## Enhancing Surface/Interface Activity and Wettability via Trimeric

## **Surfactant-Containing Mixtures**

Chao Zhang<sup>a</sup>, Jianlin Jiang<sup>a</sup>, Lili Zhou<sup>b\*</sup>, Bin Qin<sup>a\*</sup> and Fulin Qiao<sup>a\*</sup>

<sup>a</sup>Sinopec research institute of petroleum processing CO. LTD.

<sup>b</sup>School of Light Industry Science and Engineering, Beijing Technology and Business University, Beijing 100048, P. R. China



Figure S1. Synthetic procedure of Citric-3C12.

## Material.

*N*,*N*-Dimethyl-1,2-ethanediamine, trimethyl citrate and 1-bromododecane were obtained from Innochem and used without further purification. CH<sub>3</sub>OH, CH<sub>3</sub>COCH<sub>3</sub>, and all organic solvents were purchased from Beijing Chemical Co. Milli-Q (18.2 M $\Omega$  M $\Omega$ ·cm) water was used in all experiments.

## Synthesis.

Compound **Citric-3C12** was synthesized as outlined in **Figure S1** and was characterized by <sup>1</sup>H NMR, mass spectrum. 2.34 g trimethyl citrate was dispersed in

20 mL N,N-Dimethyl-1,2-ethanediamine and stirred at 100  $^\circ\!\mathrm{C}$  for 4 h. The N,N-

Dimethyl-1,2-ethanediamine was then removed under reduced pressure, and the residue was recrystallized three times from methanol/ethyl acetate to afford intermediate **A**.

Intermediate A (4 g) and 1-bromododecane (8.9 g) were added to 100 mL of CH<sub>3</sub>OH/

CH<sub>3</sub>COCH<sub>3</sub> (v: v = 1:1) and heated at 45 °C for 72 h. The solvent was removed under

reduced pressure, and the residue was recrystallized three times from methanol/ethyl acetate, yielding **Citric-3C12** as white solid in 41% yield. <sup>1</sup>H NMR (CD<sub>3</sub>OD, 400 MHz):  $\delta$  = 0.88 (m, 9H), 1.12—1.39 (m, 46H), 1.80 (s, 8H), 2.52-2.64 (m, 4H), 2.71—2.82(m, 4H), 3.00—3.06 (m, 5H), 3.06—3.17 (m, 12H), 3.17—3.29 (m, 4H), 3.28—3.50 (m, 12H), 3.51—3.69 (m, 8H). MS-ESI (m/z): calculated: 1146.64; found: 495.40 ([M-2Br]<sup>2+</sup>/2). FTIR (KBr, cm<sup>-1</sup>): 3225, 2907, 2827, 1648, 1458, 1349, 1228,1073, 1015, 969, 906, 739.



Figure S2. <sup>1</sup>H NMR spectra of **Citric-3C12** in  $D_2O$ .



Figure S3. FT-IR spectra of Citric-3C12.

**Table S1.** Values of the CMC, interaction parameters ( $\beta^{\sigma}$ ,  $\beta^{m}$ ), interfacial composition ( $X^{\sigma}$ ), micellar composition ( $X^{m}$ ) Surface Excess ( $\Gamma_{max}$ ), and Area Per Molecule ( $A_{min}$ ) for

the Citric-3C12/9-EOS	mixture in aqueous	solution at 2	2 <b>5.0</b> ℃
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X <sub>Citric-3C12</sub>	CMC (mM)	pC20	X <sup>m</sup>	₿ <sup>m</sup>	Χσ	βσ	Γ×10 <sup>-6</sup> (mol m <sup>-</sup> ²)	A(Ų/mole cule)	A <sub>ideal</sub> (Ų/mol ecule)
1	0.25	1.1	-	-	-	-	1.699	97.763	-
0.9	0.052	1.9	0.491	-3.721	0.42	-4.012	1.835	90.526	97.847
0.8	0.038	2.2	0.419	-3.665	0.365	-4.431	2.015	82.418	97.855
0.7	0.022	2.4	0.398	-5.148	0.383	-7.605	2.303	72.104	97.889
0.6	0.018	2.5	0.38	-5.656	0.362	-7.678	2.031	81.778	97.855
0.5	0.00455	2.6	0.334	-8.123	0.323	-6.773	1.855	89.543	97.861
0.4	0.018	2.8	0.309	-4.969	0.404	-11.050	1.683	98.679	97.849
0.3	0.018	3.1	0.279	-4.922	0.334	-9.764	1.208	137.563	97.859
0.2	0.017	2.8	0.256	-5.302	0.292	-8.746	1.713	96.9632	97.866
0.1	0.012	2.8	0.275	-7.702	0.277	-9.694	1.574	105.547	97.868
0	0.022	2.4	-	-	-	-	1.697	97.914	-



Figure S4. Contact angle photographs of water on oil-wet silica gel surfaces



**Figure S5.** Contact angle photographs of oil-wet silica gel surfaces treated with **Citric-3C12/9-EOS** at at  $X_{\text{Citric-3C12}} = 1.0, 0.7, 0.5, 0.3 \text{ and } 0.$