

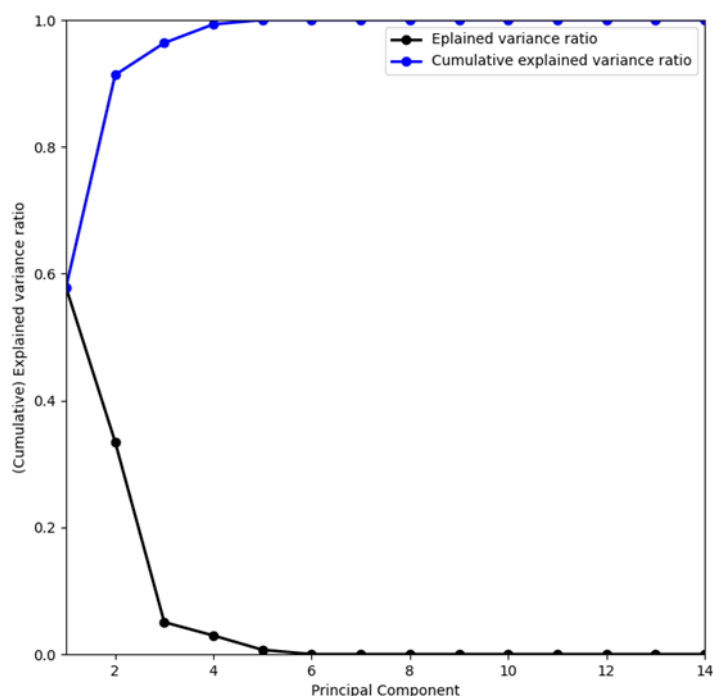
Enhancing the performance of heterogeneous palladium based catalysts in the mild reductive depolymerization of Soda lignin through addition of a non-noble metal and tuning of the preparation strategy.

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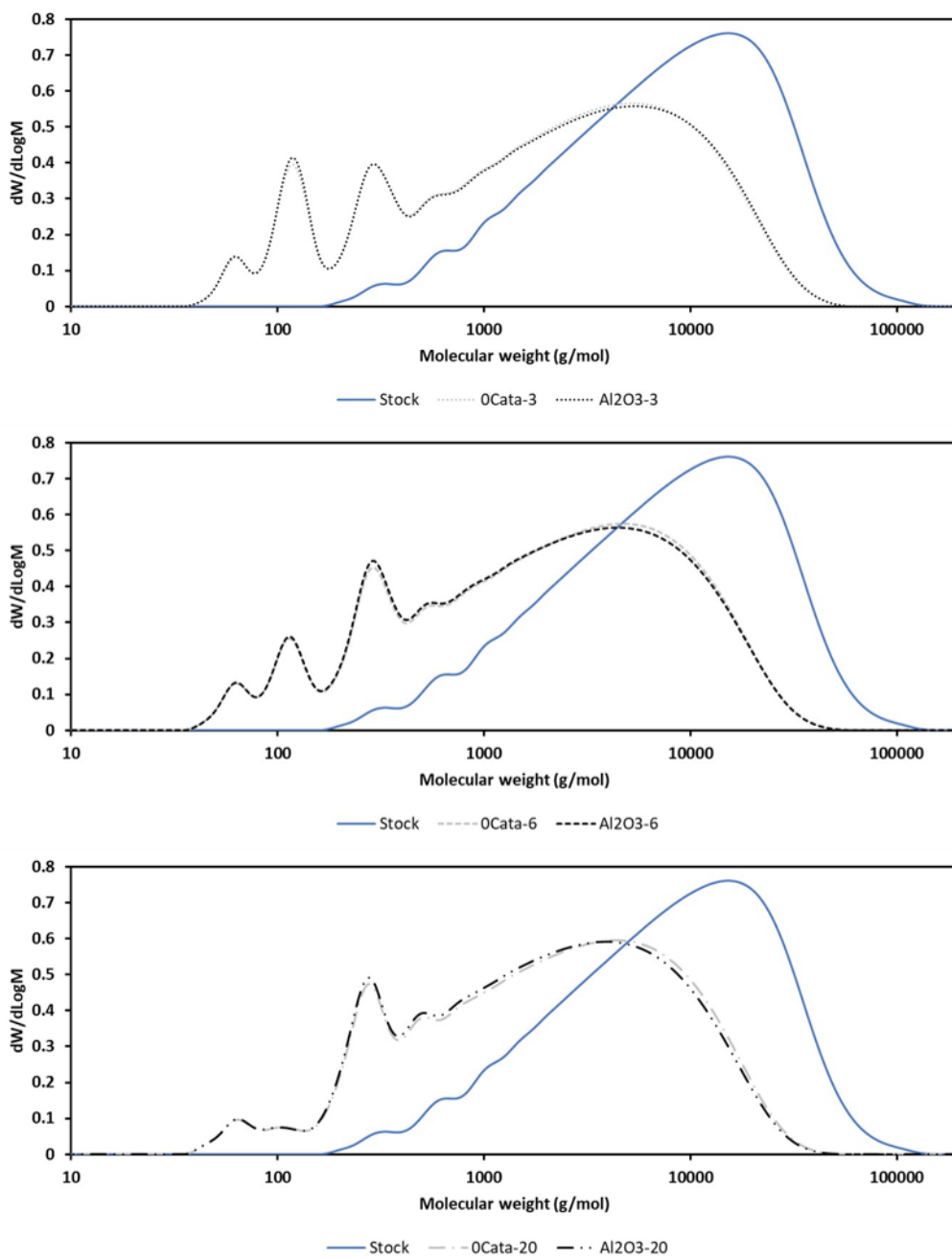
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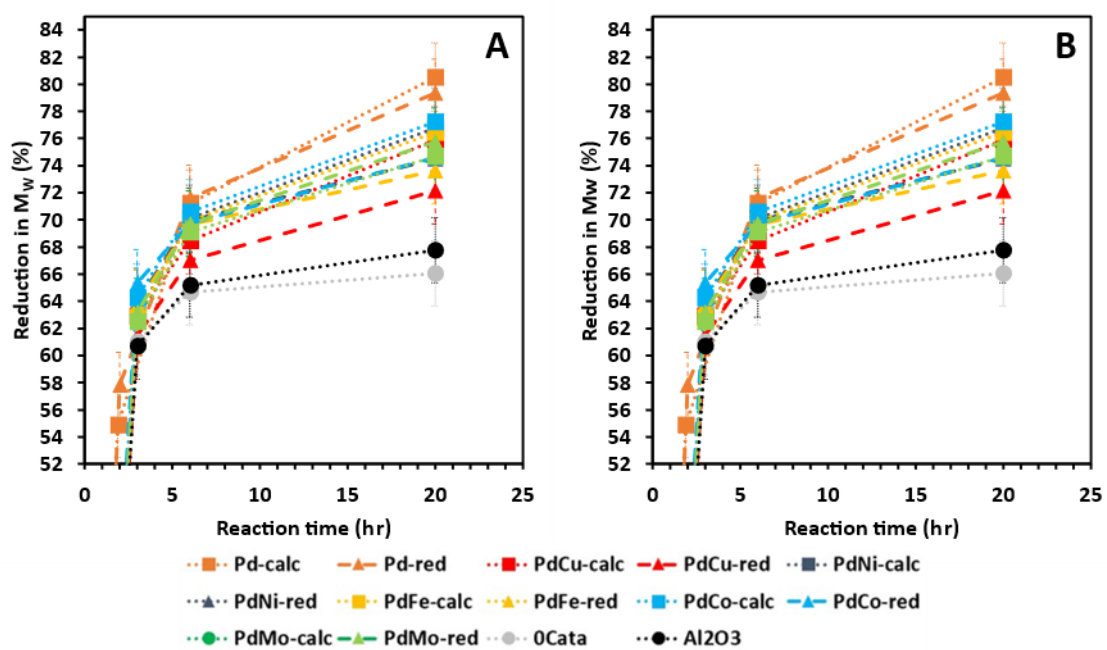
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**Figure S1:** Scree plot representing the explained variance ratio (black) and the cumulated explained variance ratio (blue) as a function of the number of principal components.



