Supporting Information for

Size-Controlled Antimicrobial Peptide Drug Delivery Vehicles through Complex Coacervation

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1. SAXS form and structure factors for the fuzzy-surface complex coacervate model

1.1. Structure factors

In the case of micellar clusters, a structure factor is used to explain the scattering at low Q. The structure factor, $S(Q)_{cluster}$, is described by the following equations¹ (Equation S1-S6).

$$
p = N_{clu} - floor(N_{clu})
$$
(S1)

$$
D = 2 \cdot R_{tot} \cdot f_{dist}
$$
(S2)

$$
S_x = \frac{\sin (Q \cdot D)}{Q \cdot D}
$$
(S3)

$$
S_n = \frac{2}{1 - S_x} - 1 - \frac{2(1 - S_x^{floor(N_{clu})}) \cdot S_x}{floor(N_{clu}) \cdot (1 - S_x)^2}
$$
(S4)

$$
S_{n1} = \frac{2}{1 - S_x} - 1 - \frac{2(1 - S_x^{(floor(N_{clu}) + 1)}) \cdot S_x}{(floor(N_{clu}) + 1) \cdot (1 - S_x)^2}
$$
(S5)

$$
S(Q)_{cluster} = (1 - p) \cdot S_n + p \cdot S_{n1}
$$
 (S6)

In which *Nclu* is the number of cluster units, and *fdist* is the distance correlation of the clusters. To describe the correlation peak at high Q values caused by polyelectrolyte charge correlations in the core of the C3Ms, a pseudo-structure factor is introduced (*S(Q)internal*) (Equation S7).2,3

$$
S(Q)_{internal} = \frac{|C(Q)| \cdot e^{- (Q - Q_{local})^2}}{W \cdot Q_{local} \cdot \sqrt{2\pi}}
$$
 (S7)

In which $C(Q)$ is the fractal scattering, Q_{local} is the structure peak position, and *W* is the relative width of the local Q.

1.2. Form factors

To describe the fuzzy-surface complex coacervate micelle, a form factor for fuzzy spheres was used (*PCoa(Q)*) (Equation S8-S12).4,5

$$
F_0
$$
\n
$$
= \left(\frac{R_{in}}{\sigma_{in}^2} + \frac{1}{\sigma_{in}}\right) \cdot \frac{\cos\left(Q \cdot (R_{in} + \sigma_{in})\right)}{Q^4} - \frac{3 \cdot \sin\left[\frac{\pi}{2}\right]\left(Q \cdot (R_{in} + \sigma_{in})\right)}{Q^5 \cdot \sigma_{in}^2} + \left(\frac{R_{in}^2 \cdot \sigma_{in}^2}{Q^4}\right)\right)
$$
\n
$$
\cdot \frac{\cos\left[\frac{\pi}{2}\right]\left(Q\left(R_{in} - \sigma_{in}\right)\right)}{Q^4}
$$
\n
$$
\tag{S8}
$$

$$
F_1 = \frac{-3 \cdot \sin \left(Q \cdot \left(R_{in} + \sigma_{in}\right)\right) + 6 \cdot \sin \left(Q \cdot R_{in}\right)}{Q^5 \cdot {\sigma_{in}}^2} - \frac{2 \cdot R_{in} \cdot \cos[0](Q \cdot R_{in})}{Q^4 \cdot {\sigma_{in}}^2}
$$
\n
$$
(S9)
$$

$$
V_n = \frac{R_{in}^{3}}{3} + \frac{R_{in} \cdot \sigma_{in}^{2}}{6}
$$
 (S10)

$$
A_{core}(Q) = \frac{F_0 + F_1}{V_n}
$$
 (S11)

$$
P_{Coa}(Q) = A_{core}(Q)^2 \qquad (S12)
$$

The form factor includes spheres with graded interfaces described by the radius *Rin* and width of the interface, σ_{in} . $A_{core}(Q)$ is the scattering amplitude of the fuzzy core, which, to obtain the form factor, is squared. Lastly, the Debye form factor for polymers and polyelectrolytes (*PDebye(Q)*) is given in Equation S13.⁶

$$
P_{Debye}(Q) = \frac{2(e^{-Q^2 \cdot R_g^2} - 1 + Q^2 \cdot R_g^2)}{(Q^2 \cdot R_g^2)^2}
$$
 (S13)

In which the R_g is the radius of gyration of the polymer/polyelectrolyte. From the fits, we can calculate some of the features of the C3Ms. We start with the average molecular weight of the complex coacervate micelle (M_w) in Equation S14.

