

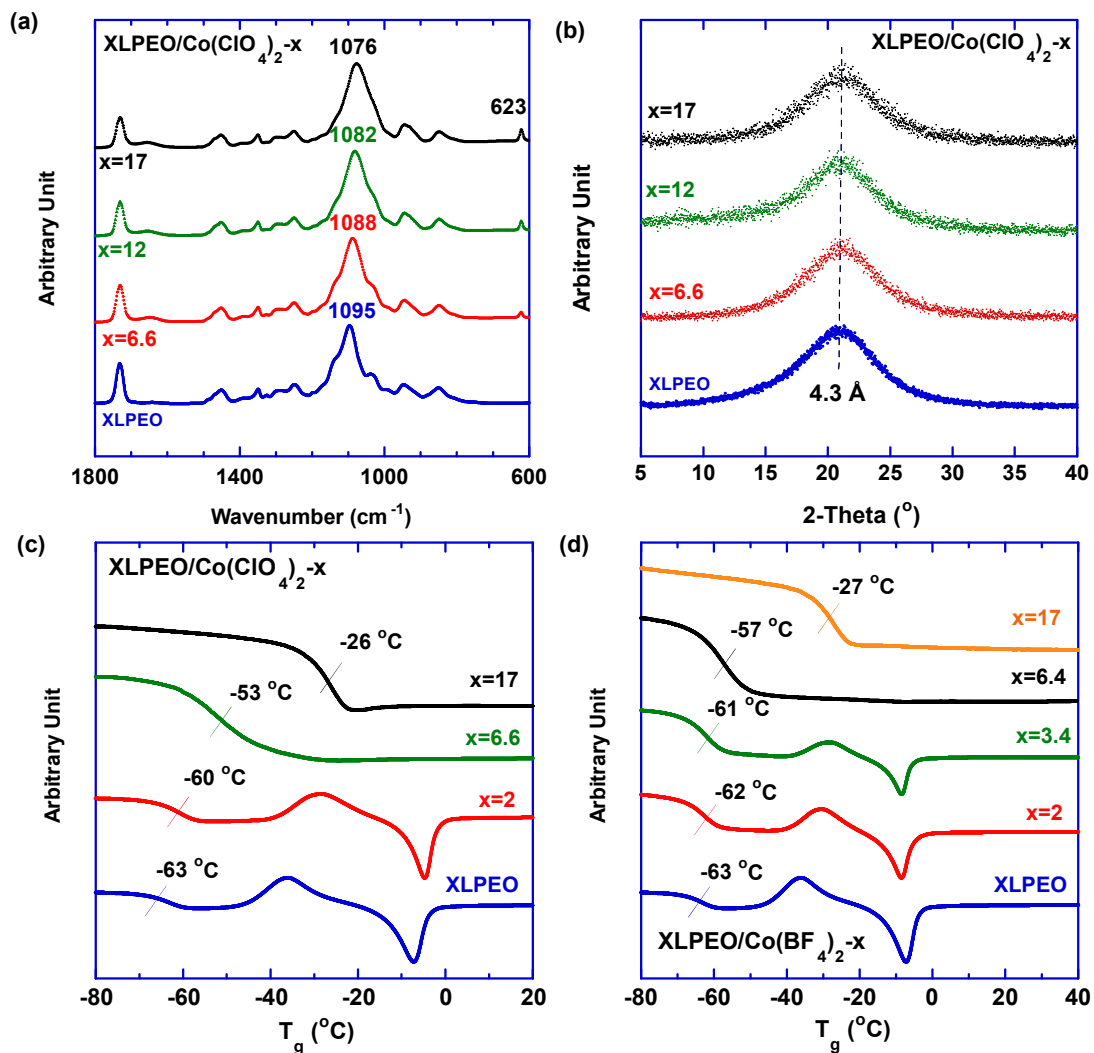
## Supporting Information

### **Retarded O<sub>2</sub> transport in Co<sup>2+</sup>-coordinated supramolecular polymer networks for membrane CO<sub>2</sub>/O<sub>2</sub> separations**

