

Supplementary Information

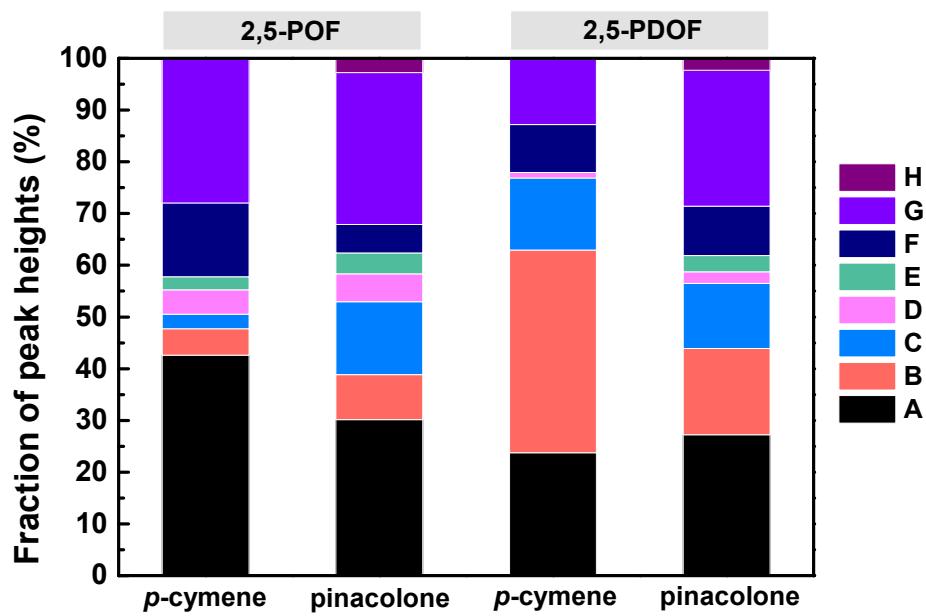
**Enzymatic Polymerization of Furan-Based Polymers in Biobased Solvents**

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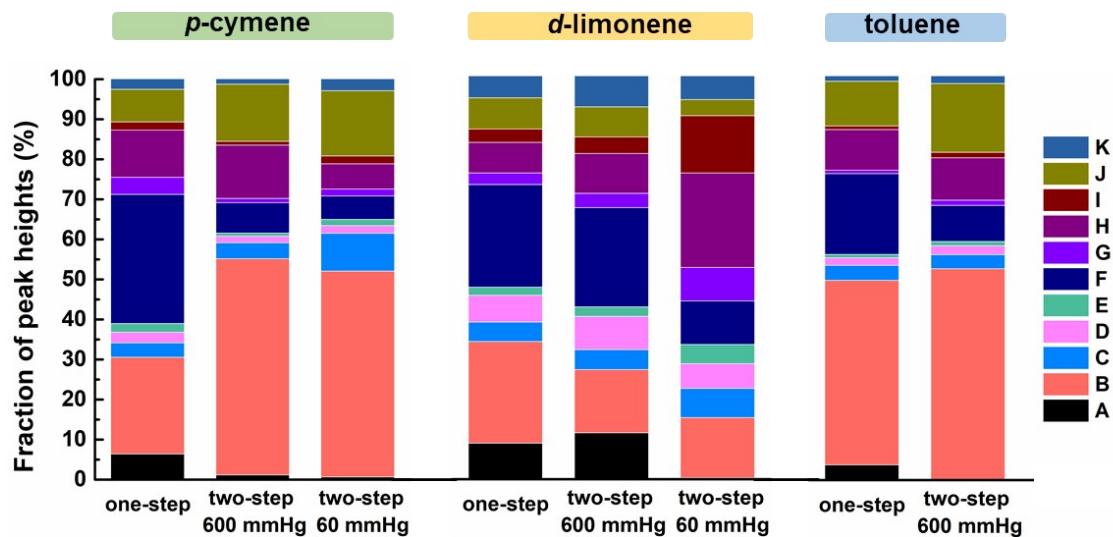
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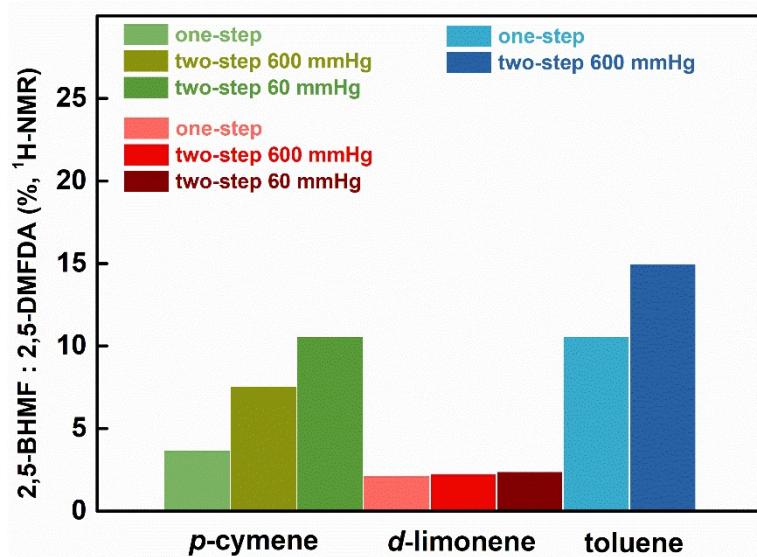
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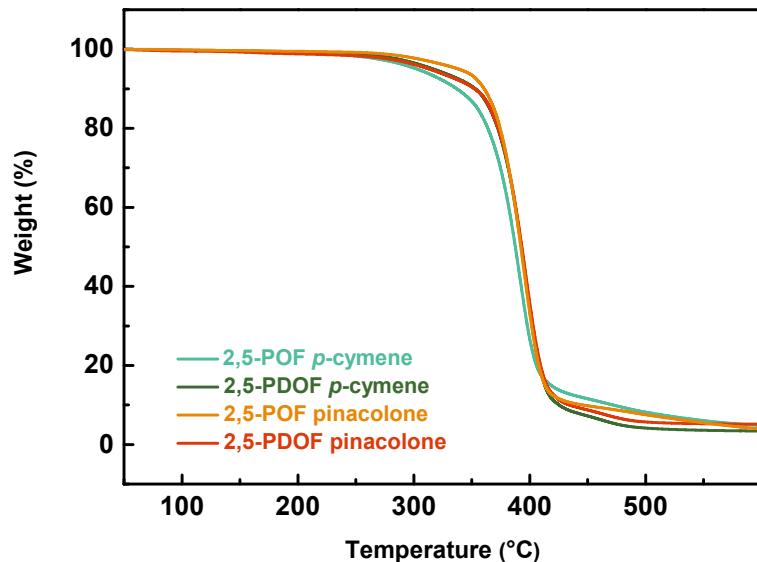
**Figure S1.** The Sum of peak height intensities of furan-based polyesters species from MALDI-ToF MS analysis and details of each species A to H are provided in Table 2.