 M_w

$$
= P \cdot \left(f_{mix0poly1} \cdot \left(M_{PMAA_1} + M_{PEO_1} \right) + f_{mix0poly2} \cdot \left(M_{PMAA_2} + M_{PEO_2} \right) + \left(1 - \frac{1}{2} \right) \right)
$$
\n
$$
(S14)
$$

Where P is the aggregation number (number of chains per micelle), $f_{mix0polyl}$ is the molar fraction of polymer 1 in the micelles, *MPMAA1* is the molecular weight of the PMAA block of polymer

1, *MPEO1* is the molecular weight of the PEO block of polymer 1, *fmix0poly2* is the molar fraction of polymer 2 in the micelles, *MPMAA2* is the molecular weight of the PMAA block of polymer 2, *MPEO2* is the molecular weight of the PEO block of polymer 2, *fmix0* is the total molar fraction of polymer in the micelles, *MCol0* is the molecular weight of colistin without the sulfate salt. Furthermore, we can also calculate the fraction of water inside the core with the following equations (S15-S17). We start with the dry volumes of the PMAA and colistin core (V_{cp}) , PEO (V_{sp}) and the total dry volume (V_{tot}) :

$$
M_{PMAA_1}
$$
\n
$$
V_{cp} = \frac{(1 - f_{mix0}) \cdot \frac{M_{Col0}}{d_{Col}} + f_{mix0poly1} \cdot \frac{M_{PMAA_1}}{d_{PMAA}} + f_{mix0poly2} \cdot \frac{M_{PMAA_2}}{d_{PMAA_2}}}{M_{H_{M_{M_{M_{M}}}}}
$$
\n
$$
M_{PEO_1}
$$
\n
$$
M_{PEO_2}
$$
\n
$$
V_{sp} = \frac{f_{mix0poly1} \cdot \frac{M_{PEO_1}}{d_{PEO}} + f_{mix0poly2} \cdot \frac{M_{PEO_2}}{d_{PEO_2}}}{M_{A}}
$$
\n
$$
(S16)
$$

$$
V_{tot} = V_{cp} + V_{sp} \tag{S17}
$$

Where d_{Col} is the density of colistin, d_{PMAA} is the density of the PMAA block, d_{PEO} is the density of the PEO block, and *N^A* is Avogadro's number. Now we can calculate the wet volume of the micelle and determine the water fraction in the micelle through Equations S18 and S19.

$$
V_c = 4\pi \cdot \left(\frac{R_{in}}{3} + \frac{R_{in} \cdot \sigma_{in}^2}{6}\right)_{\text{(S18)}}
$$

$$
f_w = 1 - \left(\frac{V_{tot}}{V_c}\right)_{\text{(S19)}}
$$

Where R_{in} is the core radius of the micelle and σ_{in} is its density distribution, both obtained from the fits. For a more detailed explanation of the model features and mass balance calculations, we refer to a previous publication.³ A summary of the parameters used and their categorisation (fixed, calculated, experimental, or fitting parameters) is given in Table S1.

Model	Category	fixed Values for	Description
parameters / f_{+}		parameters	
M_{Col} (Da)	Fixed	1405.65	Molecular weight colistin sulfate
$\Delta \rho_{Col}$ (cm ⁻²)	Fixed	$1.24 \cdot 10^{11}$	SLD colistin
$\Delta\rho_{PMAA}$ (cm ⁻²)	Fixed	$1.09 \cdot 10^{11}$	SLD colistin
$\Delta\rho_{PEO}$ (cm ⁻²)	Fixed	$1.11 \cdot 10^{11}$	SLD PEO block
$\Delta\rho_{Core}$ (cm ⁻²)	Calculated		SLD average in core
$\Delta\rho_{Solvent}$ (cm ⁻²)	Fixed	$9.43 \cdot 10^{10}$	SLD solvent buffer

Table S1. Summary of parameters used in the fitting model.

