Electronic Supplementary Information (ESI)

Crystal Reconstruction and Defects Healing Enabled High-quality Sb₂Se₃ Films for Solar Cell Applications

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Supplementary Note

Note 1

Principles of Deep-level Transient Spectroscopy

We conducted deep-level transient spectroscopy (DLTS) to detect the deep-level defects properties and the defect level is identified from DLTS signal using Fourier deconvolution algorithm. The active energy (E_a , E_C - E_T or E_T - E_V) and capture cross section of electron traps and hole traps can be calculted by the Arrhenius Equations (1) and (2),

$$ln(\tau_e v_{th,p} N_V) = \frac{E_T - E_V}{k_B} \frac{1}{T} - \ln\left(X_p \sigma_p\right) \#(1)$$

$$ln(\tau_e \nu_{th,n} N_C) = \frac{E_C - E_T}{k_B} \frac{1}{T} - \ln(X_n \sigma_n) \# (2)$$

where τ_{e} , N_{V} , N_{C} , E_{T} , E_{V} , and E_{C} are emission time constant, valence band state density, conduction band state density, trap energy level, valence band, and conduction band, respectively. $v_{th,n/p}$, $X_{n/p}$, and $\sigma_{n/p}$ are thermal velocity, entropy factor, and capture cross-section for electron and hole, respectively. E_{a} can be obtained from the slope of the corresponding line, and the σ can be obtained from the y-intercept of the lines. The trap concentration (N_{T}) can be acquired from Equation (3):

$$N_T = 2N_S \frac{\Delta C}{C_R} \#(3)$$

where $N_{\rm S}$ is the shallow donor concentration, $C_{\rm R}$ is the capacitance under reverse bias, while ΔC represents the amplitude of transient capacitance.

The product of the capture cross section and the defect density ($\sigma \times N_T$) is considered as an important parameter for the intuitive evaluation of the carrier lifetime, and the relationship follows the following Equation (4):

$$\tau_{trap} = \frac{1}{\upsilon \sigma N_T} \#(4)$$

where v is the thermal velocity of electric charge.^{1,2}

b а Tube furnace С Intensity (a.u.) Zone 2 Zone 1 Sb₂Se₃ film Sulfur powder 500 nm T1 **T2** Configuration I 10 30 40 50 20 29 (degree) d f е Tube furnace ntensity (a.u.) Zone 1 +++ Sb₂Se₃ film Sulfur powder **** 500 nm **T1** 10 20 30 40 20 (degree) 40 50 Configuration \mathbf{I}

Supplementary Figure

Fig. S1. Characterization of structure and surface morphology of the films. XRD pattern of the Sb₂Se₃ films sulfurized using b) configuration I and e) configuration II; Surface SEM images of the Sb₂Se₃ films sulfurized using b) configuration I and e) configuration II.



Fig. S2. Magnified view of the (021), (211),(221) and (301) XRD peaks of the films.



Fig. S3. Raman spectra of the Control-Sb₂Se₃ and Sulfur-Sb₂Se₃ films.



Fig. S4. The cross-sectional SEM image of the films. a) Control-Sb₂Se₃ films, b) Sulfur-Sb₂Se₃

films.





Fig. S5. UV-visible absorption characterization of the films. a) plot of $(\alpha hv)^2$ versus (hv) of

Sb₂Se₃ films. b) Bandgap value extracted from EQE spectra.



Fig. S6. Surface composition of the films. Analysis of the surface chemical states of Control-Sb₂Se₃ and Sulfur-Sb₂Se₃ films for Se 3d XPS peaks.



Fig. S7. Elementary composition of the films. a) EDX line scan of Sulfur-Sb₂Se₃; b) Atomic fraction ratio of S/(S+Se) in Sb₂Se₃ films.

Temperature/	V _{OC} /V	J _{SC} /mA cm ⁻²	FF/%	PCE/%
°C				
370	0.43	28.21	52.75	6.45
360	0.43	27.56	53.81	6.43
350	0.43	27.21	55.36	6.50

Table S1 Effect of annealing temperature on device performance

Table S2 Effect of annealing time on device performance

Time/min	V _{OC} /V	J _{SC} /mA cm ⁻²	FF/%	PCE/%

10	0.43	27.12	55.36	6.50
20	0.46	29.70	48.98	6.66
30	0.41	29.57	38.05	4.60

Table S3 Effect of the amount of sulfur on device performance

Amount/g	V _{OC} /V	$J_{ m SC}/ m mA~cm^{-2}$	FF/%	PCE/%
0.025	0.41	29.95	54.75	6.71
0.05	0.40	31.31	53.16	6.71
0.1	0.45	30.06	56.23	7.63
0.2	0.44	31.15	54.99	7.53
0.3	0.42	31.48	53.68	7.12
0.4	0.40	30.15	53.76	6.43

Table S4 Effect of the thickness on device performance

Thickness/nm	V _{OC} /V	$J_{\rm SC}/{ m mA~cm^{-2}}$	FF/%	PCE/%
400	0.42	28.81	52.55	6.38
1000	0.45	30.06	56.23	7.63
1200	0.40	31.47	56.41	7.05
1500	0.42	29.75	57.63	7.12

Temperature/	V _{OC} /V	J _{SC} /mA cm ⁻²	FF/%	PCE/%
°C				
250	0.45	30.06	56.23	7.63
275	0.40	32.19	54.02	6.90
300	0.38	30.60	56.88	6.59

Table S5 Effect of the temperature of sulfur on device performance

Table S6 The element proportion of Sulfur-Sb₂Se₃ and Control-Sb₂Se₃

Sb ₂ Se ₃ films	Se (at.%)	S (at.%)	Sb (at.%)	(Se+S)/Sb	S/(Se+S)
Sulfur-Sb ₂ Se ₃	33.0	8.6	21.5	1.93	0.21
Control-Sb ₂ Se ₃	47.5	0	26.3	1.81	0

Table S7 Parameters fitted from electrochemical impedance spectra of devices

Sample	$R_{S}(\Omega)$	$R_{REC}(\Omega)$
Control-Sb ₂ Se ₃	54.2	15411
Sulfur-Sb ₂ Se ₃	77.74	3046

References

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