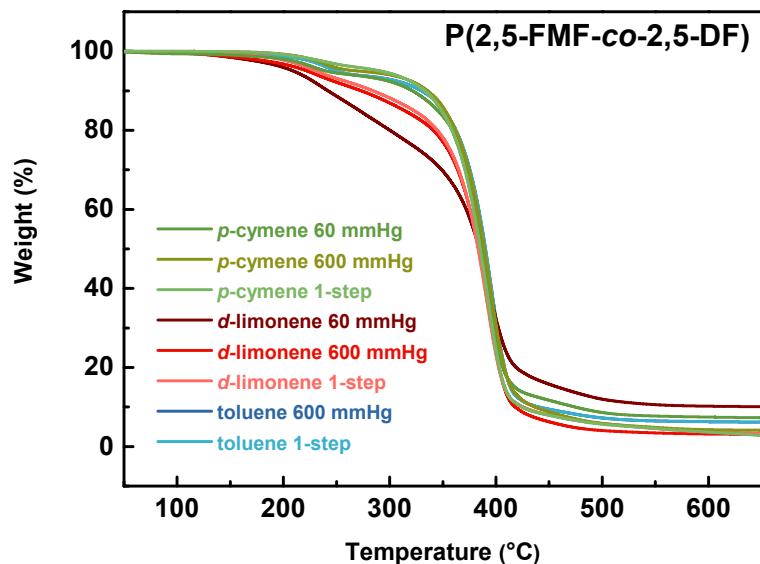
**Figure S2.** The Sum of peak height intensities of P(2,5-FMF-co-2,5-DF) species from MALDI-ToF MS analysis and details of each species A to K are provided in Table 3.



