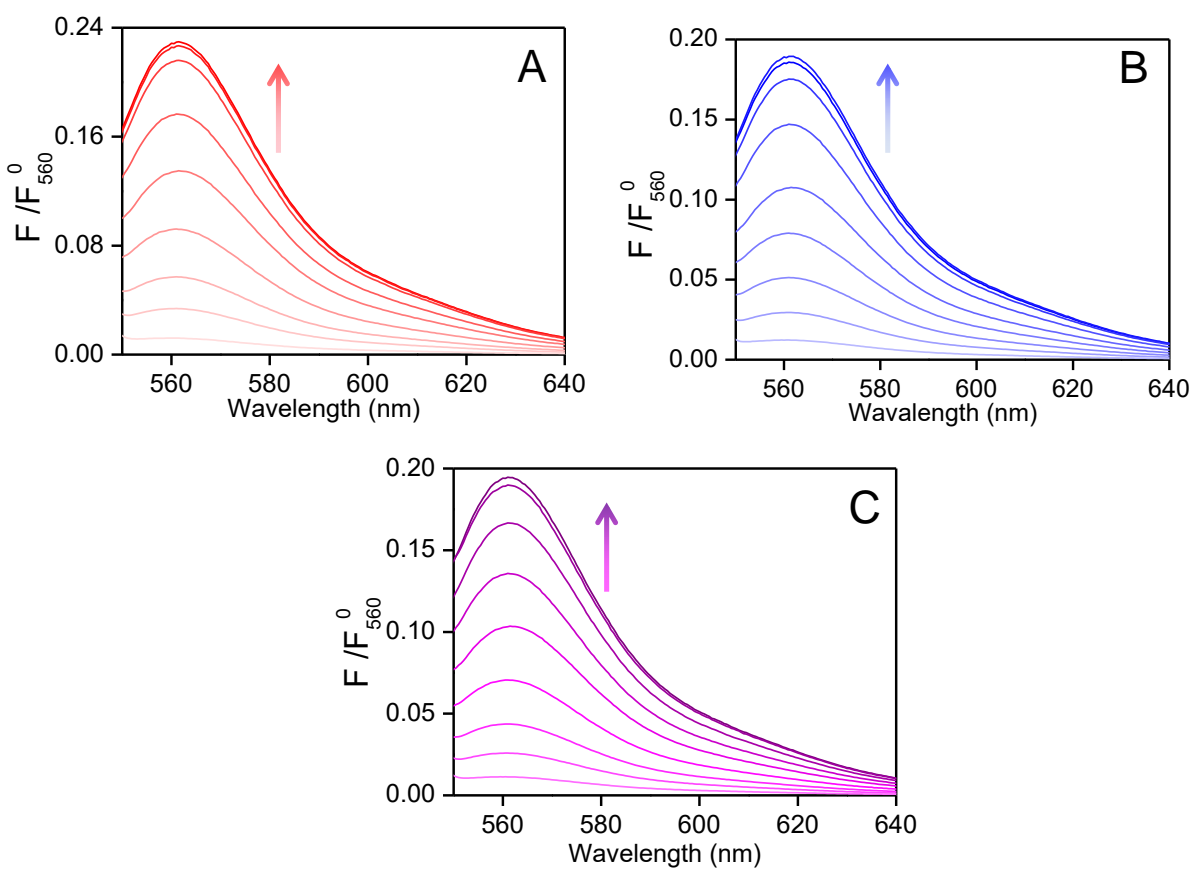


Fig. S10. LUV-concentration dependence (lipid, 0–1.0 mM) of normalized fluorescence spectra of RGG (1 μM) in the presence of LUVs: (A) DMPG, (B) DOPG and (C) POPG LUVs in 2 mM HEPES, pH 7.0 at 25°C. The increases in intensities by increasing the LUV concentrations are indicated in arrows. F represents the fluorescence intensity at pH 7.0, and F_{560}^0 represents F at pH 5.5. Excitation wavelength was 530 nm.