**Figure S2:** Molecular weight distributions for OCata (grey) and Al<sub>2</sub>O<sub>3</sub> (black) compared to the lignin stock solution (blue) after 3 h (top), 6 h (middle) and 20 h (bottom) of reaction time.



**Figure S3:** Reduction in Mw (A) and increase in PDI (B) as a function of reaction time for 0Cata, Al<sub>2</sub>O<sub>3</sub> and all Pd(X)-calc/red variations.

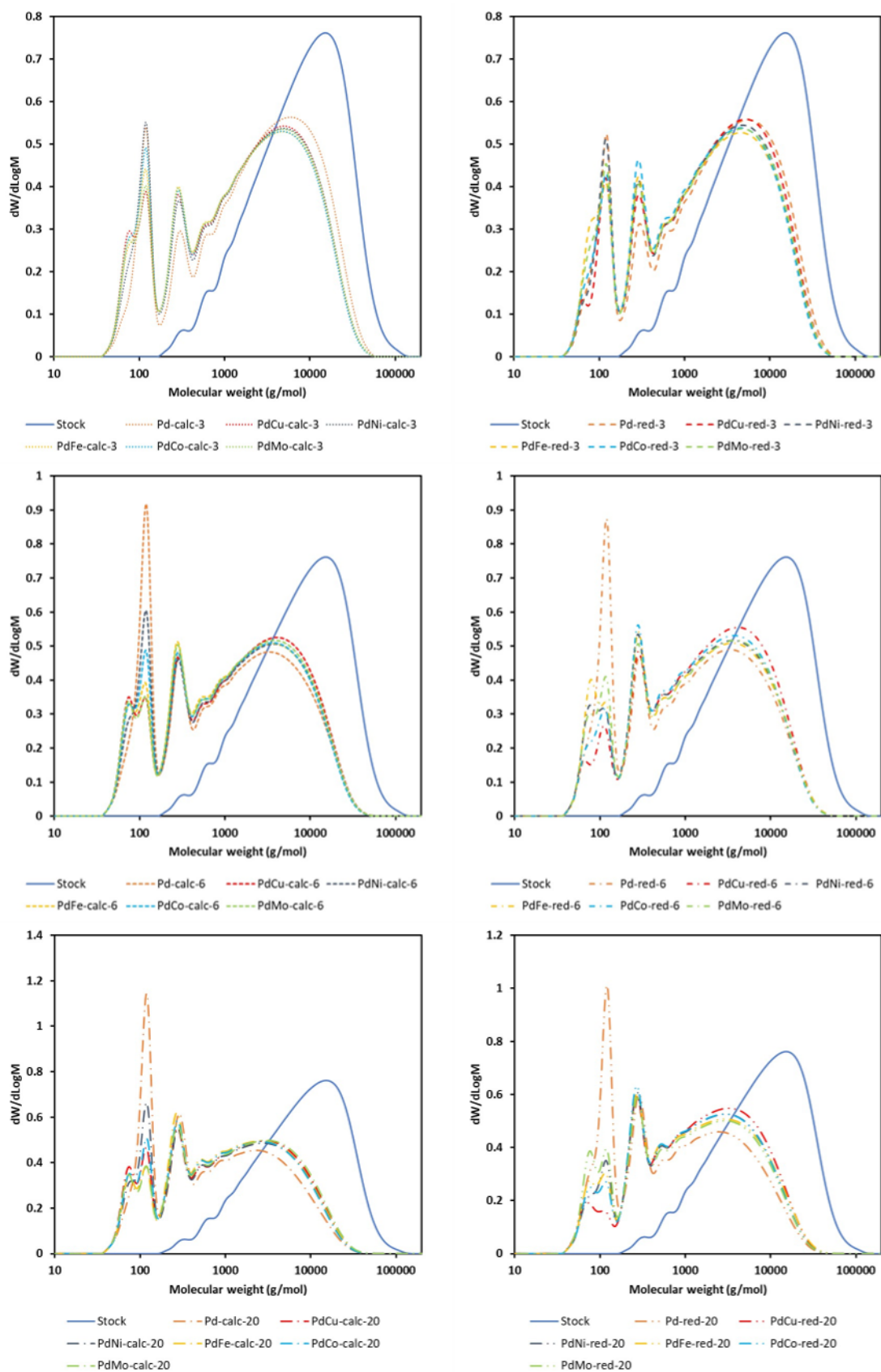
**Table S1:** reaction time, batch time, reduction in Mw (with numerical Mw values), Mw reduction rate, increase in PDI (with numerical PDI values) and PDI increase rate for all performed experiments.