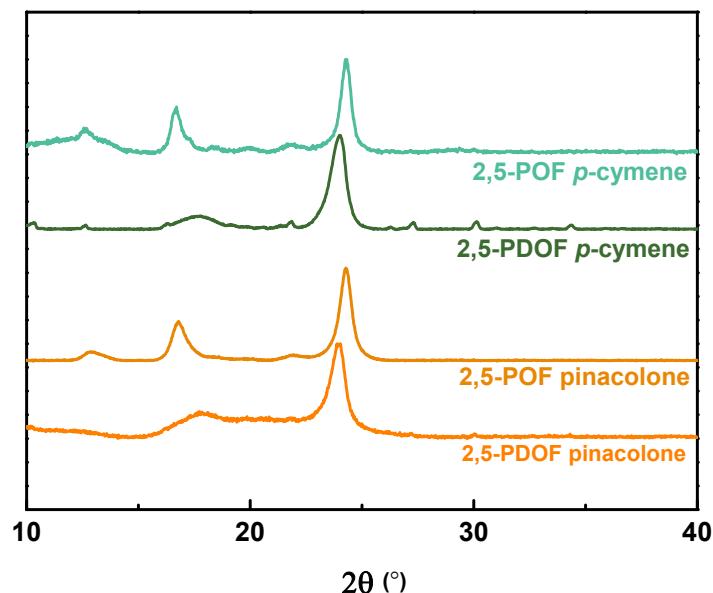
**Figure S3.** The molar fractions percentage of 2,5-BHMF comparison between P(2,5-FMF-*co*-2,5-DF) obtained in various method determined by  $^1\text{H}$ -NMR spectroscopy



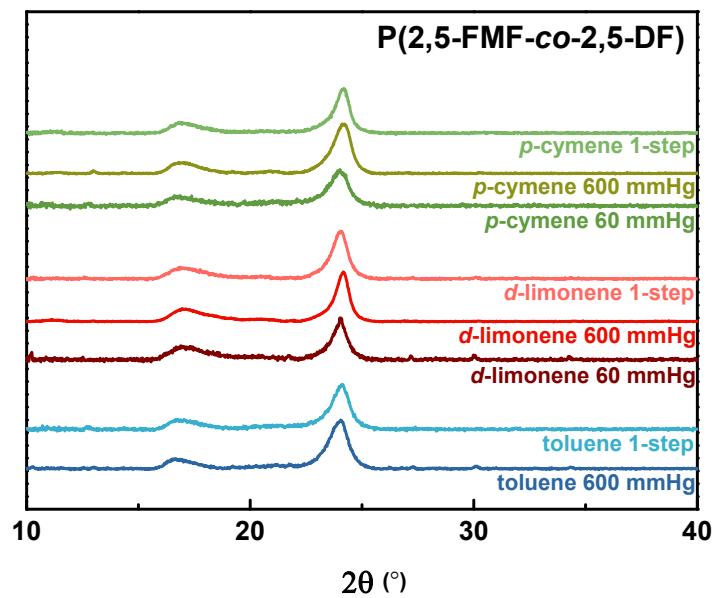
**Figure S4.** TGA curves of the obtained furan-based polyesters



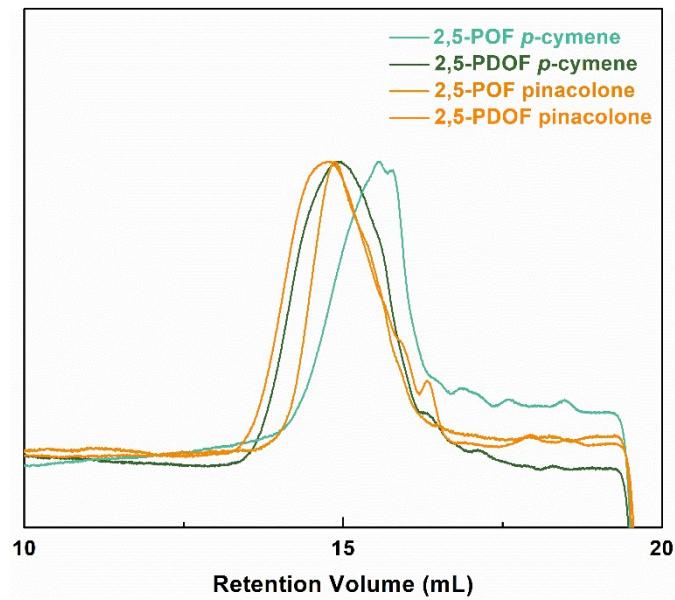
**Figure S5.** TGA curves of the obtained P(2,5-FMF-*co*-2,5-DF)