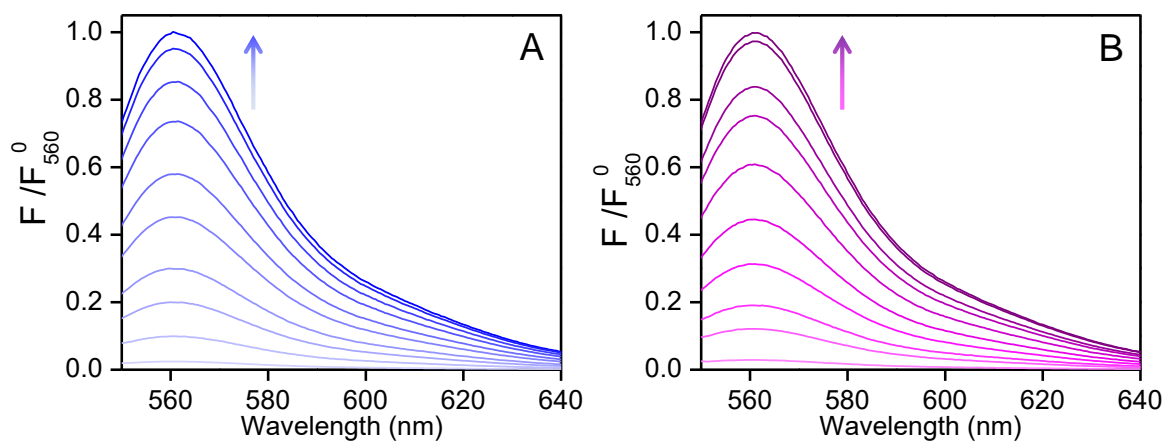


Fig. S11. Normalized fluorescence spectra of RGG (1 μM) in the presence of binding saturated concentration of (A) DOPG and (B) POPG LUV (1 mM) at various pH: 7.50, 7.25, 7.00, 6.80, 6.60, 6.40, 6.20, 6.00, 5.75, 5.50. F represents the fluorescence intensity, and F^0_{560} represents F at pH 5.5. The intensity changes by decreasing pH are shown in arrows. Excitation wavelength was 530 nm.

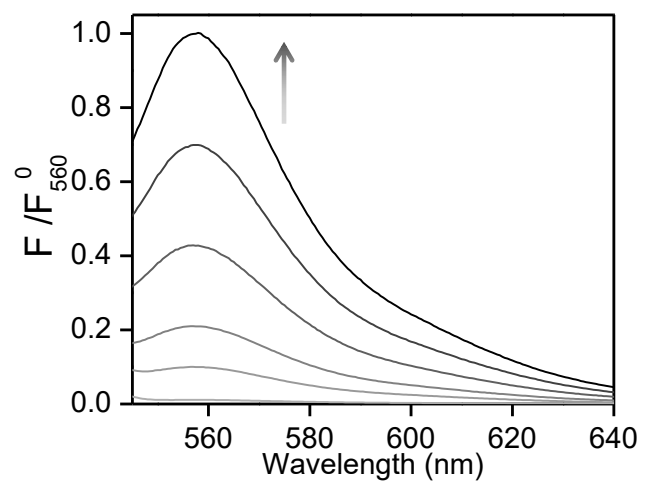


Fig. S12. Normalized fluorescence spectra of RGG (1 μ M) in in 2 mM Britton–Robinson buffer containing 30% (v/v) ethanol at various pH: 4.50, 4.00, 3.80, 3.40, 3.00 and 2.25. F represents the fluorescence intensity, and F_{560}^0 represents F at pH 5.5. The intensity changes by decreasing pH are shown in arrows. Excitation wavelength was 530 nm.