Sample	Reaction time (h)	Batch time (mmol Pd*s)	Reduction in Mw (%) (Mw (g/mol))	Mw reduction rate %/(mmol Pd*s)	Increase in PDI (%) (PDI)	PDI increase rate %/(mmol Pd*s)
Stock solution	0	0	0 (12938)	Not applicable	0% (4)	Not applicable
OCata	3	Not applicable	61.03 +- 2.45 (5042)	Not applicable	131.75 +- 11.27 (9.27)	Not applicable
	6	Not applicable	64.66 +- 2.38 (4572)	Not applicable	96.75 +- 11.84 (7.87)	Not applicable
	20	Not applicable	66.05 +- 2.39 (4392)	Not applicable	66 +- 9.08 (6.64)	Not applicable
Al2O3	3	Not applicable	60.68 +- 2.45 (5087)	Not applicable	137.25 +- 11.66 (9.49)	Not applicable
	6	Not applicable	65.19 +- 2.38 (4504)	Not applicable	97.25 +- 11.89 (7.89)	Not applicable
	20	Not applicable	67.75 +- 2.4 (4172)	Not applicable	64 +- 8.88 (6.56)	Not applicable
Pd-calc	2	100.98	54.89 +- 2.41 (5836)	0.544 +- 0.024	168.25 +- 13.86 (10.73)	1.666 +- 0.137
	6	322.72	71.22 +- 2.44 (3724)	0.221 +- 0.008	160 +- 16.65 (10.4)	0.496 +- 0.052
	20	1075.74	80.52 +- 2.53 (2520)	0.075 +- 0.002	110 +- 12.92 (8.4)	0.102 +- 0.012
Pd-red	2	104.53	57.82 +- 2.43 (5457)	0.553 +- 0.023	168.5 +- 13.88 (10.74)	1.612 +- 0.133
	6	313.59	71.6 +- 2.44 (3674)	0.228 +- 0.008	155.75 +- 16.35 (10.23)	0.497 +- 0.052
	20	1045.29	79.33 +- 2.51 (2674)	0.076 +- 0.002	117.75 +- 13.53 (8.71)	0.113 +- 0.013
PdCu-calc	3	99.45	62.91 +- 2.47 (4799)	0.633 +- 0.025	167.25 +- 13.79 (10.69)	1.682 +- 0.139
	6	198.91	68.44 +- 2.41 (4083)	0.344 +- 0.012	161 +- 16.72 (10.44)	0.809 +- 0.084
	20	663.03	75.88 +- 2.48 (3121)	0.114 +- 0.004	126.75 +- 14.22 (9.07)	0.191 +- 0.021
PdCu-red	3	110.11	61.48 +- 2.46 (4984)	0.558 +- 0.022	140.5 +- 11.89 (9.62)	1.276 +- 0.108
	6	220.22	67.04 +- 2.4 (4265)	0.304 +- 0.011	106.75 +- 12.66 (8.27)	0.485 +- 0.057
	20	734.07	72.13 +- 2.44 (3606)	0.098 +- 0.003	80.25 +- 10.42 (7.21)	0.109 +- 0.014
PdNi-calc	3	120.77	62.87 +- 2.47 (4804)	0.521 +- 0.02	165 +- 13.63 (10.6)	1.366 +- 0.113
	6	241.53	70.12 +- 2.43 (3866)	0.29 +- 0.01	149 +- 15.86 (9.96)	0.617 +- 0.066
	20	805.11	76.76 +- 2.49 (3007)	0.095 +- 0.003	115.5 +- 13.35 (8.62)	0.143 +- 0.017
PdNi-red	3	94.38	63.94 +- 2.48 (4665)	0.678 +- 0.026	145 +- 12.21 (9.8)	1.536 +- 0.129
	6	188.76	69.93 +- 2.42 (3891)	0.37 +- 0.013	132.25 +- 14.63 (9.29)	0.701 +- 0.078
	20	629.21	74.52 +- 2.47 (3296)	0.118 +- 0.004	89.5 +- 11.23 (7.58)	0.142 +- 0.018
PdFe-calc	3	188.76	62.84 +- 2.47 (4808)	0.673 +- 0.026	167.75 +- 13.83 (10.71)	1.797 +- 0.148

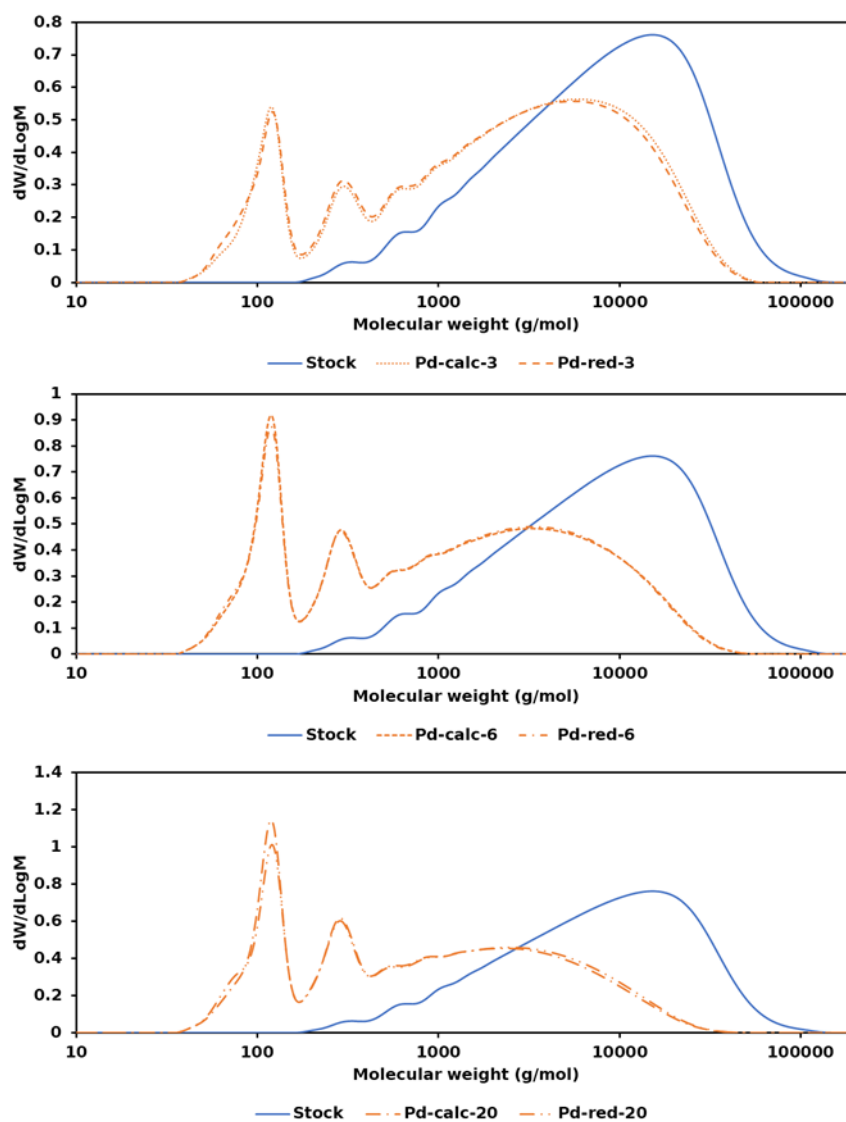
	6	93.37	69.95 +- 2.42 (3888)	0.375 +- 0.013	142.5 +- 15.39 (9.7)	0.763 +- 0.082
	20	622.44	76.43 +- 2.48 (3049)	0.123 +- 0.004	102.75 +- 12.34 (8.11)	0.165 +- 0.02
PdFe-red	3	106.56	63.8 +- 2.48 (4683)	0.599 +- 0.023	165.5 +- 13.67 (10.62)	1.553 +- 0.128
	6	213.12	69.67 +- 2.42 (3924)	0.327 +- 0.011	144.75 +- 15.55 (9.79)	0.679 +- 0.073
	20	710.39	73.64 +- 2.46 (3411)	0.104 +- 0.003	101.5 +- 12.24 (8.06)	0.143 +- 0.017
	3	94.38	64.26 +- 2.48 (4624)	0.681 +- 0.026	167 +- 13.78 (10.68)	1.769 +- 0.146
PdCo-calc	6	188.76	70.57 +- 2.43 (3808)	0.374 +- 0.013	144 +- 15.5 (9.76)	0.763 +- 0.082
	20	629.21	77.25 +- 2.49 (2944)	0.123 +- 0.004	111.5 +- 13.04 (8.46)	0.177 +- 0.021
	3	83.22	65.34 +- 2.49 (4484)	0.785 +- 0.03	139.5 +- 11.82 (9.58)	1.676 +- 0.142
PdCo-red	6	166.43	69.74 +- 2.42 (3915)	0.419 +- 0.015	112.25 +- 13.1 (8.49)	0.674 +- 0.079
	20	554.78	74.6 +- 2.47 (3286)	0.134 +- 0.004	86.25 +- 10.95 (7.45)	0.155 +- 0.02
	3	96.92	62.5 +- 2.47 (4852)	0.645 +- 0.025	167.75 +- 13.83 (10.71)	1.731 +- 0.143
PdMo-calc	6	193.84	69.1 +- 2.42 (3998)	0.356 +- 0.012	143.75 +- 15.48 (9.75)	0.742 +- 0.08
	20	646.12	74.68 +- 2.47 (3276)	0.116 +- 0.004	116 +- 13.39 (8.64)	0.18 +- 0.021
	3	92.35	63.2 +- 2.47 (4761)	0.684 +- 0.027	162.25 +- 13.44 (10.49)	1.757 +- 0.146
PdMo-red	6	184.70	69.72 +- 2.42 (3918)	0.377 +- 0.013	139 +- 15.13 (9.56)	0.753 +- 0.082
	20	615.67	75.76 +- 2.48 (3136)	0.123 +- 0.004	112.5 +- 13.12 (8.5)	0.183 +- 0.021
	Cu-calc	Not applicable	0% (12938)	Not applicable	0% (4)	Not applicable
Ni-calc	Not applicable	61.03 +- 2.45 (5042)	Not applicable	131.75 +- 11.27 (9.27)	Not applicable	
Fe-calc	Not applicable	64.66 +- 2.38 (4572)	Not applicable	96.75 +- 11.84 (7.87)	Not applicable	
Co-calc	Not applicable	66.05 +- 2.39 (4392)	Not applicable	66 +- 9.08 (6.64)	Not applicable	
Mo-calc	Not applicable	60.68 +- 2.45 (5087)	Not applicable	137.25 +- 11.66 (9.49)	Not applicable	
Cu-red	Not applicable	65.19 +- 2.38 (4504)	Not applicable	97.25 +- 11.89 (7.89)	Not applicable	
Ni-red	Not applicable	67.75 +- 2.4 (4172)	Not applicable	64 +- 8.88 (6.56)	Not applicable	
Fe-red	Not applicable	54.89 +- 2.41 (5836)	Not applicable	168.25 +- 13.86 (10.73)	Not applicable	
Co-red	Not applicable	71.22 +- 2.44 (3724)	Not applicable	160 +- 16.65 (10.4)	Not applicable	
Mo-red	Not applicable	80.52 +- 2.53 (2520)	Not applicable	110 +- 12.92 (8.4)	Not applicable	