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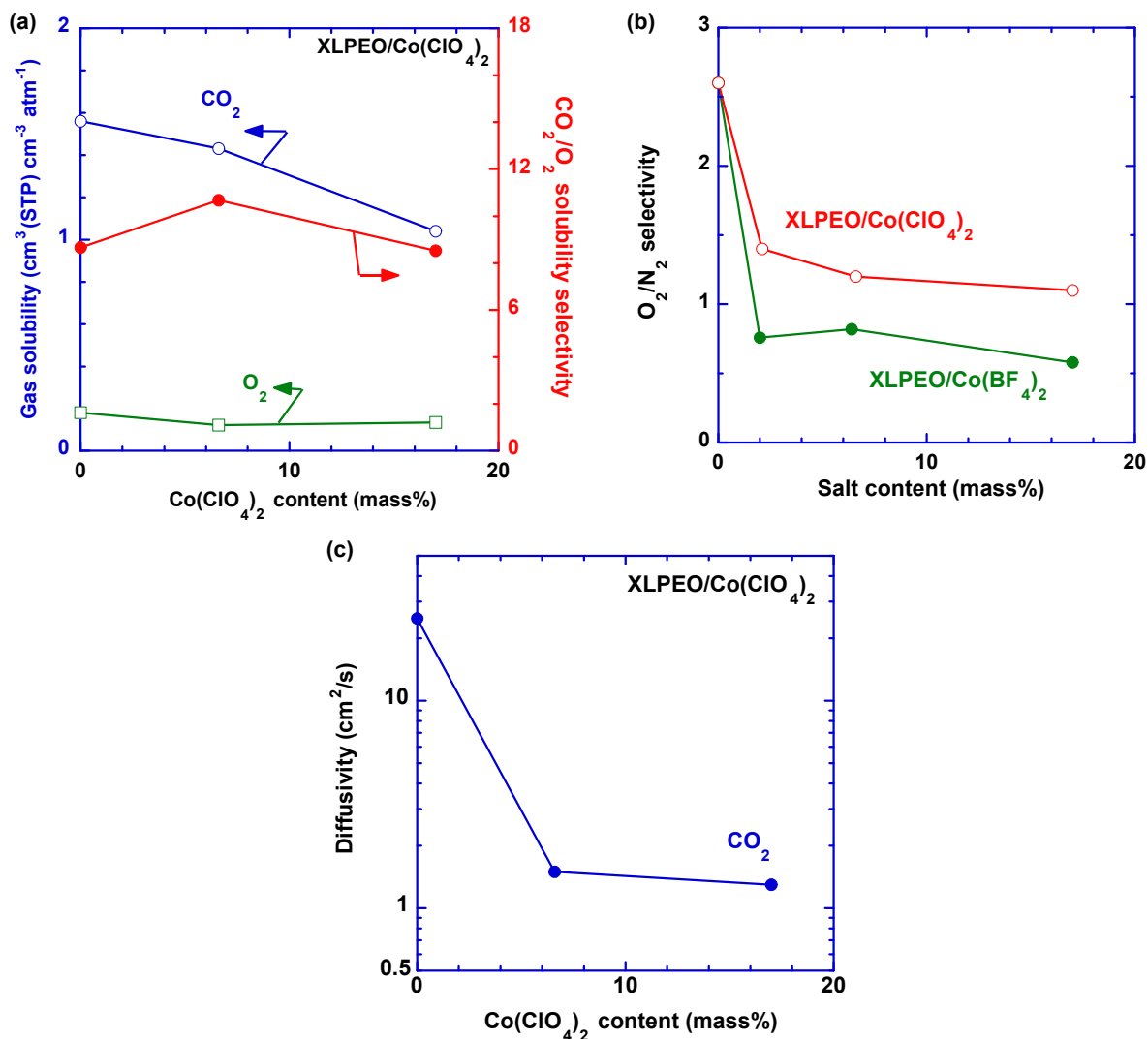
**Fig. S1.** (a) FTIR spectra, (b) XRD patterns, and (c) DSC thermograms of XLPEO/Co(ClO<sub>4</sub>)<sub>2</sub> SPNs. (d) DSC thermograms of XLPEO/Co(BF<sub>4</sub>)<sub>2</sub> SPNs.

**Table S1.** Physical properties, including  $r$  (molar ratio of ether oxygens to cations),  $\rho_{PE}$ ,  $T_g$ , and pure-gas permeability and selectivity at 35 °C in XLPEO/Co(ClO<sub>4</sub>)<sub>2-x</sub>.