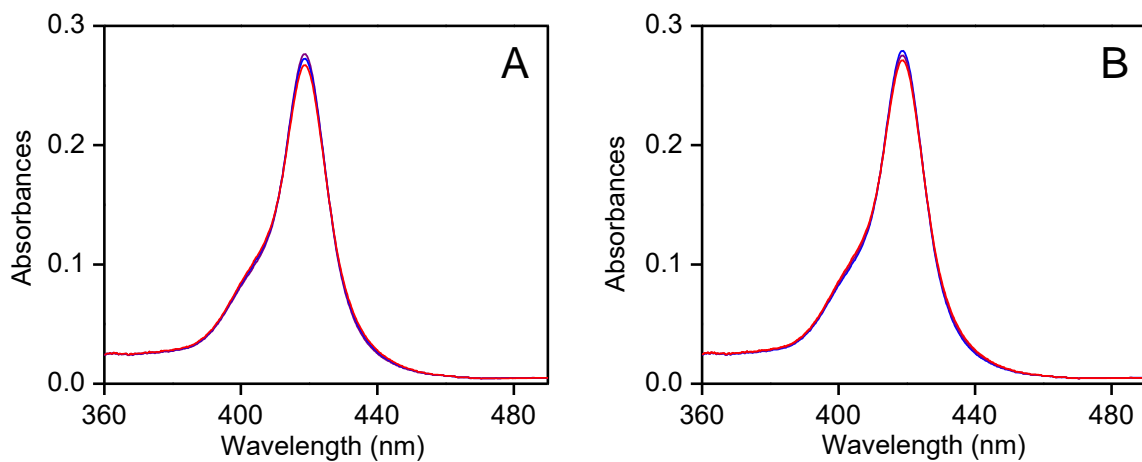


Fig. S13. UV-Vis absorption spectra of GPP (2.5 μ M) in the presence of binding saturated concentration of DMPG (red), DOPG (blue) and POPG (purple) LUV (1 mM) in 2 mM HEPES buffer at different pH values: (A) 7.0 and (B) 6.0 at 25 $^{\circ}$ C.