**Table S2:** Concentration, UV/VIS area and RID area for lignin monomer model compounds after injection onto GPC-RID-UV/VIS.

compound	compound CAS	Conc. (mg/L)	UV/VIS area (a.u.)	RID area (a.u.)
Phenol	108-95-2	100	4106.0	3233.8
Para cresol	106-44-5	100	3330.5	6189.9
Guaiacol	90-05-1	100	4128.4	4842.3
Vanillin	121-33-5	100	11284.3	12940.7
Eugenol	97-53-0	100	3287.4	9524.8
Isoeugenol	97-54-1	100	9856.7	15794.3
3-(4-hydroxyphenyl)-1-propanol	10210-17-0	100	2186.9	9045.3
Coniferyl alcohol	458-35-5	100	6668.7	11586.4
Sinapyl alcohol	20675-96-1	100	7432.9	10076.1
Trans ferulic acid	537-98-4	100	7309.8	15987.9
4-Hydroxy-3-methoxycinnamaldehyde	458-36-6	100	3170.8	18586.2
Vanillin alcohol	498-00-0	100	2857.2	10741.5
<b>AVG</b>		100	5468.3	10712.4
<b>%RSD (STDEV.S)</b>		100	54.5%	43.3%

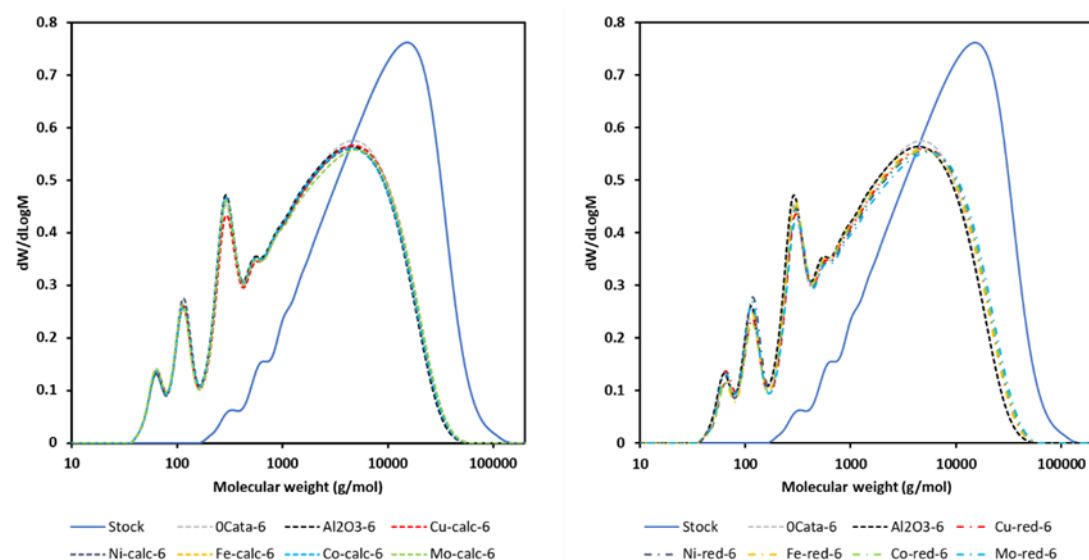


**Figure S4:** Molecular weight distributions, as determined through GPC-RID, for the calcined (left column) and reduced (right column) catalysts at 3 h (top row), 6 h (middle row) and 20 h (bottom row) reaction times.