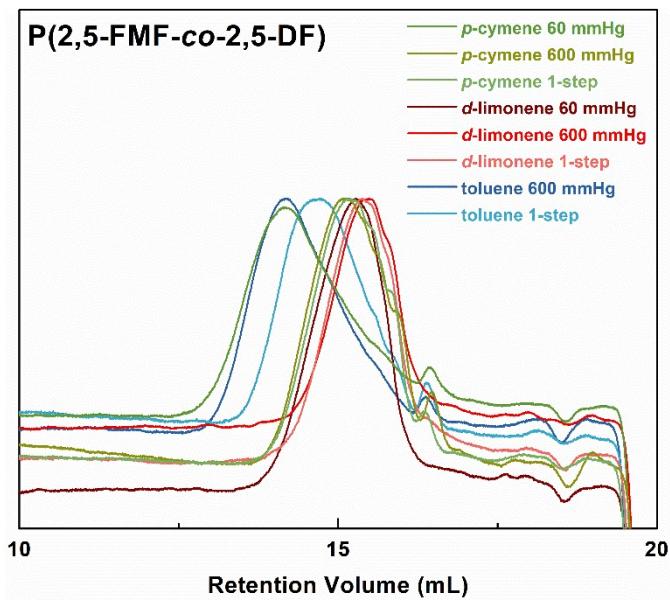
**Figure S6.** WAXD spectra of the obtained furan-based polyesters



**Figure S7.** WAXD spectra of the obtained P(2,5-FMF-*co*-2,5-DF)



**Figure S8.** SEC elugrams of the obtained furan-based polyesters



**Figure S9.** SEC elugrams of the obtained P(2,5-FMF-*co*-2,5-DF)

**Table S1.** Molecular weights and yields of the obtained furan-based polyesters in comparison with previous results.<sup>[14,39]</sup>

| x <sup>a</sup> | Polymer  | Solvent             | $\bar{M}_n^a$<br>[g mol <sup>-1</sup> ] | $\bar{M}_w^a$<br>[g mol <sup>-1</sup> ] | Yield <sup>b</sup><br>[%] |
|----------------|----------|---------------------|---|---|---------------------------|
| 8              | 2,5-POF  | p-cymene            | 3350                                    | 5000                                    | 35                        |
|                |          | Pinacolone          | 2400                                    | 4100                                    | 17                        |
|                |          | Toluene             | 4450                                    | 8200                                    | 53                        |
|                |          | BMIMPF <sub>6</sub> | 1550                                    | 2000                                    | 30                        |
|                |          | U-ChCl              | 1800                                    | 2500                                    | 35                        |
| 12             | 2,5-PDOF | p-cymene            | 4100                                    | 7000                                    | 63                        |
|                |          | Pinacolone          | 3700                                    | 6200                                    | 48                        |
|                |          | Toluene             | 5700                                    | 13300                                   | 60                        |
|                |          | BMIMPF <sub>6</sub> | 2700                                    | 5400                                    | 56                        |
|                |          | U-ChCl              | 1900                                    | 2700                                    | 41                        |

<sup>a</sup> The number-average molecular weight ( $\bar{M}_n$ ) and weight-average molecular weight ( $\bar{M}_w$ ) were determined by SEC using CHCl<sub>3</sub> as the eluent; <sup>b</sup> isolated yield.

**Table S2.** Molecular weights and yields of the obtained P(2,5-FMF-*co*-2,5-DF)

| Method                 | Solvent    | $\bar{M}_n^a$<br>[g mol <sup>-1</sup> ] | $\bar{M}_w^a$<br>[g mol <sup>-1</sup> ] | Yield <sup>b</sup><br>[%] |
|------------------------|------------|---|---|---------------------------|
| One-step               | p-cymene   | 3000                                    | 4400                                    | 42                        |
|                        | d-limonene | 2400                                    | 3500                                    | 49                        |
|                        | Toluene    | 4300                                    | 7100                                    | 48                        |
| Two-step<br>(600 mmHg) | p-cymene   | 3500                                    | 5900                                    | 47                        |
|                        | d-limonene | 2300                                    | 3400                                    | 49                        |
|                        | Toluene    | 5800                                    | 11400                                   | 54                        |
| Two-step<br>(60 mmHg)  | p-cymene   | 5600                                    | 12800                                   | 61                        |
|                        | d-limonene | 3000                                    | 4500                                    | 60                        |

<sup>a</sup> The number-average molecular weight ( $\bar{M}_n$ ) and weight-average molecular weight ( $\bar{M}_w$ ) were determined by SEC using CHCl<sub>3</sub> as the eluent; <sup>b</sup> isolated yield.