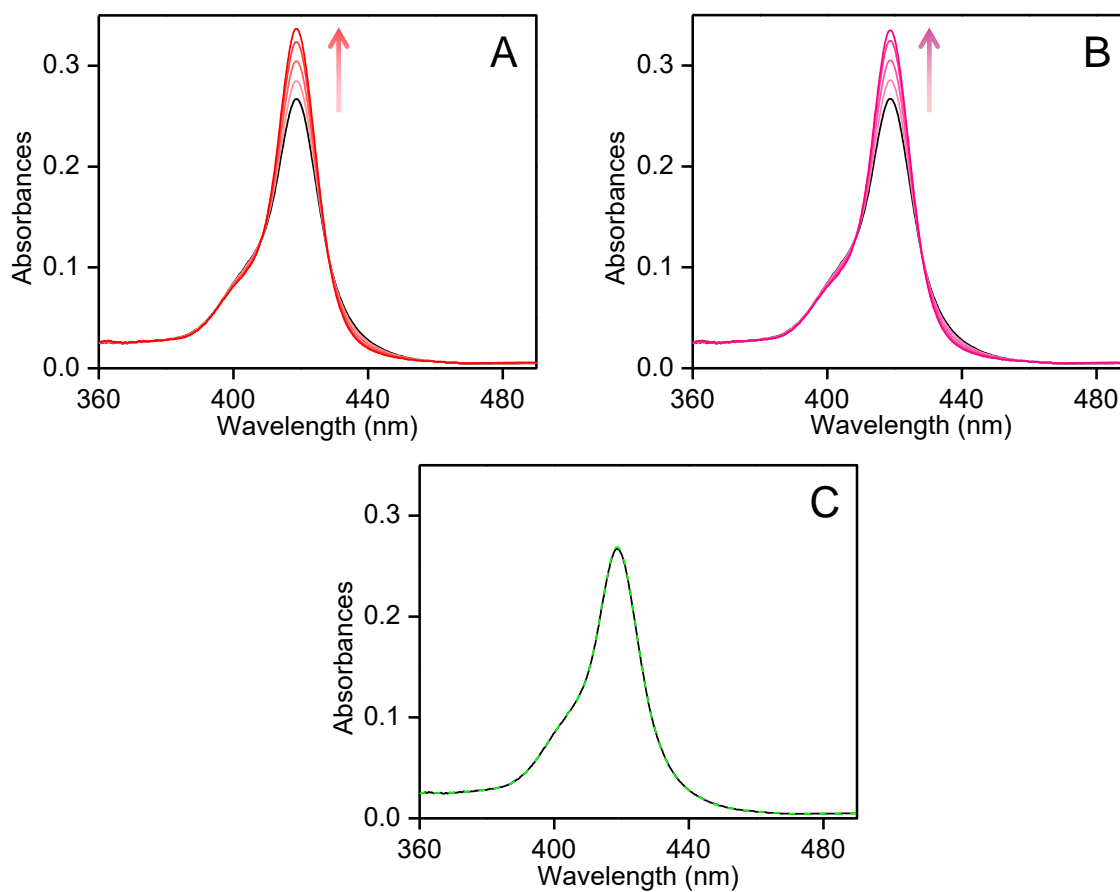


Fig. S14. (A) Ca^{2+} , (B) Mg^{2+} and (C) Na^{+} ion concentration (0.03–0.5 mM for $\text{Ca}^{2+}/\text{Mg}^{2+}$; 1.0 mM for Na^{+}) dependent UV-Vis absorption spectra of GPP (2.5 μM) in the presence of binding saturated concentration of DMPG LUV (1 mM) in 2 mM HEPES buffer, pH values at 25 °C. (A–C) The spectra in the absence of metal ions are depicted by black. (A,B) The intensity changes by increasing metal ion concentration are shown in arrows.

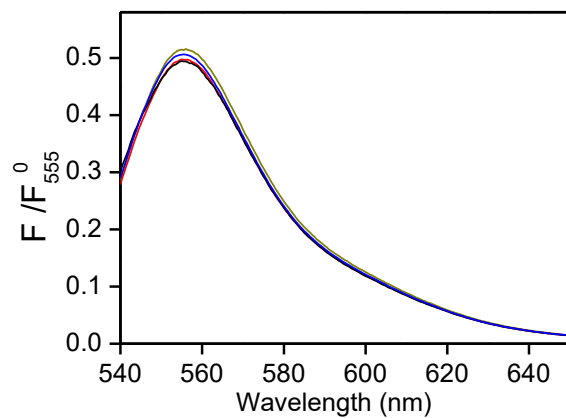


Fig. S15. Normalized fluorescence spectra of RGG (1 μM) in the presence of different bivalent metal ions (0.5 mM) in 2 mM Britton–Robinson buffer, pH 3.7: red, Ca^{2+} ; blue, Mg^{2+} and orange, Zn^{2+} . The spectrum in the absence of metal ions is shown by black. F represents the fluorescence intensity, and F_{555}^0 represents F at pH 2.4. Excitation wavelength was 530 nm.