2. Results and Discussion

2.1. SAXS fitting parameters of the binary polymer-colistin C3Ms

In Table S2, S3, and S4, the final states of the three different PEO-b-PMAA-colistin C3Ms are compared at 0.50 wt%, 0.25 wt% and 0.125 wt%, respectively.

Table S2. The most important fit parameters for PEO₄₅-b-PMAA₄₁-colistin C3Ms at 0.50 wt%, 0.25 wt% and 0.125 wt% were obtained from the fuzzy-surface complex coacervate model.

[a] It was not possible to extract a reliable width parameter for these systems, most likely due to the increased polydispersity. The size is still extracted well based on the *Rin* parameter.

Table S4. The most important fit parameters for PEO₁₁₄-b-PMAA₈₁-colistin C3Ms at 0.50 wt%, 0.25 wt% and 0.125 wt% were obtained from the fuzzy-surface complex coacervate model.

2.2. Swelling ternary coacervate systems

[[]a] It was not possible to extract a reliable width parameter for these systems, most likely due to the increased polydispersity. The size is still extracted well based on the *Rin* parameter.

In Figure S1, the SAXS patterns and fits of the ternary complex coacervate systems T5, T6, T7, and T8 are shown at 0.125 wt%, including their *Rtot* fit parameter (from the fuzzy-surface complex coacervate model) for the complete polymer molar fraction set $(0 \le \Psi \le 1)$. In Figure S2, T3 and T4 ternary coacervate systems are shown to not swell at the complete polymer molar fraction set based on the fits from the fuzzy-surface complex coacervate model. Lastly, in Figure S3, the *PDI* behaviour (from the fuzzy-surface complex coacervate model) for the complete polymer molar fraction set is shown for T5, T6, T7, and T8 complex coacervate systems.

Figure S1. SAXS patterns of colistin-C3Ms formed from four different ternary coacervate formulations at 0.125 wt%. C3Ms were prepared from a combination of colistin and two other polymers: $PMAA_{46}/PEO_{45}$ -b-PMA A_{81} (T5, grey symbols, A), $PMAA_{46}/PEO_{114}$ -b- $PMAA_{81}$ (T6, purple symbols, B), PEO_{45} -b- $PMAA_{15}/PEO_{45}$ -b-PMAA₈₁ (T7, orange symbols, C), and PEO₄₅-b-PMAA₁₅/PEO₁₁₄-b-PMAA₈₁ (T8, yellow symbols, D). The

complete polymer molar fraction set $0 \le \Psi \le 1$ was investigated and fitted with the fuzzy-surface C3M model, while for visibility reasons only a couple of SAXS patterns are shown. The fit parameter R_{tot} is shown for T5, T6, T7, and T8 along the complete polymer molar fraction set (E).

Figure S2. From the fuzzy-surface C3M model fits, the C3M water content (f_w) was plotted against the molar fraction of PEO₄₅-b-PMAA₄₁ (T3, red squares and T4, green circles).

Figure S3. Polydispersity index (*PDI*) behaviour for T5 (grey squares), T6 (purple circles), T7 (orange triangles), and T8 (yellow rhombuses) ternary colistin complex coacervate systems for the complete polymer molar fraction set $(0 \le \Psi \le 1)$. From the fuzzy-surface C3M model fits, the *PDI* values were plotted against the molar fraction of either $PEO₄₅$ -b-PMAA₁₅ or PMAA₄₆.

2.3. Formation kinetics of ternary coacervate systems

In Figure S4, the power law behaviour of T1-T4 ternary coacervate systems for *Rtot* and *P* (from the fuzzy-surface complex coacervate model) is shown from the kinetic growth process over time, analysed with a TR-SAXS setup at 0.125 wt% and Ψ = 0.50.

Figure S4. Power law relation of R_{tot} and P on log-log scale for colistin-C3Ms from T1 (pink squares), T2 (blue circles), T3 (red triangles) and T4 (green rhombuses) analysed through TR-SAXS, including fits from the fuzzysurface complex coacervate model. 1/3 fits for the first two are added to visualise the power law of equal density growth. We observe some density changes in the growth process of complex coacervates.

3. References

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