Supplementary Information

Concentration-Dependent Kinetic Study of Graphene Oxide (GO) Reduction Using Biophenol and Electrochemical Analysis

Saad Zafar[†], Sanjana Krishna Mani[†], Monisha Monisha, Bimlesh Lochab*

Materials Chemistry Laboratory, Department of Chemistry, School of Natural Sciences, Shiv Nadar Institution of Eminence, Gautam Buddha Nagar, Uttar Pradesh, 201314, India



Figure S1. pH variation of the GO and NH_4OH solution with the sequential addition of gallic acid to obtain rGO_{50} to rGO_{300} .



Figure S2. UV-vis. plot of gallic acid.

Ratio	Slope (nm/h)	
rGO ₅₀	19.25	
rGO ₇₅	28.39	
rGO ₁₀₀	21.65	
rGO ₂₀₀	12.21	
rGO ₃₀₀	8.571	

Table S1. Variation of slope in plot within the initial first hour of the reduction reaction.



Figure S3. FT-IR spectra of gallic acid.



Figure S4. (a) XPS wide survey spectra of GO, and reduced GOs, and (b) carbon to oxygen ratio evaluated from atomic weight percentage in XPS wide spectra.

Table S2. Comparative specific capacitance values reported for rGO obtained using different sustainable reducing agents.

Sample	Reducing agent	Capacity (F g ⁻¹)	Current density (A g ⁻¹)
rGO ¹	Urea	255	0.5
rGO ²	Caffeic acid	136	1.0
rGO ³	Ascorbic acid	48.3	0.1
rGO (our work)	Gallic acid	301.7	1.0



Figure S5. (a) Plot depicting specific capacitance against current density, (b) Ragone plot, and (c) EIS spectra of GO and rGO_{100} (inset shows magnified EIS spectra).

Reference:

- 1. Z. Lei, L. Lu and X. S. Zhao, *Energy Environ. Sci.*, 2012, 5, 6391-6399.
- Z. Bo, X. Shuai, S. Mao, H. Yang, J. Qian, J. Chen, J. Yan and K. Cen, *Sci. Rep.*, 2014, 4, 4684.
- 3. S. Rai, R. Bhujel, M. Khadka, R. L. Chetry, B. P. Swain and J. Biswas, *Mater. Today Chem.*, 2021, **20**, 100472.