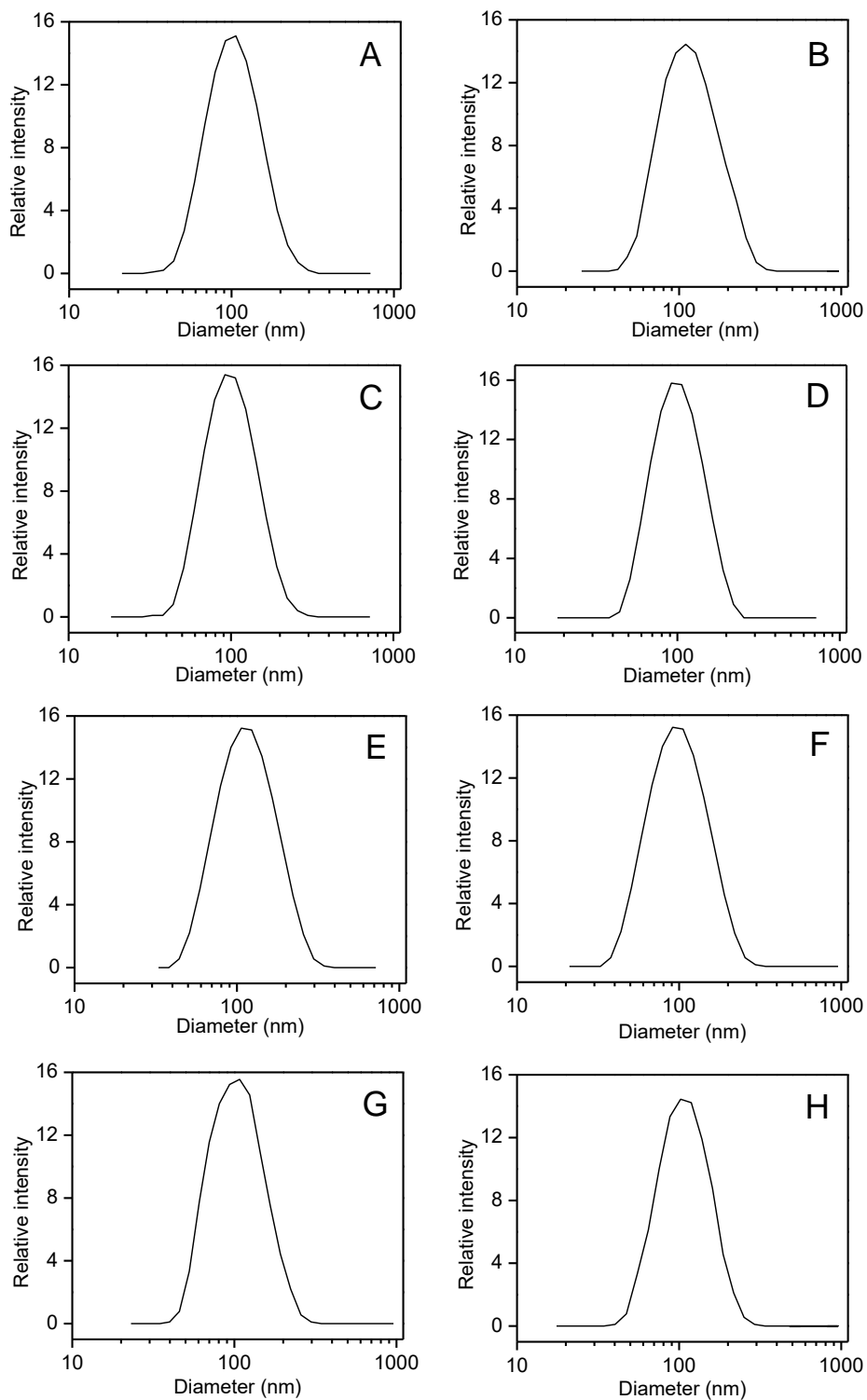


Fig. S16. DLS measurement showing particle size distribution profile of DOPG LUV (1 mM) in the (A,B) absence and (C–H) presence of various mono-valent (1 mM) and bi-valent metal ions (0.3 mM) at pH (A, C–E, H) 7.5 and (B, F, G) 6.0: (C,F) Ca^{2+} , (D,G) Mg^{2+} , (E) Zn^{2+} and (H) Na^{+} . Each of these spectra is an average of 48 scan.

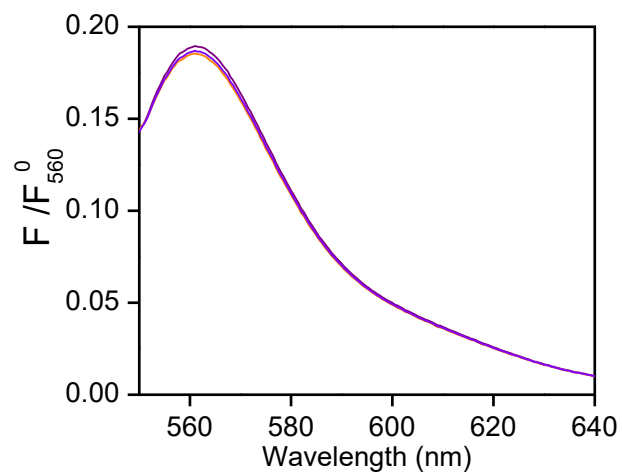


Fig. S17. Normalized fluorescence spectra of RGG (1 μ M) in the presence of binding saturated concentration of POPG LUV (1 mM) containing 1.0 mM NaCl (orange) or KCl (violet) in 2 HEPES, pH 7.0. The spectrum in the absence of salts is shown by purple. F represents the fluorescence intensity, and F_{560}^0 represents F at pH 5.5. Excitation wavelength was 530 nm.