$x$ (mass%)	$r$	$\rho_{PE}$ (g/cm <sup>3</sup> )	$T_g$ (°C)	$P_A$ (Barrer)			Selectivity		
				N <sub>2</sub>	O <sub>2</sub>	CO <sub>2</sub>	CO <sub>2</sub> /N <sub>2</sub>	CO <sub>2</sub> /O <sub>2</sub>	O <sub>2</sub> /N <sub>2</sub>
0	0	1.149	-63	10	26	510	51	20	2.6
2	273	1.167	-60	8.1	11	140	17	13	1.4
6.6	83	1.207	-53	1.9	2.3	29	15	13	1.2
17	34	1.304	-26	0.38	0.43	18	46	37	1.1

**Table S2.** Comparison of  $a$  values for various polymer/salt networks.

Salts in SPNs	$a$ (10 <sup>-4</sup> m <sup>3</sup> kmol <sup>-1</sup> K <sup>-1</sup> )	Ref.
Co salts	8.1	This work
LiClO <sub>4</sub>	2.6	1
Ni(BF <sub>4</sub> ) <sub>2</sub>	9.3	1
Cu(BF <sub>4</sub> ) <sub>2</sub>	3.2	1



**Fig. S2.** (a) Effect of the Co(ClO<sub>4</sub>)<sub>2</sub> and Co(BF<sub>4</sub>)<sub>2</sub> content in XLPEO on O<sub>2</sub>/N<sub>2</sub> selectivity at 35 °C and 50 psig. (b) CO<sub>2</sub> and O<sub>2</sub> solubility and CO<sub>2</sub>/O<sub>2</sub> solubility selectivity and (c) CO<sub>2</sub> diffusivity as a function of Co(ClO<sub>4</sub>)<sub>2</sub> loading. Gas solubility ( $S_A$ , cm<sup>3</sup> (STP) cm<sup>-3</sup> atm<sup>-1</sup>) at any equilibrium pressure ( $p_A$ , atm) is calculated by  $S_A = C_A/p_A$ , where  $C_A$  is the gas concentration in the polymer (cm<sup>3</sup> (STP) cm<sup>-3</sup>).

**Table S3.** Pure-gas solubility and diffusivity in XLPEO/Co salts at 35 °C.

SPNs	$S_A$		$S_{CO_2}/S_{O_2}$	$D_A$ ( $10^{-7}$ cm <sup>2</sup> /s)		$D_{CO_2}/D_{O_2}$
	CO <sub>2</sub>	O <sub>2</sub>		CO <sub>2</sub>	O <sub>2</sub>	
XLPEO	1.5	0.19	8.2	26	11	2.5
XLPEO/Co(BF <sub>4</sub> ) <sub>2</sub> -6.4	1.3	6.7	0.20	0.83	0.0048	173
XLPEO/Co(BF <sub>4</sub> ) <sub>2</sub> -17	0.93	10	0.090	0.69	0.0011	650
XLPEO/Co(ClO <sub>4</sub> ) <sub>2</sub> -6.6	1.4	0.12	11	1.5	1.5	1.0
XLPEO/Co(ClO <sub>4</sub> ) <sub>2</sub> -17	1.0	0.14	8.5	1.3	0.23	5.7

$S_A$ : cm<sup>3</sup>(STP) cm<sup>-3</sup> atm<sup>-1</sup>.

**Table S4.** Pure-gas permeability in XLPEO/Co(BF<sub>4</sub>)<sub>2</sub>-5 at 35 °C during a continuous test. The film had a thickness of ~300 μm, and it was kept under vacuum for ~4 months before the test. The films were tested with CO<sub>2</sub> at 50 psig and O<sub>2</sub> at 100 psig alternatively. The films were exposed to a vacuum for at least 8 h during the gas change.

	CO <sub>2</sub>	O <sub>2</sub>	CO <sub>2</sub>		O <sub>2</sub>
	Day 1	Day 2	Day 3	Day 6	Day 8
CO <sub>2</sub> permeability (Barrer)	75		15	12	
O <sub>2</sub> permeability (Barrer)		0.74			0.25
CO <sub>2</sub> /O <sub>2</sub> selectivity		100			48

**Reference**

1. T. Alebrahim, A. Chakraborty, L. Hu, S. Patil, S. Cheng, D. Acharya, C. M. Doherty, A. J. Hill, T. R. Cook and H. Lin, *J. Membr. Sci.*, 2022, **644**, 120063.