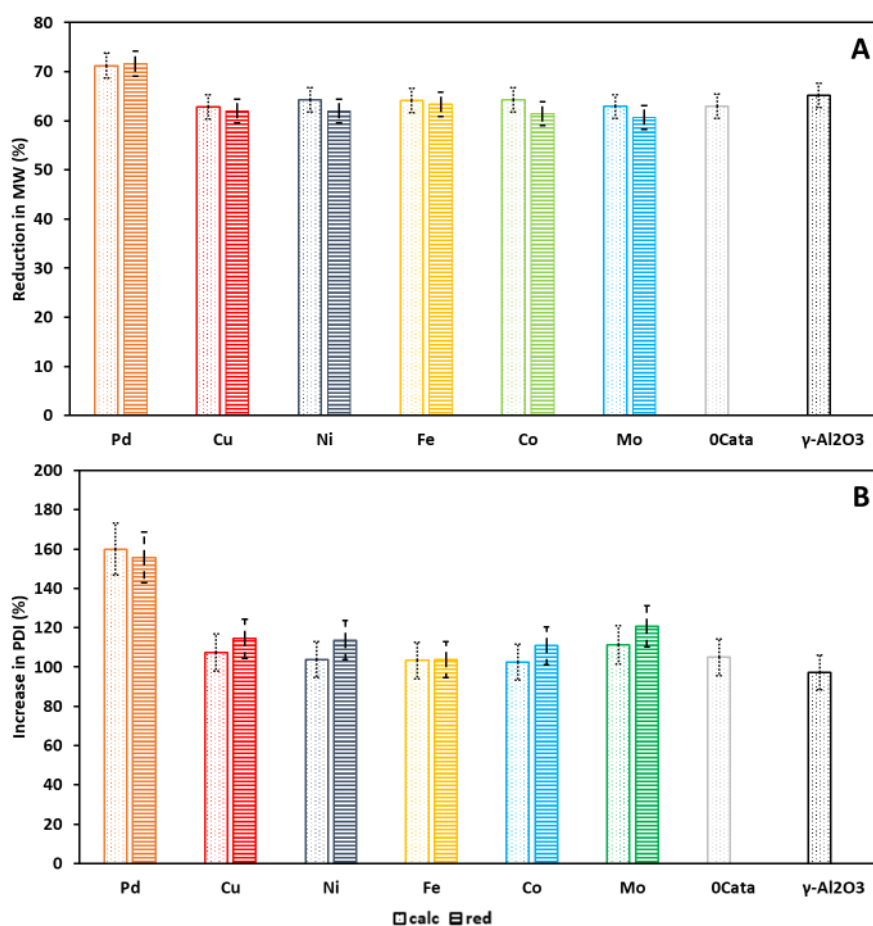


**Figure S5:** Molecular weight distribution, as determined through GPC-RID, for Pd-calc and Pd-red after 2 h (left), 6 h (middle) and 20 h (right) of reaction time.

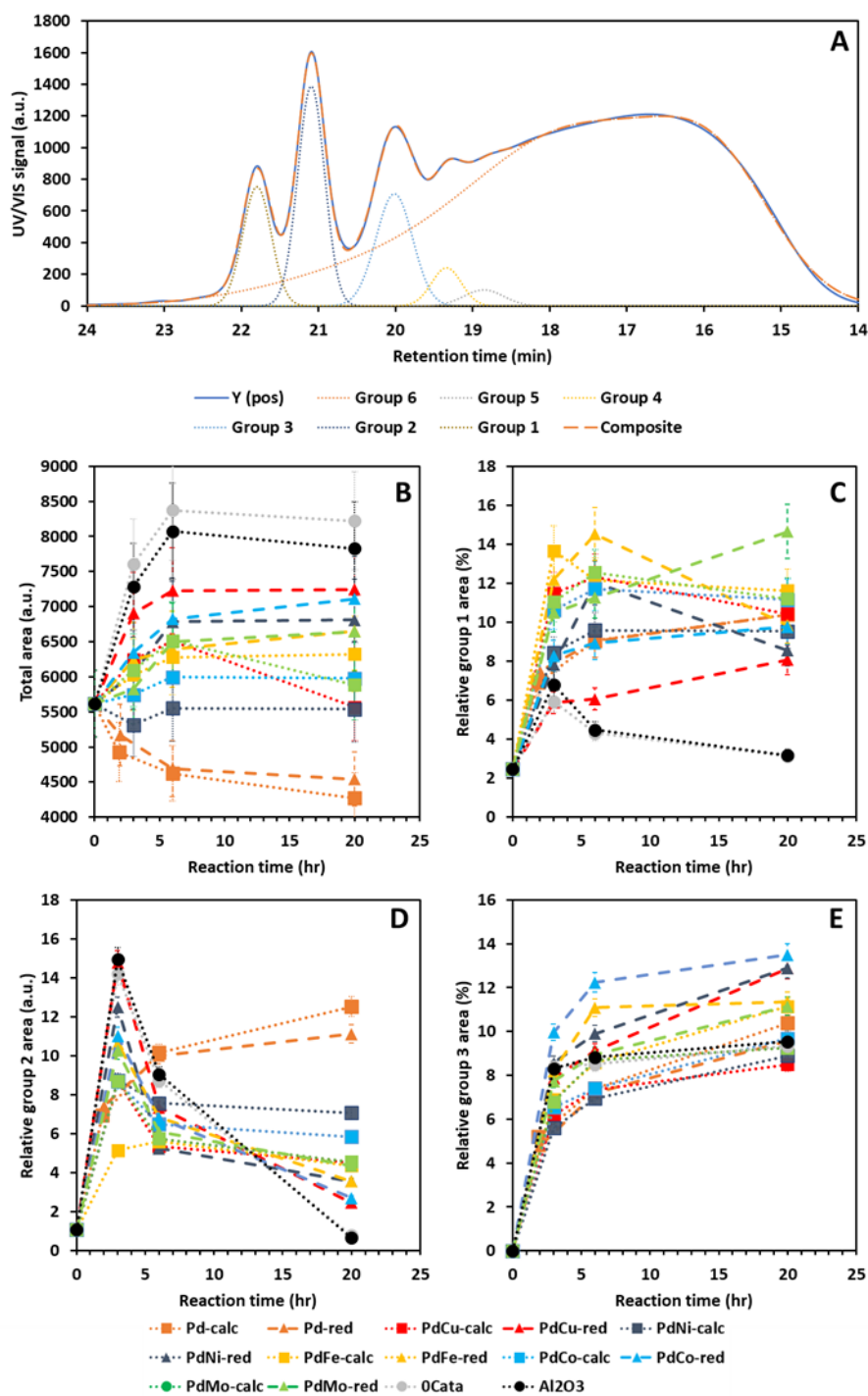




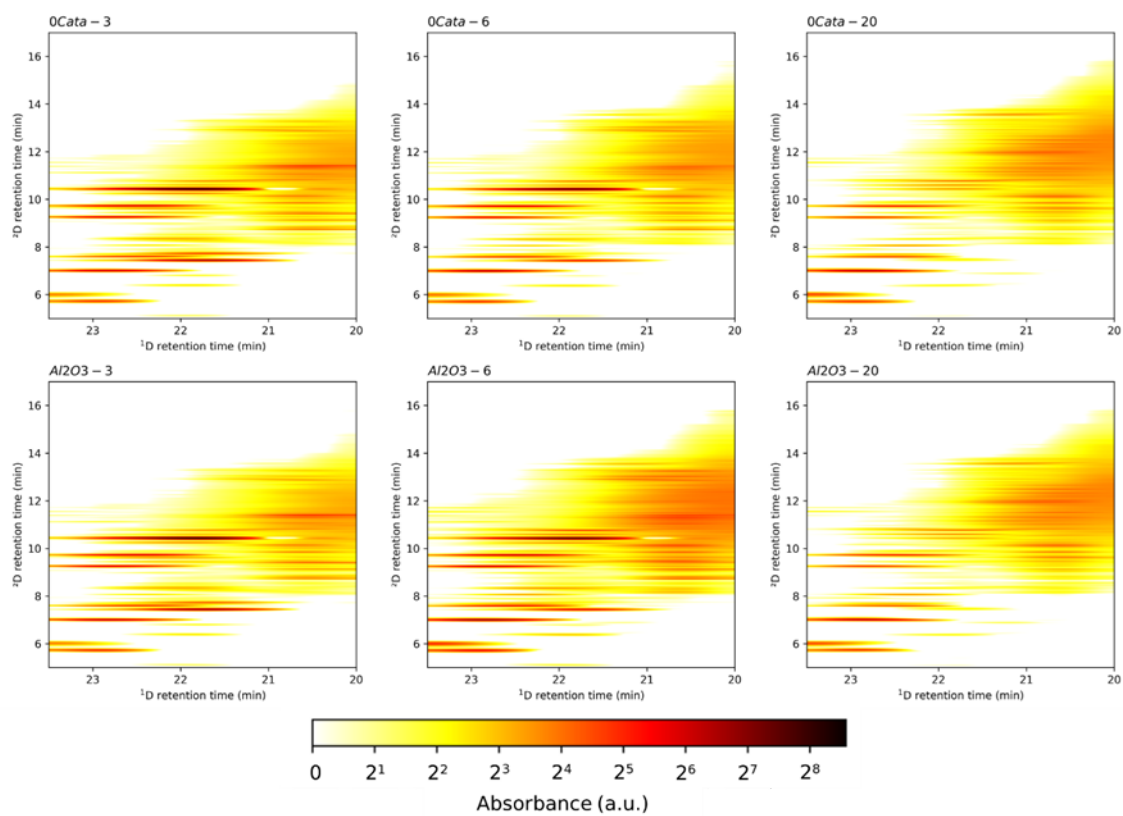
**Figure S6:** Molecular weight distribution, as determined through GPC-RID, for Cu (red), Ni (dark blue), Fe (yellow), Co (light blue) and Mo (green) as the calcined variant (left) and the reduced variant (right), in addition to 0Cata and Al<sub>2</sub>O<sub>3</sub>, after 6 h of reaction time.