**Table S3.** Properties of selected solvents

| Solvent    | Kamlet-Taft <sup>a</sup> |         | $V_m^b$<br>(mL/mol) | Log P | References                 |
|------------|--------------------------|---------|---------------------|-------|----------------------------|
|            | $\beta$                  | $\pi^*$ |                     |       |                            |
| Toluene    | 0.12                     | 0.50    | 107.10              | 2.58  | Clark et al. <sup>44</sup> |
| p-cymene   | 0.13                     | 0.39    | 156.06              | 3.47  | Clark et al. <sup>44</sup> |
| Pinacolone | 0.58                     | 0.59    | 125.20              | 1.21  | Byrne et al. <sup>33</sup> |
| d-limonene | 0.00                     | 0.16    | 162.17              | 4.57  | Clark et al. <sup>44</sup> |

<sup>a</sup> measured via UV spectroscopy taking the wavelength of maximum absorption,  $\lambda_{\text{max}}$ , in the relevant solvent, i.e., hydrogen bond accepting ability ( $\beta$ ) using 4-nitroanisole, polarizability ( $\pi^*$ ) using N,N-diethyl-4-aniline;

<sup>b</sup> molar volume calculated as the ratio between molecular weight ( $M_w$ ) and density ( $\rho$ ) of the solvent

**Table S4.** Thermal and crystalline properties of the obtained furan-based polyesters

| Polyester | Solvent          | DSC             |                 |                 | TGA             | WAXD                         |
|-----------|------------------|-----------------|-----------------|-----------------|-----------------|------------------------------|
|           |                  | $T_g^a$<br>(°C) | $T_m^b$<br>(°C) | $T_m^b$<br>(°C) | $T_c^c$<br>(°C) | $T_{d\text{-max}}^d$<br>(°C) |
| 2,5-POF   | <i>p</i> -cymene | 10              | 138; 141        | 129; 138        | 118             | 394                          |
|           | Pinacolone       | 5               | 121; 127        | 116; 126        | 104             | 390                          |
| 2,5-PDOF  | <i>p</i> -cymene | 5               | 57; 102         | 94; 102         | 79              | 396                          |
|           | Pinacolone       | 6               | 55; 100         | 91; 100         | 76              | 396                          |

<sup>a</sup>  $T_g$  = glass transition temperature from the modulated DSC heating scan, <sup>b</sup>  $T_m$  = melting temperature from the second heating scan, <sup>c</sup>  $T_c$  = crystallization temperature from the cooling scan; <sup>d</sup>  $T_{d\text{-max}}$  = temperature at the maximum rate of decomposition; <sup>e</sup> The degree of crystallinity ( $\chi_c$ ) was calculated from WAXD

**Table S5.** Thermal and crystalline properties of the obtained P(2,5-FMF-*co*-2,5-DF

| Method               | Solvent          | DSC             |                 |                 | TGA                | WAXD            |                              |            |
|----------------------|------------------|-----------------|-----------------|-----------------|--------------------|-----------------|------------------------------|------------|
|                      |                  | $T_g^a$<br>(°C) | $T_m^b$<br>(°C) | $T_m^c$<br>(°C) | $T_{cc}^c$<br>(°C) | $T_c^d$<br>(°C) | $T_{d\text{-max}}^e$<br>(°C) | $\chi_c^f$ |
| One-step             | <i>p</i> -cymene | -5              | 64; 99          | 80; 87; 98      | - g                | 63; 73          | 238; 390                     | 37         |
|                      | (+)-limonene     | -5              | 59; 73; 91      | 72; 81; 92      | - g                | 50; 65          | 227; 390                     | 35         |
|                      | Toluene          | -2              | 60; 101         | 84; 90; 101     | - g                | 64; 70          | 239; 393                     | 35         |
| Two-step<br>600 mmHg | <i>p</i> -cymene | -1              | 60; 91; 101     | 87; 100         | - g                | 69              | 238; 393                     | 37         |
|                      | (+)-limonene     | -2              | 58; 73; 88      | 71; 89          | 25                 | 46              | 232; 393                     | 32         |
|                      | Toluene          | -3              | 55; 100         | 99              | 37                 | 53              | 239; 393                     | 35         |
| Two-step<br>60 mmHg  | <i>p</i> -cymene | -1              | 57; 99          | 87; 100         | 35                 | 58              | 227; 391                     | 37         |
|                      | (+)-limonene     | -3              | 69; 82          | 84              | - g                | 9               | 237; 393                     | 30         |

<sup>a</sup>  $T_g$  = glass transition temperature from the modulated DSC heating scan, <sup>b</sup>  $T_m$  = melting temperature from the first heating scan, <sup>c</sup>  $T_m$  = melting temperature and  $T_{cc}$  = cold crystallization temperature from the second heating scan, <sup>d</sup>  $T_c$  = crystallization temperature from the cooling scan; <sup>e</sup>  $T_{d\text{-max}}$  = temperature at the maximum rate of decomposition; <sup>f</sup> The degree of crystallinity ( $\chi_c$ ) was calculated from WAXD; <sup>g</sup> not determined