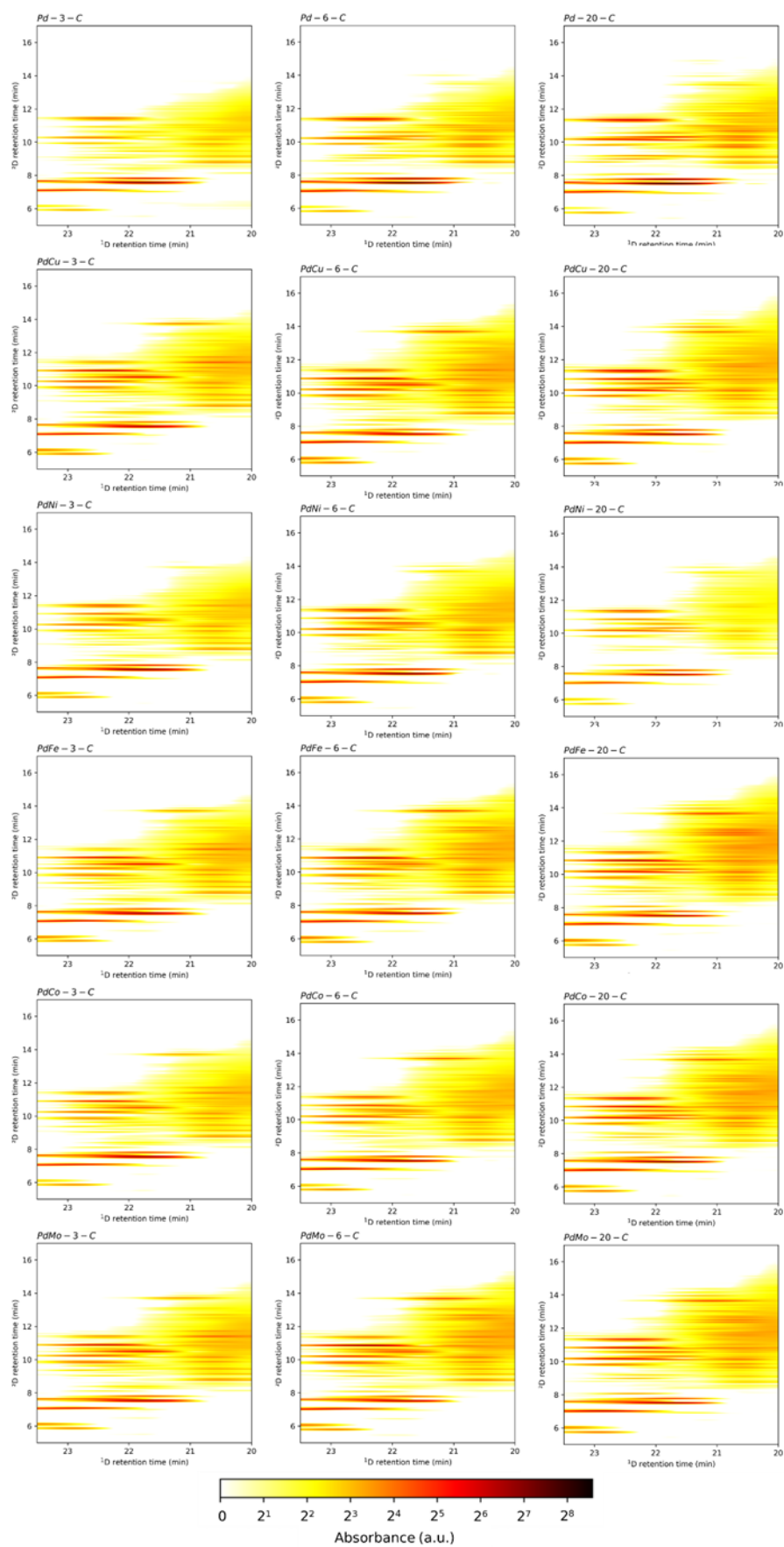
**Figure S7:** Reduction in mass averaged molar mass ( $M_w$ ) (A) and increase in polydispersity index (PDI) (B) for Pd-calc, Cu-calc, Ni-calc, Fe-calc, Co-calc, Mo-calc, Pd-red, Cu-red, Ni-red, Fe-red, Co-red, Mo-red, 0Cata and Al<sub>2</sub>O<sub>3</sub> after 6 h of reaction time. Dotted bars are used for calcined catalysts; striped bars indicate reduced catalysts.



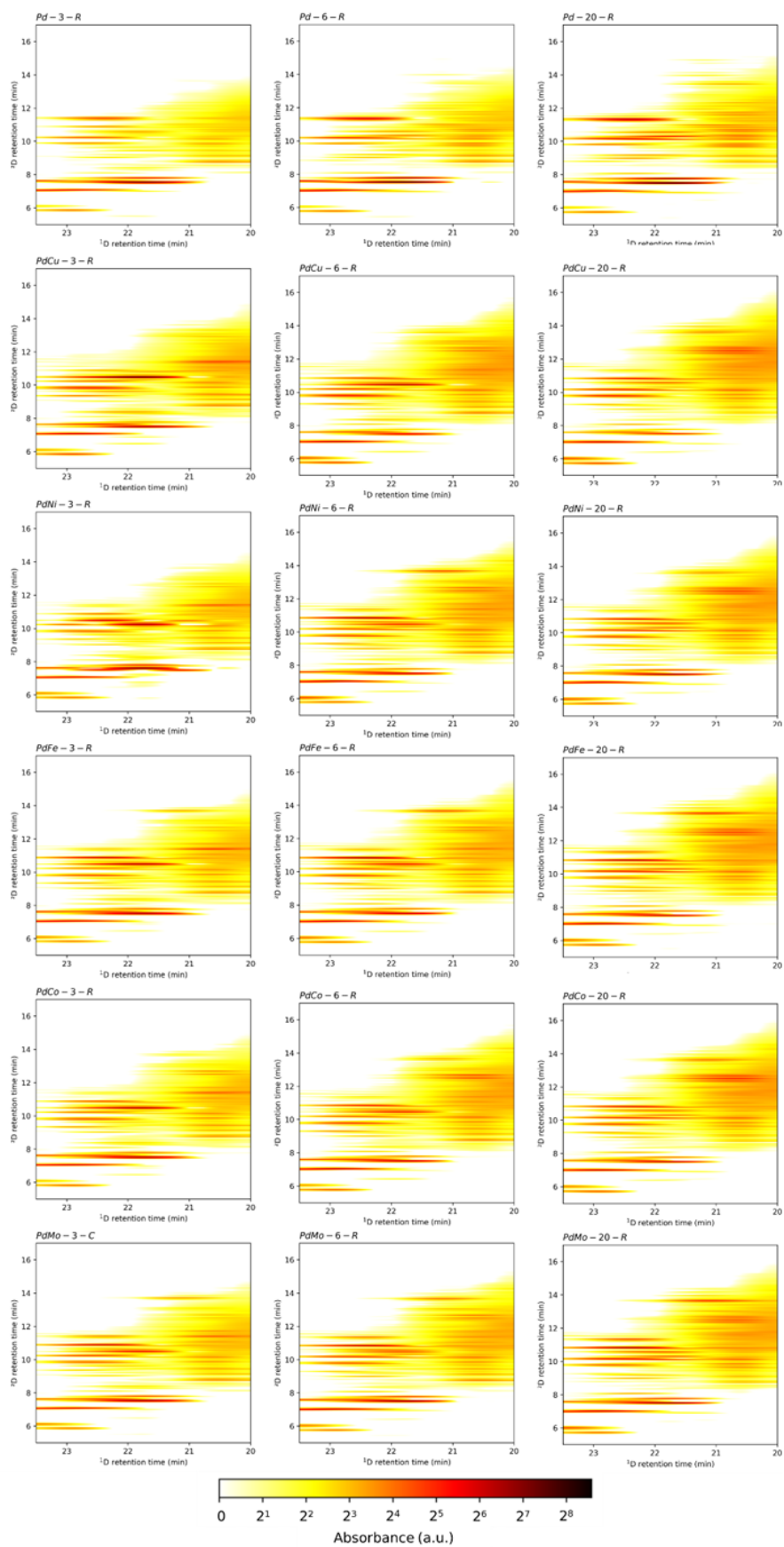
**Figure S8:** Example deconvolution of GPC-UV/VIS chromatogram (A), total surface area under GPC-UV/VIS chromatogram (B) and relative areas for groups 1-3 (C-E), as determined through deconvolution of GPC-UV/VIS, as a function of reaction time for OCata, Al<sub>2</sub>O<sub>3</sub> and all Pd(X)-calc/red variations.



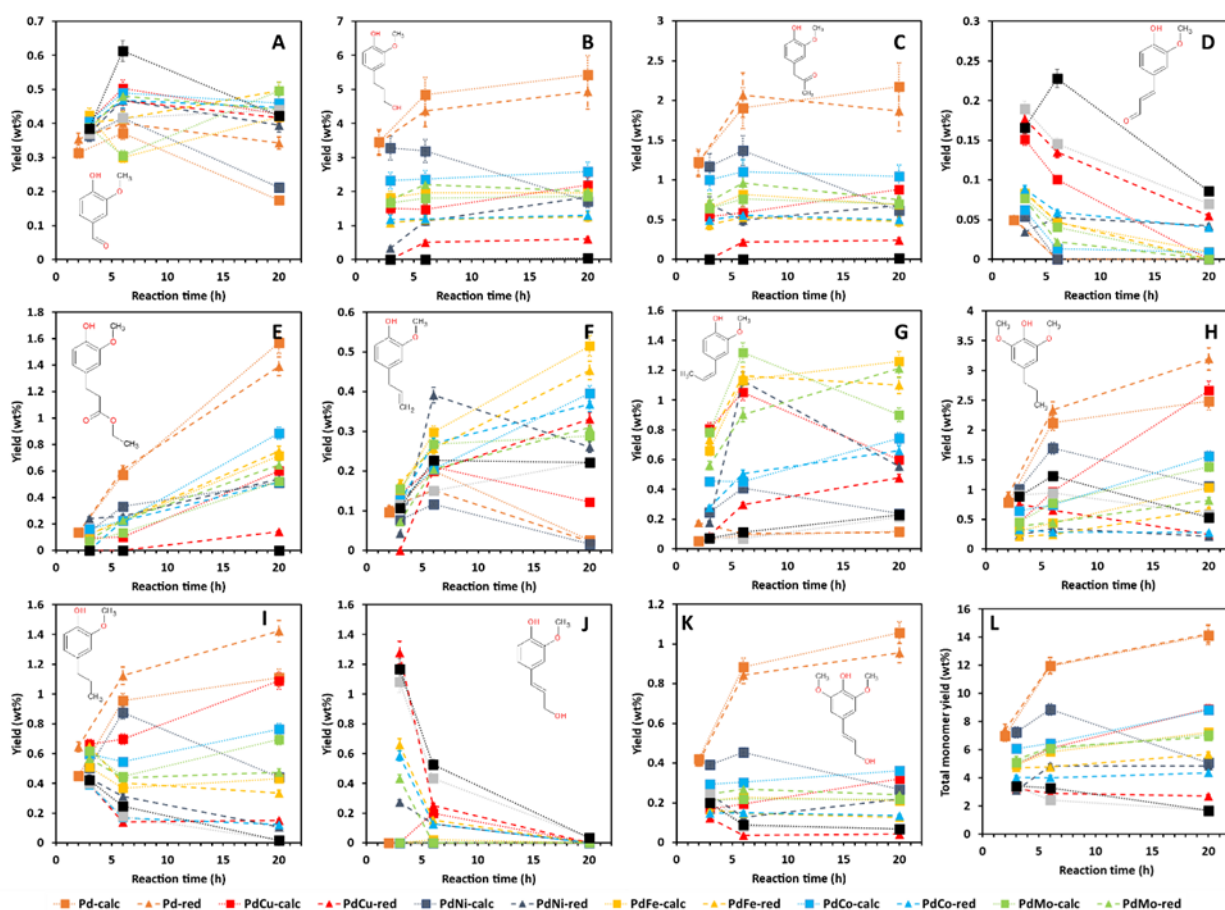
**Figure S9:** GPC-HPLC-UV/VIS heatmaps for the low molecular weight region, i.e.,  $t_R = [20;23.5]$  min, for the Ocata and Al<sub>2</sub>O<sub>3</sub> variations after 3 h (left column), 6 h (middle column) and 20 h (right column) of reaction time.



**Figure S10:** GPC-HPLC-UV/VIS heatmaps of the low molecular weight region, i.e.,  $^1t_R = [20;23.5]$  min, for the all PdX-calc catalysts at all reaction times.



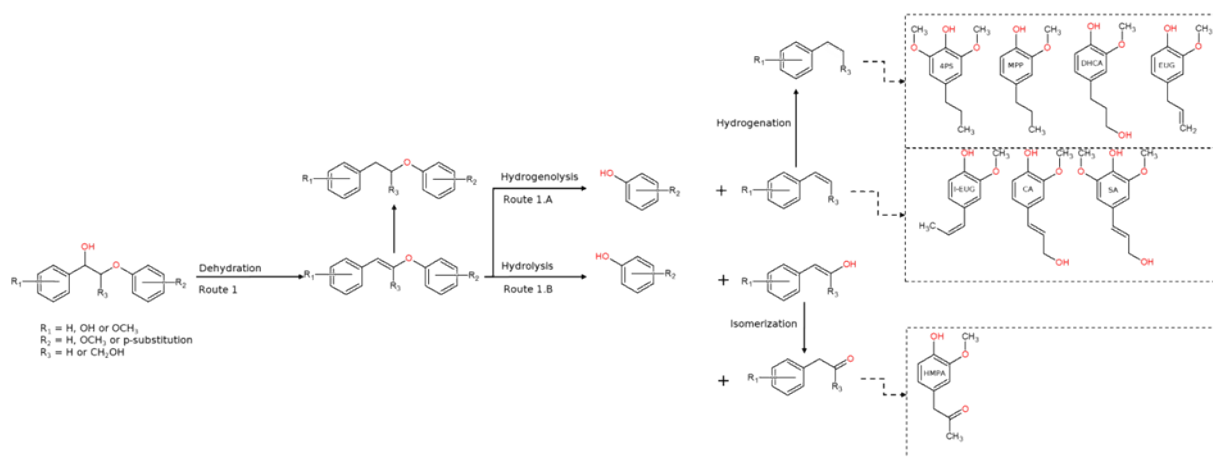
**Figure S11:** GPC-HPLC-UV/VIS heatmaps of the low molecular weight region, i.e.,  ${}^1t_R = [20;23.5]$  min, for the all PdX-red catalysts at all reaction times.



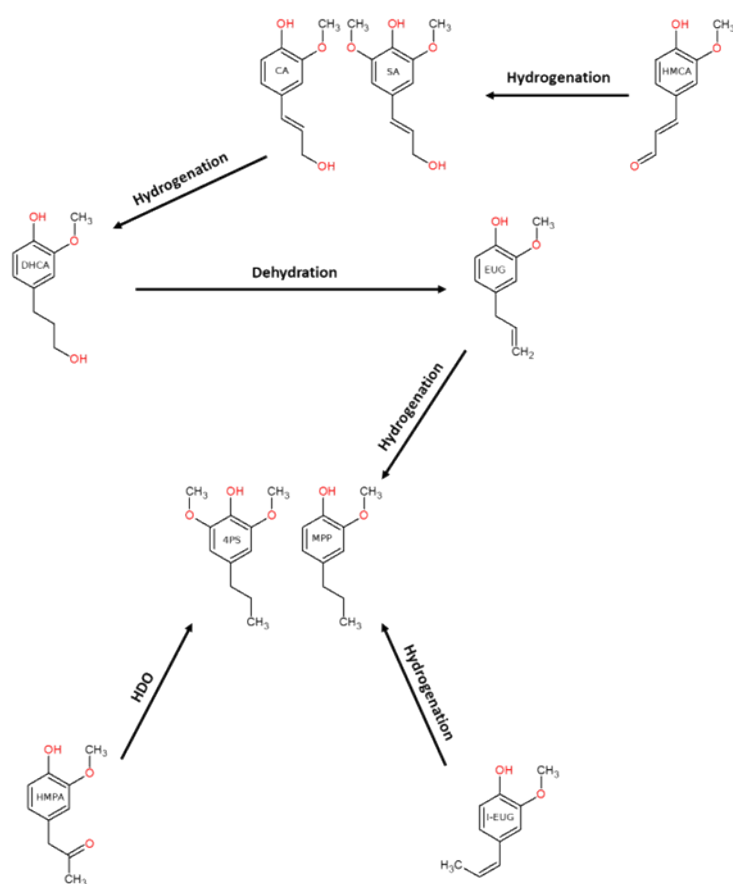
**Figure S12:** Mass yields of V (A), DHCA (B), HMPA (C), HMCA (D), EHF (E), EUG (F), I-EUG (G), 4PS (H), MPP (I), CA (J) and SA (K) and the total monomer mass yield as a function of reaction time for OCat, Al<sub>2</sub>O<sub>3</sub> and all Pd(X)-calc/red variations.

### Main pathways for cleavage of the $\beta$ -O-4 linkage

While the precise reaction mechanisms for each of the linkages within lignin are not fully identified yet in literature, model compound research has provided an indication of the main pathways for cleavage of the  $\beta$ -O-4 linkage, the most abundant linkage in native lignin structures.<sup>[1]</sup> The monomers identified and quantified within this work can be related to dehydration of the OH group on the alpha carbon ( $\alpha$ -C) within the  $\beta$ -O-4 linkage, followed by either hydrogenolysis (route 1.A) or hydrolysis (route 1.B) of the ether linkage between the  $\beta$ -C of the para substitution of one aromatic center and another aromatic center. If hydrogenolysis occurs, an aliphatic double bond is formed between the  $\alpha$ -C and  $\beta$ -C of the original  $\beta$ -O-4 linkage, resulting in monomers like isoeugenol (I-EUG), coniferyl alcohol (CA) or sinapyl alcohol (SA), depending on the presence of an aliphatic OH group on the  $\gamma$ -C. This aliphatic double bond can be further hydrogenated under the reductive environment, leading to monomers like 4-propylsyringol (4PS), 2-methoxy-4-propylphenol (MPP), dihydroconiferylalcohol (DHCA) or eugenol (EUG), depending on the presence of an aliphatic OH group on the  $\gamma$ -C and whether this group has undergone dehydration. If route 1.B, i.e., hydrolysis is followed, an aliphatic double bond between the  $\alpha$ -C and  $\beta$ -C is formed without removal of the aliphatic OH group on the  $\beta$ -C. The only monomer related to this pathway within this work is 4-hydroxy-3-methoxyphenylacetone (HMPA) wherein the aliphatic double bond is hydrogenated and the aliphatic OH group on the  $\beta$ -C is dehydrogenated.



**Figure S13:** Reaction mechanism for the cleavage of a  $\beta$ -O-4 linkage through dehydration and hydrogenolysis or hydrolysis, with indication of 4PS, MPP, DHCA, EUG, I-EUG, CA, SA and HMPA, quantified through GPC-HPLC-UV/VIS analysis.



**Figure S14:** Indication of the chemical relationships between the monomeric products quantified through GPC-HPLC-UV/VIS due to hydrodeoxygenation and/or hydrogenation reactions occurring during lignin depolymerization.

- [1] B. Zhang, Z. Qi, X. Li, J. Ji, L. Zhang, H. Wang, X. Liu, C. Li, *Green Chemistry* **2019**, *21*, 5556–5564.